

# IspC as Target for Antiinfective Drug Discovery: Synthesis, Enantiomeric Separation, and Structural Biology of Fosmidomycin Thia Isosters

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**(5)** Supporting Information

**ABSTRACT:** The emergence and spread of multidrug-resistant pathogens are widely believed to endanger human health. New drug targets and lead compounds exempt from cross-resistance with existing drugs are urgently needed. We report on the synthesis and properties of "reverse" thia analogs of fosmidomycin, which inhibit the first committed enzyme of a metabolic pathway that is essential for the causative agents of tuberculosis and malaria but is absent in the human host. Notably, IspC displays a high level of enantioselectivity for an  $\alpha$ -substituted fosmidomycin derivative.



## INTRODUCTION

The discussion on the current and future management of malaria, a dominant cause of morbidity and mortality, is characterized by apparently antipodal points of view. Optimistic speculation about the feasibility of eradication coexists with the actual problem of progressing resistance against most or all available drugs and the paucity of recent pharmacological innovation.<sup>1–4</sup>

Fosmidomycin (1), a phosphonate antibiotic with a hydroxamate pharmacophore that was originally isolated from *Streptomyces lavendulae*,<sup>5,6</sup> is known to have antibacterial and herbicidal activity and was more recently shown to be active against the *Plasmodium* species causing malaria.<sup>7,8</sup> Its molecular target, IspC, catalyzes the first committed step in the non-mevalonate isoprenoid biosynthesis pathway that is essential in *Plasmodium* spp. but absent in mammals (Scheme 1).<sup>9,10</sup> 1 can cure human malaria and has a favorable toxicity profile but has shortcomings with regard to pharmacokinetic aspects, which prompted structural modifications in several laboratories.<sup>11–27</sup> Also of note, 1 and its analogs are expected to be exempt from target-related toxicity and from cross-resistance with established antimalarials and may even be able to target the malaria liver

stages that are insufficiently addressed by current antimalarials.<sup>28,29</sup> We report the synthesis of thia isosters of reversed hydroxamic acid analogs of **1** and their enzymatic, antiparasite, and structural biology features. Furthermore, we show that IspC has a high degree of enantioselectivity for a reverse  $\alpha$ -aryl derivative of **1**.

#### RESULTS AND DISCUSSION

**Chemistry.** The retrosynthetic analysis of the target structures **5a**–**e** and **7a**–**c** suggested  $\alpha$ -mercaptophosphonates (**3a**–**e**), 2-chloro-*N*-hydroxyacetamide, and 2-chloro-*N*-hydroxy-*N*-methylacetamide as suitable building blocks. The alkylating reagents 2-chloro-*N*-hydroxyacetamide and 2-chloro-*N*-hydroxy-*N*-methylacetamide were obtained by published procedures.<sup>30,31</sup> Previously unreported  $\alpha$ -mercaptophosphonates **3b**–**d** were accessible according to a known procedure developed for the preparation of building block **3a**.<sup>32</sup> S-Alkylation of  $\alpha$ -mercaptophosphonates (**3a**–**e**) with 2-



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Scheme 2. Synthesis of Fosmidomycin Thia Isosters



**a**: R = Ph; **b**: R = 3,4-F-Ph; **c**: R = 3,4-Cl-Ph; **d**: R = naphthalene-1-yl; **e**: R = 4-CH<sub>3</sub>-Ph

<sup>a</sup>Reagents and conditions: (a) (1) *n*-BuLi, S, THF, -78 °C, 1 h; (2) -20 °C, THF, 1 h; (3) rt, THF, 1 h. (b)4a-e: 2-chloro-*N*-hydroxy-*N*-methylacetamide, DMF, Na<sub>2</sub>CO<sub>3</sub>, 0 °C  $\rightarrow$  rt. (c) (1) TMSBr, CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 1 h; (2) CH<sub>2</sub>Cl<sub>2</sub>, rt, 47 h; (3) THF/H<sub>2</sub>O, rt, 45 min.

chloro-*N*-hydroxyacetamides afforded partially protected intermediates 4a-e and 6a-c (Scheme 2). Deprotection of their respective phosphonate moieties by treatment with TMSBr afforded the racemic thia analogs 5a-e and 7a-c as solids or oils.<sup>33</sup>

**Biological Evaluation.** Inhibition of IspC from *Plasmodium falciparum* (*Pf*IspC) by thia analogs **5** and 7 was monitored using a photometric assay. IC<sub>50</sub> values for inhibition of *Pf*IspC and drug sensitivity assay for two different *P. falciparum* strains were in the low nanomolar to micromolar range (Tables 1 and 2 and Supporting Information Figure S1). Typical dose–response curves are shown in Figure 1. In parallel with earlier findings, the *N*-methyl substituted hydroxamates **5** showed lower IC<sub>50</sub> values than the respective unsubstituted hydroxamic acids 7.<sup>21,34</sup> Recently we reported reverse carba and oxa analogs of **1** that act as potent inhibitors of IspC and of *P. falciparum* growth.<sup>20,21,35</sup>

In order to unequivocally establish the generic impact of sulfur, oxygen, or methylene in the  $\beta$ -position of  $\alpha$ -phenylated

reverse derivatives of 1, the inhibitory activity of small compound libraries comprising isosteric sets of thia, oxa, and carba analogs were determined using IspC orthologs from Escherichia coli (EcIspC), Mycobacterium tuberculosis (MtIspC), and P. falciparum. (See Tables 1 and 2 for an overview of all IC<sub>50</sub> values.) By comparison with the carba compounds, the thia isosters show increased inhibition (by more than a factor of 10) for IspC proteins of bacterial origin (black ellipse, Figure 2a) but decreased inhibition for *Pf*IspC (red ellipse, Figure 2a). The oxa isosters were less potent inhibitors than the cognate carba compounds for all enzymes under study (red and dashed ellipses, Figure 2a). The correlation between inhibition of IspC protein and of P. falciparum blood stages is illustrated in Figure 2b and Figure 2c. One cannot observe a perfect match between activity against the enzyme and against the parasite in vitro. Consistently, the IC<sub>50</sub> values of the isolated enzyme are lower than the IC<sub>50</sub> values for parasite growth. This is derived from the fact that the compounds must penetrate several membranes in order to reach the site of IspC in the apicoplast (the

	<i>Ec</i> IspC IC <sub>50</sub> [n	[a] M]		<i>Mt</i> IspC IC <sub>50</sub> [nN	[a] /【]		<i>Pf</i> IspC IC <sub>50</sub> [n	[a] M]	
X =	S	0	С	S	0	С	S	0	С
$R_1 = Ph$ $R_2 = CH_3$	5a 8.2 ± 0.8	<b>5aO</b> 940 ± 50	<b>5aC</b> 243 ± 30	5a 280 ± 30	<b>5aO</b> 16000 ± 1000	<b>5aC</b> 2000 ± 100	<b>5a</b> 24 ± 2	<b>5aO</b> 37 ± 2	<b>5aC</b> 3.1 ± 0.3
$R_1 = 3,4\text{-F-Ph}$ $R_2 = CH_3$	<b>5b*</b> 8.5 ± 1	<b>5bO</b> 240 ± 10	<b>5bC</b> 117 ± 7	<b>5b</b> 42 ± 2	<b>5bO</b> 4600 ± 200	<b>5bC</b> 770 ± 60	<b>5b</b> 14 ± 1	<b>5bO</b> 12 ± 1	<b>5bC</b> 3.4 ± 0.4
$R_1 = 3,4-Cl-Ph$ $R_2 = CH_3$	<b>5c</b> 5.9 ± 0.6	5cO 200 ± 20	208 ± 16	<b>5c*</b> 9.2 ± 0.6	5cO 1600 ± 100	280 ± 30	5c 4.5 ± 0.4	5cO 14 ± 1	2.8 ± 0.4
$R_1 = $ naphthalene-1-yl R_2 = CH_3	<b>5d</b> 140 ± 10	<b>5dO</b> 4600 ± 300	3830 ± 170	<b>5d*</b> 550 ± 10	<b>5dO</b> 15000 ± 1000	$13000 \\ \pm \\ 1000$	<b>5d*</b> 9.8 ± 0.9	<b>5dO</b> 39 ± 4	9.4 ± 1.3
$R_1 = 4\text{-}CH_3\text{-}Ph$ $R_2 = CH_3$	5e 33 ± 9	<b>5eO</b> 320 ± 20	286 ± 19	<b>5e</b> 110 ± 10	5eO 4000 ± 200	1800 ± 100	<b>5e</b> 18 ± 4	5eO 25 ± 3	14 ± 2
$R_1 = Ph$ $R_2 = H$	7 <b>a*</b> 600 ± 90	$20000 \pm 2000$	592 ± 25	7a* 15000 ± 1000	$466000 \\ \pm \\ 60000$	$^{\pm}_{1000}$	<b>7a</b> 110 ± 20	$\begin{array}{c} 1500 \\ \pm \ 100 \end{array}$	12 ± 3
$R_1 = 3,4\text{-}F\text{-}Ph$ $R_2 = H$	7b* 77 ± 6	$49000 \pm 2000$	$\begin{array}{c} 178 \\ \pm \ 20 \end{array}$	<b>7b*</b> 1700 ± 100	> 500000	$\begin{array}{c} 3400 \\ \pm \ 300 \end{array}$	7 <b>b</b> 15 ± 2	$\begin{array}{c} 3900 \\ \pm \ 200 \end{array}$	3.9 ± 0.4
$R_1 = 3,4-Cl-Ph$ $R_2 = H$	<b>7c</b> 44 ± 2	n.s.	n.s.	<b>7c*</b> 720 ± 50	n.s.	n.s.	<b>7c</b> 200 ± 10	n.s.	n.s.
Fosmidomycin	$221\pm14.4^{[b]}$			$230\pm20$		$160 \pm 20^{[c]}$			

# Table 1. IC<sub>50</sub> Values of Thia Analogs 5 and 7 versus IspC of Escherichia coli, Mycobacterium tuberculosis, and P. falciparum<sup>d</sup>

<sup>*a*</sup>Enzyme assay. Values were calculated from at least eight data points. In general two to three independent determinations have been performed.<sup>36,37</sup> The asterisk (\*) indicates single determination. <sup>*b*</sup>IC<sub>50</sub> value according to ref 20. <sup>*c*</sup>IC<sub>50</sub> value according to ref 35. <sup>*d*</sup>n.s.: not synthesized.

