Silicon-Tethered Cyclocarbonylation of Alkynes

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ABSTRACT



Temporary union of alkynes in the Fe(CO)₅-promoted cyclocarbonylation provides a method for carbonylative cross-coupling of alkynes to construct unsymmetrically substituted cyclopentadienones.

It is known that $Fe(CO)_5$ -promoted cyclocarbonylation¹ can provide a method for the construction of five-membered carbocyclic ring systems, which is potentially useful owing to the occurrence of such rings in a diverse range of natural product molecules.² Although there are a number of examples of homocoupling via this reaction where two alkynes react with Fe(CO)₅ in CO atmosphere to give the symmetrically substituted cyclopentadienone-Fe(CO)₃ complexes (eq 1),³ no example of cross-coupling of two alkynes has been reported to the best of our knowledge.



We report here a method for cross-coupling of alkynes via "silicon tethered"⁴ Fe(CO)₅-promoted cyclocarbonylation, which consists of linking two alkynes into a silyl acetal, subjecting the resulting diyne to cyclocarbonylation, and disengaging the metal and the tether moiety. Tethering the two alkyne reaction partners of Fe(CO)₅-promoted cyclo-

carbonylation makes this reaction intramolecular, thereby leading to better yield and regioselectivity than are generally associated with intermolecular reactions. Shibata et al. have reported a related silicon-tethered diyne *homo*coupling promoted by octacarbonyldicobalt.⁵ Our strategy was first tested with a simple substrate **1**, which was obtained by the reaction of trimethylsilylpropynol⁶ with dichlorodiphenylsi-

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lane in the presence of base. The substrate 1, which has two alkyne reaction units connected by a silyl acetal moiety, underwent cyclocarbonylation in good yield considering that the reaction involves seven-membered ring formation.⁷

Treatment of the resulting cyclopentadienone $-Fe(CO)_3$ complex 2 with Me₃NO⁸ afforded two products, 3 and 4,⁹ without the known possible problems³ such as dimerization of cyclopentadienone or migration of one double bond and accompanying desilylation. Removal of the tether moiety in 3 was completed by treatment with HF (Scheme 1).¹⁰ Also,



a sterically more demanding substrate **5** was prepared from trimethysilylacetylene by a one-pot process involving sequential treatment with *n*-BuLi,¹¹ acetaldehyde, and dichlorodiphenylsilane. The cyclocarbonylation of diyne **5** over 48 h gave all the possible diastereomers **6a**, **6b**, and **6c** in a ratio of ca. 1:1:1.2, which gave two diastereomers **7a** and *rac*-**7b** after demetalation and concomitant desilylation by Me₃NO treatment (Scheme 2). However, the same reaction for 24 h gave a 1:1 mixture of **6a** and **6b** in 44% yield, with only a trace amount of **6c**. Me₃NO treatment of this mixture gave only **7a** in 42% yield.

The inseparable mixture of $Fe(CO)_3$ complexes **6a**, **6b**, and **6c** was analyzed by comparison of the ¹H NMR spectrum of the mixture with that of (*S*,*S*)-**6c**, which was synthesized from commercially available (*S*)-3-butynol (Scheme 3).

These results indicate that the meso diastereomer of **5** reacts faster than the (S,S)- or (R,R)-diastereomer. The



putative intermediate, the tricarbonyl[bis-(η^2 -alkyne)]iron complex, may be considered to account for the reactivity difference between two diastereomers. The tricarbonyl[bis-(η^2 -alkyne)]iron complexes,¹² which would be formed by the sequential replacement of two CO ligands by two alkyne units of the diyne **5**, are depicted by **RS-9** and **SS-9** (Figure 1).

Due to the *pseudo* 1,3-diaxial interaction between phenyl and methyl that is unavoidable in **SS-9**, the formation of **SS-9**



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Figure 1. Tricarbonyl[bis- $(\eta^2$ -alkyne)]iron complexes (**RS-9** and SS-9) and the diyne 10.

is expected to be slower than that of **RS-9**. As a result, the corresponding Fe(CO)₃ complex 6c is formed more slowly. The same reasoning also explains the observation that 10, which has two additional methyl groups compared to the substrate 5, did not undergo the cyclocarbonylation but was recovered unreacted. Divne 11, which was prepared by the same method as used for 5, reacted with $Fe(CO)_5$ to give the inseparable diastereomeric mixture 12. The anticipated mixture of two diastereomers 13a,b was obtained in ca. 1.1:1 ratio (Scheme 4). The ¹H NMR spectrum of the mixture 12 was not useful for determining the ratio of the diastereomers.





SiMe₃

HC

8h



We did not examine the potential selectivity of this conversion for a shorter reaction time because of the poorer overall yield.

Since we found that the silvl acetal moiety is a good tether for Fe(CO)₅-promoted cyclocarbonylation, we prepared the divnes 16 and 17, which have two different alkyne units tethered by the silvl acetal moiety, according to the route depicted in Scheme 5. Monoalkoxylation was accomplished by using a 5-fold excess of dichlorodiphenylsilane, the excess of which could be recovered by distillation of the resulting

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mixture.¹³ The addition of another alkynol (8^{14} or 15^{15}) to the obtained monochlorosilane 14 by using 2,6-lutidine as a base afforded the diynes 16 and 17, which underwent cyclocarbonylation to give the mixture of two diastereomers 18a,b and 19a,b, respectively. Treatment of the diastereomeric mixtures of the Fe(CO)₃ complexes with Me₃NO gave single products 20 and 21.

The diols **20** and **21** are equivalent to the products of an intermolecular version of this reaction that would give not

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only the cross-coupled product but also the homocoupled products and possibly their regioisomers.

In conclusion, we have demonstrated that "silicontethered" Fe(CO)₅-promoted cyclocarbonylation can provide an effective method for carbonylative cross-coupling of alkynes to furnish cyclopentadienones with variable substituents.

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Supporting Information Available: Experimental procedures, spectroscopic data, copies of ¹H NMR and ¹³C NMR spectra for new compounds. This material is available free of charge via the Internet at http://pubs.acs.org.

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