# ChemComm

Accepted Manuscript





This is an Accepted Manuscript, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about Accepted Manuscripts in the <u>author guidelines</u>.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the ethical guidelines, outlined in our <u>author and reviewer resource centre</u>, still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this Accepted Manuscript or any consequences arising from the use of any information it contains.



Published on 11 May 2017. Downloaded by Cornell University Library on 11/05/2017 20:17:37.

DOI: 10.1039/C7CC02212C



#### **Journal Name**

## COMMUNICATION

# Bioorthogonal Double-Fluorogenic Siliconrhodamine Probes for Intracellular Super-resolution Microscopy<sup>‡</sup>

Received 00th January 20xx, Accepted 00th January 20xx

E. Kozma, G. Estrada Girona, G. Paci, E. A. Lemke and P. Kele\*

DOI: 10.1039/x0xx00000x

www.rsc.org/

A series of double-fluorogenic siliconrhodamine probes was synthesized. These tetrazine-functionalized, membrane-permeable labels allowed site-specific bioorthogonal tagging of genetically manipulated intracellular proteins and subsequent imaging using super-resolution microscopy.

Recent years have brought substantial progress in the field of super-resolution microscopy (SRM) enabling the exploration of biomolecular processes in the sub-diffraction range. Although the emerged SRM techniques have revealed fine details of live organisms, there is still room for further improvements. Nowadays, the unavailability of dyes suitable for site-specific tagging of intracellular biomolecules is considered as the biggest limitation of SRM techniques.<sup>2</sup> Ideal small-sized organic fluorophores can be synthetically tailored in order to meet the criteria necessary for intracellular SRM imaging of live cells. Such criteria are: (a) both the labels themselves and the chemistry applied for tagging should be biocompatible, (b) their photophysical characteristics should allow minimal background and autofluorescence in order to result in high signal-to-noise (S/N) ratio, (c) their fluorescence should be bright and not or minimally impaired by photobleaching and (d) they should be able to transit through membranes. As of the chemistry applied for installation of such probes, bioorthogonal reactions are extremely useful as they are fully biocompatible.<sup>3</sup> Among these transformations, inverse electron demand Diels-Alder (IEDDA) reactions between tetrazines and strained unsaturated ring systems enable the fastest reactions.4 In order to minimize phototoxicity and autofluorescence, far-red/near-infrared (NIR) emitting fluorophores (650-900 nm) are especially well suited, while background fluorescence of unreacted probes bound nonspecifically to hydrophobic surfaces can be reduced efficiently

by the use of fluorogenic species.<sup>5,6</sup>

Our continuing efforts to develop fluorescent probes that combine the above listed features have resulted in several bioorthogonally applicable fluorogenic probes with various excitation/emission characteristics. Lately, our attention turned to tetrazine quenched fluorescent frameworks as tetrazines can act both as bioorthogonal handles and quencher moieties.7d Tetrazines can exert their quenching effect either by Förster-resonance energy transfer (FRET) or through-bond energy transfer (TBET) mechanisms.<sup>8,9</sup> FRET-based fluorogenic probes are limited to tetrazine-fluorophore pairs where the absorption band of the tetrazine (typically between 520-540 nm) overlaps with the emission band of the fluorophore. TBET based systems, on the other hand, are not constrained by the emission band of the fluorescent unit and, in theory, any kind of fluorophore-tetrazine pairs are suitable upon proper design. Moreover, TBET-based quenching is much faster and more efficient, thus enhancement of the fluorescence signal can reach up to 11,000-fold upon IEDDA reaction with dienophiles.  $^{\rm 9a}$  The tetrazine-based TBET-quenching approach has already been applied to various fluorophores, yet, it has not been performed on SRM compatible dyes. 9 However, techniques like SRM would greatly benefit from the high signal-to-noise ratio offered by TBET-mechanism.

