

Available online at www.sciencedirect.com



Tetrahedron

Tetrahedron 61 (2005) 1449-1457

Regio- and stereoselective reactions between cyclic Baylis–Hillman type adducts and N-nucleophiles and P-nucleophile

Ewa Krawczyk, Krzysztof Owsianik and Aleksandra Skowrońska*

Centre of Molecular and Macromolecular Studies, Polish Academy of Sciences, 90-363 Łódź, Sienkiewicza 112, Poland

Received 20 September 2004; revised 8 November 2004; accepted 2 December 2004

Available online 16 December 2004

Abstract—New important organic compounds multifunctionalized cyclic 6-membered and 7-membered allylic amines, azide and phosphonates have been obtained via regio- and diastereoselective reactions of cyclic Baylis–Hillman type adducts 1 with N-nucleophiles and P-nucleophile. We have found that the reactions proceed by $S_N 2$ or $S_N 2'$ processes exclusively, or by both processes simultaneously. The $S_N 2'$ process occurs with *anti* stereochemistry.

© 2004 Elsevier Ltd. All rights reserved.

1. Introduction

Baylis–Hillman acyclic adducts are versatile multifunctional molecules. They have proved useful synthetic intermediates¹ and undergo a variety of organic transformations with regio- and stereochemical control.² Several successful examples of such transformations have already been reported, including synthesis of allylic amino compounds.³ But little work has been described on cyclic Baylis–Hillman adducts. They are mainly connected with addition reactions of metalloorganic reagents.⁴

We have previously reported the synthesis of novel cyclic Baylis–Hillman type adducts 1 of defined stereochemistry (Fig. 1).⁵



Figure 1.

In continuation of our interest in the synthesis of multifunctional cyclic compounds, we have studied regio- and stereoselectivity of the reaction of adducts 1 with nucleophiles. They lead to functionalized 6 and 7-membered ring allylic amines, allylic azide and allylic phosphonates of defined stereochemistry.

2. Results and discussion

Our synthetic approach to Baylis–Hillman type adducts 1 involves reduction of the carbonyl group in readily available thiophosphates 2^5 by NaBH₄ in the presence of MeI, subsequent oxidation of intermediate 3 to sulphoxide 4 and *cis* elimination from the latter (Scheme 1).



Scheme 1.

Initially we studied the reactions between adducts **1a** and **1b** with different amines, sodium azide and trimethyl phosphite. Treatment of **1a** with benzylamine and diethylamine at room temperature furnished the corresponding secondary

Keywords: Baylis–Hillman type adducts; Multifunctionalized allylic compounds; $\rm S_N2'$ reaction; Stereochemistry.

^{*} Corresponding author. Tel.: +48 42 6843120; fax: +48 42 6847126; e-mail: askow@bilbo.cbmm.lodz.pl

^{0040–4020/\$ -} see front matter @ 2004 Elsevier Ltd. All rights reserved. doi:10.1016/j.tet.2004.12.008



Scheme 2.

5 and tertiary cyclic allylic amines **6** in good yield (82% and 66%, respectively). Reaction of **1a** and **1b** with optically active *S*-(-)- α -methylbenzylamine and *R*-(+)- α -methylbenzylamine in methanol under the same reaction conditions provided 6-membered cyclic amines **7a(b)** in 80% yield and 7-membered cyclic amines **8a(b)** in 60% yield, as a mixture of two diastereoisomers in the ratio 1:1 and 1.5:1, respectively (Scheme 2).

Reaction of **1a** with sodium azide in DME solution at room temperature afforded a new allylic azide **9** in good yield. Reduction of azide **9** under standard conditions with $Ph_3P/THF/H_2O^{3b}$ gave exclusively the primary allylic amine **10**. The phosphate **1a** was also subjected to reaction with trimethyl phosphite in boiling toluene, providing phosphonate **11** in 69% yield (Scheme 3).

The structure of the amines, azide and phosphonate obtained was confirmed by ¹H NMR analysis and mass spectrometry.

The quasiequatorial position of allylic hydrogen (H⁽¹⁾) in **5** was determined on the basis of characteristic vicinal coupling constants ${}^{3}J_{\text{HaHe}'}=3.8 \text{ Hz}$, ${}^{3}J_{\text{HeHe}'}=3.8 \text{ Hz}$.⁶ Determination of the structures of amines **6**, **7a(b)** and **8a(b)** was impossible, they exhibited allylic protons peaks at 3.75, 3.66, and 3.91 ppm as multiplets. Configuration of amines **7** was determined by hydrogenolysis, which afforded of 2-amino-cyclohexanecarboxylic acid ethyl ester of known configuration.⁷

2.1. Regio- and stereoselectivity of reactions between adducts 1c,d and N-nucleophiles and P-nucleophile

Yamamoto et al. reported that phosphate ester is an excellent leaving group in $S_N 2'$ selective reactions between organocopper reagents and acyclic allylic alcohol derivatives.⁸ These authors have also found that acyclic allylic phosphates gave, with Grignard reagents and in the presence





Scheme 4.





of an iron catalyst, cross-coupling products with high $S_N 2$ selectivity.⁹

Therefore, it was of interest to test the regio- and stereoselectivity of reactions of adducts **1c** and **1d** (which are functionalized allylic phosphates) containing an additional alkyl or aryl substituent in the ring, with nucleophiles. Nucleophilic attack may follow an $S_N 2$ or $S_N 2'$ pathway, or both pathways simultaneously. The stereochemical course of the $S_N 2'$ reaction, of crucial importance in synthesis, is clear in many instance but much less clear in others. In many cases entering and leaving groups departed from the same side of the C=C plane.¹⁰ However, there are several experiments which show that the *anti* mode is also possible.¹¹ The factors which favor the *anti* mode are not yet completely understood, but there

appears to be no doubt that stereoelectronic effects play an important role in these reactions.¹²

We have found that in the reaction of **1c** with benzylamine and sodium azide both processes ($S_N 2$ and $S_N 2'$) operated, giving the mixture of regioisomers **12** and **13**, and **14** and **15** in the ratio 1:1.6 and 2:1, respectively (Scheme 4). The $S_N 2$ and $S_N 2'$ product distribution of these reactions is not solvent dependent.¹³ In polar solvents like methanol, as well as in much less polar solvents like DME, a similar ratio of final products was observed. The resulting regioisomers could be easily separated by silica gel chromatography.

In contrast, reaction of **1c** with diethylamine performed in methanol at r.t. is fully regio- and diastereoselective providing one regioisomer **16** ($S_N 2'$ product) in high yield (Scheme 5).

The extension of our investigation to the reactions of adduct **1d** with N-nucleophiles has shown that, with benzylamine and diethylamine alongside amounts of cyclic conjugated diene **19**,¹⁴ allylic amine **17** ($S_N 2'$ product) and allylic amine **18** ($S_N 2$ product) were formed (Scheme 6).

Reactions between adducts **1c** and **1d** with P-nucleophile such as trimethylphosphite lead to one regioisomer of the



corresponding novel allylic phosphonate **20** in 70% yield and allylic phosphonate **21** in 90% yield, exclusively (Scheme 7).



Scheme 7.

