

Thiol-ene Click Reaction as a General Route to Functional Trialkoxysilanes for Surface Coating Applications

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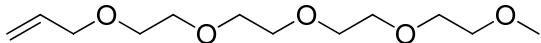
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1. General information

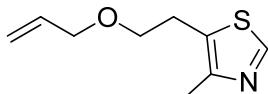
Ferric chloride hexahydrate ($\text{FeCl}_3 \cdot 6\text{H}_2\text{O}$), ferrous chloride tetrahydrate ($\text{FeCl}_2 \cdot 4\text{H}_2\text{O}$), ammonium hydroxide (NH_4OH), oleic acid, sodium sulfate (Na_2SO_4), magnesium sulfate (MgSO_4), hydrochloric acid (HCl), and sodium hydroxide (NaOH) were purchased from Fisher Scientific (Pittsburgh, PA). 3-mercaptopropyltriethoxysilane (MPTES), 3-mercaptopropyltrimethoxysilane (MPTMS), allyltriethoxysilane (ATES) were purchased at >95% purity from Gelest Inc. (Morrisville, PA). Diisopropylethylamine, potassium *tert*-butoxide (*t*-BuOK), and tetraethyleneglycol monomethyl ether were purchased from Acros Organics. Allyl bromide, 5-(2-hydroxyethyl)-4-methylthiazole, 10-undecenoic acid, di-*tert*-butyl dicarbonate, *N,N*-dimethylaminepyridine, *N*-Boc-trans-4-hydroxy-L-proline methyl ester, 4-penteneoic anhydride, 3-aminoquinuclidine dihydrochloride, dicyclohexylcarbodiimide (DCC), L-serine-*tert*-butyl ester, *tert*-butyl bromoacetate, benzyl hydrosulfide, 2,2-dimethoxy-1,2-diphenylethanone, Celite® 500 fine, and all other alkene substrates were purchased from Sigma Aldrich (St. Louis, MO). All solvents were HPLC grade and purchased from either EMD Biosciences, Inc. (Gibbstown, NJ) or Fisher. Dry THF, dioxane, and toluene were prepared by mixing the solvent with MgSO_4 and filtering into hot dry glassware just prior to use. All other solvents and chemicals were used as received.

Hand-held rare-earth magnets, recycled from computer hard drives, were used to separate magnetic nanoparticles from solution. Sonication was performed using a Branson 2510 ultrasonicator. Fourier transform infrared spectroscopy (FT-IR) was performed using pressed KBr pellets, or thin liquid films between AgCl plates on a Jasco FT/IR-420. Elemental analysis was performed on a Flash 1112 series elemental analyzer (Thermo Scientific). Magnetic measurements of particles were made using a Quantum Design MPMS 5XL Super Quantum Interference Device (SQUID) magnetometer. Transmission electron microscopy (TEM) images were obtained with an 80-300 kV tunable Titan TEM/STEM. ^1H , ^{13}C NMR spectra were obtained using either a Bruker AV300 spectrometer, or Bruker ARX400 spectrometer with a sample changer and CDCl_3 or CD_3CN as solvents. High resolution mass spectra were acquired on a Waters LCT Premier XE time-of-flight mass spectrometer; gas chromatography/mass spectrometry (GC/MS) was performed on a Agilent 6890-5975 GC-MS with autosampler, equipped with an HP-5 column.

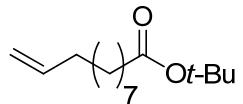
2. Alkene precursor synthesis



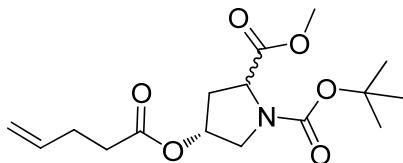
2,5,8,11,14-Pentaoxaheptadec-16-ene. An oven-dried 100 mL round bottom flask was charged with *t*-BuOK (13 mmol), dry THF (30 mL), purged with argon and cooled to 0°C in an ice bath. Tetraethyleneglycol monomethyl ether (10 mmol) was then added dropwise to the solution and the reaction stirred at 0°C for 1 h under argon. Allyl bromide (13 mmol) was added dropwise, and the solution was warmed to R.T. and allowed to stir for 24 h under argon. Deionized water (2 mL) was then added and the reaction stirred for 10 min. All liquid were then removed *in vacuo* and the residue redissolved in ethyl acetate (50 mL). The organic was washed three times with water (20 mL), dried over Na₂SO₄, filtered over Celite and then concentrated *in vacuo* to yield clear/light yellow oil (47%).
¹H NMR (400 MHz, CDCl₃): δ = 5.90 (m, 1H), 5.23 (dd, 1H, ³J_{HHtrans} = 17.3 Hz, , ²J_{HHgem} = 1.7 Hz), 5.15 (dd, 1H, ³J_{HH} = 10.4 Hz, , ²J_{HHgem} = 1.5 Hz), 4.00 (dd, 2H, ³J_{HH} = 5.89 Hz, , ⁴J_{HH} = 1.5 Hz), 3.65 (m, 12H), 3.59 (m, 2H), 3.53 (m, 2H), 3.36 (s 3H) ppm.
¹³C NMR (400 MHz, CDCl₃): δ = 134.8, 117.0, 72.2, 71.9, 70.6, 70.6, 70.6, 70.5, 69.4, 59.0 ppm. HRMS-ESI: Calculated [M+Na]⁺: 271.1516; found: 271.1516.



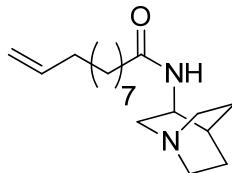
5-(2-(Allyloxy)ethyl)-4-methylthiazole. An oven-dried 100 mL round bottom flask was charged with *t*-BuOK (20.8 mmol), dry THF (25 mL), purged with argon and cooled to 0°C in an ice bath. 5-(2-Hydroxyethyl)-4-methylthiazole (16 mmol) was then added dropwise to the solution and the reaction stirred at 0 °C for 1 h under argon. Allyl bromide (20.8 mmol) was added dropwise, and the solution was warmed to R.T. and allowed to stir for 20 h under argon. Deionized water (4 mL) was then added and the reaction stirred for 10 min. All liquid were then removed *in vacuo* and the residue redissolved in ethyl acetate (50 mL). The organic phase was washed three times with water (20 mL), dried over Na₂SO₄, filtered over Celite, and then concentrated *in vacuo* to yield a yellow oil (96%).
¹H NMR (400 MHz, CDCl₃): δ = 8.52 (s, 1H), 5.89 (m, 1H), 5.23 (dd, 1H, ³J_{HHtrans} = 17.4 Hz, ²J_{HHgms} = 1.8 Hz), 5.13 (dd, 1H, ³J_{HHcis} = 10.3 Hz, ²J_{HHgms} = 1.6 Hz), 3.96 (dt, 2H, ³J_{HH} = 5.7 Hz, ⁴J_{HH} = 1.4 Hz), 3.57 (t, 2H, ³J_{HH} = 6.9 Hz), 2.99 (t, 2H, ³J_{HH} = 6.9 Hz), 2.37 (s, 3H) ppm.
¹³C NMR (400 MHz, CDCl₃): δ = 149.6, 149.2, 134.5, 127.9, 17.0, 71.9, 70.0, 27.0, 14.9 ppm. HRMS-ESI: Calculated [M+H]⁺: 184.0791; found: 184.0787.



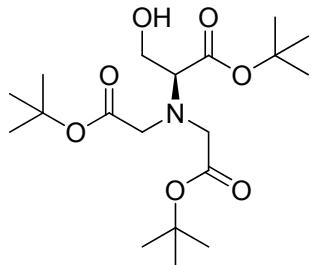
tert-Butyl undec-10-enoate. An oven-dried 50 mL round bottom flask was charged with 10-undecenoic acid (5 mmol), di-*tert*-butyl dicarbonate (10 mmol) and *tert*-butanol (10 mL). *N,N*-dimethylaminopyridine (1.5 mmol) was added to the stirred mixture and the reaction was allowed to stir for 20 h at R.T.. The reaction mixture was concentrated *in vacuo* and the residue purified by silica gel chromatography using ethyl acetate/hexanes (2:8) to afford the product as a light yellow oil (58%). $R_f = 0.98$. ^1H NMR (400 MHz, CDCl_3): $\delta = 5.81$ (m, 1H), 4.96 (dq, 1H, $^3J_{HH} = 17.1$ Hz, $^4J_{HH} = 1.6$ Hz), 4.91 (dq, 1H, $^3J_{HH} = 10.2$ Hz, $^4J_{HH} = 1.2$ Hz), 2.19 (t, 2H, $^3J_{HH} = 7.7$ Hz), 2.02 (q, 2H, $^3J_{HH} = 7.0$ Hz), 1.28-1.56 (m, 21H) ppm. ^{13}C NMR (400 MHz, CDCl_3): $\delta = 173.3$, 139.2, 114.1, 79.9, 35.6, 33.8, 29.3, 29.2, 29.0, 28.9, 28.1, 25.1 ppm. GC/MS (EI): Calculated [M-C₄H₉]⁺: 184.1; found: 184.1.



N-Boc 2-methyl (4R)-4-(pent-4-enyloxy)pyrrolidine-1,2-dicarboxylate. An oven-dried 50 mL round bottom flask was charged with *N*-Boc-trans-4-hydroxy-L-proline methyl ester (1 mmol), *N,N*-dimethylaminopyridine (0.15 mmol), pyridine (3 mmol), dry acetonitrile (5 mL), and placed under an argon atmosphere. 4-Pentenoic anhydride (3 mmol) was then added dropwise and the reaction was warmed to 50 °C and stirred under argon for 44 h. The reaction mixture was then concentrated *in vacuo*, and the residue dissolved in THF (5 mL), deionized water (3 mL), pyridine (0.5 mL) and allowed to stir an additional 24 h at 50 °C. The reaction mixture was then concentrated *in vacuo*, the residue dissolved in DCM (20 mL), and the DCM washed 1 M HCl (15 mL), water (15 mL), 1 M NaOH (15 mL), and finally with water (15 mL). The organic was dried over Na_2SO_4 , filtered over Celite, and then concentrated *in vacuo* to yield a yellow oil (71%). Diastereomeric mixture (59:41). ^1H NMR (400 MHz, CDCl_3): $\delta = 5.80$ (m, 1H), 5.29 (m, 1H), 5.03 (m, 2H), 4.39 (dt, 1H, $^3J_{HH} = 8.7$ Hz), 3.71 (s, 3H), 3.70-3.05 (m, 2H), 2.37 (m, 5H), 2.17 (m, 1H), 1.46 (s, 3H), 1.41 (s, 6H) ppm. ^{13}C NMR (400 MHz, CDCl_3): $\delta = 172.9$, 172.7, 172.3, 172.1, 154.0, 153.3, 136.2, 115.6, 80.3, 72.6, 71.8, 57.8, 57.4, 52.2, 52.0, 51.9, 36.5, 35.5, 33.2, 30.2, 29.6, 28.7, 28.2, 28.1 ppm. HRMS-ESI: Calculated [M+Na]⁺: 350.1574; found: 350.1568.

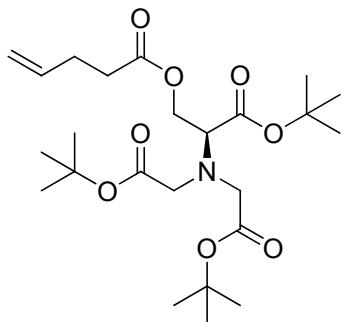


N-(quinuclidin-3-yl)undec-10-enamide. An oven-dried 50 mL round bottom flask was charged with 3-aminoquinuclidine dihydrochloride (2 mmol), 10-undecenoic acid (2 mmol), *N,N*-dimethylaminopyridine (0.3 mmol), diisopropylethylamine (4 mmol), and dry DMF (15 mL). The flask was purged with argon and cooled to 0°C in an ice bath. Dicyclohexycarbodiimide (2.2 mmol) was then added to the solution, the flask repurged with argon, and the mixture warmed to R.T. and stirred for 46 h. The reaction mixture was concentrated *in vacuo* and 30 ml diethyl ether added to the residue. The solid was filtered off over Celite and the organic phase concentrated *in vacuo*. The residue was purified by silica gel chromatography using ethyl acetate/hexanes (3:7) to afford the product as a white solid (23%). $R_f = 0.44$. ^1H NMR (400 MHz, CDCl_3): δ = 5.73 (m, 1H), 4.93 (dd, 1H, $^3J_{HH} = 17.0$ Hz, $^4J_{HH} = 1.5$ Hz), 4.87 (d, 1H, $^3J_{HH} = 10.3$ Hz), 3.71(m, 1H), 2.10 (t, 2H, $^3J_{HH} = 7.8$ Hz), 1.97 (q, 2H, $^3J_{HH} = 7.3$ Hz), 1.85 (m, 2H), 1.54-1.68 (m, 5H), 1.02-1.34 (m, 15H) ppm. ^{13}C NMR (400 MHz, CDCl_3): δ = 172.2, 139.1, 114.1, 48.6, 48.0, 37.0, 34.0, 33.7, 33.2, 29.3, 29.2, 29.0, 28.8, 25.9, 25.7, 25.5, 25.0, 24.9 ppm. HRMS-ESI: Calculated [M-C₂H₂]⁺: 266.2358; found: 266.2474.

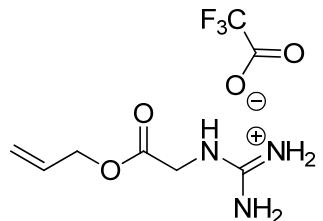


(S)-tert-Butyl-2-di(tert-butoxycarbonylmethyl)amino-3-hydroxypropanoate. This material was prepared following a previously reported protocol (Meunier, S.; Cristau, P.; Taran, F. *Synth. Commun.* (2005), 35,2415-2425). An oven-dried 100 mL round bottom flask was charged with L-serine *tert*-butyl ester hydrochloride (4 mmol), diisopropylethylamine (24 mmol), and acetonitrile (25 mL). *tert*-butyl bromoacetate (16 mmol) was then added dropwise and the entire mixture refluxed in air for 24 h. The reaction mixture was concentrated *in vacuo* and ethyl acetate (40 mL) added and the resulting precipitate filtered off over Celite. The ethyl acetate was then concentrated *in vacuo* and the residue purified by silica gel chromatography using ethyl acetate/hexanes (2:8) to afford the product as a light yellow solid (68%). $R_f = 0.56$. ^1H NMR (400 MHz, CDCl_3): δ = 4.21 (d, 1H, $^3J_{HH} = 10.9$ Hz), 3.69 (td, 1H, $^2J_{HH} = 10.8$ Hz, $^3J_{HH}$

= 13.6 Hz), 3.42 (m, 6H), 1.40 (s, 27H) ppm. $^{13}\text{CNMR}$ (400 MHz, CDCl_3): δ = 171.3, 107.2, 81.7, 81.3, 68.2, 60.1, 54.9, 28.1, 28.0 ppm. HRMS-ESI: Calculated $[\text{M}+\text{H}]^+$: 390.2488; found: 390.2470.



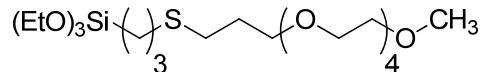
(S)-tert-Butyl-2-di(tert-butyloxycarbonylmethyl)amino-3-(pent-4-enyloxy)-propanoate. An oven-dried 50 mL round bottom flask was charged with (S)-*tert*-butyl-2-di(*tert*-butyloxycarbonylmethyl)amino-3-hydroxypropanoate (1 mmol), *N,N*-dimethylaminopyridine (0.3 mmol), diisopropylethylamine (3 mmol), acetonitrile (5 mL), and purged with argon. 4-pentenoic anhydride (3 mmol) was then slowly added dropwise and the reaction mixture stirred under agron at 55 °C for 42 h. The reaction mixture was concentrated *in vacuo* and the residue dissolved in THF (5 mL), deionized water (1 mL), pyridine (0.5 mL), capped, and stirred at 55 °C for 24 h. Solvents were concentrated *in vacuo* and the residue purified by silica gel chromatography using ethyl acetate/hexanes (1:9) to afford the product as a light yellow oil (96%). R_f =0.49. $^1\text{HNMR}$ (400 MHz, CDCl_3): δ = 5.80 (m, 1H), 5.05 (dq, 1H, $^3J_{HH}$ = 17.2 Hz, $^4J_{HH}$ = 1.6 Hz), 4.97 (dq, 1H, $^3J_{HH}$ = 10.2 Hz, $^4J_{HH}$ = 1.4 Hz), 4.42 (dd, 1H, $^2J_{HH}$ = 10.9 Hz, $^3J_{HH}$ = 5.9 Hz), 4.29 (dd, 1H, $^2J_{HH}$ = 11.2 Hz, $^3J_{HH}$ = 6.7 Hz), 3.69 (t, 1H, $^3J_{HH}$ = 5.6 Hz), 3.55 (s, 2H), 3.54 (s, 2H), 2.36 (m, 4H), 1.45 (s, 9H), 1.44 (s, 18H) ppm. $^{13}\text{CNMR}$ (400 MHz, CDCl_3): δ = 172.4, 170.6, 169.6, 136.6, 115.5, 81.7, 80.8, 63.8, 63.5, 54.3, 33.4, 28.7, 28.1 ppm. HRMS-ESI: Calculated $[\text{M}+\text{H}]^+$: 472.2905; found: 472.2924.



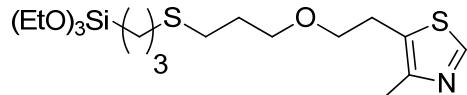
((2-(allyloxy)-2-oxoethyl)amino)(amino)methaniminium 2,2,2-trifluoroacetate. An oven-dried 100 mL round bottom flask was charged with guanidineacetic acid (4.32 mmol), trifluoroacetic acid (20 mL), and allyl alcohol (15 mL) and the reaction mixture refluxed 48 h. The reaction mixture was concentrated *in vacuo* and the residue purified by silica gel chromatography chloroform/methanol (7:3) to afford the product as a viscous yellow oil (52%). $R_f = 0.82$. ^1H NMR (400 MHz, CD_3CN): $\delta = 7.79$ (bt, 1H, $^3J_{HH} = 5.3$ Hz), 7.15 (bs, 1H), 5.90 (m, 1H), 5.33 (dd, 1H, $^3J_{HH} = 17.3$ Hz, $^4J_{HH} = 1.4$ Hz), 5.23 (dd, 1H, $^3J_{HH} = 10.3$ Hz, $^4J_{HH} = 1.4$ Hz), 4.61 (d, 2H, $^3J_{HH} = 5.6$ Hz), 4.01 (d, 2H, $^3J_{HH} = 6.1$ Hz) ppm. ^{13}C NMR (400 MHz, CD_3CN): $\delta = 169.6, 159.2, 159.2, 159.1, 132.9, 132.8, 132.8, 119.0, 67.1, 67.0, 66.9, 43.7, 43.6$ ppm. ^{13}C NMR (400 MHz, CD_3CN): $\delta = -77.5$ ppm. HRMS-ESI: Calculated [M] $^+$: 158.0930; found: 158.0927.

3. Triethoxysilane synthesis

General procedure. A cooled oven-dried 10-mL round bottom was charged with MPTES (1 equiv.), alkene (1 equiv.) and 2 mol% 2,2-dimethoxy-1,2-diphenylethanone (0.02 equiv.), known commercially as Irgacure®651. The reaction was run neat unless the alkene did not fully dissolve, in which case a tiny amount of dry chloroform or methanol was added. The reaction mixture was then capped with a septum and purged briefly with argon or nitrogen. The flask was placed next to an 15 W, 18"-long blacklight having a total UV output of 2.6 W and $\lambda_{\text{max}} = 368$ nm. The flask was positioned so that one side rested against the center of the bulb. The UV flux at $\lambda_{\text{max}} = 368$ nm was measured to be 30% attenuated after passing through standard thickness Pyrex glass beaker. Both the flask and blacklight were wrapped in aluminum foil and the reaction mixture irradiated on average for 24 h with gentle stirring to yield product silanes in high purity. If solvent was added to the reaction, it was removed *in vacuo* upon completion of the reaction. Product conversion and purity were determined by ^1H NMR or GC/MS.

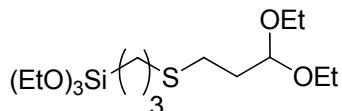


PEG triethoxysilane (1). Following the general procedure above, MPTES (2 mmol), 2,5,8,11,14-pentaoxaheptadec-16-ene (2 mmol), and Irgacure®651 (0.04 mmol) were mixed, purged with argon, and irradiated 24 h (>99% conversion, 96% pure). ^1H NMR (400 MHz, CDCl_3): $\delta = 3.78$ (q, 2H, $^3J_{HH} = 6.7$ Hz), 3.35-3.65 (m, 18H), 3.37 (s, 3H), 2.56 (t, 2H, $^3J_{HH} = 7.6$ Hz), 2.52 (t, 2H, $^3J_{HH} = 7.2$ Hz), 1.82 (m, 2H), 1.68 (m, 2H), 0.726 (t, $^3J_{HH} = 8.5$ Hz) ppm. ^{13}C NMR (400 MHz, CDCl_3): $\delta = 70.6, 70.6, 70.6, 70.6, 70.5, 70.5, 70.2, 69.8, 59.0, 58.4, 35.2, 29.7, 28.6, 23.2, 18.3, 9.9$ ppm. HRMS-ESI: Calculated [M+Na] $^+$: 509.2575; found: 509.2552.

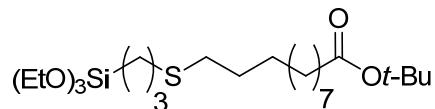


Thiazole triethoxysilane (2). Following the general procedure above, MPTES (5 mmol), 5-(2-(allyloxy)ethyl)-4-methylthiazole (5 mmol), and Irgacure®651 (0.1 mmol) were mixed, purged with argon, and irradiated 24 h (>99% conversion, 95% pure). ^1H NMR (400 MHz, CDCl_3): $\delta = 8.54$ (s, 1H), 3.78 (q, 6H, $^3J_{HH} = 7.2$ Hz), 3.56 (t, 2H, $^3J_{HH} = 6.6$ Hz), 3.50 (t, 2H, $^3J_{HH} = 6.1$ Hz), 2.98 (t, 2H, $^3J_{HH} = 6.4$ Hz), 2.56 (t, 2H, $^3J_{HH} = 7.2$ Hz), 2.50 (t, 2H, $^3J_{HH} = 7.2$ Hz), 2.37 (s, 3H), 1.14 (t, 9H, $^3J_{HH} = 6.7$ Hz), 0.71 (t, 2H, $^3J_{HH} = 7.7$ Hz)

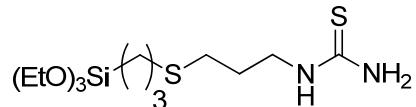
Hz) ppm. $^{13}\text{CNMR}$ (400 MHz, CDCl_3): δ = 149.6, 149.1, 128.0, 70.6, 69.5, 58.4, 35.1, 29.8, 28.6, 27.0, 23.0, 18.3, 14.9, 9.9 ppm. HRMS-ESI: Calculated $[\text{M}+\text{H}]^+$: 422.1850; found: 422.1850.



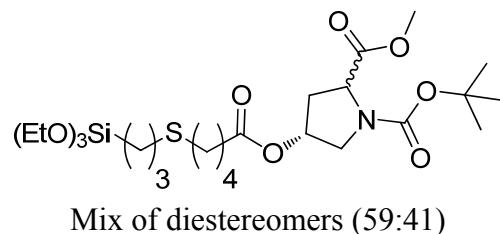
Diethyl acetal triethoxysilane (3). Following the general procedure above, MPTEs (2 mmol), acrolein diethyl acetal (2 mmol), and Irgacure®651 (0.04 mmol) were mixed, purged with argon, and irradiated 24 h (>99% conversion, 96% pure). $^1\text{HNMR}$ (400 MHz, CDCl_3): δ = 4.95 (t, 1H, $^3J_{HH}$ = 5.6 Hz), 3.80 (q, 6H, $^3J_{HH}$ = 7.1 Hz), 3.64 (m, 2H), 3.49 (m, 2H), 2.53 (q, 4H, $^3J_{HH}$ = 6.6 Hz), 1.88 (m, 2H), 1.70 (m, 2H), 1.20 (m, 15H), 0.72 (t, 2H, $^3J_{HH}$ = 8.1 Hz) ppm. $^{13}\text{CNMR}$ (400 MHz, CDCl_3): δ = 101.17, 61.5, 58.4, 35.2, 33.9, 27.1, 23.2, 18.3, 15.3, 9.9 ppm. HRMS-ESI: Calculated $[\text{M}+\text{Na}]^+$: 391.1945; found: 391.1932.



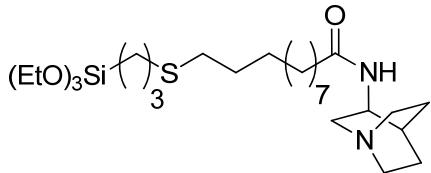
tert-Butyl undecanoate triethoxysilane (4). Following the general procedure above, MPTEs (2 mmol), *tert*-butyl undec-10-enoate (2 mmol), and Irgacure®651 (0.04 mmol) were mixed, purged with argon, and irradiated 24 h (>98% conversion, 96% pure). $^1\text{H NMR}$ (400 MHz, CDCl_3): δ = 3.81 (q, 6H, $^3J_{HH}$ = 6.9 Hz), 2.52 (t, 2H, $^3J_{HH}$ = 7.2 Hz), 2.48 (t, 2H, $^3J_{HH}$ = 7.2 Hz), 2.19 (t, 2H, $^3J_{HH}$ = 7.6 Hz), 1.68 (m, 2H), 1.56 (m, 4H), 1.43 (s, 9H), 1.36 (m, 12H), 1.22 (t, 9H, $^3J_{HH}$ = 7.2 Hz), 0.73 (t, 2H, $^3J_{HH}$ = 8.4 Hz) ppm. $^{13}\text{CNMR}$ (400 MHz, CDCl_3): δ = 173.3, 79.9, 58.4, 35.6, 35.2, 32.0, 29.8, 29.5, 29.4, 29.3, 29.2, 29.1, 29.0, 28.1, 25.1, 23.3, 18.3, 9.9 ppm. HRMS-ESI: Calculated $[\text{M}+\text{Na}]^+$: 501.3046; found: 501.3050.



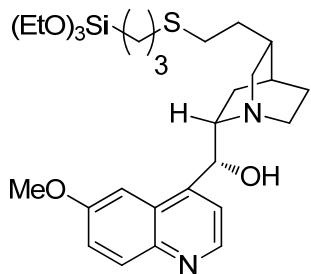
Thiourea triethoxysilane (5). Following the general procedure above, MPTES (2 mmol), *N*-allylthiourea (2 mmol), Irgacure®651 (0.04 mmol), and dry MeOH (0.6 mL) were mixed, purged with argon, and irradiated 24 h (>99% conversion, 95% pure). ¹HNMR (400 MHz, CDCl₃): δ = 6.61 (bs, 1H), 5.94 (bs, 1H), 3.77 (q, 6H, ³J_{HH} = 6.6 Hz), 3.51 (bs, 1H), 2.51 (q, 4H, ³J_{HH} = 7.3 Hz), 2.16 (s, 2H), 1.77 (m, 2H), 1.62 (m, 2H), 1.59 (t, 9H, ³J_{HH} = 7.4 Hz), 0.68 (t, 2H, ³J_{HH} = 8.2 Hz) ppm. ¹³CNMR (400 MHz, CDCl₃): δ = 182.9, 58.2, 34.7, 28.9, 22.9 ppm. HRMS-ESI: Calculated [M+Na]⁺: 377.1365; found: 377.1368.



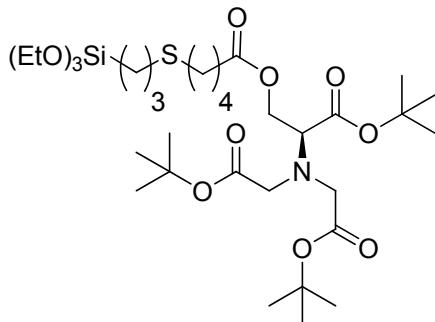
N-Boc-(4R)-4-hydroxy-L-proline methyl ester triethoxysilane (6). Following the general procedure above, MPTES (0.66 mmol), *N*-Boc 2-methyl (4R)-4-(pent-4-enyloxy)pyrrolidine-1,2-dicarboxylate (0.66 mmol), and Irgacure®651 (0.013 mmol) were mixed, purged with argon, and irradiated 24 h (96% conversion, 94% pure). ¹HNMR (400 MHz, CDCl₃): δ = 5.28 (s, 1H), 4.34 (dt, 1H, ³J_{HH} = 8.3 Hz), 3.82 (q, 6H, ³J_{HH} = 6.9 Hz), 3.49-3.74 (m, 5H), 2.51 (q, 4H, ³J_{HH} = 7.3 Hz), 2.36 (m, 1H 2.32 (t, 2H, ³J_{HH} = 7.7 Hz), 2.19, (m, 1H), 1.68 (m, 6H), 1.45 (s, 3H), 1.41 (s, 6H), 1.22 (t, 9H, ³J_{HH} = 7.2 Hz), 0.73 (t, 2H, ³J_{HH} = 8.0) ppm. ¹³CNMR (400 MHz, CDCl₃): δ = 173.1, 172.8, 172.7, 154.2, 153.5, 80.5, 72.5, 71.8, 58.4, 57.9, 57.5, 52.3, 52.2, 52.1, 52.0, 36.6, 35.6, 35.1, 33.8, 31.4, 29.0, 28.3, 28.2, 24.0, 23.2, 18.3, 9.9 ppm. HRMS-ESI: Calculated [M+Na]⁺: 588.2639; found: 588.2649.



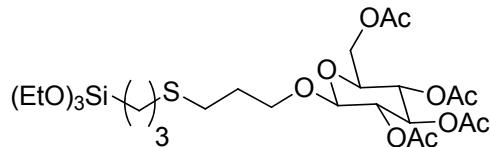
Quinuclidine triethoxysilane (7). Following the general procedure above, MPTES (0.41 mmol), *N*-(quinuclidin-3-yl)undec-10-enamide (0.41 mmol), Irgacure®651 (0.008 mmol), and dry CHCl₃ (0.5 mL) were mixed, purged with argon, and irradiated 24 h (>98% conversion, 96% pure). ¹H NMR (400 MHz, CDCl₃): δ = 5.34 (s, 1H), 3.80 (m, 6H), 2.52 (t, 2H, ³J_{HH} = 7.5 Hz), 2.48 (t, 2H, ³J_{HH} = 7.5 Hz), 2.11 (t, 2H, ³J_{HH} = 7.8 Hz), 1.57-1.89 (m, 12H), 1.23 (m, 26H), 0.73 (t, 2H, ³J_{HH} = 8.2 Hz) ppm. ¹³C NMR (400 MHz, CDCl₃): δ = 172.1, 129.1, 77.2, 67.1, 58.4, 58.2, 48.0, 37.1, 35.2, 33.3, 32.0, 29.8, 29.5, 29.4, 29.3, 29.2, 29.0, 25.9, 25.6, 24.9, 23.2, 18.3, 9.9 ppm. HRMS-ESI: Calculated [M-C₃H₆NO]⁺: 458.3124; found: 458.3134.



