# Design and Synthesis of a Series of 6-Substituted-2-pyridinylmethylamine Derivatives as Novel, High-Affinity, Selective Agonists at 5-HT<sub>1A</sub> Receptors

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A search for novel, selective agonists with high intrinsic activity at the 5-HT<sub>1A</sub> subtype of serotonin (5-HT) receptors was undertaken. Mechanistic and thermodynamic considerations led to the design of 6-substituted-2-pyridinylmethylamine as a potential 5-HT<sub>1A</sub> pharmacophore. Various adducts derived from the 6-substituted-2-pyridinylmethylamine moiety were tested for their affinity at 5-HT<sub>1A</sub>,  $\alpha_1$ -adrenergic, and D<sub>2</sub>-dopaminergic receptors. Compounds with high affinity for 5-HT<sub>1A</sub> receptors (p $K_i \ge 8$ ) were examined for agonist properties by measuring their ability to inhibit forskolin-stimulated cAMP production in HA7 cells (i.e., HeLa cells permanently transfected with the h5-HT<sub>1A</sub> receptor gene and expressing the h5-HT<sub>1A</sub> receptor protein). Several compounds of the type aryl{4-[(6-substituted-pyridin-2-ylmethylamino)methyl]piperidin-1-yl}methanone had nanomolar affinity for 5-HT<sub>1A</sub> binding sites and were more than 500-fold selective with respect to  $\alpha_1$  and  $D_2$  sites. Importantly, their 5-HT<sub>1A</sub> agonist properties were demonstrated in HA7 cells where they behaved as potent inhibitors of cAMP accumulation. In particular, (3,4-dichlorophenyl){4-[(6-oxazol-5-ylpyridin-2-ylmethylamino)methyl]piperidin-1-yl}methanone (70) and (3,4-dichlorophenyl){4-[(6-azetidinopyridin-2-ylmethylamino)methyl]piperidin-1-yl}methanone (36) appeared to be more potent than, and at least as efficacious as, the prototypical 5-HT<sub>IA</sub> agonist  $(\pm)$ -8-OH-DPAT. SAR studies revealed that the pyridine nitrogen atom and the nature and the position of the substituents on the pyridine ring were critically involved in the ability of the compounds to recognize and activate 5-HT<sub>1A</sub> receptors. Structural modifications of the nonpharmacophoric part of the molecule showed, however, that the entire structure was required for affinity at 5-H $T_{1A}$  binding sites.

# **Introduction**

Serotonin (5-HT) recognizes several, distinct cell-surface receptors. Among these, 5-HT $_{1A}$  receptors, at which 5-HT has high affinity, have been implicated in many psychiatric and neurological disorders. Clinical evidence of a link between 5-HT $_{1A}$  receptor function and, for example, depression is now becoming available. However, to represent a significant advance in antidepressant therapy, 5-HT $_{1A}$  receptor ligands must achieve two objectives not yet met by currently available treatments: (1) a fast onset of action and (2) a low percentage of "non-responders".

Our research program in the 5-HT $_{1A}$  area is guided by the hypothesis that very few, if any, of the ligands reported so far possess the high level of intrinsic activity needed to exert rapid and robust antidepressant effects. Thus, we postulate that a 5-HT $_{1A}$  agonist with an optimized level of intrinsic activity may offer a more rapid and effective antidepressant treatment than existing drugs, including 5-HT $_{1A}$  partial agonists. Numerous ligands at 5-HT $_{1A}$  receptors have been identified. They can be divided, however, into a small number of chemical classes. Our effort is aimed at discovering a structurally novel family of 5-HT $_{1A}$  ligands, assuming this to be the best approach to obtain 5-HT $_{1A}$  agonists

with an improved profile. Here, we describe the design, synthesis, and preliminary pharmacological characterization of a series of 6-substituted-pyridin-2-ylmethylamine derivatives. Affinity of members of this class for 5-HT $_{1A}$  binding sites was compared with their affinity for  $\alpha_1$ -adrenergic and  $D_2$ -dopaminergic sites. Furthermore, their agonist activity at 5-HT $_{1A}$  receptors was examined in vitro by measuring their ability to inhibit forskolin-stimulated cAMP production in HA7 cells. Several of the compounds had an affinity for, and a selectivity at, 5-HT $_{1A}$  receptors at least as high as the prototypical 5-HT $_{1A}$  agonist  $(\pm)$ -8-OH-DPAT. Finally, several compounds appeared to have greater intrinsic activity at 5-HT $_{1A}$  receptors than  $(\pm)$ -8-OH-DPAT.

## Chemistry

Products of general formula 4 were prepared according to the methods depicted in Scheme 1. The Schiff base 1, obtained by condensation of the commercially available 4-(aminomethyl)piperidine with the appropriate carboxaldehydes, enabled the secondary amino group to be selectively acylated. Reduction of the crude imine 2 with KBH4 in methanol afforded the desired derivatives 4 (method A). Pyridine-type compounds 4 (X = N), containing an N-bound 6-substituent (i.e., 6-alkylamino, 6-dialkylamino, 6-cycloalkylamino, 6-pyrrol-1-yl, or 6-imidazol-1-yl), were conveniently prepared from the 6-halogeno derivative 3 and the appropriate amine or sodium salt of the amine, via a SNAr at the last step of the synthesis (method B). The homologues

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# Scheme 1a

<sup>a</sup> Conditions: (a) toluene, 110 °C; (b) ArCOCl, THF, TEA; (c) KBH<sub>4</sub>, CH<sub>3</sub>OH; (d) NH<sub>3</sub>, RNH<sub>2</sub>, RRNH, EtOH, 100 °C or nitrogen heterocycles, HNa, DMF, 60 °C.

#### Scheme 2<sup>a</sup>

<sup>a</sup> Conditions: (a) Swern oxidation; (b) 2-, 3-, or 4-pyridinylethylamine, PhCH<sub>3</sub>, 110 °C; (c) KBH<sub>4</sub>, CH<sub>3</sub>OH.

# Scheme 3<sup>a</sup>

 $^a$  Conditions: (a) tBuOCH(NMe2)2, 150 °C; (b) NaIO4, H2O, 25 °C; (c) HOCH2CH2OH, PTSA, PhCH3, 110 °C; (d) HNa, pyrazole, DMF, 60 °C; (e) HCO<sub>2</sub>H, H<sub>2</sub>O, CuSO<sub>4</sub>, 65 °C.

**6** were obtained by reductive amination of aldehydes **5** with the commercially available 2-(aminoethyl)pyridine (Scheme 2).

Known aldehydes of the type **8** (Schemes 3-7), such as 2-pyridinepropionaldehyde,7 6-methoxy-2-pyridinecarboxaldehyde,<sup>8</sup> 6-phenyl-2-pyridinecarboxaldehyde,<sup>9</sup>

#### Scheme 4<sup>a</sup>

<sup>a</sup> Conditions: (a) SEMCl, DIEA, CH<sub>2</sub>Cl<sub>2</sub>; (b) BzOH, HNa, DMF, 40 °C; (c) H<sub>2</sub>, Raney Ni, EtOH; (d) (CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>O, pyridine; (e) ArB(OH)<sub>2</sub>, Tl<sub>2</sub>CO<sub>3</sub>, Pd(PPh<sub>3</sub>)<sub>4</sub>; (f) nBu<sub>4</sub>NF, THF, HMPA; (g) MnO<sub>2</sub>, CH<sub>2</sub>Cl<sub>2</sub>, 40 °C.

### Scheme 5<sup>a</sup>

<sup>a</sup> Conditions: (a) MeONa, MeOH, 25 °C; (b) NH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>NH<sub>2</sub>· HCl, EtOH, 80 °C or NH<sub>2</sub>CH<sub>2</sub>CH<sub>2</sub>SH, 130 °C; (c) MeMgBr, THF; (d) (Me)<sub>2</sub>NCH(OMe)<sub>2</sub>, 110 °C; (e) NH<sub>2</sub>NH<sub>2</sub>·H<sub>2</sub>O, EtOH, 80 °C; (f) BaMnO<sub>4</sub> or NiO<sub>2</sub>, benzene, 80 °C; (g) HCO<sub>2</sub>H, H<sub>2</sub>O, 65 °C.

6-isopropyl-2-pyridinecarboxaldehyde, 10 3-(dimethylamino)benzaldehyde,11 and 3-(pyrrol-1-yl)benzaldehyde<sup>12</sup> used as intermediates in the synthesis of compounds 4, have been prepared as previously reported. The synthesis of novel aldehydes 8 was carried out as follows: the enamine formed by condensation of 6-fluoro-2-picoline and *tert*-butoxy(dimethylamino)methane<sup>13</sup> was oxidatively cleaved14 by NaIO4 to give the 6-fluoro-2pyridinecarboxaldehyde (8a) (Scheme 3). Protection of

#### Scheme 6a

OHC 
$$\frac{a}{7m}$$
  $O$   $\frac{a}{b}$   $\frac{7n}{R} = \frac{0}{0}$ 

 $^{\it a}$  Conditions: (a) TOSMIC,  $K_2CO_3,$  MeOH, 65 °C; (b) HCO<sub>2</sub>H,  $H_2O,$  65 °C.

## Scheme 7<sup>a</sup>

<sup>a</sup> Conditions: (a) Na, nBuOH, 100 °C; (b) NaSMe, DMF, 100 °C; (c) NaIO<sub>4</sub>, MeOH; (d) HCO<sub>2</sub>H, H<sub>2</sub>O, 65 °C.

the aldehyde **8a** as a dioxolane **7a** followed by displacement of the 6-fluoro substituent by the sodium salt of pyrazole gave **7b**. Hydrolysis of the acetal group under acidic conditions furnished the aldehyde **8b**.

The preparation of the 6-heteroaryl-2-pyridinecarbox-aldehydes 8c-e (Scheme 4) began by protection of the 6-chloro-2-pyridinemethanol<sup>15</sup> (9) as a (trimethylsi-lyl)ethoxymethyl ether 10. Replacement of the 6-chloro substituent by a hydroxyl group was performed by the method of Sieburth<sup>16</sup> (i.e., formation of the benzyl ether 11 and then hydrogenolysis of the benzyl group leading to the pyridol 12). Conversion of 12 into the triflate 13 and then Suzuki cross-coupling reaction with the boronic acid of the appropriate heteroaryl derivatives yielded compounds 14. Cleavage of the SEM group<sup>17</sup> and oxidation of the resulting primary alcohol 15, under mild conditions (MnO<sub>2</sub>/CH<sub>2</sub>Cl<sub>2</sub>), provided the 6-heteroaryl-2-pyridinecarboxaldehydes 8c-e.

The 6-(1,3-dioxolan-2-yl)-2-pyridinecarbonitrile (**7f**) (Scheme 5), derived from the known 6-cyano-2-pyridinecarboxaldehyde, <sup>18</sup> was the common starting material in the synthesis of aldehydes **8h,k,l**. At first, the nitrile function of **7f** was activated as an imidoyl ether (not shown in Scheme 5) which, by reaction with ethylenediamine or cysteamine, afforded the imidazoline **7i** and thiazoline **7j**, respectively. Oxidation of **7i**,**j**<sup>19,20</sup> and cleavage of the acetal group produced the aldehydes **8k,l**. Addition of CH<sub>3</sub>MgBr on the nitrile **7f** gave the methyl ketone **7g** which was converted to the pyrazole **7h** using known chemistry. <sup>21</sup> Deprotection of the carbonyl function of **7h** generated the 6-(pyrazol-3-yl)-2-pyridinecarboxaldehyde (**8h**).