Reactions presented in Scheme 4 are partially regioselective giving products via $S_N 2$ and $S_N 2'$ processes, whereas reactions presented in Schemes 5-7 are fully regioselective providing allylic amines 16, 17 and allylic phosphonates 20, 21 via an $S_N 2'$ process. It is difficult to explain unambiguously the observed different regioselectivity in the investigated reactions. However, it is very likely that reactions of 1c with diethylamine, benzylamine and azide are controlled by the following factors: the nature of nucleophile, conformational equilibrium in the starting material and the steric hindrance of the methyl group in the transition states of the competing $S_N 2$ and $S_N 2'$ reaction pathways. The ratio of regioisomers was established by ¹H NMR. The structure and configuration of compounds prepared were determined on the basis of ¹H, ¹³C NMR (including COSY experiments), in particular characteristic values of vicinal H–H coupling constants: $J_{ae'}$ and $J_{ee'}$, vicinal ³¹P–¹³C coupling constants values, and in some cases (for 17, 20 and 21) were confirmed by the data obtained from calculations performed with the AM-1 program.¹⁵ The values of the vicinal coupling constants between axial-quasiaxial protons are generally higher than these for an axial-quasiequatorial orientation.⁶ The values of coupling constants observed also revealed a cis relationship between the methyl (phenyl) substituent and functional groups in 12, 13, 14, 17, 18, 20 and 21. Karpluslike dependence of vicinal ${}^{31}P-{}^{13}C$ coupling on the dihedral angle has been established for the (MeO)₂P(O) group.¹⁶ The small values of ${}^{3}J_{C(4)P}$ observed for compounds 11, 20 and 21, indicated a quasiaxial orientation for the phosphonate group.17

It was not possible to determine *syn* or *anti* stereochemistry of $S_N 2'$ process for all products obtained. However, it seems reasonable, on the basis of presented NMR data, to assume that nucleophilic attack of benzylamine, azide anion and trimethyl phosphite on the allylic system of adducts **1c** and **1d** is *anti* to the leaving phosphate group via a chairlike transition state. A non-concerted $S_N 2'$ process is not

excluded. In such cases *anti* attack of the nucleophile creates an anion stabilized by the presence of a carboester group in the α -position. The next step should involve elimination of phosphoric acid with formation of the final product. We excluded the possibility of allylic rearrangement of adducts **1** under reaction conditions, which consequently provided a final product with *anti* stereo-chemistry.^{11a,18}

3. Conclusion

Reactions of cyclic Baylis–Hillman type adducts 1 with nucleophiles presented here constitute an excellent route to novel synthetically important compounds: sterically defined 6-membered and 7-membered cyclic multifunctional allylic amines, azide and phosphonates. The results constitute new examples of S_N2' reactions proceeding with *anti* stereo-chemistry. The whole spectrum, from *syn* to *anti*, is to be expected depending mainly in any particular case, on the nature of displacing and displaced groups.

4. Experimental

4.1. General

¹H, ¹³C and ³¹P NMR spectra were recorded on a Bruker AC 200 Spectrometer at 200.13, 50.32 and 81.02 MHz, respectively (using deuterochloroform as solvent) unless otherwise noted IR spectra were measured on an Ati Mattson Infinity FT IR 60. GC spectra were performed on a Hewlett-Packard 5890. MS spectra (EI, CI, and HRMS) were recorded on a Finnigan MAT 95 spectrometer. Optical rotation values were measured in 100 mm cell on Perkin Elmer 241 MC under Na lamp radiation.

All the reactions were carried out using anhydrous conditions and under an atmosphere of argon. Chromatographic purification was performed on silica gel columns (Merck, Kieselgel 70–230 mesh) with indicated eluent. Chemicals and solvents were obtained from commercial sources and distilled or dried according to standard methods. All phosphates, i.e. 6-(diethoxy-phosphoryloxy)-cyclohex-1-enecarboxylic acid ethyl ester (**1a**) were prepared as described.⁵

4.2. Synthesis of allylic amines 5, 6, 7a,a', 7b,b', 8a,a' 8b,b', 12, 13, 16, 17, 18 and allylic azides 9, 14, 15

General procedure. To a solution of appropriate allylic phosphates (0.3 mmol) in dry methanol (5 mL) or, in the case of azide, in dry dimethoxyethane (5 mL), 0.6 mmol of amines or azide was added at room temperature under argon. Progress of the reaction was monitored by TLC chromatography. When the reaction was complete (24 h for amines and three days for azide), solvent was removed in vacuo, residue was diluted water and extracted with ether $(3 \times 5 \text{ mL})$. The organic layer was dried (MgSO₄) and the solvent was removed to leave colorless oils, which were purified by silica gel chromatography (petroleum ether/EtOAc). **4.2.1. 6-Benzyloamino-cyclohex-1-enecarboxylic acid** ethyl ester (5). R_f (ethyl acetate/petroleum ether 1:1)= 0.46. Yield: 82% (63.7 mg from 92.0 mg of **1a**); colorless oil; ¹H NMR (CDCl₃): δ =1.28 (t, ³ J_{HH} =7.1 Hz, 3H, OCH₂CH₃), 1.35–1.67 (m, 2H, CH₂), 1.72–1.95 (m, 2H, CH₂), 1.82 (s, 1H, NH), 2.10–2.38 (m, 2H, CH₂), 3.63 (dd, ³ J_{HH} =3.8, 3.8 Hz, 1H, CHNH), 3.82 (AB, ² J_{HH} =12.8 Hz, 2H, NHCH₂), 4.18 (q, ³ J_{HH} =7.1 Hz, 2H, COCH₂), 7.08 (dd, ³ J_{HH} =3.9, 3.9 Hz, 1H, CH=C), 7.18–7.42 (m, 5H, CH_{arom}) ppm. ¹³C NMR (CDCl₃): δ =14.3 (OCH₂CH₃), 23.2, 25.9, 29.7 (CH₂), 52.1 (NCH₂Ph), 53.8 (CHNH), 60.3 (OCH₂CH₃), 126.7, 127.5, 128.1 (C_{arom}), 130.3 (C=CH), 140.3 (CH=C), 143.4 (C_{*ipso*}), 168.0 (C=O). IR (film, cm⁻¹): 3387 ν (NH), 1707 ν (C=O). MS (CI-isobutane): *m/z* 260 [M+H]⁺. HRMS (CI) calc. for C₁₆H₂₁O₂N+H [M+ H]⁺ 260.1650; found: 260.1659.

4.2.2. 6-Diethylamino-cyclohex-1-enecarboxylic acid ethyl ester (6). $R_{\rm f}$ (ethyl acetate/petroleum ether 1:1) = 0.67. Yield: 66% (42.7 mg from 88.0 mg of **1a**); colorless oil; ¹H NMR (CDCl₃): δ =0.99 (t, ³ $J_{\rm HH}$ =7.0 Hz, 6H, NCH₂CH₃), 1.29 (t, ³ $J_{\rm HH}$ =7.1 Hz, 3H, OCH₂CH₃), 1.40– 1.85 (m, 4H, CH₂), 2.05–2.20 (m, 2H, CH₂), 2.50 (q, ² $J_{\rm HH}$ =7.0 Hz, 4H, NHCH₂), 3.75 (m, 1H, CHN), 4.20 (q, ³ $J_{\rm HH}$ =7.1 Hz, 2H, COCH₂), 6.75 (ddd, ³ $J_{\rm HH}$ =4.0, 4.0 Hz, ^{4 $J_{\rm HH}$ =1.5 Hz, 1H, CH=C) ppm. ¹³C NMR (CDCl₃): δ = 14.2 (OCH₂CH₃), 19.6, 21.0 (CH₂), 25.5 (NCH₂CH₃), 29.6 (CH₂), 44.0 (NCH₂CH₃), 53.4 (HCN), 65.8 (OCH₂CH₃), 121.9 (HC=C), 136.7 (HC=C), 167.4 (C=O). IR (film, cm⁻¹): 1715 ν (C=O). MS (EI, 70ev): m/z 225 [M]⁺. HRMS (CI) calc. for C₁₃H₂₃O₂N+H [M+H]⁺ 226.1807; found: 226.1814.}