Quinine triethoxysilane (8). Following the general procedure above, MPTES (1 mmol), quinine (1 mmol), Irgacure®651 (0.02 mmol), and dry CHCl₃ (1 mL) were mixed, purged with argon, and irradiated 27 h (>98% conversion, 96% pure). ¹H NMR (400 MHz, CDCl₃): δ = 8.63 (d, 1H, ³J_{HH} = 4.6 Hz), 7.95 (d, 1H, ³J_{HH} = 9.2 Hz), 7.47 (d, 1H, ³J_{HH} = 5.1 Hz), 7.31 (dd, 1H, ³J_{HH} = 9.2 Hz, ⁴J_{HH} = 2.6 Hz), 7.22 (d, 1H, ⁴J_{HH} = 2.0 Hz), 5.51 (s, 1H), 3.98 (s, 3H), 3.78 (q, 6H, ³J_{HH} = 7.1 Hz), 3.58 (bs, 1H), 3.42 (bs, 1H), 3.06 (m, 2H), 2.65 (m, 1H), 2.45 (t, 2H, ³J_{HH} = 7.1 Hz), 2.39 (t, 2H, ³J_{HH} = 7.1 Hz), 1.42-1.75 (m, 10H), 1.19 (t, 9H, ³J_{HH} = 6.6 Hz), 0.67 (t, 2H, ³J_{HH} = 8.2 Hz) ppm. ¹³C NMR (400 MHz, CDCl₃): δ = 157.8, 147.6, 144.3, 131.6, 126.6, 121.5, 118.4, 117.5, 101.3, 77.2, 72.0, 59.8, 58.4, 43.2, 35.2, 34.7, 34.7, 30.0, 28.2, 25.7, 23.2, 21.7, 18.3, 9.9 ppm. HRMS-ESI: Calculated [M+H]⁺: 563.2969; found: 563.2977.



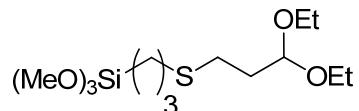
Nitrilotriacetate (NTA) triethoxysilanesilane (9). Following a similar procedure to the one described above, MPTES (0.5 mmol), (S)-*tert*-butyl-2-di(*tert*-butyloxycarbonylmethyl)amino-3-(pent-4-enoyloxy)propanoate (0.5 mmol), and Irgacure®651 (0.015 mmol, 3 mol %) were mixed, purged with argon, and then gently warmed in an oil bath until the Irgacure®651 dissolved. The mixture was then irradiated 45 h. Following, an additional amount of Irgacure®651 (0.01 mmol, 2 mol%) was added to the reaction, the flask purged, and the mixture irradiated an additional 22 h (94% conversion, 90% pure). ^1H NMR (400 MHz, CDCl_3): δ = 4.41 (dd, 1H, $^2J_{HH}$ = 10.9 Hz, $^3J_{HH}$ = 5.8 Hz), 4.27 (dd, 1H, $^2J_{HH}$ = 10.9 Hz, $^3J_{HH}$ = 6.9 Hz), 3.81 (q, 6H, $^3J_{HH}$ = 6.9 Hz), 3.68 (t, 1H, $^3J_{HH}$ = 6.6 Hz), 3.54 (d, 3H, $^4J_{HH}$ = 2.8 Hz), 2.50 (m, 4H), 2.30 (t, 2H, $^3J_{HH}$ = 7.5 Hz), 1.57-1.71 (m, 6H), 1.45 (m, 28H), 1.21 (t, 9H, $^3J_{HH}$ = 7.1 Hz), 0.72 (t, 2H, $^3J_{HH}$ = 8.3 Hz) ppm. ^{13}C NMR (400 MHz, CDCl_3): δ = 172.7, 170.5, 169.6, 81.8, 80.8, 63.8, 63.4, 58.4, 54.2, 35.1, 33.6, 31.5, 29.1, 28.1, 24.0, 23.2, 18.3, 9.9 ppm. HRMS-ESI: Calculated [M+H] $^+$: 710.3964; found: 710.3981 ppm.



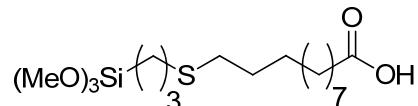
Tetra-*O*-acetyl- β -D-glucopyranoside triethoxysilane (10). Following the general procedure above, MPTES (1 mmol), allyl-*O*-acetyl- β -D-glucopyranoside (1 mmol), Irgacure®651 (0.02 mmol), and dry CHCl_3 (0.6 mL) were mixed, purged, and irradiated 25 h (>99% conversion, 97% pure). ^1H NMR (400 MHz, CDCl_3): δ = 5.30 (s, 2H), 5.20 (t, 1H, $^3J_{HH}$ = 8.9 Hz), 5.07 (t, 1H, $^3J_{HH}$ = 9.5 Hz), 4.97 (t, 1H, $^3J_{HH}$ = 8.9 Hz), 4.50 (d, 1H, $^3J_{HH}$ = 8.9 Hz), 4.28 (dd, 1H, $^3J_{HH}$ = 12.5 Hz, $^3J_{HH}$ = 4.8 Hz), 4.14 (dd, 1H, $^3J_{HH}$ = 12.5 Hz, $^3J_{HH}$ = 2.4 Hz), 3.70 (dt, 1H, $^3J_{HH}$ = 10.1 Hz, $^3J_{HH}$ = 5.4 Hz), 3.82 (q, 6H, $^3J_{HH}$ = 7.2 Hz), 3.62 (m, 2H), 2.53 (q, 4H, $^3J_{HH}$ = 7.7 Hz), 2.08 (s, 3H), 2.04 (s, 3H), 2.02 (s, 3H), 2.00 (s, 3H), 1.80 (m, 2H), 1.67 (m, 2H), 1.22 (t, 9H, $^3J_{HH}$ = 7.1 Hz) 0.73 (t, 2H, $^3J_{HH}$ = 8.3 Hz) ppm. ^{13}C NMR (400 MHz, CDCl_3): δ = 170.7, 170.3, 169.4, 169.3, 101.0, 72.9, 71.8, 71.3, 68.5, 62.0, 58.4, 53.4, 53.1, 29.4, 28.2, 23.2, 20.8, 20.7, 20.6, 20.6, 18.3, 10.0 ppm. HRMS-ESI: Calculated [M+Na] $^+$: 422.1850; found: 422.1850.

4. Silane synthesis using MPTMS or thiol/ATES.

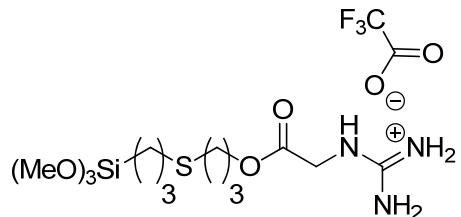
General procedure. A cooled oven-dried 10-mL round bottom was charged with either MPTMS, alkene (1 equiv.) and 2 mol% Irgacure®651, or ATES (1 equiv.), or thiol (1 equiv.) and 2 mol% Irgacure®651. The reaction was run neat unless one of the reactants did not fully dissolve; in which case a tiny amount of dry chloroform, or methanol was added. The reaction mixture was then capped with a septum and purged briefly with argon or nitrogen. The flask was placed next to an 18" 15 W blacklight, with total UV output of 2.6 W and $\lambda_{\text{max}} = 368$ nm, so that the flask's side rested against the center of the bulb. The UV flux at $\lambda_{\text{max}} = 368$ nm was measured to be attenuated ~30% after passing through standard thickness Pyrex glass. Both the flask and UV blacklight were wrapped in aluminum foil and the reaction mixture irradiated for 24 hours with gentle stirring to yield products in high purity. If solvent was added to the reaction, it was removed upon reaction completion *in vacuo*.



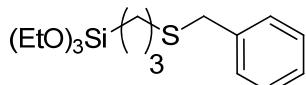
Acetal trimethoxysilane (11). Following the general procedure above, MPTMS (2 mmol), acrolein diethyl acetal (2 mmol), Irgacure®651 (0.04 mmol), were mixed, purged, and irradiated 24 h (>99% conversion, 96% pure). ^1H NMR (400 MHz, CDCl_3): $\delta = 4.59$ (t, 1H, $^3J_{HH} = 5.5$ Hz), 3.64 (m, 2H), 3.54 (s, 9H), 3.49 (m, 2H), 2.52 (q, 4H, $^3J_{HH} = 7.5$ Hz), 1.86 (m, 2H), 1.69 (m, 2H), 1.19 (t, 6H, $^3J_{HH} = 7.0$ Hz), 0.74 (t, 2H, $^3J_{HH} = 8.1$ Hz) ppm. ^{13}C NMR (400 MHz, CDCl_3): $\delta = 101.7, 61.5, 50.5, 35.1, 33.8, 27.1, 22.9, 15.3, 8.6$ ppm. HRMS-ESI: Calculated $[\text{M}+\text{Na}]^+$: 349.1475; found: 349.1478.



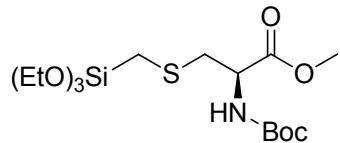
Undecanoic acid trimethoxysilane (12). Following the general procedure above, MPTMS (2.5 mmol), 10-undecenoic acid (2.5 mmol), Irgacure®651 (0.05 mmol), and dry CHCl_3 (0.5 mL) were mixed, purged, and irradiated 24 h (>99% conversion, 96% pure). ^1H NMR (400 MHz, CDCl_3): $\delta = 3.56$ (s, 9H), 2.50 (m, 4H), 2.34 (t, 2H, $^3J_{HH} = 7.8$ Hz), 1.60 (m, 6H), 1.32 (m, 12H), 0.75 (t, 2H, $^3J_{HH} = 8.1$ Hz) ppm. ^{13}C NMR (400 MHz, CDCl_3): $\delta = 179.5, 117.7, 50.5, 35.1, 34.0, 32.0, 29.7, 29.4, 29.3, 29.2, 29.0, 28.9, 24.7, 23.0, 8.6$ ppm. HRMS-ESI: Calculated $[\text{M}-\text{H}]^-$: 379.1980; found: 379.1992.



Guanidinium triethoxysilane (13). Following the general procedure above, MPTMS (1.0 mmol), ((2-(allyloxy)-2-oxoethyl)amino)(amino)methaniminium 2,2,2-trifluoroacetate (1.0 mmol), Irgacure®651 (0.05 mmol), and dry CH₃CN (0.7 mL) were mixed, purged, and irradiated 24 h (>99% conversion, 95% pure). The final product contained some dimerized silane due to reaction of **13** with small amounts of water trapped in the guanidinium salt. Formation of the dimer was determined from analysis of the high resolution mass spectrum, ¹H NMR, and ¹³C NMR spectra. ¹H NMR (400 MHz, CD₃CN): δ = 7.81 (bs, 1H), 7.16 (bs, 4H), 4.20 (t, 2H, ³J_{HH} = 6.4 Hz), 3.96 (bs, 2H), 3.49 (s, 5H), 2.52 (m, 4H), 1.87 (m, 2H), 1.62 (m, 2H), 0.72 (t, 2H, ³J_{HH} = 7.9 Hz) ppm. ¹³C NMR (400 MHz, CD₃CN): δ = 168.7, 168.6, 168.6, 158.2, 64.2, 7, 42.7, 34.2, 28.4, 27.5, 22.9, 22.8, 9.5, 9.4, 8.0 ppm. ¹³F NMR (400 MHz, CD₃CN): δ = -76.6 ppm. HRMS-ESI: Calculated [M-C₂O₂F₃]⁺: 354.1519; found: 354.1518. Calculated for dimer [2M- 2C₂O₂F₃-2MeO-H]⁺: 661.2535; found: 661.2549.



Benzyl triethoxysilane (14). Following the general procedure above, ATES (2 mmol), benzyl hydrosulfide (2 mmol), Irgacure®651 (0.04 mmol), were mixed, purged, and irradiated 24 h (>99% conversion, 97% pure). ¹H NMR (400 MHz, CDCl₃): δ = 7.30 (m, 4H), 7.23 (m, 1H), 3.80 (q, 6H, ³J_{HH} = 7.0 Hz), 3.70 (s, 2H), 2.44 (t, 2H, ³J_{HH} = 7.3 Hz), 1.66 (m, 2H), 1.22 (t, 9H, ³J_{HH} = 7.0 Hz), 0.70 (t, 2H, ³J_{HH} = 8.3 Hz) ppm. ¹³C NMR (400 MHz, CDCl₃): δ = 138.7, 128.8, 128.4, 126.9, 58.4, 36.1, 34.4, 22.8, 18.3, 9.9 ppm. HRMS-ESI: Calculated [M-C₄H₉O]⁺: 255.0870; found: 255.0874.



Boc-cysteine-OMe triethoxysilane (15). Following the general procedure above, ATES (2 mmol), *N*-(*tert*-butoxycarbonyl)-L-cysteine methyl ester (2 mmol), Irgacure®651 (0.04 mmol), were mixed, purged, and irradiated 24 h (>99% conversion, 97% pure). ^1H NMR (400 MHz, CDCl_3): δ = 5.43 (d, 1H, $^3J_{HH}$ = 7.7 Hz), 4.51 (q, 1H, $^2J_{HH}$ = 14.5 Hz, $^3J_{HH}$ = 4.9 Hz), 3.80 (q, 6H, $^3J_{HH}$ = 7.0 Hz), 3.74 (s, 3H), 2.93 (d, 2H, $^3J_{HH}$ = 4.6 Hz), 2.52 (t, 2H, $^3J_{HH}$ = 7.3 Hz), 1.65 (m, 2H), 1.43(s, 9H), 1.20 (t, 9H, $^3J_{HH}$ = 6.6 Hz), 0.69 (t, 2H, $^3J_{HH}$ = 8.4 Hz) ppm. ^{13}C NMR (400 MHz, CDCl_3): δ = 171.6, 155.1, 80.0, 58.3, 53.2, 52.4, 35.6, 34.3, 28.3, 23.1, 18.3, 9.8 ppm. HRMS-ESI: Calculated [M+H] $^+$: 440.2138; found: 440.2123.

5. Large-scale synthesis of 3

A cooled oven-dried Pyrex glass tube (7.5" tall, 1" diameter) with a 29/42 joint was charged with MPTES (107 mmol), acrolein diethyl acetal (107 mmol), Irgacure®651 (2.1 mmol), and then capped with a septum purged briefly with argon. The small tube diameter made it possible for the maximum amount of light to illuminate the reaction mixture. The tube was placed next to two 15 W, 18" long blacklights, each having a total UV output of 2.6 W ($\lambda_{\text{max}} = 368 \text{ nm}$). The tube was positioned so that each blacklight just rested against each side of the tube, (See Figure S1 below.) The flask and blacklights were wrapped in aluminum foil and the reaction mixture irradiated for 24 h with gentle stirring to yield product silanes in high purity (>99% conversion, 93% pure). ^1H NMR (400 MHz, CDCl_3): $\delta = 4.95$ (t, 1H, $^3J_{HH} = 5.6 \text{ Hz}$), 3.80 (q, 6H, $^3J_{HH} = 7.1 \text{ Hz}$), 3.64 (m, 2H), 3.49 (m, 2H), 2.53 (q, 4H, $^3J_{HH} = 6.6 \text{ Hz}$), 1.88 (m, 2H), 1.70 (m, 2H), 1.20 (m, 15H), 0.72 (t, 2H, $^3J_{HH} = 8.1 \text{ Hz}$) ppm. ^{13}C NMR (400 MHz, CDCl_3): $\delta = 101.17, 61.5, 58.4, 35.2, 33.9, 27.1, 23.2, 18.3, 15.3, 9.9$ ppm.

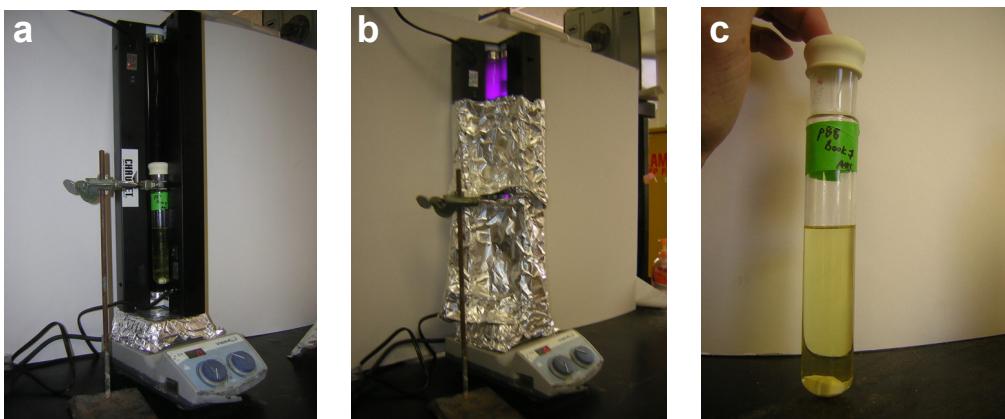


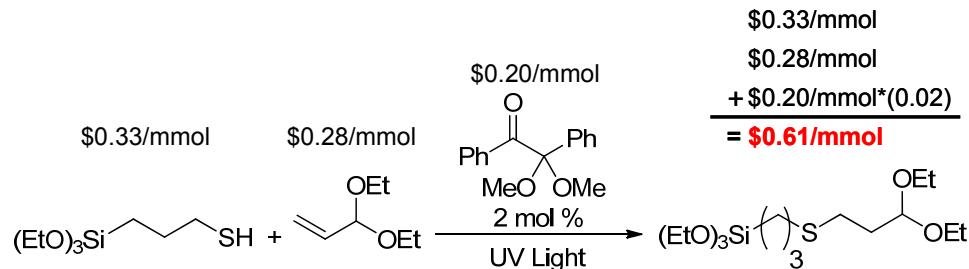
Figure S1. Images of setup for large-scale synthesis of 3: a) prior to, and b) during reaction. c) Image of reaction tube containing product 3 after the reaction was complete.

6. Cost analysis for synthesizing triethoxysilanes 3 and 7

The estimates provided here of the material cost to produce triethoxysilanes **3** and **7** via the thiol-ene reaction were simplified by only considering the cost of the chemicals. [The blacklights, stir plate, glassware, and energy required to power the lamps were not included because the equipment is reusable and the blacklights we used are very inexpensive.]

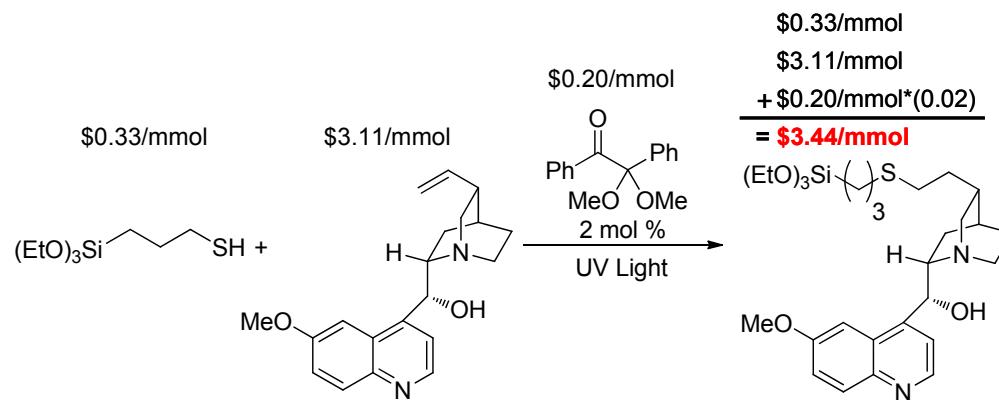
The cost estimates for producing **3** and **7** per mmol material are shown below in Schemes S1 and S2, respectively. Starting materials to prepare **3** and **7** included 25 g acrolein diethyl acetal ($\geq 96\%$), 5 g quinine (suitable for fluorescence, anhydrous, $\geq 98.0\%$), 50 g 2,2-dimethoxy-1,2-diphenylethanone ($>99\%$) purchased from Sigma Aldrich, and 25g MPTES ($\geq 95\%$) purchased from Gelest Inc.

The estimated cost to prepare **3** is \$0.61/mmol **3**. The retail price of the chemically similar, commercially available triethoxysilylundecanal ethylene glycol acetal (Gelest catalog #: SIT8194.5) is 13 times higher.



Scheme S1. Synthesis of diethyl acetal triethoxysilane **3** and cost of materials.

The estimated cost to prepare **7** is \$3.44/mmol. For comparison, the price of (R)-*N*-triethoxysilylpropyl quinine urethane (Gelest catalog #: SIT8192.4) is 4 times higher.



Scheme S2. Synthesis of quinine triethoxysilane **7** and cost of materials.

7. UV-visible spectra of silane **9** and photoinitiator.

Figure S2 shows an overlay of the UV-visible spectra of silane **9** (0.3 M) and the Irgacure® 651 photoinitiator (0.003 M). At 368 nm, λ_{\max} for the blacklight used to drive the thiol-ene reaction, there is a little overlap of the absorption bands of the photoinitiator and product. Despite the large difference in molar extinction coefficients at 368 nm, one must remember that the photoinitiator is present at only at 2 mol% while the silane is extremely concentrated. Consequently, there is significant absorbance overlap at 365 nm. This absorbance overlap is believed to be a factor that prevents the reaction of MPTES (0.5 mmol) and (S)-*tert*-butyl-2-di(*tert*-butyloxycarbonylmethyl)amino-3-(pent-4-enyloxy)propanoate from reaching completion. The other contributing factor is thought to be the increase in the solution viscosity of the reaction as it proceeds, which limits mass transport.

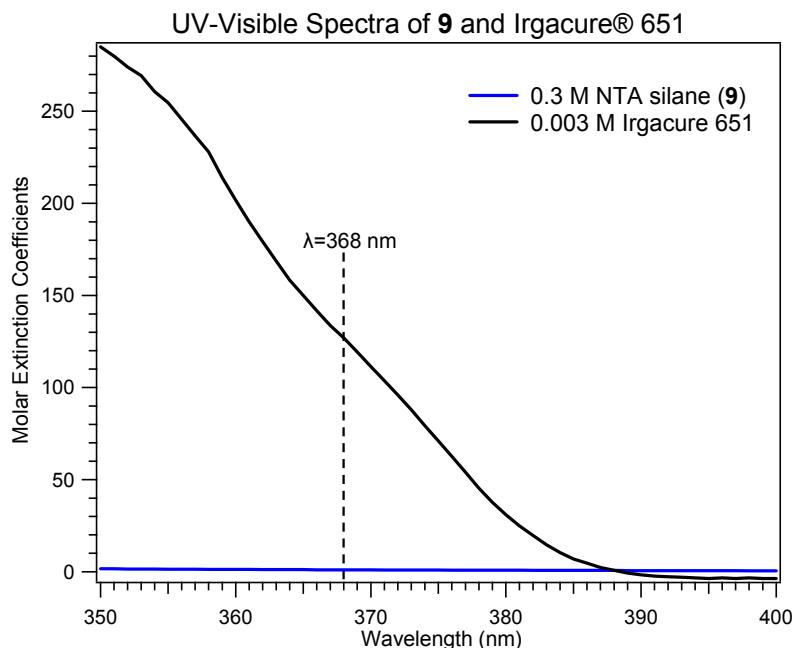


Figure S2. UV-visible spectra of 0.3 M NTA silane **9** in chloroform (black line) and 0.003 M Irgacure® 651 in chloroform (blue line) with the y-axis normalized to concentration and sample path length..

8. Nanoparticle synthesis and silane surface coating procedure.

8.1-Preparation of oleic acid-coated Fe_3O_4 superparamagnetic nanoparticles (SPNs). Core magnetite SPNs were synthesized by the method previously described.¹ Briefly, 1 mL of an aqueous 3.0 M FeCl_2 , 2 M HCl solution was mixed with 4 mL of 1.5 M FeCl_3 , 2 M HCl on a magnetic stir plate at R.T. The iron chloride solution was then slowly titrated dropwise with 50 mL of 1.05 M NH_4OH over a period of about 30 min, producing a deep black particle suspension. The superparamagnetic magnetite nanoparticles were separated using a hand-held neodymium rare-earth magnet and washed three times with water, three times with acetone, and then air-dried for at least 3 h. The collected particles were crushed, weighed and placed in a screw-cap glass vial. An equal weight of oleic acid was added to the vial along with 1 mL ethyl acetate *per 0.1 g SPNs*. The vial was capped and sonicated for 2 h at 60 °C. The suspension was cooled, and the particles were magnetically separated, washed three times with excess ethyl acetate, three times with methanol, and then dried under vacuum. The final product was crushed to yield a black/dark brown powder.

8.2- General procedure for functionalizing Fe_3O_4 SPNs with triethoxysilanes 1–10. The goal of these experiments was simply to determine whether the triethoxysilanes 1–10 could be covalently attached to magnetite SPN surfaces and detected. Because we were not concerned with the relative surface reactivities of the silanes, the amount of silane used in the reaction was not tightly controlled. In most cases, the triethoxysilanes were made on ≥ 1 mmol scale, and 1 mmol was used to coat the SPNs. In a few cases, < 1 mmol was synthesized, and all of the product was used to coat the SPNs.

Oleic acid-coated SPNs (120 mg), and either toluene or dioxane (60 mL), passed over MgSO_4 just prior to use, were added to a hot, oven-dried 100 mL round bottom flask. The flask was immediately capped and purged with argon for 5 min. The flask was then placed in an ultrasonication bath at R.T. and sonicated for 20 min to redisperse the oleic acid-coated SPNs. Then a stir bar was quickly added to the flask, which was immediately capped and purged 5 min with argon while vigorously stirring the particle suspension to prevent the SPNs from sticking to the stir bar. An argon balloon was placed on the flask and a triethoxysilane (0.4 to 1 mmol; see Table S1 for exact quantity of added silane) was added to the flask dropwise. The flask was placed in a 75 °C oil bath and heated for 24 h with vigorous stirring. The stir bar was removed and the particles were separated using a hand-held rare-earth magnet, washed three times with either diethyl ether or CHCl_3 , and then dried under vacuum.

9. Nanoparticle spectroscopic data.

9.1- Elemental analysis of silane loadings on coated SPNs. All triethoxysilanes used to coat the SPNs contained at least one sulfur atom present in the thioether bridge formed during synthesis of the silane. This allowed quantification of silane loadings on coated SPNs by sulfur elemental analysis. The experimentally-determined loadings are given in Table S1. Nearly all of the thiol-ene reactions used to produce these silanes reached $\geq 98\%$ conversion. The crude products contained virtually no MPTES, which, if present, could bind to the surface and contribute to the measured sulfur content. Hence, for the reactions with $>98\%$ reactant conversions, we have assumed that the measured SPN sulfur content of the silane-coated particles derived only from the *functional triethoxysilane*. For the two cases in which the reactions reached only 94% or 96% conversion (**6**, **9**), we have assumed that the percentage of the measured sulfur content that is attributable to surface-bound functional triethoxysilane is at best equal to the percent purity of the as-synthesized crude silane .

Table S1. Triethoxysilane loadings on SPNs.^a

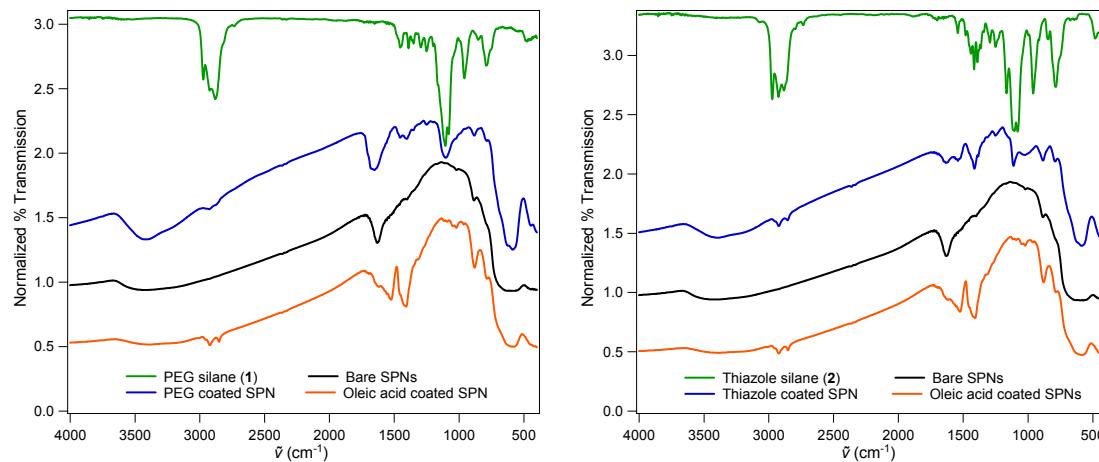
Coated silane	Quantity used (mmol)	Silane loading (mmol/g SPN)
1	1.0	0.23
2	1.0	0.34
3	1.0	0.79
4	1.0	0.16
5	1.0	0.31
6	0.5	0.23 ^b
7	0.4	0.27
8	1.0	0.27
9	0.5	0.13 ^c
10	1.0	0.15

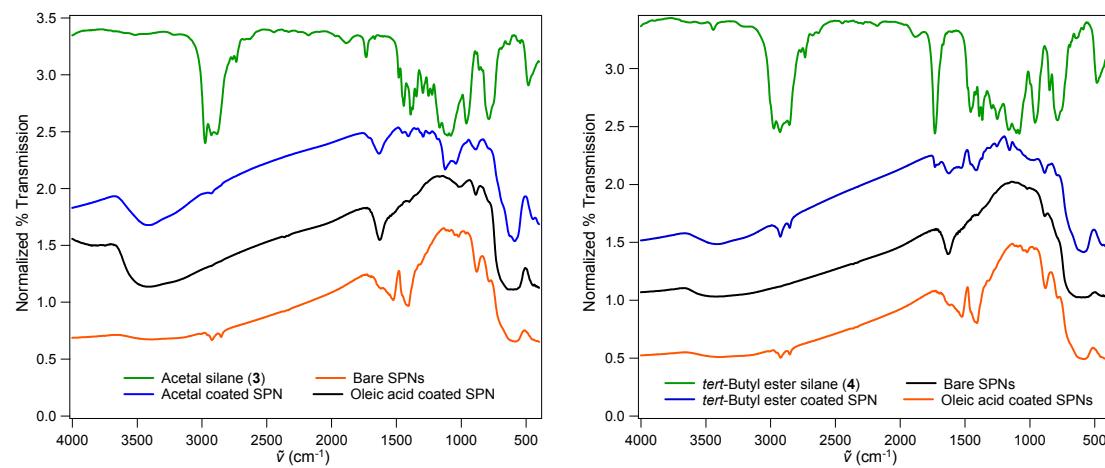
^aLoadings determined by sulfur elemental analysis.