Siliconrhodamine (SiR) is a widely used membrane-permeable NIR dye for SRM applications. <sup>10</sup> Besides the high photostability and brightness, the unique environment-dependent fluorescence of carboxy-SiRs due to a polarity dependent lactone-formation offers an appealing opportunity to distinguish between specific and non-specific labeling, (polarity-based fluorogenicity). <sup>10</sup> When bound to polar protein surfaces, carboxyl-SiRs exist in a fluorescent zwitterionic form, while upon non-specific adhesion to hydrophobic surfaces, it closes to non-fluorescent spirolactone (Figure 1). SiR derivatives were applied in voltage sensing, *in vivo* tumor imaging, photodynamic therapy and multicolour imaging studies as well. <sup>11</sup> Despite their wide applications, there are only a few examples of SiR-based probes for site-specific labeling of proteins. <sup>10a,12,13</sup> Furthermore, no TBET-based carboxyl-SiR fluorogenic probes were reported yet.

Herein we report the first set of bioorthogonally applicable, NIR-emitting, membrane permeable, double fluorogenic

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/x0xx00000x

a. "Lendület" Chemical Biology Research Group, Institute of Organic Chemistry, Research Centre for Natural Sciences, Hungarian Academy of Sciences, Magyar tudósok krt. 2, 1117 Budapest, Hungary. E-mail: kele.peter@ttk.mta.hu

b. Structural and Computational Biology Unit, Cell Biology and Biophysics Unit European Molecular Biology Laboratory D-69117, Heidelberg, Meyerhofstrasse 1., Germany

COMMUNICATION

Published on 11 May 2017. Downloaded by Cornell University Library on 11/05/2017 20:17:37.

DOI: 10.1039/C7CC02212C Journal Name

probes suitable for SRM imaging with exceptionally high turnon rates compared to previously reported tetrazine-based NIR dyes. 9d We demonstrate the power of the developed SiR-dyes in site-specific labeling of intracellular skeletal proteins and subsequent SRM imaging applications.

TBET-based fluorogenic probes are composed of a donor fluorophore and an acceptor tetrazine unit that are linked together with a conjugated but electronically decoupled (twisted) linker. 14 The tetrazine moiety can be installed onto the phenyl group of SiR directly or via alkene, alkyl and phenyl linkers. Former examples showed that incorporation of the tetrazine through direct conjugation or via alkyl-linkers to fluorophores is less efficient.  $^{9c,d,f}$  Therefore we decided to apply alkene and phenyl linkers to connect the tetrazine to carboxy-SiR. Devaraj and co-workers and also our group have recently reported Heck- and Suzuki-type cross-coupling reactions to connect methyl-substituted tetrazines to fluorophores. 7d,9b To this end we have synthesized Brsubstituted intermediate 6. The general synthetic route for SiR derivatives proceeds through a Si-xantone intermediate which is reacted with aryl lithium reagents via nucleophilic addition. 10a Through this route, however, introduction of halogen-substituted aromatic rings is problematic. Therefore we proceeded via an aldehyde condensation reaction developed by Wang et al. (Scheme 1) and acquired Br-SiR derivative **6**.15a

Scheme 1. Synthesis of SiR dyes 11-16. Reaction conditions: a) i) n-BuLi, THF, -78 °C, 2 h; ii) SiCl<sub>2</sub>Me<sub>2</sub>, rt, 16 h; b) CuBr<sub>2</sub>, 140 °C, 16 h; c) Pd<sub>2</sub>(dba)<sub>3</sub>, QPhos, (Cy)2NMe, DMF, 50 °C, 40 min, MW; d) Pd2(dba)3, QPhos, DIPEA, 1,4-dioxane, 90-110 °C, 16 h; e) PdCl<sub>2</sub>(dppf), Cs<sub>2</sub>CO<sub>3</sub>, 1,4-dioxane, 90 °C, 16 h.

