

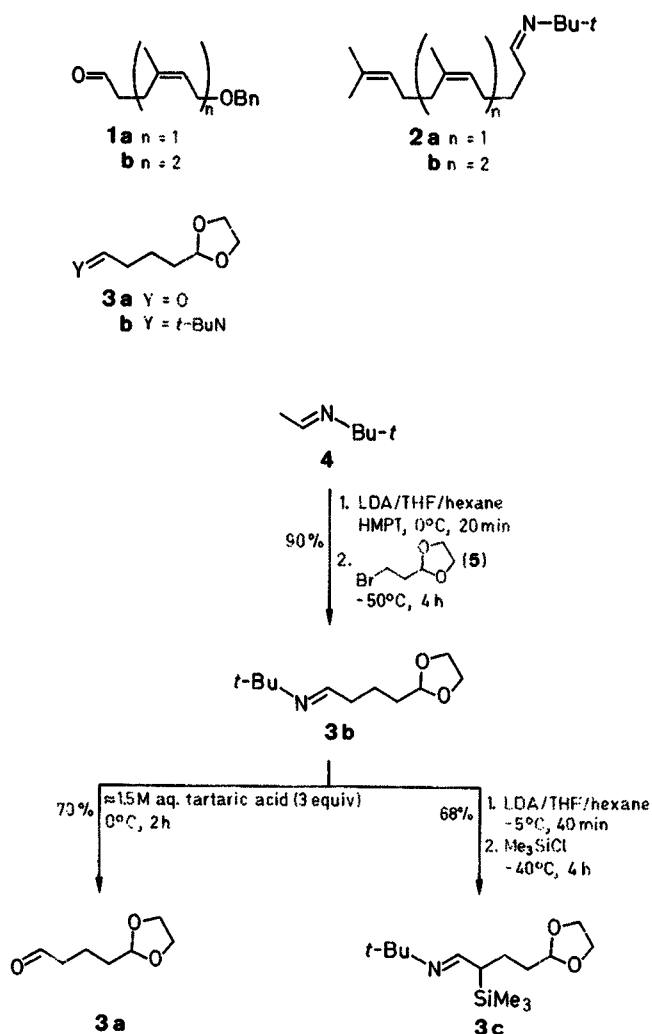
# Glutaraldehyde Derivatives as Building Blocks for Stereoselective (Z)-C<sub>5</sub> Elongation of a Regular Isoprenoid Chain

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5,5-Ethylenedioxy-pentanal and its *tert*-butylimine may serve as versatile non-isoprenoid synthons for the construction of a regular *cisoid* terpenoid chain using the controlled crossed aldol condensation methodology.

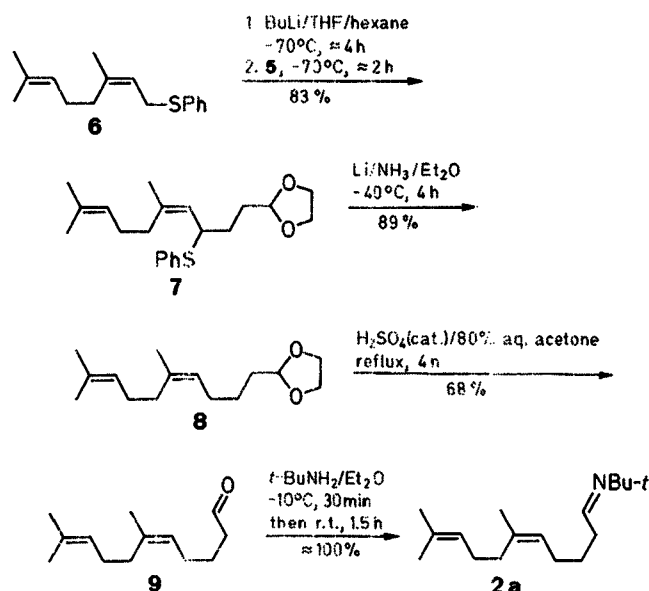
One of effective approaches to the construction of linear isoprenoid molecules containing inner (Z)-C<sub>5</sub> units consists of the controlled crossed aldol condensation of trisnorprenal derivatives of type 1 with trishomoprenal imines such as 2.<sup>1-5</sup> Here we report the use of the same methodology for highly stereoselective *cis*-C<sub>5</sub> homologation of a terpenoid chain using readily available glutaraldehyde derivatives 3. The versatility of such building blocks for the elongation of both the "tail" and the "head" moieties of terpenoid molecules is illustrated below the conversion of aldehyde 1a and imine 2a into respective pre-nologues 1b and 2b (Bn = CH<sub>2</sub>Ph).