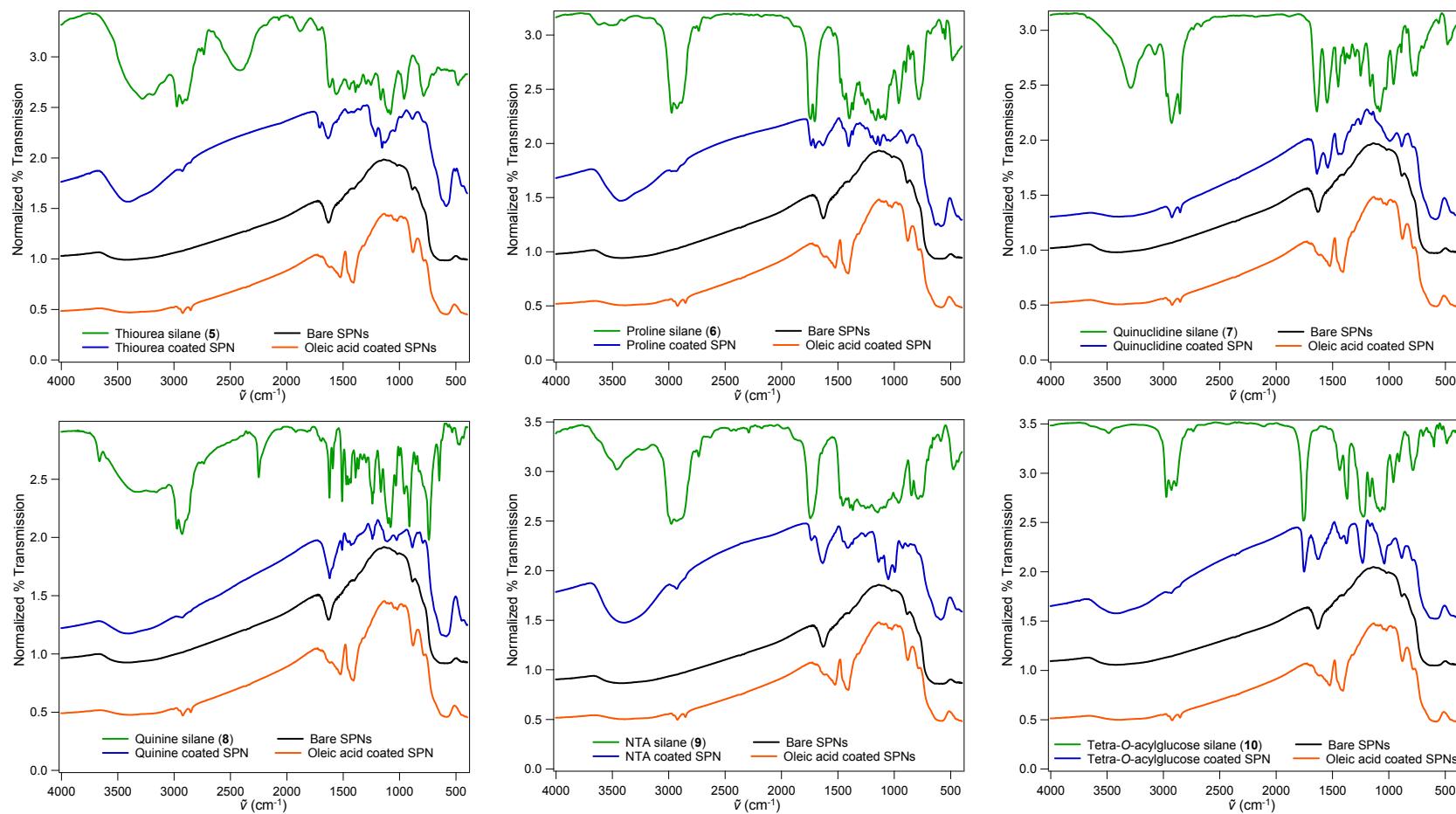
^bSilane loading assuming 94% of the SPN sulfur content is from the functional silane and the rest from MPTES.

^cSilane loading assuming 90% of the SPN sulfur content is from the functional silane and the rest from MPTES.

9.2- FT-IR spectroscopic analysis of silane-coated SPNs. FT-IR spectra of silane-coated SPNs pressed into KBr pellets were obtained to confirm that that silanes had been immobilized on the particle surfaces. The figures shown below depict overlaid FT-IR spectra of the pure silanes, silane-coated SPNs, oleic acid-coated SPNs, and bare magnetite SPNs. The spectra verify that all of the triethoxysilanes **1–10** were successfully immobilized onto the surface of oleic-acid coated SPNs. A key indication that silanes have been covalently immobilized on the nanoparticle surface is the presence of a new Si-O-Si stretching vibration between 1060 and 1140 cm^{-1} .







9.3- Magnetic properties of SPNs. The magnetic properties of the synthesized SPNs were measured at R.T. on a supercritical quantum interference device (SQUID). Figure S3 shows magnetic hysteresis curves for all of the SPNs: the bare magnetite, the oleic acid-coated particles, and the silane-modified SPNs. All of the particles exhibit low coercivity, <20 Oe, indicating that they are superparamagnetic. Both the coated and uncoated particles have excellent saturation magnetizations, >34 emu/g, which means they are strongly attracted to magnetic fields produced by standard rare-earth magnets.

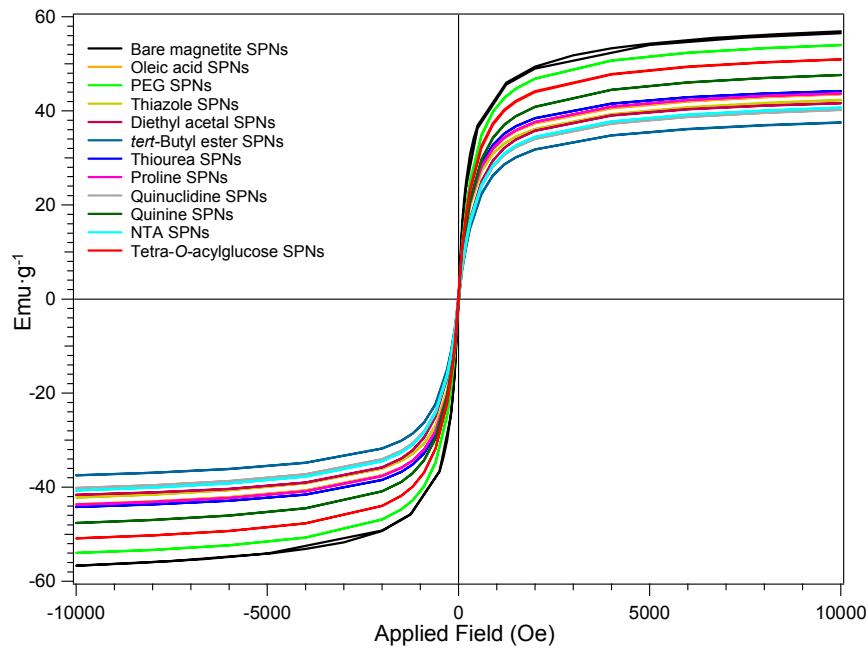


Figure S3. Magnetic hysteresis curves of all synthesized SPNs.

9.4- Transmission electron microscopy (TEM) of SPNs. The TEM images in Figure S4 show silane-coated SPNs ranging in size from ~10 to 25 nm. This polydispersity is expected from the protocol that was used to prepare these Fe₃O₄ nanoparticles. Single particles, clusters and large aggregates are typically present. We did not systematically investigate the reasons for the varying extent of particle aggregation, because many factors can influence this process and some are interdependent. These factors include: a) the composition of the surface coating; b) the surface coverage of the silane coating; c) the uniformity of silane coating; d) interactions between the particles, which are solvent-dependent; e) the dispersing solvent; and f) the cleanliness and surface charge of the carbon grids onto which the particle suspensions are deposited for TEM analysis.

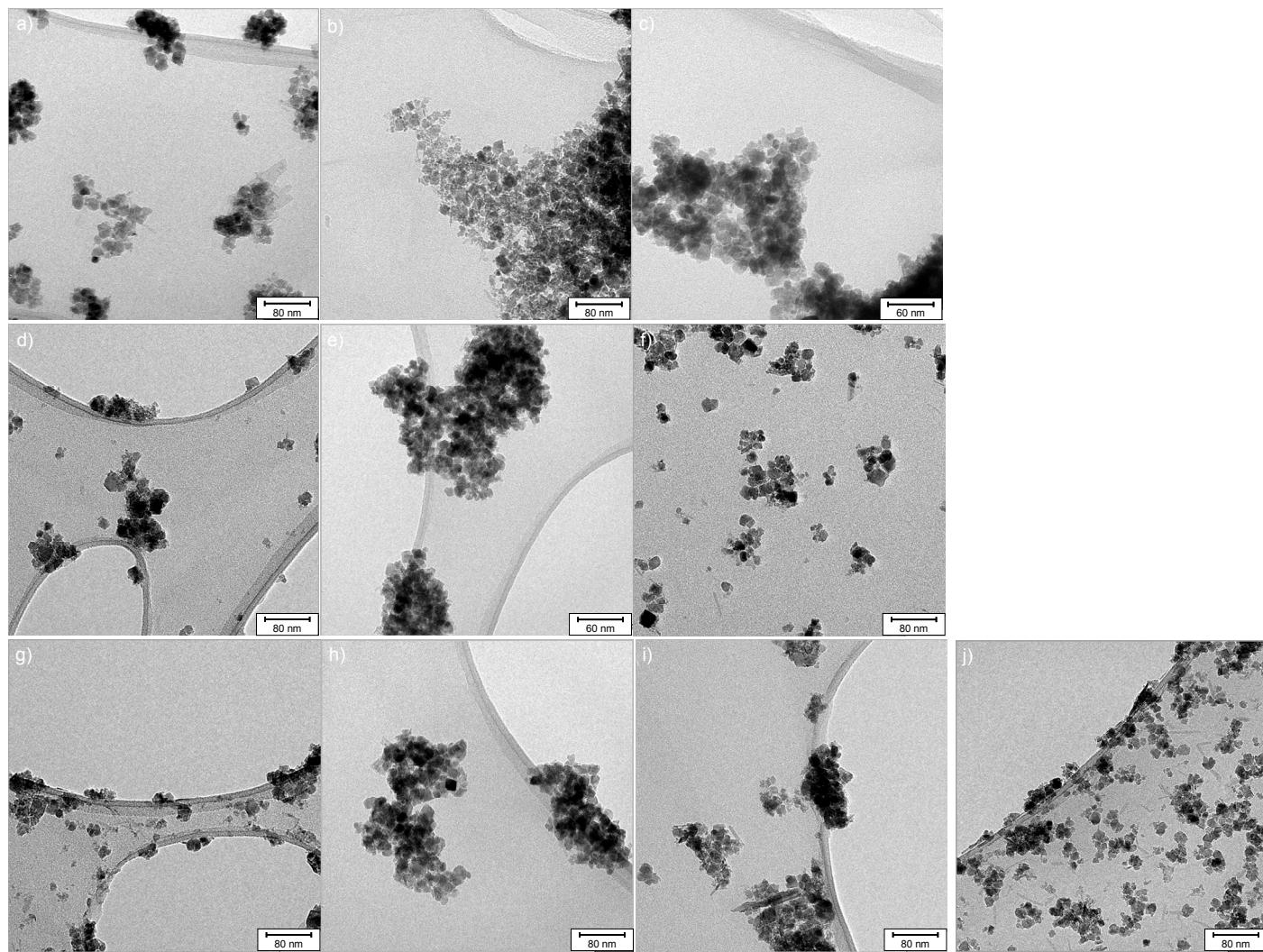
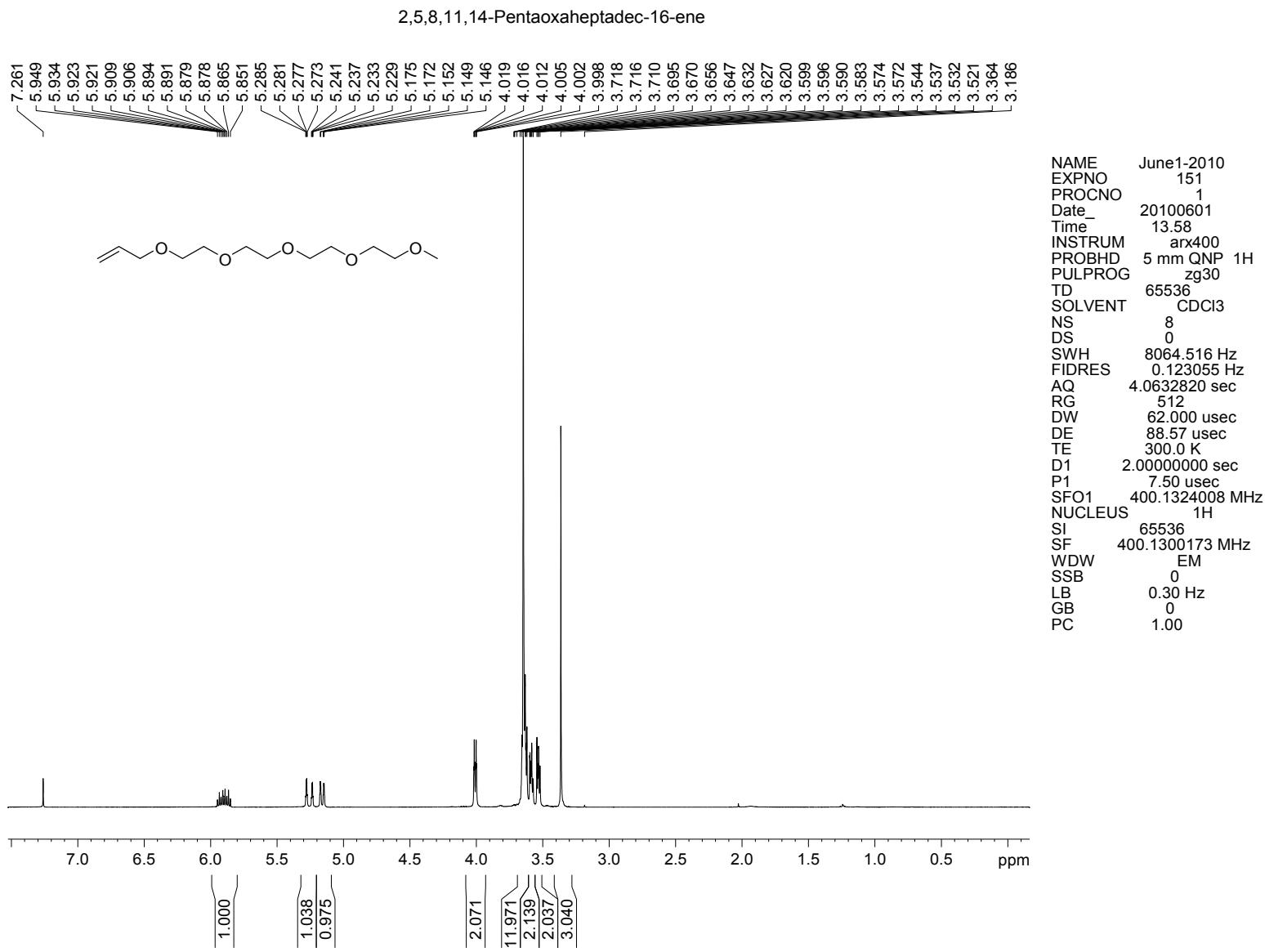
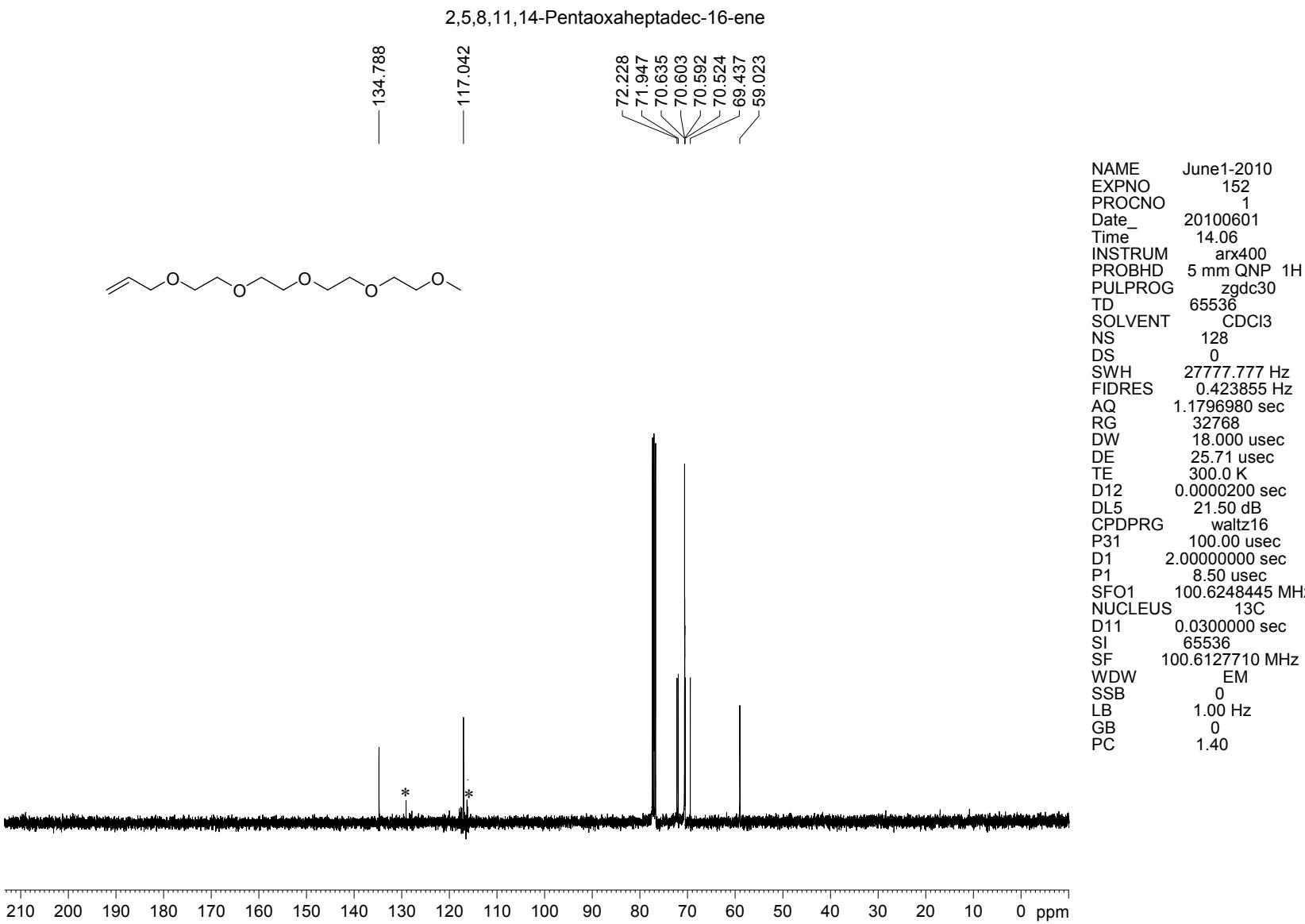


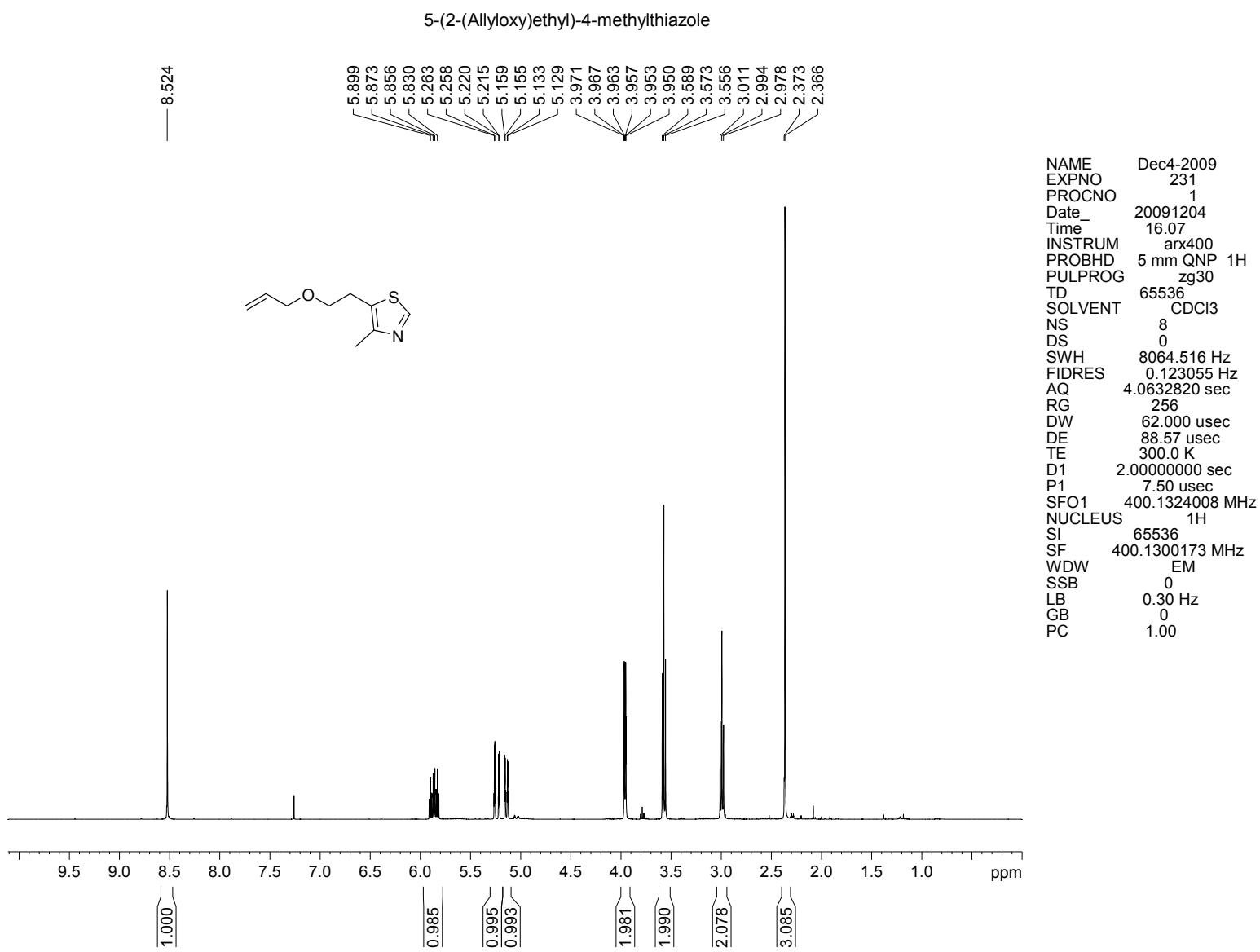
Figure S4. TEM images of a) PEG SPNs, **1**, b) thiazole SPNs, **2**, c) aectal SPNs, **3**, d) *tert*-butyl ester SPNs, **4**, e) thiourea SPNs, **5**, f) proline SPNs, **6**, g) quinuclidine SPNs, **7**, h) quinine SPNs, **8**, i) NTA SPNs, **9**, j) tetra-*O*-acylglucose, **10**, SPNs.

10. ^1H and ^{13}C NMR spectra of synthesized compounds.

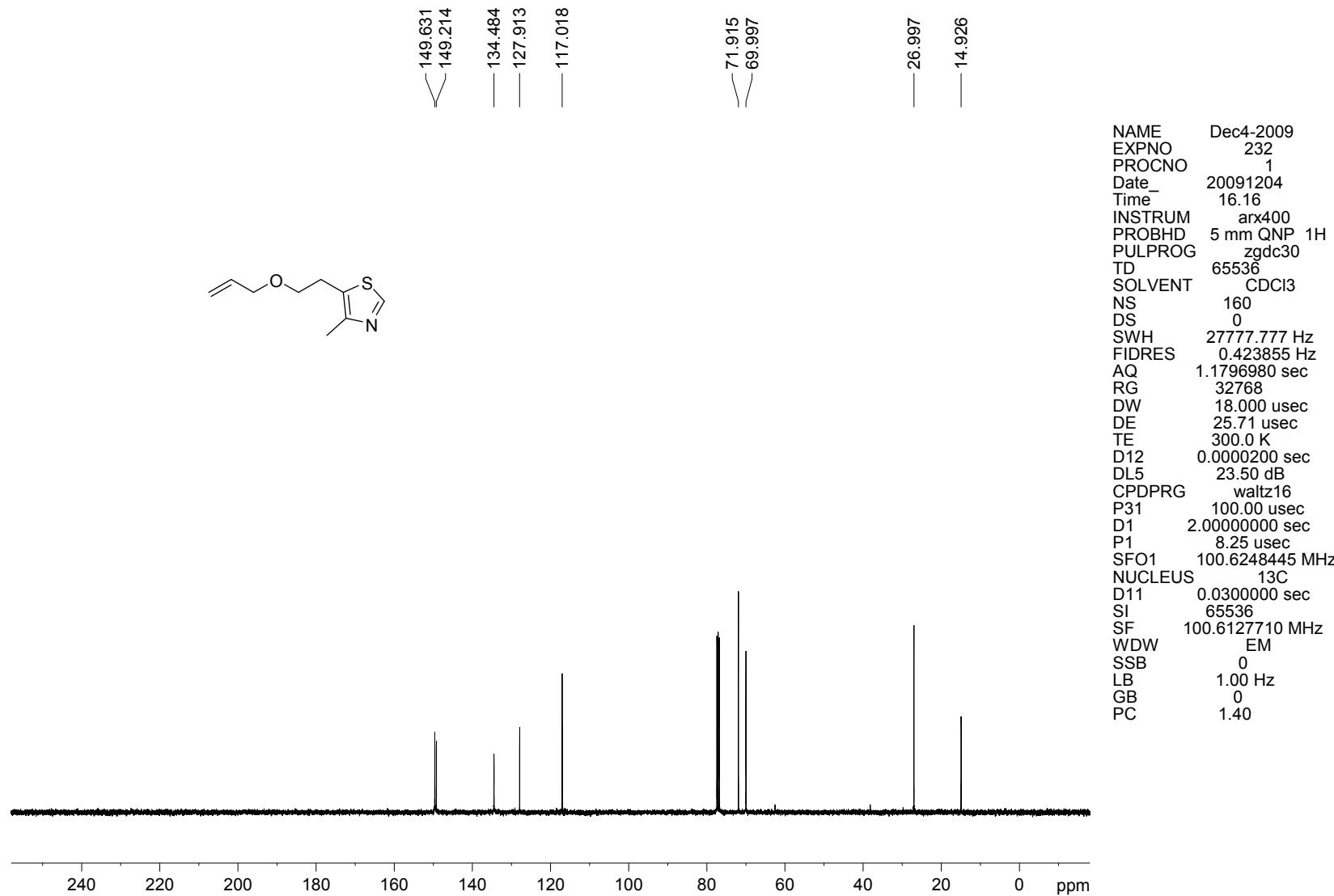
Shown below are ^1H NMR and ^{13}C NMR spectra of the synthesized compounds. Note that ^{13}C NMR spectra obtained with our 400 MHz spectrometer sometimes showed a significant amount of noise at \sim 118 and 129 ppm. Although this noise may look like one or more real spectral features, it is an artifact produced by the spectrometer, and cannot be corrected or compensated for, due to the instrument's age. These artifacts, denoted with asterisks (*) in the ^{13}C NMR spectra, are more noticeable when the signal-to-noise ratio is low. We have verified that the features are not assignable to vinyl carbons from unreacted alkenes by observing that ^1H NMR spectra of the same samples usually have extremely weak or no signals from the vinyl protons. Most importantly, high resolution mass spectra of the synthesized compounds confirm their identities.

