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Dinuclear Asymmetric Zn Aldol Additions: Formal Asymmetric Synthesis of Fostriecin

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Direct asymmetric aldol reactions constitute a powerful methodology for the synthesis of biologically active compounds. With the use of our dinuclear asymmetric zinc complexes we recently demonstrated the feasibility of using alkynyl methyl ketones as donors. One of the advantages of using terminal benzyldimethylsilyl alkynes as donors in this reaction is the potential for employing the resultant adducts directly in subsequent cross-coupling reactions.² The strategic potential offered by these reactions led us to target fostriecin (CI-920, 1), a cytotoxic phosphate ester isolated from Streptomyces pulveraceus.3 The natural product displays potent anticancer activity against leukemia and many other cell lines.⁴ The cytotoxic properties of 1 are attributed to its selective inhibition of protein phosphatase 2A (PP2A).5 The absolute and relative stereochemistry of 1 was established by Boger,6 who also disclosed the first total synthesis of the natural product⁷ as well as preliminary SAR studies of this molecule.⁸ Several total syntheses have been published, 9 as well as a number of synthetic approaches. 10 Herein we report our efforts in this area, which culminated in a short and efficient synthesis of dephospho-fostriecin 2.

As shown in Scheme 1, we targeted the tetraol 2, a key intermediate in Boger's synthesis. We envisioned that 2 could be disconnected between C13 and C14 to give silane 3 and iodide 4. In the synthetic direction, an alkenylsilane cross-coupling reaction would be employed to forge the labile triene unit. Further disconnection at the C7–C8 bond gives a vinyl metal species 5 and ketone 6. Finally, ketone 6 would be derived from the aldehyde 7 and ynone 8, utilizing the direct Zn-catalyzed asymmetric aldol reaction developed in this group.

Our synthesis commenced with the preparation of the metalation precursor of **5**. Thus, ethyl propiolate was converted to the β -E-iodoacrolein **10** according to literature procedures (Scheme 2). ¹² The labile aldehyde **10** was allylated according to Brown's procedure, ¹³ giving the allylic alcohol in high yield and ee (81% and 95%, respectively). Finally, silylation under standard conditions afforded the target vinyl iodide **11**.

With an efficient route to 11 in hand, attention shifted to the preparation of the C8-C13 portion of fostriecin (1) (Scheme 3).

Here we wished to establish the C9 stereochemistry by enantio-selective direct aldol reaction as reported recently. Thus, the requisite alkynyl ketone **8** was prepared by the addition of ethynyl-magnesium bromide to benzyldimethylchlorosilane (BDMSCI) followed by deprotonation and subsequent acylation. Ketone **12** was treated with aldehyde **7** under our standard Zn-catalyzed direct aldol reaction (3 mol % **13**, 6 mol % Et₂Zn), affording the aldol adduct **14** in an excellent ee (99%). It is worth noting that this reaction could routinely be performed on 50 mmol scale with no deterioration in yield or ee. Use of 5 mol % catalyst bumped the yield to 73%. Next, a diasteroselective reduction was envisaged. Unfortunately, various substrate-controlled methods failed to deliver the desired *anti*-diol in useful yield. Reduction under Noyori's Ru-

Scheme 1. Retrosynthetic Analysis

Scheme 2. Synthesis of Vinyl Iodide 11a

$$EtO_2C \longrightarrow \underbrace{a} EtO_2C \longrightarrow \underbrace{b, c} O \longrightarrow \underbrace{d, e} O \longrightarrow \underbrace{OTES}$$

 a Reagents and conditions: (a) NaI, AcOH, 70 °C, (77%); (b) DIBAL-H, CH₂Cl₂, −78 °C; (c) BF₃·Et₂O, CH₂Cl₂, (95:5 *E:Z*); (d) (+)-Ipc₂BOMe, allylmagnesium chloride, Et₂O, −90 °C, (49% over 3 steps, 95% ee); (e) TESCl, imidazole, DMF, (95%).

Scheme 3. Synthesis of Alkynyl Silane^a

^a Reagents and conditions: (a) ethynylmagnesium bromide, THF, 0 °C (99%); (b) nBuLi, MeCON(Me)OMe, THF, −78 to −15 °C (90%); (c) **13** (3 mol %), Et₂Zn (6 mol %), 4 Å MS, **7**, rt (58−67%, 99% ee); (d) **15** (1 mol %), iPrOH, rt (88%, >10:1 dr); (e) TBSCl, Im, DMF, then CSA, Me₂CO (77% 2 steps); (f) DMBOCH₂Cl, TBAI (14 mol %), DIEA, DMF, 40 °C (97%); (g) MgBr₂, THF, rt, then **11**, iPrMgCl, sBuLi, THF, −78 °C (75%, >20:1 dr); (h) HF, MeOH, MeCN, rt (91%); (i) acryloyl chloride, DIEA, CH₂Cl₂ (99%); (j) TESOTf, DIEA, CH₂Cl₂ (99%); (k) Grubbs 1 (10 mol %), CH₂Cl₂, reflux (93%).

