



A short route to functionalized imidazo[4,5-*c*]carbazoles. Synthesis of the first example of the imidazo[4,5-*c*]β-carboline ring system

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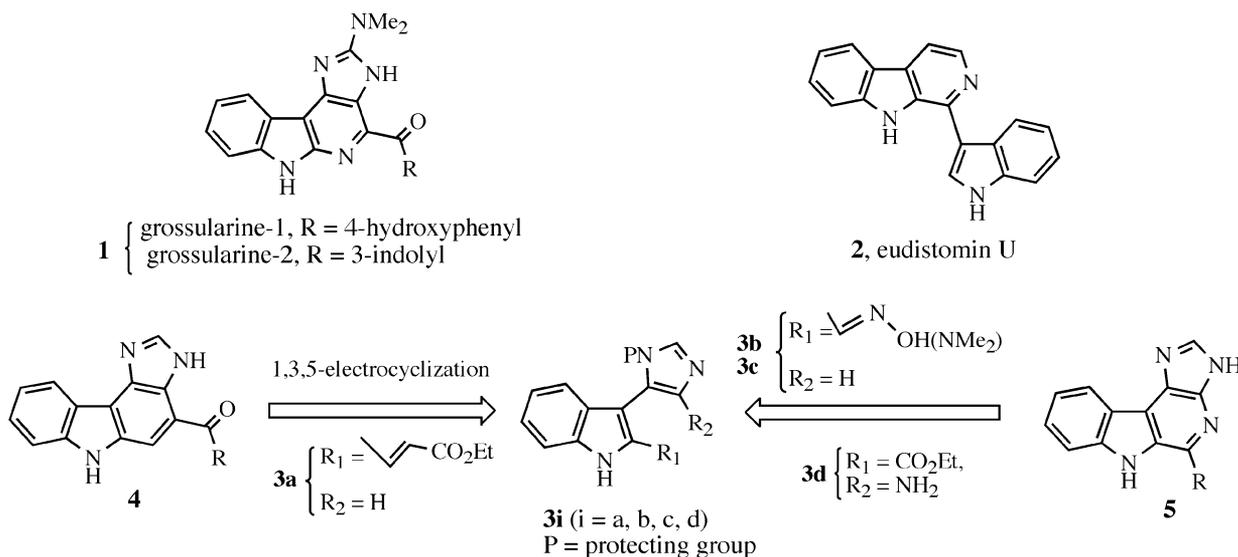
Abstract—A new synthetic route to functionalized imidazo[4,5-*c*]carbazoles (**4**) via intramolecular electrocyclization of indolo-1,3,5-hexatriene system (**3a**) is described. Thermally induced ring-closure of 2-ethylcarboxylate-(4-amino-5-imidazolyl)-3-indole (**3d**) led to the previously unknown imidazo[4,5-*c*]β-carboline ring system (**5**). These heterocycles were efficiently converted into analogs of both the marine cytotoxic agents grossularines-1 and -2 (**1**) and the antimicrobial alkaloid eudistomin U (**2**). © 2001 Elsevier Science Ltd. All rights reserved.

The carbazole and β-carboline moieties are important structural subunits which occur as components of many biologically interesting, natural as well as non-natural, compounds.¹

As part of an investigation into the synthesis of structurally related derivatives of the marine cytotoxic α-carbolines, grossularines-1 and -2² (**1**, Scheme 1), we were

interested in developing a ready access to carbazole and β-carboline containing analogs of both these natural products and the antimicrobial alkaloid eudistomin U (**2**).³

Previous synthesis of imidazo[4,5-*c*]carbazole⁴ has made use of a 3-amino-carbazole derivative as starting material, which was elaborated further into the target



Scheme 1.

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heterocycle via regioselective nitration and subsequent imidazole ring construction. In recent years, the thermal- and photo-induced electrocyclizations of 1,3,5- or 1-aza-3,5-hexatrienes have provided an efficient alternative method for accessing the carbazole and β -carboline ring systems,⁵ respectively. We report herein the application of such a methodology to the 2-acrylate-3-imidazolylindole derivative **3a** en route to the corresponding imidazolocarbazole **4** (Scheme 1).

The conversion of 2-vinyl-3-imidazoloindole **9** into carbazole **10** was first attempted. Access to compound **9** involved Stille cross-coupling of 2-formyl-3-iodoindole⁶ **6** with imidazolostannane **7** by a route similar to that we reported previously⁷ to furnish the expected imidazoloindole **8** in good yield.

