

Enantioselective [4 + 2]-Annulation of Chiral Crotylsilanes: Application to the Synthesis of a C1–C22 Fragment of Leucascandrolide A

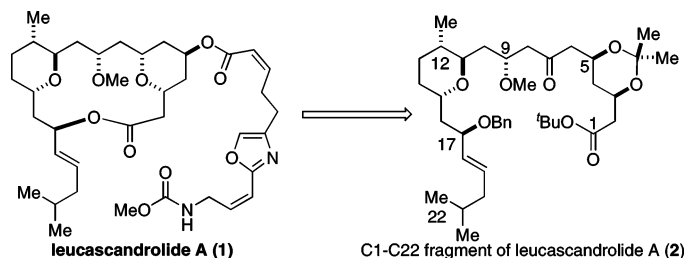
Les A. Dakin and James S. Panek*

Department of Chemistry and Center for Chemical Methodology and Library
Development, Boston University, 590 Commonwealth Avenue,
Boston, Massachusetts 02215

panek@chem.bu.edu

Received August 21, 2003

ABSTRACT



The asymmetric synthesis of a C1–C22 fragment (2) of leucascandrolide A is described. Synthetic highlights include the construction of the C9–C22 pyran fragment using a formal [4 + 2]-annulation of a chiral organosilane. A diastereoselective Mukaiyama aldol was used to introduce the C9 stereocenter and complete the assembly of the macrocycle's carbon skeleton.

Leucascandrolide A **1** is a doubly O-bridged 18-membered macrolide isolated from the calcareous sponge *Leucascandra caveolata*, obtained from the northeastern coast of New Caledonia, Coral Sea.¹ This macrolide exhibits high in vitro cytotoxicity against human KB and P388 tumor cell lines displaying low IC₅₀ values of 0.05 and 0.26 $\mu\text{g/mL}$, respectively. The natural product also possesses potent antifungal ability against *Candida albicans*, a pathogenic yeast that attacks AIDS patients and other immunocompromised individuals. A subsequent report indicates that leucascandrolide A is no longer available from its initial natural source.² It has been postulated that **1** is not a metabolite of *Leucascandra caveolata* but rather an opportunistic bacteria as evidenced by the large amounts of dead tissue in the initial

harvest of *Leucascandra caveolata*. Currently there is no natural source of leucascandrolide A.

The lack of a natural source of **1**, together with its unique doubly oxygenated 18-membered macrolide, has made leucascandrolide A, a target of interest for the synthetic community. Following the first total synthesis by Leighton,³ there have been additional reports detailing total,⁴ formal,⁵ and fragment syntheses of **1**.⁶ Herein we discuss our approach

(1) D'Ambrosio, M.; Guerriero, A.; Debitus, C.; Pietra, F. *Helv. Chim. Acta.* **1996**, *79*, 51–60.

(2) (a) D'Ambrosio, M.; Tato, M.; Pocsfalvi, G.; Debitus, C.; Pietra, F. *Helv. Chim. Acta* **1999**, *82*, 347–353. (b) D'Ambrosio, M.; Tato, M.; Pocsfalvi, G.; Debitus, C.; Pietra, F. *Helv. Chim. Acta* **1999**, *82*, 1135.

(3) Hornberger, K. R.; Hamblett, C. L.; Leighton, J. L. *J. Am. Chem. Soc.* **2000**, *122*, 12894–12895.