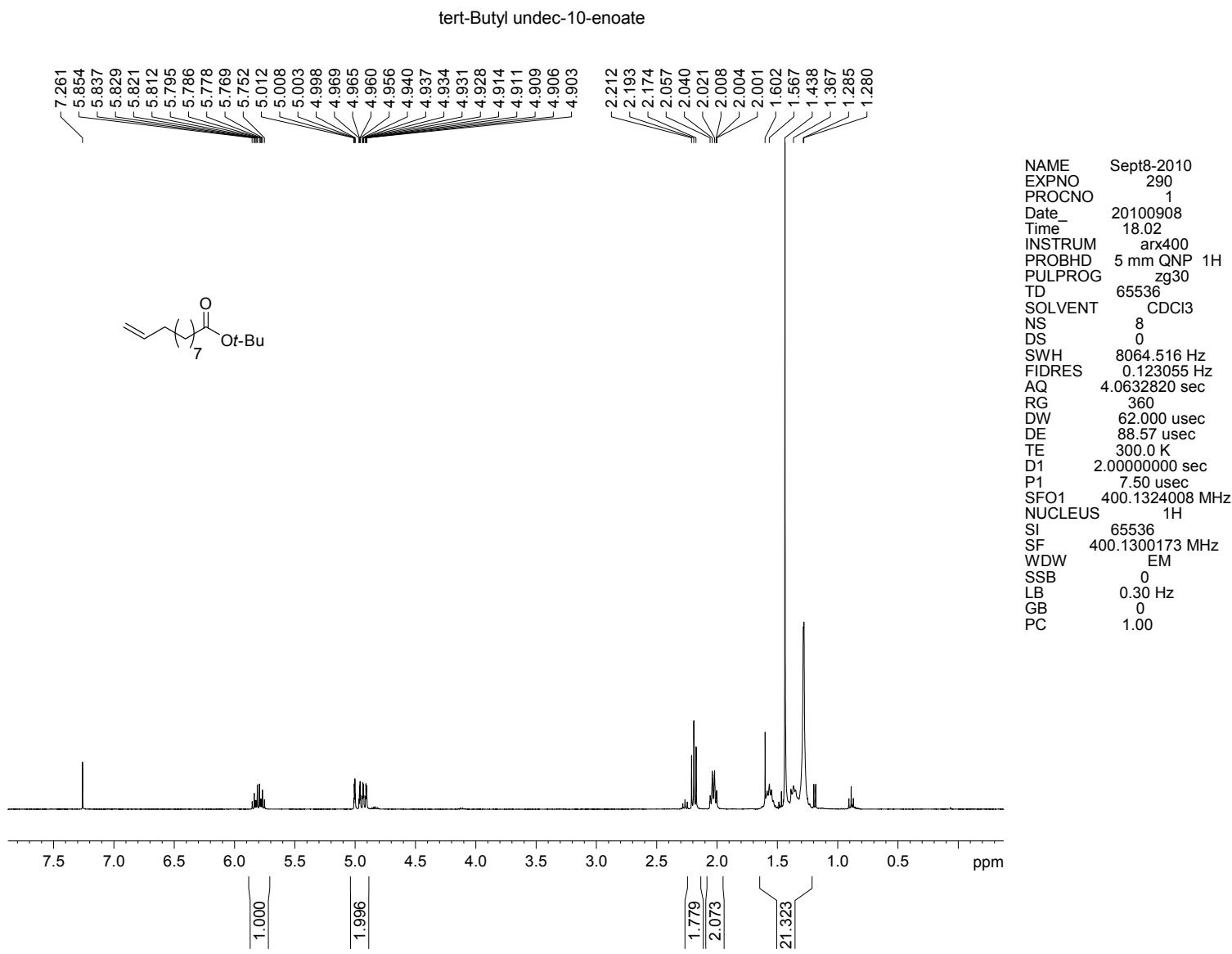


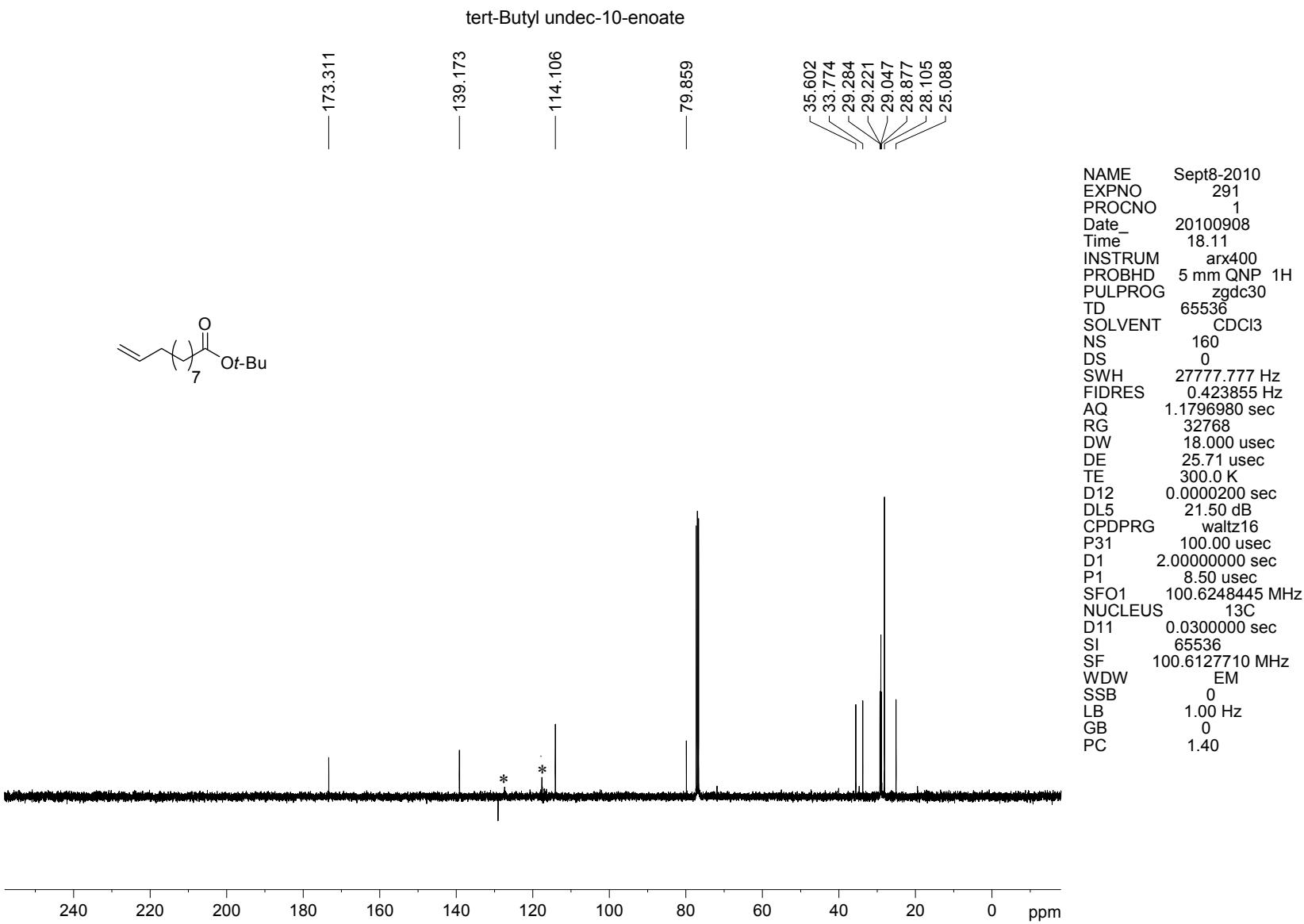


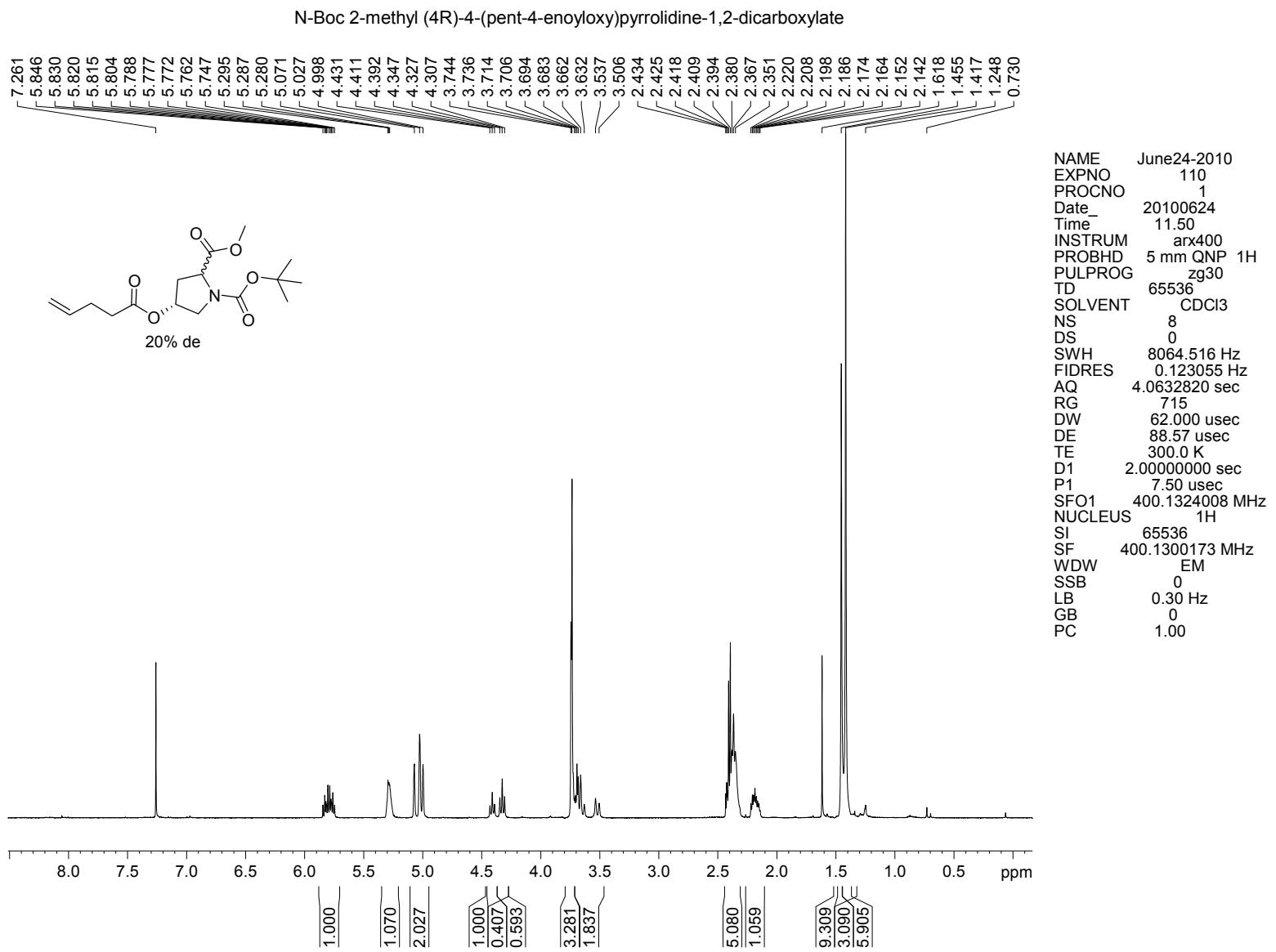


5-(2-(Allyloxy)ethyl)-4-methylthiazole

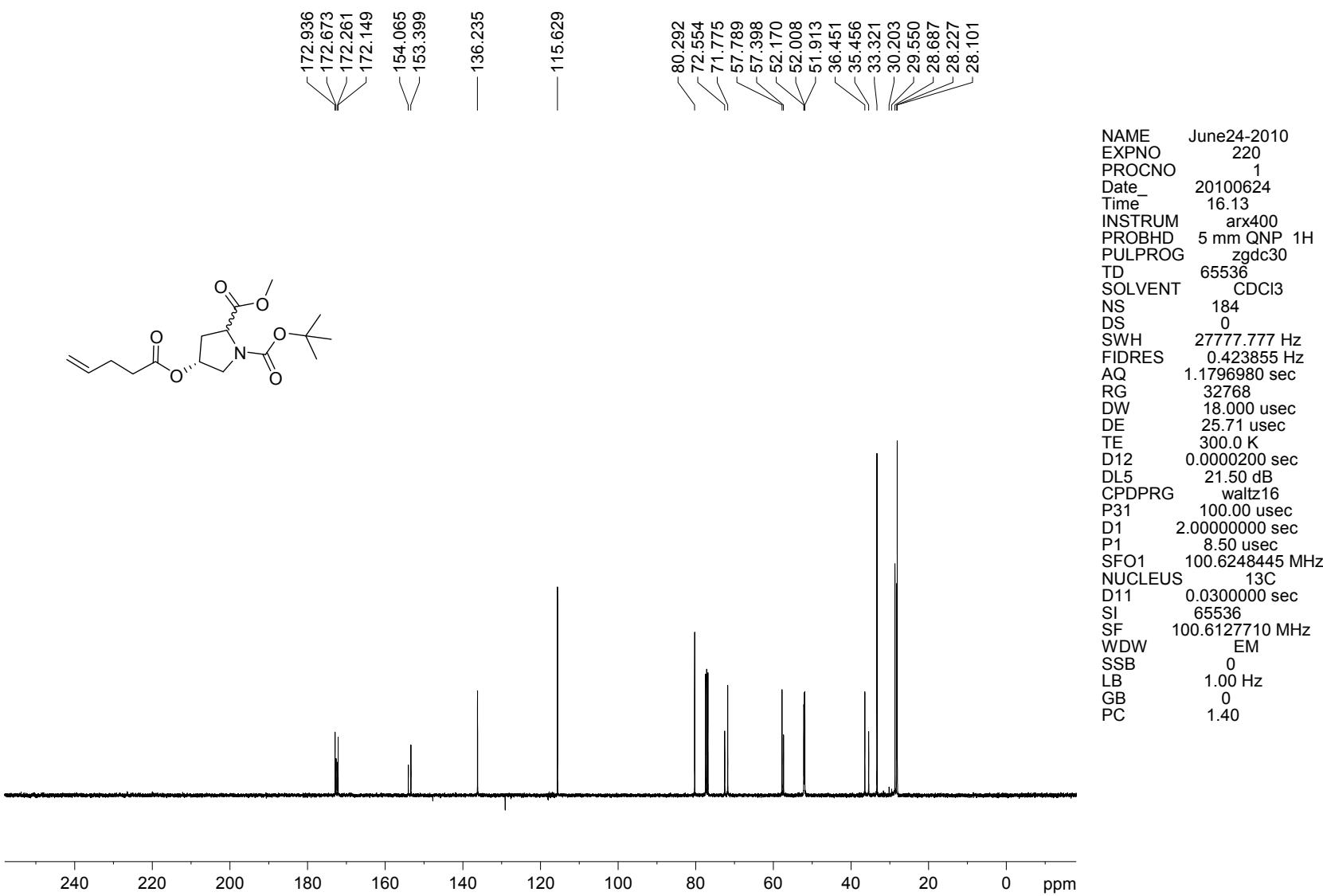




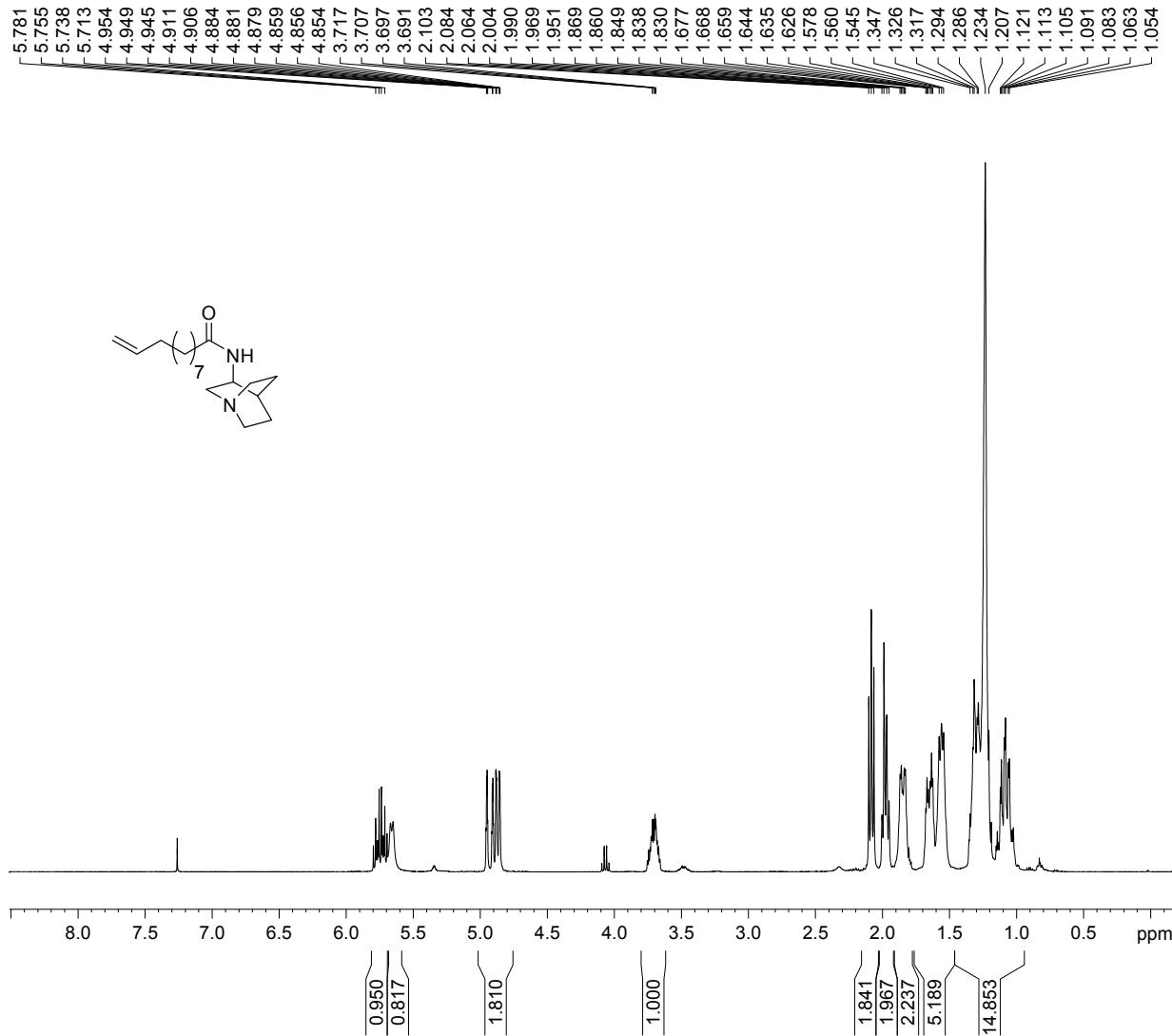
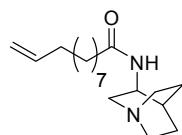




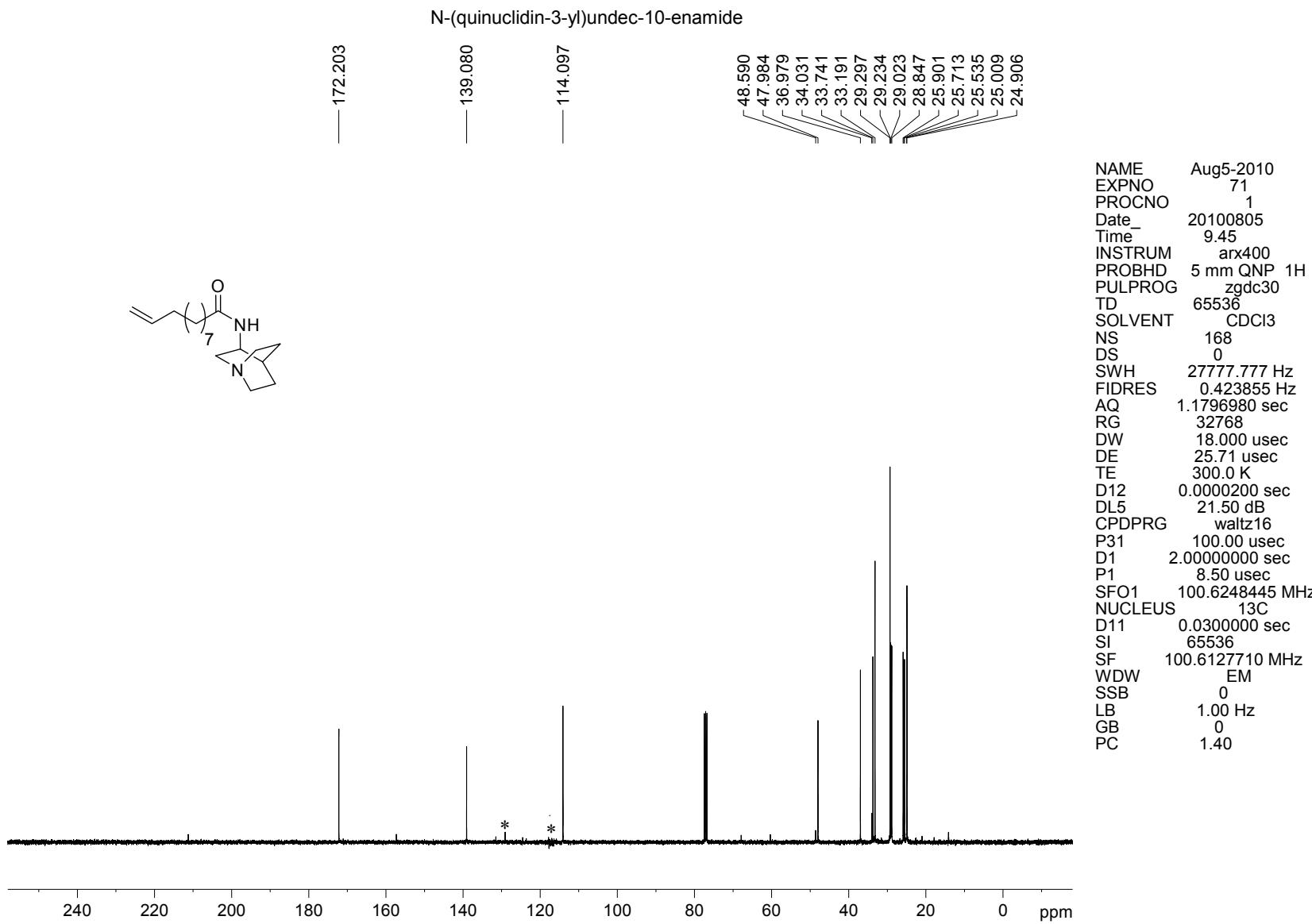
N-Boc 2-methyl (4R)-4-(pent-4-enyloxy)pyrrolidine-1,2-dicarboxylate



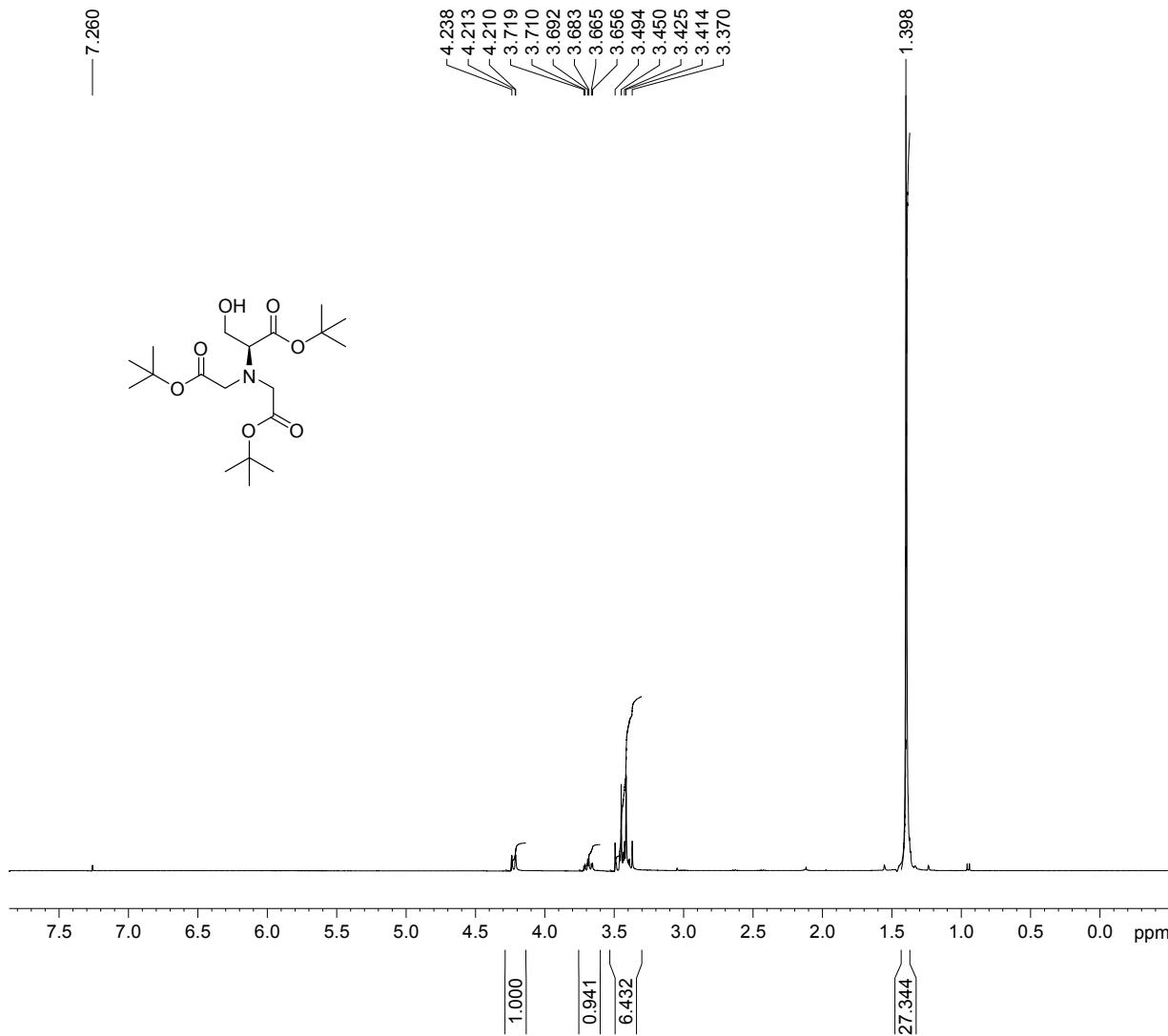
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FIDRES	0.123055 Hz
AQ	4.0632820 sec
RG	90
DW	62.000 usec
DE	88.57 usec
TE	300.0 K
D1	2.0000000 sec
P1	7.50 usec
SFO1	400.1324008 MHz
NUCLEUS	1H
SI	65536
SF	400.1300173 MHz
WDW	EM
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GB	0
PC	1.00



(S)-tert-Butyl-2-di(tert-butyloxycarbonylmethyl)amino-3-hydroxypropanoate

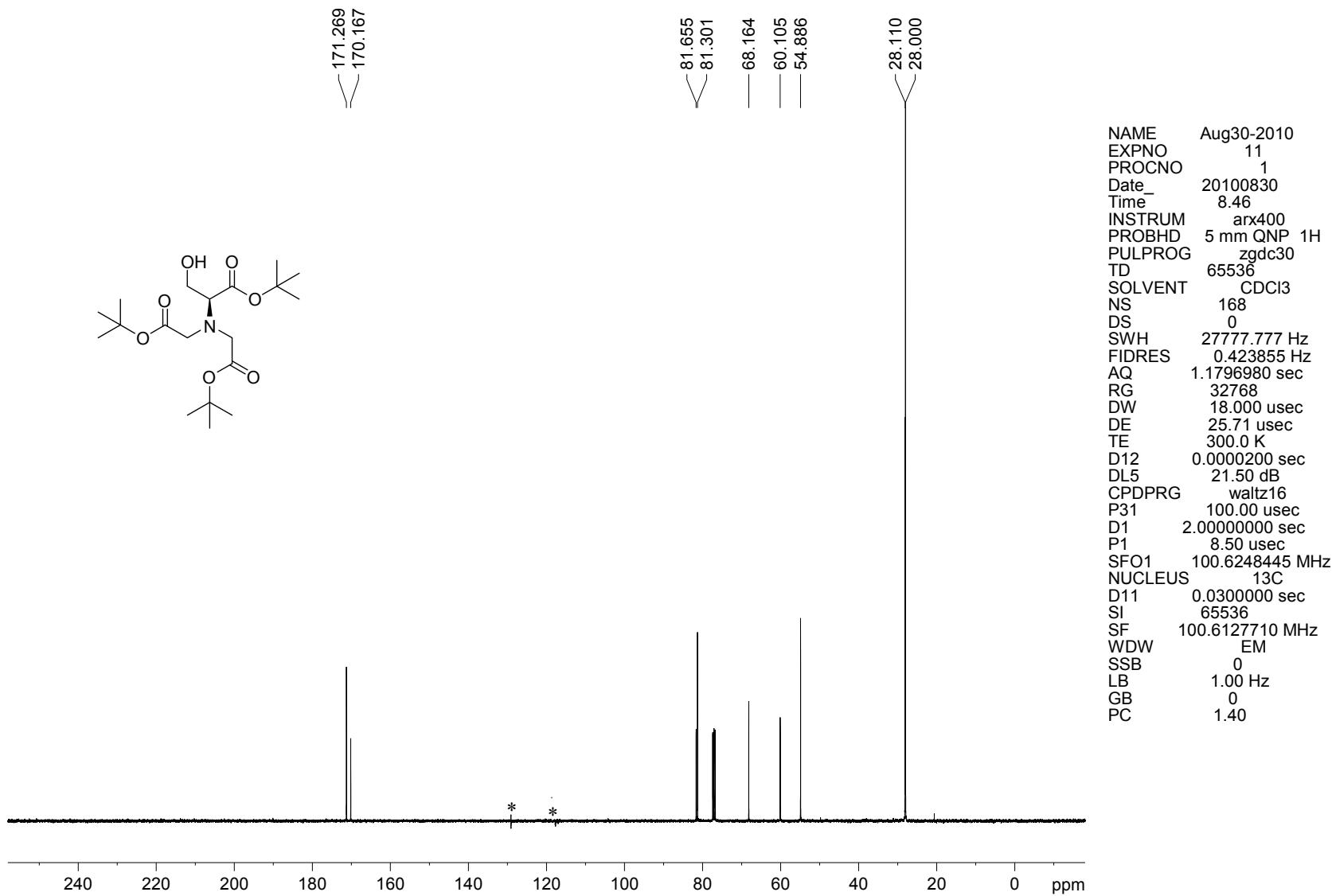


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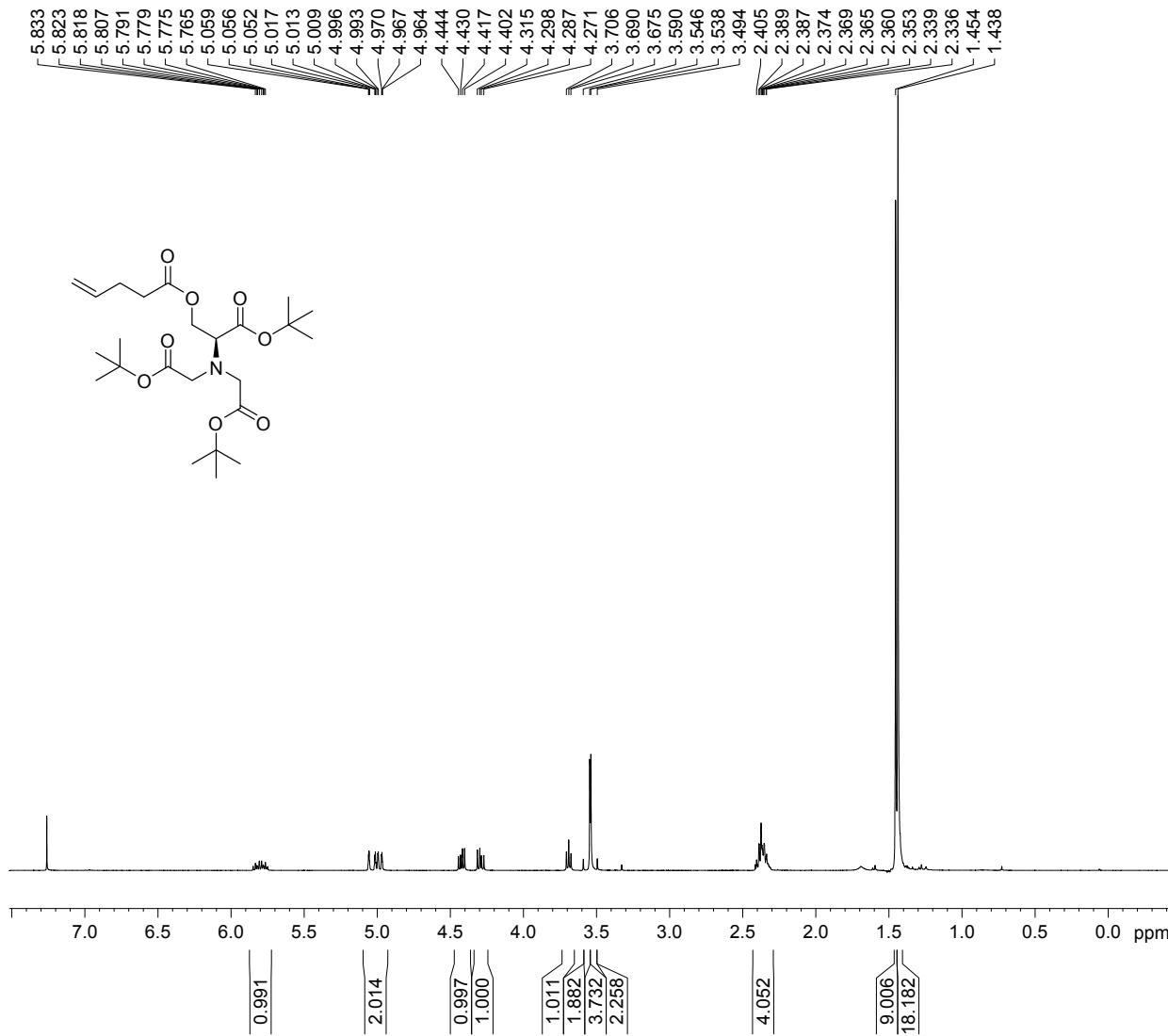
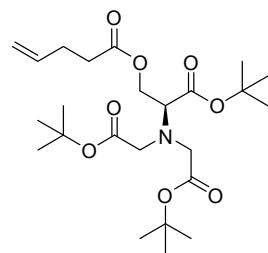
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FIDRES   0.123055 Hz
AQ        4.0632820 sec
RG        90
DW        62.000 usec
DE        88.57 usec
TE        300.0 K
D1        2.0000000 sec
P1        7.50 usec
SFO1     400.1324008 MHz
NUCLEUS  1H
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SF        400.1300173 MHz
WDW      EM
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(S)-tert-Butyl-2-di(tert-butyloxycarbonylmethyl)amino-3-hydroxypropanoate