4.2.3. (-)-(1S,6S)-6-(1-Phenyl-ethylamino)-cyclohex-1ene-carboxylic acid ethyl ester (7a). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.64. Yield: 40% (40 mg from 112.0 mg of **1a**); colorless oil; ¹H NMR (CDCl₃, 500 MHz): $\delta = 1.28$ (t, ${}^{3}J_{HH} = 7.1$ Hz, 3H, OCH₂CH₃), 1.33 (d, ${}^{3}J_{HH} = 6.6$ Hz, 3H, CH₃CH), 1.39 $(dd, {}^{3}J_{HH} = 4.1, 7.0 \text{ Hz}, 1.5\text{H}, CH_{2}), 1.53 (dd, {}^{3}J_{HH} = 2.4,$ 13.3 Hz, 2H, CH₂), 1.62–1.70 (m, 2H, CH₂), 1.97–2.05 (10 lines, ${}^{3}J_{HH} = 3.8$, 4.5 Hz, 1H, CH₂), 2.16–2.20 (two lines, 1H, CH₂), 3.66 (br s, 1H, CHNH), 3.93 (q, ${}^{3}J_{HH}$ =6.6 Hz, 1H, CH₃CH), 4.17 (q, ${}^{3}J_{HH}$ =7.1 Hz, 1H, COCH₂), 4.19 (q, ${}^{3}J_{\text{HH}} = 7.1 \text{ Hz}, 1\text{H}, \text{ COCH}_{2}, 6.99 \text{ (dd, } {}^{3}J_{\text{HH}} = 3.3, 3.8 \text{ Hz},$ 1H, CH=C), 7.16–7.36 (m, 5H, CH_{arom}) ppm. ¹³C NMR $(CDCl_3): \delta = 14.3 (OCH_2CH_3), 24.1 (CHCH_3), 25.9, 27.4,$ 29.6 (CH₂), 49.8 (HNCH), 57.7 (HC-CH₃), 60.9 (OCH₂CH₃), 126.8, 126.9, 127.3, 128.2 (C_{arom}), 132.5 (HCCH₂=*C*), 142.4 (H*C*=*C*), 146.5 (C_{*ipso*}), 167.4 (C=O). IR (film, cm⁻¹): 3417 ν (NH), 1703 ν (C=O). MS (CI, isobutane): m/z 274 [M+H]⁺. HRMS (CI) calc. for $C_{17}H_{23}O_2N+H$ [M+H]⁺ 274.1807; found: 274.1813. $[\alpha]_{D}^{20} = -105.3 \text{ (CHCl}_{3}, c = 0.85) \text{ [from } S - (-) \text{PhCH(CH}_{3}) - \text{NH}_{2}, [\alpha]_{D}^{20} = -37.5 \text{ (MeOH, } c = 3.0) \text{]}.$

4.2.4. (-)-(1*S*,6*R*)-6-(1-Phenyl-ethylamino)-cyclohex-1enecarboxylic acid ethyl ester (7a'). Single diastereoisomer: R_f (ethyl acetate/petroleum ether 1:2)=0.48. Yield: 40% (39.8 mg from 112.0 mg of 1a); colorless oil; ¹H NMR (CDCl₃): δ =1.28 (t, ³J_{HH}=7.1 Hz, 3H, OCH₂CH₃), 1.39 (d, ³J_{HH}=6.6 Hz, 3H, CH₃CH), 1.79 (br s, 2H, CH₂), 1.91 (br s, 1H, CH₂), 2.08–2.13 (seven lines, ³J_{HH}=4.0, 5.5, 6.0 Hz, 2H, CH₂), 2.28 (dd, ${}^{3}J_{HH}$ =9.6, 3.9 Hz, 1H, CH₂), 3.39 (br s, 1H, CHNH), 3.99 (q, ${}^{3}J_{HH}$ =6.6 Hz, 1H, CH₃CH), 4.17 (q, ${}^{3}J_{HH}$ =7.1 Hz, 1H, OCH₂), 4.18 (q, ${}^{3}J_{HH}$ =7.1 Hz, 1H, OCH₂), 7.04 (dd, ${}^{3}J_{HH}$ =3.8, 3.8 Hz, 1H, CH=C), 7.34–7.38 (m, 5H, CH_{aron}) 13 C NMR (CDCl₃): δ =14.2 (OCH₂CH₃), 24.2 (CHCH₃), 25.8, 29.1, 30.3 (CH₂), 49.0 (HNCH), 55.1 (HCCH₃), 60.9 (OCH₂CH₃), 126.9, 127.9, 128.4, 128.8 (C_{arom}), 133.3 (HCCH₂=*C*), 145.9 (H*C*=C), 148.0 (C_{*ipso*}), 167.32 (C=O). IR (film, cm⁻¹): 3387 ν (NH), 1699 ν (C=O). MS (CI, isobutane): *m*/*z* 274 [M+H]⁺. HRMS (CI) calc. for C₁₇H₂₃O₂N+H [M+H]⁺ 274.1807; found: 274.1815. [α]_D²⁰= -13.9 (CHCl₃, *c*=0.9) [from *S*-(-) PhCH(CH₃)NH₂, [α]_D²⁰= -37.5 (MeOH, *c*=3.0)].

4.2.5. (+)-(1*R*,6*R*)-6-(1-Phenyl-ethylamino)-cyclohex-1enecarboxylic acid ethyl ester (7b). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2) = 0.64. Yield: 40% (28.6 mg from 81.0 mg of 1a); colorless oil; ¹H NMR (CDCl₃): δ = 1.32 (t, ³J_{HH} = 7.2 Hz, 3H, OCH₂CH₃), 1.37 (d, ³J_{HH} = 6.7 Hz, 3H, CH₂), 3.66 (br s, CHNH), 3.97 (q, ³J_{HH} = 6.5 Hz, 1H, CH₃CH), 4.22 (q, ³J_{HH} = 7.1 Hz, 1H, COCH₂), 4.23 (q, ³J_{HH} = 7.1 Hz, 1H, COCH₂) 7.03 (dd, ³J_{HH} = 3.4, 4.0 Hz, 1H, CH=C), 7.20–7.41 (m, 5H, CH_{arom}) ppm. [α]_D²⁰ = +106.3 (CHCl₃, c=0.45) [from *R*-(+) PhCH(CH₃)NH₂, [α]_D²⁰ = +27.4 (MeOH, *c*=2.4)]. IR and MS spectra were identical to those of the isomer **7a**.

4.2.6. (+)-(1*R*,6*S*)-6-(1-Phenyl-ethylamino)-cyclohex-1enecarboxylic acid ethyl ester (7b'). Single diastereoisomer: R_f (ethyl acetate/petroleum ether 1:2) = 0.48. Yield: 40% (28.2 mg from 81.0 mg of 1a); colorless oil; ¹H NMR (CDCl₃): δ =1.27 (t, ³ J_{HH} =7.1 Hz, 3H, OCH₂CH₃), 1.38 (d, ³ J_{HH} =6.6 Hz, 3H, CH₃CH), 1.76–2.25 (m, 6H, CH₂), 3.39 (br s, 1H, CHNH) 3.99 (q, ³ J_{HH} =6.6 Hz, 1H, CH₃CH), 4.18 (q, ³ J_{HH} =7.1 Hz, 1H, OCH₂), 4.19 (q, ³ J_{HH} =7.1 Hz, 1H OCH₂), 7.06 (dd, ³ J_{HH} =3.8, 3.8 Hz, 1H, CH=C), 7.24–7.37 (m, 5H, CH_{arom}) ppm. [α]_D²⁰ = +20.6 (CHCl₃, c=0.35) [from R-(+)PhCH(CH₃)NH₂, [α]_D²⁰ = +27.4 (MeOH, c=2.4)]. IR and MS spectra were identical to those of the isomer 7a'.

