

Note

Carbon-Phosphorous Bond Formation on Anilines Mediated by a Hypervalent Iodine Reagent

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Carbon-Phosphorous Bond Formation on Anilines Mediated by a Hypervalent Iodine Reagent

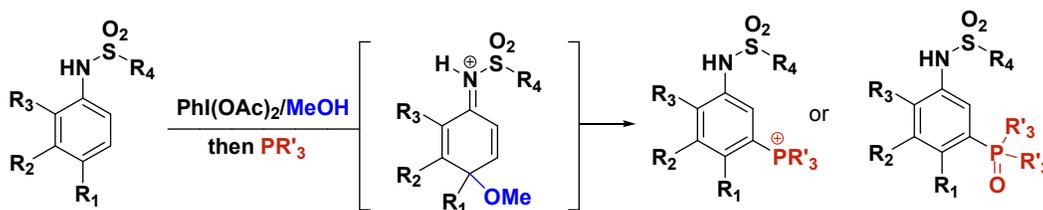
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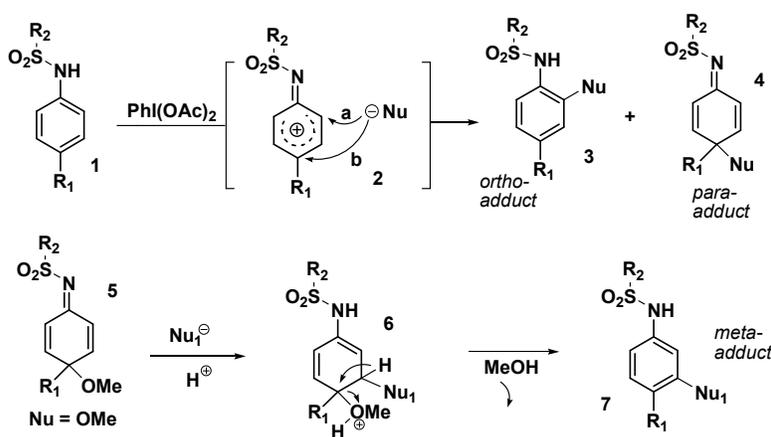
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Abstract: Substituted anilines containing a sulfonyl group may be oxidized *in situ* in the presence of methanol and a hypervalent iodine reagent to form an active iminium species. Subsequent addition of phosphines or phosphites in the same pot produces *meta*-substituted anilines in good yields. This formal C-H bond functionalization is a direct and efficient means of selectively substituting the *meta*-position of anilines to produce aromatic phosphonium ions or phosphonates.

Methods for selective and mild activation of aromatic C-H bonds have elicited strong interest in the scientific community since these methods enable one-step formation of more highly-functionalized

1 systems without pre-activation. However, most of these efficient processes are restricted in scope and
 2 require catalytic amounts of a heavy metal.¹ Therefore, the discovery of novel, efficient, and
 3 environmentally benign methodologies enabling similar transformations remains an active field of
 4 research. As an alternative to methods requiring metal catalysts, formal C-H bond activation methods
 5 mediated by hypervalent iodine reagents² have been developed for electron-rich aromatic systems.
 6 Several methods using iodanes have recently been reported in the literature, examples of which include
 7 the noteworthy work of Kita,³ Gaunt,⁴ and Fan⁵ and co-workers. Indeed, aromatic sulfonamides appear
 8 to be competent substrates, enabling selective coupling or nucleophilic addition at any position on the
 9 aromatic ring. The direct oxidation of anilines containing a sulfonyl group **1** in the presence of a
 10 nucleophile may occur at the *para* or the *ortho* positions depending on the nature of the nucleophile.
 11 With a hindered carbon-based nucleophile, the reaction occurs at the most accessible *ortho* position,⁶
 12 leading to **3**. In contrast, with smaller nucleophiles such as alcohols (methanol) the attack occurs at the
 13 most substituted position (*para*) to yield the dienimide⁷ **4**. This functionalized system (Nu = MeO) may
 14 be strategically involved in a further 1,4-addition-rearomatization to produce a *meta*-substituted
 15 aromatic in an indirect manner.⁸ Depending on the strategy and the nucleophile used, any position of an
 16 aniline may be selectively attacked, Scheme 1.



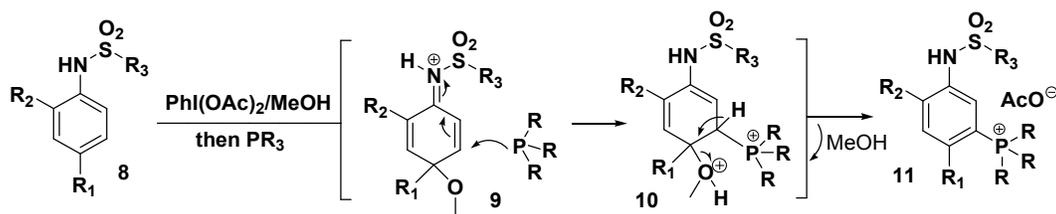
Scheme 1. Selective Substitution of Aniline derivatives

56 One interesting avenue would be to develop a method for one-pot *meta*-activation of the aniline
 57 derivative **7**. For this purpose, a reactive nucleophile stable under acidic and protic conditions (in the
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1 presence of methanol and acetic acid) is required. Candidates of choice include elements in the third
2 period of the periodic table, which are recognized for their nucleophilicity and poor basicity. In addition,
3 metal-free direct carbon-phosphorous bond formation on unactivated aromatics is not often reported in
4 the literature. Phosphonium salts are interesting reagents involved in noteworthy transformations⁹ such
5 as the Wittig reaction; they may also be employed as phase transfer catalysts¹⁰ or as potentially chiral
6 cationic counterions.¹¹

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14 In this paper, we describe a formal C-H bond functionalization mediated by an iodane-(III) species and
15 methanol to selectively produce good yields of *meta*-substituted anilines through a phosphorous moiety.
16 This oxidative process involves the *in-situ* formation of a dienimine intermediate, which is trapped by a
17 phosphorous nucleophile followed by a domino rearomatization. This transformation represents an
18 aromatic variant of the Michaelis-Arbuzov reaction¹² on an unactivated aromatic *sp*² center.

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26 Iodanes¹³ catalyze remarkable transformations in electron-rich aromatic systems, including
27 dearomatization processes and cross coupling transformations. Our interest in producing carbon-carbon
28 bonds^{6c} in such systems led us to explore new coupling routes with the formation of carbon-heteroatom
29 bonds. Carbon-oxygen and carbon-nitrogen bond formation has been extensively reported in the
30 literature,¹³ and we were therefore motivated to investigate less well-explored techniques such as the
31 direct functionalization of aromatic C-H bonds using a phosphorous group.¹⁴ However, due to the direct
32 reactivity of phosphines with iodanes to produce phosphonium oxides, it was obvious that these reagents
33 could not be mixed. A previous report indicated that activation of aniline derivatives with iodanes in the
34 presence of methanol would generate a dienimine species⁷ that could readily be transformed into an
35 iminium species by the acid released from the hypervalent iodine reagent reduction. This intermediate **9**
36 appears to be an excellent Michael acceptor. A nucleophilic species insensitive to the acidic conditions
37 generated during the oxidation could be added to the same pot to initiate a 1,4- addition process with the
38 dearomatized species **9**. Phosphines and phosphites are valuable nucleophiles in this process, leading to
39 enamines **10** which may subsequently be transformed into phosphonium ions **11** through a
40 rearomatization process beginning with methanol elimination, Scheme 2.



Scheme 2. *meta*-Carbon-Phosphorous Bond Formation

Several conditions including a variety of phosphines, sulfonamides, solvents, and iodanes were investigated. DIB was chosen as the hypervalent iodine reagent of choice but PIFA (a perfluorinated and more expensive analog) may also be used with similar yields. However, DIB appears to be a milder and safer reagent in the presence of sensitive functional groups. Methanol was selected as the solvent of choice but the reaction may also be performed in ethanol or TFE at the cost of slightly lower yields. A series of representative experimental conditions is summarized in Table 1.

entry	R ₁	R ₂	R'	R	yield (%)
a	Me	H	Me	Ph*	36
b	2-NO ₂ -Ph	Me	Me	Ph*	56
c	Me	Me	Me	<i>n</i> -Bu	85
d	4-Tol	Me	Me	<i>n</i> -Bu	78
e	2-NO ₂ -Ph	Me	Me	<i>n</i> -Bu	94
f	4-NO ₂ -Ph	Me	Me	<i>n</i> -Bu	95
g	Me	Me	Et	<i>n</i> -Bu	88
h	Me	Me	CF ₃ CH ₂	<i>n</i> -Bu	84

*48 h are required

Table 1. Conditions Optimization

The best results were obtained using electron-poor sulfonamides such as nosylamides, probably due to their electron-withdrawing effect on the Michael-acceptor moiety. A positive point is these sulfonamides can be easily removed using a Fukuyama's process.¹⁵ Aliphatic phosphines such as PBu₃ were much more reactive than the less nucleophilic aromatic derivative PPh₃. Phosphites were also competent reagents for this transformation. In our trials methanol was the best and least expensive solvent for this transformation, enabling direct functionalization of the *meta*-position. With these

conditions in hand, several substituted nosylamides were transformed into phosphonium salts in good to excellent yields. In the case of non-symmetrical anilines, coupling occurred on the less hindered alkene for steric reasons, Table 2.