Table 2. IC <sub>50</sub> Values of Thia Analogs 5 and 7 versus <i>Plasmodium</i> Strains 3D7	and Dd2 <sup>c</sup>
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HO,	Pf3D7 <sup>[a]</sup> IC <sub>50</sub> [nM]			<i>Pf</i> Dd2 <sup>[a]</sup> IC <sub>50</sub> [nM]		
X =	S	0	С	S	0	С
$R_1 = Ph$	5a	5aO	5aC	5a	5aO	5aC
$R_2 = CH_3$	30	1200	90	75	700	74
$R_1 = 3,4-F-Ph$	5b	5bO	5bC	5b	5bO	5bC
$R_2 = CH_3$	111	540	122	88	130	40
$R_1 = 3,4-Cl-Ph$ $R_2 = CH_3$	<b>5c</b> 91	<b>5cO</b> 240	n.d.	<b>5c</b> 85	<b>5cO</b> 140	n.d.
$R_1 = naphthalene-1-yl$ $R_2 = CH_3$	<b>5d</b> 380	<b>5dO</b> 520	n.d.	<b>5d</b> 510	<b>5dO</b> 350	n.d.
$\begin{array}{l} R_1 = 4\text{-}CH_3\text{-}Ph \\ R_2 = CH_3 \end{array}$	<b>5e</b> 111	<b>5eO</b> 180	n.d.	<b>5e</b> 95	<b>5eO</b> 190	n.d.
$\begin{aligned} R_1 &= Ph \\ R_2 &= H \end{aligned}$	7 <b>a</b> 2900	> 50000	400	<b>7a</b> 3770	n.d.	570
$R_1 = 3,4\text{-F-Ph}$ $R_2 = H$	<b>7b</b> 160	> 50000	75	7 <b>b</b> 330	n.d.	65
$R_1 = 3,4-Cl-Ph$ $R_2 = H$	7c 540	n.s.	n.s.	7c 870	n.s.	n.s.
Fosmidomycin			880 <sup>[b]</sup>			810 <sup>[b]</sup>

<sup>*a*</sup>In vitro assay. Values are the mean of two independent double determinations. <sup>*b*</sup>IC<sub>50</sub> value according to ref 35. <sup>*c*</sup>n.d.: not determined. n.s.: not synthesized.

membranes of the erythrocyte and the parasitophorous vacuole, the outer membrane of the parasite cell, and four membranes of the apicoplast).<sup>38</sup> In agreement with the enzyme assays, the N-methyl substituted derivatives 5 also showed higher inhibitory



Figure 1. Inhibition of PfIspC by compounds 5a (blue), 7c (brown), 7a (green), 7b (orange): residual reaction velocity vs inhibitor concentration.



**Figure 2.** (a) Inhibition of IspC orthologs by minilibraries of carba, oxa, and thia analogs. Abscissa is the log  $IC_{50}$  of carba compound. Ordinate is the activity gain with respect to loss caused by sulfur and with respect to oxygen in the  $\beta$ -position. The sulfur isosters show increased inhibition for *EcIspC* and *MtIspC* (black ellipse) but decreased inhibition for *PfIspC* (red ellipse). The oxa isosters exhibit decreased inhibition throughout (red and dashed ellipses). (b, c) Enzyme inhibition compared to inhibition of two different *P. falciparum* strains: (b) strain 3D7; (c) strain Dd2. Abscissa is the IC<sub>50</sub> for *PfIspC*. Ordinate is the difference of log IC<sub>50</sub> (enzyme) and log IC<sub>50</sub> (parasite proliferation). "C" and "O" after compound numbers designate carba and oxa isosters, respectively. See Tables 1 and 2 for details on the IC<sub>50</sub> values.

activity toward strains 3D7 and Dd2 than the free hydroxamic acids 7. The most active thia analog (**5a**) showed an IC<sub>50</sub> value of 30 nM against the *Plasmodium* 3D7 strain. Differences in membrane permeability of the compounds might also be the reason for deviating results between activities against the *Pf*IspC and the whole parasite in vitro, explaining the correlation discrepancy of the assays.

Next, we determined the three-dimensional structure of *Pf*IspC in complex with  $Mn^{2+}$ , NADPH, and **5a** at a resolution of 2.0 Å (PDB code 4KP7,  $R_{\text{free}} = 24.9\%$ ), applying Patterson search calculations with the coordinates of PfIspC (PDB code 3AU8<sup>39</sup>) as starting model. Details on data collection and refinement statistics can be found in Table 4 in the Experimental Section. The overall structure is similar to previously reported *Pf*IspC structures (root-mean-square deviation of  $C_a$  of <0.6 Å for PDB code 3AU8). The reversed hydroxamate moiety of the bound 5a chelates the metal ion (Mn<sup>2+</sup>), which is also coordinated by Asp231, Glu233, and Glu315 of the enzyme (Figure 3a and Figure 3b). In contrast to *EcIspC*, where the coordination of  $Mn^{2+}$  is octahedral because of an additional water molecule, the metal ion is 5-fold coordinated in the case of PfIspC and MtIspC complex structures (see also Supporting Information Figures S2-S4). The phosphonate group forms hydrogen bonds with Ser269, Ser270, Ser306, Asn311, Lys312, and two water molecules. Apart from Ser269, the amino acids involved in direct contact with the ligand are strictly conserved (Supporting Information Figure S5). The loop that covers the ligand is well-defined, whereas it is typically disordered in previously reported IspC structures.<sup>40</sup> The Trp296 and Met298 residues in this region further stabilize the bound ligand by hydrophobic interactions. As shown by the Connolly surface representation (Figure 3c), 5a, Mn<sup>2+</sup>, and two water molecules are enclosed in the active site cavity. The loop region over the active site is highly mobile and can thereby accept more bulky residues like the phenyl group in 5a or the naphthyl group in 5d.

As seen in the complex structure, the sulfur atom of the ligand is involved in hydrophobic interactions with the strictly conserved Met298. In contrast, the oxygen atom at the  $\beta$ -methylene position would have a repulsive effect with the sulfur



**Figure 3.** (a) Stereoview of **5a** (green) bound to the active site of *Pf*IspC (gray) at 2.0 Å resolution. NADPH is shown in gray,  $Mn^{2+}$  in orange, and the electron density (purple mesh) is displayed at  $1.0\sigma$  contour level. Ligands have been excluded prior to phase calculating. (b) Interactions of amino acid residues in contact with **5a** and  $Mn^{2+}$ . Distances are given in Å. van der Waals interactions are indicated by colored curves. (c) Stereoview of the Connolly surface representation of the active site cavity.

atom of Met298. This is in good agreement with the corresponding  $IC_{50}$  values. The carbon atom at this position is not in contact with the protein. Since the  $IC_{50}$  values for the thia analogs are significantly lower than for the carba derivatives, it can be assumed that the increased flexibility caused by the sulfur introduction facilitates the binding process

of the ligand into the active site cavity. This effect can be clearly seen for *Ec*IspC as well as *Mt*IspC. However, this correlation is less distinctive regarding *Pf*IspC.

Whereas cocrystallization had been performed with the racemic inhibitor, the omit map clearly illustrates that the enzyme selectively binds the S-enantiomer. Notably, the high



**Figure 4.** Different contour levels for the electron density map (purple mesh) of S-**5**a/PfIspC/Mn/NADPH calculated at 2.0 Å resolution. Density is displayed at (a) 1.0 $\sigma$ , (b) 2.0 $\sigma$ , and (c) 3.0 $\sigma$  contour level.

contrast provided by the electron-rich sulfur strongly supports the chirality assignment. To illustrate the strength of the crystallographic information, different contour levels for the electron density of the sulfur atom are provided in Figure 4. In previous crystallographic studies using several IspC orthologs in complex with  $\alpha$ -phenylated derivatives of 1, *S*-configuration of the enzyme-bound ligands had been the preferred interpretation,<sup>21,22,41</sup> but a definitive chirality assignment had not been possible up to now.

In order to quantitatively assess the enantioselectivity toward IspC, the enantiomers of **5a** were separated by chiral HPLC (with ee > 99.9%). The apparent IC<sub>50</sub> values of the (+)- and (-)-enantiomers measured against *Pf*IspC are 9.4 nM (+) and >10  $\mu$ M (-), respectively (Table 3, Figure 5), a similar range

# Table 3. Inhibition of IspC Enzymes of E. coli, M. tuberculosis and P. falciparum by 5a

	$\operatorname{IC}_{50}[\mathrm{nM}]^{a}$				
5a	EcIspC	<i>Mt</i> IspC	PfIspC		
R- $(-)$	$4000 \pm 700$	$83000 \pm 13000$	$12000 \pm 1000$		
S-(+)	$5.1 \pm 1.6$	$120 \pm 20$	$9.4 \pm 0.6$		
racemate	$8.2 \pm 0.8$	$280 \pm 30$	$24 \pm 2.0$		

<sup>a</sup>Enzyme assay. Values were calculated from at least eight data points. In general two to three independent determinations have been performed. See Supporting Information Figure S3 and Table S1 for details on the multiple reaction monitoring.



**Figure 5.** Inhibition of *Pf*IspC by inhibitor **5a**: *S*-(+)-enantiomer, orange; *R*-(-)-enantiomer, green; racemate, blue.  $I_e$  is an effective inhibitor concentration, which is calculated according to  $I_e = I/(1 + [S]/K_m)$  where *I* is the inhibitor concentration, [S] is the substrate concentration, and  $K_m$  is the Michaelis–Menten constant. At a residual reaction velocity of 50%,  $I_e$  is equal to the  $K_i$  (inhibition constant) of the inhibitor.

that has also been recorded for the bacterial enzymes. On the basis of the kinetic data and the crystallographic complex structure, the more active enantiomer can now be unequivocally assigned as S-(+)-**5a**. Thus, because of the strong enantioselectivity, the inhibitory activity of  $\alpha$ -substituted derivatives of **1** can be virtually doubled by racemate separation, while undesirable effects can be diminished by the removal of the less active enantiomer.

#### CONCLUSION

In summary, we provided kinetic and crystallographic evidence for the mode of action of a novel type of reverse hydroxamatebased IspC inhibitors. We could show that certain analogs of 1 inhibit IspC with IC<sub>50</sub> values in the single-digit nanomolar range while inhibiting the parasite growing in human erythrocytes with IC<sub>50</sub> values in the low nanomolar range. The introduction of a sulfur atom at the  $\beta$ -methylene group of the main chain of the ligands led to an improvement of the  $IC_{50}$ values in the cases of EcIspC and MtIspC, compared to the respective carba and oxa ligands. The presented results highlight the more potent inhibitory activity of the S-(+)-enantiomer; thus, the long pending question of the stereochemistry of chiral IspC ligands is now clarified. The achievements in the mode of action to block IspC activity might be a helpful guide for the development of more potent chiral inhibitors.

#### EXPERIMENTAL SECTION

General Procedures. All solvents and chemicals were used as purchased without further purification. The progress of all reactions was monitored on Merck precoated silica gel plates (with fluorescence indicator  $UV_{254}$ ) using ethyl acetate/*n*-hexane as solvent system. Column chromatography was performed with Fluka silica gel 60 (230-400 mesh ASTM) with the solvent mixtures specified in the corresponding experiment. Spots were visualized by irradiation with ultraviolet light (254 nm). Melting points (mp) were taken in open capillaries on a Stuart melting point apparatus SMP11 and are uncorrected. Proton (1H) and carbon (13C) NMR spectra were recorded on a Bruker Avance 500 (500.13 MHz for <sup>1</sup>H; 125.76 MHz for  ${}^{13}C$ ) using DMSO- $d_6$  as solvent. Chemical shifts are given in parts per million (ppm) ( $\delta$  relative to residual solvent peak for <sup>1</sup>H and <sup>13</sup>C). Elemental analysis was performed on a Perkin-Elmer PE 2400 CHN elemental analyzer and a vario MICRO cube elemental analyzer, Elementar Analysensysteme GmbH, Hanau Germany. IR spectra were recorded on a Varian 800 FT-IR Scimitar series. Optical rotation was determined by a Krüss P8000 polarimeter. Specific rotations  $[\alpha]_{\rm D}^{20}$  are given in deg  $cm^3$  g<sup>-1</sup> dm<sup>-1</sup>. High-resolution mass spectrometry (HRMS) analysis was performed using a UHR-TOF maXis 4G instrument (Bruker Daltonics, Bremen, Germany). If necessary, the purity was determined by high performance liquid chromatography (HPLC). Purity of all final compounds was 95% or higher. The instrument was an Elite LaChrom system [Hitachi L-2130 (pump) and L-2400 (UV-detector)]. The column was a Phenomenex Luna C-

18(2), 5  $\mu$ m particle size (250 mm × 4.6 mm), supported by Phenomenex Security Guard cartridge kit C18 (4.0 mm ×3.0 mm). Chiral HPLC separation of compound **5a** was performed on HP series 1100 (Agilent): column, Chiralcel OZ-H (Chiral Technologies Europe), 5  $\mu$ m particle size (250 mm × 4.6 mm); eluent A, *n*heptane; eluent B, propan-2-ol with 0.1% trifluoroacetic acid, isocratic (70:30) with a flow rate of 1 mL min<sup>-1</sup> and detection at 240 nm; column temperature of 25 °C; injection of 50  $\mu$ L of 34.3 mM **5a** in *n*heptane/propan-2-ol (70:30).

