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An allylation-metathesis sequence as a convergent strategy towards enantiopure equivalents of highly functionalized cyclic dienes

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Dedicated to Professor Georges Hoornaert on the occasion of his retirement from the KUL.

Abstract—Highly enantioenriched equivalents of cyclic dienes have been readily prepared by asymmetric allylation of unsaturated aldehydes using a chiral allyltitanium reagent, followed by a ring-closing metathesis. The resulting β -hydroxy allylsilanes react stereoselectively with a wide variety of electrophilic reagents. © 2003 Elsevier Ltd. All rights reserved.

We have previously reported a two-step reaction sequence consisting of an asymmetric allylation of an aldehyde followed by a diastereoselective silicondirected [2+2] cycloaddition (Scheme 1).¹ This illustrated a convergent strategy for the enantioselective synthesis of polycyclic compounds.

More recently we became interested in a variation of this strategy which combines an asymmetric metallation (step 1) with a ring-closing metathesis (RCM) as a general route to enantiomerically pure β -hydroxy-allyl-silanes 1 which can be looked upon as *chiral equivalents* of cyclic conjugated dienes (Scheme 2). The dense functionality of these products make them prone to undergo useful transformations into a wide range of enantiomerically enriched mono- and polycyclic molecules.

Examples of sequential asymmetric allylation-RCM have been reported, but in these cases the olefinic partners for the RCM were linked by a heteroatom which became part of the newly formed ring.² The sequence of Scheme 2 has been very recently investigated by Roush et al. who showed that it could be elegantly applied to the synthesis of conduritols and inositols of high enantiomeric purity.³ This report



H N + PhMe₂Si Cp Ti Taddol

enriched synthetic intermediates.

prompts us to quickly describe our own studies on the

enantioselective synthesis of β -hydroxy allylsilanes 1

and their transformation into a wide variety of enantio-



Scheme 1.



Scheme 2.

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In our studies, the allylmetallation (step 1 of Scheme 2) of the unsaturated aldehydes $3\mathbf{a}-\mathbf{c}^4$ was effected with the (*E*)- γ -dimethylphenylsilyl substituted titanium reagent **2** generated from allyldimethylphenylsilane, *n*-BuLi and (*R*,*R*)-Cl-TiCpTADDOL (Table 1).⁵

The reaction gave only the *anti*- β -hydroxyallylsilanes 4a-c in good yields and high enantiomeric purities. Silylation of 4a-c yielded the corresponding silyl ethers 5a-c which were submitted to the ring-closing metathe-

 Table 1. Asymmetric allylmetallation of 3-butenal, 4-pentenal and 5-hexenal with 2



Reagents and conditions: (a) *n*-BuLi (1.1 equiv.) in THF, 20 min, rt then (R,R)-Cl-TiCpTADDOL (1.15 equiv.), 30 min, -78° C to give a solution of **2**; (b) aldehyde **3** (1 equiv.), 3 h, -78° C, 40% aqueous NH₄F, 15 h, rt, flash chromatography (100% CH₂Cl₂), ee determined by HPLC on a Chiralcel AS or OD column.

Table 2. Ring-closing metathesis of 4b and 5a,c

sis (Table 2). The configurations of the two new stereogenic centers (3S, 4R) of 4a were assigned by X-ray diffraction analysis.¹ The absolute configurations of 5a-c were assigned by analogy. *t*-Butyldimethylsilyl ethers 5a,b underwent an extremely fast RCM reaction in the presence of the first generation Grubbs'catalyst 8 (entries 1 and 2). Thus, with 5b, the RCM was completed in one hour in refluxing dichloromethane with as little as 0.05 mol% catalyst (entry 3). The reaction was much slower with the free alcohol 4b which is a better coordinating group than the corresponding TBS ether and could therefore deactivate the catalyst (entry 4).^{2a} The RCM of the homologous diene 5c was much slower, requiring higher temperatures and more catalyst (entry 5). However the use of 2 mol% of the new generation Grubbs' catalyst 9 enabled the reaction to take place at room temperature in 2 hours (entry 6).