Condensation of TOSMIC on the known aldehyde  $7m^{22}$  (Scheme 6) gave the derivative  $7n.^{23}$  Removal of the protecting group led to the expected aldehyde 8n. Finally, acetals 7o.p (Scheme 7) were prepared via a SNAr between the 6-fluoro derivative 7a and the sodium salt of the appropriate alcohol or thiol. Acetal cleavage furnished the aldehydes 8o.p. The aldehyde 8q was obtained by a chemoselective oxidation at the sulfur atom before the deprotection of the latent aldehyde function.

# **Pharmacology**

Binding affinities for the different receptors were determined by ligand displacement assays using the conditions summarized in the Experimental Section. All experiments were performed in triplicate. IC<sub>50</sub> values were determined using nonlinear regression.  $K_i$  values were calculated using the equation  $K_i = IC_{50}/1 + [C]/K_D$ , where [C] is the concentration of the radioligand. The results are expressed as mean p $K_i$  values  $\pm$  SEM of 2–3 independent determinations, each performed in triplicate.

The HeLa cell line permanently transfected with the human 5-HT<sub>1A</sub> receptor gene and permanently expressing the 5-HT<sub>1A</sub> receptor protein (HA7), as described by Fargin,<sup>24</sup> was commercially obtained from Duke University (Durham, NC). Concentration-effect relationships are expressed as -log [M] of the test compound versus the cAMP content expressed as a percentage of forskolin-stimulated cAMP control values. IC<sub>50</sub> values for compounds were estimated by linear interpolation between the logarithms of the concentrations that inhibited forskolin-stimulated cAMP with amounts bordering 50% of the maximal inhibition observed with the compound. The potency and maximal inhibition values represent the mean of 2 independent determinations (each performed in triplicate) for all compounds, except (±)-8-OH-DPAT whose values represent the mean of 5 determinations.

#### **Results and Discussion**

(a) **Design of the Pharmacophore.** At the outset of the project, we wondered about the role of the indole nitrogen lone pair of 5-HT in its ability to activate 5-HT<sub>1A</sub> receptors. However, because of the inaccuracy in homology modeling methods<sup>25</sup> and the lack of knowledge about the conformational landscape of ligandreceptor complexes,26 we took an approach that avoided speculations about the protein residues involved and about the nature of the interactions engaged in the complex. On the basis of the model of 5-HT<sub>1A</sub> receptor activation developed by Weinstein,<sup>27</sup> we reasoned that the intrinsic activity of a 5-HT<sub>1A</sub> ligand may be related to the ability of its pharmacophore to stabilize a positive charge generated on the receptor protein upon agonist binding. Hence, one way of increasing the intrinsic activity of a 5-HT<sub>1A</sub> agonist may be to enhance the ability of its pharmacophore to stabilize a positive charge.

According to Weinstein's model, the protein cation involved is a soft nitrogen cation derived from a basic amino acid (i.e., ammonium, imidazolium, or guanidinium ion). Since soft nitrogen cations are soluble in pyridine, which means that enthalpically favorable interactions occur within  $N^+$ -pyridine pairs,  $^{28}$  we suggested that a pharmacophore containing a pyridine core could efficiently stabilize the positive charge generated on the protein side. Thus, a pyridine appropriately located in the active site of the receptor could shift the equilibrium toward its G-protein-coupled state(s).

Having selected pyridine as the core component of the pharmacophore, our next concern was to find its optimal position relative to a stronger basic amino function that could serve as the prime anchoring point of the ligand inside the receptor protein.<sup>27,29</sup> At first, we looked at

**Table 1.** Effects of the Py-NH Linker Length and the Chain Orientation on 5-HT<sub>1A</sub> Affinity

	chain			5-HT <sub>1A</sub> affinity
compd	position	n	R	pKi <sup>a,b</sup>
16	2	1	Н	$7.58 \pm 0.15$
17	2	2	Н	$8.92 \pm 0.05$
18	2	3	Н	$7.37 \pm 0.03$
19	3	1	Н	$6.38 \pm 0.16^b$
20	3	2	Н	$6.03 \pm 0.01^b$
21	4	2	Н	<6.0
22	2	1	6-F	$8.05 \pm 0.10$
23	2	2	6-F	$8.50 \pm 0.05$
45	2	1	4-pyrazol-1-yl	$6.19 \pm 0.02^b$
58	2	1	6-pyrazol-1-yl	$8.78 \pm 0.09$
61	2	2	6-pyrazol-1-yl	$8.45 \pm 0.06$
24	2	1	4-Cl	$8.04 \pm 0.04$
25	2	1	6-Cl	$8.34 \pm 0.12$
33	2	1	4-(CH <sub>3</sub> ) <sub>2</sub> N	$5.58 \pm 0.01^b$
34	2	1	6-(CH <sub>3</sub> ) <sub>2</sub> N	$8.85 \pm 0.11$

<sup>&</sup>lt;sup>a</sup> See Experimental Section for details. The compound with a pK<sub>i</sub> value indicated with a smaller than sign (<) has no binding affinity at the given concentration. Each value is the mean  $\pm$  SEM of three determinations. <sup>b</sup> Two determinations only.

the distance between the pyridine and the secondary nitrogen atoms in (3,4-dichlorophenyl){[pyridinyl(*n*alkylamino)methyl]piperidin-1-yl}methanones. The choice of such side chains was motivated by the availability in our laboratory of the 1-(3,4-dichlorobenzoyl)piperidine-4-methanol precursor<sup>30</sup> and by its successful use with various other types of pharmacophores.<sup>30,31</sup> From displacement data shown in Table 1, it appears that unsubstituted 2-pyridinylmethyl, ethyl, and propylamino derivatives of (3,4-dichlorophenyl)(4-methylpiperidin-1-yl)methanone (16–18) have affinity for 5-HT<sub>1A</sub> receptors. When the pyridine is monosubstituted (R = H), a two-methylene linkage is the optimum distance between the pyridine ring and the secondary amino group (17 compared with 16, 18). However, in disubstituted pyridine, the nature of the R substituent influenced affinity only with ligands having a monomethylene linker between the pyridine ring and the amino group. Affinity varied over 1.2 log units for compounds of the 2-pyridinylmethylamine type (16, 22, 58) and only over 0.4 log unit for ethyl homologues (17, **23**, **61**). Assuming that for agonists, the elements of the pharmacophore involved in binding are also involved in the control of intrinsic activity, a positive relationship between the affinity and the intrinsic activity is expected for compounds whose pattern of interactions with the receptor is homogeneous. 32,33 Thus, a 6-substitutedpyridin-2-ylmethylamine seemed a more suited motif than a 6-substituted-pyridin-2-ylethylamine to examine structure—intrinsic activity relationships and was therefore used subsequently. Another interesting feature of a 6-substituted-pyridin-2-ylmethylamine pharmacophore is that the distance between the pyridine nucleus and the side-chain amino group is less than the distance required for recognition as predicted by steric models of 5-HT<sub>1A</sub> receptors.<sup>29</sup> The 2-pyridinylethyl and 2-pyridinylpropylamine homologues (17 and 18, respectively) of compound 16 can be viewed as open-chain analogues of aminotetralins or partial ergots.

Apparently, the range of substituents accommodated in the 6-position of the pyridine ring (25, 34, 58) is much wider than that in the 4-position (24, 33, 45), even though C-4 and C-6 may be regarded as isoelectronic. The 5-HT<sub>1A</sub> binding affinity of the benzene derivatives **72–74** was significantly lower than that of the pyridine counterparts 22, 34, and 43 (Table 2) and decreased

**Table 2.** Comparison of Effects of a Pyridyl versus a Phenyl Group on  $5\text{-}HT_{1A}$  Affinity

$$\mathbb{R} \xrightarrow{H} \mathbb{Q} \mathbb{Q}$$

compd	X	R	5-HT <sub>1A</sub> affinity pKi <sup>a,b</sup>
72	СН	F	$7.83 \pm 0.03$
22	N	F	$8.05 \pm 0.10$
73	СН	$(CH_3)_2N$	$7.12 \pm 0.14^b$
34	N	$(CH_3)_2N$	$8.85 \pm 0.11$
74	СН	pyrrol-1-yl	$6.53 \pm 0.01^b$
43	N	pyrrol-1-yl	$8.37 \pm 0.12$

<sup>&</sup>lt;sup>a</sup> See footnotes of Table 1.

markedly as the R substituent became larger (72-74). These results suggest that the pyridine ring contributes to the binding of the ligand to 5-HT<sub>1A</sub> receptors. Moreover, the nitrogen atom of the pyridine may be a major site of interactions of the pharmacophore with the receptor protein because the only tolerated orientation for the side chain was the 2-position of the pyridine (16 and 17 compared to 19 and 20, 21, respectively, Table 1).

(b) Influences of the Chain and of the Terminal **Amide Functional Group.** To explore the influence on binding of the nonpharmacophoric part of the molecule (i.e., the part other than the 6-R-pyridin-2-ylmethylamino fragment), we tested diverse representative chains attached to the 6-(pyrazol-1-yl)-2-pyridinylmethylamine moiety. Usually, known classes of 5-HT<sub>1A</sub> pharmacophores can accommodate various chains with only minor consequences for their in vitro biological activities. However, results summarized in Table 3 demonstrate the opposite for derivatives built around a 6-substituted-2-pyridinylmethylamine pharmacophore. Besides compound 58 (Table 1), only derivatives 78 and 82 (Table 3) achieved significant affinity for 5-HT<sub>1A</sub> receptors. In addition, the drop in binding observed with the open-chain acetylenic derivative 83 and the piperidine regioisomer 85 illustrates the tight geometric constraints imposed on that part of the molecule. Compared with the conformationally restricted piperidine (48, 53 in Table 4 and 58 in Table 1), the open-chain molecules **81–83** showed only weak affinity for the 5-HT<sub>1A</sub> receptors, indicating they cannot adopt suitable conformation(s) for optimal binding to the receptors. In line with these findings, changing the amide to a sulfonamide group or reducing the carbonyl oxygen atom of the amide function suppressed binding affinity (60 and 59, respectively, Table 4). Hence, the amide function represents a compromise between steric and conformational factors: the sulfone group being too sterically hindered, whereas the reduced amide might suffer a loss of entropic binding energy.

The comparison of binding data of compounds 46-**48** (Table 4) indicates that the lipophilicity of the carbonyl amide substituent is an important property of that group. On the other hand, the 10-fold gain in affinity observed with monosubstituted benzamides 50, **53**, and **54**, compared with the unsubstituted derivative **48**, suggested that an accessory hydrophobic effect occurred between the group in the meta position and the receptor protein. Moreover, examination of the benzamide monosubstitution pattern revealed that this interaction was subjected to strict directional and size requirements. Substitution in the meta position was the only one allowed (50 compared to 49, 51), and the size of the group should be between that of a fluorine (52) and that of an ethoxy group (55). Finally, it seemed that there was no electrostatic component contributing to the meta effect (56). Taking into account the deleterious consequences on binding of a para substituent alone (51), it was puzzling that some meta-para substituent combinations on the benzamide ring could act synergistically (58, Table 1, compared to 50). This synergistic effect was only observed when the substituent in the para position was not larger than a methyl group (58, 63, 64 compared to 65, 66). With the backbone of the ligand now roughly in place, we attempted to refine our understanding of the contribution of the substituent in the 6-position of the pyridine ring.