4.2.7. (–)-7-(1-Phenyl-ethylamino)-cyclohept-1-ene-car**boxylic acid ethyl ester (8a).** Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.65. Yield: 38% (30.5 mg from 90.0 mg of 1b); colorless oil; ¹H NMR (CDCl₃): $\delta = 1.22$ (t, ${}^{3}J_{\text{HH}} = 7.1$ Hz, 3H, OCH₂CH₃), 1.34 (d, ${}^{3}J_{HH}$ = 6.6 Hz, 3H, CH₃CH), 1.35–2.08 (m, 6H, CH₂, NH), 2.16–2.60 (m, 3H, CH₂), 3.82 (q, ${}^{3}J_{HH}$ = 6.6 Hz, 1H, CH₃CH), 3.91 (m, 1H, CHNH), 4.13 (q, ${}^{3}J_{HH}$ = 7.1 Hz, 2H, OCH₂), 7.17 (m, 1H, CH=C), 7.19–7.50 (m, 5H, CH_{arom}) ppm. ¹³C NMR (CDCl₃): $\delta = 16.0$ (OCH₂CH₃), 23.7 (CHCH₃), 24.8, 27.6, 29.7, 30.4 (CH₂), 53.0, 56.6 (CHN or CHCH₃), 60.8 (OCH₂CH₃), 124.9 (C=CH), 125.9, 127.0, 128.5 (Carom), 145.1 (C=CH), 146.7 (Cipso), 168.1 (C=O). IR (film, cm⁻¹): 3460 ν (NH), 1703 ν (C=O). MS (CI, isobutane): m/z 288 $[M+H]^+$. HRMS (CI) calc. for $C_{18}H_{25}O_2N+H [M+H]^+$ 288.1963; found: 288.1957. $[\alpha]_{\rm D}^{20} = -36.5$ $(CHCl_3, c=0.2)$ [from S-(-)PhCH(Me)NH₂, $[\alpha]_D^{20} = -37.5$ (CHCl₃, c = 3.0)].

4.2.8. (+)-7-(1-Phenyl-ethylamino)-cyclohept-1-ene-carboxylic acid ethyl ester (8a'). Single diastereoisomer: R_f

(ethyl acetate/petroleum ether 1:2)=0.45. Yield: 25% (20.0 mg from 90.0 mg of **1b**); colorless oil; ¹H NMR (CDCl₃): δ =1.24 (t, ³J_{HH}=7.1 Hz, 3H, OCH₂CH₃), 1.38 (d, ³J_{HH}=6.5 Hz, 3H, CH₃CH), 1.46–2.10 (m, 6H, CH₂, NH), 2.15–2.68 (m, 3H, CH₂), 3.85 (q, ³J_{HH}=6.5 Hz, 1H, CH₃CH), 3.92 (m, 1H, CHNH), 4.12 (q, ³J_{HH}=7.1 Hz, 2H, OCH₂), 7.12 (dd, ³J_{HH}=5.7, 5.7 Hz, 1H, CH=C), 7.23–7.37 (m, 5H, CH_{arom}) ppm. ¹³C NMR (CDCl₃): δ =14.2 (OCH₂CH₃), 23.5 (CH₃CH), 24.1, 26.0, 27.3, 28.1 (CH₂), 52.6, 55.6 (CHN or CHCH₃), 60.7 (OCH₂CH₃), 125.6 (C=CH), 126.6, 127.4, 128.3 (C_{arom}), 144.8 (C=CH), 146.5 (C_{ipso}), 168.0 (C=O). IR (film, cm⁻¹): 3400 ν (NH), 1704 ν (C=O). MS (CI, isobutane): *m*/z 288 [M+H]⁺. HRMS (CI) calc. for C₁₈H₂₅O₂N+H [M+H]⁺ 288.1963; found: 288.1967; [α]_D²⁰= +4.9 (CHCl₃, *c*=1.3) [from *S*-(-) PhCH(Me)NH₂, [α]_D²⁰= -37.5 (CHCl₃, *c*=3.0)].

4.2.9. (+)-7-(**1**-Phenyl-ethylamino)-cyclohept-1-ene-carboxylic acid ethyl ester (**8b**). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.65. Yield: 35% (18.7 mg from 60.0 mg of **1b**); colorless oil; ¹H NMR (CDCl₃): δ =1.23 (t, ³ $J_{\rm HH}$ =7.1 Hz, 3H, OCH₂CH₃), 1.41 (d, ³ $J_{\rm HH}$ =6.5 Hz, 3H, CH₃CH), 1.58–1.85 (m, 6H, CH₂, NH), 2.25–2.49 (m, 3H, CH₂), 3.84 (m, 1H, CHNH), 3.92 (q, ³ $J_{\rm HH}$ =6.3 Hz, 1H, CH₃CH), 4.12 (q, ³ $J_{\rm HH}$ =7.1 Hz, 1H, OCH₂), 4.14 (q, ³ $J_{\rm HH}$ =7.1 Hz, 1H, OCH₂), 7.16–7.37 (m, 6H, CH=C, CH_{arom}) ppm. $[\alpha]_{\rm D}^{20}$ = +33.5 (CHCl₃, c=0.2) [from *R*-(+) PhCH(Me)NH₂, $[\alpha]_{\rm D}^{20}$ = +27.4 (MeOH, c= 2.4)]. IR and MS spectra were identical to those of the isomer **8a**.

4.2.10. (-)-7-(1-Phenyl-ethylamino)-cyclohept-1-enecarboxylic acid ethyl ester (8b'). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.45. Yield: 25% (13.0 mg from 60.0 mg of 1b); colorless oil; ¹H NMR (CDCl₃): δ =1.25 (t, ³ $J_{\rm HH}$ =7.2 Hz, 3H, OCH₂CH₃), 1.40 (d, ³ $J_{\rm HH}$ =6.4 Hz, 3H, CH₃CH), 1.57–2.04 (m, 6H, CH₂, NH), 2.23–2.53 (m, 3H, CH₂), 3.85–3.90 (four lines, ³ $J_{\rm HH}$ = 2.1, 6.3 Hz, 2H, CH₃CH, CHNH), 4.12 (q, ³ $J_{\rm HH}$ =7.2 Hz, 1H, OCH₂), 4.13 (q, ³ $J_{\rm HH}$ =7.1 Hz, 1H, OCH₂), 7.11–7.23 (m, 1H, CH=C), 7.28–7.35 (m, 5H, CH_{arom}) ppm. [α]_D²⁰= -6.8 (CHCl₃, c=0.25) [from *R*-(+) PhCH(Me)NH₂, [α]_D²⁰= +27.4 (MeOH, c=2.4)]. IR and MS spectra were identical to those of the isomer **8a**'.

4.2.11. 6-Azido-cyclohex-1-enecarboxylic acid ethyl ester (9). R_f (ethyl acetate/petroleum ether 1:1)=0.74. Yield: 65% (40.0 mg from 97.0 mg of **1a**); colorless oil; ¹H NMR (CDCl₃): δ =1.32 (t, ³ J_{HH} =7.1 Hz, 3H, OCH₂CH₃), 1.50–1.80 (m, 4H, CH₂), 1.90–2.42 (m, 2H, CH₂), 4.25 (q, ³ J_{HH} =7.1 Hz, 2H, OCH₂), 4.45 (m, 1H, CHN₃), 7.23 (dd, ³ J_{HH} =2.4, 2.4 Hz, 1H, CH=C) ppm. ¹³C NMR (CDCl₃): δ =14.3 (OCH₂CH₃), 22.8, 24.1, 26.2 (CH₂), 55.2 (CHN₃), 60.9 (OCH₂CH₃), 131.8 (C=CH), 140.3 (HC=C), 167.8 (C=O). IR (film, cm⁻¹): 2100 ν (N₃), 1706 ν (C=O). MS (CI, isobutane): m/z 196 [M+H]⁺. HRMS (CI) calc. for C₉H₁₃O₂N₃+H [M+H]⁺ 196.1086; found: 196.1081.