catalyzed transfer hydrogenation conditions¹⁴ (cf. **15**) gave the desired alcohol **16** in high yield and selectivity. Selective silylation and hydrolysis of the ketal furnished ketone **17**, which was protected as the DMBO-acetal¹⁵ **18** in high yield with no detectable epimerization at the α -stereocenter. Thus, the stage was set for the introduction of vinyl metal species **5**. A chelation-controlled addition

^a Reagents and conditions: (a) DDQ, CH₂Cl₂:H₂O (10:1) (94%); (b) **23**, NaHCO₃, MeOH, rt (53%, 72% brsm); (c) **4**, Pd₂(dba)₃·CHCl₃ (5 mol %), TBAF (4 equiv slow addition), THF, 0 °C to room temperature (54%).

of the corresponding magnesiate species¹⁶ of **11** afforded **19** as a single diastereomer in very good yield (75%). The unsaturated lactone moiety was installed by selective removal of the allylic TES-group, followed by acylation with acrolyl chloride giving **20**, which was subsequently exposed to Grubbs first-generation catalyst, yielding key intermediate **21** (83% over three steps).

With an ample supply of 21 in hand, the stage was set for examination of the final steps toward the target, fostriecin precursor, tetraol 2. Thus, the acetal protecting group was removed by the addition of DDQ, providing alkyne 22 in 94% yield (Scheme 4).

A *cis* reduction was required to install the vinyl silane. Attempts to effect this transformation were hampered by either poor yield and/or over-reduction. Eventually, using diimide precursor 23¹⁷ under mildly basic conditions afforded the *cis*-vinyl silane 24. Finally the assembly of the triene unit called upon a Pd-catalyzed vinyl silane cross-coupling reaction. The susceptibility of the unsaturated lactone toward base complicated this cross-coupling. After much experimentation, it was found that the sensitive triene functionality could be assembled by adding TBAF slowly to a combination of vinyl silane 24, vinyl iodide 4,¹⁸ and catalytic amounts of Pd₂(dba)₃·CHCl₃ in THF. As expected, the cross-coupling also led to concomitant deprotection of the silicon protecting groups, giving the dephosphofostriecin 2 in good yield. The conversion of 2 to fostriecin (1) has been demonstrated by Boger.⁷

In conclusion, we have completed the synthesis of dephosphofostriecin **2**, and thereby a formal synthesis of fostriecin **1**, in 14 steps for the longest linear sequence and 8.5% overall yield. This work illustrates for the first time the use of the direct asymmetric Zn-catalyzed aldol reaction, and the utility of the corresponding aldol adducts as building blocks for complex molecule synthesis. It also exemplifies the extraordinary utility of the Pd-catalyzed vinyl silane cross-coupling as an alternative to more mainstream Pd-catalyzed cross-coupling reactions, in the synthesis of complex molecules.

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Supporting Information Available: Experimental details and characterization for all new compounds (PDF). This material is available free of charge via the Internet at http://pubs.acs.org