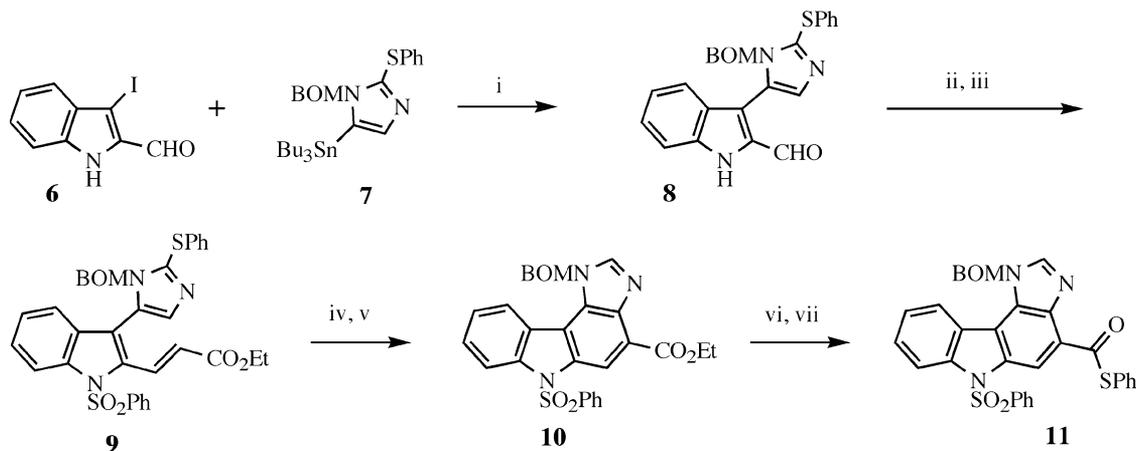
Wittig reaction of aldehyde **8** gave the 2-vinyl compound **9** which in turn was transformed, in good yield, into the corresponding carbazole **10** through a one-pot two-step process involving electrocyclization followed by in situ aromatization with MnO_2 and Raney-nickel desulfurization (Scheme 2).

Finally, conversion of ester **10** into the requisite ketones **14** and **17** was carried out by using the thiolester-

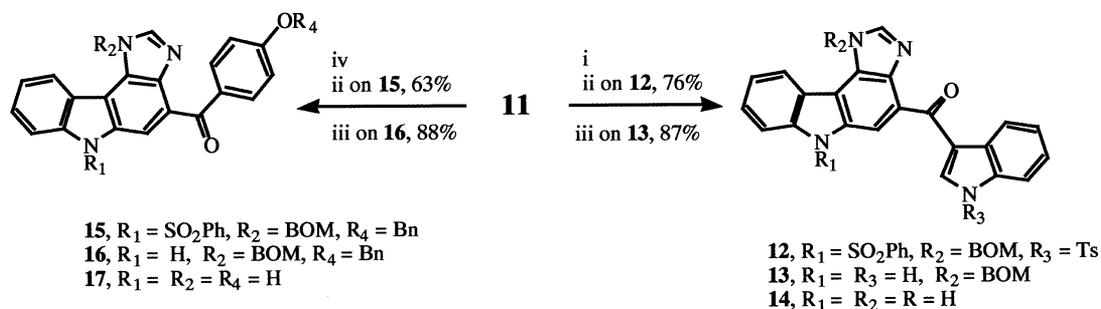
boronic acid cross-coupling approach⁸ recently reported by Liebeskind and Srogl.

To this end, thiol ester **11** was obtained in a straightforward manner by saponification of **10** and treatment of the resulting acid with carbonyl-diimidazole and thiophenol.⁹ The thiol ester **11** was then subjected to palladium-catalyzed cross-coupling with the known 1-tosyl-3-indoleboronic acid¹⁰ **31** (Scheme 4) to provide ketone **12** in a modest 55% yield. Deprotection of **12** under standard conditions furnished the analog **14** of grossularine-2 (**1**) in good yield. The analog **17** of grossularine-1 (**1**) was similarly obtained through cross-coupling with 4-benzyloxyphenylboronic acid¹¹ **28** and subsequent deprotection of ketone **15** (Scheme 3).