4.2.12. 6-Benzylamino-5-methyl-cyclohex-1-enecar-boxylic acid ethyl ester (12). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.71. Yield: 31% (31.0 mg from 130.0 mg of **1c**); colorless oil; ¹H NMR

4.2.13. 6-Benzylamino-3-methyl-cyclohex-1-enecarboxylic acid ethyl ester (13). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.49. Yield: 49% (50.6 mg from 130.0 mg of **1c**); colorless oil; ¹H NMR (CDCl₃): δ =1.05 (d, ³ $J_{\rm HH}$ =7.3 Hz, 3H, CH₃CH), 1.22 (m, 1H, CH₂), 1.28 (t, ³ $J_{\rm HH}$ =7.1 Hz, 3H, OCH₂CH₃), 1.70 (m, 1H, CH₂), 1.80 (m, 1H, CH₂), 1.87 (bs, 1H, NH), 1.95 (m, 1H, CH₂), 2.40 (m, 1H, CH), 3.61 (dd, ³ $J_{\rm HH}$ =5.1, 5.1 Hz, 1H, CHNH), 3.79 (AB, ³ $J_{\rm HH}$ =7.1 Hz, 2H, NHCH₂), 4.19 (q, ³ $J_{\rm HH}$ =7.1 Hz, 2H, OCH₂), 6.92 (d, ³ $J_{\rm HH}$ =5.0 Hz, 1H, CH=C), 7.20–7.36 (m, 5H, CH_{arom}) ppm. ¹³C NMR (CDCl₃): 14.5 (OCH₂CH₃), 19.7 (CH₂), 20.5 (CH₃), 24.7 (CH₂), 30.0 (CHCH₃), 52.0 (NCH₂Ph), 55.4 (CHNH), 60.1 (OCH₂CH₃), 126.1 (*C*=CH), 126.9, 127.1, 128.0 (C_{arom}), 141.70 (C_{*ipso*}), 142.2 (HC=C), 165.5 (C=O). IR (film, cm⁻¹): 3400 ν (NH), 1705 ν (C=O). MS (CI, isobutane): *m*/*z* 274 [M+H]⁺. HRMS (CI) calc. for C₁₇H₂₄O₂N+H [M+H]⁺ 274.1807; found: 274.1810.

4.2.14. 6-Azido-5-methyl-cyclohex-1-enecarboxylic acid ethyl ester (14) and 6-Azido-3-methyl-cyclohex-1-enecarboxylic acid ethyl ester (15). Ratio of regioisomers: 2:1: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.73 and 0.66. Yield: 70% (41.0 mg from 95.0 mg of 1c); colorless oil; ¹H NMR (CDCl₃): $\delta = 0.97$ (d, ${}^{3}J_{HH} = 7.7$ Hz, 3H, CH₃CH, major), 1.05 (d, ${}^{3}J_{HH}$ = 7.3 Hz, 3H, CH₃CH, minor), 1.32 (t, ${}^{3}J_{\rm HH} = 7.1$ Hz, 6H, OCH₂CH₃), 1.34–1.50 (m, 2H, CH₂), 1.55–2.08 (m, 5H, CH₂), 2.15–2.35 (m, 2H, CH₂, CH), 2.41 (m, 1H, CH), 4.05 (d, ${}^{3}J_{HH}$ = 3.8 Hz, 1H, CHN₃, major), 4.25 (m, 4H, OCH₂), 4.38 (m, 1H, CHN₃, minor), 7.07 (d, ${}^{3}J_{\text{HH}}$ =4.1 Hz, CH=C, 1H, minor), 7.19 (dd, ${}^{3}J_{\text{HH}}$ =3.9, 3.9 Hz, 1H, CH=C, major) ppm. 13 C NMR (CDCl₃): δ = 13.6, 14.1 (OCH₂CH₃), 14.8 (CH₃, major), 20.9, 22.4, 24.3, 25.1 (CH₂, major and minor), 30.6, 35.1 (CHCH₃, major and minor), 54.2 (CHN₃, minor), 57.8 (CHN₃, major), 61.7 (OCH₂CH₃, major), 63.1 (OCH₂CH₃, minor), 131.5 (C=CH, minor), 133.0 (C=CH, major), 140.3 (HC=C, major), 143.2 (HC=C, minor), 167.8 (C=O, major and minor). IR (film, cm⁻¹): 2112, 2110 ν (N₃), 1710 ν (C=O). MS (CI, isobutane): m/z 210 [M+H]⁺. HRMS (CI) calc. for $C_{10}H_{15}O_2N_3 + H [M+H]^+$ 210.1242; found: 210.1247.

4.2.15. 6-Diethylamino-3-methyl-cyclohex-1-enecarboxylic acid ethyl ester (16). $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.78. Yield: 78% (75.0 mg from 128.0 mg of **1c**); colorless oil; ¹H NMR (CDCl₃): δ =0.97 (t, ³ $J_{\rm HH}$ = 7.0 Hz, 6H, NCH₂CH₃), 0.99 (d, ³ $J_{\rm HH}$ =7.1 Hz, 3H, CH₃CH), 1.28 (t, ³ $J_{\rm HH}$ =7.1 Hz, 3H, OCH₂CH₃), 1.52 (m, 1H, CH₂), 1.67–1.95 (m, 2H, CH₂), 2.08–2.33 (m, 2H, CH, CH₂), 2.30–2.72 (m, 4H, NCH₂), 3.80 (m, 1H, CHN), 4.18 (q, ${}^{3}J_{\text{HH}}$ =7.1 Hz, 2H, OCH₂), 6.44 (d, ${}^{3}J_{\text{HH}}$ =2.7 Hz, 1H, CH=C) ppm. 13 C NMR (CDCl₃): δ =14.1 (OCH₂CH₃), 20.0 (CH₂), 21.6 (CH₃), 27.2 (CH₂), 29.7 (NCH₂CH₃), 30.2 (CHCH₃), 45.8 (NCH₂CH₃), 56.1 (CHNH), 61.8 (OCH₂CH₃), 126.4 (C=CH), 135.5 (CH=C), 168.0 (C=O). IR (film, cm⁻¹): 1715 ν (C=O). MS (CI, isobutane): m/z 240 [M+H]⁺. HRMS (CI) calc. for C₁₄H₂₅O₂N+H [M+H]⁺ 240.1963; found: 240.1959.

4.2.16. 6-Benzylamino-5-phenyl-cyclohex-1-enecarboxylic acid ethyl ester (17). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.65. Yield: 58% (70.0 mg from 140.0 mg of 1d); colorless oil; ¹H NMR (CDCl₃, 500 MHz): $\delta = 1.32$ (t, ${}^{3}J_{HH} = 7.1$ Hz, 3H, OCH₂CH₃), 1.82 (dd, ${}^{3}J_{HH} = 6.7$, 12.8 Hz, 1H, CH₂), 2.04 (ddd, ${}^{3}J_{\rm HH}$ =4.0, 8.2, 13.5 Hz 1H, CH₂), 2.16 (ddd, ${}^{3}J_{\rm HH}$ = 4.1, 8.6, 12.9 Hz 1H, CH₂), 2.23 (2×q, ${}^{3}J_{\rm HH}$ =4.4 Hz, 1H, CH₂), 2.30 (bs, 1H, NH), 3.19 (dt, ${}^{3}J_{HH} = 4.1$, 7.3 Hz, 1H, PhCH) 3.79 (AB, ${}^{2}J_{HH} = 12.6$ Hz, 2H, CH₂NH), 3.93 (d, ${}^{3}J_{\text{HH}}$ =4.0 Hz, 1H, *CH*NH), 4.25 (q, ${}^{3}J_{\text{HH}}$ =7.0 Hz, 2H, OCH₂), 7.15 (dd, ${}^{3}J_{\text{HH}}$ =3.8, 3.8 Hz, 1H, CH=C), 7.19– 7.32 (m, 10H, CH_{arom}) ppm. COSY (¹H-¹H): cross peak $[\delta = 3.19 \text{ (PhCH) and } \delta = 3.93 \text{ (CHNH)}, \delta = 7.15 \text{ (CH=C)}$ and $\delta = 2.04, 2.23 \text{ (CH_2)}]$. ¹³C NMR (CDCl₃): 14.3 (OCH₂CH₃), 23.6, 25.7 (CH₂), 42.5 (CHPh), 51.8 (NCH₂Ph), 55.9 (CHNH), 60.5 (OCH₂CH₃), 126.2, 126.9, 127.5, 128.3, 128.3 (C_{arom}), 132.5 (C=CH), 140.4 (C_{ipso}), $142.7 (CH=C), 143.7 (C_{ipso}), 167.4 (C=O). IR (film, cm^{-1}):$ 3407 v(N-H), 1715 v(C=O). MS (CI, isobutane): m/z 336 $[M+H]^+$. HRMS (CI) calc. for $C_{22}H_{25}O_2N + H [M+H]^+$ 336.1963; found: 336.1969.