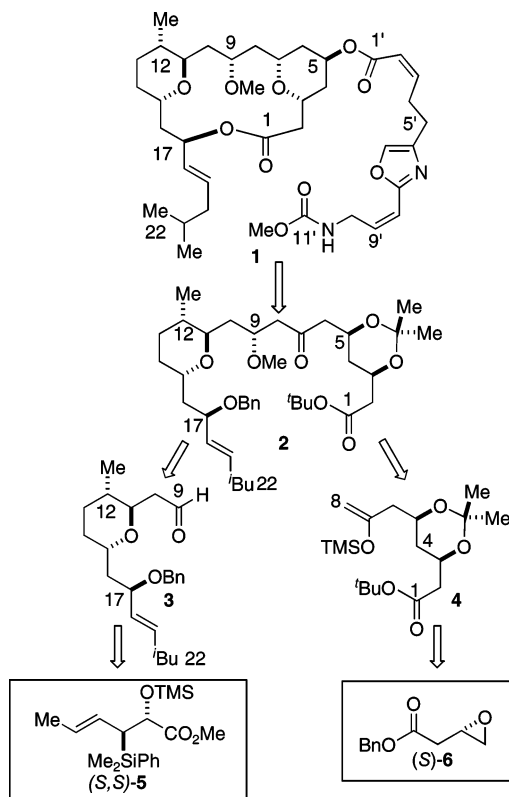
(4) (a) Fettes, A.; Carreira, E. M. *Angew. Chem., Int. Ed.* **2002**, *41*, 4098–4101. (b) Paterson, I.; Tudge, M. *Angew. Chem., Int. Ed.* **2003**, *42*, 343–347. (c) Wang, Y.; Janic, J.; Kozmin, S. A. *J. Am. Chem. Soc.* **2002**, *124*, 13670–13671.

(5) (a) Kopecky, D. J.; Rychnovsky, S. D. *J. Am. Chem. Soc.* **2001**, *123*, 8420–8421. (b) Wipf, P.; Reeves, J. T. *Chem. Commun.* **2002**, 2066–2067.

(6) (a) Crimmins, M. T.; Carroll, C. A.; King, B. W. *Org. Lett.* **2000**, *2*, 597–599. (b) Wipf, P.; Graham, T. H. *J. Org. Chem.* **2001**, *66*, 3242–3245. (c) Dakin, L. A.; Langille, N. F.; Panek, J. S. *J. Org. Chem.* **2002**, *67*, 6812–6815.

to macrolide and describe the synthesis of the C1–C22 advanced intermediate. Concerning the retrosynthesis of **1**, disconnection of the C5 ester gives the macrocyclic lactone, which could be derived from the C1–C22 fragment **2** (Scheme 1). Further analysis of **2** suggests that it could be

Scheme 1. Retrosynthetic Analysis of Leucascandrolide A



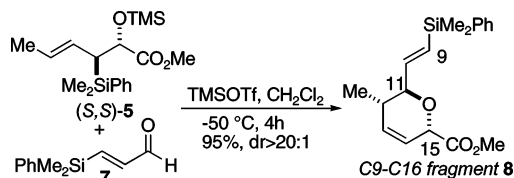
divided at the C8–C9 bond to give the Mukaiyama aldol-coupling partners: pyran **3** and silyl enol ether **4**.⁷ The C9–C22 pyran **3** could be synthesized using the [4 + 2] annulation of chiral crotylsilanes developed in our laboratories between chiral silane (*S,S*)-**5** and an appropriate aldehyde.⁸

Silyl enol ether **4**, could be synthesized in a straightforward manner starting with enantioenriched epoxide (*S*)-**6** as an initial building block, readily available via Jacobsen's hydrolytic kinetic resolution (HKR).⁹

To begin the synthesis of fragment **3**, a [4 + 2] annulation of (*S,S*)-**5** with selected aldehydes capable of serving as a precursor to the C9 aldehyde was evaluated. After a brief examination of the reaction, it was determined that reaction of (*S,S*)-**5** with aldehyde **7** catalyzed by TMSOTf at –50 °C gave the requisite dihydropyran **8** in excellent yield (95%) and diastereoselectivity (dr > 20:1).¹⁰

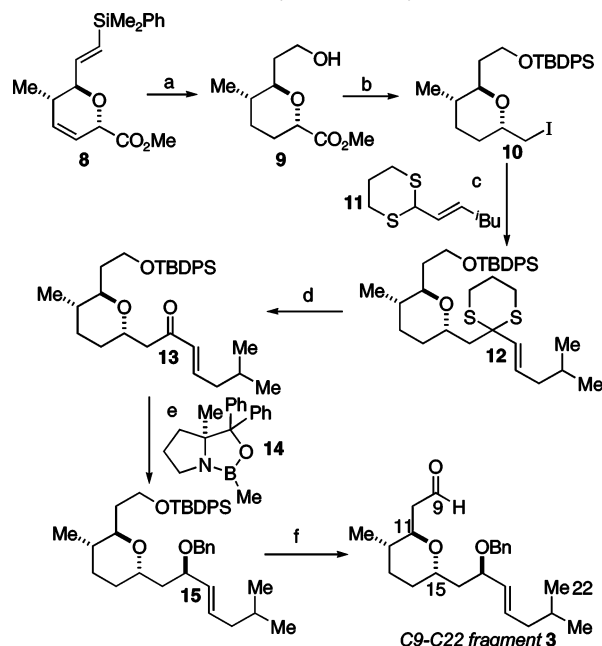
The utility of the annulation is illustrated in the efficient manner in which it installs the C11, C12, and C15 stereo-centers of the target molecule (Scheme 2).