Scheme A

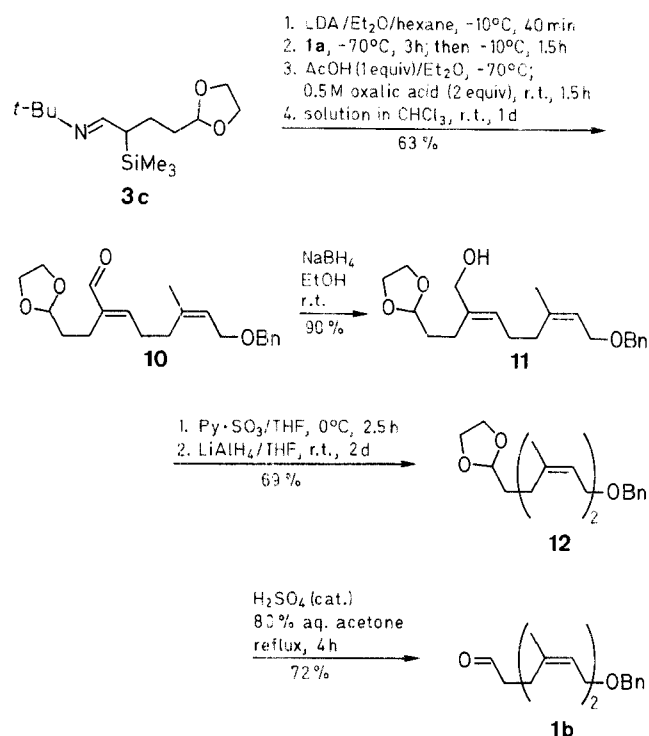
Both of these two transformations were previously unknown. The protected glutaraldehyde derivatives 3a-c were prepared by deprotonated acetaldehyde *N*-*tert*-butylimine<sup>6</sup> (4) with lithium diisopropylamide (LDA) and hexamethylphosphoric triamide (HMPT) followed by alkylation with 3-bromopropanal ethylene acetal<sup>7</sup> (5) to give imine 3b which was selectively hydrolyzed to aldehyde 3a. Further, imine 3b was readily transformed into its unstable trimethylsilyl derivative 3c according to a previously reported procedure<sup>8</sup> (Scheme A).

The previously unknown aldimine 2a used in the second transformation (2a → 2b) was obtained by a four-step sequence (Scheme B) recently developed by us for the synthesis of all-(*E*)-farnesyl and all-(*E*)-geranylgeranyl analogs of 2:<sup>2,3</sup> alkylation of neryl phenyl sulfide<sup>9</sup> (6) with 5 followed by reductive desulfurization of the intermediate 7 to the acetal 8 smoothly gave aldehyde 9, which was converted into imine 2a (≈ 50% overall yield from 6). The structure of 2a was confirmed by spectral (IR, <sup>1</sup>H-NMR) data.



Scheme B

The Peterson-type reaction of the silylated aldimine 3c with the known nerol derivative 1a<sup>10</sup> under the action of LDA proceeds smoothly to give, after mild acidic work-up and chromatography of the reaction mixture on silica gel, the (4*E*)-4-formyldecadienalmonoacetal 10 in 63% yield and in a stereochemical purity of ≈ 85% (Scheme C).



Scheme C

The purity of **10** was estimated by comparison in its <sup>1</sup>H-NMR spectrum of the integral intensities corresponding to aldehyde proton signals of the main (4*E*)-**10** and minor (4*Z*)-**10** isomers at  $\delta$  = 9.3 and 10.1, respectively (cf. Ref. 11). It was found that the portion of the former isomer is increased up to 95–97% on storage of a freshly prepared sample of **10** as its ca. 10% (v/v) chloroform solution under argon at room temperature for 1 day.