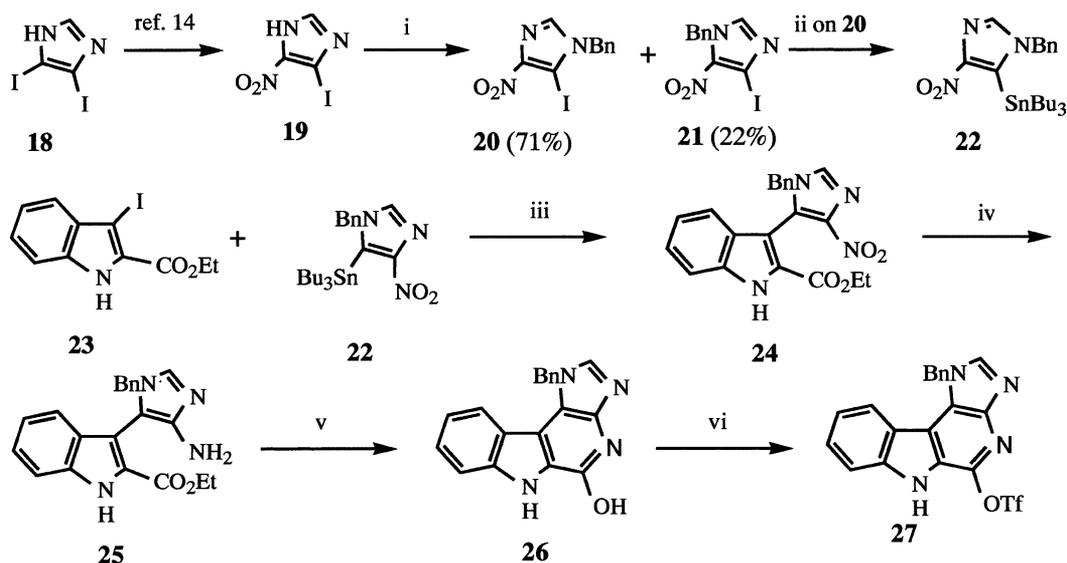
With carbazoles **14** and **17** in hand, we then turned our attention to the synthesis of their β -carboline counterparts. Our synthetic efforts toward this end focused initially on the transformation of oxime **3b** ($\text{R}_1 = \text{CHNOH}$, $\text{R}_2 = \text{H}$) and hydrazone **3c** ($\text{R}_1 = \text{HCNNMe}_2$, $\text{R}_2 = \text{H}$), respectively, into the β -carboline ring-system through thermal electrocyclization.^{12,13} Compounds **3b** and **3c** failed, however, to undergo thermal electrocyclization, suffering instead extensive decomposition.



Scheme 2. Reagents and conditions: (i) **7** (1.5 equiv.), $\text{PdCl}_2(\text{PPh}_3)_2$ (6 mol%), DMF, 120°C, 2.5 h, 75–80%; (ii) NaH (1.3 equiv.), THF, $(\text{EtO})_2\text{P}(\text{O})\text{CH}_2\text{CO}_2\text{Et}$, 0°C, 15 min, add **8** (0.55 equiv.), 0°C then rt, 2 h, 93%; (iii) NaH (1.1 equiv.), DMF, 0°C, 30 min, PhSO_2Cl (1.2 equiv.), 0°C then rt, 2 h, 71%; (iv) *o*- Cl_2Ph , reflux, 1 h, then add MnO_2 (8 equiv.), reflux, 1 h, 72%; (v) Raney-Ni (xs), EtOH–THF (2:1), reflux, 2 h, 84%; (vi) LiOH (7 equiv.), THF:H₂O:MeOH (5:2:1), rt, 5 h, 91%; (vii) CDI (1.9 equiv.), DMF, –10°C, 2 h, then PhSH (xs), –10°C, 2 h, 58%.



Scheme 3. Reagents and conditions: (i) **31** (1.1 equiv.), CuTc (1.5 equiv.), $\text{Pd}_2\text{dba}_3\cdot\text{CHCl}_3$ (1.2 mol%), trifurylphosphine (10 mol%), THF, 50°C, 18 h, 55%; (ii) 10% aq. KOH, THF:EtOH (2:1), reflux, 4 h; (iii) Pd/C (10%), $\text{HCO}_2\text{-NH}_4^+$, EtOH, reflux, 3 h; (iv) **28** (2 equiv.), CuTc (3 equiv.), $\text{Pd}_2\text{dba}_3\cdot\text{CHCl}_3$ (3 mol%), trifurylphosphine (26 mol%), THF, 50°C, 18 h, 60%.



Scheme 4. Reagents and conditions: (i) K_2CO_3 , MeCN, BnBr, reflux, 3 h; (ii) $(n\text{-Bu}_3\text{Sn})_2$, $\text{PdCl}_2(\text{PPh}_3)_2$, DMF, 120°C , 2 h, 57%; (iii) **23** (1.2 equiv.), $\text{Pd}(\text{PPh}_3)_4$ (4 mol%), CuI (8 mol%), DMF, 80%, 2.5 h, 62%; (iv) Raney-Ni, H_2 , MeOH, rt, 2.5 h, 64%; (v) $o\text{-Cl}_2\text{C}_6\text{H}_4$, reflux, 48 h, 79%; (vi) Tf_2O , pyridine, 0°C , 30 min, rt, overnight, 87%.