We positioned the Br-substituent at the 6-position as previous reports showed superior fluorogenic properties of 6substituted rhodamine derivatives compared to substitution. 9d,f Pd-catalyzed cross-coupling reactions of **6** with methyl vinyltetrazine precursor (8) and methyltetrazine phenylboronic acid pinacol ester (10) resulted in alkene- (11)

and phenyl-linked SiR-tetrazine (15), respectively, with moderate to good yields (11-48%). To improve reaction rate of alkene-linked SiR-tetrazine in IEDDA, we designed a derivative with trifluoromethyl-phenyl substituent. Electron-withdrawing groups are known to markedly increase the reactivities of tetrazines in IEDDA reactions. 4a To this end, we synthesized trifluormethyl-tetrazine (9) in two steps via Zn(OTf)2 catalyzed Pinner-synthesis and subsequent O-mesylation. 15b Heckreaction of tetrazine 9 with compound 6 proceeded with moderate yields (32%). It is known that different N-alkylation patterns on the rhodamine core can markedly alter spectral characteristics and brightness of probes. 16 Rotation about the C-N bond in -NMe2 substituted rhodamines leads to faster decay of the excited state thus lowered quantum yield. Thus we synthesized -NEt2 substituted derivatives (12, 14, 16) starting from Br-SiR derivative 7, which have restricted C-N bond rotation compared to -NMe<sub>2</sub> derivatives.

Next, we characterized the spectral properties of SiRtetrazines 11-16. Measurements were conducted in water containing 0.01% SDS to prevent aggregation of the dyes. 10a Excitation maxima were found to be at around 645 nm and 655 nm, and fluorescence emission was detected at around 665 nm and 670 nm for the dimethylamino- and diethylaminoderivatives, respectively (Figure S1, see ESI). We measured the fluorogenic properties of the probes in an IEDDA reaction with OxTCO (17) (Figure 1, S2, Table 1).7d To our delight, all derivatives showed much larger fluorescence enhancements (FE) than formerly reported fluorogenic tetrazine NIR dyes (cf. 5.6x). 9d We propose that this is attributed to the double fluorogenic nature of our SiR dyes. Among dimethylamine derivatives, phenyl-linked 15 showed the highest turn-on ratio (31x). As expected, ethyl substitution on the N-atom improved turn-on rates of the dyes, and up to 49-fold increase was observed. Although energy transfer occurs from the excited state, interestingly, the extinction coefficients were measured to be lower than  $\varepsilon$  of parent fluorophores (Table 1).

Table 1. Photophysical properties of SiR dyes 11-16

Dye	λ <sub>exc(max)</sub> [nm] <sup>a</sup>	λ <sub>em(max)</sub> [nm] <sup>a</sup>	$\epsilon_{max} [M^{ ext{-}1} cm^{ ext{-}1}]^{a,b}$	<b>φ</b> <sub>F</sub> <sup>a,c</sup>	$\Delta\phi_{\scriptscriptstyle F}{}^{\sf d}$	FE <sup>e</sup>
11	643 (645)	665 (667)	19,700 (19,200)	0.016 (0.34)	21	22
12	653 (652)	671 (672)	20,600 (19,800)	0.015 (0.80)	50	49
13	647 (648)	662 (665)	26,200 (22,500)	0.028 (0.26)	9	6
14	655 (654)	673 (672)	29,000 (25,600)	0.037 (0.27)	7	8
15	645 (644)	660 (660)	23,000 (21,600)	0.012 (0.34)	28	31
16	651 (652)	671 (670)	26,000 (23,900)	0.015 (0.53)	35	40

values refer to tetrazine-dyes while values in parenthesis are that of the respective conjugates with 17; b measured at absorption maxima of the species; c cresyl violet in MeOH was used as quantum yield reference ( $\phi_E$  = 0.53); <sup>d</sup> quantum yield ratios of conjugates with 17 and tetrazine-dyes in 0.1% SDS in water; ' fluorescence intensity enhancements upon reaction with 17 in 0.1% SDS in water at products emission maxima

Published on 11 May 2017. Downloaded by Cornell University Library on 11/05/2017 20:17:37.