(c) Modulation of the Substituent in the 6-Position of the Pyridine. Data summarized in Table 5 showed that the R substituent modulated affinity values over more than 2 orders of magnitude. In general, electron-withdrawing groups bound poorly to the receptor (22, 31, 38, 39), whereas electron-releasing groups (n-donors: 28, 30, 32, 34, 36, 37) and particularly  $\pi$ -donors, such as five-membered heteroaromatic rings C-bound to pyridine (41, 42, 67, 70, 71), had high affinity for 5-HT<sub>1A</sub> sites.<sup>34</sup> Qualitatively, the effect of the R substituent is consistent with what would be expected if the pharmacophore was acting by stabilization of a positive charge on the protein. We assumed that derivative 27 may combine favorable inductive donor and steric effects. The size range of the R substituent was well-delineated, extending from an upper exclusion limit of four contiguous atoms (29, 35) to a lower one of two atoms, below which some favorable hydrophobic interactions may be lost (16, 25, 26).

Fortunately, the most active compounds were among the most selective toward  $\alpha_1$  and  $D_2$  receptors, the degree of selectivity being, in general, similar to that of (±)-8-OH-DPAT. Selectivity is important because interactions with other receptors, in particular  $\alpha_1$  and  $D_2$ , are known to influence the in vivo expression of 5-HT $_{1A}$  agonist activity.  $^{35,36}$ 

Derivatives shown in Table 5 displayed pEC $_{50}$  values to inhibition of forskolin-stimulated cAMP that differed by more than 2 log units, reminiscent of the trend seen with affinity values. Moreover, derivatives having nanomolar affinity for the receptor also inhibited the accumulation of cAMP with greater efficacy ( $E_{\rm max}$ ) than ( $\pm$ )-8-OH-DPAT (Table 5, footnote). Hence, common elements of the pharmacophore may participate in recognition and activation of the 5-HT $_{\rm 1A}$  receptor. Again, compounds containing a C-bound five-membered

Table 3. Effects of the Nature of the Chain on 5-HT<sub>1A</sub> Affinity

compd	R <sub>1</sub>	R <sub>2</sub>	5-HT <sub>1A</sub> affinity pKi <sup>ab</sup>	mp °C °
75	nC <sub>3</sub> H <sub>9</sub>	nC₃H₀	<5.0	90-95 <sup>d</sup>
76	Н	осн	$5.69 \pm 0.01^b$	170-172 <sup>e</sup>
77	Н	OCH <sub>3</sub>	$6.18 \pm 0.08^b$	123-125 <sup>e</sup>
78	Н	N N	$7.56 \pm 0.15^b$	195-197 <sup>d</sup>
79	Н		$6.96 \pm 0.03^b$	165-167 <sup>e</sup>
80	н		$6.82 \pm 0.03^b$	145-147 <sup>e</sup>
81	Н		$6.77 \pm 0.02^b$	179-181 <sup>d</sup>
82	Н		$7.85 \pm 0.05^b$	184-186 <sup>d</sup>
83	Н	, , , , , , , , , , , , , , , , , , ,	$6.61 \pm 0.02^b$	162-164 <sup>f</sup>
84	R <sub>2</sub>	N H CI	$5.55 \pm 0.03^b$	160-162 <sup>e</sup>
85	Н	, c1	$6.39 \pm 0.17^b$	191-193 <sup>e</sup>

 $<sup>^{</sup>a,b}$  See footnotes of Table 1.  $^c$  Experimental Section.  $^d$  Oxalate.  $^e$  Fumarate.  $^f$  Maleate.

heteroaromatic ring substituent emerged as especially potent (pEC<sub>50</sub>): **41**, **42**, **67**, **70**, and **71**. Interestingly, the two exceptions in this subset were the 6-pyrrol-2-yl and the 6-imidazol-2-yl derivatives 44 and 69, respectively. Their modest in vitro profiles compared with that of the close analogue 67 suggest that the hydrogenbonding ability of the group in the 2'-position of the fivemembered ring pyridine substituent may be one of the factors influencing the affinity and the potency of compounds belonging to this subset. This group was a hydrogen-bond donor in 44 and 69 but a hydrogen-bond acceptor in 67.37

Taken together, the in vitro results pointed to a C-bound five-membered heteroaromatic ring as the 6-substituent of choice for pyridine, although some 6-amino derivatives also deserve consideration (32, 34, **36**). However, it still remains unclear which properties of the 6-substituted-2-pyridinylmethylamine system (e.g., hydrogen-bonding ability, charge-transfer or chelating potentials, etc.) determine intrinsic activity.

The novel 5-HT<sub>1A</sub> ligands described here were found to have high intrinsic activity, and because of the hypothesized importance of intrinsic activity in the antidepressant effect of 5-HT<sub>1A</sub> agonists, future studies will examine their effects in vivo.

# **Conclusions**

In this paper, we summarized briefly the pharmacological rationale that prompted the search for novel, potent, and efficacious 5-HT<sub>1A</sub> agonists. We then described the basic concept that guided the design of a structurally novel 5-HT<sub>1A</sub> pharmacophore. Importantly, the approach adopted did not rely on assumptions concerning the geometry of the receptor protein. Finally, we reported the preparation of various novel compounds and discussed their preliminary in vitro pharmacological properties.

Taken together, the results show that this novel class of ligands is characterized by a high binding affinity for 5-HT<sub>1A</sub> receptors and a high selectivity versus  $\alpha_1$  and

**Table 4.** Effects of the Nature of the N-Piperidine Substituent on 5-HT<sub>1A</sub> Affinity

compd	A	R	5-HT <sub>1A</sub> affinity pKi <sup>a,b</sup>
59	CH <sub>2</sub>	3,4-dichlorophenyl	$7.75 \pm 0.03^b$
60	$SO_2$	3,4-dichlorophenyl	$6.70 \pm 0.09^b$
46	СО	cyclohexyl	$7.53 \pm 0.03^b$
47	СО	1-adamantyl	$7.88 \pm 0.06^b$
48	СО	phenyl	$7.48 \pm 0.02^b$
49	СО	2-chlorophenyl	$6.88 \pm 0.06^b$
50	СО	3-chlorophenyl	$8.64 \pm 0.11$
51	СО	4-chlorophenyl	$7.31 \pm 0.02^b$
52	СО	3-fluorophenyl	$7.87 \pm 0.02^b$
53	СО	3-methylphenyl	$8.46 \pm 0.04$
54	CO	3-trifluoromethylphenyl	$8.45 \pm 0.02$
55	CO	3-ethoxyphenyl	$7.97 \pm 0.02$
56	CO	3-cyanophenyl	$7.75 \pm 0.05^b$
57	CO	2,3-dichlorophenyl	$7.00 \pm 0.17^b$
62	CO	3,5-dichlorophenyl	$7.58 \pm 0.07^b$
63	CO	3-chloro, 4-fluorophenyl	$8.81 \pm 0.04$
64	CO	3-chloro, 4-methylphenyl	$8.85 \pm 0.04$
65	CO	3-chloro, 4-methoxyphenyl	$8.04 \pm 0.01$
66	CO	3-chloro, 4-methoxycarbonylphenyl	$6.78 \pm 0.08^b$

a,b See footnotes of Table 1.

D<sub>2</sub> receptors. Several members of this class behave in vitro as more potent inhibitors of forskolin-induced cAMP accumulation than the prototypical agonist  $(\pm)$ -8-OH-DPAT. From a structural standpoint, the 6-substituted-2-pyridinylmethylamine motif was used in an effort to obtain a novel 5-HT<sub>1A</sub> pharmacophore. The importance of the 6-substituent on the pyridine nucleus in the recognition and activation of 5-HT<sub>1A</sub> receptors was established. In addition, examples were given of the unexpected sensitivity of 5-HT<sub>1A</sub> binding to the structural integrity of the entire molecule. The very interesting in vitro profile of the most active derivatives warrants a further examination of their properties in biochemical and behavioral studies. Already, these derivatives and their analogues may provide innovative tools that could contribute to the further characterization of 5-HT<sub>1A</sub> receptors and, perhaps, to a better understanding of their functional roles and therapeutic significance.

## **Experimental Section**

Melting points were determined on a Büchi 530 melting point apparatus and were not corrected.  $^1H$  NMR spectra were recorded on a Bruker AC200 (200-MHz) instrument. Chemical shifts are reported in  $\delta$  value (ppm) relative to an internal standard of tetramethylsilane in CDCl $_3$  or DMSO- $d_6$ . Infrared (IR) spectra were obtained on a Nicolet FT 510 P spectrophotometer. Microanalyses were obtained on a Fison EA 1108/CHN analyzer, and the results obtained were  $\pm 0.4\%$  of the theoretical values. Analytical thin-layer chromatography was carried out on precoated plates (silica gel, 60 F 254 Merck). SDS silica gel (0.040–0.063 mm) was used for flash chromatography. Organic extracts were dried over MgSO $_4$  unless otherwise noted.

Method A. (3,4-Dichlorophenyl)(4-{[(6-fluoropyridin-2-ylmethyl)amino]methyl}piperidin-1-yl)methanone (22). A solution of 8a (0.58 g, 4.62 mmol), 4-(aminomethyl)piperidine (0.53 g, 4.62 mmol), and toluene (25 mL) was heated under reflux for 2 h with water separation by a Dean—Stark trap. The solvent was removed under vacuum, and the residue was taken up in THF (10 mL) and TEA (0.68 mL, 4.9 mmol). To

 $\textbf{Table 5.} \quad \text{Receptor Binding Profile and Agonist Activity of Compounds of the Type} \\ (3,4-\text{Dichlorophenyl}) \\ (4-\{[(6\text{-substituted-pyridin-2-ylmethyl})\text{amino}]\text{methyl}\} \\ \text{piperidin-1-yl})\text{methanone} \\ (3,4-\text{Dichlorophenyl}) \\ (4-\{(6\text{-substituted-pyridin-2-ylmethyl})\text{amino}]\text{methyl}\} \\ \text{piperidin-1-yl})\text{methanone} \\ (3,4-\text{Dichlorophenyl}) \\ (4-\{(6\text{-substituted-pyridin-2-ylmethyl})\text{amino}]\text{methyl}\} \\ \text{piperidin-1-yl})\text{methanone} \\ \text{methyl} \\ \text{piperidin-1-yl})\text{methanone} \\ \text{methyl} \\ \text{piperidin-1-yl})\text{methanone} \\ \text{methyl} \\ \text{piperidin-1-yl}) \\ \text{methyl} \\ \text{piperidin-1-yl})\text{methanone} \\ \text{methyl} \\ \text{piperidin-1-yl})\text{methanone} \\ \text{methyl} \\ \text{piperidin-1-yl})$ 

