

Tetrahedron: Asymmetry 9 (1998) 4113–4115

An efficient synthesis of (R)-3-hydroxytetradecanoic acid

Guangfei Huang and Rawle I. Hollingsworth*

Department of Chemistry, Michigan State University, East Lansing, MI 48824-1322, USA

Received 19 October 1998; accepted 20 October 1998

Abstract

A short, efficient synthesis of (*R*)-3-hydroxytetradecanoic acid, a key component of bacterial endotoxins, using (*R*)-oxirane acetic acid ethyl ester as the source of chirality is described. The method is general and can be used in the preparation of other chiral 3-hydroxy acids. © 1998 Elsevier Science Ltd. All rights reserved.

(*R*)-3-Hydroxytetradecanoic acid is the most common fatty acid constituent of the lipid A component of bacterial lipopolysaccharides (LPS) (Fig. 1). The lipid A is responsible for most of the endotoxic activities of LPS.^{1,2}

Although several methods have been reported for the formation of 3-hydroxy acids,³ current syntheses for (R)-3-hydroxytetradecanoic acid either give products in low optical purity,⁴ in low yield⁵ or are limited to small scale preparations.⁶ 3-Hydroxytetradecanoic acid of very high optical purity is essential for

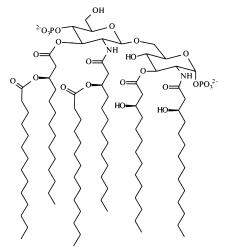
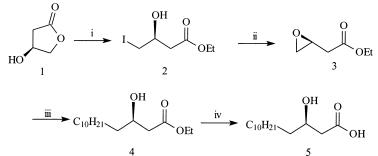


Figure 1. The structure of a typical bacterial lipid A

^{*} Corresponding author. E-mail: rih@argus.cem.msu.edu

^{0957-4166/98/\$ -} see front matter @ 1998 Elsevier Science Ltd. All rights reserved. PII: S0957-4166(98)00441-8

the synthesis of lipid A because at least four such fatty acid residues are present. A synthesis employing a fatty acid preparation that was 93% optically pure would yield a lipid A product that was only 75% pure containing 16 isomers. Here we report an efficient synthesis for (*R*)-3-hydroxytetradecanoic acid in high enantiomeric excess from (*S*)-3-hydroxy- γ -butyrolactone (1).^{7,8} It should be useful in the preparation of other chiral 3-hydroxy acids. As shown in the scheme, (*S*)-3-hydroxy- γ -butyrolactone was treated with NaI–TMSCI–CH₃CN⁹ at room temperature to give the iodohydrin **2**. This was converted to the oxiraneacetic acid ester **3** by treatment with Ag₂O in CH₃CN at room temperature.¹⁰ Compound **3** was selectively ring-opened with decyl magnesium bromide and CuI in anhydrous THF at -30° C to give (*R*)-3-hydroxytetradecanoic acid ethyl ester **4** in 97% yield¹¹ [99.3% ee; NMR spectrum of (*S*)-(–)- α -methoxy- α -(trifluoromethyl)phenylacetyl ester¹²]. Finally, the 3-hydroxy ester was saponified to give (*R*)-3-hydroxytetradecanoic acid **5**.



Reaction Conditions, Reagents and Yields: (i) NaI, TMSCl, CH₃CN, then CH₃CH₂OH, 63%; (ii) Ag₂O, CH₃CN, 94%; (iii) CuI, n-C₁₀H₂₁MgBr, THF, -30°C, 97%; (iv) KOH in 90% ethanol, 76%.

Acknowledgements

We are grateful for the financial support of Synthon Corporation and the Michigan State University Research Excellence Fund.

References

- (a) Galanos, C.; Luderitz, O.; Rietschel, E. T.; Westphal, O. In *International Review of Biochemistry, Biochemistry of Lipids II*; Goodwin, T. W., Ed. University Park Press: Baltimore, 1977; Vol. 14, p. 239. (b) Luderitz, O.; Galanos, C.; Lehmann, V.; Myer, H.; Rietschel, E. T.; Wechesser, J. *Naturwissenschaften* 1978, 65, 578.
- Raetz, C. H. R. In *Bacterial Outer Membranes As Model Systems*; Inouye, D. M., Ed. John Wiley: New York, 1987; pp. 229–245.
- (a) Curran, D. P.; Scanga, S. A.; Fenk, C. J. J. Org. Chem. 1984, 49, 3474. (b) Deng, M.-Z.; Lu, D.-A.; Xu, W.-H. J. Chem. Soc., Chem. Commun. 1985, 21, 1478. (c) Vining, L. C.; Taber, W. A. Can. J. Chem. 1962, 40, 1579.
- 4. (a) Basavaiah, D.; Bharathi, T. K. Synth. Commun. 1989, 19, 2035. (b) Tai, A.; Nakahata, M.; Harada, T.; Izumi, Y.; Kusumoto, S.; Inage, M.; Shiba, T. Chem. Lett. 1980, 1125–1126.
- 5. Utaka, M.; Watabu, H.; Higashi, H.; Sakai, T.; Tsuboi, S.; Torii, I. J. Org. Chem. 1990, 55, 3917.
- 6. Sugai, T.; Ritzen, H.; Wong, C.-H. Tetrahedron: Asymmetry 1993, 4, 1051.
- 7. Hollingsworth, R. I. United States Patent 5,292,939, 1994.
- 8. Huang, G.; Hollingsworth, R. I. Tetrahedron 1998, 54, 1355.
- 9. Olah, G. A.; Narane, S. C.; Gupta, B. G. B.; Malhora, R. J. Org. Chem. 1979, 44, 1247.
- 10. Larcheveque, M.; Henrot, S. Tetrahedron 1990, 46, 4277.
- 11. To a suspension of CuI (3.81 g, 20.0 mmol) in 50 mL anhydrous THF (dry nitrogen) was added dropwise decylmagnesium bromide (40 mL, 1.0 M in ether) at -30°C with stirring. After 30 min, the epoxide 3 (2.60 g, 20.0 mmol) in 10 mL anhydrous THF was added dropwise and the reaction mixture was stirred for 1 h at -30°C. It was then allowed to reach

room temperature and was stirred overnight. The reaction was quenched with saturated NH₄Cl in water and the phases separated. The aqueous layer was extracted three times with ether. The combined ether extracts were washed with saturated NaCl solution, dried (Na₂SO₄), concentrated and separated on a silica column. The yield was 97%. IR (CDCl₃): 3027 (s), 2928 (s), 2857 (s), 1213 (s), 1208 (s) cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 4.15 (2H, q, *J*=7.2 Hz), 3.97 (1H, m), 2.48 (1H, dd, *J*=16.5, 3.0 Hz), 2.37 (1H, dd, *J*=16.5, 9.0 Hz), 1.40 (2H, m), 1.23 (21H, br, s), 0.86 (3H, t, *J*=6.6 Hz); ¹³C NMR (75 MHz, CDCl₃): 173.15, 68.01, 60.65, 41.24, 36.47, 31.88, 29.60, 29.55, 29.50, 29.32, 25.45, 22.66, 14.14, 14.09; HRMS Exact mass: calcd for C₁₆H₃₃O₃, [M+H]⁺, 273.2430. Found 273.2438.

12. Dale, J. A.; Dull, D. L.; Mosher, H. S. J. Org. Chem. 1969, 34, 2543.