Figure 1. a) Fluorogenicity of SiR dyes: IEDDA reaction between 11 and 17 resulting in fluorescence turn-on upon the formation of conjugate 18 due to the transformation of quenching tetrazine moiety into dihydropyridazine during reaction. b) Corresponding fluorescence emission spectra of free dye 11 (black) and conjugate 18 (red) showing fluorescence turn-on upon reaction. c) Polarity-dependent fluorogenicity of SiR dyes: conjugate 18 is present in fluorescent open, zwitterionic form in polar environment, and in non-fluorescent closed spirolactone form in apolar solvents. d) Normalized integral of fluorescence spectra of 18 in water-dioxane mixtures as a function of dielectric constant. Conjugate 18 is in fluorescent zwitterionic form in solvents with dielectric constant >50.

The second order rate constants were also determined and were in good agreement with  $k_2$  values of previously reported methyl-tetrazine derivatives with similarly reactive TCO\* (for values see ESI). As expected, trifluoromethylphenyl derivatives, 13 and 14 reacted much faster than methyl derivatized congeners with OxTCO (17) (Figure S3). Due to the lower electron density of the tetrazines in 13 and 14 IEDDA reactions went to completion within minutes compared to methyl substituted probes where reaction required ca. 1.5 h.

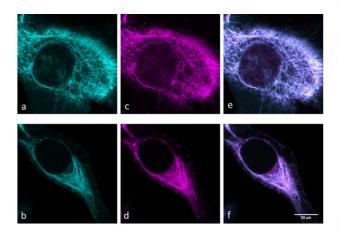


Figure 2. Confocal images of live cell SiR-labeling of vimentin BCNendo—mOrange with dyes **11** (a,c,e) and **12** (b,d,f). Left to right: reference channel (mOrange-cyan) (a,b), labeling channel (c, d) and overlay (e,f). Labeling was performed at 3 µM for 10 min at 37 °C.

We also investigated the polarity dependent absorbance of the Diels-Alder product of alkene-SiR (18). In agreement with previous findings, in solvents with lower dielectric constants, absorption maximum, characteristic of the spirolactone form at 290 nm was dominant. In polar solvents, however, the zwitterionic form with a characteristic absorption at 645 nm

dominated the spectrum (Figure S4). As a consequence, in solvents with dielectric constants <50 conjugate **18** was non-fluorescent while fluorescence gradually reinstated with increasing polarity (Figure 1).

With the SiR-dyes in hand, showing exceptional fluorogenicity in the NIR region, we were eager to evaluate their performance in the cellular environment. We sought to utilize genetic code expansion (GCE) technology for site-specific labeling with the newly developed double-fluorogenic probes. We and others have recently shown that non-canonical amino acids (ncAA) with strained alkene and alkyne moieties can be genetically encoded in mammalian cells using the pyrrolysine tRNA<sup>Pyl</sup>/PyIRS pair from *Methanosarcina* species.<sup>18</sup> We used the previously reported skeletal protein vimentin 116TAG mOrange to test the intracellular labeling ability of the dyes. 13,18e Using orthogonal tRNA Pyl/PylRS AF pair we genetically encoded commercially available cyclooctynylated-lysine (Lys(arepsilon-N-BCN<sup>endo</sup>), Figure S5) into vimentin 116TAG-mOrange and performed live cell labeling experiments with dyes 11-16 in mammalian cells. Clear live cell labeling was observed with dye 11 and 12 in concentrations as low as 1.5  $\mu$ M within 10 min at 37 °C (Figure 2, for additional images at different concentrations and labeling times see Figure S6, S7, and for a comparison of different post-labeling washing time see Figure S8). The labeling was specific with no strong background fluorescence (as quantified by channel colocalisation in Figure S9). Less efficient labeling was observed with phenyl-linked dyes 15 and 16, which is probably due to their slower kinetics (Figure S10). Although CF3-dyes 13 and 14 showed the fastest labeling properties in vitro, they did not give any specific labeling in vivo (results not shown). We attribute this to the low polarity of these dyes, which probably impairs their ability to cross cell-membrane or to label the protein specifically.