The transformation of formyldecadienal monoacetal **10** into the methyl derivative **12** was performed according to Corey<sup>12</sup> in 62% overall yield by the treatment of **10** with sodium borohydride in aqueous ethanol followed by lithium aluminum hydride reduction of the *O*-sulfate derivative of allylic alcohol without its isolation. Hydrolytic cleavage of the acetal **12** completes the preparation of **1b**, the known oxidative product of (*Z,Z*)-farnesyl benzyl ether.<sup>2</sup>

Starting with the aldimine **2a** and glutaraldehyde monoacetal **3a** (*Z,Z*)-trishomofarnesal derivative **2b** was then obtained (Scheme D). Thus, silylation of **2a** and Peterson-type condensation of the unstable intermediate **2c**, without its isolation, with **3a** furnished, after mild acidic treatment of the reaction mixture and equilibration of the crude product in chloroform solution, the (5*E*)-6-formylpentadecatrienal monoacetal **13** in 45% isolated yield with a stereochemical purity of > 95%, according to the above <sup>1</sup>H-NMR criteria. Successive hydride reduction of **13** to the hydroxymethyl derivative **14**, its further conversion into the triene acetal **15**, and hydrolytic cleavage of the latter gave (*Z,Z*)-trishomofarnesal (**16**) was followed by smooth transformation to the aldimine **2b** under the action of excess *tert*-butylamine.

The assigned structures of all new compounds prepared were consistent with their microanalytical and spectral data.

In summary, glutaraldehyde derivatives **3** can be regarded as versatile building blocks for the highly stereoselective *cis*-C<sub>5</sub> homologation of linear terpenoid aldehydes by means of controlled crossed aldol condensations.

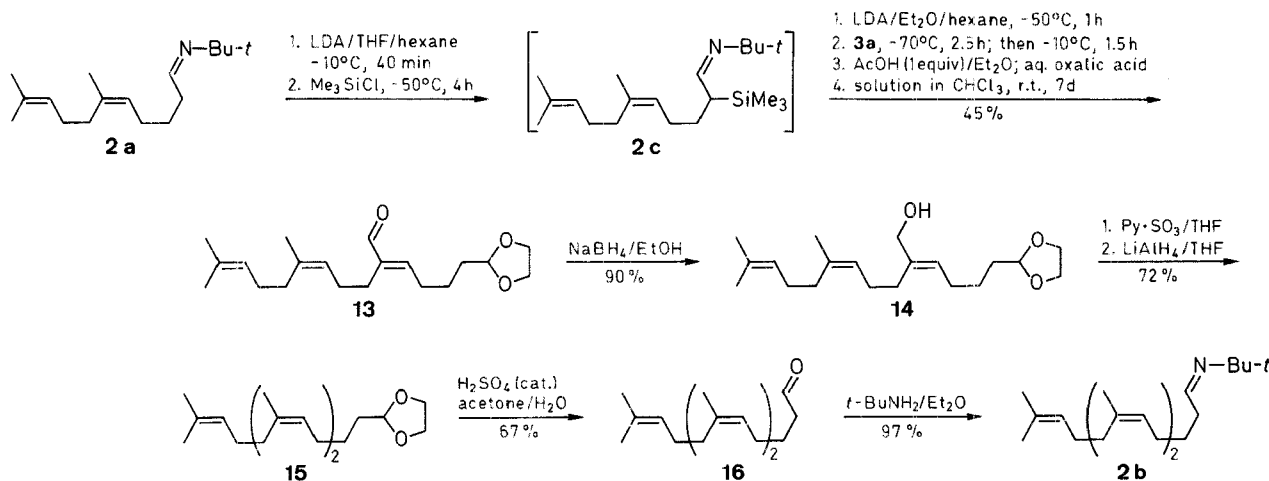
IR spectra were recorded on a Perkin-Elmer 577 spectrophotometer in Nujol (**11** and **14** were measured in CCl<sub>4</sub> solution), and UV spectra on a Perkin Elmer UV-VIS 552 spectrometer in EtOH. <sup>1</sup>H-NMR spectra were measured with a Bruker WM 250 spectrometer at 250 MHz.