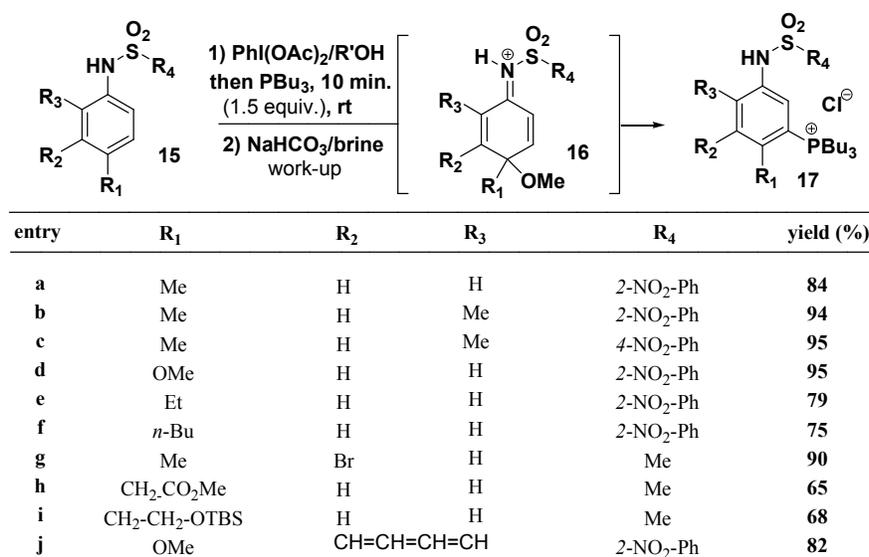
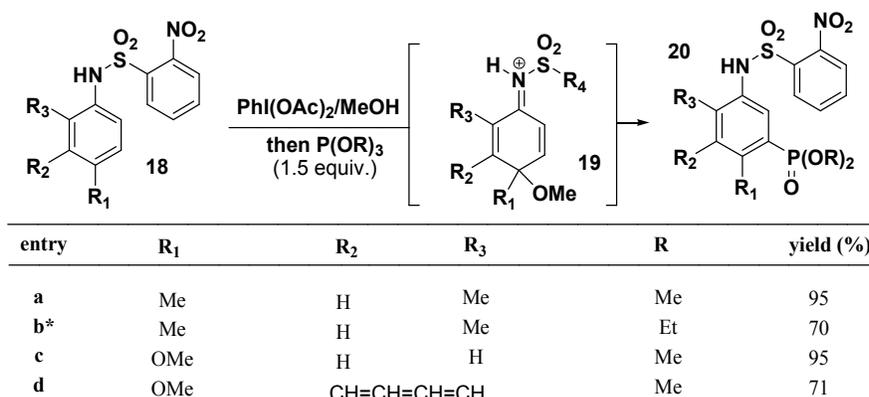


Table 2. Phosphonium Salt Formation

As the process appears quite efficient in generating aromatic phosphonium ions, we were interested in extending the transformation to the formation of phosphonates. The introduction of phosphites to the reaction resulted in formation of the expected aryl phosphonates in good yields. In the case of compounds with structure **20b**, the reaction was carried out in ethanol rather than methanol. This change avoided partial trans-esterification of the phosphonate moiety, but reduced the yield. Representative experimental conditions are described in Table 3.



* EtOH has been used instead of MeOH to avoid partial trans-esterification

Table 3. Phosphonate Formation Conditions

In order to confirm the mechanism of this transformation and the efficiency of the one pot process, the results were compared to a two-step approach in which dienimine **21** was isolated and the 1,4-addition rearomatization representing the second part of the cascade was carried out separately under aprotic conditions (acetonitrile) at 82 °C. The expected *meta*-substituted anilines were obtained in fair to good yields. While the same products were formed in each protocol, the one-pot transformation occurred more rapidly and in better yield (examples **22a**, **22b** and **22e**), demonstrating that the process is more efficient under protic conditions mediated by an iminium species **16** as Michael acceptor, Table 4.

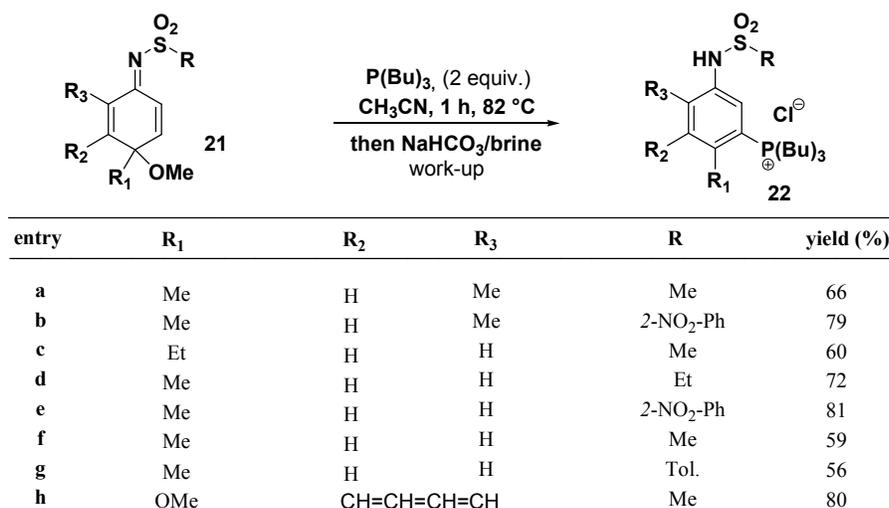


Table 4. Tributyl-phosphine 1,4-addition-Rearomatization Domino Process

We have developed a selective oxidative method enabling production of a carbon-phosphorous bond at the *meta*-position of aniline derivatives. This process occurs in one pot via the formation of a dieniminium intermediate that is readily transformed into the substituted aromatic system through a domino 1,4-addition and rearomatization process. This transformation is equivalent to a C-H bond functionalization without the need for metal catalysts.

Experimental Section:

Unless otherwise indicated, ¹H and ¹³C NMR spectra were recorded at 300 and 75 MHz, respectively, in CDCl₃ solutions. Chemical shifts are reported in ppm on the δ scale. Multiplicities are described as s (singlet), d (doublet), dd, ddd, etc. (doublet of doublets, doublet of doublets of doublets, etc.), t (triplet), q (quartet), quin (quintuplet), m (multiplet), and further qualified as app (apparent), br (broad). Coupling

1 constants, J, are reported in Hz. HRMS were measured in the electrospray (ESI) mode on a LC-MSD
2 TOF mass analyzer.
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4 **a) General procedure for the preparation of sulfonamide products 12/15**
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8 To a solution of the corresponding aniline (1 mmol, 1.0 equiv.) and pyridine (2.4 mmol, 2.4 equiv.) in
9 CH₂Cl₂ (10 mL) at 0 °C the corresponding sulfonyl chloride or nitrobenzenesulfonyl chloride (1.1
10 mmol, 1.1 equiv.) was added slowly. The reaction mixture was allowed to warm to room temperature
11 and stirred overnight. After complete conversion of the starting material (followed by TLC), the reaction
12 was quenched with a solution of HCl (1M). The phases were separated and the aqueous phase extracted
13 with CH₂Cl₂. The organic layers was washed with brine then, the organic phases were dried over
14 Na₂SO₄, filtered and concentrated under vacuum. The residue was purified by silica gel chromatography
15 (DCM/hexanes or EtOAc/hexanes)/recrystallization in ethanol or without purification to give the desired
16 sulfonamides products **12/15**.
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27 **b) General procedure for the formation of phosphonium/phosphonate product 14/17/20:**
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30 Iodobenzene diacetate ("DIB", 0.15 mmol, 1.5 equiv.) was added to a vigorously stirred solution of
31 sulfonamide **12/15/18** (0.1 mmol, 1 equiv.) in a minimum of MeOH (0.5 mL) at room temperature. The
32 mixture was then stirred for 5 min, and phosphine or phosphite (0.15 mmol, 1.5 equiv.) was added (the
33 reaction was followed by TLC with a mixture of ethyl acetate/hexane, 10 min. with PBU₃ or phosphite
34 and 48 hours with PPh₃). The crude was concentrated under vacuum and the residue was quenched with
35 a saturated solution of NaHCO₃. The phases were separated and the aqueous phase extracted with
36 CH₂Cl₂. The organic layers was washed with brine then, the organic phases were dried over Na₂SO₄,
37 filtered and concentrated under vacuum. The residue was purified by silica gel chromatography with a
38 mixture of CH₂Cl₂/MeOH excepted for compounds **20a-d** (mixture of ethyl acetate/hexane) to give
39 corresponding phosphonium/phosphonate products **14/17/20**.
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49 **c) General procedure for the formation of compounds 21:**
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53 Iodobenzene diacetate («DIB», 2 mmol, 2 equiv.) was added to a vigorously stirred solution of
54 sulfonamide (1 mmol, 1 equiv.) in MeOH (5 mL) at 0°C. The mixture was then stirred for 5 min.
55 (followed by TLC with a mixture of ethylacetate/hexane) and then concentrated under vacuum. The
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residue was purified by silica gel chromatography with a mixture of ethyl acetate/hexane to give product **21**.

d) General procedure for the formation of phosphonium product 22:

To a solution of **21** (0.1 mmol, 1 equiv.) in acetonitrile (1 mL) was added tributylphosphine (2.0 mmol, 2.0 equiv.). The mixture was then stirred at 82°C between 20 min. and 2 hours. After completion, the reaction mixture was concentrated under reduced pressure. The residue was purified by silica gel chromatography with a mixture of CH₂Cl₂/MeOH or by recrystallization in diethyl ether to afford **22**.

N-p-tolylmethanesulfonamide (12a)¹⁶: was obtained without purification as a white solid: 0.9 mmol, 167 mg, 90% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.15 (br, 4H), 7.08 (s, 1H), 2.98 (s, 3H), 2.32 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 135.6, 134.2, 130.3, 121.7, 39.0, 20.9.