		гес	eptor binding (p	Ki) <sup>a,b</sup>	agonist activity <sup>c</sup>
compd	R	5-HT <sub>1A</sub>	$D_2$	$\alpha_1$	pEC <sub>50</sub>
16	Н	$7.58 \pm 0.15$	NT	NT	NT
22	F	$8.05 \pm 0.10$	<5.0	$5.85 \pm 0.05$	$6.43 \pm 0.08$
25	Cl	$8.34 \pm 0.12$	$5.20 \pm 0.09$	$6.24 \pm 0.04$	$6.75 \pm 0.06$
26	CH <sub>3</sub>	$8.39 \pm 0.05$	$5.32 \pm 0.05$	$6.32 \pm 0.02$	$7.41 \pm 0.05$
27	$(CH_3)_2CH$	$9.20 \pm 0.12$	$6.44 \pm 0.09$	$6.75 \pm 0.02$	$7.77 \pm 0.12$
28	CH <sub>3</sub> O	$8.46\pm0.02$	<5.0	$6.43 \pm 0.02$	$7.93 \pm 0.28$
29	$nC_4H_9O$	$7.73 \pm 0.02^b$	NT	NT	NT
30	CH₃S	$8.59 \pm 0.10$	$5.90 \pm 0.04$	$6.56 \pm 0.04$	$7.34 \pm 0.09$
31	CH₃SO	$7.33 \pm 0.02^b$	NT	NT	NT
32	CH₃NH	$8.76 \pm 0.06$	$5.51 \pm 0.05$	$6.87 \pm 0.04$	$7.66 \pm 0.12$
34	$(CH_3)_2N$	$8.85 \pm 0.11$	$6.28 \pm 0.04$	$6.92 \pm 0.03$	$7.69 \pm 0.14$
35	$nC_3H_7(CH_3)N$	$7.93 \pm 0.04^b$	NT	NT	NT
36	azetidino	$9.43 \pm 0.05$	$6.22 \pm 0.13$	$6.64 \pm 0.04$	$8.15 \pm 0.03$
37	pyrrolidino	$8.80 \pm 0.17$	$6.43 \pm 0.17$	$6.98 \pm 0.01$	$7.18 \pm 0.33$
38	N=-	$7.30 \pm 0.14^b$	NT	NT	NT
39	H <sub>2</sub> NCO	$7.33 \pm 0.09^b$	NT	NT	NT
43	pyrrol-1-yl	$8.37 \pm 0.12$	$6.30 \pm 0.06$	$6.93 \pm 0.02$	$7.45 \pm 0.12$
58	pyrazol-1-yl	$8.78 \pm 0.09$	$6.00 \pm 0.03$	$6.48 \pm 0.13$	$7.23 \pm 0.13$
68	imidazol-1-yl	$8.13 \pm 0.12$	$5.24 \pm 0.07$	$6.26 \pm 0.06$	$7.43 \pm 0.02$
44	pyrrol-2-yl	$8.22 \pm 0.03$	$6.61 \pm 0.03$	$7.14 \pm 0.02$	$7.15 \pm 0.17$
69	imidazol-2-yl	$8.57 \pm 0.04$	$5.43 \pm 0.04$	$6.45 \pm 0.03$	$7.44 \pm 0.18$
67	pyrazol-3-yl	$9.27 \pm 0.10$	$6.01 \pm 0.03$	$6.66 \pm 0.07$	$8.52 \pm 0.11$
42	thien-2-yl	$8.86 \pm 0.04$	$6.54 \pm 0.07$	$6.76 \pm 0.05$	$8.24 \pm 0.06$
71	thiazol-2-yl	$9.46 \pm 0.08$	$5.47\pm0.07$	$6.75 \pm 0.05$	$8.30 \pm 0.13$
41	furan-2-yl	$9.14 \pm 0.06$	$6.34 \pm 0.08$	$6.71 \pm 0.07$	$8.28 \pm 0.35$
70	oxazol-5-yl	$9.03 \pm 0.01$	$6.06 \pm 0.04$	$6.55 \pm 0.02$	$8.70 \pm 0.02$
40	phenyl	$8.13 \pm 0.12$	$5.53 \pm 0.08$	$6.68 \pm 0.03$	$6.78 \pm 0.24$
8-O	H-DPAT	$8.85 \pm 0.07$	$6.26 \pm 0.03$	$5.88 \pm 0.01$	$7.60 \pm 0.17^d$

a.b See footnotes of Table 1. c See Experimental Section for details. Each value is the mean  $\pm$  SEM of two independent determinations, each performed in triplicate. d Mean  $\pm$  SEM of five determinations. NT, not tested. Forskolin-stimulated cAMP levels in HA7 cells were inhibited by ( $\pm$ )-8-OH-DPAT ( $E_{max} = 71 \pm 4.3\%$ ) and to a greater extent ( $E_{max} > 80\%$ ) by all the other compounds (data not shown).

 $\textbf{Table 6.} \quad \textbf{Physical Data of Substituted-4-[(pyridinylalkylamino)methyl] piperidines}$ 

$$R = \left( \frac{1}{N} \right)^{\frac{1}{N}} \times \left( \frac{1}{N} \right$$

Compd	х	R	Chain position	n	$R_1$	Method	% yield	Formula <sup>a</sup>	mp °C
16	N	Н	2	1	3,4-dichlorobenzoyl	Α	57	$C_{19}H_{21}Cl_2N_3O.C_4H_4O_4{}^b$	144-146
17	N	Н	2	2	3,4-dichlorobenzoyl	C	54	$C_{20}H_{23}Cl_{2}N_{3}O.C_{2}H_{2}O_{4}{}^{\circ}$	173-175
18	N	Н	2	3	3,4-dichlorobenzoyl	3,4-dichlorobenzoyl A		$C_{21}H_{25}Cl_2N_3O.C_4H_4O_4^d$	140-142
19	N	Н	3	1	3,4-dichlorobenzoyl	Α	42	$C_{19}H_{21}Cl_2N_3O.1.5C_2H_2O_4{}^c$	159-161
20	N	Н	3	2	3,4-dichlorobenzoyl	C	47	$C_{20}H_{23}Cl_2N_3O.1.5C_4H_4O_4{}^d$	141-143
21	N	Н	4	2	3,4-dichlorobenzoyl	C	58	$C_{20}H_{23}Cl_2N_3O\cdot 1.5C_4H_4O_4{}^d$	149-151
22	N	6-F	2	1	3,4-dichlorobenzoyl	Α	40	$C_{19}H_{20}Cl_2FN_3O.C_2H_2O_4{}^c$	182-184
23	N	6-F	2	2	3,4-dichlorobenzoyl	C	52	$C_{20}H_{22}Cl_2FN_3O.C_4H_4O_4^b$	155-157
24	N	4-C1	2	1	3,4-dichlorobenzoyl	Α	51	$C_{19}H_{20}Cl_3N_3O.C_4H_4O_4^{\ b}$	150-152
25	N	6-Cl	2	1	3,4-dichlorobenzoyl	Α	37	$C_{19}H_{20}Cl_3N_3O.C_4H_4O_4^{\ d}$	174-176
26	N	6-CH <sub>3</sub>	2	1	3,4-dichlorobenzoyl	Α	56	$C_{20}H_{23}Cl_2N_3O.C_4H_4O_4{}^d$	156-158
27	N	6-CH(CH <sub>3</sub> ) <sub>2</sub>	2	1	3,4-dichlorobenzoyl	Α	49	$C_{22}H_{27}Cl_2N_3O.C_4H_4O_4{}^d$	148-150
28	N	6-OCH <sub>3</sub>	2	1	3,4-dichlorobenzoyl	Α	34	$C_{20}H_{23}Cl_2N_3O_2.C_4H_4O_4{}^b$	150-152
29	N	6-OC₄H₀n	2	1	3,4-dichlorobenzoyl	Α	41	$C_{23}H_{29}Cl_2N_3O_2.C_4H_4O_4{}^d$	160-162
30	N	6-SCH <sub>3</sub>	2	1	3,4-dichlorobenzoyl	Α	35	$C_{20}H_{23}Cl_2N_3OS.C_2H_2O_4^{\ c}$	142-144
31	N	6-S(O)CH <sub>3</sub>	2	1	3,4-dichlorobenzoyl	Α	30	$C_{20}H_{23}Cl_{2}N_{3}O_{2}.S.C_{2}H_{2}O_{4}{}^{c}$	114-116
32	N	6-NHCH <sub>3</sub>	2	1	3,4-dichlorobenzoyl	$B_1$	48	$C_{20}H_{24}Cl_2N_4O.C_4H_4O_4{}^b$	117-119
33	N	4-N(CH <sub>3</sub> ) <sub>2</sub>	2	1	3,4-dichlorobenzoyl	$\mathbf{B}_1$	53	$C_{21}H_{26}Cl_2N_4O.C_2H_2O_4{}^c$	169-171
34	N	6-N(CH <sub>3</sub> ) <sub>2</sub>	2	1	3,4-dichlorobenzoyl	$B_1$	61	$C_{21}H_{26}Cl_2N_4O.C_4H_4O_4{}^d$	175-177
35	N	6-N(CH <sub>3</sub> )C <sub>3</sub> H <sub>7</sub> n	2	1	3,4-dichlorobenzoyl	$B_1$	36	$C_{23}H_{30}Cl_2N_4O.C_2H_2O_4^{\ c}$	186-188
36	N	6-azetidino	2	1	3,4-dichlorobenzoyl	$B_1$	52	$C_{22}H_{26}Cl_2N_4O.C_2H_2O_4^{\ c}$	219-221
37	N	6-pyrrolidino	2	1	3,4-dichlorobenzoyl	$\mathbf{B}_1$	54	$C_{23}H_{28}Cl_2N_4O.C_2H_2O_4{}^c$	203-205
38	N	6-cyano	2	1	3,4-dichlorobenzoyl	Α	48	$C_{20}H_{20}Cl_2N_4O.C_4H_4O_4^b$	144-146
39	N	6-carboxamido	2	1	3,4-dichlorobenzoyl	$\mathbf{B}_1$	27	$C_{20}H_{22}Cl_2N_4O_2.C_4H_4O_4{}^d$	211-213
40	N	6-phenyl	2	1	3,4-dichlorobenzoyl	Α	52	$C_{25}H_{25}Cl_2N_3O.C_4H_4O_4{}^d$	128-130
41	N	6-furan-2-yl	2	1	3,4-dichlorobenzoyl	Α	47	$C_{23}H_{23}Cl_2N_3O_2.0.5C_4H_4O_4{}^d$	178-180
42	N	6-thien-2-yl	2	1	3,4-dichlorobenzoyl	Α	53	$C_{23}H_{23}Cl_2N_3OS.C_2H_2O_4{}^c$	208-210
43	N	6-pyrrol-1-yl	2	1	3,4-dichlorobenzoyl	$B_2$	57	$C_{23}H_{24}Cl_2N_4O.0.5C_4H_4O_4{}^d$	174-176
44	N	6-pyrrol-2-yl	2	1	3,4-dichlorobenzoyl	Α	41	$C_{23}H_{24}Cl_{2}N_{4}O.C_{2}H_{2}O_{4}{}^{c}\\$	200-210
45	N	4-pyrazol-1-yl	2	1	3,4-dichlorobenzoyl	$\mathbf{B}_2$	54	$C_{22}H_{23}Cl_2N_5O.C_4H_4O_4^{\ b}$	118-120
46	N	6-pyrazol-1-yl	2	1	cyclohexylcarbonyl	Α	37	$C_{22}H_{31}N_5O.C_2H_2{O_4}^c\\$	197-199
47	N	6-pyrazol-1-yl	2	1	1-adamantylcarbonyl	Α	43	$C_{26}H_{35}N_5O.C_2H_2{O_4}^c\\$	224-226
48	N	6-pyrazol-1-yl	2	1	benzoyl	Α	51	C <sub>22</sub> H <sub>25</sub> N <sub>5</sub> O.1.5C <sub>4</sub> H <sub>4</sub> O <sub>4</sub> <sup>d</sup>	161-163

Table 6 (Continued)