All experiments were carried out under argon atmosphere and with freshly distilled and dried solvents.

#### 5,5-Ethylenedioxyptental (3a):

**5,5-Ethylenedioxyptental *tert*-Butylimine (3b):** A solution of acetaldehyde *tert*-butylimine (**4**;<sup>6</sup> 9.9 g, 100 mmol) in THF (20 mL) is added dropwise (40 min) to a stirred solution of LDA (110 mmol) in hexane (70 mL), THF (30 mL), and HMPT (17.6 mL, 100 mmol) at 0°C. The mixture is stirred for 20 min and cooled at -70°C. A solution of 3-bromopropanal ethylene acetal<sup>7</sup> (**5**; 16.3 g, 90 mmol) in THF (20 mL) is then added during 20 min, and after 10 min the temperature is raised to -50°C and stirring is continued for 4 h. The reaction mixture is then treated at -5°C with ice-cold H<sub>2</sub>O (50 mL). The aqueous phase is separated and extracted with Et<sub>2</sub>O (3 × 70 mL). The combined organic layer is washed with sat. aq. NaCl (2 × 40 mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). The solvent is removed *in vacuo*, and the residue is kept at 25°C and 1 Torr for 6 h to give the imine **3b** as a colorless oil (Table); yield: 18.43 g (90%).

**5,5-Ethylenedioxyptental (3a):** The above reaction mixture is treated at 5°C, instead of ice-cold H<sub>2</sub>O, with a solution of tartaric acid (29.25 g, 0.21 mol) in H<sub>2</sub>O (0.2 L) and is vigorously stirred at 0°C for 2 h. The aqueous phase is then separated and extracted with Et<sub>2</sub>O (5 × 100 mL). The combined organic layer is washed with sat. aq. NaCl (3 × 70 mL).