N-(2,4-dimethylphenyl)-2-nitrobenzenesulfonamide (12b): was obtained without purification as a brown solid: 0.95 mmol, 291 mg, 95% yield; Mp: 102-104 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.87 (d, *J* = 7.2 Hz, 1H), 7.82 – 7.67 (m, 2H), 7.61 (dt, *J* = 7.6, 3.8 Hz, 1H), 7.05 (br, 1H), 7.03 (s, 1H), 6.97 (br, 1H), 6.91 (d, *J* = 8.0 Hz, 1H), 2.27 (s, 3H), 2.16 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 148.0, 137.5, 133.9, 133.5, 132.8, 131.9, 131.5, 131.1, 127.5, 126.2, 125.3, 20.98, 17.83; LRMS (ESI): Calc. for C₁₄H₁₅N₂O₄S (M+H)⁺: 307; found: 307.

N-(2,4-dimethylphenyl)methanesulfonamide (12c)¹⁷: was obtained without purification as a white solid : 0.94 mmol, 187 mg, 94% yield; Mp: 131-133 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.29 (d, *J* = 8.0 Hz, 1H), 7.04 (s, 1H), 7.01 (d, *J* = 8.2 Hz, 1H), 6.67 (s, 1H), 2.98 (s, 3H), 2.31 (s, 3H), 2.30 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 136.5, 132.0, 131.9, 127.8, 124.5, 39.7, 20.9, 18.1.

N-(2,4-dimethylphenyl)-4-methylbenzenesulfonamide (12d)¹⁸: was obtained by silica gel chromatography with a mixture of EtOAc/Hexane (1:1) as a white solid: 0.88 mmol, 242 mg, 88% yield; Mp: 93-94 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.62 (d, *J* = 8.3 Hz, 2H), 7.21 (d, *J* = 8.0 Hz, 2H), 7.15 (d, *J* = 8.0 Hz, 1H), 6.95 – 6.87 (m, 2H), 6.75 (s, 1H), 2.38 (s, 3H), 2.25 (s, 3H), 1.98 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 143.7, 136.9, 136.3, 132.3, 131.8, 131.5, 129.6, 127.5, 127.3, 125.3, 21.6, 20.9, 17.6.

N-(2,4-dimethylphenyl)-4-nitrobenzenesulfonamide (12f): was obtained by recrystallization in ethanol as a white solid : 0.95 mmol, 291 mg, 95% yield; Mp: 133-134 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.27 (d, *J* = 8.8 Hz, 2H), 7.90 (d, *J* = 8.8 Hz, 2H), 7.08 (d, *J* = 8.7 Hz, 1H), 7.00 – 6.89 (m, 2H), 6.77 (s, 1H), 2.27 (s, 3H), 1.97 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 150.2, 145.5, 137.6, 133.0, 132.0, 130.6, 128.6, 127.9, 125.9, 124.3, 21.0, 17.7. LRMS (ESI): Calc. for C₁₄H₁₅N₂O₄S (M+H)⁺: 307; found: 307.

(2-methyl-5-(methylsulfonamido)phenyl)triphenylphosphonium chloride (14a): was obtained by silica gel chromatography with a mixture of CH₂Cl₂/MeOH (9:1) as a brown solid: 0.036 mmol, 16.1 mg, 36% yield; Mp: 67-69 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.79 (t, *J* = 6.9 Hz, 3H), 7.71 – 7.51 (m, 13H), 7.20 (t, *J* = 7.9 Hz, 1H), 6.94 (dd, *J* = 18.0, 1.9 Hz, 1H), 2.83 (s, 3H), 1.78 (s, 3H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 150.5, 150.3, 135.0, 135.0, 134.4, 134.3, 134.2, 134.1, 130.6, 130.4, 129.1, 129.0, 128.3, 128.2, 126.8, 126.8, 120.0, 118.9, 114.0, 112.8, 39.7, 22.1, 22.0; ³¹P NMR (122 MHz, CDCl₃) δ 22.6 (br); HRMS (ESI): Calc. for C₂₆H₂₅NO₂PS⁺ (M)⁺: 446.1338; found: 446.1324.

(2,4-dimethyl-5-(2-nitrophenylsulfonamido)phenyl)triphenylphosphonium chloride (14b): was obtained by silica gel chromatography with a mixture of CH₂Cl₂/MeOH (97:03) as an yellow solid: 0.056 mmol, 31.8 mg, 56% yield; Mp: 247-249 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.86 – 7.70 (m, 4H), 7.68 – 7.57 (m, 6H), 7.55 (s, 2H), 7.54 – 7.48 (m, 4H), 7.31 – 7.25 (m, 2H), 7.24 – 7.16 (m, 2H), 7.11 (d, *J* = 7.3 Hz, 1H), 3.28 (s, 1H), 2.29 (s, 3H), 1.75 (s, 3H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 148.1, 147.2, 147.0, 141.6, 139.9, 135.1, 134.9, 134.8, 134.8, 134.2, 134.1, 131.0, 130.8, 130.5, 130.4, 130.2, 129.9, 128.6, 128.4, 122.7, 120.3, 119.1, 110.2, 109.0, 22.1, 19.4; ³¹P NMR (122 MHz, CDCl₃) δ 21.8 (br); HRMS (ESI): Calc. for C₃₂H₂₈N₂O₄PS⁺ (M)⁺: 567.1502; found: 567.1496.

Tributyl(2,4-dimethyl-5-(methylsulfonamido)phenyl)phosphonium chloride (14c): was obtained by silica gel chromatography with a mixture of CH₂Cl₂/MeOH (97:03) as an orange solid: 0.085 mmol, 34 mg, 85% yield; Mp: 162-163 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.76 (d, *J* = 15.2 Hz, 1H), 7.04 (d, *J* = 6.2 Hz, 1H), 4.13 (br, 1H), 2.82 (s, 3H), 2.51 – 2.37 (m, 6H), 2.35 (s, 3H), 2.28 (s, 3H), 1.56 – 1.30 (m, 12H), 0.89 (br, 9H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 147.2, 147.0, 139.9, 139.8, 135.2, 135.0, 129.2, 129.1, 124.5, 124.4, 109.9, 108.9, 40.0, 24.2, 24.1, 24.1, 24.0, 23.8, 21.1, 21.1, 21.0, 20.9, 20.5, 20.5, 19.3, 19.3, 13.5; ³¹P NMR (122 MHz, CDCl₃) δ 30.0 (br); HRMS (ESI): Calc. for C₂₁H₃₉NO₂PS⁺ (M)⁺: 400.2434; found: 400.2415.