**Materials.** Diethylphosphonates **2a–e**,  $\alpha$ -mercaptophosphonates **3a** and **3e**, 2-chloro-N-hydroxy-N-methylacetamide, and 2-chloro-N-hydroxyacetamide have been prepared according to known procedures.<sup>30,31</sup>  $\alpha$ -Mercaptophosphonates **3a–e** were synthesized according to the procedure of Mikolajczyk.<sup>32</sup> Compound **3a** has been previously described by Mikolajczyk<sup>32</sup> and compound **3e** by Creary.<sup>42</sup>

Diethyl ((3,4-Difluorophenyl)(mercapto)methyl)phosphonate (3b). Yellow oil; yield 67% (9.931 g, 33.52 mmol). <sup>1</sup>H NMR (500.13 MHz,  $[D_6]$ DMSO):  $\delta = 7.57-7.17$  (m, 3H), 4.81  $(d, {}^{2}J(H,P) = 20.0 \text{ Hz}, 0.31\text{H}, PCH), 4.76 (d, {}^{2}J(H,P) = 19.2 \text{ Hz},$ 0.26H, PCH), 4.59 (d,  ${}^{2}J(H,P) = 19.7$  Hz, 0.21H, PCH), 4.57 (d, <sup>2</sup>*J*(H,P) = 19.7 Hz, 0.22H, PCH), 4.15–3.99 (m, 2H, POCH<sub>2</sub>), 3.99– 3.76 (m, 2H, POCH<sub>2</sub>), 3.70-3.64 (m, SH), 1.31-1.14 (m, 3H, CH<sub>3</sub>), 1.14–0.97 ppm (m, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (125.76 MHz, [D<sub>6</sub>]DMSO):  $\delta = 149.84$  (m), 147.82 (m), 135.75 (m), 132.36 (m), 132.03 (m), 126.63 (m), 125.70 (m), 118.52 (m), 117.46 (m), 62.90 (m, POCH<sub>2</sub>), 62.63 (d,  ${}^{2}J(C,P) = 6.9$  Hz, POCH<sub>2</sub>), 47.85 (d,  ${}^{1}J(C,P) = 138.6$  Hz, PC), 47.67 (d, <sup>1</sup>*J*(C,P) = 141.3 Hz, PC), 35.33 (d, <sup>1</sup>*J*(C,P) = 146.2 Hz, PC), 16.24–15.79 ppm (m, CH<sub>3</sub>). IR (NaCl):  $\tilde{\nu}$  =2985 (s;  $\nu$ (C–  $H_{aliph}$ )), 2913 (m;  $\nu$ (C- $H_{aliph}$ )), 2507 (w;  $\nu$ (S-H)), 1250 (s;  $\nu$ (P= O)), 1050 (s;  $\nu$ (P–O)), 1025 cm<sup>-1</sup> (s;  $\nu$ (P–O)). HRMS (ESI): m/z calculated for C<sub>11</sub>H<sub>15</sub>F<sub>2</sub>O<sub>3</sub>PS + H<sup>+</sup> [M + H<sup>+</sup>]: 297.052 03. Found: 297.052 15.

Diethyl ((3,4-Dichlorophenyl)(mercapto)methyl)phosphonate (3c). Yellow oil; yield 62% (10.201 g, 30.99 mmol). <sup>1</sup>H NMR (500.13 MHz,  $[D_6]$ DMSO):  $\delta = 8.09$  (d, J = 2.1 Hz, 0.05H), 7.93 (d, J = 2.1 Hz, 0.02H), 7.92 (d, J = 2.2 Hz, 0.03H), 7.85 (s, 0.04H), 7.83 (s, 0.04H), 7.80-7.77 (m, 0.08H), 7.73-7.61 (m, 1.2H), 7.60–7.48 (m, 0.7H), 7.45–7.31 (m, 0.84H), 5.28 (d,  ${}^{2}J$ (H,P) = 20.5 Hz, 0.06H, PCH), 5.04 (d,  ${}^{2}J(H,P) = 19.8$  Hz, 0.1H, PCH), 5.02 (d,  ${}^{2}I(H,P) = 20.1 \text{ Hz}, 0.11 \text{H}, PCH), 4.93 (d, {}^{2}I(H,P) = 20.2 \text{ Hz}, 0.14 \text{H},$ PCH), 4.88 (d,  ${}^{2}J(H,P) = 20.1$  Hz, 0.33H, PCH), 4.79 (d,  ${}^{2}J(H,P) =$ 19.1 Hz, 0.26H, PCH), 4.17-3.79 (m, 4H, POCH<sub>2</sub>), 1.31-1.15 (m, 3.34H, CH<sub>3</sub>), 1.13–0.98 ppm (m, 2.66H, CH<sub>3</sub>). <sup>13</sup>C NMR (125.76 MHz,  $[D_6]DMSO$ :  $\delta$  = 135.66 (d, <sup>2</sup>J(C,P) = 4.6 Hz), 135.40 (d,  ${}^{4}J(C,P) = 2.9 \text{ Hz}$ , 131.44 (m), 130.70 (m), 129.75 (m), 63.03 (m, POCH<sub>2</sub>), 62.49 (m, POCH<sub>2</sub>), 47.61 (d,  ${}^{1}J(C,P) = 137.5$  Hz, PC), 16.05 ppm (m, CH<sub>3</sub>). IR (NaCl):  $\tilde{\nu}$  =2983 (s;  $\nu$ (C-H<sub>aliph</sub>)), 2910 (s;  $\nu$ (C-H<sub>aliph</sub>)), 2562 (w;  $\nu$ (S-H)), 1251 (s;  $\nu$ (P=O)), 1028 cm<sup>-1</sup> (s;  $\nu$ (P–O)). HRMS (ESI): m/z calculated for C<sub>11</sub>H<sub>15</sub>Cl<sub>2</sub>O<sub>3</sub>PS + H<sup>+</sup> [M + H<sup>+</sup>]: 328.992 93. Found: 328.992 98.

**Diethyl (Mercapto(naphthalene-1-yl)methyl)phosphonate** (3d). Yellow oil; yield 84% (13.028 g, 41.98 mmol). <sup>1</sup>H NMR (500.13 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 8.30 (s, 1H), 7.95 (d, *J* = 7.6 Hz, 1.20H), 7.88 (d, *J* = 7.9 Hz, 1.80H), 7.64–7.46 (m, 3H), 5.32–5.18 (m, 1H, PCH), 4.19–4.10 (m, 2H, POCH<sub>2</sub>), 3.87–3.77 (m, 1H, POCH<sub>2</sub>), 3.70–3.60 (m, 1H, POCH<sub>2</sub>), 3.59–3.52 (m, SH), 1.25 (t, <sup>3</sup>*J*(H,H) = 7.0 Hz, 3H, CH<sub>3</sub>), 0.85 ppm (t, <sup>3</sup>*J*(H,H) = 7.0 Hz, 3H, CH<sub>3</sub>), 1.3C NMR (125.76 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 133.84 (m), 133.24, 130.51 (d, <sup>3</sup>*J*(C,P) = 7.6 Hz), 128.60, 128.02, 127.08 (m), 126.29 (m), 125.81 (m), 125.26 (m), 123.54 (m), 62.92 (d, <sup>2</sup>*J*(C,P) = 7.0 Hz, POCH<sub>2</sub>), 32.23 (m, PC), 31.02 (m, PC), 16.29 (m, CH<sub>3</sub>), 15.80 ppm (m, CH<sub>3</sub>). IR (NaCl):  $\tilde{\nu}$  =2981 (s;  $\nu$ (C–H<sub>aliph</sub>)), 2517 (w;  $\nu$ (S–H)), 1249 (s;  $\nu$ (P=O)), 1051 (s;  $\nu$ (P–O)), 1024 cm<sup>-1</sup> (s;  $\nu$ (P–O)). HRMS (ESI): *m*/*z* calculated for C<sub>15</sub>H<sub>19</sub>O<sub>3</sub>PS + H<sup>+</sup> [M + H<sup>+</sup>]: 311.086 53. Found: 311.086 91.

General Procedure for the Synthesis of Intermediates 4a-eand 6a-c. After purification, compounds 3a-e were immediately used for the synthesis of 4a-e and 6a-c. Under a nitrogen atmosphere, the respective  $\alpha$ -mercaptophosphonate 3a-e (1 equiv, 10 mmol) was dissolved in dry dimethylformamide (10 mL). After

addition of Na<sub>2</sub>CO<sub>3</sub> (1.25 equiv, 1.32 g, 12.5 mmol) the reaction mixture was cooled to 0 °C. In the case of the preparation of 4a-e, a solution of 2-chloro-N-hydroxy-N-methylacetamide (1 equiv, 1.24 g, 10 mmol) in dry dimethylformamide (2 mL) was added dropwise. For the synthesis of 6a-c, 2-chloro-N-hydroxyacetamide (1 equiv, 1.10 g, 10 mmol) in dry dimethylformamide (2 mL) was used as alkylating agent. The reaction mixture was allowed to warm to room temperature and stirred for 2 h while monitoring the progress of the reaction by thin layer chromatography. Ethyl acetate (40 mL) and 1 M hydrochloric acid (10 mL) were added, and the aqueous DMF layer was extracted and separated. Next the aqueous DMF layer was extracted three more times with ethyl acetate  $(3 \times 30 \text{ mL})$ . The ethyl acetate was washed with 1 M hydrochloric acid (10 mL), dried over Na<sub>2</sub>SO<sub>4</sub>, filtered, and evaporated to dryness. The residues were purified by column chromatography on silica gel using ethyl acetate/nhexane (10:90  $\rightarrow$  100:0) to give hydroxamic acids 4a-e and 6b as white solids and compounds 6a,c as colorless oils.