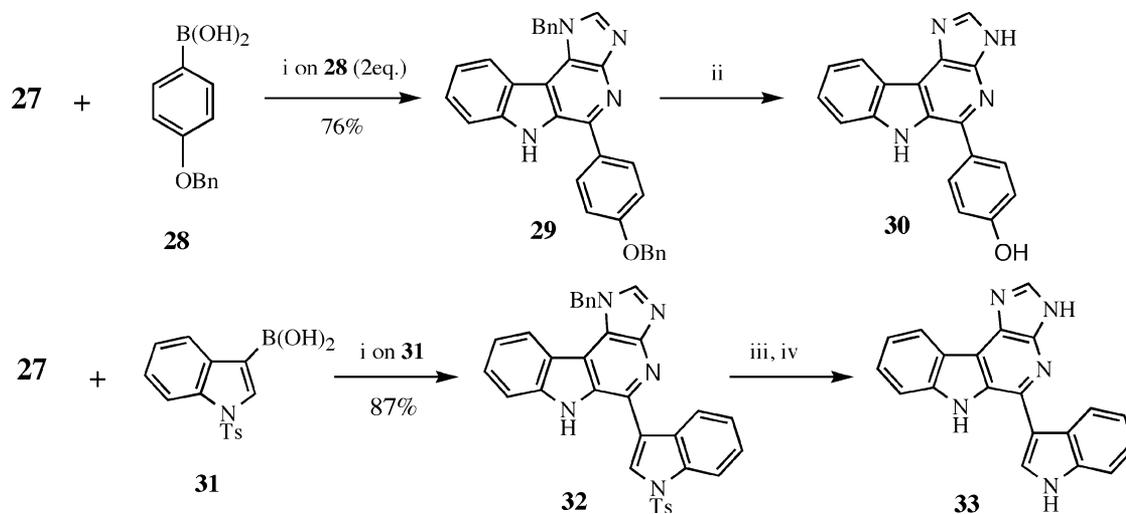
Accordingly, an alternative access to the imidazo- β -carboline ring system was sought, based on thermally-induced ring closure of 2-ethylcarboxylate-3-(4-amino-5-imidazolyl)-indole **25** (Scheme 4). Synthesis of **24**, the precursor of **25**, was envisaged through coupling of imidazolostannane **22** with 3-iodoindole **23** (Scheme 4).

Protection of the known (4,5)-iodonitroimidazole¹⁴ **19** by way of *N*-benzyl derivatives formation¹⁵ led to a mixture of compounds **20** and **21**, the purification of which allowed efficient recovery of isomer **20** subsequently transformed into stannane **22**.

Cross-coupling of stannane **22** with iodindole **23** furnished the expected 2-ethylcarboxylate-3-(4-nitro-5-imidazolyl)indole **24**,

the subsequent reduction of which gave 4-amino derivative **25**. Intramolecular ring-closure of **25** required prolonged heating in refluxing *o*-dichlorobenzene to provide pyridone **26**, which was then transformed into triflate **27** upon exposure to triflic anhydride in pyridine. Coupling of triflate **27** with boronic acids **28** and **31** provided the analogs **29** and **32**, which upon routine deprotection furnished the analogs **30** and **33** of eudistomin U (**2**) (Scheme 5).¹⁶

In the course of this work we have been able to develop a new short approach toward the functionalized imidazo[4,5-*c*]carbazoles **10** and have, as well, achieved the first synthesis of the previously unknown imidazo- β -carboline ring system **26**. In addition, these heterocycles have been successfully converted into the potentially



Scheme 5. Reagents and conditions: (i) $\text{Pd}(\text{PPh}_3)_4$ (8 mol%), K_2CO_3 (2 equiv.) in H_2O , LiCl (2 equiv.), dioxane, reflux, 2–3 h; (ii) $\text{HCO}_2\text{-NH}_4^+$, Pd/C (10%), EtOH, reflux, 3 h, 92%; (iii) 10% aq. KOH, EtOH:THF (2:1), reflux, 2.5 h, 67%; (iv) ii on **32**, reflux, 8 h, 77%.

interesting analogs (**14**, **17**, **30**, **31**) of such biologically active natural products as grossularine-1, grossularine-2 and eudistomin U.