Scheme 2. Construction of the Left-Hand Pyran Subunit



With efficient access to dihydropyran **8**, construction of the aldehyde **3** was initiated (Scheme 3). Elaboration of **8**

Scheme 3. Synthesis of Pyran **3**^a



^a Key: (a) (i) H₂, Pd/C, EtOAc, rt, 99%; (ii) Hg(OAc)₂, CH₃CO₃H, rt, 76%. (b) (i) TBDPSCl, imidazole, DMF, rt; (ii) Dibal-H, CH₂Cl₂, 0 °C; (iii) (PhO)₃P⁺CH₃I[–], DMF, 0 °C, 77% for three steps. (c) **11**, ^tBuLi, THF/HMPA (10:1), 78 → 0 °C, 97%. (d) Dess–Martin periodinane, MeOH/H₂O/THF (8:1:1), rt, 91%. (e) (i) (*S*)-CBS (**14**), BH₃·THF, THF, –20 °C, 80%, (dr > 15:1); (ii) NaH, BnBr, *n*-Bu₄NI, DMF, rt, 90%. (f) (i) TBAF, THF, rt, 99%; (ii) Dess–Martin periodinane, pyridine, CH₂Cl₂, rt, 80%.

began with the hydrogenation of the olefins under the standard conditions (H₂, Pd/C) followed by a Fleming–Tamao oxidation¹¹ of the dimethylphenylsilyl substituent. This was best performed by treatment with Hg(OAc)₂ in

(7) Mukaiyama, T.; Banno, K.; Naraska, N. *J. Am. Chem. Soc.* **1974**, *96*, 7503–7509.

(8) Huang, H.; Panek, J. S. *J. Am. Chem. Soc.* **2000**, *122*, 9836–9837.

(9) Schaus, S. E.; Brandes, B. D.; Larrow, J. F.; Tokunaga, M.; Hansen, K. B.; Gould, A. E.; Furrow, M. E.; Jacobsen, E. N. *J. Am. Chem. Soc.* **2002**, *124*, 1307–1316.

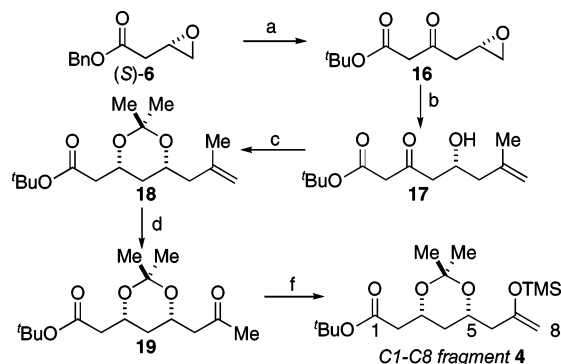
(10) Several other protected 3-hydroxy-propionaldehyde derivatives were surveyed in the annulation with (*S,S*)-**5** all giving inferior yields and diastereoselectivities.