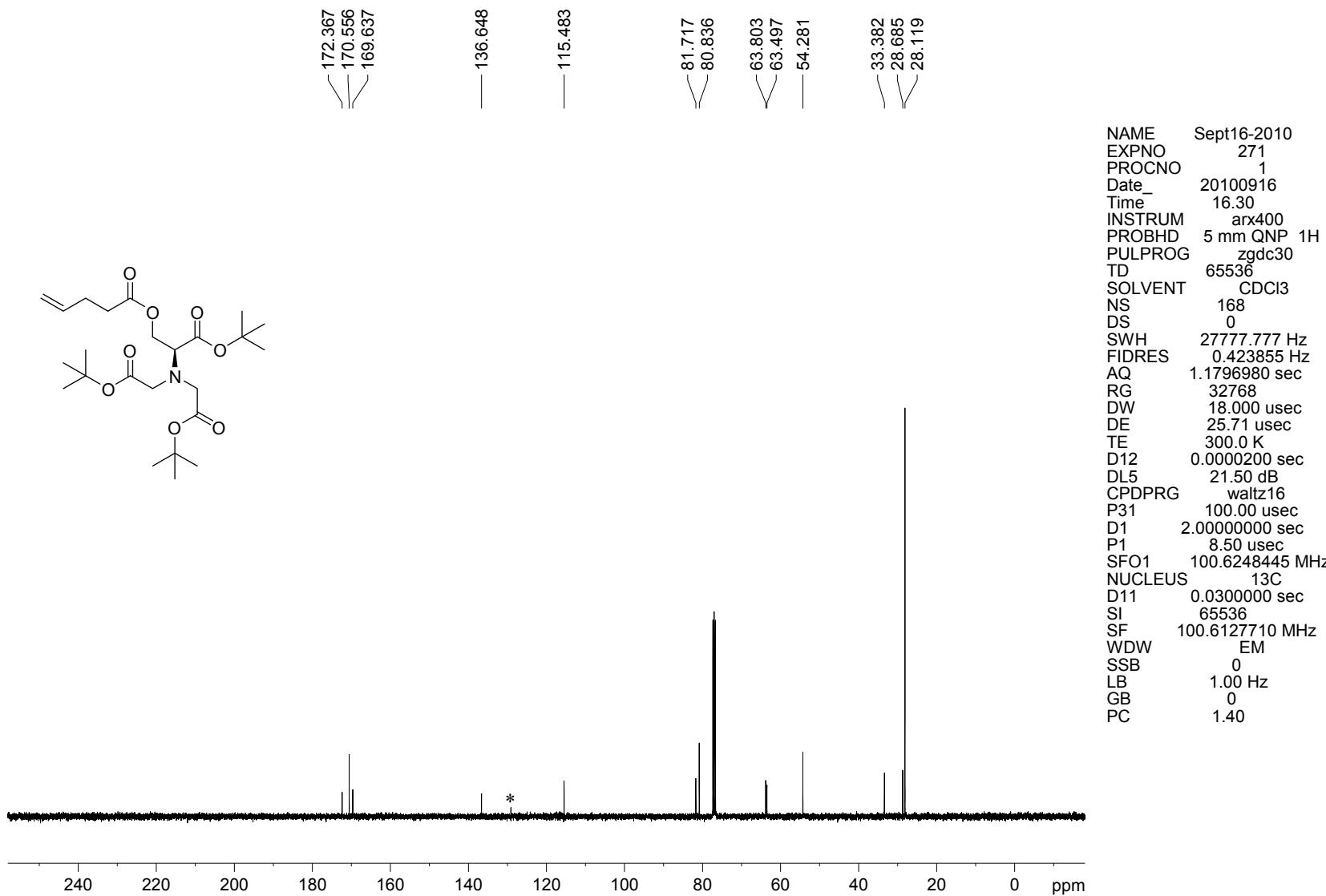


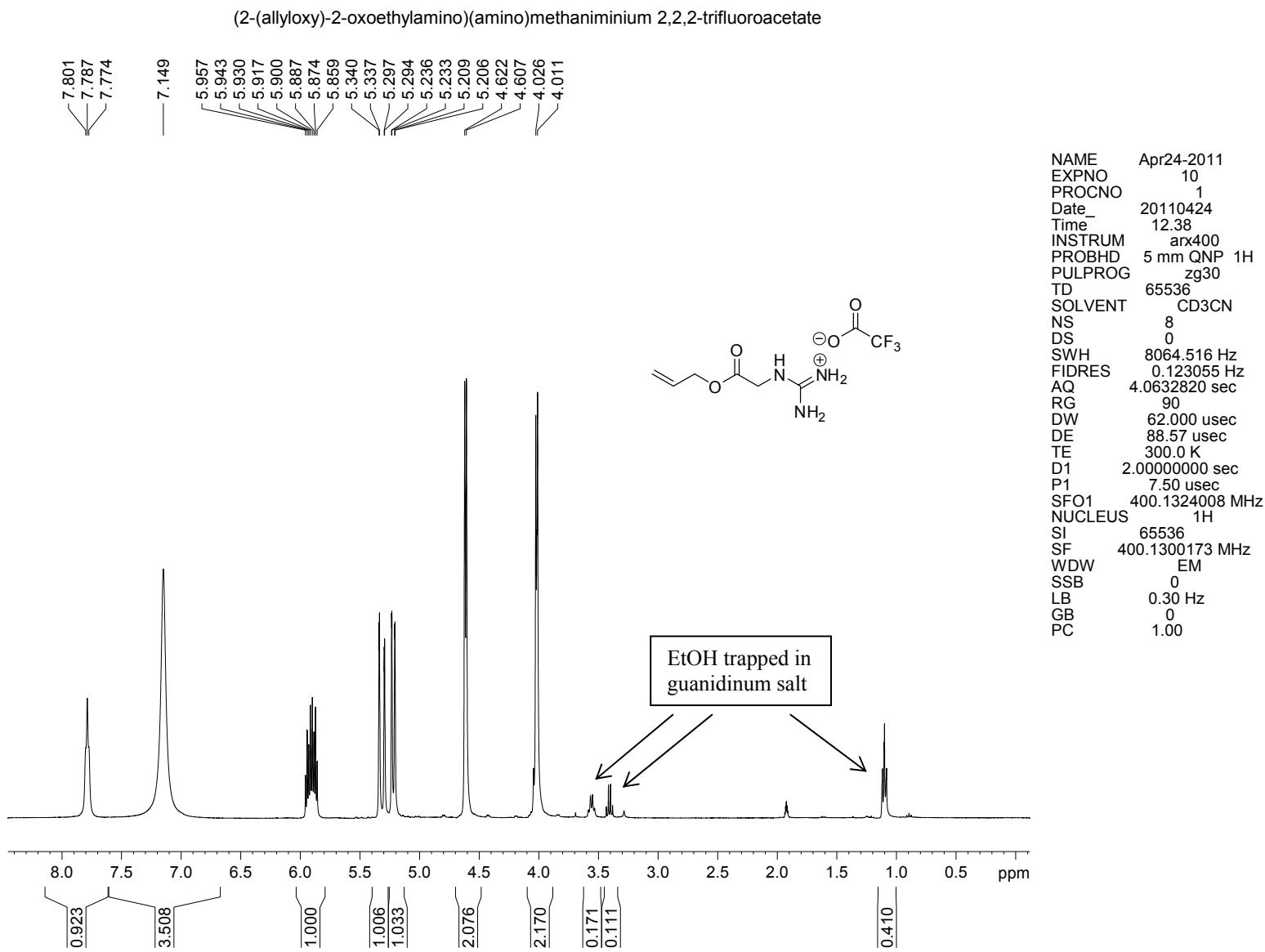
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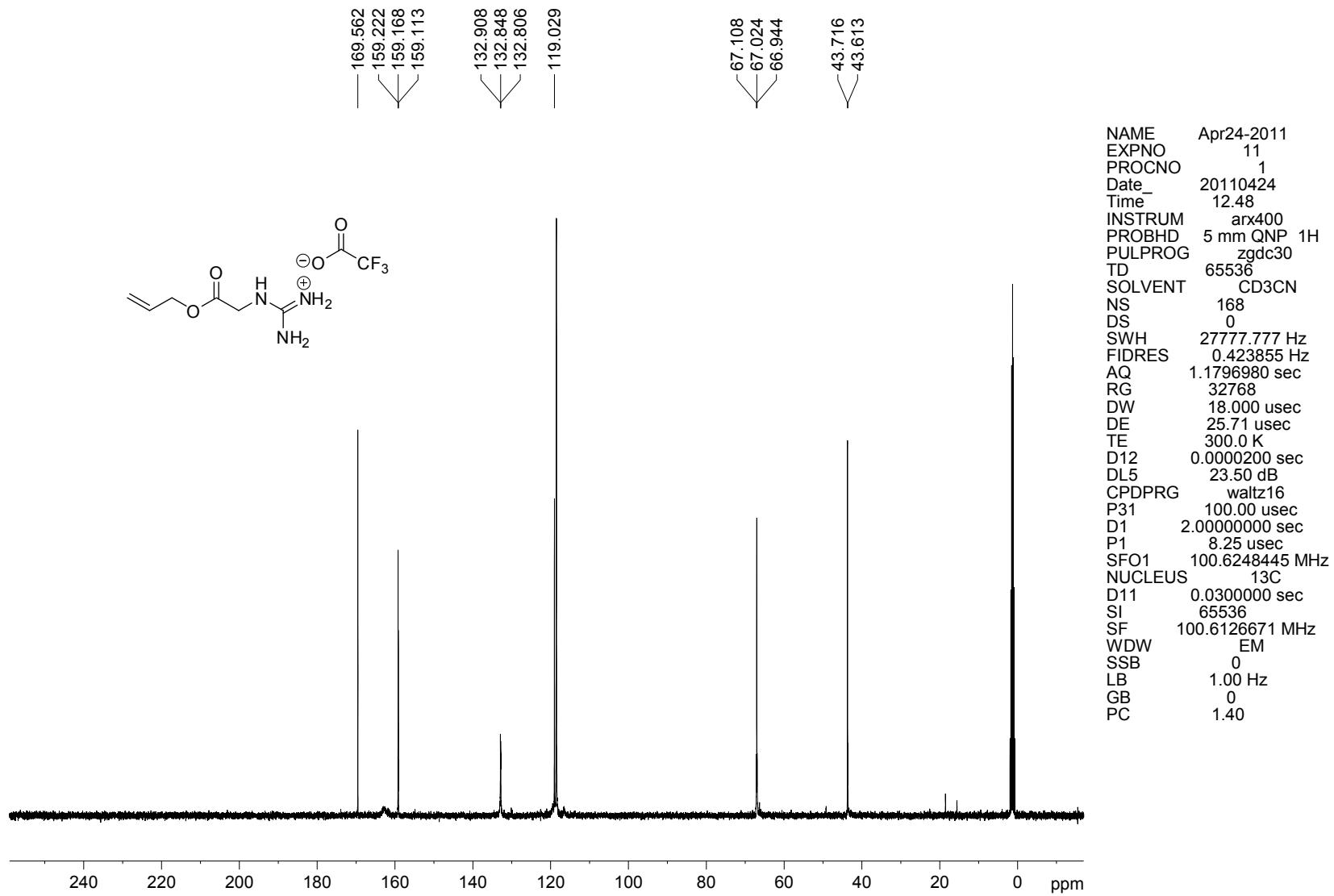
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DW	62,000 usec
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TE	300.0 K
D1	2.00000000 sec
P1	7.50 usec
SFO1	400.1324008 MHz
NUCLEUS	1H
SI	65536
SF	400.1300173 MHz
WDW	EM
SSB	0
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(S)-tert-Butyl-2-di(tert-butyloxycarbonylmethyl)amino-3-(pent-4-enyloxy)-propanoate

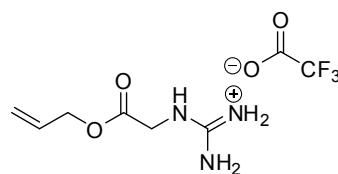




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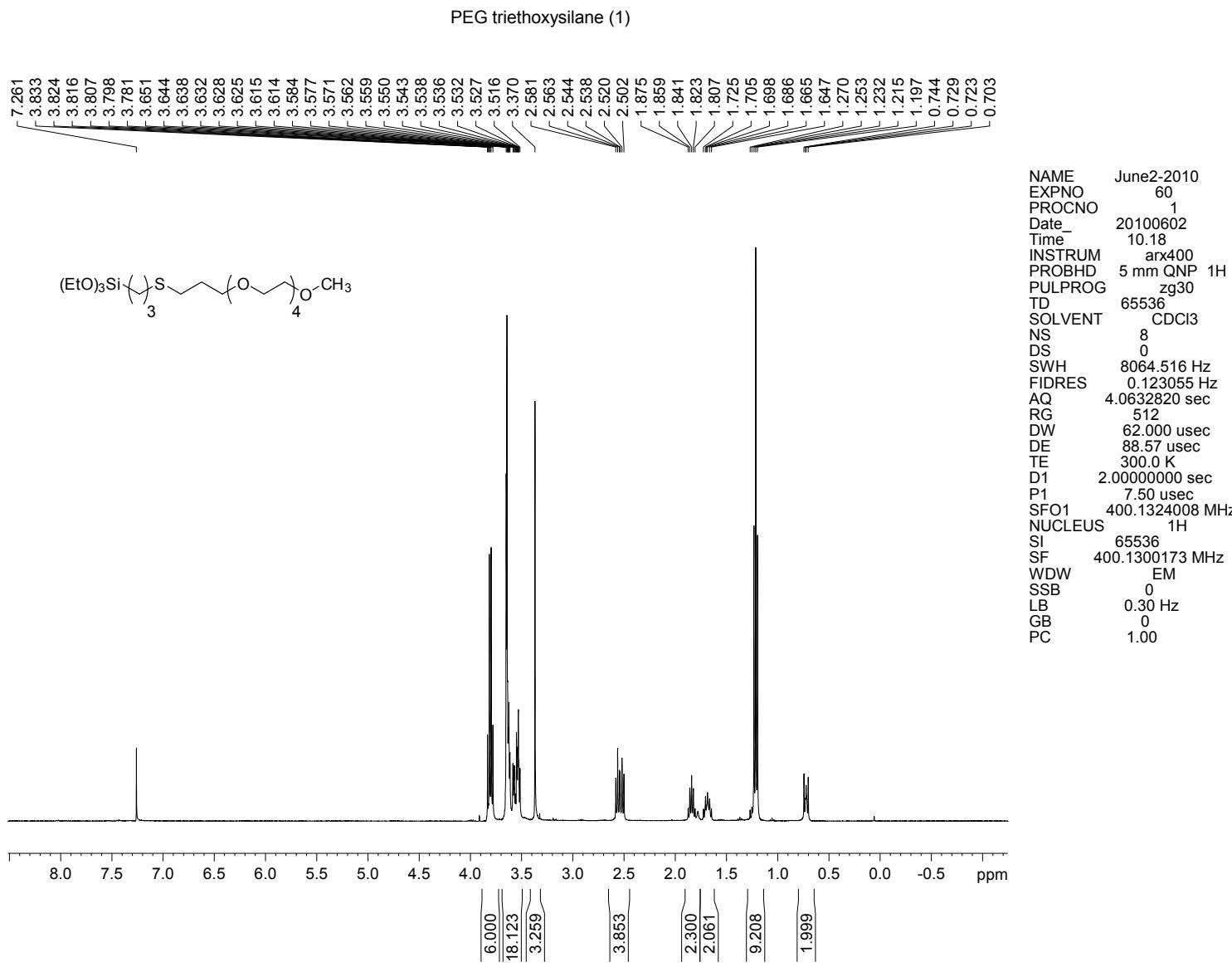
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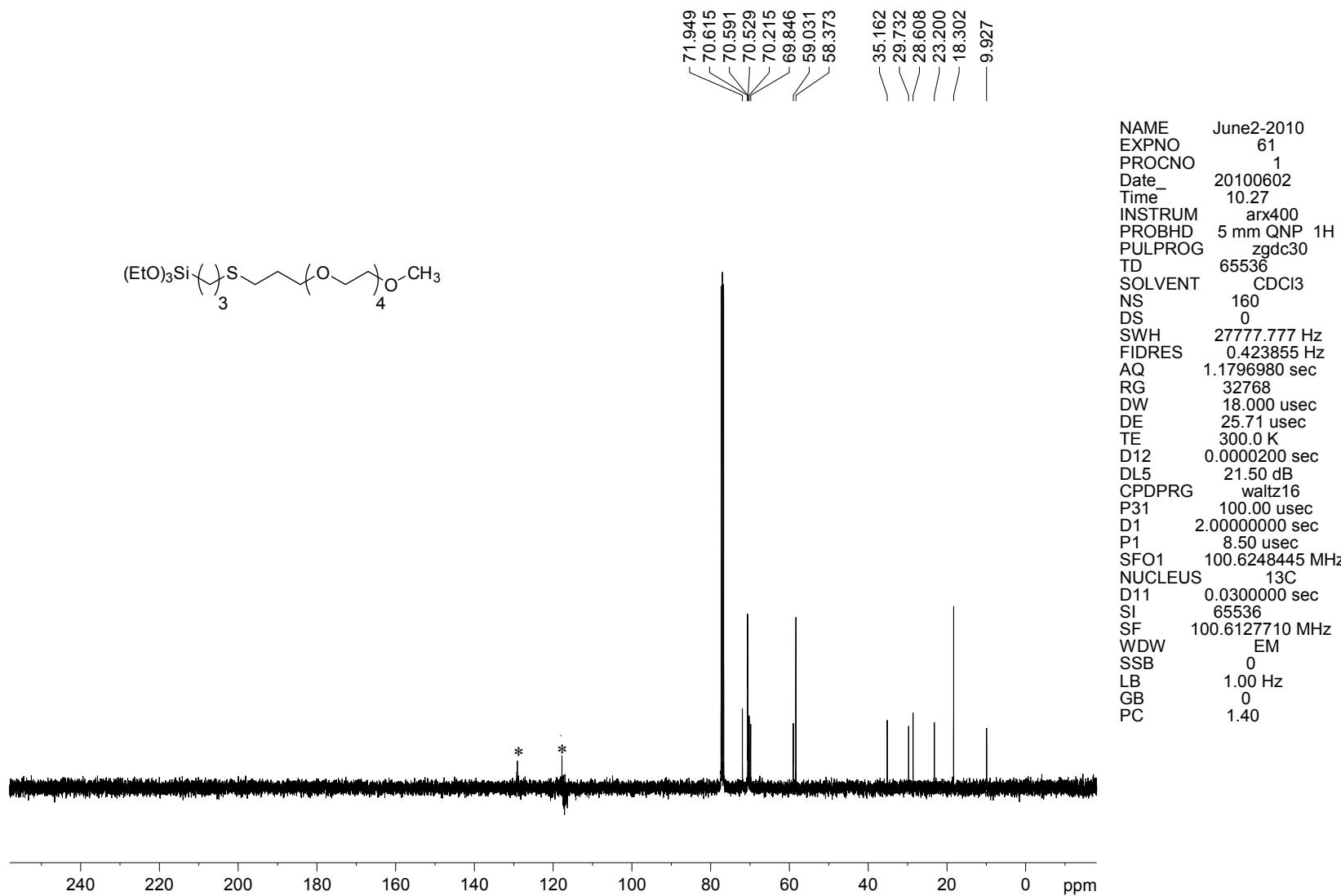
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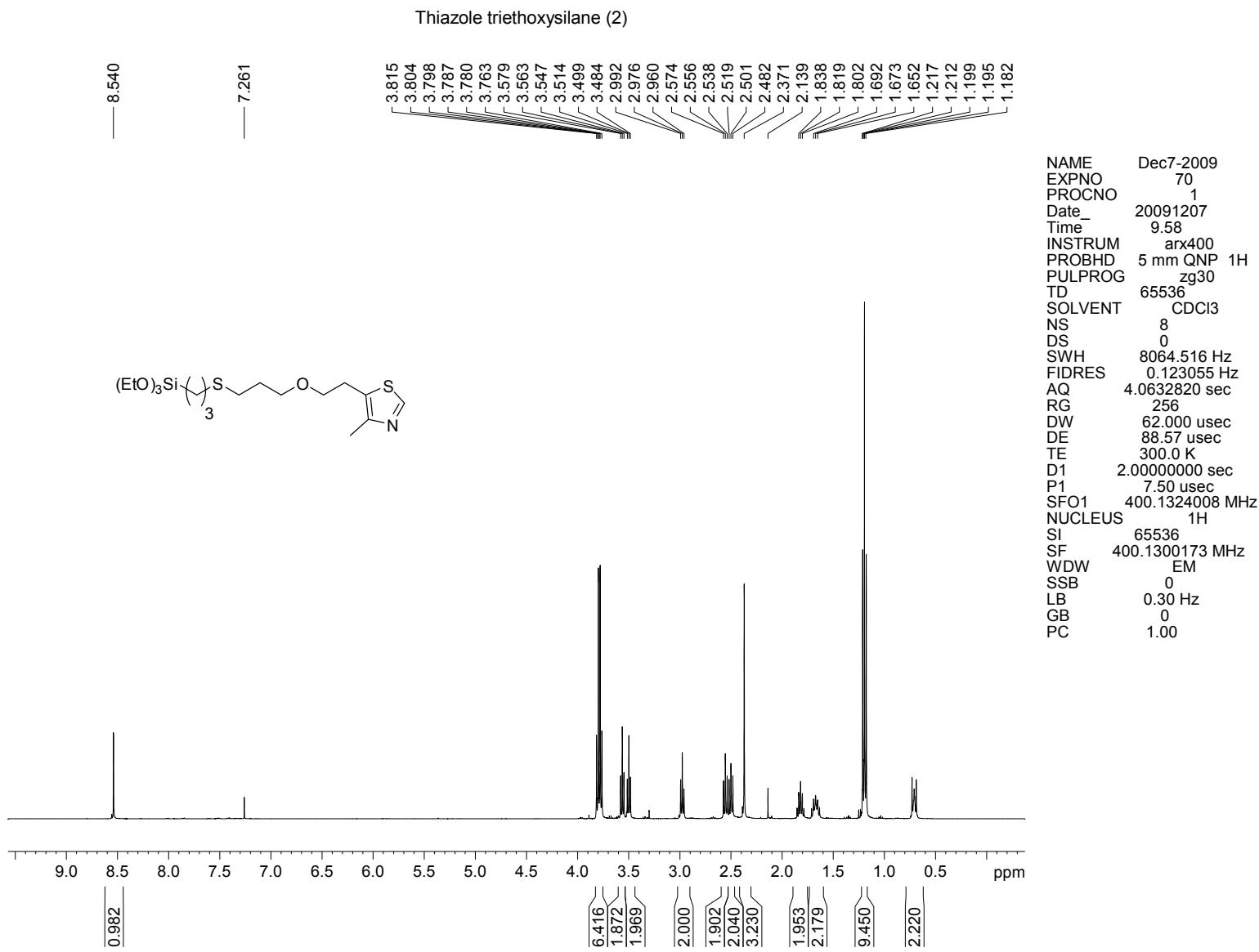
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TD	262144
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NS	8
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FIDRES	0.476837 Hz
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RG	2860
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TE	300.0 K
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P1	10.90 usec
D13	0.00000300 sec
DL5	23.50 dB
SFO1	376.4985950 MHz
NUCLEUS	¹⁹ F
CPDPRG	waltz16
P31	100.00 usec
D11	0.03000000 sec
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SF	376.4985960 MHz
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PC	1.00

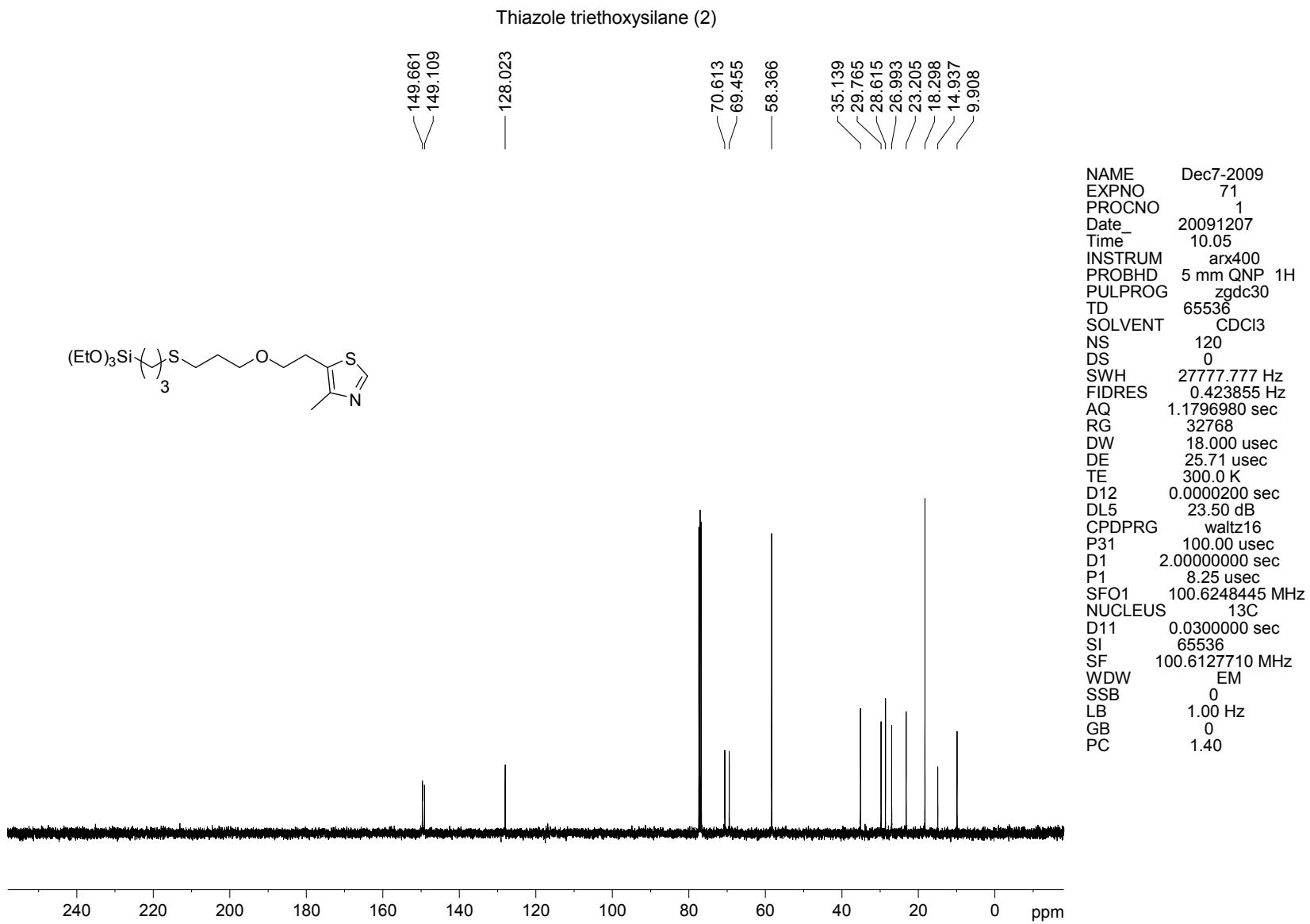
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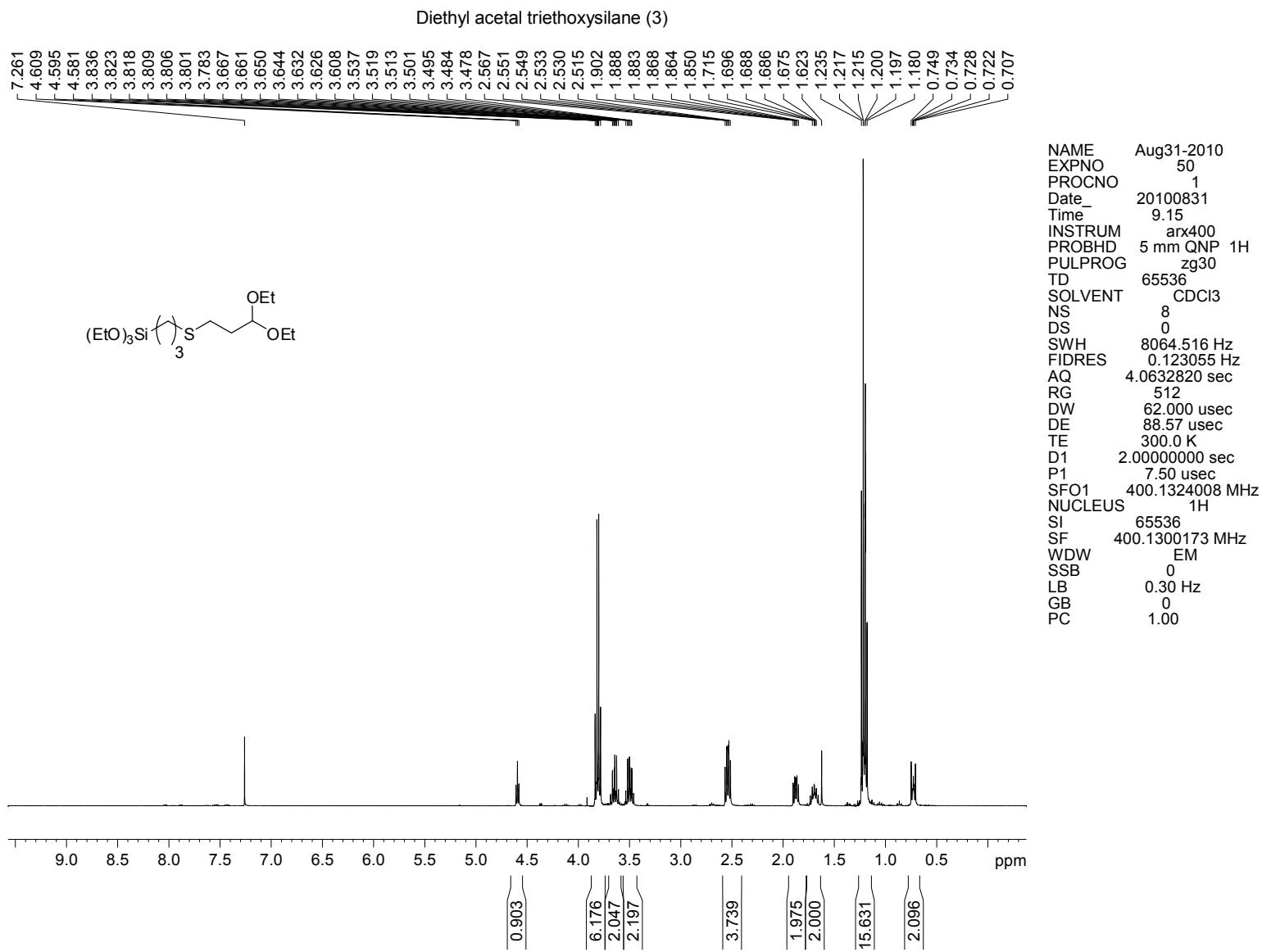


PEG triethoxysilane (1)

