DOI: 10.1002/adsc.201201111

Regio- and Stereoselective Synthesis of Tri- and Tetrasubstituted Enamides *via* **Palladium-Catalyzed Silaboration of Ynamides**

Nozomi Saito,^{a,*} Keiichi Saito,^a Hiroyasu Sato,^b and Yoshihiro Sato^{a,*}

^a Faculty of Pharmaceutical Sciences, Hokkaido University, Sapporo 060-0812, Japan

Fax: (+81)-11-706-4982; e-mail: nozomi-s@pharm.hokudai.ac.jp or biyo@pharm.hokudai.ac.jp

^b Rigaku Corporation, Akishima, Tokyo 196-8666, Japan

Received: December 19, 2012; Published online: March 15, 2013

Supporting information for this article is available on the WWW under http://dx.doi.org/10.1002/adsc.201201111.

Abstract: The palladium-catalyzed silaboration of ynamides is demonstrated. The silaboration proceeds in a highly regioselective manner to give the corresponding tri- and tetrasubstituted enamide derivatives having both a silyl group and a boryl group on the alkene. Furthermore, the silaborated enamide could be utilized as a coupling partner in Suzuki–Miyaura coupling with aryl iodides to give the corresponding cross-coupling product in good yield.

Keywords: boron; metallation; palladium; silicon; ynamides

The transition metal-catalyzed addition of an element-element σ -bond of a bimetallic reagent (X-Y) to carbon-carbon unsaturated compounds (i.e., bismetallation) is one of the most important methods for the synthesis of 1,n-dimetallo compounds, which are expected to be useful building blocks for further organic transformations. Group 10 metals are widely employed for bismetallation reactions.^[1] The generally accepted mechanism of bismetallation is shown in Scheme 1. First, oxidative addition of a bimetallic reagent to a low-valent transition metal (M) proceeds to give the the X-M-Y complex I. Insertion of a multiple bond in the substrate into either the M-X or the M-Y σ -bond in I occurs to produce intermediate II or II', from which reductive elimination affords bismetallated product III.

The *N*-alkynylamide (ynamide) moiety has been recognized as a versatile building block in recent synthetic organic chemistry,^[2–4] and we have already reported a new transformation of ynamides by virtue of the use of transition metal catalysis.^[5] The aforementioned bismetallation of ynamides is expected to be an attractive strategy for the synthesis of multi-substi-



Scheme 1. Transition metal-catalyzed bismetallation of carbon-carbon unsaturated bonds.

tuted enamides having two metal components, and palladium-catalyzed silastannylation and bisstannylation of ynamides have been reported to date.^[6,7] However, only a terminal alkyne has been employed in those studies, and no examples of the bismetallation of an ynamide having an internal alkyne structure have been reported so far. In this context, we planned the silaboration^[8-11] of an ynamide as shown in Scheme 2. Thus, if silaboration of ynamide 1 with silylborane (Si-B, 2) proceeds in the presence of a transition metal catalyst, the silaborated enamide 3 and/or 3' would be formed. Here, we report the palladiumcatalyzed regioselective silaboration of ynamides using commercially available (dimethylphenylsilyl)pinacolatoboron [PhMe₂SiB(pin), 4]^[12,13] that led to the formation of a multi-substituted enamide derivative bearing vinyl silicon and vinyl boron moieties.



Scheme 2. Bismetallation of ynamide leading to multi-substituted enamides.



Scheme 3. Pd-catalyzed silaboration of ynamide 5a.

Referring to the previously reported procedure,^[8a] vnamide 5a was reacted with PhMe₂SiBi(pin) (4) in the presence of $Pd(OAc)_2$ and isonitrile ligand 6 in toluene under reflux conditions. As a result, silaborated enamide 7a was obtained as a single regio- and stereoisomer in 89% yield (Scheme 3). The structure of 7a was unambiguously determined by X-ray crystallographic analysis.^[14] It is known that Pd(0)-catalyzed silaboration of alkyne proceeds through insertion of a triple bond into the Pd–B σ -bond of **IV** formed by oxidative addition of 4 to Pd(0).^[1h,15] Therefore, this result indicates that regioselective insertion of ynamide 5a into the Pd-B bond of IV occurred to produce intermediate V, from which reductive elimination took place stereoselectively to afford the enamide derivative 7a.