Compd	X	R	Chain position	n	$R_1$	Method	% yield	Formula <sup>a</sup>	mp °C
49	N	6-pyrazol-1-yl	2	1	2-chlorobenzoyl	A	53	C <sub>22</sub> H <sub>24</sub> ClN <sub>5</sub> O.C <sub>2</sub> H <sub>2</sub> O <sub>4</sub> <sup>c</sup>	212-214
50	N	6-pyrazol-1-yl	2	1	3-chlorobenzoyl	Α	55	$C_{22}H_{24}ClN_5O.C_2H_2O_4{}^c$	174-176
51	N	6-pyrazol-1-yl	2	1	4-chlorobenzoyl	Α	49	C <sub>22</sub> H <sub>24</sub> ClN <sub>5</sub> O.1.5C <sub>4</sub> H <sub>4</sub> O <sub>4</sub> <sup>d</sup>	177-179
52	N	6-pyrazol-1-yl	2	1	3-fluorobenzoyl	Α	51	C <sub>22</sub> H <sub>24</sub> FN <sub>5</sub> O.1.5C <sub>4</sub> H <sub>4</sub> O <sub>4</sub> <sup>d</sup>	183-185
53	N	6-pyrazol-1-yl	2	1	3-methylbenzoyl	Α	54	$C_{23}H_{27}N_5O.C_4H_4O_4{}^d$	188-190
54	N	6-pyrazol-1-yl	2	1	3-trifluoromethylbenzoyl	Α	49	$C_{23}H_{24}F_3N_5O.C_4H_4O_4{}^d$	160-162
55	N	6-pyrazol-1-yl	2	1	3-ethoxybenzoyl	Α	50	$C_{24}H_{29}N_5O_2.C_4H_4O_4{}^d$	143-145
56	N	6-pyrazol-1-yl	2	1	3-cyanobenzoyl	Α	39	$C_{23}H_{24}N_6O.C_4H_4O_4{}^d$	209-211
57	N	6-pyrazol-1-yl	2	1	2,3-dichlorobenzoyl	Α	47	$C_{22}H_{23}Cl_2N_5O.C_4H_4O_4{}^d$	190-192
58	N	6-pyrazol-1-yl	2	1	3,4-dichlorobenzoyl	Α	53	$C_{22}H_{23}Cl_2N_5O.C_4H_4O_4{}^b$	152-154
59	N	6-pyrazol-1-yl	2	1	3,4-dichlorobenzyl	D	44	$C_{22}H_{25}Cl_2N_5.2C_4H_4O_4{}^d$	201-203
60	N	6-pyrazol-1-yl	2	1	3,4-dichlorophenylsulfonyl	E	35	$C_{21}H_{23}Cl_2N_5O_2S.C_4H_4O_4{}^d$	199-201
61	N	6-pyrazol-1-yl	2	2	3,4-dichlorobenzoyl	$B_2$	49	$C_{23}H_{25}Cl_2N_5O.C_4H_4O_4{}^b$	148-150
62	N	6-pyrazol-1-yl	2	1	3,5-dichlorobenzoyl	Α	48	$C_{22}H_{23}Cl_2N_5O.C_4H_4O_4{}^d$	191-193
63	N	6-pyrazol-1-yl	2	1	3-chloro, 4-fluorobenzoyl	Α	52	$C_{22}H_{23}Cl_1FN_5O.C_4H_4O_4{}^d$	181-183
64	N	6-pyrazol-1-yl	2	1	3-chloro, 4-methylbenzoyl	Α	44	$C_{23}H_{26}Cl_1N_5O.C_4H_4O_4{}^d$	206-208
65	N	6-pyrazol-1-yl	2	1	3-chloro, 4-methoxybenzoyl	Α	42	$C_{23}H_{26}ClN_5O_2.C_4H_4O_4{}^d$	173-175
66	N	6-pyrazol-1-yl	2	1	3-chloro, 4-methoxycarbonylbenzoyl	Α	31	$C_{24}H_{26}CIN_5O_3.C_2H_2O_4{}^d$	176-178
67	N	6-pyrazol-3-yl	2	1	3,4-dichlorobenzoyl	Α	54	$C_{22}H_{23}Cl_2N_5O.C_2H_2O_4^{d}$	145-150
68	N	6-imidazol-1-yl	2	1	3,4-dichlorobenzoyl	$\mathbf{B}_2$	53	$C_{22}H_{23}Cl_2N_5O.C_4H_4O_4{}^d$	125-130
69	N	6-imidazol-2-yl	2	1	3,4-dichlorobenzoyl	A	47	$C_{22}H_{23}Cl_2N_5O.C_4H_4O_4{}^d$	206-208
70	N	6-oxazol-5-yl	2	1	3,4-dichlorobenzoyl	Α	38	$C_{22}H_{22}Cl_2N_4O_2.C_2H_2O_4{}^c$	189-191
71	N	6-thiazol-2-yl	2	1	3,4-dichlorobenzoyl	Α	52	C <sub>22</sub> H <sub>22</sub> Cl <sub>2</sub> N <sub>4</sub> OS.C <sub>4</sub> H <sub>4</sub> O <sub>4</sub> <sup>d</sup>	187-189
72	СН	F	3	1	3,4-dichlorobenzoyl	Α	57	$C_{20}H_{21}Cl_{2}FN_{2}O.C_{2}H_{2}O_{4}{}^{c}$	199-201
73	СН	$N (CH_3)_2$	3	1	3,4-dichlorobenzoyl	Α	35	$C_{22}H_{27}Cl_2N_3O.C_2H_2O_4{}^c$	237-239
74	СН	pyrrol-1-yl	3	1	3,4-dichlorobenzoyl	Α	48	C <sub>24</sub> H <sub>24</sub> Cl <sub>2</sub> N <sub>3</sub> O.C <sub>4</sub> H <sub>4</sub> O <sub>4</sub> <sup>b</sup>	139-141

 $<sup>^</sup>a$  The analyses for all compounds were within 0.4% of the theoretical value for C, H, and N.  $^b$  Maleate.  $^c$  Oxalate.  $^d$  Fumarate.

Table 7. Experimental Details for Each of the Binding Assays

	[ <sup>3</sup> H]ligand		tissue (rat)		incubation		nonspecific binding			
binding site	K <sub>D</sub> (nM)	concn (nM)	type	concn (mg/mL)	time (min)	temp (°C)	drug	concn (µM)	buffer	
5-HT <sub>1A</sub> <sup>b</sup>	8-OH-DPAT (3.1)	0.2	cortex	10	30	23	5-HT	10	A	
$\mathrm{D}_2{}^c$	YM-09151-2 (0.036)	0.05	striatum	1	60	23	(+)-butaclamol	1	В	
$\alpha_1{}^b$	prazosin (0.063)	0.1	cortex	5	30	23	phentolamine	50	С	

 $<sup>^</sup>a$  Buffers: (A) Tris HCl (50 mM, pH 7.4), pargyline (10  $\mu$ M, CaCl $_2$  (4 mM), ascorbic acid (0.1%); (B) Tris HCl (50 mM, pH 7.4), NaCl (120 mM), KCl (5 mM); (C) Tris HCl (50 mM, pH 7.4).  $^b$  Reference 40.  $^c$  Reference 41.

this cooled solution (ice bath) was added 3,4-dichlorobenzoyl chloride (0.68 g, 4.62 mmol) in THF (2 mL), and stirring was continued at room temperature for 2 h. Methanol (10 mL) was added followed by  $KB\dot{H}_4$  (0.50 g, 9.27 mmol), and stirring was continued at room temperature for 4 h. The mixture was concentrated, taken up in CH2Cl2, washed with water and brine, dried, and concentrated. Purification by flash chromatography (5% MeOH-CH $_2\text{Cl}_2)$  gave 0.73 g (40%) of  $\boldsymbol{22}$  as a pale-yellow solid: mp 77-79 °C. The oxalate salt of **22** was crystallized from MeOH-EtOAc to afford a white solid (0.65 g): mp 182–184 °C; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  1.19 (m, 2H), 1.90 (m, 3H), 2.85 (m, 3H), 3.04 (m,1H), 3.47 (s, 1H), 4.24 (s, 2H), 4.39 (m, 1H), 7.20 (dd, 1H), 7.36 (dd, 1H), 7.48 (dd, 1H), 7.70 (m, 2H), 8.05 (dd, 1H), 8.15 (s, broad, 2H). Anal. (C<sub>19</sub>H<sub>20</sub>C<sub>12</sub>- $FN_3O \cdot C_2H_2O_4)$  C, H, N.

Method B1. (3,4-Dichlorophenyl)(4-{[(6-azetidin-1ylpyridin-2-ylmethyl)amino]methyl}piperidin-1-yl)methanone (36). A solution of 22 (0.70 g, 1.77 mmol), THF (10 mL), and azetidine (0.20 g, 3.54 mmol) was heated at 100 °C for 17 h. After cooling to room temperature, the mixture was concentrated under vacuum, and water was added. The product was extracted with CH2Cl2, washed with water and brine, dried, filtered, and evaporated under reduced pressure. The residue was purified by flash chromatography (10% MeOH-CH<sub>2</sub>Cl<sub>2</sub>) to give 0.40 g (52%) of 36 as a pale-yellow oil. The oxalate salt of 36 was crystallized from EtOH-EtOAc to give a white solid (0.35 g): mp 219–221 °C;  $^1\text{H}$  NMR  $(DMSO-d_6) \delta 1.19 (m, 2H), 1.80 (m, 2H) 2.00 (m, 1H), 3.35 (m, 2H) 2.00 (m, 2H), 3.35 (m, 2H) 2.00 (m, 2H), 3.35 (m, 2H)$ 2H), 2.75 (m, 1H), 2.88 (d, 2H), 3.04 (m, 1H), 3.49 (m, 1H), 3.95 (t, 4H), 4.09 (s, 2H), 4.39 (m, 1H), 6.31 (d, 1H), 6.68 (d, 1H), 7.35 (dd, 1H), 7.53 (t, 1H), 7.65 (d, 1H), 7.70 (d, 1H). Anal.  $(C_{22}H_{26}Cl_2N_4O\cdot C_2H_2O_4)$  C, H, N.