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3 **Tributyl(2,4-dimethyl-5-(4-methylphenylsulfonamido)phenyl)phosphonium chloride (14d):** was
4 obtained by silica gel chromatography with a mixture of CH₂Cl₂/MeOH (9:1) as a yellow solid: 0.078
5 mmol, 37.2 mg, 78% yield; Mp: 163-164 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.73 (d, *J* = 8.1 Hz, 2H),
6 7.21 (d, *J* = 15.3 Hz, 1H), 7.06 (d, *J* = 8.0 Hz, 2H), 6.99 (d, *J* = 6.5 Hz, 1H), 2.33 (s, 3H), 2.31 – 2.22
7 (m, 12H), 1.44 – 1.30 (m, 6H), 1.29 – 1.16 (m, 6H), 0.87 (t, *J* = 7.2 Hz, 9H); ¹³C {¹H} NMR (75 MHz,
8 CDCl₃) δ 147.0, 146.9, 143.2, 139.6, 139.6, 134.8, 134.6, 128.7, 128.6, 127.2, 124.5, 124.3, 109.6,
9 108.6, 23.9, 23.8, 23.7, 23.7, 21.4, 20.7, 20.7, 20.6, 19.9, 19.3, 13.5; ³¹P NMR (122 MHz, CDCl₃) δ 29.5
10 (br); HRMS (ESI): Calc. for C₂₇H₄₃NO₂PS⁺ (M)⁺: 476.2747; found: 476.2726.
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20 **Tributyl(2,4-dimethyl-5-(2-nitrophenylsulfonamido)phenyl)phosphonium chloride (14e):** was
21 obtained as an yellow solid: 0.095 mmol, 48.2 mg, 95% yield; Mp: 159-161 °C; ¹H NMR (300 MHz,
22 CDCl₃) δ 8.00 (d, *J* = 6.7 Hz, 1H), 7.43 – 7.27 (m, 4H), 6.99 (d, *J* = 6.2 Hz, 1H), 2.32 (m, 12H), 1.51 –
23 1.14 (m, 12H), 0.84 (t, *J* = 7.0 Hz, 9H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 149.3, 147.2, 147.1, 139.5,
24 139.4, 139.3, 134.9, 134.7, 130.6, 130.6, 130.1, 129.1, 129.0, 123.8, 123.7, 122.5, 109.8, 108.8, 23.8,
25 23.6, 20.8, 20.8, 20.5, 19.9, 19.2, 13.5; HRMS (ESI): Calc. for C₂₆H₄₀N₂O₄PS⁺ (M)⁺: 507.2441; found:
26 507.2418.
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36 **Tributyl(2,4-dimethyl-5-(4-nitrophenylsulfonamido)phenyl)phosphonium chloride (14f):** was
37 obtained by silica gel chromatography with a mixture of CH₂Cl₂/MeOH (97:03) as a dark orange solid:
38 0.095 mmol, 48.2 mg, 95% yield; Mp: 152-154 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.13 (d, *J* = 8.7 Hz,
39 2H), 8.02 (d, *J* = 8.8 Hz, 2H), 7.54 (d, *J* = 15.4 Hz, 1H), 7.04 (d, *J* = 6.6 Hz, 1H), 2.27 (br, 12H), 1.51 –
40 1.28 (m, 12H), 0.91 (t, *J* = 7.0 Hz, 9H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 153.4, 148.2, 146.7, 146.5,
41 140.3, 135.3, 135.1, 129.8, 129.7, 127.7, 125.0, 124.9, 123.4, 109.6, 108.6, 23.9, 23.9, 23.7, 20.9, 20.8,
42 20.2, 19.1, 13.4; ³¹P NMR (122 MHz, CDCl₃) δ 29.6 (br); HRMS (ESI): Calc. for C₂₆H₄₀N₂O₄PS⁺ (M)⁺:
43 507.2441; found: 507.2445.
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54 **2-nitro-N-(p-tolyl)benzenesulfonamide (15a)**¹⁹: was obtained by silica gel chromatography with a
55 mixture of EtOAc/Hexane (1:9) as an orange solid: 0.90 mmol, 263 mg, 90% yield; ¹H NMR (300 MHz,
56 CDCl₃) δ 7.88 – 7.75 (m, 2H), 7.68 (td, *J* = 7.8, 1.4 Hz, 1H), 7.56 (td, *J* = 7.7, 1.2 Hz, 1H), 7.17 (s, 1H),
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1 7.06 (s, 4H), 2.28 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 148.3, 136.8, 134.0, 132.8, 132.6, 132.3, 131.9,
2 130.1, 125.3, 123.8, 21.0.
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7 **N-(4-methoxyphenyl)-2-nitrobenzenesulfonamide (15d)**²⁰: was obtained by silica gel chromatography
8 with a mixture of CH_2Cl_2 /Hexane (5:5) as a yellow solid: 0.85 mmol, 262 mg, 85% yield; ^1H NMR
9 (300 MHz, CDCl_3) δ 7.84 (dd, $J = 7.9, 1.1$ Hz, 1H), 7.75 – 7.64 (m, 2H), 7.55 (td, $J = 7.7, 1.2$ Hz, 1H),
10 7.13 (s, 1H), 7.10 – 7.04 (m, 2H), 6.80 – 6.72 (m, 2H), 3.74 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ
11 158.7, 148.3, 134.0, 132.6, 132.2, 131.9, 127.9, 126.3, 125.3, 114.6, 55.5.
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19 **N-(4-ethylphenyl)-2-nitrobenzenesulfonamide (15e)**²¹: was obtained without purification as a brown
20 solid : 0.95 mmol, 291 mg, 95% yield; ^1H NMR (300 MHz, CDCl_3) δ 7.87 – 7.77 (m, 2H), 7.68 (t, $J =$
21 7.4 Hz, 1H), 7.56 (t, $J = 7.5$ Hz, 1H), 7.08 (br, 4H), 2.58 (q, $J = 7.6$ Hz, 2H), 1.17 (t, $J = 7.6$ Hz, 3H);
22 ^{13}C NMR (75 MHz, CDCl_3) δ 148.3, 143.1, 134.0, 133.0, 132.6, 132.4, 131.9, 128.9, 125.3, 123.8, 28.3,
23 15.4.
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31 **N-(4-butylphenyl)-2-nitrobenzenesulfonamide (15f)**: was obtained without purification as a dark red
32 solid : 0.95 mmol, 317 mg, 95% yield; Mp: 100-102 °C; ^1H NMR (300 MHz, CDCl_3) δ 7.91 – 7.77 (m,
33 2H), 7.70 (t, $J = 7.4$ Hz, 1H), 7.57 (t, $J = 7.4$ Hz, 1H), 7.26 (br, 1H), 7.09 (br, 4H), 2.55 (t, $J = 7.5$ Hz,
34 2H), 1.53 (dd, $J = 14.5, 7.3$ Hz, 2H), 1.31 (dd, $J = 14.3, 7.1$ Hz, 2H), 0.91 (t, $J = 7.1$ Hz, 3H); ^{13}C NMR
35 (75 MHz, CDCl_3) δ 148.3, 141.8, 134.0, 133.0, 132.6, 132.4, 131.9, 129.5, 125.3, 123.7, 35.1, 33.5,
36 22.3, 14.0; LRMS (ESI): Calc. for $\text{C}_{16}\text{H}_{19}\text{N}_2\text{O}_4\text{S}$ ($\text{M}+\text{H}$)⁺: 319; found: 319.
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45 **N-(3-bromo-4-methylphenyl)methanesulfonamide (15g)**^{6d}: was obtained without purification as a
46 brown solid : 0.95 mmol, 250 mg, 95% yield; ^1H NMR (300 MHz, CDCl_3) δ 7.45 (d, $J = 2.2$ Hz, 1H),
47 7.23 – 7.16 (m, 2H), 7.12 (dd, $J = 8.2, 2.2$ Hz, 1H), 3.02 (s, 3H), 2.35 (s, 3H); ^{13}C NMR (75 MHz,
48 CDCl_3) δ 135.6, 135.3, 131.6, 125.4, 124.9, 120.2, 39.4, 22.4.
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56 **Methyl 2-(4-(methylsulfonamido)phenyl)acetate (15h)**^{2f}: was obtained without purification as a
57 brown solid: 0.90 mmol, 218 mg, 90%; Mp: 114-116 °C; ^1H NMR (300 MHz, CDCl_3) δ 7.25 (d, $J = 8.7$
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1 Hz, 2H), 7.18 (d, $J = 8.6$ Hz, 2H), 7.13 (br, 1H), 3.70 (s, 3H), 3.60 (s, 2H), 2.98 (s, 3H); ^{13}C NMR (75
2 MHz, CDCl_3) δ 172.2, 136.0, 131.2, 130.7, 121.1, 52.3, 40.5, 39.3.
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7 **N-(4-(2-((tert-butyldimethylsilyloxy)ethyl)phenyl)methanesulfonamide (15i)^{6c}**: was obtained by
8 silica gel chromatography with a mixture of EtOAc/Hexane (2:8) as an yellow oil: 0.95 mmol, 313 mg,
9 95% yield; ^1H NMR (300 MHz, CDCl_3) δ 7.23 – 7.11 (m, 4H), 6.83 (s, 1H), 3.78 (t, $J = 6.8$ Hz, 2H),
10 2.97 (s, 3H), 2.79 (t, $J = 6.8$ Hz, 2H), 0.85 (s, 9H), -0.03 (s, 6H); ^{13}C NMR (75 MHz, CDCl_3) δ 137.1,
11 134.8, 130.5, 121.4, 64.4, 53.6, 39.0, 39.2, 29.8, 26.0, 18.4, -5.3.
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20 **N-(4-methoxynaphthalen-1-yl)-2-nitrobenzenesulfonamide (15j)²²**: was obtained without purification
21 as a brown solid: 0.95 mmol, 311 mg, 95% yield; ^1H NMR (300 MHz, Acetone- d_6) δ 9.17 (br, 1H), 8.21
22 (d, $J = 7.7$ Hz, 1H), 8.08 – 7.72 (m, 7.3 Hz, 4H), 7.75 – 7.54 (m, 2H), 7.57 – 7.27 (m, 4H); ^{13}C NMR
23 (75 MHz, Acetone- d_6) δ 149.0, 135.3, 135.2, 133.4, 133.2, 132.4, 131.7, 131.3, 129.0, 128.9, 127.3,
24 127.2, 126.3, 126.1, 125.8, 123.6.
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32 **Tributyl(2-methyl-5-(2-nitrophenylsulfonamido)phenyl)phosphonium chloride (17a)**: was
33 obtained by silica gel chromatography with a mixture of $\text{CH}_2\text{Cl}_2/\text{MeOH}$ (97:03) as a brown solid: 0.084
34 mmol, 41.5 mg, 84% yield; Mp: 37-40 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.11 (dd, $J = 8.0, 1.4$ Hz, 1H),
35 7.48 – 7.34 (m, 3H), 7.34 – 7.34 (m, 1H), 7.28 (s, 1H), 7.04 (t, $J = 7.3$ Hz, 1H), 4.13 (s, 1H), 2.49 – 2.36
36 (m, 6H), 2.34 (s, 3H), 1.49 – 1.22 (m, 12H), 0.85 (t, $J = 7.0$ Hz, 9H); ^{13}C $\{^1\text{H}\}$ NMR (75 MHz, CDCl_3)
37 δ 149.2, 147.4, 147.3, 138.2, 133.8, 133.6, 130.9, 130.7, 130.6, 130.2, 130.1, 127.8, 125.9, 125.8, 122.8,
38 114.1, 113.1, 23.8, 23.6, 20.9, 20.4, 19.7, 13.5; ^{31}P NMR (122 MHz, CDCl_3) δ 30.4 (br); HRMS (ESI):
39 Calc. for $\text{C}_{25}\text{H}_{38}\text{N}_2\text{O}_4\text{PS}^+$ (M)⁺: 493.2284; found: 493.2312.
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49 **Tributyl(2-methoxy-5-(2-nitrophenylsulfonamido)phenyl)phosphonium chloride (17d)**: was
50 obtained by silica gel chromatography with a mixture of $\text{CH}_2\text{Cl}_2/\text{MeOH}$ (97:03) as an orange solid:
51 0.095 mmol, 48.4 mg, 95% yield; Mp: 45-47 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.16 (dd, $J = 7.5, 1.2$
52 Hz, 1H), 7.54 – 7.37 (m, 3H), 7.35 (d, $J = 1.2$ Hz, 1H), 7.32 (dd, $J = 5.9, 1.9$ Hz, 1H), 6.82 (dd, $J = 9.0,$
53 6.4 Hz, 1H), 3.81 (s, 3H), 2.45 – 2.09 (m, 6H), 1.47 – 1.18 (m, 12H), 0.83 (t, $J = 6.9$ Hz, 9H); ^{13}C $\{^1\text{H}\}$
54 NMR (75 MHz, CDCl_3) δ 155.5, 149.0, 140.7, 140.5, 137.1, 131.2, 131.1, 130.9, 130.7, 126.7, 126.6,
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122.8, 112.5, 112.4, 103.7, 102.6, 56.2, 23.8, 23.7, 23.6, 19.6, 18.9, 13.4; ^{31}P NMR (122 MHz, CDCl_3) δ 29.7 (br); HRMS (ESI): Calc. for $\text{C}_{25}\text{H}_{38}\text{N}_2\text{O}_5\text{PS}^+$ (M) $^+$: 509.2234; found: 509.2235.