4.2.17. 6-Diethylamino-3-phenyl-cyclohex-1-enecarboxylic acid ethyl ester (18). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2)=0.70. Yield: 60% (47.0 mg from 100.0 mg of 1d); colorless oil; ¹H NMR $(CDCl_3): \delta = 0.86 \text{ (t, }^{3}J_{HH} = 7.1 \text{ Hz, } 6\text{H, NCH}_2CH_3), 1.27 \text{ (t, }$ ${}^{3}J_{\text{HH}} = 7.2 \text{ Hz}, 3\text{H}, \text{ OCH}_2\text{CH}_3), 1.61-1.89 \text{ (m, 2H, CH}_2),$ 1.92-2.31 (m, 2H, CH₂), 2.33-2.62 (m, 4H, NCH₂), 3.10 (m, 1H, PhCH), 4.02 (ddd, ${}^{3}J_{HH}$ =4.0, 2.0, 2.0 Hz, 1H, CHN), 4.20 (q, ${}^{3}J_{HH}$ =7.1 Hz, 2H, OCH₂), 6.81 (d, ${}^{3}J_{HH}$ = 1.8 Hz, 1H, CH=C), 7.21–7.51 (m, 5H, CH_{arom}) ppm. ¹³C NMR (CDCl₃): $\delta = 14.1$ (OCH₂CH₃), 21.6, 29.7 (CH₂), 27.4 (NCH₂CH₃), 41.5 (CHPh), 45.8 (NCH₂CH₃), 52.4 (CHNH), 60.8 (OCH₂CH₃), 126.9, 127.5, 128.3 (C_{arom}), 132.8 (C=CH), 142.1 (C_{ipso}), 143.4 (CH=C), 168.0 (C=O). IR (film, cm⁻¹): 1711 ν (C=O). MS (CI, isobutane): m/z 302 $[M+H]^+$. HRMS (CI) calc. for $C_{19}H_{27}O_2N + H[M+H]^+$ 302.2120; found: 302.2125.

4.2.18. 5-Phenyl-cyclohexa-1,5-dienecarboxylic acid ethyl ester (19).¹⁴ $R_{\rm f}$ (ethyl acetate/petroleum ether 1:2) = 0.90. Yield: 28% (23.0 mg from 140.0 mg of **1d**); colorless oil; ¹H NMR (CDCl₃): $\delta = 1.33$ (t, ³ $J_{\rm HH} = 7.1$ Hz, 3H, OCH₂CH₃), 2.35–2.80 (m, 4H, CH₂), 4.26 (q, ³ $J_{\rm HH} =$ 7.1 Hz, 2H, OCH₂), 6.81 (dd, ⁴ $J_{\rm HH} = 1.2$, 1.3 Hz, 1H, C=CHC) 7.01 (ddd, ³ $J_{\rm HH} = 4.6$, 4.4 Hz, ⁴ $J_{\rm HH} = 1.2$ Hz, 1H, CH₂CH=C), 7.15–7.65 (m, 5H, CH_{arom}) ppm. IR (film, cm⁻¹): 1710 ν (C=O). MS (CI, isobutane): *m*/*z* 229 [M+H]⁺.

4.3. Hydrogenolysis

Amine 7a (70 mg, 0.25 mmol), dissolved in MeOH (20 mL) was hydrogenated in the presence of Pd/C (50 mg) during 48 h at r.t. Then the catalyst was filtered off and solution was concentrated in vacuo to give the mixture (36 mg, 80%) of pure (1R, 2S)-cis^{7b} and (1S, 2S)-trans^{7c} 2-aminocyclohexanecarboxylic acid ethyl ester as colorless oil. The mixture was analyzed by GC (column hp1, 30 m, temperature gradient: 40 °C, 2 min; 10 °C/min, detector temperature 260 °C) retention time (min): minor isomer 8.21 (40.2%), major isomer 8.82 (59.8%), $[\alpha]_D^{20} = +35.6$ (EtOH, c = 0.5), ¹H NMR (CDCl₃, 500 MHz, ¹H–¹H COSY): $\delta = 1.27$ (t, ${}^{3}J_{\text{HH}} = 7.1 \text{ Hz}, 3\text{H}, \text{OCH}_{2}\text{CH}_{3}, \text{ major and minor}, 1.35-1.45$ (m, 4H, CH₂, major and minor), 1.73 (br s, 2H, CH₂, major and minor), 1.80-1.87 (m, 2H, CH₂, major and minor), 2.20 (t, ${}^{3}J_{HH} = 5.9$ Hz, 1H, CH, major), 2.33–2.38 (m, 3H, CH and NH, minor), 2.75 (t, ${}^{3}J_{HH}$ =5.9 Hz, 1H, CH, major) 3.4 (t, ${}^{3}J_{HH}$ =6.2 Hz, 1H, CH, minor), 3.77 (br s, 2H, NH major), 4.22 (q, ${}^{3}J_{HH}$ =7.1 Hz, 2H, OCH₂, major and minor). 13 C NMR (CDCl₃): δ =14.1 (OCH₂CH₃, major and minor), 24.1 (CH2, major), 24.5, 24.7 (CH2, minor), 25.6 (CH₂, major), 25.8 (CH₂, minor), 28.5 (CH₂, major), 30.0 (CH₂, minor), 32.9 (CH₂, major), 47.1, 49.6 (CH, minor), 51.5, 52.4 (CH, major), 60.1 (OCH₂CH₃, minor), 61.3 (OCH₂CH₃, major), 173.9 (C=O, major). 174.1 (C=O, minor). MS (CI, isobutane): m/z 172 [M+H]⁺, 100%; 158 [M–N]⁺, 56%. Calculated the rotary power of the mixture of the 2-amino-cyclohexanecarboxylic acid ethyl ester: $[\alpha]_{D \text{ calcd}}^{20} = (+56.3)^{7c} (0.598) + (-2.7)^{7b} (0.402) = +32.6,$ a value in good agreement with the experimental value.^{7d}

According to the described procedure the mixture of (1*S*, 2*R*)-*cis* and (1*R*, 2*R*)-*trans* 2-amino-cyclohexanecarboxylic acid ethyl ester^{7a} from amine **7a**' was obtained as colorless oil. GC: retention time (min) minor isomer 8.41 (44.8%), major isomer 8.98 (55.2%), $[\alpha]_D^{20} = -27.9$ (EtOH, c = 0.75), ¹H NMR (CDCl₃): $\delta = 1.26$ (t, ³*J*_{HH}=7.1 Hz, 3H, OCH₂CH₃, major and minor), 1.34–1.78 (m, 4H, CH₂, major and minor), 2.73–2.98 (m, 3H, NH minor, CH major), 3.02–3.36 (br s, 1H, CH, minor), 4.06–4.22 (m, 4H, NH, OCH₂, major and minor). $[\alpha]_{D\ calcd}^{20} = (-52.9)^{7a}$ (0.552)+(+0.9)^{7a} (0.448)= -28.7. MS spectrum was identical to that of the mixture from amine **7a** obtained.