Method B2. (3,4-Dichlorophenyl)(4-{[(6-pyrrol-1-ylpyridin-2-ylmethyl)amino|methyl}piperidin-1-yl)meth**anone (43).** To a solution of pyrrole (0.45 g, 6.70 mmol) in anhydrous DMF (5 mL), under a nitrogen atmosphere, was added NaH (60% oil dispersion, 0.27 g, 6.75 mmol). After the mixture stirred at room temperature for 1 h, a solution of 22 (1.32 g, 3.33 mmol) in DMF (2 mL) was added, and the mixture was stirred at 60 °C for 2 h. The cooled solution was poured into ice-water and extracted with EtOAc. The organic layer was washed with water and brine, dried, and filtered, and the solvent was removed under vacuum. Purification by flash chromatography (5% MeOH-CH<sub>2</sub>Cl<sub>2</sub>) gave an oily residue which on trituration with Et2O gave 43 as a pale-yellow solid (0.85 g; 57%): mp 108-110 °C. The hemifumarate salt of **43** was crystallized from EtOH-EtOAc to give a white solid: mp 174–176 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.12 (m, 2H), 1.81 (m, 3H), 2.60 (m, 2H), 2.94 (s, 1H), 3.60 (m, 2H), 3.95 (s, 2H), 4.59 (s, 1H), 6.25 (t, 2H), 6.65 (s, 1H), 7.01 (d, 1H), 7.14 (m, 2H), 7.39 (m, 4H), 7.64 (t, 1H). Anal. (C<sub>23</sub>H<sub>24</sub>Cl<sub>2</sub>N<sub>4</sub>O·0.5C<sub>4</sub>H<sub>4</sub>O<sub>4</sub>) C, H,

Method C. (3,4-Dichlorophenyl){4-[(2-pyridin-4-ylethylamino)methyl]piperidin-1-yl}methanone (21). A solution of **5** (0.80 g, 2.80 mmol), 4-(2-aminoethyl)pyridine (0.34 g, 2.80 mmol), and toluene (30 mL) was heated under reflux for 2 h with water separation by a Dean-Stark trap. The solvent was removed under vacuum and then the residue taken up in MeOH (30 mL) and cooled to 0-5 °C, and KBH<sub>4</sub> (0.30 g, 5.6 mmol) was added portionwise. Stirring was continued at room temperature for 18 h. The mixture was concentrated, extracted with CH<sub>2</sub>Cl<sub>2</sub>, washed with water and brine, dried, filtered, and concentrated under vacuum to give a yellow oil. Purification by flash chromatography (10% MeOH-CH<sub>2</sub>Cl<sub>2</sub>) afforded 0.64 g (58%) of 21 as a pale-yellow oil. The fumarate salt of 21 was crystallized from EtOH-EtOAc to give a white solid (0.55 g): mp 149-151 °C; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  1.34 (m, 2H), 1.75 (m, 1H), 1.95 (m, 1H), 2.14 (m, 1H), 2.90-3.25 (m, 4H), 2.33-3.54 (m, 4H), 3.74 (m, 1H), 4.53 (m, 1H), 6.60 (s, 3H), 7.32 (dd, 1H), 7.60 (d, 1H), 7.64 (d, 1H), 7.95 (d, 2H), 8.71 (d, 2H). Anal. (C<sub>20</sub>H<sub>23</sub>Cl<sub>2</sub>N<sub>3</sub>O· 1.5C<sub>4</sub>H<sub>4</sub>O<sub>4</sub>) C, H, N.

Method D. [1-(3,4-Dichlorobenzyl)piperidin-4-ylmethyl](6-pyrazol-1-ylpyridin-2-ylmethyl)amine (59). A solution of the amide 58 (0.70 g, 1.57 mmol) in THF (5 mL) was added over 15 min to a suspension of LiAlH<sub>4</sub> (90 mg, 2.37 mmol) in THF (10 mL) maintained at 0 °C. Stirring was continued for 18 h at room temperature; then the reaction was quenched by addition of  $H_2O$  and 1 N NaOH. The mixture was filtered through Celite and the solid washed with THF. The filtrate was evaporated to dryness and then the residue taken up in CH<sub>2</sub>Cl<sub>2</sub>, dried, and concentrated. The product was purified by flash chromatography (10% MeOH-CH<sub>2</sub>Cl<sub>2</sub>) to give 0.30 g (44%) of **59** as a pale-yellow oil. The fumarate salt was crystallized from EtOH to afford a white solid (0.33 g): mp 201–203 °C; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  1.23 (m, 2H), 1.77 (m, 3H), 1.99 (t, 2H), 2.76 (m, 4H), 3.50 (s, 2H), 4.22 (s, 2H), 6.52 (s, 2H), 6.57 (t, 1H), 7.27 (dd, 1H), 7.38 (d, 1H), 7.53 (m, 2H), 7.83 (d, 2H), 7.99 (t, 1H), 8.82 (d, 1H). Anal. (C<sub>22</sub>H<sub>25</sub>Cl<sub>2</sub>N<sub>5</sub>· 2C<sub>4</sub>H<sub>4</sub>O<sub>4</sub>) C, H, N.

Method E. [1-(3,4-Dichlorophenylsulfonyl)piperidin-4-ylmethyl](6-pyrazol-1-ylpyridin-2-ylmethyl)amine (60). This product was prepared from aldehyde **8b**, 4-(aminomethyl)piperidine, and 3,4-dichlorobenzenesulfonyl chloride as described for compound 22. Purification by flash chromatography (5% MeOH-CH<sub>2</sub>Cl<sub>2</sub>) followed by salt formation with fumaric acid gave the fumarate salt of 60 as a white solid (35%): mp 199–201 °C; <sup>1</sup>H NMR (DMSO- $d_6$ )  $\delta$  1.18 (m, 2H), 1.57 (m, 1H), 1.80 (d, 2H), 2.32 (t, 2H), 2.55 (d, 2H), 3.63 (d, 2H), 3.96 (s, 2H), 6.55 (s, 2H), 6.57 (d, 1H), 7.35 (d, 1H), 7.69 (dd, 1H), 7.77 (m, 2H), 7.94 (m, 3H), 8.70 (d, 1H). Anal.  $(C_{21}H_{23}Cl_2N_5O_2S\cdot C_4H_4O_4)$  C, H, N.

1-(3,4-Dichlorobenzoyl)piperidine-4-carboxaldehyde (5). To a solution of oxalyl chloride (2.72 mL, 31 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (50 mL), cooled to -70 °C, was added DMSO (4.1 mL, 58 mmol) dropwise. Stirring was continued at -60 °C for 10 min followed by addition of a solution of (3,4-dichlorophenyl)[4-(hydroxymethyl)piperidin-1-yl]methanone<sup>30</sup> (7.50 g, 26 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (20 mL). The reaction mixture was stirred for 1 h at -60 °C; then DIEA (18 mL, 104 mmol) was added dropwise and the temperature allowed to reach to room temperature. Water was added, and the organic layer was washed with 1 N HCl, water, and brine, dried, filtered, and concentrated. Purification by flash chromatography (30% EtOAc-CH<sub>2</sub>Cl<sub>2</sub>) gave 5.65 g (76%) of **5** as a white solid: mp 101-103 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.65 (m, 2H), 1.95 (m, 2H), 2.54 (m, 1H), 3.13 (m, 2H), 3.68 (m, 1H), 4.35 (m, 1H), 7.18 (dd, 1H), 7.46 (m, 2H), 9.66 (s, 1H); IR (KBr, cm<sup>-1</sup>) 1640, 1732. Anal. (C<sub>13</sub>H<sub>13</sub>-Cl<sub>2</sub>NO<sub>2</sub>) C, H, N.

6-Fluoro-2-pyridinecarboxaldehyde (8a). A solution of 6-fluoro-2-picoline (20 g, 0.18 mol) and tert-butoxybis(dimethylamino) methane (47 g, 0.27 mol) was heated, under a nitrogen atmosphere, at 140 °C for 24 h. The cooled reaction mixture was diluted with THF (100 mL), and a solution of NaIO<sub>4</sub> (77 g, 0.36 mol) in water (400 mL) was added. The mixture was stirred overnight at room temperature; then the precipitate was filtered off. THF was removed under vacuum, and the product was extracted with CH<sub>2</sub>Cl<sub>2</sub>. The organic layer was washed with brine, dried, and filtered. The solution was carefully concentrated (the boiling point of the product is low), and the residual brown oil was submitted to distillation to give 7.40 g (33%) of **8a** as a pale-yellow oil: bp  $65-70 \, ^{\circ}\text{C}$  (80 mb); <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 7.16 (dd, 1H), 7.82 (dd, 1H), 7.96 (m, 1H), 9.90 (s, 1H); IR (film, cm<sup>-1</sup>) 1713.

6-Fluoro-2-(1,3-dioxolan-2-yl)pyridine (7a). A solution of 8a (7.40 g, 59 mmol), ethanediol (11.2 mL, 180 mmol), and p-toluenesulfonic acid monohydrate (0.25 g) in benzene (80 mL) was heated under reflux for 18 h with water separation by a Dean-Stark trap. The solvent was removed under vacuum and the residue taken up in  $\text{Et}_2\text{O}$  and washed with 10%NaHCO<sub>3</sub> solution and then brine. The organic layer was dried and filtered, and the solvent was removed under reduced pressure. The oily residue was distilled to give 8 g (80%) of **7a** as a pale-yellow oil: bp 75–80 °C (0.1 mb); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  4.05 (m, 2H), 4.16 (m, 2H), 5.77 (s, 1H), 6.90 (dd, 1H), 7.39 (dd, 1H), 7.79 (dd, 1H).

6-Pyrazolo-2-(1,3-dioxolan-2-yl)pyridine (7b). To a suspension of HNa (60% dispersion in mineral oil, 7 g, 0.174 mol) in anhydrous DMF (50 mL) under a nitrogen atmosphere was added dropwise a solution of pyrazole (11.85 g, 0.174 mol) in DMF (55 mL). After 2 h, 7a (9.81 g, 0.058 mol) was added dropwise, and the mixture was stirred at 80 °C for 3 h. The reaction mixture was poured in ice-water and extracted with EtOAc. The combined extracts were washed with water and brine, dried, and filtered, and the solvent was removed under vacuum. The product was purified by flash chromatography (20% EtOAc-hexane) to give 9.50 g (75%) of 7b as a paleyellow oil: <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 4.10 (m, 2H), 4.22 (m, 2H), 5.87 (s, 1H), 6.45 (dd, 1H), 7.43 (dd, 1H), 7.73 (d, 1H), 7.86 (t, 1H), 7.98 (dd, 1H), 8.62 (d, 1H).

6-Pyrazol-1-ylpyridine-2-carboxaldehyde (8b). A mixture of **7b** (9.50 g, 43.7 mmol), CuSO<sub>4</sub>·5H<sub>2</sub>O (1.09 g, 4.37 mmol), formic acid (80 mL), and water (20 mL) was stirred at 65 °C for 5 h. The solution was concentrated under vacuum, and the residual formic acid was eliminated by azeotropric distillation with toluene. The oil was taken up in ice-water and made basic with excess K<sub>2</sub>CO<sub>3</sub>. The product was extracted with EtOAc, washed with diluted NH<sub>4</sub>OH and brine, dried, and concentrated under vacuum. The product was isolated by flash chromatography (10% EtOAc-CH2Cl2) to give 6 g (79%) of **8b** as a pale-yellow solid: mp 75-76 °C; <sup>1</sup>H NMR  $(CDCl_3)$   $\delta$  6.52  $(d\bar{d}, 1H)$ , 7.78 (s, 1H), 7.84 (dd, 1H), 7.99 (t, 1H), 8.23 (dd, 1H), 8.67 (d, 1H), 10.04 (s, 1H); IR (KBr, cm<sup>-1</sup>) 1701. Anal. (C<sub>9</sub>H<sub>7</sub>N<sub>3</sub>O) C, H, N.

2-Chloro-6-[2-(trimethylsilanyl)ethoxymethoxymethyl]pyridine (10). To a solution of 6-chloro-2-pyridinemethanol<sup>15</sup> (2.50 g, 17.4 mmol), DIEA (3.30 mL, 19.1 mmol). and CH<sub>2</sub>Cl<sub>2</sub> (20 mL), under nitrogen atmosphere and cooled to 0 °C, was added (trimethylsilyl)ethoxychloromethyl ether (3.20 mL, 18.2 mmol). Stirring was continued at room temperature for 3 h. The solvent was evaporated and the oil extracted with Et<sub>2</sub>O, washed with water, dried, filtered, and evaporated to dryness. The product was purified by flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>) to give 3.65 g (76.6%) of **10** as a colorless oil: <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.00 (s, 9H), 0.95 (t, 2H), 3.68 (t, 2H), 4.67 (s, 2H), 4.80 (s, 2H), 7.24 (d, 1H), 7.35 (d, 1H), 7.64 (t, 1H).