Diethyl (((2-Hydroxy(methyl)amino)-2-oxoethyl)thio-(phenyl)methyl)phosphonate (4a). White solid; yield 41% (1.421 g, 4.09 mmol); mp 79 °C. <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ :  $\delta = 10.02$  (br s, OH), 7.45–7.39 (m, 2H), 7.38–7.32 (m, 2H), 7.32-7.26 (m, 1H), 4.57 (d,  ${}^{2}J$ (H,P) = 19.7 Hz, 1H, PCH), 4.12-3.99 (m, 2H, POCH<sub>2</sub>), 3.95-3.84 (m, 1H, POCH<sub>2</sub>), 3.84-3.73 (m, 1H, POCH<sub>2</sub>), 3.53 (d,  ${}^{2}J(H,H) = 14.7$  Hz, 1H, SCH<sub>2</sub>), 3.34 (d,  ${}^{2}J(H,H) = 14.7 \text{ Hz}, 1H, \text{ SCH}_{2}, 3.06 \text{ (s, 3H, NCH}_{3}, 1.22 \text{ (t, }{}^{3}J(H,H))$ = 7.0 Hz, 3H, CH<sub>3</sub>), 1.04 ppm (t,  ${}^{3}J(H,H) = 7.0$  Hz, 3H, CH<sub>3</sub>).  ${}^{13}C$ NMR (125.76 MHz,  $[D_6]DMSO$ ):  $\delta = 168.41$  (C=O), 135.55 (d,  ${}^{2}J(C,P) = 4.2 \text{ Hz}$ , 129.12 (d,  ${}^{3}J(C,P) = 6.4 \text{ Hz}$ ), 128.23, 127.52, 62.50  $(d_{1}^{2}J(C,P) = 6.9 \text{ Hz}, \text{ POCH}_{2}), 62.42 (d_{1}^{2}J(C,P) = 6.9 \text{ Hz}, \text{ POCH}_{2}),$ 43.37 (d,  ${}^{1}J(C,P) = 143.6$  Hz, PC), 35.72 (NCH<sub>3</sub>), 32.12 (d,  ${}^{3}J(C,P) =$ 6.6 Hz, SCH<sub>2</sub>), 16.13 (d,  ${}^{3}J(C,P) = 5.6$  Hz, CH<sub>3</sub>), 15.92 (d,  ${}^{3}J(C,P) =$ 5.5 Hz, CH<sub>3</sub>). IR (KBr):  $\tilde{\nu}$  =3103 (s;  $\nu$ (O-H)), 2983 (s;  $\nu$ (C- $H_{aliph}$ )), 2902 (s;  $\nu$ (C- $H_{aliph}$ )), 1650 (s;  $\nu$ (C=O)), 1209 (s;  $\nu$ (P= O), 1042 cm<sup>-1</sup> (s;  $\nu$ (P–O)). Elemental analysis calculated (%) for C14H22NO5PS: C 48.41, H 6.38, N 4.03. Found: C 48.58, H 6.51, N 3.87.

Diethyl ((3,4-Difluorophenyl)((2-(hydroxy(methyl)amino)-2oxoethyl)thio)methyl)phosphonate (4b). White solid; yield 44% (1.691 g, 4.41 mmol); mp 102 °C. <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ :  $\delta = 10.02$  (br s, OH), 7.49–7.38 (m, 2H), 7.31–7.23 (m, 1H), 4.64 (d, <sup>2</sup>J(H,P) = 20.3 Hz, 1H, PCH), 4.12-4.04 (m, 2H, POCH<sub>2</sub>), 3.97-3.89 (m, 1H, POCH<sub>2</sub>), 3.89-3.82 (m, 1H, POCH<sub>2</sub>),  $3.52 (d, {}^{2}J(H,H) = 14.8 Hz, 1H, SCH_{2}), 3.38 (d, {}^{2}J(H,H) = 14.9 Hz,$ 1H, SCH<sub>2</sub>), 3.05 (s, 3H, NCH<sub>3</sub>), 1.23  $(t, {}^{3}J(H,H) = 7.0$  Hz, 3H, CH<sub>3</sub>), 1.08 ppm (t,  ${}^{3}J(H,H) = 7.0$  Hz, 3H, CH<sub>3</sub>).  ${}^{13}C$  NMR (125.76 MHz,  $[D_6]DMSO$ :  $\delta = 168.11$  (C=O), 149.75 (m), 147.81 (m), 133.57, 126.09 (m), 117.88 (dd,  ${}^{2}I(C,F) = 17.5$  Hz,  ${}^{3}I(C,F) = {}^{3}I(C,P) = 6.2$ Hz), 117.32 (d,  ${}^{2}J(C,F) = 17.3$  Hz), 62.74 (d,  ${}^{2}J(C,P) = 7.2$  Hz, POCH<sub>2</sub>), 62.66 (d,  ${}^{2}J(C,P) = 7.3$  Hz, POCH<sub>2</sub>), 42.49 (d,  ${}^{1}J(C,P) =$ 144.1 Hz, PC), 35.72 (NCH<sub>3</sub>), 32.37 (d,  ${}^{3}J(C,P) = 7.6$  Hz, SCH<sub>2</sub>), 16.12 (d,  ${}^{3}I(C,P) = 5.7$  Hz, CH<sub>3</sub>), 15.96 ppm (d,  ${}^{3}I(C,P) = 5.5$  Hz, CH<sub>3</sub>). IR (KBr):  $\tilde{\nu}$  =3169 (s;  $\nu$ (O–H)), 2989 (s;  $\nu$ (C–H<sub>aliph</sub>)), 2910 (s;  $\nu$ (C–H<sub>aliph</sub>)), 1642 (s;  $\nu$ (C=O)), 1230 (s;  $\nu$ (P=O)), 1022 cm<sup>-1</sup> (s;  $\nu(P-O)$ ). Elemental analysis calculated (%) for  $C_{14}H_{20}F_2NO_5PS$ : C 43.86, H 5.26, N 3.65. Found: C 44.09, H 5.41, N 3.55.

**Diethyl ((3,4-Dichlorophenyl)((2-(hydroxy(methyl)amino)-2oxoethyl)thio)methyl)phosphonate (4c).** White solid; yield 51% (2.119 g, 5.09 mmol); mp 86 °C. <sup>1</sup>H NMR (500.13 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 10.03 (br s, OH), 7.67–7.62 (m, 2H), 7.44–7.39 (m, 1H), 4.67 (d, <sup>2</sup>*J*(H,P) = 20.5 Hz, 1H, PCH), 4.12–4.03 (m, 2H, POCH<sub>2</sub>), 3.99–3.83 (m, 2H, POCH<sub>2</sub>), 3.51 (d, <sup>2</sup>*J*(H,H) = 14.8 Hz, 1H, SCH<sub>2</sub>), 3.40 (d, <sup>2</sup>*J*(H,H) = 14.9 Hz, 1H, SCH<sub>2</sub>), 3.05 (s, 3H, NCH<sub>3</sub>), 1.23 (t, <sup>3</sup>*J*(H,H) = 7.0 Hz, 3H, CH<sub>3</sub>), 1.09 ppm (t, <sup>3</sup>*J*(H,H) = 7.0 Hz, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (125.76 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 168.03 (C=O), 137.14 (d, <sup>2</sup>*J*(C,P) = 4.2 Hz), 130.81 (d, <sup>3</sup>*J*(C,P) = 6.5 Hz), 130.73, 130.49, 130.15 (d, <sup>4</sup>*J*(C,P) = 2.8 Hz), 129.38 (d, <sup>3</sup>*J*(C,P) = 6.2 Hz), 62.84 (d, <sup>2</sup>*J*(C,P) = 6.8 Hz, POCH<sub>2</sub>), 62.70 (d, <sup>2</sup>*J*(C,P) = 6.9 Hz, POCH<sub>2</sub>), 42.41 (d, <sup>1</sup>*J*(C,P) = 143.0 Hz, PC), 35.72 (NCH<sub>3</sub>), 32.53 (d, <sup>3</sup>*J*(C,P) = 7.6 Hz, SCH<sub>2</sub>), 16.12 (d, <sup>3</sup>*J*(C,P) = 5.7 Hz, CH<sub>3</sub>), 15.96 ppm (d, <sup>3</sup>*J*(C,P) = 5.5 Hz, CH<sub>3</sub>). IR (KBr):  $\tilde{\nu}$  =3184 (s;  $\nu$ (O–H)), 2979 (s;  $\nu$ (C–H<sub>aliph</sub>)), 2916 (s;  $\nu$ (C–H<sub>aliph</sub>)), 1648 (s;  $\nu$ (C=O)), 1226 (s;  $\nu$ (P=O)), 1052 cm<sup>-1</sup> (s;  $\nu$ (P–O)). Elemental analysis calculated (%) for C<sub>14</sub>H<sub>20</sub>Cl<sub>2</sub>NO<sub>5</sub>PS: C 40.40, H 4.84, N 3.36. Found: C 40.65, H 4.83, N 3.30.

Diethyl (((2-(Hydroxy(methyl)amino)-2-oxoethyl)thio)-(naphthalen-1-yl)methyl)phosphonate (4d). White solid; yield 70% (2.778 g, 6.99 mmol); mp 85 °C. <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ :  $\delta = 10.02$  (br s, OH), 8.22 (d, J = 7.6 Hz, 1H), 7.96 (d, J = 7.8 Hz, 1H), 7.89 (d, J = 8.0 Hz, 1H), 7.87-7.81 (m, 1H), 7.62-7.50 (m, 3H), 5.49 (d,  ${}^{2}J(H,P) = 19.6$  Hz, 1H, PCH), 4.13–4.04 (m, 2H, POCH<sub>2</sub>) 3.90-3.80 (m, 1H, SCH<sub>2</sub>), 3.76-3.64 (m, 2H, POCH<sub>2</sub>), 3.43-3.37 (m, 1H, SCH<sub>2</sub>), 3.06 (s, 3H, NCH<sub>3</sub>), 1.24-1.17 (m, 3H, CH<sub>3</sub>), 0.94–0.86 ppm (m, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (125.76 MHz,  $[D_6]DMSO$ :  $\delta = 168.60$  (C=O), 133.27 (m), 131.66 (m), 131.14 (m), 128.61, 128.08, 127.16 (m), 126.33 (m), 125.78, 125.22, 123.28 (m), 62.56 (d,  ${}^{2}J(C,P) = 7.0$  Hz, POCH<sub>2</sub>), 37.46 (d,  ${}^{1}J(C,P) = 147.51$ Hz, PC), 35.72 (NCH<sub>3</sub>), 32.29 (m, SCH<sub>2</sub>), 16.16 (d,  ${}^{3}J(C,P) = 5.8$  Hz, CH<sub>3</sub>), 15.80 ppm (d,  ${}^{3}J(C,P) = 5.7$  Hz, CH<sub>3</sub>). IR (KBr):  $\tilde{\nu} = 3154$  (s;  $\nu$ (O–H)), 2983 (s;  $\nu$ (C–H<sub>aliph</sub>)), 2921 (s;  $\nu$ (C–H<sub>aliph</sub>)), 1651 (s;  $\nu$ (C=O)), 1225 (s;  $\nu$ (P=O)), 1020 cm<sup>-1</sup> (s;  $\nu$ (P–O)). Elemental analysis calculated (%) for C<sub>18</sub>H<sub>24</sub>NO<sub>5</sub>PS: C 54.40, H 6.09, N 3.52. Found: C 54.52, H 6.14, N 3.47.