(11) (a) Fleming, I.; Henning, R.; Parker, D. C.; Plaut, H. E.; Sanderson, P. E. J. *J. Chem. Soc., Perkin. Trans. 1* **1995**, 317–337. (b) Tamao, K.; Ishida, N. *J. Organomet. Chem.* **1984**, C37–C39.

peracetic acid giving alcohol **9** in 74% yield for two steps. Protection of the primary alcohol as the *tert*-butyldiphenylsilyl ether, Dibal-H reduction of the methyl ester to the primary alcohol, and subsequent treatment with $(\text{PhO})_3\text{P}^+\text{CH}_3\text{I}^-$ in DMF gave alkyl iodide **10** (77% yield over three steps). Iodide **10** was then alkylated (97%) by treatment with the lithium anion of dithiane **11** in a THF/HMPA (10:1) solvent system.¹² At this stage it was necessary to unveil the α,β -unsaturated ketone **13**. This proved to be challenging, and after an extensive review of conditions that promote the removal of dithiane protecting groups, it was discovered that treatment of **12** with Dess–Martin periodinane in wet methanol for 12 h afforded α,β -unsaturated ketone **13** in good yield.¹³ Subsequent reduction of ketone **13** with a catalytic amount of Corey’s chiral borane, (*S*)-CBS (**14**), in the presence of $\text{BH}_3\cdot\text{SMe}_2$, cleanly installed the C17 stereocenter (80%, dr \approx 15:1).¹⁴ Protection of the emerged allylic alcohol as the benzyl ether (BnBr, *n*-Bu₄NI, DMF, 90%) and deprotection of the primary silyl ether (TBAF, 99%), followed by Dess–Martin oxidation¹⁵ of the resulting primary alcohol (80%), afforded the aldehyde, completing the synthesis of intermediate **3**.

Synthesis of the C1–C8 silyl enol ether **4** began with enantiomerically pure epoxide (*S*)-**6** readily available in multigram quantities from HKR of (\pm) **6** (Scheme 4).

Scheme 4. Synthesis of Silyl Enol Ether **4**^a



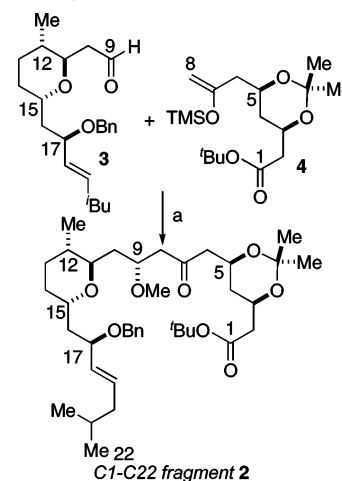
^a Key: (a) *tert*-butyl acetate, LDA, THF, -50°C , 70%. (b) Isopropenylmagnesium bromide, 20 mol % CuI, THF, -50°C , 75%. (c) (i) $\text{Zn}(\text{BH}_4)_2$, CH_2Cl_2 , -78°C , 80%, dr $> 15:1$; (ii) 2,2-dimethoxypropane, cat. *p*-TsOH, rt, 99%. (d) O_3 , MeOH, Me_2S , -78°C , 95%. (e) Lithium tetramethylpiperidine, TMSCl, -78°C , 77%.

Treatment of (*S*)-**6** with the lithium anion of *tert*-butyl acetate in THF at -50°C gave the unstable β -ketoester **16** in 70% yield. Subsequent treatment of **16** with isopropenylmagne-

sium bromide in the presence of 20 mol % CuI provided β -hydroxy ketone **17** (75%). Subsequent treatment of **17** with $\text{Zn}(\text{BH}_4)_2$ ¹⁶ in CH_2Cl_2 at -78°C afforded a 1,3-syn reduction to the 1,3-diol (80%, dr $> 15:1$). This material was then treated with 2,2-dimethoxypropane in the presence of a catalytic amount of *p*-TsOH, which provided acetone **18**. Completion of **4** was accomplished by cleavage of the terminal olefin by ozonolysis giving ketone **19**, which, when treated the bulky lithium anion of 2,2,6,6-tetramethylpiperidine, gave the desired regiochemical enolate, which was then trapped with TMS–Cl, completing the C1–C8 silyl enol ether coupling partner **4**.