2-(Benzyloxy)-6-[2-(trimethylsilanyl)ethoxymethoxy**methyl]pyridine (11).** To a suspension of HNa (60% dispersion in mineral oil, 0.91 g, 22.6 mmol) in DMF (20 mL) at 0 °C and under a nitrogen atmosphere was added dropwise a solution of benzyl alcohol (2.10 mL, 19.9 mmol) in DMF (3 mL). Stirring was continued for 1.5 h at 0 °C, and a solution of 10 (3.65 g, 13.3 mmol) in DMF (3 mL) was added dropwise. The reaction mixture was heated at 40 °C for 12 h, cooled to room temperature, poured in ice-water, and extracted with Et<sub>2</sub>O. The combined extracts were washed with water, dried, and filtered, and the solvent was evaporated to dryness. The crude product was purified by flash chromatography (40% hexane-CH<sub>2</sub>Cl<sub>2</sub>) to give 3 g (65%) of **11** as a colorless oil: <sup>1</sup>H NMR  $(CDCl_3)$   $\delta$  0.00 (s, 9H), 0.94 (t, 2H), 3.67 (t, 2H), 4.61 (s, 2H), 4.80 (s, 2H), 5.35 (s, 2H), 6.66 (d, 1H), 6.97 (d, 1H), 7.34 (m, 3H), 7,42 (m, 2H), 7.54 (t, 1H).

2-Hydroxy-6-[2-(trimethylsilanyl)ethoxymethoxymethyl]pyridine (12). To a solution of 11 (7 g, 20.26 mmol) in EtOH saturated with hydrogen chloride (75 mL) was added Raney nickel (7 g). The suspension was stirred at room temperature under hydrogen atmosphere for 1.5 h. The catalyst was filtered off through Celite and then the solvent removed under vacuum. The product was isolated by flash chromatography to give 4 g (76.8%) of 12 as a pale-yellow oil: <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.00 (s, 9H), 0.93 (t, 2H), 3.61 (t, 2H), 4.50 (s, 1H), 4.75 (s, 2H), 6.21 (d, 1H), 6.47 (d, 1H), 7.36 (d, 1H), 7.41 (d, 1H), 12.28 (s, 1H).

Trifluoromethanesulfonic Acid 6-[2-(Trimethylsilanyl)ethoxymethyl]pyridin-2-yl Ester (13). To a solution of 12 (3.80 g, 14.8 mmol) and DMAP (0.10 g, 0.82 mmol) in pyridine (35 mL) at 0 °C was added dropwise  $(CF_3SO_2)_2\hat{O}$  (2.63 mL, 14.9 mmol). After stirring for 2  $\hat{h}$  at 0 C, the mixture was poured into ice—water and extracted with Et<sub>2</sub>O. The combined organic layers were washed with an aqueous KHSO4 solution and water, dried, filtered, and evaporated to dryness. Purification by flash chromatography

(CH<sub>2</sub>Cl<sub>2</sub>) gave 4.30 g (75%) of 13 as a colorless oil: <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.00 (s, 9H), 0.93 (t, 2H), 3.66 (t, 2H), 4.68 (s, 2H), 4.80 (s, 2H), 7.05 (d, 1H), 7.52 (d, 1H), 7.87 (t, 1H).

2-[6-[2-(Trimethylsilanyl)ethoxymethoxymethyl]pyridin-2-yl]pyrrole-1-carboxylic Acid tert-Butyl Ester (14c). To a degassed solution of 13 (4 g, 10.3 mmol) in benzene (40 mL) were added  $Tl_2CO_3$  (9.68 g, 20.6 mmol), Pd  $[P(C_6H_5)_3]_4$  (1 g, 0.86 mmol), and [1-(tert-butoxycarbonyl)pyrrol-2-yl]boronic acid<sup>38</sup> (2.45 g, 11.6 mmol). After stirring at room temperature for 23 h, the mixture was filtered and the filtrate concentrated under reduced pressure. The oily residue was purified by flash chromatography (1% EtOAc-CH<sub>2</sub>Cl<sub>2</sub>) to give 3.75 g (90%) of **14c** as a pale-yellow oil:  ${}^{1}H$  NMR (CDCl<sub>3</sub>)  $\delta$  0.00 (s, 9H), 0.94 (t, 2H), 1.32 (s, 9H), 3.67 (t, 2H), 4.73 (s, 2H), 4.82 (s, 2H), 6.21 (t, 1H), 6.37 (m, 1H), 7.27 (d, 1H), 7.32 (m, 1H), 7.34 (d, 1H), 7.68 (t, 1H).

**6-(1***H***-Pyrrol-2-yl)pyridine-2-methanol (15c).** To a solution of (nC<sub>4</sub>H<sub>9</sub>)<sub>4</sub>NF (1.10 M THF solution, 3 mL), anhydrous THF (45 mL), HMPA (5 mL), and 4-Å molecular sieves (10 g) was added **14c** (3 g, 7.41 mmol). The mixture was stirred at room temperature under argon for 8 h. Insoluble materials were filtered off, and the solvent was removed under vacuum. Purification by flash chromatography (1% EtOAc-CH<sub>2</sub>Cl<sub>2</sub>) gave 0.51 g (39%) of **15c** as a white solid: mp 73-75 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.61 (s, 1H), 4.74 (s, 2H), 6.30 (dd, 1H), 6.75 (m, 1H), 6.92 (m, 1H), 6.98 (d, 1H), 7.45 (d, 1H), 7.62 (t, 1H), 9.56 (s, 1H). Anal. (C<sub>10</sub>H<sub>10</sub>N<sub>2</sub>O) C, H, N.

6-(Thien-2-yl)pyridine-2-methanol (15d). Prepared as 15c from 13 and commercially available thien-2-ylboronic acid. Purification by flash chromatography (10% EtOAc-CH<sub>2</sub>Cl<sub>2</sub>) gave **15d** as a yellow oil (45%): <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.7 (s, 1H), 4.65 (s, 2H), 7.10 (dd, 1H), 7.35 (m, 2H), 7.66 (m, 2H), 7.85 (m, 1H).

6-(Furan-2-yl)pyridine-2-methanol (15e). Prepared as 15c from 13 and commercially available furan-2-ylboronic acid. Purification by flash chromatography (10% EtOAc-CH<sub>2</sub>Cl<sub>2</sub>) gave **15e** as a yellow oil (56%): <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.99 (s, 1H), 4.74 (s, 2H), 6.50 (dd, 1H), 7.06 (m, 2H), 7.49 (d, 1H), 7.56 (d, 1H), 7.67 (t, 1H).

6-(1H-Pyrrol-2-yl)pyridine-2-carboxaldehyde (8c). To a solution of 15c (0.30 g, 1.72 mmol) and anhydrous CH<sub>2</sub>Cl<sub>2</sub> (3 mL) was added MnO<sub>2</sub> (0.75 g, 8.61 mmol), and the suspension was stirred under reflux for 1 h. Insoluble materials were filtered, and the solvent was removed under vacuum. Purification by flash chromatography (CH2Cl2) afforded 0.20 g (66%) of **8c** as a white solid: mp 112–113 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  6.31 (dd, 1H), 6.77 (m, 1H), 6.95 (m, 1H), 7.63–7.81 (m, 3H), 9.67 (s, 1H), 10.03 (s, 1H); IR (KBr, cm<sup>-1</sup>) 1701. Anal. (C<sub>10</sub>H<sub>8</sub>N<sub>2</sub>O) C, H, N.

6-(Thien-2-yl)pyridine-2-carboxaldehyde (8d): mp 48-50 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 7.15 (dd, 1H), 7.46 (dd, 1H), 7.69 (dd, 1H), 7.78–7.88 (m, 3H), 10.16 (s, 1H); IR (KBr, cm<sup>-1</sup>) 1714. Anal.  $(C_{10}H_7NOS)$  C, H, N.

6-(Furan-2-yl)pyridine-2-carboxaldehyde (8e): mp 46-48 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  6.55 (q, 1H), 7.16 (d, 1H), 7.55 (d, 1H), 7.82 (m, 3H), 10.07 (s, 1H); ÎR (KBr, cm<sup>-1</sup>) 1717. Anal.  $(C_{10}H_7NO_2)$  C, H, N.

2-(4,5-Dihydrothiazol-2-yl)-6-(1,3-dioxolan-2-yl)pyri**dine (7i).** A solution of 2-(1,3-dioxolan-2-yl)-6-cyanopyridine<sup>18</sup> (2 g, 11.35 mmol), methanol (10 mL), and sodium methoxide (0.20 g, 3.70 mmol) was stirred at room temperature for 24 h. The solvent was distilled, and the residue was taken up in EtOAc, washed with brine, dried, filtered, and concentrated under vacuum. The crude imidate (1 g, 4.80 mmol) was mixed with 2-aminoethanethiol (0.47 g, 6.10 mmol) and heated at 130 °C for 1.5 h. The cooled mixture was taken up in CH<sub>2</sub>Cl<sub>2</sub>, washed with water, dried, filtered, and concentrated. Purification by flash chromatography (2% MeOH-CH $_2$ Cl $_2$ ) gave 1.10 g (97%) of 7i as a pale-yellow solid: mp 55-57 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  3.36 (t, 2H), 4.15 (m, 4H), 4.55 (t, 2H), 5.91 (s, 1H), 7.60 (dd, 1H), 7.81 (t, 1H), 8.03 (dd, 1H). Anal. (C<sub>11</sub>H<sub>12</sub>N<sub>2</sub>O<sub>2</sub>S) C, H, N.

2-(1H-Imidazolin-2-yl)-6-(1,3-dioxolan-2-yl)pyridine (7j). Prepared from 7f and ethylenediamine (55%): 1H NMR (CDCl<sub>3</sub>)  $\delta$  4.10 (m, 8H), 5.82 (s, 1H), 6.03 (s, 1H), 7.56 (dd, 1H), 7.78 (t, 1H), 8.10 (dd, 1H).

2-(Thiazol-2-yl)-6-(1,3-dioxolan-2-yl)pyridine (7k). A solution of 7i (1.33 g, 5.62 mmol) in benzene (50 mL) was treated with NiO2 (hydrate, 10 g in 10 portions) while the reaction mixture was heated under reflux for 20 h with elimination of water with a Dean-Stark trap. The cooled reaction mixture was filtered through Celite, and the filtrate was concentrated under vacuum. The crude product was purified by flash chromatography (15% EtOAc-CH<sub>2</sub>Cl<sub>2</sub>) to give 0.42 g (31%) of 7k as a white solid: mp 70-72 °C; <sup>1</sup>H NMR (CDC $\bar{l}_3$ )  $\delta$  4.16 (m, 4H), 5.91 (s, 1H), 7.42 (d, 1H), 7.54 (dd, 1H), 7.81 (d, 1H), 7.89 (t, 1H), 8.17 (dd, 1H).

2-(1*H*-Imidazol-2-yl)-6-(1,3-dioxolan-2-yl)pyridine (7l). Prepared from 7j using BaMnO4 as oxidant to give after purification by flash chromatography (5% MeOH-CH $_2$ Cl $_2$ ) 71 as a yellow gummy solid (68%):  $^1$ H NMR (CDCl $_3$ )  $\delta$  2.24 (s, 1H), 4.12 (m, 4H), 5.83 (s, 1H), 7.09 (s, 1H), 7.18 (s, 1H), 7.44 (dd, 1H), 7.78 (t, 1H), 8.11 (dd, 1H).