Diethyl (((2-(Hydroxy(methyl)amino)-2-oxoethyl)thio)(ptolyl)methyl)phosphonate (4e). White solid; yield 27% (0.979 g, 2.71 mmol); mp 57 °C. <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ ):  $\delta =$ 10.00 (br s, OH), 7.29 (d,  ${}^{3}J(H,H) = 7.8$  Hz, 2H), 7.15 (d,  ${}^{3}J(H,H) =$ 7.8 Hz, 2H), 4.52 (d,  ${}^{2}J(H,P) = 19.6$  Hz, 1H, PCH), 4.12–3.97 (m, 2H, POCH<sub>2</sub>), 3.94-3.84 (m, 1H, POCH<sub>2</sub>), 3.84-3.72 (m, 1H, POCH<sub>2</sub>), 3.51 (d, <sup>2</sup>*J*(H,H) = 14.8 Hz, 1H, SCH<sub>2</sub>), 3.30-3.26 (m, 1H,  $SCH_2$ ), 3.06 (s, 3H, NCH<sub>3</sub>), 2.29 (s, 3H, CH<sub>3arom</sub>), 1.22 (t, <sup>3</sup>J(H,H) = 7.0 Hz, 3H, CH<sub>3</sub>), 1.06 ppm (t,  ${}^{3}J(H,H) = 7.0$  Hz, 3H, CH<sub>3</sub>).  ${}^{13}C$ NMR (125.76 MHz,  $[D_6]$ DMSO):  $\delta = 168.46$  (C=O), 136.71 (d,  ${}^{5}J(C,P) = 1.9 \text{ Hz}$ , 132.38 (d,  ${}^{2}J(C,P) = 4.5 \text{ Hz}$ ), 129.00 (d,  ${}^{3}J(C,P) =$ 6.4 Hz), 128.79, 62.42 (d,  ${}^{2}J(C,P) = 7.2$  Hz, POCH<sub>2</sub>), 62.35 (d,  ${}^{2}J(C,P) = 7.0 \text{ Hz}, \text{ POCH}_{2}, 43.06 \text{ (d, } {}^{1}J(C,P) = 144.2 \text{ Hz}, PC), 35.71$ (NCH<sub>3</sub>), 32.00 (d,  ${}^{3}J(C,P) = 6.5$  Hz, SCH<sub>2</sub>), 20.60 (CH<sub>3arom</sub>), 16.13  $(d, {}^{3}J(C,P) = 5.6 \text{ Hz}, \text{CH}_{3}), 15.95 \text{ ppm} (d, {}^{3}J(C,P) = 5.6 \text{ Hz}, \text{CH}_{3}). \text{ IR}$ (KBr):  $\tilde{\nu} = 3165$  (s;  $\nu$ (O–H)), 2983 (s;  $\nu$ (C–H<sub>aliph</sub>)), 2921 (s;  $\nu$ (C–  $H_{aliph}$ )), 1650 (s;  $\nu$ (C=O)), 1222 (s;  $\nu$ (P=O)), 1035 cm<sup>-1</sup> (s;  $\nu$ (P-O)). Elemental analysis calculated (%) for C<sub>15</sub>H<sub>24</sub>NO<sub>5</sub>PS: C 49.85, H 6.69, N 3.88. Found: C 49.74, H 6.73, N 3.70.

Diethyl (((2-Hydroxyamino)-2-oxoethyl)thio)(phenyl)methyl)phosphonate (6a). Colorless oil; yield 15% (0.500 g, 1.50 mmol). <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ ):  $\delta = 10.59$  (br s, OH), 8.94 (br s, NH), 7.44-7.38 (m, 2H), 7.37-7.34 (m, 2H), 7.31-7.28 (m, 1H), 4.62 (d,  ${}^{2}J(H,P) = 19.7$  Hz, 1H, PCH), 4.11–4.02 (m, 2H, POCH<sub>2</sub>), 3.94-3.85 (m, 1H, POCH<sub>2</sub>), 3.83-3.74 (m, 1H, POCH<sub>2</sub>) 3.08 (d,  ${}^{2}J(H,H) = 14.0$  Hz, 1H, SCH<sub>2</sub>), 2.88 (d,  ${}^{2}J(H,H) = 13.9$  Hz, 1H, SCH<sub>2</sub>), 1.22 (t,  ${}^{3}J(H,H) = 7.0$  Hz, 3H, CH<sub>3</sub>), 1.04 ppm (t,  ${}^{3}J(H,H) = 7.0 \text{ Hz}, 3H, CH_{3}$ ).  ${}^{13}C \text{ NMR} (125.76 \text{ MHz}, [D_{6}]\text{DMSO}): \delta$ = 165.80 (C=O), 135.76 (d,  ${}^{2}J(C,P)$  = 4.3 Hz), 129.61 (d,  ${}^{3}J(C,P)$  = 6.2 Hz), 128.71, 128.04 (d,  ${}^{5}J(C,P) = 1.5$  Hz), 63.04 (d,  ${}^{2}J(C,P) = 7.3$  Hz, POCH<sub>2</sub>), 62.98 (d,  ${}^{2}J(C,P) = 7.4$  Hz, POCH<sub>2</sub>), 44.39 (d,  ${}^{1}J(C,P)$ = 143.9 Hz, PC), 31.83 (d,  ${}^{3}J(C,P)$  = 6.5 Hz, SCH<sub>2</sub>), 16.60 (d,  ${}^{3}J(C,P)$ = 5.5 Hz, CH<sub>3</sub>), 16.37 ppm (d,  ${}^{3}J(C,P)$  = 5.5 Hz, CH<sub>3</sub>). IR (NaCl):  $\tilde{\nu}$ =3385 (s;  $\nu$ (N–H)), 3192 (s;  $\nu$ (O–H)), 2982 (s;  $\nu$ (C–H<sub>aliph</sub>)), 2931 (s;  $\nu$ (C–H<sub>aliph</sub>)), 1651 (s;  $\nu$ (C=O)), 1240 (s;  $\nu$ (P=O)), 1021 cm<sup>-1</sup> (s;  $\nu(P-O)$ ). Elemental analysis calculated (%) for C<sub>13</sub>H<sub>20</sub>NO<sub>5</sub>PS: C 46.84, H 6.05, N 4.20. Found: C 46.76, H 6.10, N 3.90.

**Diethyl** ((3,4-Difluorophenyl))((2-(hydroxyamino)-2oxoethyl)thio)methyl)phosphonate (6b). White solid; yield 31% (1.141 g, 3.09 mmol); mp 105 °C. <sup>1</sup>H NMR (500.13 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 10.65 (br s, OH), 10.58 (br s, OH), 10.23 (br s, OH), 10.14 (br s, OH), 9.26 (br s, NH), 9.07 (br s, NH), 8.95 (br s, NH), 7.54–7.36 (m, 2H), 7.36–7.19 (m, 1H), 4.87 (d, <sup>2</sup>J(H,P) = 19.2 Hz, 0.12H, PCH), 4.67 (d, <sup>2</sup>J(H,P) = 20.5 Hz, 0.88H, PCH), 4.16– 4.02 (m, 2H, POCH<sub>2</sub>), 4.01–3.78 (m, 2H, POCH<sub>2</sub>), 3.07 (d, <sup>2</sup>J(H,H) = 13.9 Hz, 1H, SCH<sub>2</sub>), 2.92 (d, <sup>2</sup>J(H,H) = 13.9 Hz, 1H, SCH<sub>2</sub>), 1.23 (t, <sup>3</sup>J(H,H) = 7.0 Hz, 3H, CH<sub>3</sub>), 1.08 ppm (t, <sup>3</sup>J(H,H) = 7.0 Hz, 3H, CH<sub>3</sub>). <sup>13</sup>C NMR (125.76 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 165.00 (C=O), 164.40 (C=O), 148.94 (dd, <sup>1</sup>J(C,F) = 244.6 Hz, <sup>2</sup>J(C,F) = 12.7 Hz), 148.78 (d, <sup>1</sup>J(C,F) = 246.6 Hz), 133.33 (m), 126.18 (m), 117.97 (dd, <sup>2</sup>J(C,F) = 18.0 Hz, <sup>3</sup>J(C,F) = <sup>3</sup>J(C,P) = 6.1 Hz), 117.40 (d, <sup>2</sup>J(C,F) = 17.0 Hz), 62.86 (d, <sup>2</sup>J(C,P) = 6.8 Hz, POCH<sub>2</sub>), 62.75 (d, <sup>2</sup>J(C,P) = 6.8 Hz, POCH<sub>2</sub>), 48.29 (d, <sup>1</sup>J(C,P) = 139.7 Hz, PC), 43.08 (d, <sup>1</sup>J(C,P) = 144.0 Hz, PC), 31.50 (d, <sup>3</sup>J(C,P) = 7.4 Hz, SCH<sub>2</sub>), 16.16 (d, <sup>3</sup>J(C,P) = 5.5 Hz, CH<sub>3</sub>), 15.97 ppm (d, <sup>3</sup>J(C,P) = 5.4 Hz, CH<sub>3</sub>). IR (KBr):  $\tilde{\nu}$ =3192 (s;  $\nu$ (N–H),  $\nu$ (O–H)), 2990 (s;  $\nu$ (C–H<sub>aliph</sub>)), 2918 (s;  $\nu$ (C– H<sub>aliph</sub>)), 1652 (s;  $\nu$ (C=O)), 1236 (s;  $\nu$ (P=O)), 1026 cm<sup>-1</sup> (s;  $\nu$ (P– O)). Elemental analysis calculated (%) for C<sub>13</sub>H<sub>16</sub>F<sub>2</sub>NO<sub>5</sub>PS: C 42.28, H 4.91, N 3.79. Found: C 42.08, H 4.82, N 3.86.

Diethyl ((3,4-Dichlorophenyl)((2-(hydroxyamino)-2oxoethyl)thio)methyl)phosphonate (6c). Colorless oil; yield 38% (1.528 g, 3.80 mmol). <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ ):  $\delta$ = 10.66 (br s, OH), 10.59 (br s, OH), 10.15 (br s, OH), 9.26 (br s, NH), 9.09-9.06 (m, NH), 8.99-8.90 (m, NH), 7.66 (d, J = 8.3 Hz, 1H), 7.64–7.58 (m, 1H), 7.46–7.36 (m, 1H), 4.91 (d,  ${}^{2}J(H,P) = 19.3$ Hz, 0.09H, PCH), 4.69 (d,  ${}^{2}J(H,P) = 20.7$  Hz, 0.91H, PCH), 4.16– 4.02 (m, 2H, POCH<sub>2</sub>), 3.98-3.81 (m, 2H, POCH<sub>2</sub>), 3.06 (dd, 13.8 Hz, 1H, SCH<sub>2</sub>), 1.23 (t,  ${}^{3}J(H,H) = 7.0$  Hz, 3H, CH<sub>3</sub>), 1.09 ppm  $(t, {}^{3}J(H,H) = 7.0 \text{ Hz}, 3H, CH_{3})$ .  ${}^{13}C \text{ NMR} (125.76 \text{ MHz}, 125.76 \text{ MHz})$  $[D_6]$ DMSO):  $\delta = 164.86$  (C=O), 136.87 (d, <sup>2</sup>J(C,P) = 4.1 Hz), 130.85 (d,  ${}^{3}J(C,P) = 6.4$  Hz), 130.78 (d,  ${}^{4}J(C,P) = 1.3$  Hz), 130.53 (d,  ${}^{5}J(C,P) = 1.1 \text{ Hz}$ , 130.26 (d,  ${}^{4}J(C,P) = 3.0 \text{ Hz}$ ), 129.41 (d,  ${}^{3}J(C,P) = 3.0 \text{ Hz}$ ) 6.1 Hz), 62.93 (d,  ${}^{2}J(C,P) = 6.7$  Hz, POCH<sub>2</sub>), 62.75 (d,  ${}^{2}J(C,P) = 6.8$ Hz, POCH<sub>2</sub>), 42.98 (d,  ${}^{1}J(C,P) = 143.4$  Hz, PC), 31.57 (d,  ${}^{3}J(C,P) =$ 7.5 Hz, SCH<sub>2</sub>), 16.15 (d,  ${}^{3}J(C,P) = 5.5$  Hz, CH<sub>3</sub>), 15.96 ppm (d,  ${}^{3}J(C,P) = 5.5 \text{ Hz}, \text{ CH}_{3}$ ). IR (NaCl):  $\tilde{\nu} = 3190 \text{ (s; } \nu(N-H), \nu(O-H)),$ 2985 (s;  $\nu$ (C-H<sub>aliph</sub>)), 2910 (s;  $\nu$ (C-H<sub>aliph</sub>)), 1661 (s;  $\nu$ (C=O)), 1239 (s;  $\nu(P=O)$ ), 1024 cm<sup>-1</sup> (s;  $\nu(P-O)$ ). Elemental analysis calculated (%) for  $C_{13}H_{18}Cl_2NO_5PS$ : C 38.82, H 4.51, N 3.48. Found: C 39.10, H 4.69, N 3.42.