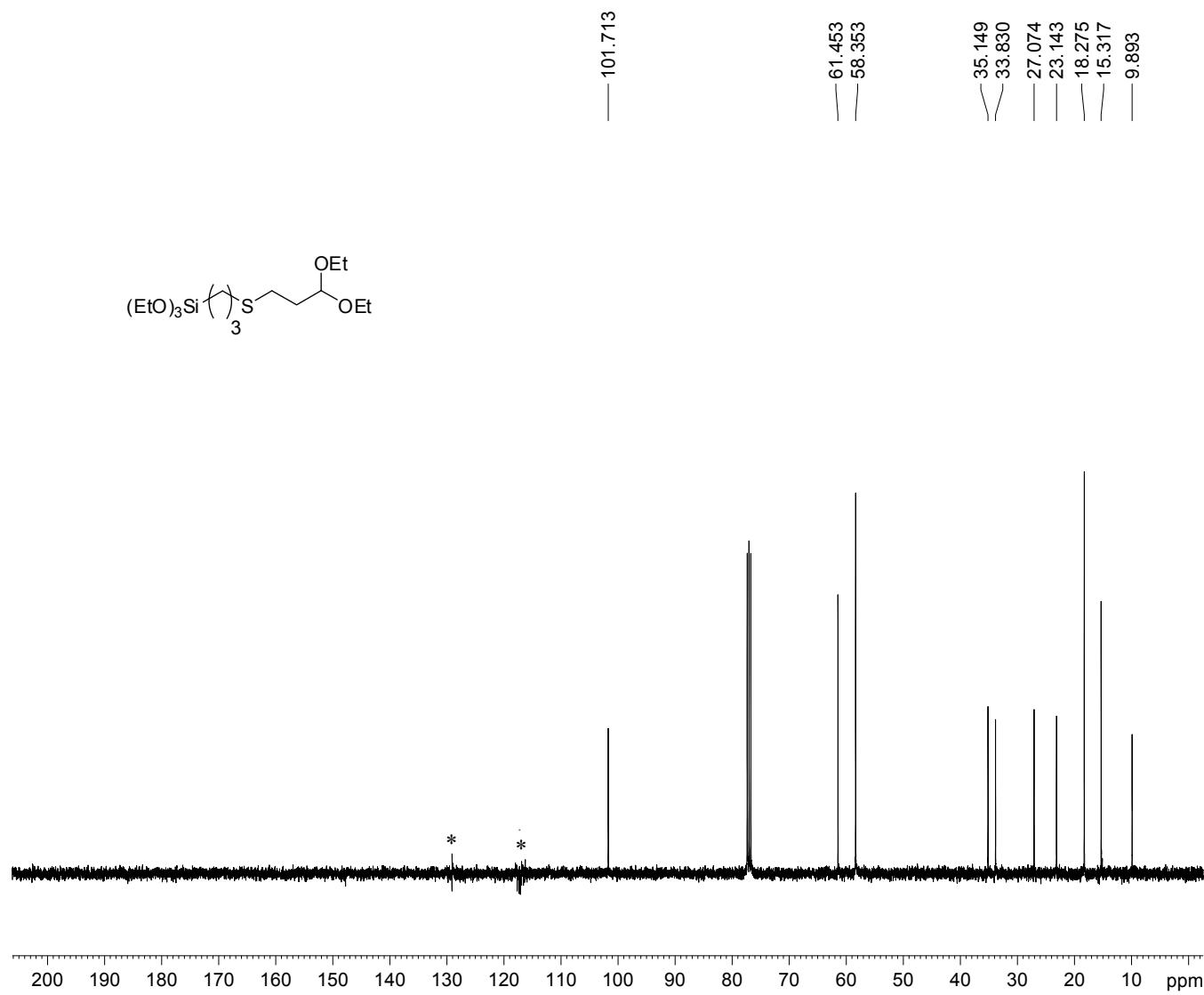




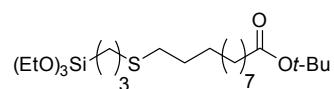




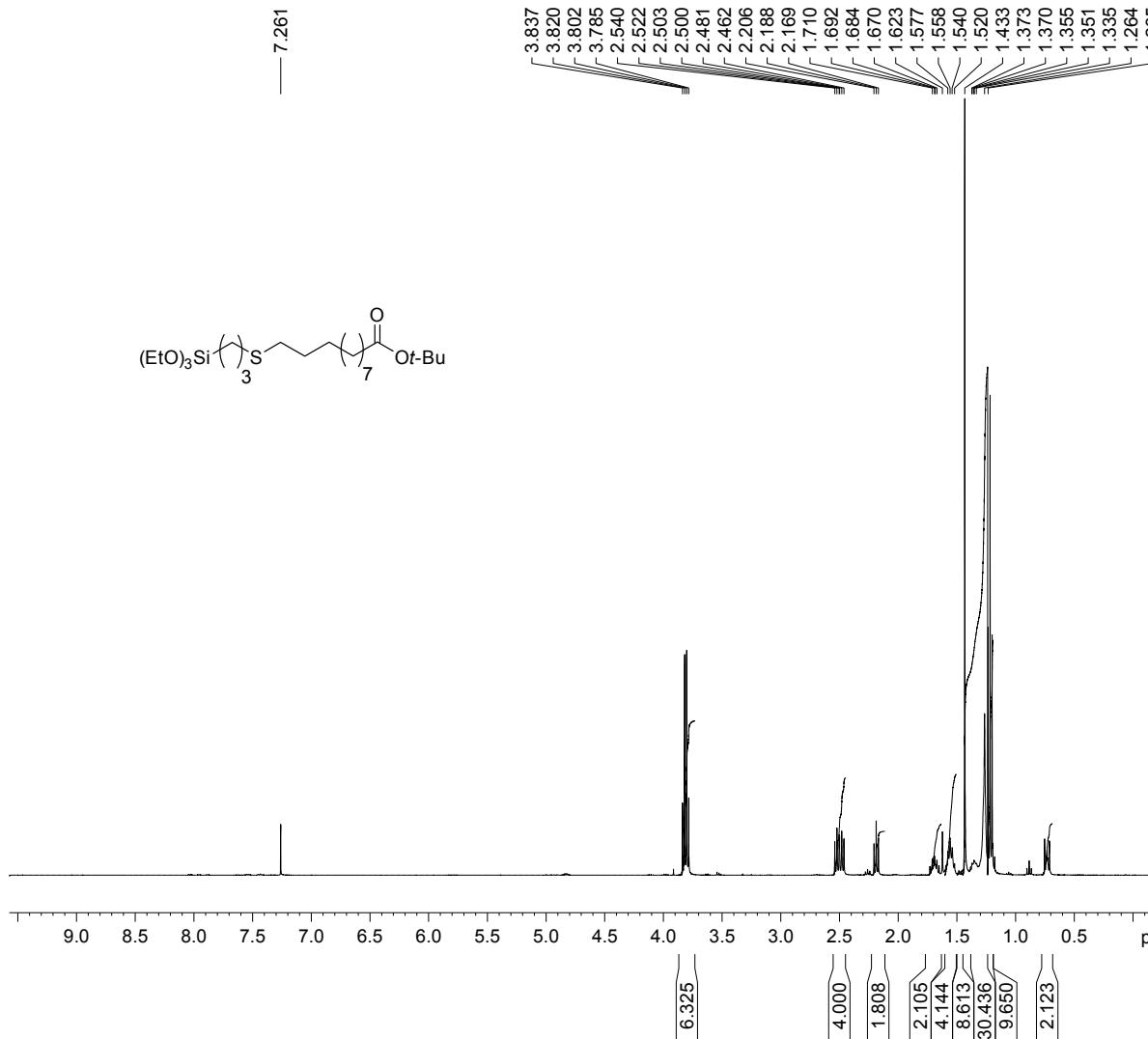
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tert-Butyl undecanoate triethoxysilane (4)

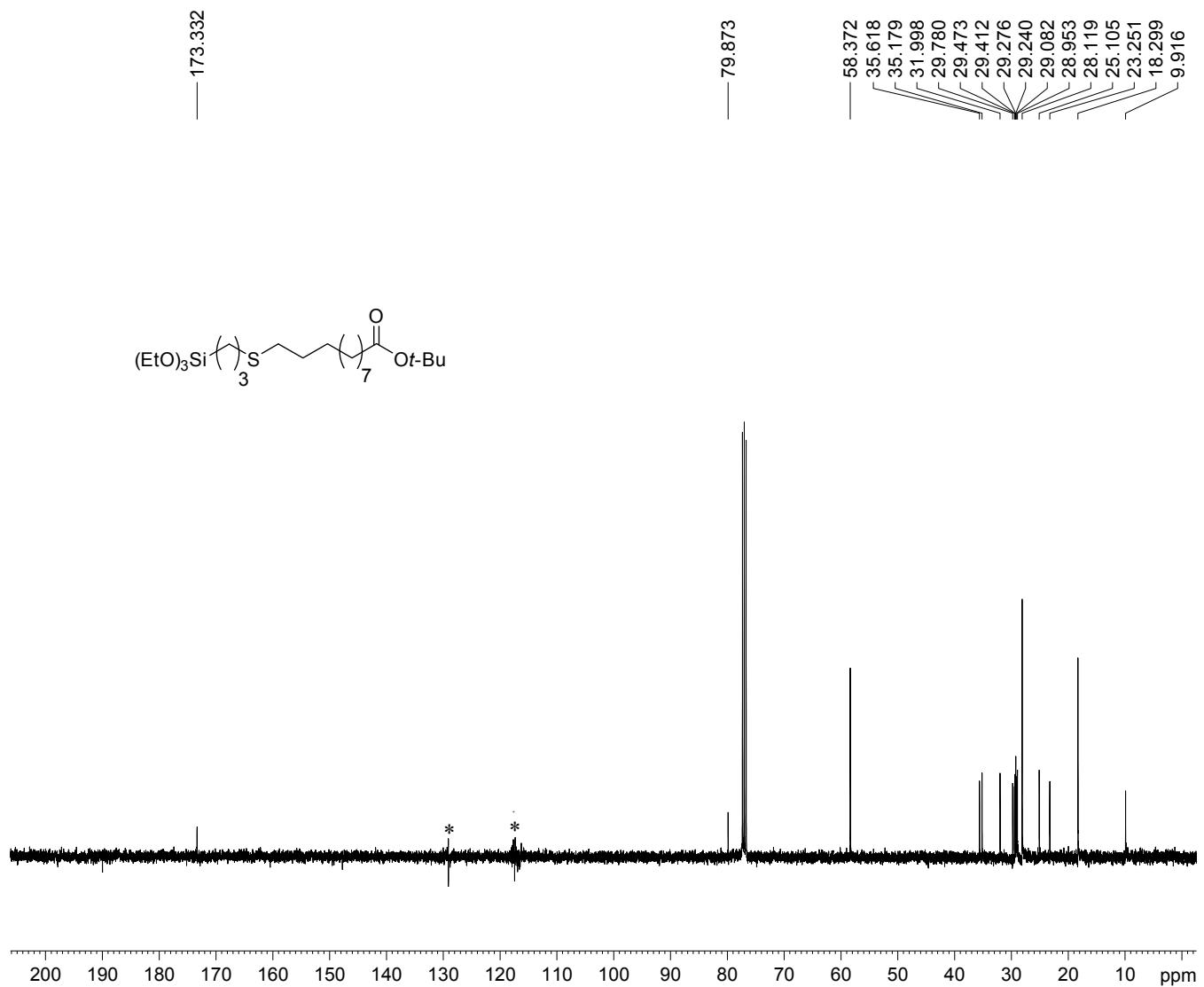


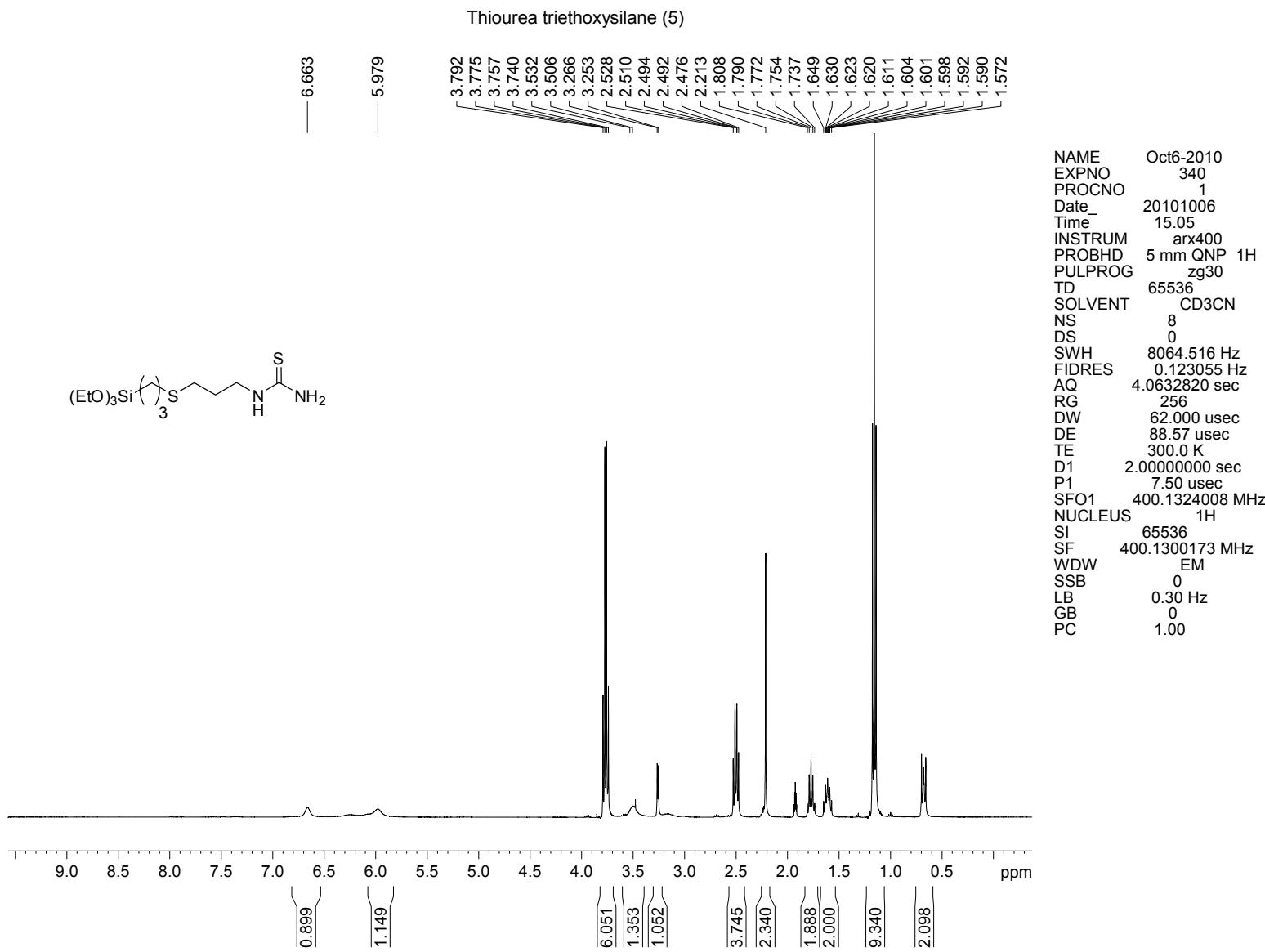
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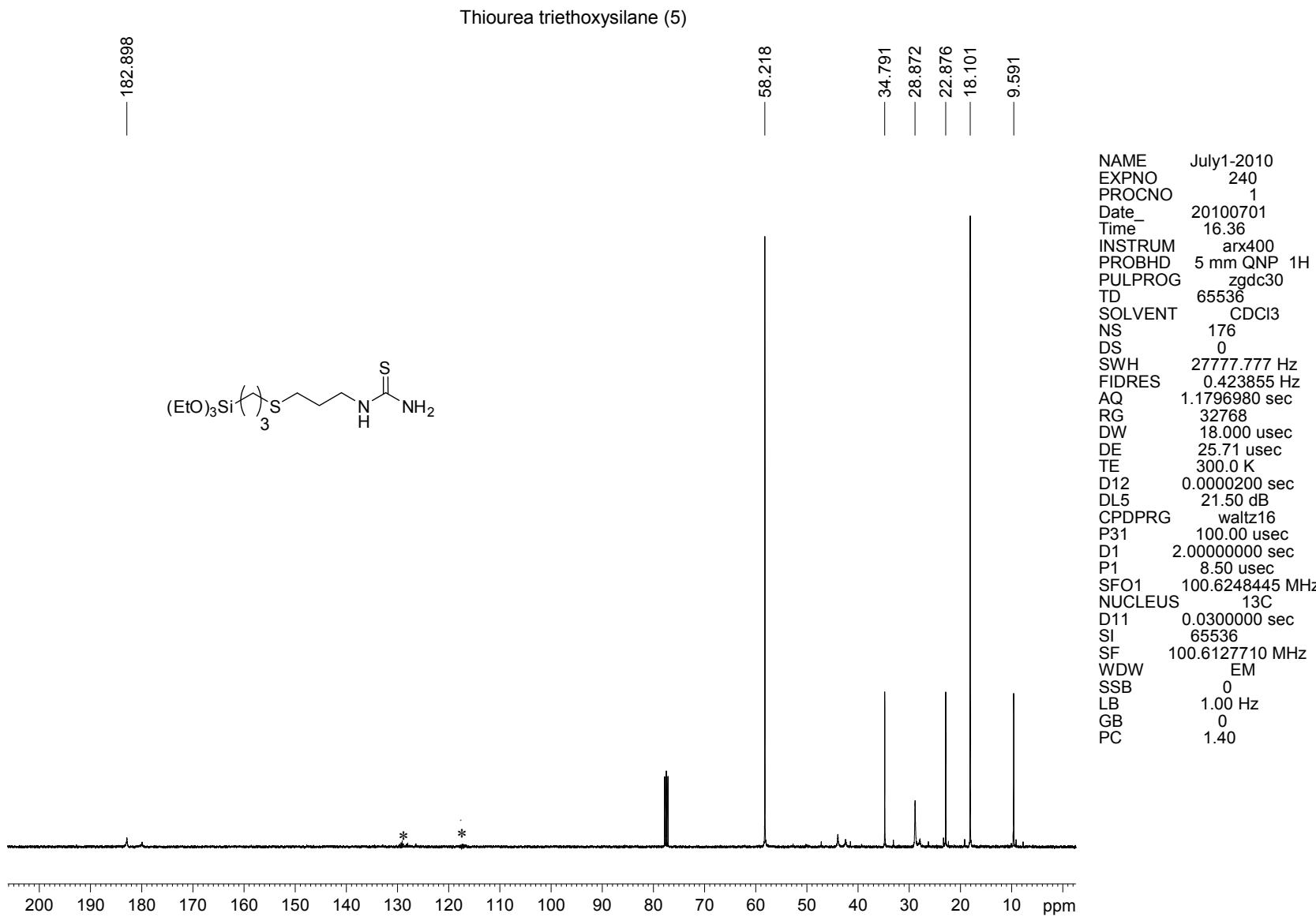


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TD	65536
SOLVENT	CDCl ₃
NS	8
DS	0
SWH	8064.516 Hz
FIDRES	0.123055 Hz
AQ	4.0632820 sec
RG	360
DW	62.0000 usec
DE	88.57 usec
TE	300.0 K
D1	2.0000000 sec
P1	7.50 usec
SFO1	400.1324008 MHz
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GB	0
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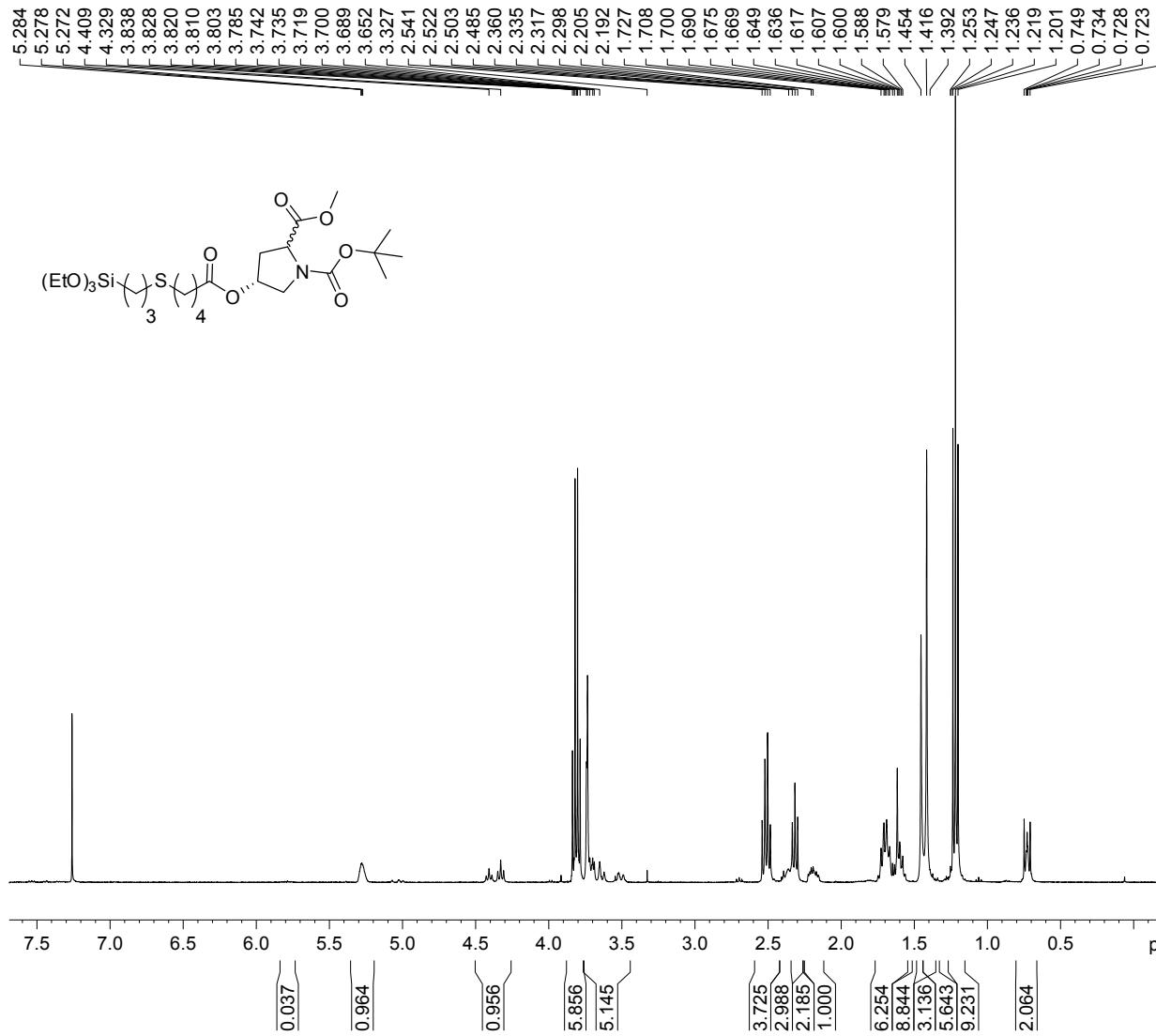
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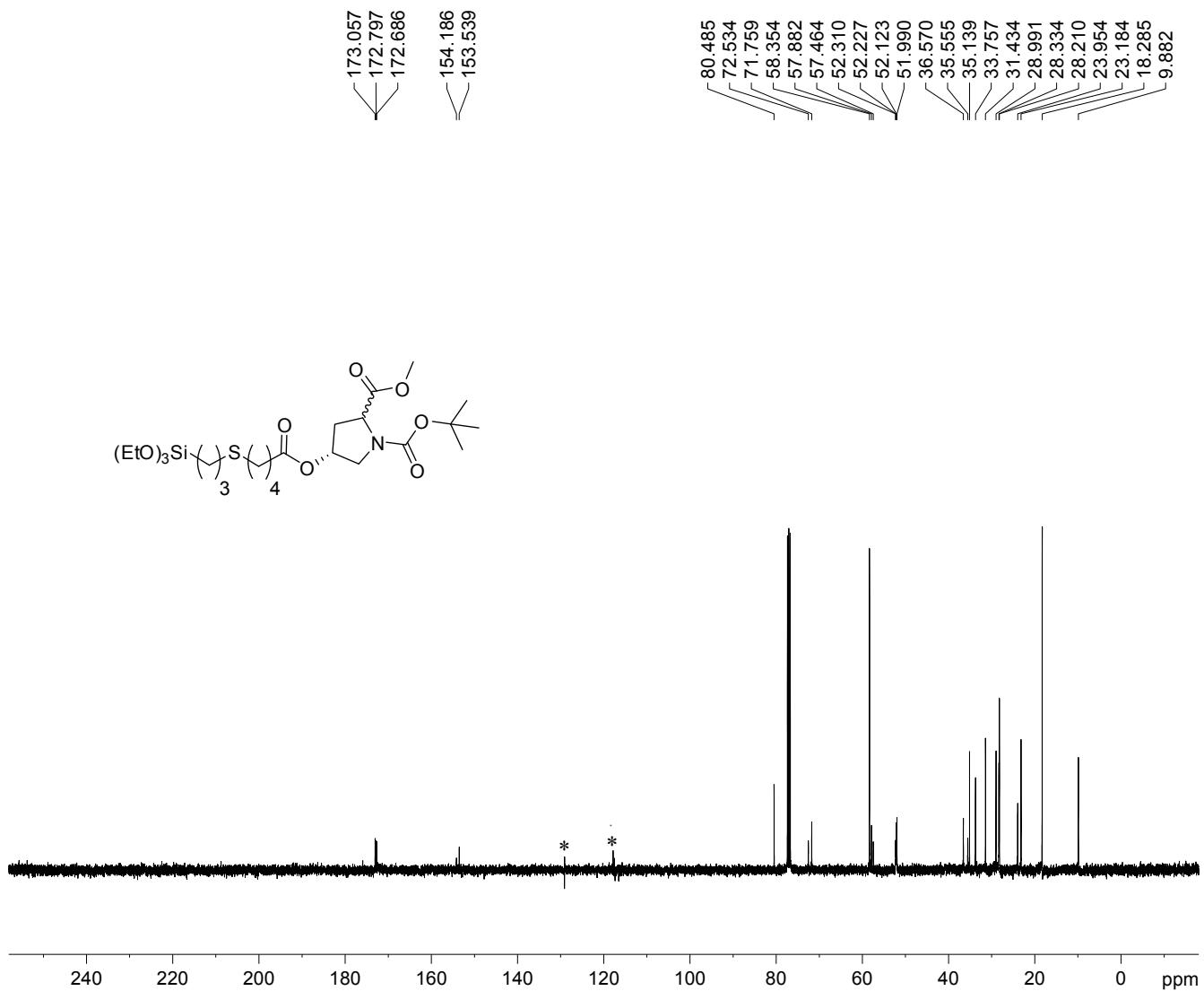


N-Boc-(4R)-4-hydroxy-L-proline methyl ester triethoxysilane (6)



NAME June29-2010
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 Time_ 10.49
 INSTRUM arx400
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 PULPROG zg30
 TD 65536
 SOLVENT CDCl3
 NS 8
 DS 0
 SWH 8064.516 Hz
 FIDRES 0.123055 Hz
 AQ 4.0632820 sec
 RG 715
 DW 62.000 usec
 DE 88.57 usec
 TE 300.0 K
 D1 2.0000000 sec
 P1 7.50 usec
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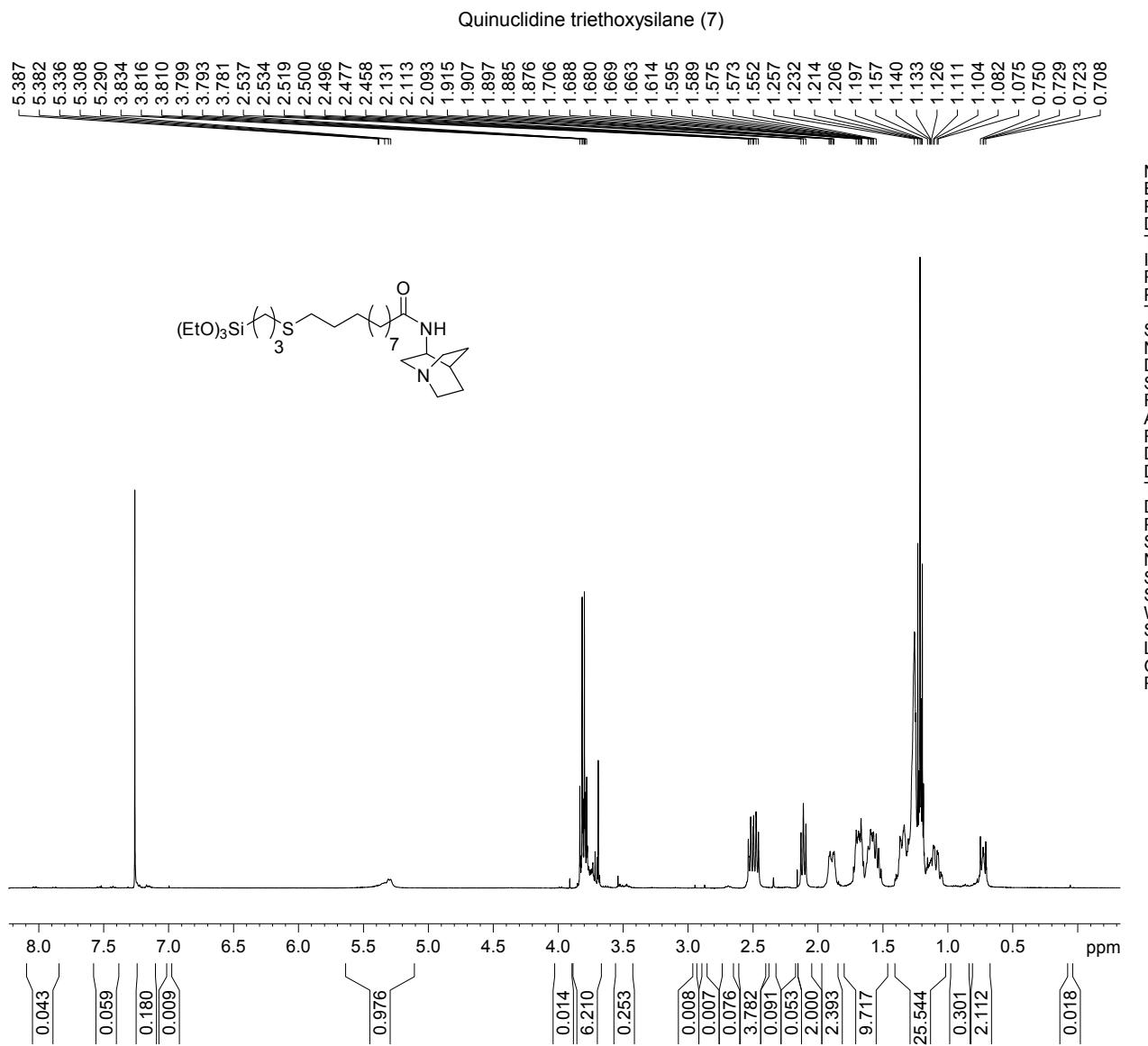
N-Boc-(4R)-4-hydroxy-L-proline methyl ester triethoxysilane (6)



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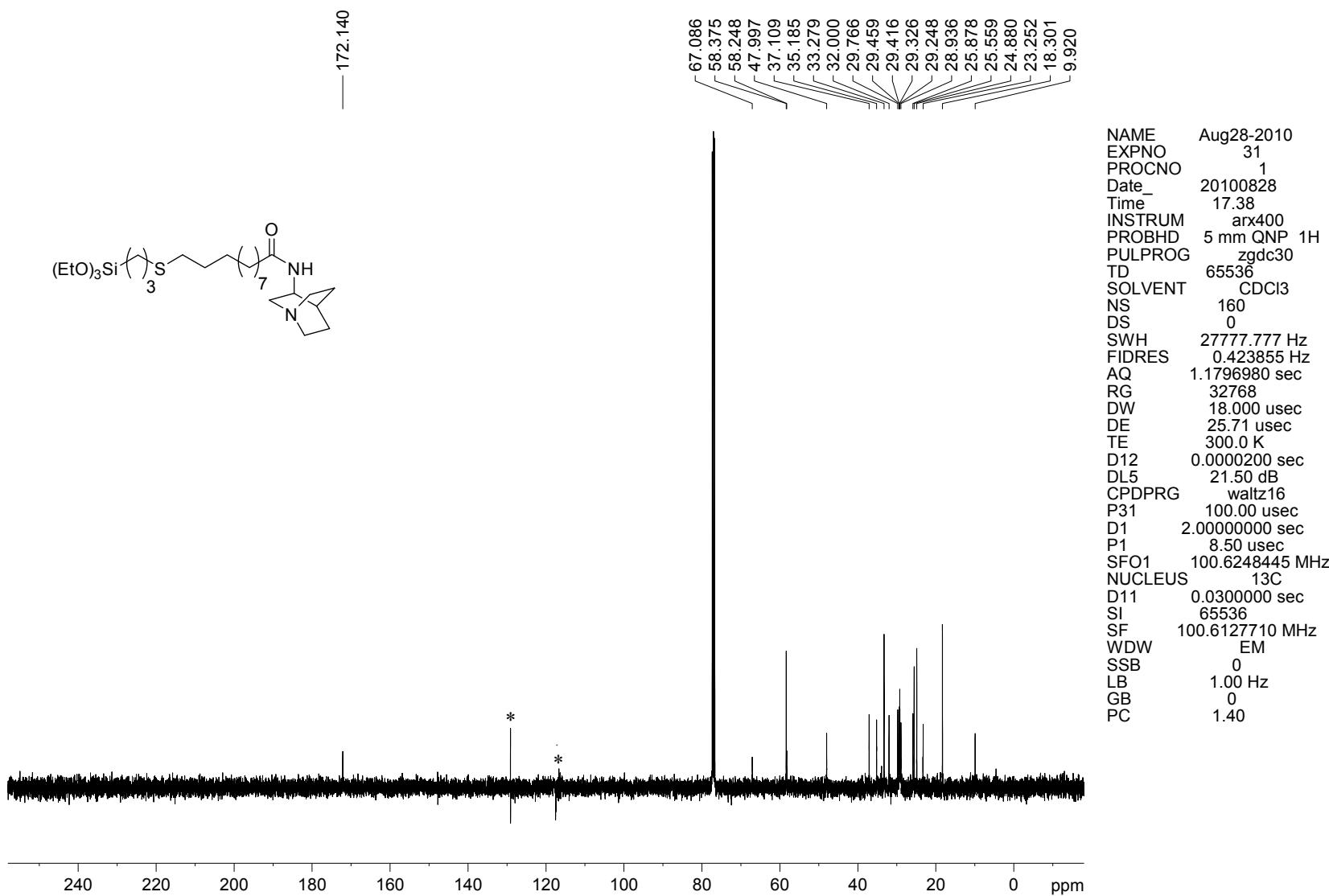
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NS        176
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FIDRES   0.423855 Hz
AQ        1.1796980 sec
RG        32768
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DE        25.71 usec
TE        300.0 K
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DLS       21.50 dB
CPDPRG   waltz16
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D1        2.0000000 sec
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PC       1.40