To date, there are only a few reports on bioorthogonally applicable, fluorogenic tetrazine-mediated intracellular labeling inside living cells and subsequent SRM. <sup>13,18e,19</sup> Among them, to the best of our knowledge, there are no reports of

DOI: 10.1039/C7CC02212C

COMMUNICATION Journal Name

site-specific labeling with TBET-based fluorogens. We set out to test the suitability of dyes 11 and 12 in SRM of vimentin. While GSDIM SRM with TIRF illumination was possible with vimentin BCNendo -mOrange labeled with dye **11**, we did not get any specific SRM signal with diethyl-derivative 12 (Figure 3). SRM imaging with SiR-11 (Figure 3E, F) clearly gives enhanced resolution compared to the reference channels (Figure 3A-D, and Figure S11).

In conclusion, we have developed double-fluorogenic siliconrhodamine-tetrazine probes with improved kinetics and enhanced fluorescence turn-on ratios in the NIR region upon inverse electron demand Diels-Alder reactions. We have demonstrated the effect of the linker and N-alkylation pattern of the SiR-tetrazine core on labeling efficiency by confocal microscopy. We successfully applied one derivative in sitespecific super-resolution imaging of a cytoskeletal protein, vimentin. Such probes with distinct spectral characteristics would allow multicolor super-resolution imaging of various intracellular structures.

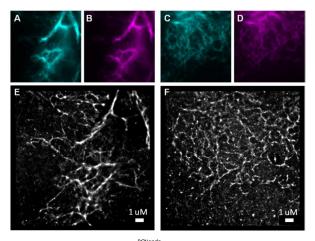


Figure 3. SRM imaging of vimentin BCNendo-mOrange labeled with dye 11. Panels A,C (cyan) and B,D (magenta) show the mOrange and SiR reference channels for the corresponding SRM images in panels E and F from dye 11 labeling (3 µM for 30 minutes at 37 °C). SRM images E and F have resolutions of 35 nm and 28 nm respectively as determined by the Fourier ring correlation criterion (FRC).<sup>20</sup>

### Notes and references

Published on 11 May 2017. Downloaded by Cornell University Library on 11/05/2017 20:17:37.

- ‡ Present work was supported by the Hungarian Scientific Research Fund (OTKA, NN-116265) and the "Lendület" Program of the Hungarian Academy of Sciences (LP2013-55/2013). E.K. is grateful for the support of The New National Excellence Program of The Ministry of Human Capacities (Hungary). EAL acknowledges the SPP1623 and SFB1129 for funding.
  - a) M. Heilemann, J. Biotechnol., 2010, 149, 243; b) B. Huang, H. Babcock and X. Zhuang, Cell, 2010, 143,
  - a) M. Fernández-Suárez and A. Y. Ting, Nat. Rev. Mol. Cell Biol., 2008, 9, 929; b) T. J. Chozinski, L. A. Gagnon and J. C. Vaughan, FEBS Lett., 2014, 588, 3603.
  - a) C. P. Ramil and Q. Lin, Chem. Commun., 2013, 49, 11007; b) J. A. Prescher and C. R. Bertozzi, Nat. Chem. Biol., 2005, 1, 13; c) E. M. Sletten and C. R. Bertozzi, Angew. Chem. Int. Ed., 2009, 48, 6974.