General Procedure for the Synthesis of Compounds 5a-e and 7a–c. The appropriate phosphonic acid diethyl ester 4a–e, 6a–c (1equiv, 1 mmol) was dissolved in dry dichloromethane (10 mL) under an atmosphere of nitrogen. The solution was cooled to 0 °C, and trimethylsilyl bromide (5 equiv, 0.66 mL, 5 mmol) was added dropwise via a syringe. After being stirred for 1 h, the reaction mixture was allowed to warm to room temperature and was stirred for further 47 h. The solvent was removed under reduced pressure, and tetrahydrofuran (10 mL) was added to the residue. The solution was treated with water (3 equiv, 0.05 mL, 3 mmol) and the reaction mixture was stirred for 45 min at room temperature. The solvent was evaporated and the residue was dried in vacuo overnight. Addition of ethyl acetate afforded phosphonic acids 5a-e as white solids. Target compounds 7a-c were obtained as colorless oils after purification of the crude products by filtration through a small amount of silica gel using ethyl acetate as the eluent. <sup>1</sup>H and <sup>13</sup>C NMR spectra for compounds 5a-e and 7a-c are provided in Figures S7-S22.

(((2-(Hydroxy(methyl)amino)-2-oxoethyl)thio)(phenyl)methyl)phosphonic Acid (5a). White solid; yield 38% (0.111 g, 0.38 mmol); mp 112 °C. <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ ):  $\delta =$ 9.99 (br s, OH), 7.47-7.35 (m, 2H), 7.34-7.26 (m, 2H), 7.26-7.18 (m, 1H), 4.25 (d,  ${}^{2}J(H,P) = 19.0$  Hz, 1H, PCH), 3.50 (d,  ${}^{2}J(H,H) =$ 14.6 Hz, 1H, SCH<sub>2</sub>), 3.23 (d,  ${}^{2}J$ (H,H) = 14.6 Hz, 1H, SCH<sub>2</sub>), 3.06 (s, 3H, NCH<sub>3</sub>). <sup>13</sup>C NMR (125.76 MHz,  $[D_6]DMSO$ ):  $\delta$  = 168.93 (C= O), 137.36 (d,  ${}^{2}J(C,P) = 3.8 \text{ Hz}$ ), 129.23 (d,  ${}^{3}J(C,P) = 5.8 \text{ Hz}$ ), 127.86, 126.78, 45.72 (d,  ${}^{1}J(C,P) = 139.2$  Hz, PC), 35.71 (NCH<sub>3</sub>), 31.85 (d,  ${}^{3}J(C,P) = 6.2$  Hz, SCH<sub>2</sub>). IR (KBr):  $\tilde{\nu} = 3192$  (s;  $\nu$ (O–H)), 2925 (s;  $\nu$ (C–H<sub>aliph</sub>)), 1630 (s;  $\nu$ (C=O)), 1191 (s;  $\nu$ (P=O)), 987 cm<sup>-1</sup> (s;  $\nu$ (P–O)). HRMS (ESI): m/z calculated for C<sub>10</sub>H<sub>14</sub>NO<sub>5</sub>PS + H<sup>+</sup> [M + H<sup>+</sup>]: 292.040 31. Found: 292.039 87. HPLC analysis: retention time = 1.870 min; peak area, 99.08%; eluent A, NH<sub>4</sub>OAc solution (5 mM); eluent B, CH<sub>3</sub>CN; isocratic (1:1) over 20 min with a flow rate of 1 mL min<sup>-1</sup>.

(R)-5a:  $[\alpha]_D^{20}$  -80.67 deg cm<sup>3</sup> g<sup>-1</sup> dm<sup>-1</sup> (c 0.011 g cm<sup>-3</sup> in methanol with 0.1% trifluoroacetic acid).

(S)-**5a**:  $[\alpha]_{20}^{0}$  +80.53 deg cm<sup>3</sup> g<sup>-1</sup> dm<sup>-1</sup> (c 0.001 g cm<sup>-3</sup> in methanol with 0.1% trifluoroacetic acid).

((3,4-Difluorophenyl)((2-(hydroxy(methyl)amino)-2oxoethyl)thio)methyl)phosphonic Acid (5b). White solid; yield 45% (0.15 g, 0.46 mmol); mp 137 °C. <sup>1</sup>H NMR (500.13 MHz,  $[D_6]$ DMSO):  $\delta = 10.04$  (br s, OH), 7.51–7.31 (m, 2H), 7.28–7.14 (m, 1H), 4.28 (d,  ${}^{2}J(H,P) = 19.6$  Hz, 0.83H, PCH), 4.21 (d,  ${}^{2}J(H,P) =$ 19.2 Hz, 0.17H, PCH), 3.46 (d,  ${}^{2}J(H,H) = 14.6$  Hz, 1H, SCH<sub>2</sub>), 3.23  $(d, {}^{2}J(H,H) = 14.6 Hz, 1H, SCH_{2}), 3.09 (s, 0.21H, NCH_{3}), 3.05 (s, 0.21H, NCH_{3}))$ 2.54H, NCH<sub>3</sub>), 3.01 ppm (s, 0.25H, NCH<sub>3</sub>). <sup>13</sup>C NMR (125.76 MHz,  $[D_6]DMSO$ :  $\delta = 168.66$  (C=O), 148.79 (dd,  ${}^1J(C,F) = 244.7$  Hz,  ${}^{2}J(C,F) = 11.7$  Hz), 148.34 (dd,  ${}^{1}J(C,F) = 244.8$  Hz,  ${}^{2}J(C,F) = 12.5$ Hz), 135.58, 126.12 (m), 117.80 (dd,  ${}^{2}J(C,F) = 17.7$  Hz,  ${}^{3}J(C,F) =$  ${}^{3}I(C,P) = 5.6 \text{ Hz}$ , 116.77 (d,  ${}^{2}I(C,F) = 17.0 \text{ Hz}$ ), 45.01 (d,  ${}^{1}I(C,P) =$ 137.8 Hz, PC), 35.70 (NCH<sub>3</sub>), 31.88 ppm (d, <sup>3</sup>J(C,P) = 7.4 Hz, SCH<sub>2</sub>). IR (KBr):  $\tilde{\nu} = 3162$  (s;  $\nu$ (O–H)), 2925 (s;  $\nu$ (C–H<sub>aliph</sub>)), 1609 (s;  $\nu$ (C=O)), 1196 (s;  $\nu$ (P=O)), 988 cm<sup>-1</sup> (s;  $\nu$ (P-O)). Elemental analysis calculated (%) for C<sub>10</sub>H<sub>12</sub>F<sub>2</sub>NO<sub>5</sub>PS: C 36.70, H 3.70, N 4.28, found: C 36.97, H 4.00, N 3.92.

((3,4-Dichlorophenyl)((2-(hydroxy(methyl)amino)-2oxoethyl)thio)methyl)phosphonic Acid (5c). White solid; yield 45% (0.162 g, 0.45 mmol); mp 134 °C. <sup>1</sup>H NMR (500.13 MHz, [D<sub>6</sub>]DMSO): δ = 10.01 (br s, OH), 7.70–7.52 (m, 2H), 7.45–7.30 (m, 1H), 4.32 (d, <sup>2</sup>J(H,P) = 19.8 Hz, 0.86H, PCH), 4.24 (d, <sup>2</sup>J(H,P) = 19.8 Hz, 0.14H, PCH), 3.47 (d, <sup>2</sup>J(H,H) = 14.6 Hz, 1H, SCH<sub>2</sub>), 3.25 (d, <sup>2</sup>J(H,H) = 14.6 Hz, 1H, SCH<sub>2</sub>), 3.12–2.97 ppm (m, 3H, NCH<sub>3</sub>). <sup>13</sup>C NMR (125.76 MHz, [D<sub>6</sub>]DMSO): δ = 168.55 (C=O), 138.99 (m), 130.84 (d, <sup>3</sup>J(C,P) = 5.8 Hz), 130.42 (d, <sup>4</sup>J(C,P) = 1.3 Hz), 130.07, 129.63 (d, <sup>3</sup>J(C,P) = 5.8 Hz), 129.42 (d, <sup>4</sup>J(C,P) = 2.3 Hz), 44.85 (d, <sup>1</sup>J(C,P) = 137.3 Hz, PC), 35.71 (NCH<sub>3</sub>), 31.96 ppm (d, <sup>3</sup>J(C,P) = 7.4 Hz, SCH<sub>2</sub>). IR (KBr):  $\tilde{\nu}$  =3175 (s;  $\nu$ (O–H)), 2923 (s;  $\nu$ (C–H<sub>aliph</sub>)), 1618 (s;  $\nu$ (C=O)), 1191 (s;  $\nu$ (P=O)), 984 cm<sup>-1</sup> (s;  $\nu$ (P–O)). Elemental analysis calculated (%) for C<sub>10</sub>H<sub>12</sub>Cl<sub>2</sub>NO<sub>5</sub>PS: C 33.35, H 3.36, N 3.89. Found: C 33.06, H 3.47, N 3.63.

(((2-(Hydroxy(methyl)amino-2-oxoethyl)thio)(naphthalen-1yl)methyl)phosphonic Acid (5d). White solid; yield 55% (0.188 g, 0.55 mmol); mp 137 °C. <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ ):  $\delta =$ 9.99 (br s, OH), 8.20 (d, J = 6.9 Hz, 1H), 7.92 (d, J = 7.9 Hz, 1H), 7.88 (s, 1H), 7.83 (d, J = 8.1 Hz, 1H), 7.59-7.44 (m, 3H), 5.19 (d,  ${}^{2}J(H,P) = 18.3$  Hz, 1H, PCH), 3.74 (d,  ${}^{2}J(H,H) = 14.2$  Hz, 1H,  $SCH_2$ ), 3.34 (d,  ${}^{2}J(H,H) = 14.5$  Hz, 1H,  $SCH_2$ ), 3.05 ppm (s, 3H, NCH<sub>3</sub>). <sup>13</sup>C NMR (125.76 MHz,  $[D_6]$ DMSO):  $\delta = 169.11$  (C=O), 133.55 (m), 133.26 (m), 131.28 (m), 128.46, 127.36, 126.95 (m), 126.00 (m), 125.47, 125.22, 123.61, 38.61 (m, PC), 35.72 (NCH<sub>3</sub>), 32.34 ppm (m, SCH<sub>2</sub>). IR (KBr):  $\tilde{\nu}$  =3225 (s;  $\nu$ (O-H)), 2928 (s;  $\nu$ (C-H<sub>aliph</sub>)), 1596 (s;  $\nu$ (C=O)), 1132 (s;  $\nu$ (P=O)), 1021 cm<sup>-1</sup> (s;  $\nu$ (P–O)). HRMS (ESI): m/z calculated for C<sub>14</sub>H<sub>16</sub>NO<sub>5</sub>PS + H<sup>+</sup> [M + H<sup>+</sup>]: 342.055 96. Found: 342.055 37. HPLC analysis: retention time = 1.860 min; peak area, 99.39%; eluent A, NH<sub>4</sub>OAc solution (5 mM); eluent B, CH<sub>3</sub>CN; isocratic (1:1) over 20 min with a flow rate of 1 mL min<sup>-1</sup> and detection at 254 nm; column temperature, rt.