Tributyl(2-ethyl-5-(2-nitrophenylsulfonamido)phenyl)phosphonium chloride (17e): was obtained by silica gel chromatography with a mixture of $\text{CH}_2\text{Cl}_2/\text{MeOH}$ (97:03) as a brown solid: 0.079 mmol, 40.1 mg, 79% yield; Mp: 130-131 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.08 (dd, $J = 6.2, 2.1$ Hz, 1H), 7.46 – 7.34 (m, 3H), 7.31 (d, $J = 8.5$ Hz, 1H), 7.18 (dd, $J = 15.5, 2.3$ Hz, 1H), 7.14 – 7.08 (m, 1H), 3.47 (s, 1H), 2.57 (q, $J = 7.3$ Hz, 2H), 2.45 – 2.30 (m, $J = 15.9, 11.8$ Hz, 6H), 1.47 – 1.23 (m, 12H), 1.18 (t, $J = 7.4$ Hz, 3H), 0.84 (t, $J = 7.0$ Hz, 9H); ^{13}C $\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 149.2, 148.1, 147.9, 138.67, 136.3, 136.2, 131.5, 131.3, 130.9, 130.6, 128.2, 128.1, 125.3, 125.1, 122.8, 113.2, 112.2, 26.4, 26.3, 23.9, 23.9, 23.8, 23.6, 21.0, 20.3, 15.7, 13.5; ^{31}P NMR (122 MHz, CDCl_3) δ 29.9 (br); HRMS (ESI): Calc. for $\text{C}_{26}\text{H}_{40}\text{N}_2\text{O}_4\text{PS}^+$ (M) $^+$: 507.2441; found: 507.2447.

Tributyl(2-butyl-5-(2-nitrophenylsulfonamido)phenyl)phosphonium chloride (17f): was obtained by silica gel chromatography with a mixture of $\text{CH}_2\text{Cl}_2/\text{MeOH}$ (97:03) as a brown oil: 0.075 mmol, 40.2 mg, 75% yield; ^1H NMR (300 MHz, CDCl_3) δ 8.10 (dd, $J = 6.2, 1.7$ Hz, 1H), 7.47 – 7.34 (m, 3H), 7.31 (d, $J = 8.5$ Hz, 1H), 7.20 (dd, $J = 15.2, 1.7$ Hz, 1H), 7.11 (dd, $J = 8.3, 6.6$ Hz, 1H), 3.34 (s, 1H), 2.50 (t, $J = 7.3$ Hz, 2H), 2.43 – 2.29 (m, 6H), 1.54 – 1.24 (m, 16H), 0.97 – 0.74 (m, 12H); ^{13}C $\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 149.2, 148.0, 147.8, 138.6, 135.1, 135.0, 132.0, 131.8, 130.9, 130.6, 130.5, 128.1, 125.4, 125.3, 122.8, 113.3, 112.3, 34.0, 33.3, 24.0, 23.9, 23.9, 23.6, 22.8, 21.1, 20.5, 14.1, 13.5; ^{31}P NMR (122 MHz, CDCl_3) δ 30.2 (br); HRMS (ESI): Calc. for $\text{C}_{28}\text{H}_{44}\text{N}_2\text{O}_4\text{PS}^+$ (M) $^+$: 535.2754; found: 535.2730.

(3-bromo-2-methyl-5-(methylsulfonamido)phenyl)tributylphosphonium chloride (17g): was obtained by silica gel chromatography with a mixture of $\text{CH}_2\text{Cl}_2/\text{MeOH}$ (98:02 to 95:05) as a yellow solid: 0.090 mmol, 41.9 mg, 90% yield; Mp: 50-52 °C; ^1H NMR (300 MHz, CDCl_3) δ 8.06 (s, 1H), 7.83 (d, $J = 14.2$ Hz, 1H), 5.77 (br, 1H), 3.00 (s, 3H), 2.69 – 2.55 (m, 6H), 2.52 (s, 3H), 1.56 – 1.35 (m, 12H), 0.92 (t, $J = 6.9$ Hz, 9H); ^{13}C $\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 142.9, 142.7, 132.4, 132.3, 129.4, 129.2, 128.9, 125.6, 125.5, 117.9, 116.9, 39.8, 24.2, 24.1, 23.9, 23.7, 21.7, 21.7, 20.9, 20.3, 13.5; ^{31}P NMR (122 MHz, CDCl_3) δ 33.1 (br); HRMS (ESI): Calc. for $\text{C}_{20}\text{H}_{36}\text{BrNO}_2\text{PS}^+$ (M) $^+$: 464.1382; found: 464.1376.

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3 **Tributyl(2-(2-methoxy-2-oxoethyl)-5-(methylsulfonamido)phenyl)phosphonium chloride (17h):** by
4 silica gel chromatography with a mixture of CH₂Cl₂/MeOH (98:02 to 95:05) as a brown foam: 0.065
5 mmol, 28.9 mg, 65% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.91 (d, *J* = 14.9 Hz, 1H), 7.79 (d, *J* = 8.5
6 Hz, 1H), 7.40 (dd, *J* = 8.6, 5.9 Hz, 1H), 6.23 (br, 1H), 3.74 (s, 2H), 3.70 (s, 3H), 3.01 (s, 3H), 2.65 –
7 2.53 (m, 6H), 1.54 – 1.36 (m, 12H), 0.91 (t, *J* = 6.9 Hz, 9H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 170.6,
8 143.3, 143.1, 134.6, 134.5, 129.9, 129.8, 126.1, 126.0, 125.4, 116.8, 115.8, 52.8, 39.8, 39.3, 29.8, 24.1,
9 24.0, 23.9, 23.7, 21.1, 20.5, 13.5; ³¹P NMR (122 MHz, CDCl₃) δ 31.9 (br); HRMS (ESI): Calc. for
10 C₂₂H₃₉NO₄PS⁺ (M)⁺: 444.2332; found: 444.2318.
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21 **Tributyl(2-(2-((tert-butyldimethylsilyl)oxy)ethyl)-5-(methylsulfonamido)-phenyl)-phosphonium**
22 **chloride (17i):** was obtained was obtained by silica gel chromatography with a mixture of
23 CH₂Cl₂/MeOH (98:02 to 95:05) as an yellow foam: 0.068 mmol, 36.1 mg, 68% yield; ¹H NMR (300
24 MHz, CDCl₃) δ 10.72 (s, 1H), 8.49 (d, *J* = 15.3 Hz, 1H), 7.93 (d, *J* = 8.6 Hz, 1H), 7.43 (dd, *J* = 8.6, 5.8
25 Hz, 1H), 3.89 (t, *J* = 6.2 Hz, 2H), 3.10 (s, 3H), 2.89 (t, *J* = 6.1 Hz, 2H), 2.64 – 2.51 (m, 6H), 1.63 – 1.46
26 (m, 12H), 0.97 (t, *J* = 6.8 Hz, 9H), 0.84 (s, 9H), 0.01 (s, 6H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 139.4,
27 139.2, 138.0, 137.9, 132.9, 132.8, 125.9, 125.7, 125.3, 117.1, 116.0, 63.1, 39.7, 36.7, 34.8, 31.7, 29.8,
28 29.2, 26.0, 25.8, 25.4, 24.2, 24.0, 23.8, 23.6, 22.7, 21.5, 20.8, 18.4, 14.2, 13.6, 11.5, -3.46, -5.27; ³¹P
29 NMR (122 MHz, CDCl₃) δ 31.5 (br); HRMS (ESI): Calc. for C₂₇H₅₃NO₃PSSi⁺ (M)⁺: 530.3248; found:
30 530.3228.
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41 **Tributyl(1-methoxy-4-(2-nitrophenylsulfonamido)naphthalen-2-yl)phosphonium chloride (17j):**
42 was obtained by recrystallization in diethyl ether as an yellow solid: 0.082 mmol, 45.9 mg, 82% yield;
43 Mp: 51-52 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.89 (dd, *J* = 6.9, 2.7 Hz, 1H), 8.14 (dd, *J* = 6.1, 1.9 Hz,
44 1H), 7.86 (dd, *J* = 6.7, 2.6 Hz, 1H), 7.58 – 7.48 (m, 2H), 7.38 – 7.28 (m, 3H), 7.15 (d, *J* = 13.7 Hz, 1H),
45 3.98 (s, 3H), 2.46 – 2.25 (m, 6H), 1.52 – 1.24 (m, 12H), 0.88 (t, *J* = 7.0 Hz, 9H); ¹³C {¹H} NMR (75
46 MHz, CDCl₃) δ 152.0, 149.5, 145.3, 145.1, 139.1, 134.7, 130.6, 130.5, 130.1, 127.4, 127.3, 126.9,
47 126.9, 122.5, 121.8, 110.3, 110.2, 103.9, 102.8, 63.8, 23.9, 23.8, 23.8, 23.7, 20.3, 19.6, 13.5; ³¹P NMR
48 (122 MHz, CDCl₃) δ 29.7 (br); HRMS (ESI): Calc. for C₂₉H₄₀N₂O₅PS⁺ (M)⁺: 559.2390; found:
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Dimethyl (2,4-dimethyl-5-(2-nitrophenylsulfonamido)phenyl)phosphonate (20a): was obtained by silica gel chromatography with a mixture of ethyl acetate/hexane (4:6 to 78:22) as a yellow solid: 0.095 mmol, 39.4 mg, 95% yield; Mp: 137-139 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.88 (d, *J* = 7.7 Hz, 1H), 7.82 – 7.69 (m, 2H), 7.61 (t, *J* = 7.6 Hz, 1H), 7.52 (d, *J* = 15.0 Hz, 1H), 7.09 (d, *J* = 6.0 Hz, 1H), 3.70 (s, 3H), 3.66 (s, 3H), 2.46 (s, 3H), 2.24 (s, 3H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 148.1, 141.5, 141.3, 139.9, 139.8, 134.6, 134.4, 134.1, 133.2, 132.7, 131.7, 131.6, 131.6, 131.5, 131.4, 125.6, 125.5, 123.2, 52.7, 52.7, 20.7, 20.7, 18.1; ³¹P NMR (122 MHz, CDCl₃) δ 20.9 (br); HRMS (ESI): Calc. for C₁₆H₂₃N₃O₇PS (M+NH₄)⁺: 432.0989; found: 432.0988.