4.4. Synthesis of primary allylic amine 10

Triphenylphosphine and traces of water were added to a solution of azide **9** (0.31 mmol, 60 mg) in THF (5 mL) and resulted mixture was stirred at room temperature for 6 h. The precipitated triphenylphosphine oxide was filtered off and after evaporation of solvent the crude product was analyzed by ¹H NMR spectroscopy.

4.4.1. 6-Amino-cyclohex-1-enecarboxylic acid ethyl ester (**10**). Yield: 82% (42 mg); colorless oil; ¹H NMR (CDCl₃): $\delta = 1.29$ (t, ³ $J_{HH} = 7.0$ Hz, 3H, OCH₂CH₃), 1.44–1.79 (m, 4H, CH₂), 2.04–2.37 (m, 4H, CH₂, NH₂), 3.92 (dd, ³ $J_{HH} =$ 6.8 Hz, 1H, CHNH₂), 4.21 (q, ³ $J_{HH} = 7.0$ Hz, 2H, OCH₂), 7.32 (dd, ³ $J_{HH} = 2.8$, 2.8 Hz, 1H, CH=C). IR (film, cm⁻¹): 3414 ν (NH), 1698 ν (C=O). MS (CI, isobutane): m/z 170 $[M+H]^+$. HRMS (CI) calc. for $C_9H_{15}O_2N+H [M+H]^+$ 170.1181; found: 170.1176.

4.5. Synthesis of allylic phosphonates 11, 20 and 21

To a solution of allylic phosphates **1a**, **1c** or **1d** (0.3 mmol) in dry toluene (5 mL), 0.6 mmol of trimethyl phosphite was added and the resulting mixture was stirred with heating up 60–90 °C under Ar. Progress of the reaction was followed by TLC chromatography. When the reaction was complete, solvent and volatile products were removed in vacuo and residue was purified by silica gel chromatography (*n*-hexane/EtOAc) to provide a pure allylic phosphonates **11**, **20** and **21** as a colorless oils.

4.5.1. 6-(Dimethoxy-phosphoryl)-cyclohex-1-enecarboxylic acid ethyl ester (11). $R_{\rm f}$ (ethyl acetate/petroleum ether 2:1)=0.50. Yield: 69% (77.0 mg from 130.0 mg of **1a**); colorless oil; ³¹P NMR (CDCl₃): δ =31.5 ppm. ¹H NMR (CDCl₃): δ =1.29 (t, ³ $J_{\rm HH}$ =7.2 Hz, 3H, OCH₂CH₃), 1.55–1.82 (m, 2H, CH₂CHP), 1.91–2.15 (m, 1H, CH₂), 2.19–2.31 (m, 3H, CH₂), 3.28 (ddd, ² $J_{\rm HP}$ =24.0 Hz, ³ $J_{\rm HH}$ = 1.3, 5.3 Hz, 1H, CHP), 3.68 (d, ³ $J_{\rm HP}$ =10.8 Hz, 3H, OCH₃), 3.74 (d, ³ $J_{\rm HP}$ =10.7 Hz, 3H, OCH₃), 4.19 (2×q, ³ $J_{\rm HH}$ = 7.2 Hz, 2H, OCH₂), 7.03 (dd, ³ $J_{\rm HH}$ =4.4, 8.4 Hz, 1H, CH=C) ppm. ¹³C NMR (CDCl₃): δ =14.2 (s, OCH₂CH₃), 17.9 (d, ⁴ $J_{\rm CP}$ =1.3 Hz, CH₂), 22.5 (d, ³ $J_{\rm CP}$ =4.4 Hz, CH₂), 24.9 (d, ² $J_{\rm CP}$ =7.3 Hz, OCH₃), 52.6 (d, ² $J_{\rm CP}$ =7.3 Hz, OCH₃), 60.6 (s, OCH₂), 126.5 (d, ² $J_{\rm CP}$ =8.5 Hz, CH=C), 141.9 (d, ³ $J_{\rm CP}$ =10.4 Hz, CH=C), 166.8 (s, C=O) ppm. IR (film, cm⁻¹): 1712 ν (C=O), 1645 ν (C=C), 1249 ν (P=O). MS (CI, isobutane): m/z 263 [M+H]⁺. HRMS (CI) calc. for C₁₁H₁₉O₅P+H [M+H]⁺ 263.1048; found: 263.1054.

4.5.2. 6-(Dimethoxy-phosphoryl)-3-methyl-cyclohex-1enecarboxylic acid ethyl ester (20). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 2:1)=0.44. Yield: 70% (63.0 mg from 107.0 mg of 1c); colorless oil; ³¹P NMR (CDCl₃): δ 32.7 ppm. ¹H NMR (CDCl₃, 500 MHz): δ = 1.02 (d, ³ $J_{\rm HH}$ = 7.4 Hz, 3H, CH₃CH), 1.28 (t, ³ $J_{\rm HH}$ = 7.0 Hz, 3H, OCH₂CH₃), 1.70-1.81 (m, 2H, CH₂CHP), 2.02-2.21 (m, 2H, CH₂CHCH₃), 2.44 (m, 1H, CH₃CH), 3.31 (ddd, ${}^{2}J_{\text{HP}}$ = 25.0 Hz, ${}^{3}J_{HH}$ =5.2, 1.5 Hz, 1H, CHP), 3.66 (d, ${}^{3}J_{HP}$ = 10.8 Hz, 6H, OCH₃), 4.20 (q, ${}^{3}J_{HH}$ =7.1 Hz, 2H, OCH₂), 6.89 (t, ${}^{3}J_{HH} = {}^{4}J_{HP} = 4.2$ Hz, 1H, CH=C) ppm. COSY (¹H-¹H): cross peak [δ =2.44 (CHCH₃) and δ =1.02 (CH₃CH) and δ =6.89 (CH=C)]. ¹³C NMR (CDCl₃, 125 MHz): $\delta = 14.1$ (s, OCH₂CH₃), 19.1 (d, ³J_{CP}=3.9 Hz, CH₂CHCH₃), 20.7 (s, CH₃CH), 25.6 (s, CH₂CHP), 28.7 (s, CH₃CH), 31.5 (d, ${}^{1}J_{CP}$ =138.4 Hz, CHP), 52.7 (d, ${}^{2}J_{CP}$ = 7.4 Hz, OCH₃), 53.0 (d, ${}^{2}J_{CP}$ =7.4 Hz, OCH₃), 60.7 (s, OCH₂), 125, 5 (d, ${}^{2}J_{CP}$ =8.6 Hz, CH=C), 147.0 (d, ${}^{3}J_{CP}$ = 10.4 Hz, *C*H=C), 174.7 (s, C=O) ppm. IR (film, cm⁻¹): 1712 v(C=O), 1250 v(P=O). MS (CI, isobutane): m/z 277 $[M+H]^+$. HRMS (CI) calc. for $C_{12}H_{21}O_5P + H [M+H]^+$ 277.1205; found: 277.1212.