Encouraged by this result, we studied the scope and limitations of silaboration of ynamides (Table 1). When vnamide having a 4-methoxyphenyl group on the alkyne part 5b was reacted with silylborane 4, the expected silaborated enamide 7b and its regioisomer 8b were produced in 71% yield and 12% yield, respectively (run 1). On the other hand, silaboration of vnamide 5c bearing an ester group on the aromatic ring afforded 7c and 8c in totally 63% yield (ratio of 2 to 1). Alkyl-substituted ynamide 5d is also applicable to the silaboration, and the corresponding enamide 7d was obtained in high yield as a single isomer (run 3). Siloxy, acetoxy and alkenyl groups in substituents on the alkyne part (5e-g) were tolerated under the reaction conditions, giving enamide derivatives 7e-g in good yields in a highly regio- and stereoselective manner (runs 4-6). Furthermore, the reaction of





- [a] Reaction conditions: PhMe₂SiB(pin) (4, 1 equiv.), Pd(OAc)₂ (3 mol%), t-BuCH₂CMe₂NC (6, 30 mol%), toluene, reflux.
- ^[b] The ratio was determined by ¹H NMR analysis (400 MHz) of the mixture of **7c** and **8c**.

terminal alkyne **5h** with **4** also proceeded smoothly to give tri-substituted enamide **7h** in 81% yield as a sole product (run 7).

Next, silaboration of oxazolidinone-derived ynamides 9 was investigated (Table 2). Ynamides 9a-

 Table 2. Silaboration of oxazolidinone-derived ynamides.^[a]



 [a] Reaction conditions: PhMe₂SiB(pin) (4, 1 equiv.), Pd(OAc)₂ (3 mol%), t-BuCH₂CMe₂NC (6, 30 mol%), toluene, reflux



Scheme 4. Suzuki–Miyaura coupling of enamide 9d with aryl iodides

c having an aryl group on the alkyne part were subjected to the optimal conditions, giving the corresponding silaborated enamides **10a–c** in high yields (runs 1–3). The silaboration of ynamide **9d** bearing an alkyl group on the alkyne part gave the desired enamide **10d** in 89% yield (run 4). On the other hand, TMS group-substituted ynamide also reacted with **4** in the presence of a palladium catalyst to give expected **10e** in 55% yield and its regioisomer **11e** in 26% yield (run 5). Terminal alkyne **9f** was also applicable to the silaboration, giving **10f** in 74% yield as a single isomer (run 6).^[16]

Having established the silaboration of ynamides, we examined the utilization of silaborated enamide as a boronic ester (Scheme 4). Suzuki–Miyaura coupling of the enamide **10d** with iodobenzene (**12a**) was carried out in the presence of Pd(dppf)Cl₂ (10 mol%) and Cs₂CO₃ (5 equiv.) in DMF/H₂O at 60 °C. As a result, cross-coupling product **13da** was obtained in 91% yield. On the other hand, the coupling reaction of **10d** and **12b** gave the corresponding enamide derivative **13db** in 66% yield.

In summary, we have found that the Pd-catalyzed silaboration of ynamides having various substituents proceeds to give various tri- and tetra-substituted enamide derivatives in a highly regio- and stereoselective manner. It was also demonstrated that the silaborated enamide could be utilized as a coupling partner in Suzuki–Miyaura coupling, and the corresponding coupling products were obtained in good yields. Further studies along this line are in progress.

Experimental Section

General Procedure for Pd-Catalyzed Silaboration of Ynamides

A solution of an ynamide, $PhMe_2SiB(pin)$ (4, 1 equiv. to the ynamide), $Pd(OAc)_2$ (3 mol% to the ynamide) and 1,1,3,3-tetramethylbutylisocyanide 6 (30 mol% to the ynamide) in toluene was stirred at 110 °C. The reaction mixture was cooled to room temperature and concentrated under vacuum, and the residue was purified by flash column chro-

matography on silica gel to give the corresponding silaborated enamide.