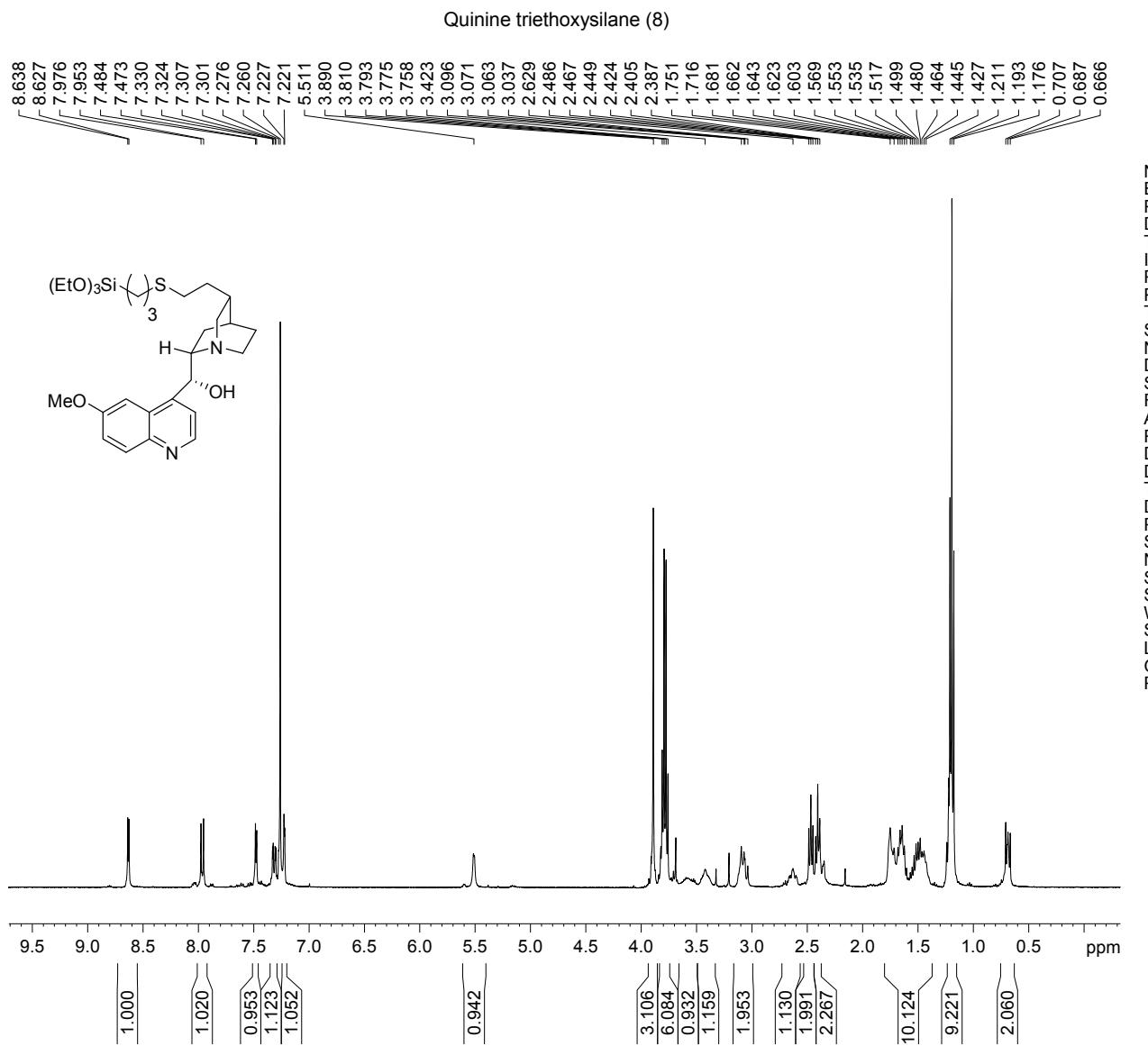
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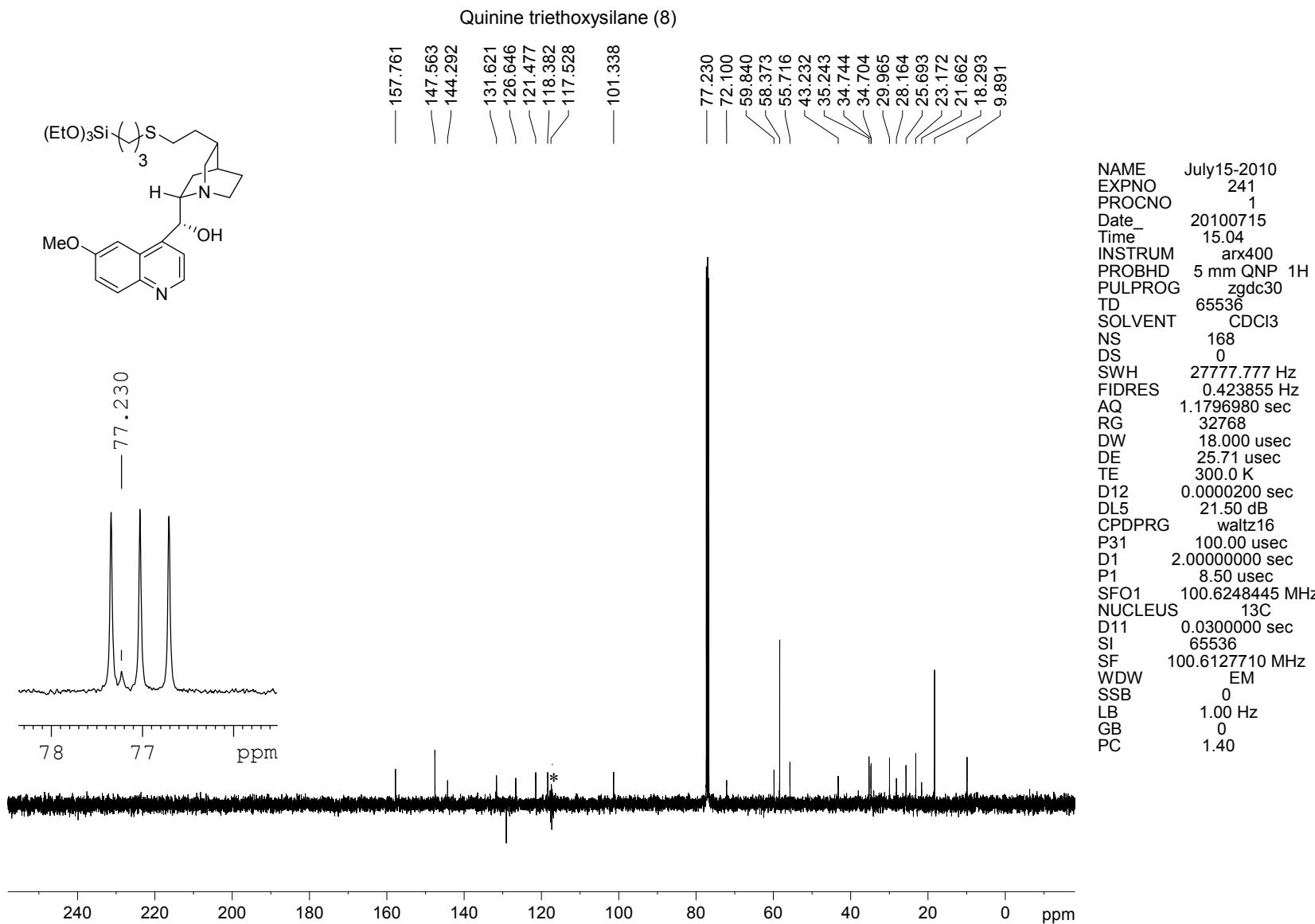
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 SOLVENT CDCl₃
 NS 24
 DS 0
 SWH 8064.516 Hz
 FIDRES 0.123055 Hz
 AQ 4.0632820 sec
 RG 360
 DW 62.000 usec
 DE 88.57 usec
 TE 300.0 K
 D1 2.0000000 sec
 P1 7.50 usec
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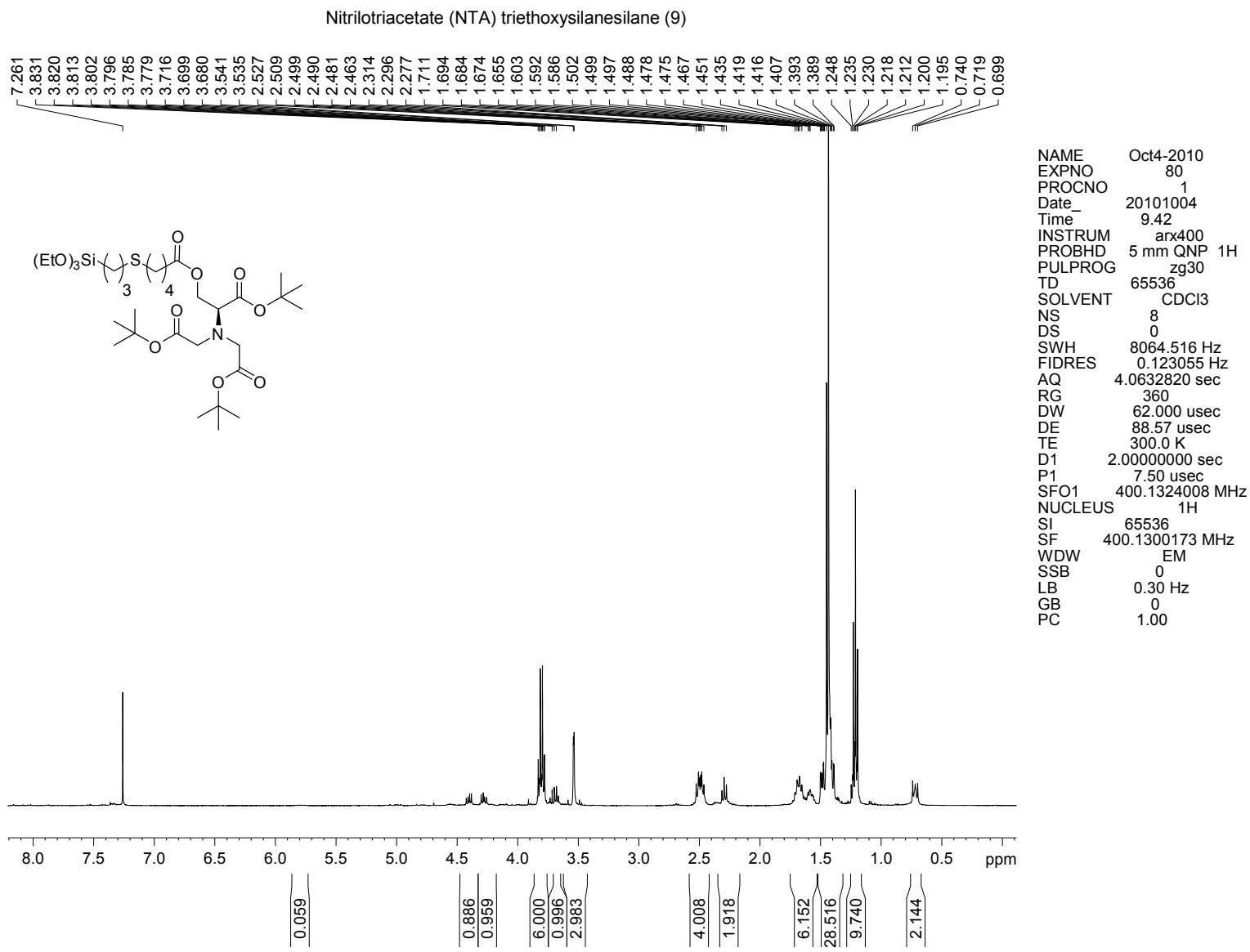
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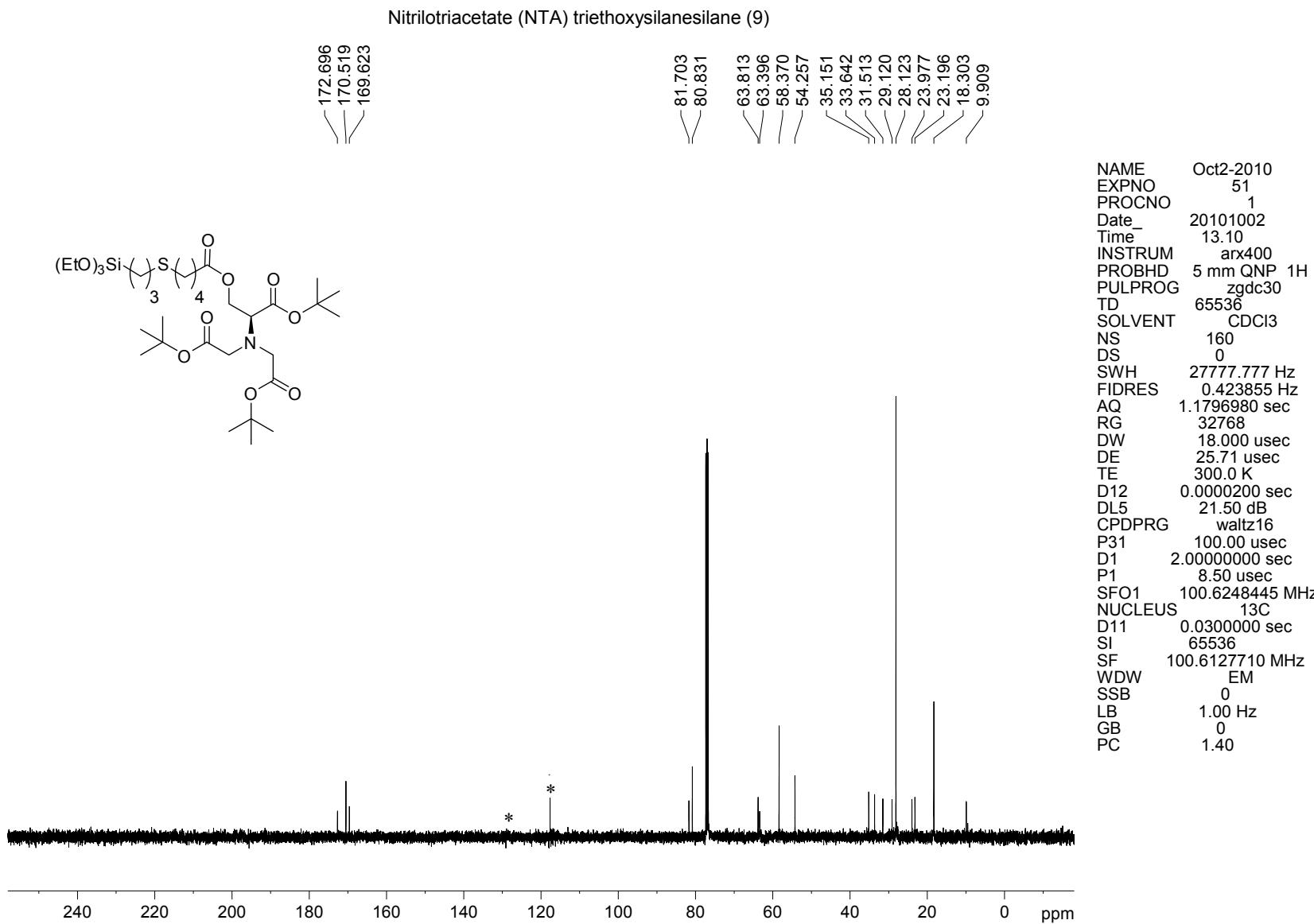




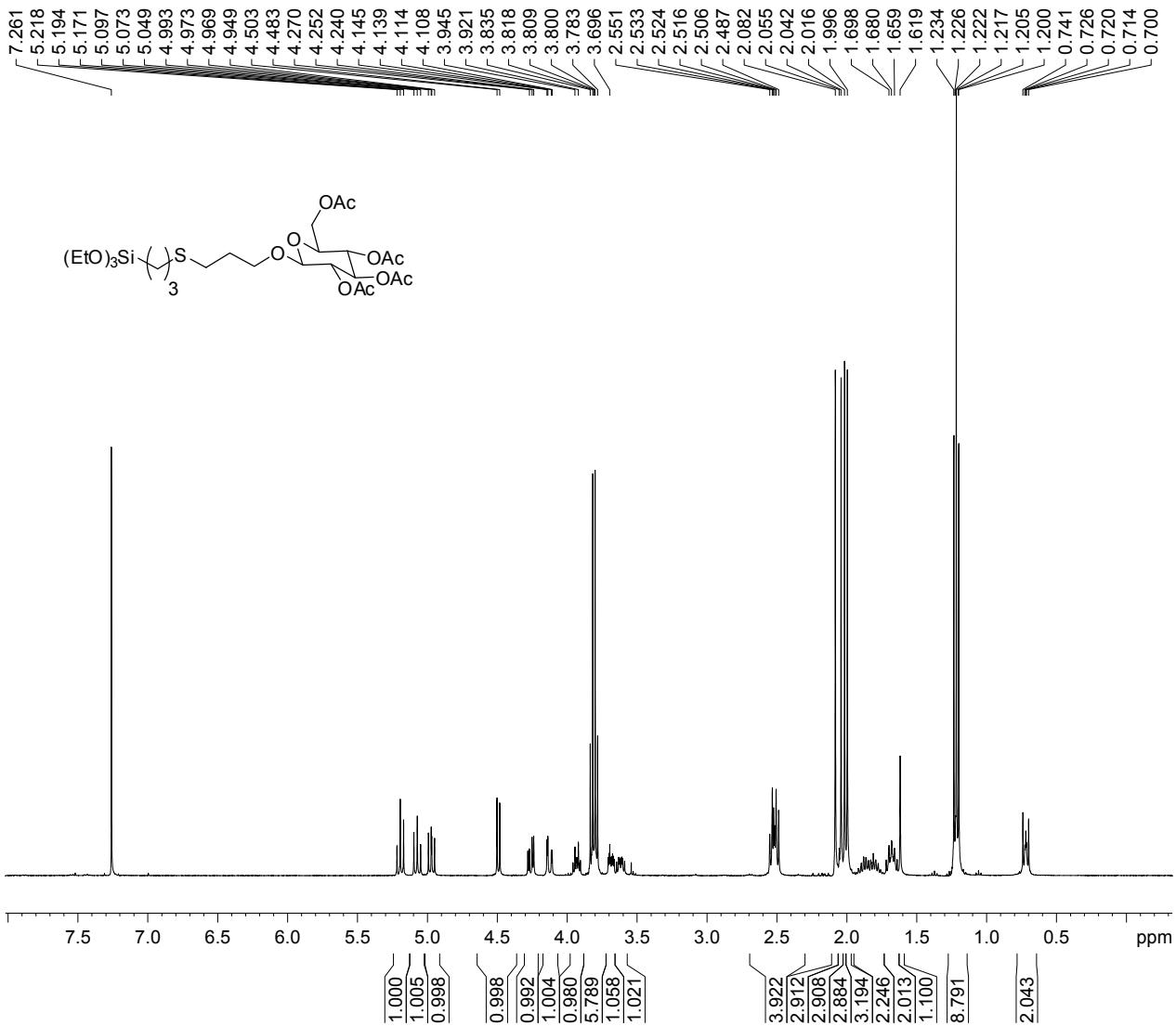
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 SOLVENT CDCl₃
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 FIDRES 0.123055 Hz
 AQ 4.0632820 sec
 RG 715
 DW 62.000 usec
 DE 88.57 usec
 TE 300.0 K
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 P1 7.50 usec
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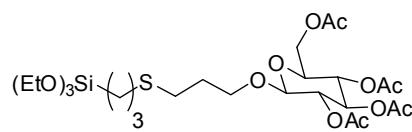




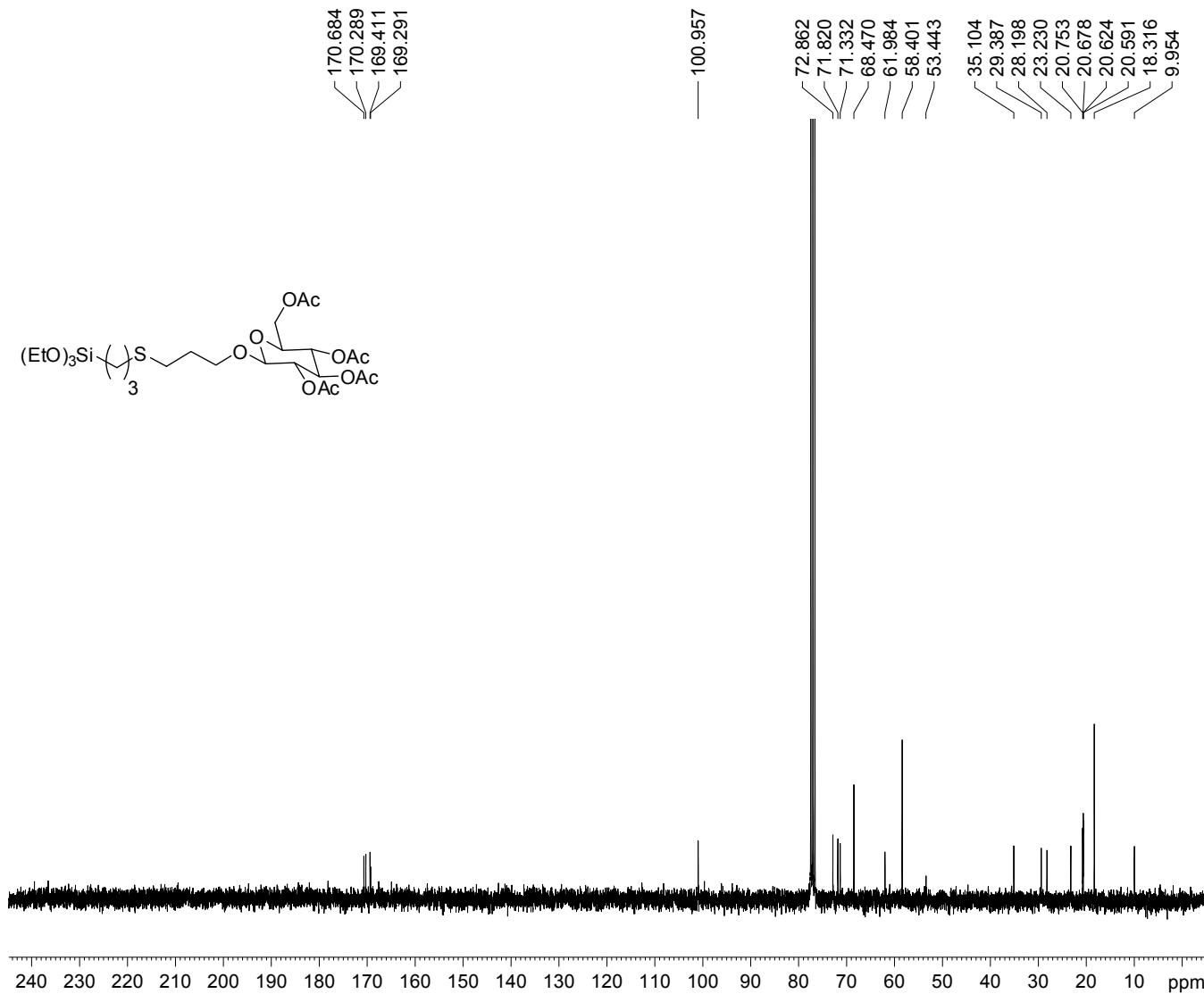
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TD	65536
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SWH	8064.516 Hz
FIDRES	0.123055 Hz
AQ	4.063280 sec
RG	715
DW	62.000 usec
DE	88.57 usec
TE	300.0 K
D1	2.0000000 sec
P1	7.50 usec
SFO1	400.1324008 MHz
NUCLEUS	1H
SI	65536
SF	400.1300173 MHz
WDW	EM
SSB	0
LB	0.30 Hz
GB	0
PC	1.00



Tetra-O-acetyl-(beta)-D-glucopyranoside triethoxysilane (10)

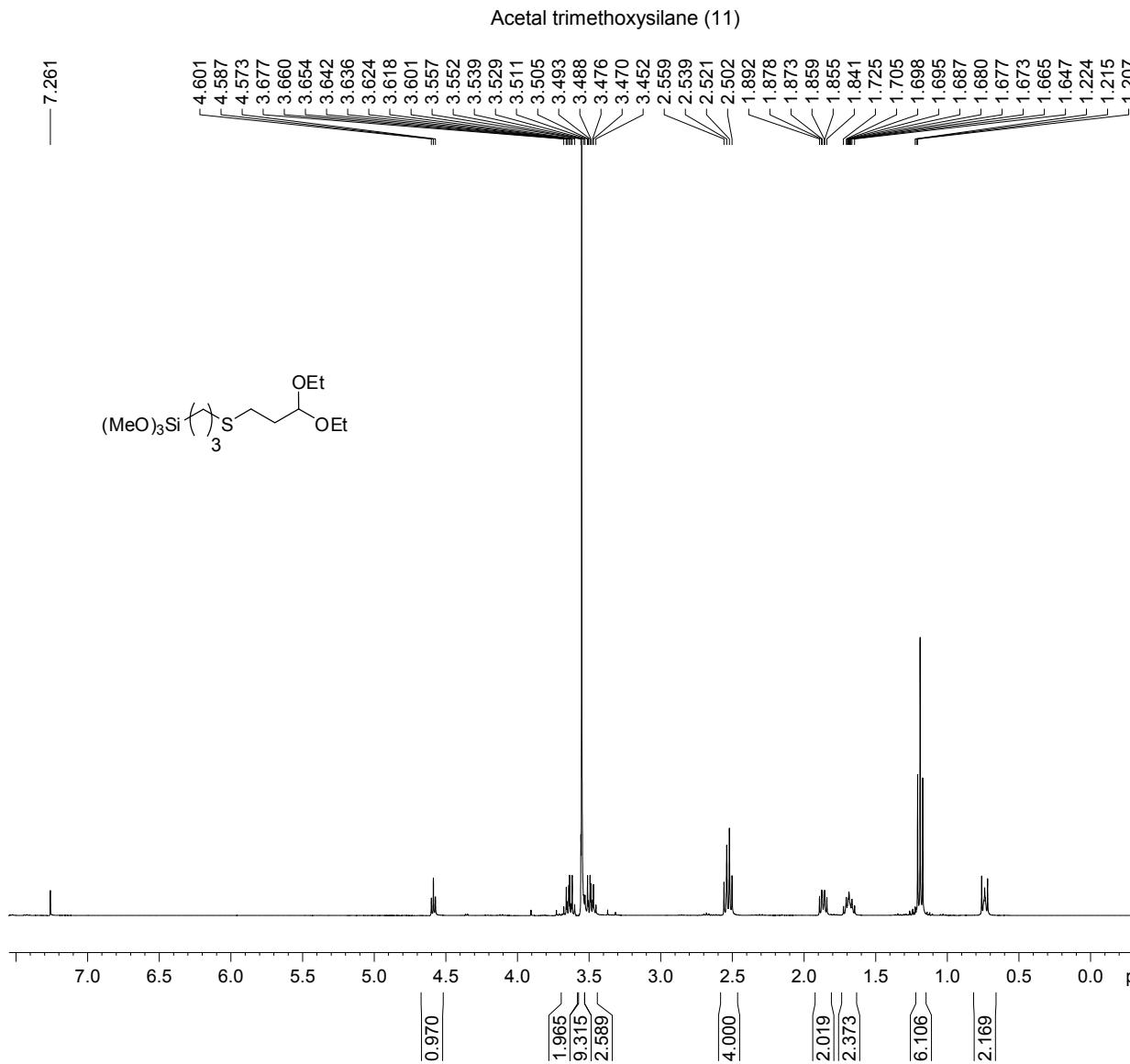


NAME	silane	Thio-enes
EXPNO	6	
PROCNO	1	
Date_	20100607	
Time	10.32	
INSTRUM	av300	
PROBHD	5 mm PABB0 BB-	
PULPROG	zgdc30	
TD	65536	
SOLVENT	CDCl3	
NS	320	
DS	0	
SWH	18832.393 Hz	
FIDRES	0.287360 Hz	
AQ	1.7400308 sec	
RG	32768	
DW	26.550 usec	
DE	6.00 usec	
TE	299.7 K	
D1	2.0000000 sec	
d11	0.03000000 sec	
TD0	1	