Diethyl (2,4-dimethyl-5-(2-nitrophenylsulfonamido)-phenyl)phosphonate (20b): was obtained by silica gel chromatography with a mixture of ethyl acetate/hexane (4:6 to 6:4) as a white solid: 0.070 mmol, 31.0 mg, 70% yield; Mp: 155-157 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.89 (dd, *J* = 7.9, 1.0 Hz, 1H), 7.80 (dd, *J* = 7.8, 1.3 Hz, 1H), 7.73 (td, *J* = 7.8, 1.4 Hz, 1H), 7.61 (td, *J* = 7.7, 1.1 Hz, 1H), 7.53 (d, *J* = 14.9 Hz, 1H), 7.15 (s, 1H), 7.09 (d, *J* = 6.0 Hz, 1H), 4.12 – 3.91 (m, 4H), 2.49 (s, 3H), 2.27 (s, 3H), 1.28 (t, *J* = 7.1 Hz, 6H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 148.2, 141.4, 141.2, 139.6, 134.6, 134.4, 134.1, 133.3, 132.8, 131.6, 131.6, 131.4, 131.3, 131.3, 127.1, 125.6, 124.6, 124.6, 62.2, 62.2, 20.8, 20.8, 18.1, 16.5, 16.4; ³¹P NMR (122 MHz, CDCl₃) δ 17.9 (br); HRMS (ESI): Calc. for C₁₈H₂₄N₂O₇PS⁺ (M+H)⁺: 443,1036; found: 443.1046.

Dimethyl (2-methoxy-5-(2-nitrophenylsulfonamido)phenyl)-phosphonate (20c): was obtained by silica gel chromatography with a mixture of ethyl acetate/hexane (75:25 to 95:05) as a white solid: 0.095 mmol, 39.6 mg, 95% yield; Mp: 147-148 °C; ¹H NMR (300 MHz, CDCl₃) δ 7.80 (td, *J* = 9.1, 0.9 Hz, 3H), 7.68 (td, *J* = 7.8, 1.2 Hz, 1H), 7.60 – 7.52 (m, 1H), 7.50 – 7.41 (m, 1H), 6.90 – 6.84 (m, 1H), 3.85 (s, 3H), 3.75 (s, 3H), 3.71 (s, 3H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 160.0, 148.4, 134.1, 132.6, 132.4, 131.7, 130.7, 130.6, 128.5, 128.3, 125.4, 117.9, 115.4, 112.3, 112.2, 56.4, 53.2, 53.2; ³¹P NMR (122 MHz, CDCl₃); HRMS (ESI): Calc. for C₁₅H₁₈N₂O₈PS (M+H)⁺: 417.0516; found: 417.0511.

Dimethyl (1-methoxy-4-(2-nitrophenylsulfonamido)naphthalen-2-yl)phosphonate (20d): was obtained by silica gel chromatography with a mixture of ethyl acetate/hexane (4:6 to 6:4) as a white solid: 0.069 mmol, 32.2 mg, 69% yield; Mp: 51-53 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.24 – 8.09 (m, 2H), 7.91 (d, *J* = 7.9 Hz, 1H), 7.77 – 7.63 (m, 3H), 7.62 – 7.44 (m, 4H), 4.09 (s, 3H), 3.80 (s, 3H), 3.76

(s, 3H); ^{13}C $\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 160.7, 148.3, 134.6, 134.1, 133.0, 132.8, 131.6, 129.6, 128.9, 128.7, 127.4, 127.3, 125.5, 123.7, 123.6, 118.0, 115.5, 64.2, 53.2, 53.1; ^{31}P NMR (122 MHz, CDCl_3) δ 29.7 (br); HRMS (ESI): Calc. for $\text{C}_{19}\text{H}_{19}\text{N}_2\text{NaO}_8\text{PS}$ ($\text{M}+\text{Na}$) $^+$: 489.0492; found: 489.0499.

(E)-N-(4-methoxy-2,4-dimethylcyclohexa-2,5-dien-1-ylidene)methanesulfonamide (21a) 8 : was obtained by silica gel chromatography with a mixture of EtOAc/Hexane (15:85) as a white foam: 0.86 mmol, 197 mg, 86% yield; ^1H NMR (300 MHz, CDCl_3) δ 7.34 (dd, J = 10.2, 2.7 Hz, 1H), 6.66 (dd, J = 10.2, 2.4 Hz, 1H), 6.48 (s, 1H), 3.16 (d, J = 2.6 Hz, 3H), 3.09 (d, J = 2.7 Hz, 3H), 1.92 (s, 3H), 1.32 (d, J = 2.4 Hz, 3H).

(E)-N-(4-methoxy-2,4-dimethylcyclohexa-2,5-dien-1-ylidene)-2-nitrobenzenesulfonamide (21b): was obtained by silica gel chromatography with a mixture of EtOAc/Hexane (15:85) as a white foam: 0.69 mmol, 232 mg, 69% yield, ^1H NMR (300 MHz, CDCl_3) δ 8.35 – 8.22 (m, 1H), 7.85 (d, J = 8.8 Hz, 1H), 7.80 – 7.67 (m, 2H), 7.49 (d, J = 10.2 Hz, 1H), 6.80 (dd, J = 10.2, 2.4 Hz, 1H), 6.57 (s, 1H), 3.16 (s, 3H), 1.89 (s, 3H), 1.39 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 165.9, 153.6, 149.1, 136.4, 135.0, 133.8, 132.6, 129.8, 124.9, 124.6, 73.2, 53.7, 26.0, 17.6; LRMS Calc. for $\text{C}_{15}\text{H}_{17}\text{N}_2\text{O}_5\text{S}$ ($\text{M}+\text{H}$) $^+$: 337; found: 337.

N-(4-ethyl-4-methoxycyclohexa-2,5-dien-1-ylidene)methanesulfonamide (21c) 8 : was obtained by silica gel chromatography with a mixture of EtOAc/Hexane (15:85) as a yellow oil: 0.85 mmol, 194 mg, 85% yield; ^1H NMR (300 MHz, CDCl_3) δ 7.44 (dd, J = 10.6, 1.8 Hz, 1H), 6.74 – 6.57 (m, 2H), 6.43 (dd, J = 10.3, 1.8 Hz, 1H), 3.16 (s, 6H), 1.73 (q, J = 7.6 Hz, 2H), 0.80 (t, J = 7.6 Hz, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 164.4, 152.0, 150.5, 131.4, 125.3, 76.5, 53.5, 43.1, 32.5, 7.9.

N-(4-methoxy-4-methylcyclohexa-2,5-dien-1-ylidene)ethanesulfonamide (21d): was obtained by silica gel chromatography with a mixture of EtOAc/Hexane (1:9) as a white solid: 0.86 mmol, 197 mg, 86% yield; Mp: 79-80 $^\circ\text{C}$; ^1H NMR (300 MHz, CDCl_3) δ 7.40 (dd, J = 10.5, 2.0 Hz, 1H), 6.74 – 6.61 (m, 2H), 6.36 (dd, J = 10.2, 2.0 Hz, 1H), 3.20 (q, J = 7.4 Hz, 2H), 3.13 (s, 3H), 1.44 (t, J = 7.4 Hz, 3H), 1.37 (s, 3H); ^{13}C NMR (75 MHz, CDCl_3) δ 164.4, 152.2, 150.8, 130.3, 124.2, 72.7, 53.6, 49.3, 26.0, 8.1; LRMS Calc. for $\text{C}_{10}\text{H}_{16}\text{NO}_3\text{S}$ ($\text{M}+\text{H}$) $^+$: 230; found: 230.