Acknowledgements

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References

1. (a) Gribble, G. W.; Berthel, S. J. In *Studies in Natural Products Chemistry*; Atta-Ur-Rahman, Ed.; Elsevier, 1993; Vol. 12, pp. 365–409; (b) Bhattacharyya, P.; Chakraborty, D. P. *Prog. Chem. Org. Nat. Prod.* **1987**, *52*, 159–209; (c) Chakraborty, D. P.; Shyamali, R. *Prog. Chem. Org. Nat. Prod.* **1991**, *57*, 71–152; (d) Suzuki, H.; Unemoto, M.; Hagiwara, M.; Ohyama, T.; Yokoyama, Y.; Murakami, Y. *J. Chem. Soc., Perkin Trans. 1* **1999**, 1717–1723; and related references cited therein; (e) Choshi, T.; Sada, T.; Fujimoto, H.; Nagayama, C.; Sugino, E.; Hibino, S. *J. Org. Chem.* **1997**, *62*, 2535–2543; (f) Link, J. T.; Raghavan, S.; Danishefsky, S. *J. Am. Chem. Soc.* **1995**, *117*, 552–553.
2. Moquin-Patey, C.; Guyot, M. *Tetrahedron* **1989**, *45*, 3445–3450.
3. Badre, A.; Boulanger, A.; Abou-Mansour, E.; Banaigs, B.; Combaut, G.; Francisco, C. *J. Nat. Prod.* **1994**, *57*, 528–533.
4. (a) Lancelot, J.-C.; Gazengel, J.-M.; Robba, M. *J. Heterocycl. Chem.* **1981**, *18*, 1281–1285; (b) Lancelot, J.-C.; Rault, S.; Robba, M. *Chem. Pharm. Bull.* **1984**, *32*, 452–456.
5. (a) Kawasaki, T.; Sakamoto, M. *J. Ind. Chem. Soc.* **1994**, *71*, 443–457; (b) see also, Mohanakrishnan, A.K.; Srinivasan, P.C. *J. Org. Chem.* **1995**, *60*, 1939–1946; (c) Choshi, T.; Matsuya, Y.; Okita, M.; Inada, K.; Sugino, E.; Hibino, S. *Tetrahedron Lett.* **1998**, *39*, 2341–2344; (d) Pindur, U.; Adam, R. *Helv. Chim. Acta* **1990**, *73*, 827–838.
6. Sakamoto, T.; Nagano, T.; Kondo, Y.; Yamanaka, Y. *Chem. Pharm. Bull.* **1988**, *36*, 2248–2252.
7. Achab, S.; Guyot, M.; Potier, P. *Tetrahedron Lett.* **1995**, *36*, 2615–2618.
8. Liebskind, L. S.; Srogl, J. *J. Am. Chem. Soc.* **2000**, *122*, 11260–11261.
9. Gais, H.-J. *Angew. Chem., Int. Ed. Engl.* **1977**, *16*, 244–246.
10. Zheng, Q.; Yang, Y.; Martin, A. R. *Heterocycles* **1994**, *37*, 1761–1772.
11. Mueller, W.; Kipfer, P.; Lowe, D. A.; Urwyler, S. *Helv. Chim. Acta* **1995**, *78*, 2026–2035.
12. (a) Yashioka, H.; Choshi, T.; Sugino, E.; Hibino, S. *Heterocycles* **1995**, *41*, 161–174; (b) Choshi, T.; Yamada, S.; Sugino, E.; Kuwada, T.; Hibino, S. *J. Org. Chem.* **1995**, *60*, 5899–5904.
13. (a) Gilchrist, T. L.; Kemmitt, P. D.; Germain, A. L. *Tetrahedron* **1995**, *51*, 9119–9126; (b) Gilchrist, T. L.; Kemmitt, P. D.; Germain, A. L. *Tetrahedron* **1997**, *53*, 4447–4456.
14. (a) Hoffer, M.; Toome, V.; Brossi, A. *J. Heterocycl. Chem.* **1966**, *3*, 454–458; (b) Dickens, J. P.; Dyer, R. L.; Hamill, B. J.; Harrow, T. A. *J. Org. Chem.* **1981**, *46*, 1781–1786; (c) Bhujanga Rao, A. K. S.; Gundu Rao, C.; Singh, B. B. *Synth. Commun.* **1994**, *24*, 353–366.
15. Liu, Z.-Z.; Chen, H. C.; Cao, S.-L.; Li, R.-T. *Synth. Commun.* **1993**, *23*, 2611–2615.
16. All compounds gave spectroscopic data in agreement with the assigned structures.