Compounds **6a–c** contain a unique combination of an allyl- and a β -silyloxysilane which should be prone to undergo interesting transformations leading to a wide variety of enantioenriched, highly functionnalized synthetic building blocks. Scheme 3 shows a selection of diastereoselective transformations illustrating the synthetic utility of **6a–c**. In these transformations, the silyl group acts as stereodirecting group favouring the attack on the opposite face. The enantiomeric purity of all compounds described in Scheme 3 were identical to those of *anti-* β -hydroxyallylsilanes **4a–c**.

Clearly the synthetic utility of this asymmetric allylmetallation-RCM sequence would be greatly enhanced if it could be successfully applied to substituted aldehydes (Scheme 4). We have examined the silyl allyltitanation of the acetonide-protected aldehyde 14 which



Entry	Diene	п	Cat. (mol%)	Conditions for (b)	Prod.	Yield (%)
1	5a	1	8 (2)	DCM. rt. 5 min	6a	98
2	5b	2	8 (1.6)	DCM, rt, 2 min	6b	99
3	5b	2	8 (0.05)	DCM, reflux, 1 h	6b	99
4	4b	2	8 (5)	DCM, rt, 3 h	7	86
5	5c	3	8 (7.5)	Toluene, 60°C, 2 h	6c	93
6	5c	3	9 (2)	DCM, rt, 2 h	6с	82

Reagents: (a) TBSCl (1.5 equiv.), imidazole (3 equiv.), 36 h, 50°C, DMF; (b) Mes=2,4,6-trimethylphenyl and Cy=cyclohexyl.



Scheme 3. Reagents and conditions: (a) N-methyl-1,2-oxido-1,2,3,4-tetrahydroiso-quinoleine tetrafluoroborate (1 equiv.), CH₂Cl₂, rt, 30 min; (b) Ethyl diazoacetate (1.2 equiv.), Cu-(acac)₂ (2 mol%), toluene, 90°C, 3 h; (c) K₂OsO₄·2H₂O (2 mol%), NMO.H₂O (1.7 equiv.), H₂O/THF, rt, 48 h; (d) Et₃BnNCl (0.1 equiv.), NaOH 50%, CHCl₃, rt, 15 h.



Scheme 4. Reagents and conditions: (a) *n*-BuLi (1.1 equiv.) in THF, 20 min, rt then (R,R)-Cl-TiCpTADDOL (1.15 equiv.), 30 min, -78°C then 14 (1 equiv.), 3 h, -78°C, 40% aqueous NH₄F, 15 h, rt; (b) pyridine (3.5 equiv.), CH₃COCl (2.5 equiv.), DCM, 4 h, rt; (c) Grubbs'catalyst 9 (8 mol%), boiling DCM, 2 h.

proceeded indeed with high diastereoselectivity to give the corresponding *anti*- β -hydroxyallylsilane **15**. The corresponding acetate **16** was subjected to RCM in the presence of the second generation Grubbs'catalyst to give 80% yield of the polysubstituted cylohexene **17** in high enantiomeric purity (ee>95%).

¹H NMR analysis and molecular modeling (Monte Carlo exploration of conformational space with MM3 force field)⁶ of **17** suggested a half-chair conformation with a pseudoaxial silyl substituent (Fig. 1). This should allow an interaction (β -effect) with the olefinic bond and favor electrophilic attack on the face opposite to



Figure 1. Lowest energy conformation of 17.

the one occupied by the silyl substituent. However, the dichlorocyclopropanation reaction under the classical conditions (PTC or *t*-BuOK+CHCl₃) did not take place, probably as a result of the steric hindrance on both faces of the double bond. It is interesting to note that the dihydroxylation of the double bond was possible when the two oxygen atoms were not part of a ring.^{3b}

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