===== CHANNEL f1 =====
NUC1 13C
P1 7.00 usec
PL1 -4.00 dB
SFO1 75.4768051 MHz

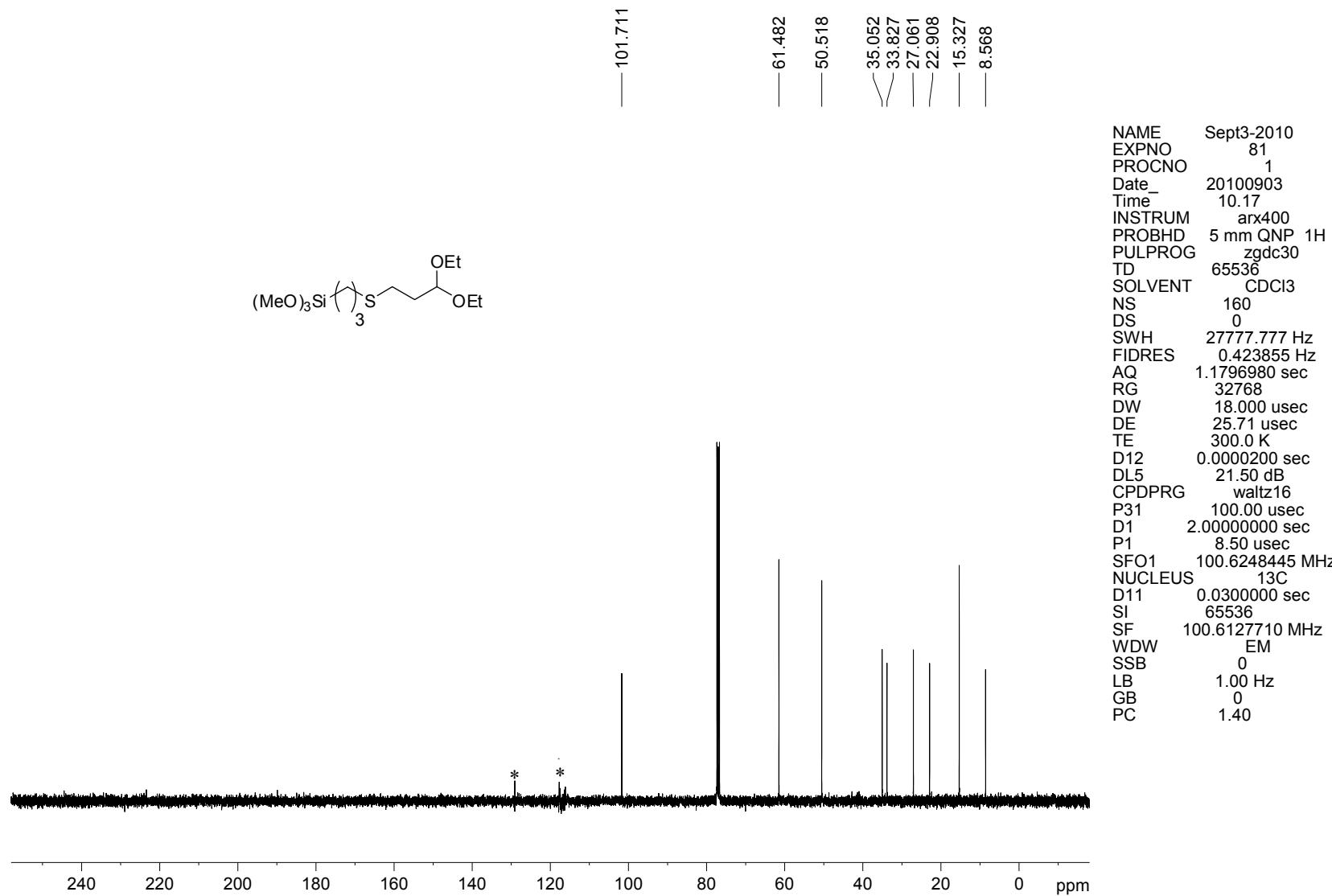
```
===== CHANNEL f2 =====
CPDPRG2      waltz16
NUC2          1H
PCPD2        80.00 usec
PL2           -2.00 dB
PL12          14.85 dB
SFO2          300.1318008 MHz
SI            32768
SF             75.4677490 MHz
WDW           EM
SSB           0
LB            1.00 Hz
GB           0
PC           140
```

— 7.261



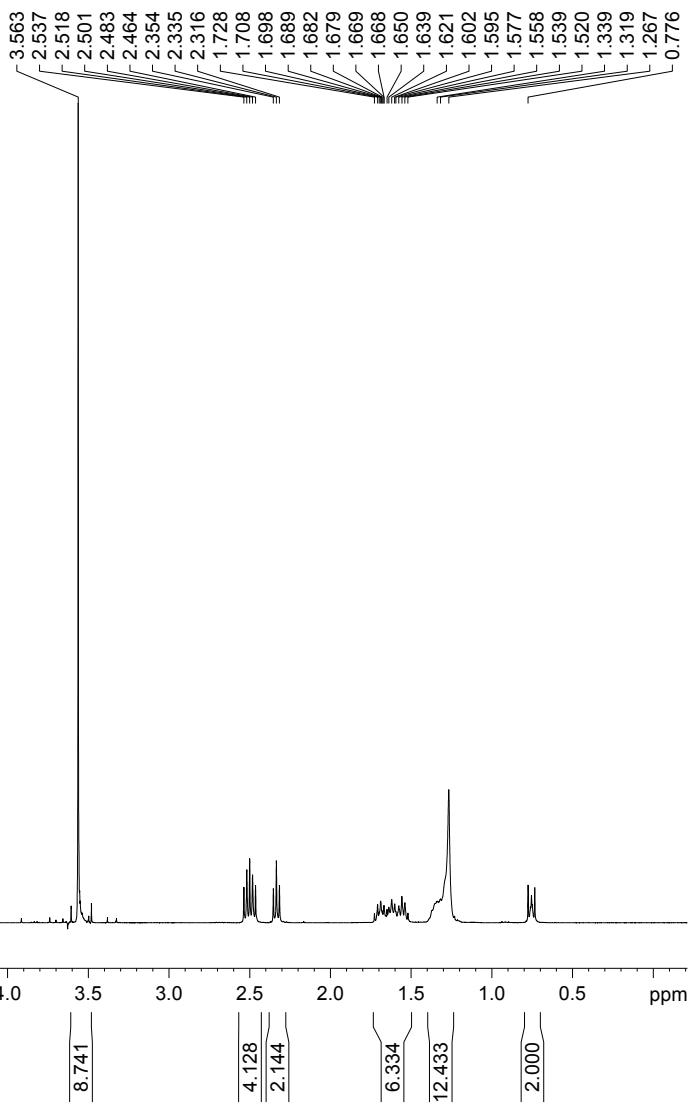
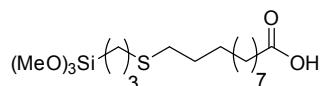
NAME Sept3-2010
EXPNO 80
PROCNO 1
Date_ 20100903
Time_ 10.08
INSTRUM arx400
PROBHD 5 mm QNP 1H
PULPROG zg30
TD 65536
SOLVENT CDCl₃
NS 8
DS 0
SWH 8064.516 Hz
FIDRES 0.123055 Hz
AQ 4.0632820 sec
RG 360
DW 62.000 usec
DE 88.57 usec
TE 300.0 K
D1 2.0000000 sec
P1 7.50 usec
SFO1 400.1324008 MHz
NUCLEUS 1H
SI 65536
SF 400.1300173 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

Acetal trimethoxysilane (11)



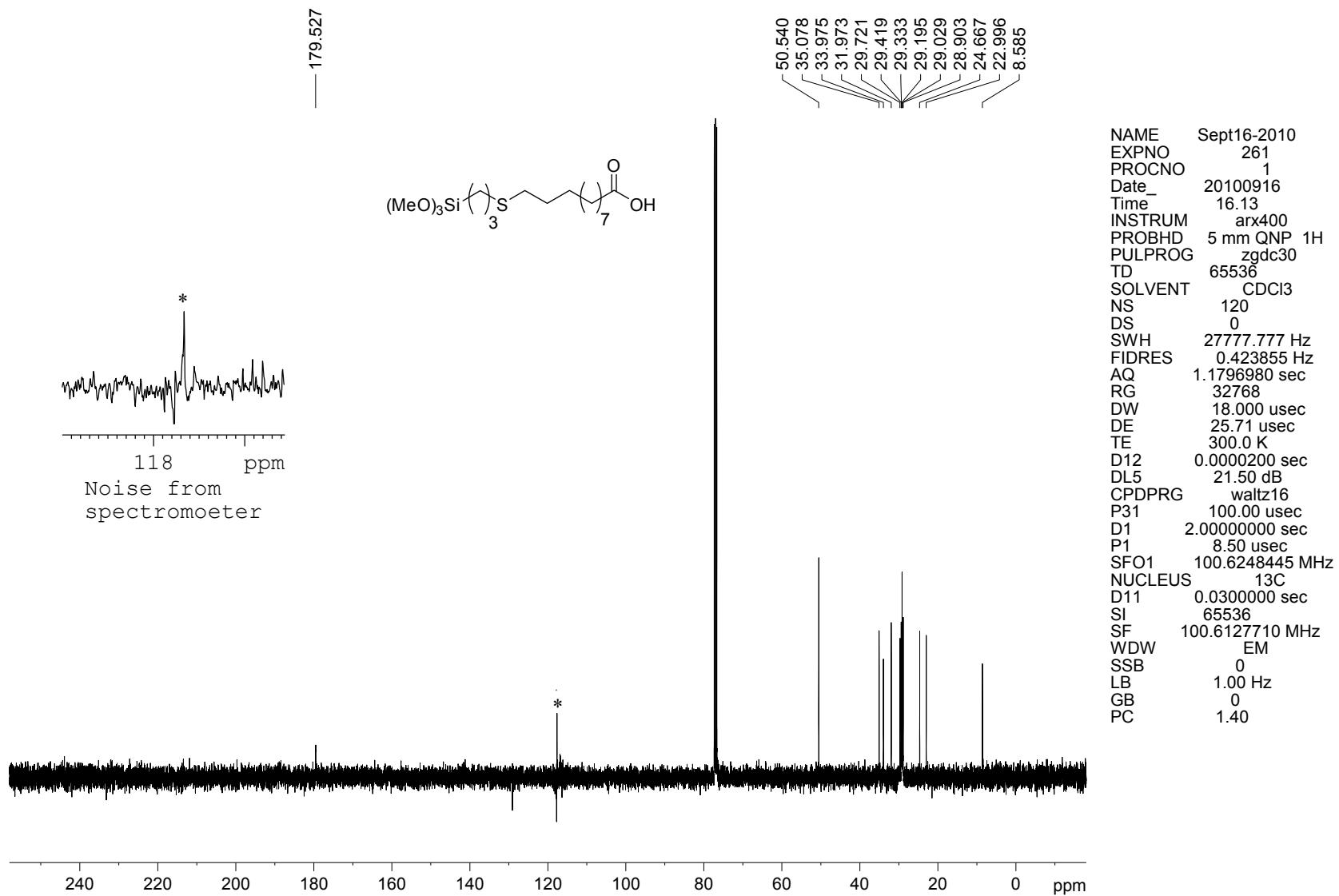
— 7.260

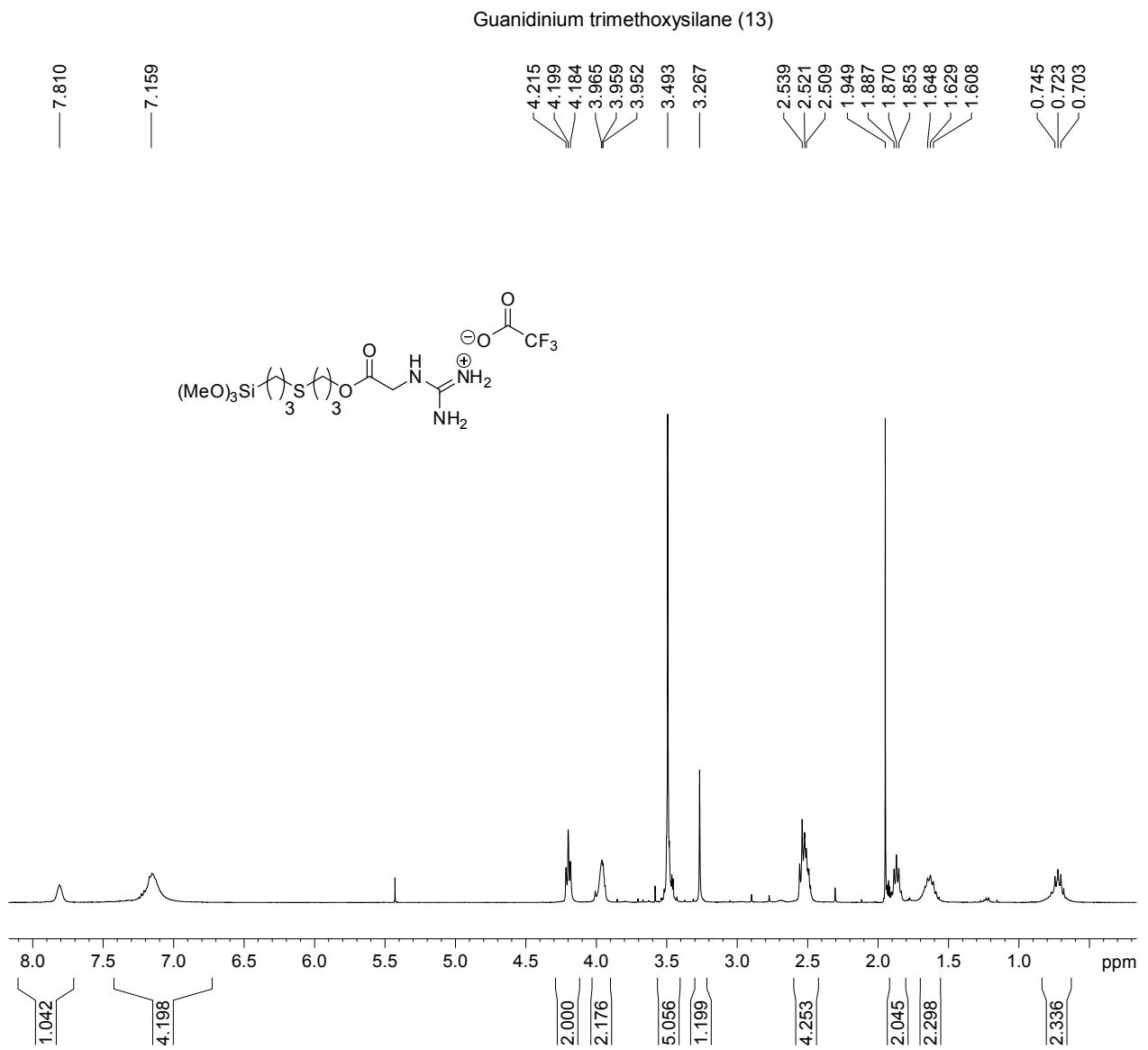
Undecanoic acid trimethoxysilane (12)



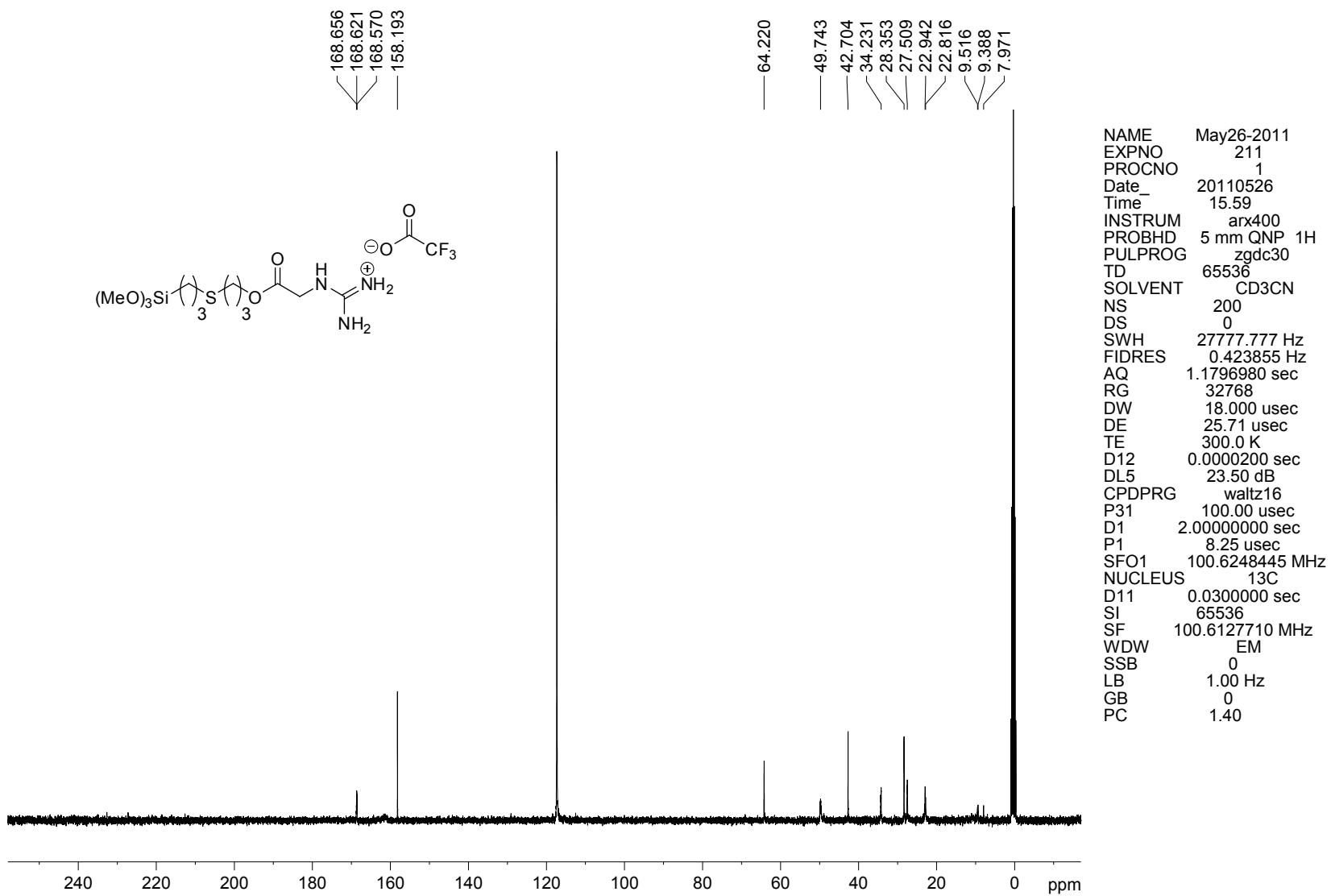
NAME Sept16-2010
EXPNO 260
PROCNO 1
Date 20100916
Time 16.05
INSTRUM arx400
PROBHD 5 mm QNP 1H
PULPROG zg30
TD 65536
SOLVENT CDCl₃
NS 8
DS 0
SWH 8064.516 Hz
FIDRES 0.123055 Hz
AQ 4.0632820 sec
RG 360
DW 62.000 usec
DE 88.57 usec
TE 300.0 K
D1 2.0000000 sec
P1 7.50 usec
SFO1 400.1324008 MHz
NUCLEUS 1H
SI 65536
SF 400.1300173 MHz
WDW EM
SSB 0
LB 0.30 Hz
GB 0
PC 1.00

Undecanoic acid trimethoxysilane (12)

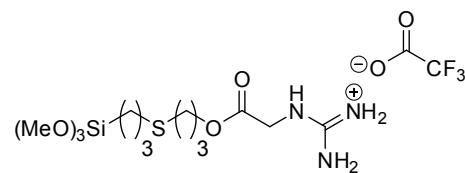




Guanidinium trimethoxysilane (13)



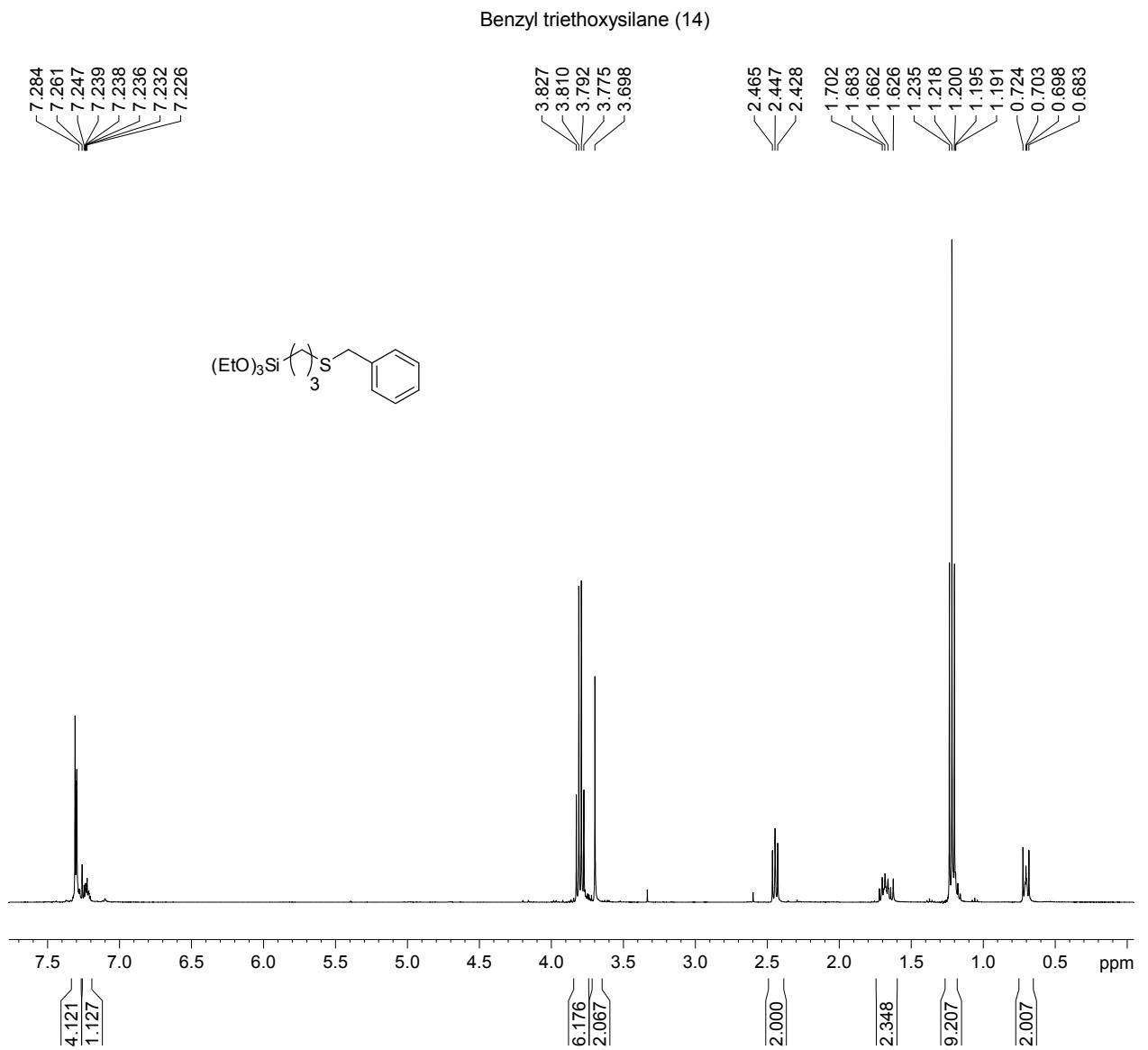
Guanidinium trimethoxysilane (13)



-76.626

NAME May27-2011
 EXPNO 60
 PROCNO 1
 Date_ 20110527
 Time 9.23
 INSTRUM arx400
 PROBHD 5 mm QNP 1H
 PULPROG zgfdc30
 TD 262144
 SOLVENT CD3CN
 NS 8
 DS 0
 SWH 125000.000 Hz
 FIDRES 0.476837 Hz
 AQ 1.0486259 sec
 RG 2860
 DW 4.000 usec
 DE 5.71 usec
 TE 300.0 K
 D1 2.0000000 sec
 D12 0.00002000 sec
 TL0 1.00 dB
 P1 10.90 usec
 D13 0.00000300 sec
 DL5 23.50 dB
 SFO1 376.4985950 MHz
 NUCLEUS 19F
 CPDPRG waltz16
 P31 100.00 usec
 D11 0.03000000 sec
 SI 262144
 SF 376.4985960 MHz
 WDW EM
 SSB 0
 LB 1.00 Hz
 GB 0
 PC 1.00



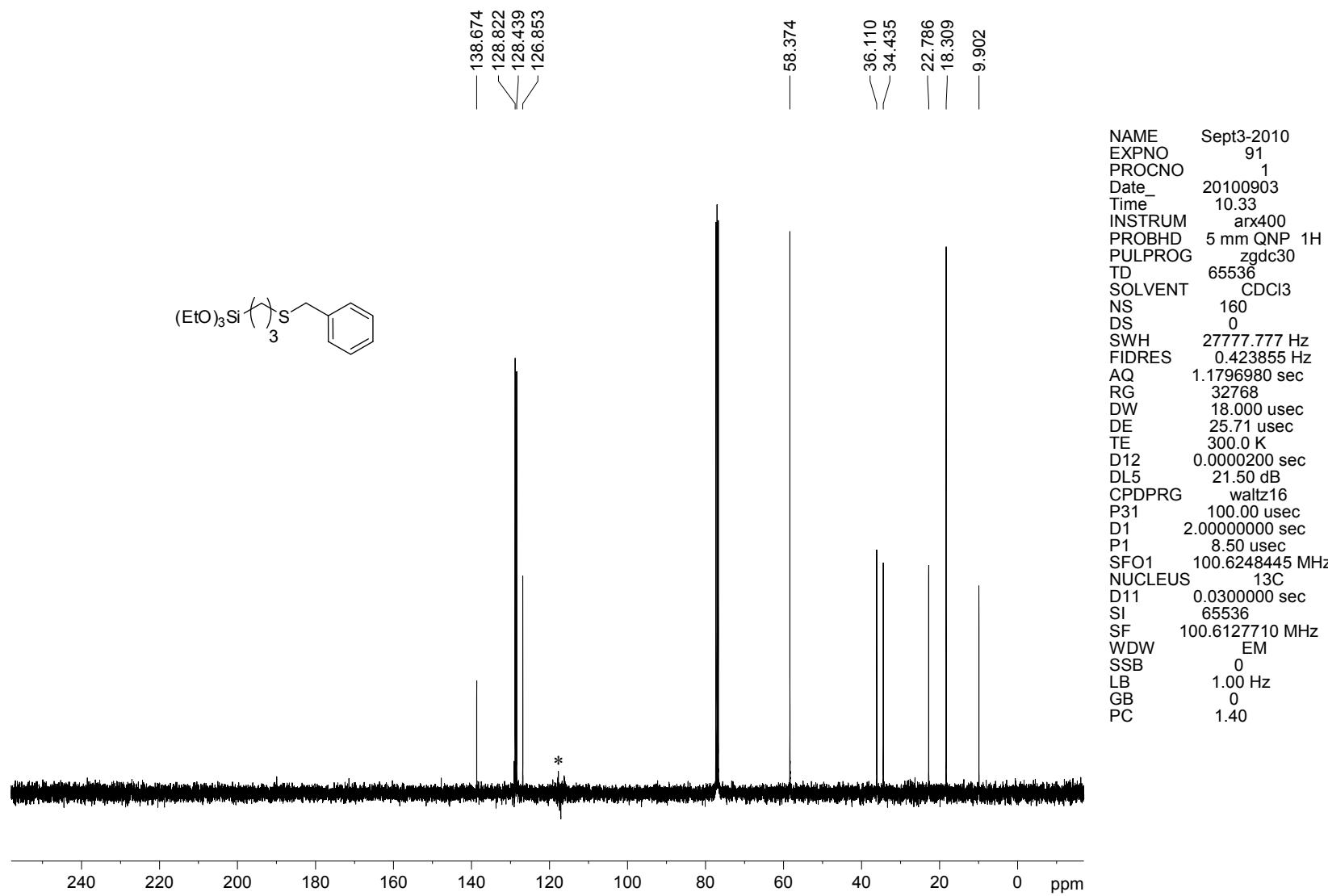


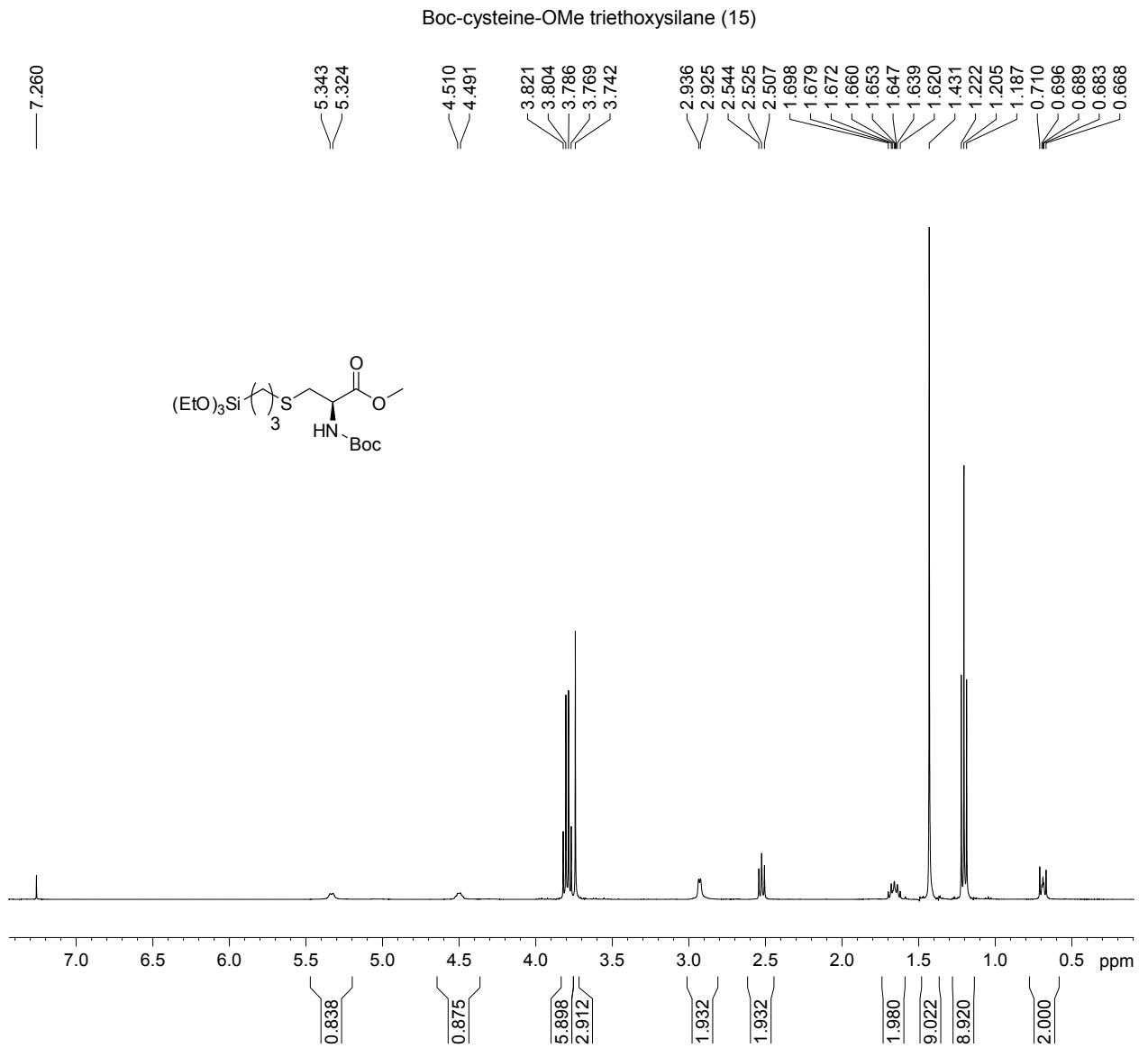
```

NAME      Sept3-2010
EXPNO    90
PROCNO   1
Date_    20100903
Time_    10.24
INSTRUM  arx400
PROBHD  5 mm QNP 1H
PULPROG zg30
TD       65536
SOLVENT  CDCl3
NS        8
DS        0
SWH     8064.516 Hz
FIDRES  0.123055 Hz
AQ      4.0632820 sec
RG       360
DW      62.000 usec
DE      88.57 usec
TE      300.0 K
D1      2.0000000 sec
P1      7.50 usec
SFO1    400.1324008 MHz
NUCLEUS 1H
SI       65536
SF      400.1300173 MHz
WDW      EM
SSB      0
LB      0.30 Hz
GB      0
PC      1.00

```

Benzyl triethoxysilane (14)





```

NAME          Oct28-2010
EXPNO         50
PROCNO        1
Date_        20101028
Time         9.29
INSTRUM      arx400
PROBHD       5 mm QNP 1H
PULPROG      zg30
TD           65536
SOLVENT      CDCl3
NS            8
DS            0
SWH          8064.516 Hz
FIDRES      0.123055 Hz
AQ           4.0632820 sec
RG            256
DW           62.000 usec
DE           88.57 usec
TE           300.0 K
D1           2.0000000 sec
P1           7.50 usec
SFO1        400.1324008 MHz
NUCLEUS      1H
SI            65536
SF          400.1300173 MHz
WDW           EM
SSB            0
LB            0.30 Hz
GB            0
PC           1.00

```