- a) A.-C. Knall and C. Slugovc, Chem. Soc. Rev., 2013, 42, 5131; b) E. Kozma, O. Demeter and P. Kele, ChemBioChem, 2017, 18, 486.
- a) G. Cserép, A. Herner and P. Kele, Methods Appl. Fluoresc., 2015, 3, 042001; b) L. D. Lavis and R. T. Raines, ACS Chem. Biol., 2014, 9, 855.
- A. Nadler and C. Schultz, Angew. Chem. Int. Ed., 2013, **52**, 2408.
- a) A. Herner, I. Nikić, M. Kállay, E. A. Lemke and P. Kele, Org. Biomol. Chem., 2013, 11, 3297; b) A. Herner, G. Estrada Girona, I. Nikić, M. Kállay, E. A. Lemke and P. Kele, Bioconjugate Chem., 2014, 25, 1370; c) O. Demeter, E. A. Fodor, M. Kállay, G. Mező, K. Németh, P. T. Szabó and P. Kele, Chem. Eur. J., 2016, 22, 6382; d) G. Knorr, E. Kozma, A. Herner, E. A. Lemke and P. Kele, Chem. Eur. J., 2016, 22, 8972.
- a) N. K. Devaraj, S. Hilderbrand, R. Upadhyay, R. Mazitschek and R. Weissleder, Angew. Chem. Int. Ed., 2010, **49**, 2869; b) J. Yang, J. Šečkutė, C. M. Cole and N. K. Devaraj, Angew. Chem. Int. Ed., 2012, 51, 7476.
- a) L. G. Meimetis, J. C. T. Carlson, R. J. Giedt, R. H. Kohler and R. Weissleder, Angew. Chem. Int. Ed., 2014, 53, 7531; b) H. Wu, J. Yang, J. Šečkute and N. K. Devaraj, Angew. Chem. Int. Ed., 2014, 53, 5805 c) A. Wieczorek, T. Buckup and R. Wombacher, Org. Biomol. Chem., 2014, 12, 4177; d) A. Wieczorek, P. Werther, J. Euchner and R. Wombacher, Chem. Sci., 2017, 8, 1506; e) P. Agarwal, B. J. Beahm, P. Shieh and C. R. Bertozzi, Angew. Chem. Int. Ed., 2015, 54, 11504; f) J. C. T. Carlson et al., Angew. Chem. Int. Ed., 2013, 52, 6917.
- 10 a) G. Lukinavičius et al., Nat. Chem., 2013, 5, 132; b) S. Uno, et al., Nat. Chem., 2014, 6, 681; c) P. Shieh, M. S. Siegrist, A. J. Cullen and C. R. Bertozzi, Proc. Natl. Acad. Sci. USA, 2014, 111, 5456
- 11 Y. Kushida, T. Nagano and K. Hanaoka, Analyst, 2015, **140**. 685.
- T. Peng and H. C. Hang, J. Am. Chem. Soc., 2016, 138, 14423.
- 13 C. Uttamapinant et al., J. Am. Chem. Soc., 2015, 137, 4602.
- a) T. G. Kim et al., J. Phys. Chem. A, 2006, 110, 20; b) M. Steeger, S. Griesbeck, A. Schmiedel, M. Holzapfel, I. Krummenacher, H. Braunschweig and C. Lambert, Phys. Chem. Chem. Phys., 2015, 17, 11848.
- 15 a) B. Wang, X. Chai, W. Zhu, T. Wang and Q. Wu, Chem. Commun., 2014, 50, 14374; b) J. Yang, M. R. Karver, W. Li, S. Sahu and N. K. Devaraj, Angew. Chem. Int. Ed., 2012, 51, 5222.
- 16 J. B. Grimm, B. P. English, J. Chen, J. P. Slaughter, Z. Zhang, A. Revyakin, R. Patel, J. J. Macklin, D. Normanno, R. H. Singer, T. Lionnet and L. D. Lavis, Nat. Methods, 2015, 12, 244.
- 17 J. E. Hoffmann, T. Plass, I. Nikić, I. V. Aramburu, C. Koehler, H. Gillandt, E. A. Lemke, C. Schultz, Chem. Eur. J. 2015, **21**, 12266.
- 18 a) T. Plass et al., Angew. Chem. Int. Ed. 2012, **51**, 4166; b) I. Nikić, T. Plass, O. Schraidt, J. Szymański, J. A. Briggs, C. Schultz and E. A. Lemke, Angew. Chem. Int. Ed. 2014, 53, 2245; c) E. Kozma, I. Nikić, B. R. Varga, I. V. Aramburu, J. H. Kang, O. T. Fackler, E. A. Lemke and P. Kele, ChemBioChem, 2016, 17, 1518; d) K. Lang, L. Davis, J. Torres-Kolbus, C. Chou, A. Deiters and J. W. Chin, Nat. Chem. 2012, 4, 298; e) I. Nikić et al., Angew. Chem. Int. Ed., 2016, 55, 16172.
- R. S. Erdmann et al., Angew. Chem. Int. Ed., 2014, 53, 10242.
- 20 N. Banterle et al., J. Struct. Biol., 2013, 183, 363.