To undertake the assembly of the C1–C22 fragment **2**, a Mukaiyama aldol between aldehyde **3** and silyl enol ether **4** was investigated. After a brief survey of the reaction, it was determined that the coupling was best effected by treatment of a mixture of **3** and **4** in CH_2Cl_2 with $\text{BF}_3\cdot\text{OEt}_2$ at -78°C for 4 h (Scheme 5). Gratifyingly, the coupling proceeded in

Scheme 5. Synthesis of the C1–C22 Fragment **2**^a



^a Key: (a) (i) $\text{BF}_3\cdot\text{OEt}_2$, CH_2Cl_2 , -78°C , 81% dr $> 15:1$; (ii) Me_3OBF_4 , Proton Sponge, 4 Å molecular sieves, CH_2Cl_2 , rt, 99%.

good yield (81%) and diastereoselectivity (dr $> 15:1$). Despite the monodentate nature of $\text{BF}_3\cdot\text{OEt}_2$, there is ample precedent for 1,3-anti induction in similar systems.¹⁷ The absolute stereochemistry at C9 was unambiguously assigned using the method of Mosher.¹⁸ The methylation of the C9 hydroxyl was then accomplished by treatment with Meerwein’s reagent and Proton sponge (99%) giving the C1–C22 fragment **2**.¹⁹

(15) (a) Dess, D. B.; Martin, J. C. *J. Org. Chem.* **1983**, *48*, 4155–4156. (b) Dess, D. B.; Martin, J. C. *J. Am. Chem. Soc.* **1991**, *113*, 7277–7287.

(16) For reviews on the synthetic applications of zinc borohydride, see: (a) Narasimhan, S.; Balakumar, R. *Aldrichimica Acta* **1998**, *31*, 19–27. (b) Hoyveda, A. H.; Evans, D. A.; Fu, G. C. *Chem. Rev.* **1993**, 1307–1370. (c) Evans, D. A.; Kim, A. S.; Metternich, R.; Novack, V. J. *J. Am. Chem. Soc.* **1998**, *120*, 5921–5942.

(17) (a) Evans, D. A.; Duffy, J. L.; Dart, M. J. *Tetrahedron Lett.* **1994**, 35, 8537–8540. (b) Paterson, I.; Smith, J. D. *J. Org. Chem.* **1992**, *57*, 3261–3264.

(18) (a) Dale, J. A.; Mosher, H. S. L. *J. Am. Chem. Soc.* **1973**, *96*, 512–519. (b) Trost, B. M.; Belletire, J. L.; Godleski, S.; McDougal, D. G.; Balkovec, J. M.; Baldwin, J. J.; Christy, M. E.; Ponticello, G. S.; Varga, S. L.; Springer, J. P. *J. Org. Chem.* **1986**, *51*, 2370–2374.

(12) For leading references on dithiane alkylation, see: Smith, A. B.; Boldi, A. M. *J. Am. Chem. Soc.* **1997**, *119*, 6925–6926. and references therein.

(13) For an initial communication and discussion of this reaction, see: Langille, N. F.; Dakin, L. A.; Panek, J. S. *Org. Lett.* **2003**, *4*, 575–578.

(14) (a) Corey, E. J.; Bakshi, R. K.; Shibata, S. *J. Am. Chem. Soc.* **1987**, *109*, 5551–5553. (b) Corey, E. J.; Bakshi, R. K. *Tetrahedron Lett.* **1990**, *31*, 611–614. (c) Corey, E. J.; Helal, C. J. *Angew. Chem., Int. Ed.* **1998**, *37*, 1986–2012.

In conclusion, we have described a convergent synthesis to an advanced intermediate of leucascandrolide A. The approach features an enantioselective synthesis of the C9–C22 fragment **3** through the use of a formal [4 + 2] annulation between chiral crotylsilane (*S,S*)-**5** and aldehyde **6** and a 1,3-anti diastereoselective Mukaiyama aldol coupling between fragments **3** and **4**, which completed the carbon framework of the macrocycle of leucascandrolide A. Studies toward the completion of leucascandrolide A are underway and will be reported at a later time.

(19) Meerwein, H.; Hinz, G.; Hofmann, P.; Kroning, E.; Pfeil, E. *J. Prakt. Chem.* **1937**, *147*, 257.

Acknowledgment. The authors are grateful to Mr. Hongbing Huang for helpful discussions. Financial support was obtained the NIH-NCI (CA56404) and NIH (P50 GM067041).

Supporting Information Available: General experimental procedures, including spectroscopic characterization of novel compounds. This material is available free of charge via the Internet at <http://pubs.acs.org>.

OL035581M