(((2-(Hydroxy(methyl)amino)-2-oxoethyl)thio)(p-tolyl)methyl)phosphonic Acid (5e). White solid; yield 36% (0.110 g, 0.36 mmol); mp 100 °C. <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ ):  $\delta =$ 10.92 (br s, OH), 10.03 (br s, OH), 7.27 (d,  ${}^{3}J(H,H) = 7.9$  Hz, 2H), 7.10 (d,  ${}^{3}J(H,H) = 7.8$  Hz, 2H), 4.19 (d,  ${}^{2}J(H,P) = 18.9$  Hz, 0.86H, PCH), 4.10 (d,  ${}^{2}J(H,P) = 19.1$  Hz, 0.14H, PCH), 3.48 (d,  ${}^{2}J(H,H) =$ 14.6 Hz, 1H, SCH<sub>2</sub>), 3.20 (d,  ${}^{2}J(H,H) = 14.6$  Hz, 1H, SCH<sub>2</sub>), 3.12-2.99 (m, 3H, NCH<sub>3</sub>), 2.34–2.21 (m, 3H, CH<sub>3arom</sub>). <sup>13</sup>C NMR (125.76 MHz,  $[D_6]DMSO$ ):  $\delta$  = 168.99 (C=O), 135.87, 134.32 (m), 129.13  $(d, {}^{3}J(C,P) = 5.7 \text{ Hz}), 128.44, 45.46 (d, {}^{1}J(C,P) = 139.5 \text{ Hz}, PC),$ 35.71 (NCH<sub>3</sub>), 31.79 (m, SCH<sub>2</sub>), 20.60 (CH<sub>3arom</sub>). IR (KBr):  $\tilde{\nu}$  =3116 (s;  $\nu$ (O-H)), 3011 (s;  $\nu$ (C-H<sub>aliph</sub>)), 2928 (s;  $\nu$ (C-H<sub>aliph</sub>)), 1625 (s;  $\nu$ (C=O)), 1190 (s;  $\nu$ (P=O)), 986 cm<sup>-1</sup> (s;  $\nu$ (P-O)). HRMS (ESI): m/z calculated for C<sub>11</sub>H<sub>16</sub>NO<sub>5</sub>PS - H<sup>+</sup> [M - H<sup>+</sup>]: 304.04140. Found: 304.041 65. HPLC analysis: retention time = 2.753 min; peak area, 95.72%; eluent, CH<sub>3</sub>CN; 20 min with a flow rate of 1 mL min<sup>-1</sup> and detection at 254 nm; column temperature, rt.

(((2-(Hydroxyamino)-2-oxoetĥyl)thio)(phenyl)methyl)phosphonic Acid (7a). Colorless oil; yield 61% (0.169 g, 0.61 mmol). <sup>1</sup>H NMR (500.13 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 10.61 (br s, NOH), 10.17 (br s, NH), 7.47–7.18 (m, 5H), 6.41 (br, s, POH), 4.33 (d, <sup>2</sup>*J*(H,P) = 18.9 Hz, 0.84, PCH), 4.18 (d, <sup>2</sup>*J*(H,P) = 18.8 Hz, 0.16H, PCH), 3.47–3.36 (m, 0.11H, SCH<sub>2</sub>), 3.29 (d, <sup>2</sup>*J*(H,H) = 14.8 Hz, 0.11H, SCH<sub>2</sub>), 3.08 (d, <sup>2</sup>*J*(H,H) = 13.9 Hz, 0.85H, SCH<sub>2</sub>), 3.00 (d, <sup>2</sup>*J*(H,H) = 14.9 Hz, 0.18H, SCH<sub>2</sub>), 2.81 (d, <sup>2</sup>*J*(H,H) = 13.9 Hz, 0.75H, SCH<sub>2</sub>). <sup>13</sup>C NMR (125.76 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 165.95 (C=O), 137.17 (d, <sup>2</sup>*J*(C,P) = 4.3 Hz), 129.33 (d, <sup>3</sup>*J*(C,P) = 5.8 Hz), 127.99, 126.95, 46.20 (d, <sup>1</sup>*J*(C,P) = 139.2 Hz, PC), 31.48 (d, <sup>3</sup>*J*(C,P) = 6.1 Hz, SCH<sub>2</sub>). IR (NaCl):  $\tilde{\nu}$  =3457 (s;  $\nu$ (N–H)), 3155 (s;  $\nu$ (O–H)), 3001 (s;  $\nu$ (C–H<sub>aliph</sub>)), 1653 (s;  $\nu$ (C=O)), 1161 (s;  $\nu$ (P=O)), 1006 cm<sup>-1</sup> (s;  $\nu$ (P–O)). HRMS (ESI): *m*/z calculated for C<sub>9</sub>H<sub>12</sub>NO<sub>3</sub>PS – H<sup>+</sup> [M – H<sup>+</sup>]: 276.010 10. Found: 276.010 106. HPLC analysis: retention time = 1.850 min; peak area, 99.82%; eluent A, NH<sub>4</sub>OAc solution (5 mM); eluent B, CH<sub>3</sub>CN; isocratic (1:1) over 20 min with a flow rate of 1 mL min<sup>-1</sup> and detection at 254 nm; column temperature, rt.

((3,4-Difluorophenyl)((2-(hydroxyamino)-2-oxoethyl)thio)methyl)phosphonic Acid (7b). Colorless oil; yield 42% (0.132 g, 0.42 mmol). <sup>1</sup>H NMR (500.13 MHz, [D<sub>6</sub>]DMSO):  $\delta = 10.57$  (br s, NOH), 9.00 (br s, NH, POH), 7.63–7.09 (m, 3H), 4.46 (d, <sup>2</sup>*J*(H,P) = 19.1 Hz, 0.09H, PCH), 4.36 (d,  ${}^{2}I(H,P) = 19.6$  Hz, 0.59H, PCH), 4.23 (d, <sup>2</sup>J(H,P) = 19.5 Hz, 0.32H, PCH), 3.76-3.62 (m, 0.02H, SCH<sub>2</sub>), 3.62-3.46 (m, 0.06H, SCH<sub>2</sub>), 3.45-3.06 (m, 0.92H, SCH<sub>2</sub>), 13.02 (d,  ${}^{2}J(H,H) = 13.8 \text{ Hz}, 0.52 \text{H}, \text{ SCH}_{2}), 2.81 \text{ ppm} (d, {}^{2}J(H,H) = 13.8 \text{ Hz},$ 0.48H, SCH<sub>2</sub>). <sup>13</sup>C NMR (125.76 MHz,  $[D_6]$ DMSO):  $\delta$  = 165.53 (C=O), 148.85 (dd,  ${}^{1}J(C,F) = 244.5$  Hz,  ${}^{2}J(C,F) = 12.2$  Hz), 148.47  $(dd, {}^{1}J(C,F) = 245.6 \text{ Hz}, {}^{2}J(C,F) = 11.4 \text{ Hz}), 135.09 (m), 126.20,$ 117.88 (dd,  ${}^{2}J(C,F) = 17.5$  Hz,  ${}^{3}J(C,F) = {}^{3}J(C,P) = 4.5$  Hz), 116.88  $(d, {}^{2}J(C,F) = 17.1 \text{ Hz}), 45.40 (d, {}^{1}J(C,P) = 137.9 \text{ Hz}, PC), 45.31 (d, {}^{2}J(C,P) = 137.9 \text{ Hz})$  ${}^{1}J(C,P) = 137.8$  Hz, PC), 33.27 (d,  ${}^{3}J(C,P) = 7.5$  Hz, SCH<sub>2</sub>), 31.35 ppm (d,  ${}^{3}J(C,P) = 7.1$  Hz, SCH<sub>2</sub>). IR (NaCl):  $\tilde{\nu} = 3418$  (s;  $\nu$ (N–H)), 3154 (s;  $\nu$ (O-H)), 3001 (s;  $\nu$ (C-H<sub>aliph</sub>)), 2926 (s;  $\nu$ (C-H<sub>aliph</sub>)), 1653 (s;  $\nu$ (C=O)), 1119 (s;  $\nu$ (P=O)), 1006 cm<sup>-1</sup> (s;  $\nu$ (P-O)). HRMS (ESI): m/z calculated for C<sub>9</sub>H<sub>10</sub>F<sub>2</sub>NO<sub>5</sub>PS + H<sup>+</sup> [M + H<sup>+</sup>]: 314.005 81. Found: 314.006 21. HPLC analysis: retention time = 1.790 min; peak area, 99.76%; eluent A, NH<sub>4</sub>OAc solution (5 mM); eluent B,  $CH_3CN$ ; isocratic (1:1) over 20 min with a flow rate of 1 mL min<sup>-1</sup> and detection at 254 nm; column temperature, rt.

((3,4-Dichlorophenyl)((2-(hydroxyamino)-2-oxoethyl)thio)methyl)phosphonic Acid (7c). Colorless oil; yield 44% (0.152 g, 0.44 mmol). <sup>1</sup>H NMR (500.13 MHz,  $[D_6]DMSO$ ):  $\delta = 10.57$  (br s, NOH), 10.06 (br s, NOH), 9.26 (br s, NH, POH), 8.98 (br s, NH, POH), 7.71-7.48 (m, 2H), 7.45-7.24 (m, 1H), 4.51 (d, <sup>2</sup>J(H,P) = 19.0 Hz, 0.06H, PCH), 4.37 (d, <sup>2</sup>J(H,P) = 19.8 Hz, 0.6H, PCH), 4.24  $(d_1^2 I(H,P) = 19.9 \text{ Hz}, 0.34 \text{H}, PCH), 3.71 - 3.48 (m, 0.28 \text{H}, SCH_2),$ 3.48-3.32 (m, 0.26H, SCH<sub>2</sub>), 3.31-3.06 (m, 0.55H, SCH<sub>2</sub>), 3.02 (d,  $^{2}J(H,H) = 13.8$  Hz, 0.47H, SCH<sub>2</sub>), 2.82 (d,  $^{2}J(H,H) = 13.9$  Hz, 0.44H, SCH<sub>2</sub>). <sup>13</sup>C NMR (125.76 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 165.42 (C=O), 138.78 (d,  ${}^{2}J(C,P) = 4.2 \text{ Hz}$ ), 130.89 (d,  ${}^{3}J(C,P) = 5.9 \text{ Hz}$ ), 130.48 (d,  ${}^{4}J(C,P) = 1.1 \text{ Hz}$ , 130.12, 129.67 (d,  ${}^{3}J(C,P) = 5.7 \text{ Hz}$ ), 129.51 (d,  ${}^{4}J(C,P) = 2.7$  Hz), 45.63 (d,  ${}^{1}J(C,P) = 137.1$  Hz, PC), 45.36 (d,  ${}^{1}J(C,P) = 137.0$  Hz, PC), 31.37 (d,  ${}^{3}J(C,P) = 7.7$  Hz, SCH<sub>2</sub>). IR (NaCl):  $\tilde{\nu} = 3192$  (s;  $\nu$ (N-H),  $\nu$ (O-H)), 2978 (s;  $\nu$ (C-H<sub>aliph</sub>)), 1651 (s;  $\nu$ (C=O)), 1134 (s;  $\nu$ (P=O)), 1028 cm<sup>-1</sup> (s;  $\nu$ (P-O)). HRMS (ESI): m/z calculated for C<sub>9</sub>H<sub>10</sub>Cl<sub>2</sub>NO<sub>5</sub>PS - H<sup>+</sup> [M - H<sup>+</sup>]: 343.932 16. Found: 343.932 51. HPLC analysis: retention time = 1.823 min; peak area, 99.95%; eluent A, NH<sub>4</sub>OAc solution (5 mM); eluent B, CH<sub>3</sub>CN; isocratic (1:1) over 20 min with a flow rate of 1 mL min<sup>-1</sup> and detection at 254 nm; column temperature, rt.