N-(4-methoxy-4-methylcyclohexa-2,5-dien-1-ylidene)-2-nitrobenzenesulfonamide (21e): was obtained by silica gel chromatography with a mixture of EtOAc/Hexane (15:85) as a white solid: 0.71 mmol, 228 mg, 71% yield; Mp: 151-152 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.33 – 8.27 (m, 1H), 7.89 – 7.82 (m, 1H), 7.79 – 7.73 (m, 2H), 7.55 (d, *J* = 9.9 Hz, 1H), 6.86 – 6.75 (m, 2H), 6.41 (d, *J* = 9.3 Hz, 1H), 3.20 (s, 3H), 1.43 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 164.6, 154.0, 152.5, 148.0, 134.5, 134.0, 132.7, 130.1, 124.9, 124.2, 73.0, 53.8, 26.0; LRMS Calc. for C₁₄H₁₅N₂O₅S (M+H)⁺: 323; found: 323.

N-(4-methoxy-4-methylcyclohexa-2,5-dien-1-ylidene)methanesulfonamide (21f)⁸: was obtained by silica gel chromatography with a mixture of EtOAc/Hexane (15:85) as a colorless oil: 0.95 mmol, 204 mg, 95% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.40 (d, *J* = 9.5 Hz, 1H), 6.76 – 6.64 (m, 2H), 6.44 – 6.31 (m, 1H), 3.17 (s, 3H), 3.16 (s, 3H), 1.39 (s, 3H).

N-(4-methoxy-4-methylcyclohexa-2,5-dien-1-ylidene)-4-methylbenzenesulfonamide (21g): was obtained by silica gel chromatography with a mixture of EtOAc/Hexane (1:9) as a white solid: 0.89 mmol, 259 mg, 89 % yield; Mp: 123-124 °C, ¹H NMR (300 MHz, CDCl₃) δ 7.86 (d, *J* = 8.1 Hz, 2H), 7.60 (d, *J* = 10.2 Hz, 1H), 7.30 (d, *J* = 8.0 Hz, 2H), 6.74 (dd, *J* = 10.3, 2.2 Hz, 1H), 6.65 (dd, *J* = 10.0, 2.2 Hz, 1H), 6.35 (d, *J* = 10.0 Hz, 1H), 3.13 (s, 3H), 2.40 (s, 3H), 1.37 (s, 3H); ¹³C NMR (75 MHz, CDCl₃) δ 163.7, 152.6, 150.9, 143.8, 138.2, 130.5, 129.6, 127.2, 123.9, 72.7, 53.6, 26.0, 21.6; LRMS Calc. for C₁₅H₁₈NO₃S (M+H)⁺: 292; found: 292.

(E)-N-(4,4-dimethoxynaphthalen-1(4H)-ylidene)methanesulfonamide (21h)^{2g}: was obtained by silica gel chromatography with a mixture of EtOAc/Hexane (1:9) as a brown oil: 0.95 mmol, 267 mg, 95% yield; ¹H NMR (300 MHz, CDCl₃) δ 8.22 (dd, *J* = 8.0, 0.9 Hz, 1H), 7.75 (dd, *J* = 7.9, 1.1 Hz, 1H), 7.71 (d, *J* = 10.7 Hz, 1H), 7.66 (td, *J* = 7.6, 1.3 Hz, 1H), 7.52 – 7.44 (m, 1H), 6.86 (d, *J* = 10.7 Hz, 1H), 3.30 (s, 3H), 3.18 (s, 6H); ¹³C NMR (75 MHz, CDCl₃) δ 163.4, 143.9, 139.4, 133.6, 130.9, 129.2, 126.9, 126.4, 94.4, 51.3, 43.3.

Tributyl(2-ethyl-5-(methylsulfonamido)phenyl)phosphonium chloride (22c): was obtained by silica gel chromatography with a mixture of CH₂Cl₂/MeOH (95:05) as a brown oil: 0.060 mmol, 24.0 mg, 60% yield; ¹H NMR (300 MHz, CDCl₃) δ 7.61 (d, *J* = 8.6 Hz, 1H), 7.43 (dd, *J* = 15.2, 1.8 Hz, 1H), 7.24

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– 7.18 (m, 1H), 4.51 (br, 1H), 2.92 (s, 3H), 2.62 (q, $J = 7.3$ Hz, 2H), 2.54 – 2.39 (m, 6H), 1.50 – 1.37 (m, 12H), 1.24 (t, $J = 7.5$, 3H), 0.90 (t, $J = 6.8$ Hz, 9H); ^{13}C $\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 147.5, 135.9, 131.9, 131.8, 126.5, 125.5, 125.4, 113.4, 112.4, 39.8, 29.8, 26.6, 26.6, 24.2, 24.1, 24.0, 23.8, 21.4, 20.7, 16.0, 14.2, 13.6; ^{31}P NMR (122 MHz, CDCl_3) δ 29.7 (br); HRMS (ESI): Calc. for $\text{C}_{21}\text{H}_{39}\text{NO}_2\text{PS}^+$ (M) $^+$: 400,2434; found: 400.2440.

Tributyl(5-(ethylsulfonamido)-2-methylphenyl)phosphonium chloride (22d): was obtained by recrystallization in diethyl ether as a white solid: 0.072 mmol, 28.8 mg, 72% yield; Mp: 61-63 °C; ^1H NMR (300 MHz, CDCl_3) δ 7.40 (d, $J = 13.8$ Hz, 2H), 7.08 (t, $J = 7.5$ Hz, 1H), 3.04 (q, $J = 7.4$ Hz, 2H), 2.37 – 2.25 (m, 9H), 1.54 – 1.43 (m, $J = 3.6$ Hz, 12H), 1.39 (t, $J = 7.4$ Hz, 3H), 0.96 (t, $J = 6.8$ Hz, 9H); ^{13}C $\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 150.8, 150.6, 133.7, 133.5, 126.5, 126.4, 125.9, 125.1, 125.0, 113.2, 112.2, 45.8, 23.9, 23.8, 23.7, 23.5, 20.6, 19.9, 13.3, 9.2; ^{31}P NMR (122 MHz, CDCl_3) δ 29.7 (br); HRMS (ESI): Calc. for $\text{C}_{21}\text{H}_{39}\text{NO}_2\text{PS}^+$ (M) $^+$: 400,2434; found: 400.2428.

Tributyl(2-methyl-5-(methylsulfonamido)phenyl)phosphonium chloride (22f): was obtained by silica gel chromatography with a mixture of $\text{CH}_2\text{Cl}_2/\text{MeOH}$ (9:1) as a yellow solid: 0.059 mmol, 22.8 mg, 59% yield; Mp: 86-89 °C; ^1H NMR (300 MHz, CDCl_3) δ 7.47 (d, $J = 8.8$ Hz, 1H), 7.36 – 7.29 (m, 1H), 7.11 (t, $J = 7.7$ Hz, 1H), 2.96 (s, 3H), 2.39 – 2.26 (m, 9H), 1.54 – 1.41 (m, $J = 3.7$ Hz, 12H), 0.96 (t, $J = 6.7$ Hz, 9H); ^{13}C $\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 150.9, 150.7, 133.8, 133.7, 126.3, 126.2, 125.3, 125.2, 113.3, 112.3, 39.2, 23.9, 23.8, 23.7, 23.5, 20.6, 19.9, 13.3; ^{31}P NMR (122 MHz, CDCl_3) δ 29.6 (br); HRMS (ESI): Calc. for $\text{C}_{20}\text{H}_{37}\text{NO}_2\text{PS}^+$ (M) $^+$: 386,2277; found: 386.2271.

Tributyl(2-methyl-5-(4-methylphenylsulfonamido)phenyl)phosphonium chloride (22g): was obtained by recrystallization in diethyl ether as a white solid: 0.056 mmol, 25.9 mg, 56% yield; Mp: 114-117 °C; ^1H NMR (300 MHz, CDCl_3) δ 7.82 (d, $J = 8.0$ Hz, 2H), 7.30 (d, $J = 8.4$ Hz, 1H), 7.18 (d, $J = 15.7$ Hz, 1H), 7.10 (d, $J = 7.9$ Hz, 2H), 7.04 – 6.95 (m, 1H), 2.41 – 2.26 (m, 12H), 1.44 – 1.25 (m, 12H), 0.87 (t, $J = 7.0$ Hz, 9H); ^{13}C $\{^1\text{H}\}$ NMR (75 MHz, CDCl_3) δ 150.1, 149.9, 143.5, 139.6, 133.7, 133.5, 128.8, 127.1, 127.0, 125.2, 125.1, 113.1, 112.1, 23.9, 23.7, 21.4, 20.6, 20.0, 13.5; ^{31}P NMR (122 MHz, CDCl_3) δ 30.0 (br); HRMS (ESI): Calc. for $\text{C}_{26}\text{H}_{41}\text{NO}_2\text{PS}^+$ (M) $^+$: 462,2590; found: 462.2598.

Tributyl(1-methoxy-4-(methylsulfonamido)naphthalen-2-yl)phosphonium chloride (22h): was obtained by recrystallization in diethyl ether as a brown solid: 0.080 mmol, 36.2 mg, 80% yield; Mp: 157-159 °C; ¹H NMR (300 MHz, CDCl₃) δ 8.85 – 8.78 (m, 1H), 7.89 – 7.82 (m, 1H), 7.57 – 7.49 (m, 2H), 7.37 (d, *J* = 14.1 Hz, 1H), 4.01 (s, 3H), 3.01 (s, 3H), 2.45 – 2.33 (m, 6H), 1.56 – 1.47 (m, 12H), 0.96 (t, *J* = 6.7 Hz, 9H); ¹³C {¹H} NMR (75 MHz, CDCl₃) δ 152.4, 144.8, 144.7, 144.6, 134.8, 134.8, 127.5, 127.4, 127.4, 127.2, 126.9, 126.79, 126.8, 121.8, 110.7, 63.8, 63.8, 39.3, 24.1, 24.0, 23.9, 23.7, 20.7, 20.1, 13.4; ³¹P NMR (122 MHz, CDCl₃) δ 29.6 (br); HRMS (ESI): Calc. for C₂₄H₃₉NO₃PS⁺ (M)⁺: 452.2383; found: 452.2393.