Acknowledgements

This work was partly supported by a Grant-in-Aid for Young Scientist (B) (No. 24790002) and by a Grant-in-Aid for Scientific Research (B) (No. 23390001) from JSPS and by a Grant-in-Aid for Scientific Research on Innovative Areas "Molecular Activation Directed toward Straightforward Synthesis" (No. 23105501) from MEXT, Japan. N.S. acknowledges Takeda Science Foundation for financial support.

References

- For reviews on bismetallation of unsaturated bonds, see: a) I. Beletskaya, C. Moberg, Chem. Rev. 1999, 99, 3435; b) L.-B. Han, M. Tanaka, Chem. Commun. 1999, 395; c) Y. Ito, J. Organomet. Chem. 1999, 576, 300; d) M. Suginome, Y. Ito, J. Organomet. Chem. 2003, 680, 43; f) T. Ishiyama, N. Miyaura, Chem. Rec. 2004, 3, 271; g) I. Beletskaya, C. Moberg, Chem. Rev. 2006, 106, 2320; h) T. Ohmura, M. Suginome, Bull. Chem. Soc. Jpn. 2009, 82, 29; i) C. Pubill-Ulldemolins, A. Bonet, C. Bo, H. Gulyás, E. Fernández, Org. Biomol. Chem. 2010, 8, 2667; j) M. Oestreich, E. Hartmann, M. Mewald, Chem. Rev. 2013, 113, 402.
- [2] For reviews on the chemistry of ynamides, see: a) C. A. Zificsak, J. A. Mulder, R. P. Hsung, C. Rameshkumar, L.-L. Wei, *Tetrahedron* 2001, 57, 7575; b) J. A. Mulder, K. C. M. Kurtz, R. P. Hsung, *Synlett* 2003, 1379; c) In: *Tetrahedron Symposia-in-Print No.* 118, (Ed: R. P. Hsung), *Tetrahedron* 2006, 62, 3783; d) G. Evano, A. Coste, K. Jouvin, *Angew. Chem.* 2010, 122, 2902; *Angew. Chem. Int. Ed.* 2010, 49, 2840; e) K. A. De-Korver, H. Li, A. G. Lohse, R. Hayashi, Z. Lu, Y. Zhang, R. P. Hsung, *Chem. Rev.* 2010, 110, 5064.
- [3] For the synthesis of ynamides reported in 2012, see:
 a) X. Jin, K. Yamaguchi, N. Mizuno, *Chem. Commun.* 2012, 48, 4974; b) K. Jouvin, J. Heimburger, G. Evano, *Chem. Sci.* 2012, 3, 756; c) J. A. Souto, P. Becker, Á. Iglesias, K. Muñiz, *J. Am. Chem. Soc.* 2012, 134, 15505; d) K. Jouvin, A. Coste, A. Bayle, F. Legrand, G. Karthikeyan, K. Tadiparthi, G. Evano, *Organometallics* 2012, 31, 7933; e) Y. Yang, X. Zhang, Y. Liang, *Tetrahedron Lett.* 2012, 53, 6557.
- [4] For the utilization of ynamides in organic synthesis reported in 2012, see: a) R. B. Dateer, B. S. Shaibu, R.-S. Liu, Angew. Chem. 2012, 124, 117; Angew. Chem. Int. Ed. 2012, 51, 113; b) S. N. Karad, S. Bhunia, R.-S. Liu, Angew. Chem. 2012, 124, 8852; Angew. Chem. Int. Ed. 2012, 51, 8722; c) D. L. Smith, W. R. F. Goundry, H. W. Lam, Chem. Commun. 2012, 48, 1505; d) G. Compain, K. Jouvin, A. Martin-Mingot, G. Evano, J. Marrot, S. Thibaudeau, Chem. Commun. 2012, 48, 5196; e) R. B. Dateer, K. Pati, R.-S. Liu, Chem. Commun. 2012, 48, 7200; f) P. Garcia, Y. Evanno, P. George, M. Sevrin, G. Ricci, M. Malacria, C. Aubert, V. Gandon, Chem. Eur.