4.5.3. 6-(Dimethoxy-phosphoryl)-5-phenyl-cyclohex-1-enecarboxylic acid ethyl ester (21). Single diastereoisomer: $R_{\rm f}$ (ethyl acetate/petroleum ether 2:1)=0.47. Yield: 90% (119.0 mg from 150.0 mg of **1d**); colorless oil; ³¹P NMR (CDCl₃): δ 30.9 ppm. ¹H NMR (CDCl₃, 500 MHz):
$$\begin{split} &\delta\!=\!1.42 \ (t,\ {}^{3}J_{\rm HH}\!=\!7.1 \ {\rm Hz},\ 3{\rm H},\ {\rm OCH}_{2}{\rm C}H_{3}),\ 1.90{-}2.13 \ ({\rm m},\ 2{\rm H},\ {\rm CH}_{2}),\ 2.35 \ ({\rm m},\ 1{\rm H},\ {\rm CH}_{2}),\ 2.50 \ ({\rm m},\ 1{\rm H},\ {\rm CH}_{2}),\ 3.70{-}3.75 \ ({\rm m},\ 1{\rm H},\ {\rm seven lines},\ {}^{3}J_{\rm HP}\!=\!12.5 \ {\rm Hz},\ {\rm PhCH}),\ 3.82 \ ({\rm d},\ {}^{3}J_{\rm HP}\!=\!10.8 \ {\rm Hz},\ 3{\rm H},\ {\rm POCH}_{3}),\ 3.85 \ ({\rm d},\ {}^{3}J_{\rm HP}\!=\!10.8 \ {\rm Hz},\ 3{\rm H},\ {\rm POCH}_{3}),\ 3.85 \ ({\rm d},\ {}^{3}J_{\rm HP}\!=\!10.8 \ {\rm Hz},\ 3{\rm H},\ {\rm POCH}_{3}),\ 3.86 \ ({\rm dd},\ {}^{2}J_{\rm HP}\!=\!23.0 \ {\rm Hz},\ {}^{3}J_{\rm HH}\!<\!1 \ {\rm Hz},\ 1{\rm H},\ {\rm CHP}),\ 4.37 \ ({\rm m},\ 2{\rm H},\ {\rm OCH}_{2}),\ 7.15 \ ({\rm dd},\ {}^{4}J_{\rm HP}\!=\!4.0 \ {\rm Hz},\ {}^{3}J_{\rm HH}\!=\!3.8 \ {\rm Hz},\ 1{\rm H},\ {\rm CHP}),\ 7.20{-}7.38 \ ({\rm m},\ 5{\rm H},\ {\rm CH}_{\rm arom})\ {\rm ppm}.\ ^{1}{\rm H}\ \{^{31}{\rm P}\}\ {\rm NMR}:\ \delta\!=\!3.73 \ ({\rm six}\ {\rm lines},\ {\rm CHPh}),\ 3.82 \ ({\rm s},\ {\rm POCH}_{3}),\ 3.85 \ ({\rm s},\ {\rm POCH}_{3}),\ 3.86 \ ({\rm s},\ {}^{3}J_{\rm HH}\!<\!<1 \ {\rm Hz},\ 1{\rm H},\ {\rm CHP}),\ 7.15 \ ({\rm d},\ {}^{3}J_{\rm HH}\!=\!3.8 \ {\rm Hz},\ 1{\rm H},\ {\rm CHP}),\ 3.82 \ ({\rm s},\ {\rm POCH}_{3}),\ 3.85 \ ({\rm s},\ {\rm POCH}_{3}),\ 3.86 \ ({\rm s},\ {}^{3}J_{\rm HH}\!<\!<1 \ {\rm Hz},\ 1{\rm Hz}$$

Acknowledgements

This work was supported by the State Committee for Scientific Research, Poland (No 7T09A 148 21).

References and notes

- (a) Janecki, T. Synth. Commun. 1993, 23, 641. (b) Hoffman, H. M. R.; Rabe, J. J. J. Org. Chem. 1985, 50, 3899. (c) Hoffman, H. M. R.; Rabe, J. J. Angew. Chem., Int. Ed. Engl. 1985, 34, 94. (d) Daude, N.; Eggerd, V.; Hoffman, H. M. R. J. Chem. Soc., Chem. Commun. 1988, 206. (e) Bailey, M.; Markó, I. E.; Ollis, W. D.; Rasmussen, P. R. Tetrahedron Lett. 1990, 31, 4509.
- (a) Atkinson, R. S.; Faweett, J.; Russel, D. R.; Williams, P. J. J. Chem. Soc., Chem. Commun. 1994, 203. (b) Bailey, M.; Markó, I. E.; Ollis, W. D. Tetrahedron Lett. 1991, 32, 2686. (c) Hoveyda, A. H.; Evans, D. A.; Fu, G. C. Chem. Rev. 1993, 43, 1907. (d) Brzeziński, L. J.; Rafel, S.; Leaky, J. W. Tetrahedron 1997, 53, 16423.
- (a) Basavaiah, D.; Rao, P. D.; Hyma, R. S. *Tetrahedron* 1996, *52*, 8001. (b) Foucaud, A.; Guemmout, F. E. *Bull. Soc. Chim. Fr.* 1989, 403. (c) Basavaiah, D.; Rao, A. J.; Satyanarayana, T. *Chem. Rev.* 2003, *103*, 811 and references therein.
- 4. (a) Amri, H.; Rambaud, M.; Villiéras, J. *Tetrahedron* 1990, 46, 3535. (b) Dambrin, V.; Villiéras, M.; Janvier, P.; Toupet, L.; Amri, H.; Lebreton, J.; Villiéras, J. *Tetrahedron* 2001, 57, 2155 and references therein.
- Krawczyk, E.; Owsianik, K.; Skowrońska, A.; Wieczorek, M.; Majzner, W. New J. Chem. 2002, 26, 1703.
- Cohen, T.; Ruffner, R. J.; Shull, D. W.; Daniewski, W. M.; Ottenbrite, R. M.; Alston, P. V. J. Org. Chem. 1978, 43, 4052.
- (a) Cimarelli, C.; Palmieri, G. J. Org. Chem. 1996, 61, 5557.
 (b) Xu, D.; Prasad, K.; Repic, O.; Blacklock, T. J. Tetrahedron Asym. 1997, 8, 1445. (c) Armarego, W. L. F.; Kobayashi, T. J. Chem. Soc. C 1969, 1635. (d) Demailly, G.; Solladie, G. J. Org. Chem. 1981, 46, 3102.
- 8. (a) Yanagisawa, A.; Noritake, Y.; Nomura, N.; Yamamoto, H.

Synlett **1991**, 251. (b) Yanagisawa, A.; Noritake, Y.; Nomura, N.; Yamamoto, H. *Synthesis* **1991**, 1130.

- 9. Yamagisawa, A.; Namura, N.; Yamamoto, H. Synlett 1991, 513.
- (a) Magid, R. M. *Tetrahedron* **1980**, *36*, 1901. (b) De Wolfe,
 R. H.; Young, W. G. *Chem. Rev.* **1956**, *56*, 753. (c) Paquette,
 L. A.; Stirling, C. J. M. *Tetrahedron* **1992**, *48*, 7383.
- (a) Stork, G.; Kreft, A. F. III J. Am. Chem. Soc. 1977, 99, 3850.
 (b) Stork, G.; Kreft, A. F. III J. Am. Chem. Soc. 1977, 99, 8373.
 (c) Oritani, T.; Overton, K. H. J. Chem. Soc., Chem. Commun. 1978, 454.
 (d) Stohrer, W. D. Angew. Chem., Int. Ed. Engl. 1983, 22, 613.
 (e) Bach, R. D.; Wolber, G. J. J. Am. Chem. Soc. 1985, 107, 1352.
- 12. Deslongchamps, P. Stereoelectronic Effects in Organic Chemistry; Pergamon: Oxford, 1983; p 174.

- 13. Buchholz, R.; Hoffmann, H. M. R. *Helv. Chim. Acta* **1991**, *74*, 1213.
- 14. Best, W. M.; Widdowson, D. A. Tetrahedron 1989, 5943.
- 15. The conformations of *cis* and *trans* isomers were fully explored and minimized using the AM-1 program of the Hyper Chem package (Release 6.03 for Windows, Molecular Modeling System).
- Quin, L. D. In *Phosphorus-31 NMR Spectroscopy in Stereo-chemical Analysis: Organic Compounds and Metal Complex*; Verkade, J. G., Quin, L. D., Eds.; VCH: Weinheim, 1987; Chapter 12, p 402.
- 17. Fiaud, J.-C. J. Chem. Soc., Chem. Commun. 1983, 1055.
- Dobbie, A. H.; Overton, K. H. J. Chem. Soc., Chem. Commun. 1977, 722.