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Supporting Information Available: The Supporting Information is available free of charge on the ACS Publications website ¹H and ¹³C NMR spectral data of all compounds.

References and Footnotes:

¹ Davies, H. M. L.; Morton, D. *J. Org. Chem.* **2016**, *81*, 343.

² (a) Tamura, Y.; Yakura, T.; Haruta, J.; Kita, Y. *J. Org. Chem.* **1987**, *52*, 3927. (b) Farid, U.; Malmedy, F.; Claveau, R.; Albers, L.; Wirth, T. *Angew. Chem. Int. Ed.* **2013**, *52*, 7018. (c) Zhdankin, V. V. *Hypervalent Iodine Chemistry: Preparation, Structure, and Synthetic Applications of Polyvalent Iodine Compounds*; John Wiley & Sons: Chichester, UK, **2013**. (d) Coffinier, R.; El Assal, M.; Peixoto, P. A.; Bosset, C.; Miqueu, K.; Sotiropoulos, J.-M.; Pouységu, L.; Quideau, S. *Org. Lett.* **2016**, *18*, 1120. (e) Jacquemot, G.; Maertens, G.; Canesi, S. *Chem. Eur. J.* **2015**, *21*, 7713. (f) Coulibali, S.; Godou, T.; Canesi, S. *Org. Lett.* **2016**, *18*, 4348. (g) Coulibali, S.; Deruer, E.; Godin, E.; Canesi, S. *Org. Lett.* **2017**, *19*, 1188. (h) Martínez, C.; Muñoz, K. *Angew. Chem. Int. Ed.*, **2015**, *54*, 8287. (i) Antien, K.; Viault, G.; Pouysegu, L.; Peixoto, P. A.; Quideau, S. *Tetrahedron* **2017**, *73*, 3684. (j) Uyanik, M.; Sasakura, N.; Mizuno, M.; Ishihara, K. *ACS Catal.* **2017**, *7*, 872. (k) Mizar, P.; Niebuhr, R.; Hutchings, M.; Farooq, U.; Wirth, T. *Chem. - Eur. J.* **2016**, *22*, 1614. (l) Shimogaki, M.; Fujita, M.; Sugimura, T. *Angew. Chem., Int. Ed.* **2016**, *55*, 15797.

³ (a) Dohi, T.; Ito, M.; Morimoto, K.; Iwata, M.; Kita, Y. *Angew. Chem., Int. Ed.* **2008**, *47*, 1301. (b) Kita, Y.; Morimoto, K.; Ito, M.; Ogawa, C.; Goto, A.; Dohi, T. *J. Am. Chem. Soc.* **2009**, *131*, 1668. (c) Dohi, T.; Ito, M.; Yamaoka, N.; Morimoto, K.; Fujioka, H.; Kita, Y. *Angew. Chem., Int. Ed.* **2010**, *49*, 3334. (d) Morimoto, K.; Yamaoka, N.; Ogawa, C.; Nakae, T.; Fujioka, H.; Dohi, T.; Kita, Y. *Org. Lett.* **2010**, *12*, 3804. (e) Dohi, T.; Ito, M.; Itani, I.; Yamaoka, N.; Morimoto, K.; Fujioka, H.; Kita, Y. *Org. Lett.* **2011**, *13*, 6208. (f) Morimoto, K.; Sakamoto, K.; Onishi, Y.; Miyamoto, T.; Ito, M.; Dohi, T.; Kita, Y. *Chem–Eur. J.* **2013**, *19*, 8726.

⁴ (a) Phipps, R. J.; Gaunt, M. J. *Science*, **2009**, *323*, 1593. (b) Ciana, C.-L.; Phipps, R. J.; Brandt, J. R.; Meyer, F.-M.; Gaunt, M. J. *Angew. Chem. Int. Ed.*, **2011**, *50*, 458. (c) Duong H. A.; Gilligan R. E.; Cooke M. L.; Phipps R. J.; Gaunt M. J. *Angew. Chem. Int. Ed.* **2011**, *50*, 463.

⁵ (a) Zheng, C.; Chen, J.-J.; Fan, R. *Org. Lett.* **2014**, *16*, 816. (b) Han, D.; He, Q.; Fan, R. *Angew. Chem. Int. Ed.*, **2015**, *54*, 14013. (c) Wang, L. F.; Fan, R. *Org. Lett.* **2012**, *14*, 3596. (d) Feng, X.; Wang, H.; Yang, B.; Fan, R. *Org. Lett.* **2014**, *16*, 3600.

⁶ (a) Ito, M.; Kubo, H.; Itani, I.; Morimoto, K.; Dohi, T.; Kita, Y. *J. Am. Chem. Soc.* **2013**, *135*, 1478; (b) Jean, A.; Cantat, J.; Bérard, D.; Bouchu, D.; Canesi, S. *Org. Lett.* **2007**, *9*, 2553. (c) Jacquemot, G.; Ménard, M.-A.; L'Homme, C.; Canesi, S. *Chem. Sci.* **2013**, *4*, 1287. (d) Deruer, E.; Canesi, S. *Org. Biomol. Chem.* **2017**, *15*, 3736.

⁷ (a) Zawada, P. V.; Banfield, S. C.; Kerr, M. A. *Synlett* **2003**, 971. (b) Wells, G.; Berry, J. M.; Bradshaw, T. D.; Burger, A. M.; Seaton, A.; Wang, B.; Westwell, A. D.; Stevens, M. F. G. *J. Med. Chem.* **2003**, *46*, 532.

⁸ Giroux, M. A.; Guérard, K. C.; Beaulieu, M. A.; Sabot, C.; Canesi, S. *Eur. J. Org. Chem.* **2009**, 3871.

⁹ (a) Marcoux, D.; Charette, A. B. *J. Org. Chem.* **2008**, *73*, 590. (b) Marcoux, D.; Charette, A. B. *Adv. Synth. Catal.* **2008**, *350*, 2967. (c) Deng, Z.; Lin, J.-H.; Xiao, J.-C. *Nat. Commun.* **2016**, *7*, 10337. (d) Toda, Y.; Komiyama, Y.; Kikuchi, A.; Suga, H. *ACS Catal.* **2016**, *6*, 6906. (e) Kim, Y.; Gabbai, F. J. *J. Am. Chem. Soc.* **2009**, *131*, 3363. (f) Moritz, R.; Wagner, M.; Schollmeyer, D.; Baumgarten, M.; Müllen, K. *Chem. - Eur. J.* **2015**, *21*, 9119.

¹⁰ Yonemori, S.; Hayashi, Y.; Kumai, S.; Wada, A. *Nippon Kagaku Kaishi* **1991**, *8*, 1146; b) Phase-Transfer Catalysis: Fundamentals, Applications, and Industrial Perspectives (Eds.: C. M. Starks, C. L. Liotta, M. Halpern), Chapman & Hall, New York, 1994, p. 132.

¹¹ He, R.; Wang, X.; Hashimoto, T.; Maruoka, K. *Angew. Chem., Int. Ed.* **2008**, *47*, 9466.

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- ¹² (a) Bhattacharya, A. K.; Thyagarajan, G. *Chem. Rev.* **1981**, *81*, 415. (b) Huang, Y.; Chew, R. J.; Pullarkat, S. A.; Li, Y.; Leung, P.-H. *J. Org. Chem.* **2012**, *77*, 6849. (c) Stewart, I. C.; Bergman, R. G.; Toste, F. D. *J. Am. Chem. Soc.* **2003**, *125*, 8696. (d) Guérard, K. C.; Hamel, V.; Guérinot, A.; Bouchard-Aubin, C.; Canesi, S. *Chem. Eur. J.* **2015**, *21*, 18068,
- ¹³ Yoshimura, A.; Zhdankin, V. V. *Chem. Rev.* **2016**, *116*, 3328
- ¹⁴ Budnikova, H.; Sinyashin, O. G. *Russ. Chem. Rev.* **2015**, *84*, 917.
- ¹⁵ Kan, T.; Fukuyama, T. *Chem. Commun.* **2004**, 353.
- ¹⁶ Kloeckner, U.; Nachtsheim, B. J. *Chem. Commun.* **2014**, *50*, 10485.
- ¹⁷ Jayalakshmi, K. L.; Thimme Gowda, B. *Zeitschrift fuer Naturforschung, A: Physical Sciences* **2004**, *59*, 491.
- ¹⁸ Massah, A. R.; Kazemi, F.; Azadi, D.; Farzaneh, S.; Aliyan, H.; Naghash, H. J.; Momeni, A. R. *Let. Org. Chem.* **2006**, *3*, 235.
- ¹⁹ Newcomer, R.; McKee, J.; Zanger, M. *Synth. Commun.* **2016**, *46*, 949.
- ²⁰ Ramírez-Martínez, J. F.; González-Chávez, R.; Guerrero-Alba, R.; Reyes-Gutiérrez, P. E.; Martínez, R.; Miranda-Morales, M.; Espinosa-Luna, R.; González-Chávez, M. M.; Barajas-López, C. *Molecules* **2013**, *18*, 894.
- ²¹ Moore, J. T.; Soldi, C.; Fettingner, J.; Shaw, J. T. *Chem. Sci.*, **2013**, *4*, 292.
- ²² Sokolov, V. V.; Butkevich, A. N.; Yuskovets, V. N.; Tomashevskii, A. A.; Potekhin, A. A. *Russ. J. Org. Chem.* **2005**, *41*, 1023.