J. 2012, 18, 4337; g) C. Schotes, M. Althaus, R. Aardoom, A. Mezzetti, J. Am. Chem. Soc. 2012, 134, 1331; h) W. Gati, M. M. Rammah, M. B. Rammah, F. Couty, G. Evano, J. Am. Chem. Soc. 2012, 134, 9078; i) F. Gomes, A. Fadel, N. Rabasso, J. Org. Chem. 2012, 77, 5439; j) Z. Lu, W. Cui, S. Xia, Y. Bai, F. Luo, G. Zhu, J. Org. Chem. 2012, 77, 9871; k) Y. Minko, M. Pasco, L. Lercher, M. Botoshansky, I. Marek, Nature 2012, 490, 522; l) J. Cao, Y. Kong, Y. Deng, G. Lai, Y. Cui, Z. Hu, G. Wang, Org. Biomol. Chem. 2012, 10, 9556; m) K. A. DeKorver, X.-N. Wang, M. C. Walton, R. P. Hsung, Org. Lett. 2012, 14, 1768; n) P. Huang, Z. Chen, Q. Yang, Y. Peng, Org. Lett. 2012, 14, 2790; o) K. A. De-Korver, R. P. Hsung, W.-Z. Song, X.-N. Wang, M. C. Walton, Org. Lett. 2012, 14, 3214; p) D. L. Smith, S. R. Chidipudi, W. R. Goundry, H. W. Lam, Org. Lett. 2012, 14, 4934; q) S. Bhunia, C.-J. Chang, R.-S. Liu, Org. Lett. 2012, 14, 5522; r) C. Schotes, R. Bigler, A. Mezzetti, Synthesis 2012, 44, 513; s) Z. Lu, X. Xu, Z. Yang, L. Kong, G. Zhu, Tetrahedron Lett. 2012, 53, 3433; t) S. J. Heffernan, D. R. Carbery, Tetrahedron Lett. 2012, 53, 5180. See also, refs.^[3c,3e]

- [5] For our reports on transition metal catalysis utilizing ynamide as a platform, see: a) N. Saito, Y. Sato, M. Mori, Org. Lett. 2002, 4, 803; b) M. Mori, H. Wakamatsu, N. Saito, Y. Sato, R. Narita, Y. Sato, R. Fujita, Tetrahedron 2006, 62, 3872; c) N. Saito, T. Katayama, Y. Sato, Org. Lett. 2008, 10, 3829; d) N. Saito, T. Katayama, Y. Sato, Heterocycles 2011, 82, 1181; e) N. Saito, K. Saito, M. Shiro, Y. Sato, Org. Lett. 2011, 13, 2718; f) N. Saito, T. Ichimaru, Y. Sato, Org. Lett. 2012, 14, 1914.
- [6] For silastannylation and bisstannylation of an ynamide bearing a terminal alkyne moiety, see: a) S. Minière, J.-C. Cintrat, *Synthesis* 2001, 705; b) S. Naud, J.-C. Cintrat, *Synthesis* 2003, 1391.
- [7] For related silylcupration of an ynamide, see: H. Yasui, H. Yorimitsu, K. Oshima, *Bull. Chem. Soc. Jpn.* 2008, *81*, 373.
- [8] For silaboration of alkynes, see: a) M. Suginome, H. Nakamura, Y. Ito, Chem. Commun. 1996, 2777; b) S.-y. Onozawa, Y. Hatanaka M. Tanaka, Chem. Commun. 1997 1229; c) M. Suginome, T. Matsuda, H. Nakamura, Y. Ito, Tetrahedron 1999, 55, 8787; d) J. C. A. Da Silva, M. Birot, J.-P. Pillot, M. Pétraud, J. Organomet. Chem. 2002, 646, 179; e) M. Suginome, H. Noguchi, T. Hasui, M. Murakami, Bull. Chem. Soc. Jpn. 2005, 78, 323; f) T. Ohmura, K. Oshima, M. Suginome, Chem. Commun. 2008, 1416; g) T. Ohmura, K. Oshima, H. Taniguchi, M. Suginome, J. Am. Chem. Soc. 2010, 132, 12194; h) H. Zhou, C. Moberg, J. Am. Chem. Soc. 2012, 134, 15992.
- [9] For silaboration of simple alkenes, alkylidenecyclopropanes, vinylcyclopropanes and vinylcyclobutanes see:
 a) M. Suginome, H. Nakamura, Y. Ito, Angew. Chem. 1997, 109, 2626; Angew. Chem. Int. Ed. Engl. 1997, 36, 2516;
 b) T. Ohmura, H. Furukawa, M. Suginome, J. Am. Chem. Soc. 2006, 128, 13366;
 c) M. Suginome, T.