6-(Thiazol-2-yl)pyridine-2-carboxaldehyde (8k). Prepared from 7k and obtained as a white solid (78%): mp 90-92 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 7.48 (d, 1H), 7.96 (m, 3H), 8.38 (m, 1H), 10.09 (s, 1H); IR (KBr, cm<sup>-1</sup>) 1705. Anal. (C<sub>9</sub>H<sub>6</sub>N<sub>2</sub>OS)

6-(1H-Imidazol-2-yl)pyridine-2-carboxaldehyde (8l). Prepared from 71 and obtained as a yellow solid (34%): mp 147-149 °C; ¹H NMR (CDCl<sub>3</sub>) δ 7.23 (m, 2H), 7.89 (m, 2H), 8.37 (dd, 1H), 10.05 (s, 1H), 10.77 (s, 1H). Anal. (C<sub>9</sub>H<sub>7</sub>N<sub>3</sub>O) C, H, N.

1-[6-(1,3-Dioxolan-2-yl)pyridin-2-yl]ethanone (7g). To a solution of 7f (0.88 g, 4.99 mmol) in anhydrous THF (10 mL), under an argon atmosphere and cooled at −10 °C, was added a solution of methylmagnesium bromide (3 M in Et<sub>2</sub>O, 3.5 mL, 10.5 mmol). After stirring at room temperature for 3 h, the mixture was quenched with a saturated aqueous ammonium chloride solution and vigorously stirred at room temperature for 30 min. The mixture was extracted with EtOAc, washed with brine, dried, and filtered, and the solvent was evaporated under vacuum. Purification by flash chromatography (CH<sub>2</sub>Cl<sub>2</sub>) gave 0.80 g (83%) of 7g as a yellow oil:  ${}^{1}H$  NMR (CDCl<sub>3</sub>)  $\delta$ 2.72 (s, 3H), 4.15 (m, 4H), 5.89 (s, 1H), 7.70 (dd, 1H), 7.86 (t, 1H), 8.01 (dd, 1H); IR (film, cm<sup>-1</sup>) 1700.

2-(1*H*-Pyrazol-3-yl)-6-(1,3-dioxolan-2-yl)pyridine (7h). A solution of 7g (0.80 g, 4.14 mmol) and DMF dimethyl acetal (1 mL, 7.53 mmol) was heated under reflux for 12 h. Excess DMF acetal was removed under vacuum, and the residual oil was taken up in EtOH (5 mL). Hydrazine hydrate (0.80 mL, 25.7 mmol) was added, and the solution was heated under reflux for 10 min. The solvent was evaporated, and water was added. The mixture was extracted with EtOAc, washed with water, dried, filtered, and distilled under vacuum. Purification by flash chromatography (2% MeOH-CH2Cl2) gave 0.65 g (72%) of **7h** as a yellow gum:  ${}^{1}$ H NMR (CDCl<sub>3</sub>)  $\delta$  4.15 (m, 4H), 5.90 (s, 1H), 6.80 (d, 1H), 7.48 (dd, 1H), 7.65 (d, 1H), 7.76 (m, 2H), 11.23 (s, 1H).

6-(1H-Pyrazol-3-yl)pyridine-2-carboxaldehyde (8h). Prepared from 7h and obtained as an amorphous white solid (60%): <sup>1</sup>H NMR (DMSO-*d*<sub>6</sub>) δ 6.91 (d, 1H), 7.80 (m, 2H), 8.03 (t, 1H), 8.20 (d, 1H), 9.99 (s, 1H), 13.17 (s, 1H); IR (KBr, cm<sup>-1</sup>)

2-(Oxazol-5-yl)-6-(1,3-dioxolan-2-yl)pyridine (7n). A mixture of 7m (1 g, 5.58 mmol), tosylmethyl isocyanide (1.10 g, 5.63 mmol), K<sub>2</sub>CO<sub>3</sub> (0.80 g, 5.79 mmol), and methanol (15 mL) was heated under reflux for 2 h. The solvent was evaporated, and water was added. The product was extracted with EtOAc, washed with brine, dried, filtered, and concentrated under vacuum. Purification by flash chromatography (50% EtOAc-hexane) gave 1.13 g (96%) of **7n** as a yellow oil:  $^{1}$ H NMR (CDCl<sub>3</sub>)  $\delta$  1.34 (d, 6H), 3.10 (m, 1H), 7.37 (m, 1H), 7.73 (d, 2H), 10.03 (s, 1H); IR (film, cm<sup>-1</sup>) 1712.

6-(Oxazol-5-yl)pyridine-2-carboxaldehyde (8n). Prepared from **7n** and obtained as a pale-yellow solid (41%): mp 150-152 °C; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  7.81-8.01 (m, 5H), 10.08 (s, 1H); IR (KBr, cm<sup>-1</sup>) 1703. Anal. (C<sub>9</sub>H<sub>6</sub>N<sub>2</sub>O<sub>2</sub>) C, H, N.

6-(Methylthio)-2-(1,3-dioxolan-2-yl)pyridine (7p). A mixture of 7a (1.36 g, 5.91 mmol), NaSCH<sub>3</sub> (0.83 g, 11.82 mmol), and DMF (4 mL) was heated at 100 °C for 18 h. The cooled mixture was diluted with ice-water and extracted with EtOAc. The combined organic layers were washed with 1 N NaOH and water, dried, and filtered, and the solvent was removed under vacuum to give 0.85 g (73%) of **7p** as a yellow oil:  ${}^{1}H$  NMR (CDCl<sub>3</sub>)  $\delta$  2.56 (s, 3H), 4.03–4.20 (m, 4H), 5.82 (s, 1H), 7.16 (t, 2H), 7.50 (t, 1H).

6-(Methylsulfinyl)-2-(1,3-dioxolan-2-yl)pyridine (7q). To a solution of NaIO $_4$  (0.72 g, 3.36 mmol) in water (15 mL) and maintained at 0 °C was added dropwise a solution of 7p (0.60 g, 3.04 mmol) in methanol (10 mL). The mixture was allowed to stir at room temperature for 18 h. The methanol was removed under vacuum; the residue was taken up in water and extracted with CH<sub>2</sub>Cl<sub>2</sub>. The combined extracts were washed with water and brine, dried, and filtered, and the solvent was removed under vacuum. The product was purified by flash chromatography (50% EtOAc-CH<sub>2</sub>Cl<sub>2</sub>) to give 0.48 g (67%) of **7q** as a pale-yellow oil:  ${}^{1}H$  NMR (CDCl<sub>3</sub>)  $\delta$  2.82 (s, 3H), 4.01-4.18 (m, 4H), 5.79 (s, 1H), 7.58 (dd, 1H), 7.97 (m,

6-(Methylthio)pyridine-2-carboxaldehyde (8p). Prepared from **7p** and obtained as a yellow oil (45%): <sup>1</sup>H NMR  $(CDCl_3)$   $\delta$  2.62 (s, 3H), 7.37 (dd, 1H), 7.60 (m, 2H), 9.99 (s, 1H); IR (film, cm<sup>-1</sup>) 1709.

6-(Methylsulfinyl)pyridine-2-carboxaldehyde (8q). Prepared from 8q and obtained as a pale-yellow oil (60%): <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  2.92 (s, 3H), 8.02 (dd, 1H), 8.14 (t, 1H), 8.27 (dd, 1H), 10.03 (s, 1H); IR (film, cm<sup>-1</sup>) 1710.

6-n-Butoxy-2-(1,3-dioxolan-2-yl)pyridine (7o). To a solution of Na (0.30 g, 0.013 atg) in n-butanol (3 mL) was added 7a (0.91 g, 5.39 mmol), and the mixture was heated at 100 °C for 18 h. After cooling to room temperature, the mixture was diluted with water and extracted with EtOAc. The organic layer was washed with water and brine, dried, and filtered, and the solvent was removed under vacuum. Residual nbutanol was removed by azeotropic distillation with cyclohexane to give 1.08 g (90%) of **70** as a yellow oil: <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.96 (t, 3H), 1.40-1.80 (m, 4H), 4.13 (m, 4H), 4.31 (t, 2H), 5.77 (s, 1H), 6.69 (d, 1H), 7.06 (d, 1H), 7.57 (dd, 1H).

6-n-Butoxypyridine-2-carboxaldehyde (80). Prepared from **70** and obtained as a yellow oil (75%): <sup>1</sup>H NMR (CDCl<sub>3</sub>) δ 0.99 (t, 3H), 1.50 (m, 2H), 1.77 (m, 2H), 4.42 (m, 2H), 6.95 (dd, 1H), 7.54 (dd, 1H), 7.72 (t, 1H), 8.09 (s, 1H); IR (film, cm<sup>-1</sup>)

Preparation of 6-Pyrazol-1-yl-2-pyridinylmethylamino **Derivatives 75-85 (Table 3).** To a mixture of **8b** and an equimolar quantity of the corresponding amine<sup>39</sup> in 1,2dichloroethane was added 1.4 equiv of sodium triacetoxyborohydride. The mixture was stirred at room temperature for 24 h and then quenched by adding 1 N NaOH. The product was extracted with EtOAc, purified by flash chromatography, and converted to the appropriate salts.

Radioligand Binding. Binding affinities for the different receptors were determined by means of ligand displacement assays using the conditions summarized in Table 1. The reactions were stopped by rapid filtration through Whatman GF/B glass fiber filters, and the filters were washed with appropriate buffer. The radioactivity retained on the filters was measured by scintillation spectroscopy in 4 mL of scintillation fluid (Emulsifier Safe, Packard).

Cyclic AMP in HA7 Cells. HA7 cells were grown in Dulbecco's modified Eagle's medium (DMEM) (GIBCO) supplemented with 10% fetal calf serum, gentamic in (100  $\mu g/mL$ ), Geneticin (G418) (400  $\mu$ g/mL) in 5% CO<sub>2</sub> at 37 °C in a watersaturated atmosphere. The cells were plated in 6-well culture plates and used in the experiments at a confluency of 80-90%. Culturing medium [DMEM, 10% fetal calf serum, gentamicin (100  $\mu$ g/mL), G418 (400  $\mu$ L/mL)] was replaced by DMEM supplemented with 10% fetal calf serum without antibiotics 24 h before experimentation.

Cells were preincubated with DMEM, 10 mM Hepes for 10 min at room temperature. Drugs, at concentrations ranging from 0.1 nM to 100  $\mu$ M, and appropriate vehicle controls [i.e., water or dimethyl sulfoxide (DMSO)] were then added in DMEM, 10 mM Hepes, 100  $\mu$ M forskolin, 100  $\mu$ M 3-isobutyl-1-methylxanthine (IBMX) to the cells. At the end of the treatment (10 min, room temperature), the reaction was stopped by aspiration of the medium and addition of 0.1 N HCl. Cellular extract was diluted 1:500 or 1:400 in radioimmunoassay buffer, and cyclic AMP (cAMP) content was measured by using a commercially available kit (DuPont NEN: NEK-033). Basal cAMP levels were  $10 \pm 0.9$  pmol/well (n = 8).

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Supporting Information Available: Additional physical data (<sup>1</sup>H NMR and elemental analyses) on final compounds and the synthetic sequences used to prepare intermediates 81, 83, and 85 (13 pages). Ordering information is given on any current masthead page.

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