Biological Evaluation. Gene Expression and Protein Purification. The recombinant *E. coli* strain M15 pREP4 pQEIspCplasart<sup>20</sup> was grown to an optical density of 0.6 at 37 °C in LB medium supplemented with ampicillin (180 mg L<sup>-1</sup>) and kanamycin (50 mg L<sup>-1</sup>). Isopropylthiogalactoside was added to a final concentration of 1 mM, and the cell suspension was cultivated overnight at 30 °C under shaking (120 rpm). Cells were harvested by centrifugation, washed with 0.9% NaCl solution, and resuspended in loading buffer (100 mM Tris hydrochloride, pH 7.4, containing 0.5 M NaCl and 20 mM imidazole). The suspension was passed through a cell disruption device (basic Z model, Constant Systems Limited, Northamptonshire, U.K.) and was then centrifuged. The supernatant was subjected to a Ni<sup>2+</sup>-chelating Sepharose fast flow column (volume of 40 mL) that had been pre-equilibrated with loading buffer. The column was flushed with 100 mM Tris hydrochloride, pH 7.4, containing 0.5 M NaCl and 100 mM imidazole to remove unbound protein and was then developed with 100 mM Tris hydrochloride, pH 7.4, containing 0.5 M NaCl and 250 mM imidazole. Fractions were combined and dialyzed overnight against 50 mM Tris hydrochloride, pH 7.4, containing 2 mM DTT and 0.02% NaN<sub>3</sub>. *Pf* IspC was obtained with approximately 95% purity, as estimated by SDS–PAGE analysis.

**IC**<sub>50</sub> **Determination Using the Photometric Assay.** Assays were conducted in 96-well plates (Nunc, catalog no. 781602) with transparent flat bottoms. Assay mixtures (total volume of 200 μL) contained 100 mM Tris hydrochloride, pH 7.6, 5 mM MgCl<sub>2</sub>, 5 mM DTT, 0.5 mM NADPH, 0.02 U of IspC protein, and test compounds. Dilution series (1:3) of inhibitors approximately covered the concentration range of 200–0.003 μM. The reaction was started by addition of 1-deoxy-D-xylulose 5-phosphate in 100 mM Tris hydrochloride, pH 8.0, to a final concentration of 0.5–5.0 mM. The reaction was monitored photometrically (room temperature) at 340 nm using a plate reader (SpectraMax M5, Molecular Devices, Biberach an der Riss, Germany). IC<sub>50</sub> values (mean ± standard deviation calculated from eight or more measuring points) were determined using nonlinear regression analysis as described earlier.<sup>41,42</sup> The IC<sub>50</sub> values are provided in Tables 1 and 3.

In Vitro Drug Sensitivity Assay. The chloroquine sensitive *P. falciparum* laboratory strain 3D7 and the chloroquine-resistant strain Dd2 were kept in continuous culture as described previously.<sup>43</sup> In brief, parasites were kept in an incubator at 37 °C in an atmosphere of 5% CO<sub>2</sub> and 5% oxygen at 5% hematocrit in erythrocytes of blood group 0 rhesus + in complete culture medium (RPMI 1640, 25 mM 4- (2-hydroxyethyl)piperazine-N'-(4-butanesulfonic acid), 2 mM L-glutamine, 50  $\mu$ g mL<sup>-1</sup> gentamycin, and 0.5% w/v albumax) with daily change of medium. Synchronization of parasite culture was performed by sorbitol.<sup>44</sup>

Tested compounds were dissolved in DMSO at stock concentrations between 25 and 100 mM, and further dilutions were done with complete culture medium. Final concentrations of solvent did not interfere with parasite growth (data not shown).

Activity of compounds was evaluated in a drug sensitivity assay by measurement of histidine rich protein 2 (HRP2) with an enzymelinked immunsorbent assay (ELISA) according to standard procedures.<sup>45</sup> In brief, 96-well plates were precoated with a 3-fold dilution of the respective compound before adding ring stage parasites at a parasitemia of 0.05% and a hematocrit of 1.5% in a total volume of 225  $\mu$ L per well. After an incubation time of 3 days plates were freeze– thawed twice before performance of HRP2-ELISA. As internal control chloroquine was evaluated in each assay. Experiments were performed in duplicate.

**Statistics.** The 50% inhibitory concentrations ( $IC_{50}$ ) was determined by analyzing the log of the concentration–response curves by nonlinear regression analysis using the drc-package, version 0.9.0, of R, version 2.10.1 (R Development Core Team).<sup>46</sup> The  $IC_{50}$  values are provided in Table 2.

Crystallization and Structure Determination. Pf IspC protein (11 mg mL<sup>-1</sup> in 50 mM Tris hydrochloride, pH 7.4, containing 2 mM DTT and 0.02% NaN<sub>3</sub>) was incubated with 1.5 mM 5a (racemate), 640  $\mu$ M MnCl<sub>2</sub>, and 800  $\mu$ M NADPH prior to crystallization. Crystals were prepared by using the sitting drop vapor diffusion method at 20 °C. The droplets comprised 0.1  $\mu$ L of protein and 0.1  $\mu$ L of reservoir solution (2% PEG4000, 100 mM Na acetate, and 15% MPD, pH 5.0). Crystals grew to a final size of  $500 \times 100 \times 50 \ \mu m^3$  within 6 weeks. PfIspC cocrystals were supercooled without further cryoprotectant in a stream of nitrogen gas at 100 K (Oxford Cryo Systems). A native data set at 2.0 Å resolution was collected using synchrotron radiation at the X06SA beamline, Swiss Light Source, Villigen, Switzerland. Data were processed using the program package XDS,<sup>47</sup> and data reduction was performed with XSCALE with a final R<sub>free</sub> value of 24.9% and rootmean-square deviation (rmsd) bond and angle values of 0.09 Å and 1.2°, respectively (Table 4). Crystal structure analysis was carried out by molecular replacement using coordinates of PfIspC deposited at

Table 4. Data Collection and Refinement Statistics

	S-5a/PfIspC/Mn/NADPH
	Crystal Parameter
space group	P2 <sub>1</sub>
cell dimensions	$a = 51.5$ Å, $b = 78.1$ Å, $c = 100.0$ Å; $\beta = 91.5^{\circ}$
molecules per AU <sup>a</sup>	2
	Data Collection
beamline	SLS, X06SA
wavelength (Å)	1.0
resolution range $(Å)^b$	40-2.0 (2.1-2.0)
observed/unique <sup>c</sup> reflections	196791/52039
completeness (%) <sup>b</sup>	97.1 (96.7)
$R_{\text{merge}} (\%)^{b,d}$	8.2 (53.3)
$I/\sigma(I)^{b}$	12.7 (3.0)
	Refinement
resolution (Å)	15-2.0
$R_{\rm work}/R_{\rm free}^{e}$	18.5/24.9
no. atoms	6790
protein	6468
ligand	84
Mn <sup>2+</sup>	2
water	236
B-factors	37.0
rmsd <sup>f</sup>	
bond length (Å)	0.009
bond angle (deg)	1.20
Ramachandran (%) <sup>g</sup>	97.3/2.7/0.0
PDB accession code	4KP7

<sup>a</sup>Asymmetric unit. <sup>b</sup>Values in parentheses of resolution range, completeness,  $R_{merge'}$  and  $I/\sigma(I)$  correspond to the last resolution shell. <sup>c</sup>Friedel pairs were treated as identical reflections. <sup>d</sup> $R_{merge}(I) = \sum_{hkl} \sum_{j} |[I(hkl)_j - I(hkl)]| / \sum_{hkl} I_{hkp}$  where  $I(hkl)_j$  is the measurement of the intensity of reflection hkl and  $\langle I(hkl) \rangle$  is the average intensity. <sup>e</sup> $R = \sum_{hkl} ||F_{obs}| - |F_{calc}|| / \sum_{hkl} |F_{obs}|$ , where  $R_{free}$  is calculated without a  $\sigma$  cutoff for a randomly chosen 5% of reflections, which were not used for structure refinement, and  $R_{work}$  is calculated for the remaining reflections. <sup>f</sup>Root mean square deviations from ideal bond lengths/ angles. <sup>g</sup>Number of residues in favored region/allowed region/outlier region.

the Protein Data Bank (PDB code 3AU8<sup>37</sup>). The anisotropy of diffraction was corrected by TLS refinement using the program REFMAC5.<sup>48</sup> Conventional crystallographic rigid body, positional, and temperature factor refinements were calculated with REFMAC5. The model was completed using the interactive three-dimensional graphic program MAIN.<sup>49</sup> The atomic coordinates for *Pf*IspC in complex with *S*-**5a**, Mn<sup>2+</sup>, and NADPH have been deposited in the Protein Data Bank, Research Collaboratory for Structural Bioinformatics at Rutgers University, under the accession code 4KP7.

LC–MS System. LC–MS analysis was performed with an Agilent (Santa Clara, CA, USA) 1200 HPLC system consisting of a binary pump (G1312B), cooled autosampler (G1367D), thermostated column compartment (G1316), and degasser (G1379B) coupled to an ABSciex (Framingham, MA, USA) API4000 mass spectrometer equipped with a Turbo V IonSpray source (electrospray ionization) equipped with a reversed phase Phenomenex Synergy C<sub>12</sub> column (150 mm × 2 mm, 4  $\mu$ m, 80 Å). Samples were chromatographed isocratically using water/methanol (3:1 v/v) with 0.1% formic acid, at 0.2 mL min<sup>-1</sup>. The retention volume for the **5a** racemate as well as for each enantiomer was 0.74 mL. Data were acquired and analyzed using ABSciex "Analyst" software (version 1.6.2, Framingham, MA, USA).

Identification and Evaluation of Compound 5a Enantiomers 1 and 2. Mass spectra of enantiomers were obtained using the API4000 in multiple-reaction-monitoring (MRM) mode. The precursor ion with m/z value of 290 (negative mode) and its diagnostic product ions were defined using an automatic compound optimization algorithm. MRM parameters are shown in Supporting Information Table S1. Peak identification was performed by comparison of retention times and usage of 5-MRM-transitions (Supporting Information Figure S6). The concentration of enantiomers 1 and 2 in stock solution was determined using peak integration of the most intensive MRM transition (m/z 243). The original **5a** sample (racemic mixture) was used as an external standard.

**Polarimetry.** Optical rotation was measured using a P8000 polarimeter from Krüss Optronic GmbH (Hamburg, Germany). Measurements were performed at 20 °C. The solvent was methanol containing 0.1% trifluoroacetic acid.

### ASSOCIATED CONTENT

#### **S** Supporting Information

Additional figures illustrating inhibition data, superpositions, and alignment of IspC from different organisms, details of multiple reaction monitoring, and NMR spectra. This material is available free of charge via the Internet at http://pubs.acs.org.

#### **Accession Codes**

PDB code for PfIspC with bound 5a is 4KP7.

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#### **Author Contributions**

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#### Notes

The authors declare no competing financial interest.

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## ABBREVIATIONS USED

br s, broad signal; *n*-BuLi, *n*-butyllithium; DMAPP, dimethylallyl pyrophosphate; DMF, *N*,*N*-dimethylformamide; DMSO, dimethyl sulfoxide; DOXP, 1-deoxy-D-xylulose 5-phosphate; Dxr (IspC), 1-deoxy-D-xylulose 5-phosphate reductoisomerase; ee, enantiomeric excess; IPP, isopentenyl pyrophosphate; MEP, 2*C*-methyl-D-erythritol 4-phosphate; mp, melting point; NADPH, nicotinamide adenine dinucleotide phosphate; rt, room temperature; S, sulfur; THF, tetrahydrofuran; TMSBr, trimethylsilyl bromide

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