Matsuda, Y. Ito, J. Am. Chem. Soc. 2000, 122, 11015; d) T. Pohlmann, A. de Meijere, Org. Lett. 2000, 2, 3877; e) M. Suginome, T. Matsuda, T. Yoshimoto, Y. Ito, Organometallics 2002, 21, 1537; f) T. Ohmura, H. Taniguchi, Y. Kondo, M. Suginome, J. Am. Chem. Soc. 2007, 129, 3518; g) T. Ohmura, H. Taniguchi, M. Suginome, Org. Lett. 2009, 11, 2880.

- [10] For silaboration of 1,2-dienes, see: a) M. Suginome, Y. Ohmori, Y. Ito, Synlett 1999, 1567; b) S.-y. Onozawa, Y. Hatanaka, M. Tanaka, Chem. Commun. 1999, 1863; c) M. Suginome, Y. Ohmori, Y. Ito, J. Organomet. Chem. 2000, 611, 403; d) M. Suginome, Y. Ohmori, Y. Ito, J. Am. Chem. Soc. 2001, 123, 4601; e) M. Suginome, T. Ohmura, Y. Miyake, S. Mitani, Y. Ito, M. Murakami, J. Am. Chem. Soc. 2003, 125, 11174; f) K.-J. Chang, D. K. Rayabarapu, F.-Y. Yang, C.-H. Cheng, J. Am. Chem. Soc. 2005, 127, 126; g) T. Ohmura, M. Suginome, Org. Lett. 2006, 8, 2503; h) T. Ohmura, H. Taniguchi, M. Suginome, J. Am. Chem. Soc. 2006, 128, 13682; i) Y. Abe, K. Kuramoto, M. Ehara, H. Nakatsuji, M. Suginome, M. Murakami, Y. Ito, Organometallics 2008, 27, 1736.
- [11] For silaboration of 1,3-dienes, see: a) M. Suginome, H. Nakamura, T. Matsuda, Y. Ito, *J. Am. Chem. Soc.* 1998, *120*, 4248; b) M. Suginome, T. Matsuda, T. Yoshimoto, Y. Ito, *Org. Lett.* 1999, *1*, 1567; c) M. Gerdin, M. Moberg, *Adv. Synth. Catal.* 2005, *347*, 749; d) M. Gerdin, M. Moberg, *Org. Lett.* 2006, *8*, 2929; e) M. Gerdin, M. Penhoat, R. Zalubovskis, C. Pétermann, C. Moberg, *J. Organomet. Chem.* 2008, *693*, 3519.
- [12] M. Suginome, T. Matsuda, Y. Ito, *Organometallics* 2000, 19, 4647.
- [13] Recently, we reported nickel-catalyzed asymmetric coupling of 1,3-dienes, aldehydes, and PhMe₂SiBi(pin) (4). See: N. Saito, A. Kobayashi, Y. Sato, *Angew. Chem.* 2012, *124*, 1258; *Angew. Chem. Int. Ed.* 2012, *51*, 1228.
- [14] CCDC 884205 contains the supplementary crystallographic data for this paper (7a). These data can be obtained free of charge from The Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.
- [15] For a mechanistic study on silaboration of alkynes mediated by a stoichiometric amount of platinum complex, see: T. Sagawa, Y. Asano, F. Ozawa, *Organometallics* **2002**, *21*, 5879. For a related theoretical study on palladium-catalyzed silaboration of 1,2-diene, see: ref.^[10i]
- [16] Although the reason for the high regioselectivity of silaboration of yanmides is not clear yet, coordination of oxygen functionality of the amide group to the palladium center might affect the direction of the insertion pathway. Furthermore, steric repulsion between the bulky boryl group of **IV** and the amide group of the substrate might operate on the regioselective insertion of the triple bond into the Pd–B σ -bond.

856