References

- (1) Trost, B. M.; Fettes, A.; Shireman, B. T. *J. Am. Chem. Soc.* **2004**, *126*, 2660 and references therein to other direct asymmetric aldol reactions.
- (2) Trost, B. M.; Machacek, M. R.; Ball, Z. T. Org. Lett. 2003, 5, 1895.
- (3) (a) Tunac, J. B.; Graham, B. D.; Dobson, W. E. J. Antibiot. 1983, 36, 1595. (b) Stampwala, S. S.; Bunge, R. H.; Hurley, T. R.; Willmer, N. E.; Brankiewicz, A. J.; Steinman, C. E.; Smitka, T. A.; French, J. C. J. Antibiot. 1983, 36, 1601. (c) Hokanson, G. C.; French, J. C. J. Org. Chem. 1985, 50, 462.
- (4) Jackson, R. C.; Fry, D. W.; Boritzki, T. J.; Roberts, B. J.; Hook, K. E.; Leopold, W. R. *Adv. Enzyme Reg.* **1985**, *23*, 193.
- (5) (a) Lewy, D. S.; Gauss, C. M.; Soenen, D. R.; Boger, D. L. Curr. Med. Chem. 2002, 9, 2005. (b) de Jong, R. S.; Mulder, N. H.; Uges, D. R. A.; Sleijfer, D. T.; Hoppener, F. J. P.; Groen, H. J. M.; Willemse, P. H. B.; van der Graaf, W. T. A.; de Vries, E. G. E. Br. J. Cancer 1999, 79, 882.
- (6) Boger, D. L.; Hikota, M.; Lewis, B. M. J. Org. Chem. 1997, 62, 1748.
- (7) Boger, D. L.; Ichikawa, S.; Zhong, W. J. Am. Chem. Soc. 2001, 123, 4161.
- (8) Buck, S. B.; Hardouin, C.; Ichikawa, S.; Soenen, D. R.; Gauss, C. M.; Hwang, I.; Swingle, M. R.; Bonness, K. M.; Honkanen, R. E.; Boger, D. L. J. Am. Chem. Soc. 2003, 125, 15694.
- (9) (a) Chavez, D. E.; Jacobsen, E. N. Angew. Chem., Int. Ed. 2001, 40, 3667.
 (b) Reddy, Y. K.; Falck, J. R. Org. Lett. 2002, 4, 969. (c) Miyashita, K.; Ikejiri, M.; Kawasaki, H.; Maemura, S.; Imanishi, T. Chem. Commun. 2002, 742. (d) Miyashita, K.; Ikejiri, M.; Kawasaki, H.; Maemura, S.; Imanishi, T. J. Am. Chem. Soc. 2003, 125, 8238. (e) Wang, Y. G.; Kobayashi, Y. Org. Lett. 2002, 4, 4615. (f) Esumi, T.; Okamoto, N.; Hatakeyama, S. Chem. Commun. 2002, 3042. (g) Fujii, K.; Maki, K.; Kanai, M.; Shibasaki, M. Org. Lett. 2003, 5, 733.
 (10) (a) Just G.; Oconnor B. Tetrahedron Lett. 1988, 29, 753. (b) Liu, S. V.
- (10) (a) Just, G.; Oconnor, B. Tetrahedron Lett. 1988, 29, 753. (b) Liu, S. Y.; Huang, D. F.; Huang, H. H.; Huang, L. Chin. Chem. Lett. 2000, 11, 957.
 (c) Cossy, J.; Pradaux, F.; BouzBouz, S. Org. Lett. 2001, 3, 2233. (d) Kiyotsuka, Y.; Igarashi, J.; Kobayashi, Y. Tetrahedron Lett. 2002, 43, 2725. (e) Marshall, J. A.; Bourbeau, M. P. Org. Lett. 2003, 5, 3197. (f) Ramachandran, P. V.; Liu, H. P.; Reddy, M. V. R.; Brown, H. C. Org. Lett. 2003, 5, 3755.
- (11) For a review, see: Denmark, S. E.; Sweis, R. F. Acc. Chem. Res. 2002, 35, 835.
- (12) Marek, I.; Meyer, C.; Normant, J. F. Org. Synth. 1997, 74, 194.
- (13) Brown, H. C.; Jadhav, P. K. J. Am. Chem. Soc. 1983, 105, 2092.
- (14) (a) Matsumura, K.; Hashiguchi, S.; Ikariya, T.; Noyori, R. J. Am. Chem. Soc. 1997, 119, 8738. (b) Haack, K.-J.; Hashiguchi, S.; Fujii, A.; Ikariya, T.; Noyori, R. Angew. Chem., Int. Ed. Engl. 1997, 36, 285.
- (15) Kigoshi, H.; Suenaga, K.; Mutou, T.; Ishigaki, T.; Atsumi, T.; Ishiwata, H.; Sakakura, A.; Ogawa, T.; Ojika, M.; Yamada, K. J. Org. Chem. 1996, 61, 5326.
- (16) (a) Inoue, A.; Kitagawa, K.; Shinokubo, H.; Oshima, K. J. Org. Chem. 2001, 66, 4333. (b) Kitagawa, K.; Inoue, A.; Shinokubo, H.; Oshima, K. Angew. Chem., Int. Ed. 2000, 39, 2481. (c) Knochel, P.; Dohle, W.; Gommermann, N.; Kneisel, F. F.; Kopp, F.; Korn, T.; Sapountzis, I.; Vu, V. A. Angew. Chem., Int. Ed. 2003, 42, 4302.
- (17) Cusack, N. J.; Reese, C. B.; Risius, A. C.; Roozpeikar, B. Tetrahedron 1976, 32, 2157.
- (18) Vinyl iodide 4 was prepared from ester 9, via DIBAL-H reduction, Horner-Wadsworth-Emmons reaction and subsequent DIBAL-H reduction, which afforded 4 in 33% overall yield, see Supporting Information for more details.

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