Scheme D

Table. Physical Data of New Compounds Prepared

Prod- uct	Yield <sup>a</sup> (%)	bp (°C) Torr	Molecular Formula <sup>b</sup>	IR $\nu$ (cm <sup>-1</sup> )	<sup>1</sup> H-NMR (CDCl <sub>3</sub> /TMS) $\delta$ , $J$ (Hz)
<b>2a</b>	~100		C <sub>17</sub> H <sub>31</sub> N (249.4)	2970, 2930, 2860, 2725, 1730, 1670, 1450, 1380, 1360, 1215, 1140, 1110, 1070, 980, 835, 740, 690	1.15 (s, 9H, <i>t</i> -Bu); 1.50 (m, 2H, CH <sub>2</sub> ); 1.57 (s, 3H, <i>cis</i> -CH <sub>3</sub> ); 1.67 (s, 6H, <i>trans</i> -CH <sub>3</sub> ); 2.00 (m, 6H, CH <sub>2</sub> C=C); 2.22 (m, 2H, CH <sub>2</sub> C=N); 5.10 (m, 2H, HC=C); 7.59 (t, 1H, $J = 5.0$ , CH=N)
<b>2b</b>	97		C <sub>22</sub> H <sub>39</sub> N (317.5)	2960, 2920, 2860, 2720, 1730, 1665, 1560, 1450, 1380, 1230, 1215, 1110, 1070, 985, 835, 740	1.16 (s, 9H, <i>t</i> -Bu); 1.53 (m, 2H, CH <sub>2</sub> ); 1.60 (s, 3H, <i>cis</i> -CH <sub>3</sub> ); 1.68 (s, 9H, <i>trans</i> -CH <sub>3</sub> ); 2.02 (m, 10H, CH <sub>2</sub> C=C); 2.22 (m, 2H, CH <sub>2</sub> C=N); 5.11 (m, 3H, HC=C); 7.58 (t, 1H, $J = 5.0$ , CH=N)
<b>3a</b>	70	63/1	C <sub>7</sub> H <sub>12</sub> O <sub>3</sub> (144.2)	3020, 2960, 2890, 2725, 2670, 2620, 1720, 1480, 1440, 1410, 1360, 1340, 1310, 1270, 1240, 1140, 1070, 1040, 1020, 970, 950, 890, 790, 700	1.70 (m, 4H, CH <sub>2</sub> ); 2.45 (dt, 2H, $J_d = 1.8$ , $J_t = 6.6$ , CH <sub>2</sub> C=O); 3.85 (m, 4H, CH <sub>2</sub> O); 4.82 (t, 1H, $J = 4.8$ , OCHO); 9.72 (t, 1H, $J = 1.8$ , CHO)
<b>3b</b>	90	78/1	C <sub>11</sub> H <sub>21</sub> NO <sub>2</sub> (199.3)	2960, 2930, 2870, 2760, 1730, 1670, 1480, 1460, 1410, 1360, 1220, 1140, 1040, 945, 880, 735, 640	1.10 (s, 9H, <i>t</i> -Bu); 1.61 (m, 4H, CH <sub>2</sub> ); 2.20 (dt, 2H, $J_d = 5.0$ , $J_t = 5.3$ , CH <sub>2</sub> C=N); 3.85 (m, 4H, CH <sub>2</sub> O); 4.82 (t, 1H, $J = 5.0$ , OCHO); 7.55 (t, 1H, $J = 5.0$ , HC=N)
<b>3c</b>	68	100.5/1	C <sub>14</sub> H <sub>29</sub> NO <sub>2</sub> Si (271.5)	3380, 2960, 2930, 2870, 2750, 1650, 1620, 1470, 1455, 1405, 1360, 1250, 1225, 1135, 1035, 940, 885, 840, 750, 690, 615	-0.02 (s, 9H, SiMe <sub>3</sub> ); 1.12 (s, 9H, <i>t</i> -Bu); 1.63 (m, 4H, CH <sub>2</sub> ); 1.84 (m, 1H, CHSi); 3.85 (m, 4H, CH <sub>2</sub> O); 4.80 (t, 1H, $J = 4.5$ , OCHO); 7.40 (d, 1H, $J = 6.2$ , HC=N)
<b>7</b>	83	142-143/ 0.03 <sup>c</sup>	C <sub>21</sub> H <sub>30</sub> O <sub>2</sub> S (346.5)	3030, 2960, 2930, 2880, 1660, 1585, 1480, 1450, 1440, 1410, 1380, 1300, 1260, 1230, 1215, 1140, 1090, 1025, 940, 885, 750, 690	1.58 (s, 3H, <i>cis</i> -CH <sub>3</sub> ); 1.61 (m, 2H, CH <sub>2</sub> ); 1.69 (s, 6H, <i>trans</i> -CH <sub>3</sub> ); 1.8 (m, 6H, CH <sub>2</sub> ); 3.9 (m, 5H, CH <sub>2</sub> O, CHS); 4.83 (t, 1H, $J = 4.3$ , OCHO); 5.02 (m, 2H, HC=C); 7.35 (m, 5H <sub>arom</sub> )
<b>8</b>	89	110.5/2	C <sub>15</sub> H <sub>26</sub> O <sub>2</sub> (238.4)	2965, 2935, 2890, 2870, 1680, 1460, 1410, 1380, 1140, 1050, 950, 830, 790	1.46 (m, 2H, CH <sub>2</sub> ); 1.60 (s, 3H, <i>cis</i> -CH <sub>3</sub> ); 1.64 (m, 2H, CH <sub>2</sub> ); 1.68 (s, 6H, <i>trans</i> -CH <sub>3</sub> ); 2.03 (m, 6H, CH <sub>2</sub> C=C); 3.9 (m, 4H, CH <sub>2</sub> O); 4.83 (t, 1H, $J = 4.7$ , OCHO); 5.10 (t, 2H, $J = 6.5$ , HC=C)
<b>9</b>	68	91-92/1.5	C <sub>13</sub> H <sub>22</sub> O (194.3)	2970, 2940, 2860, 2720, 1730, 1455, 1415, 1380, 1240, 1140, 1115, 1070, 985, 835, 670	1.52 (m, 2H, CH <sub>2</sub> ); 1.60 (s, 3H, <i>cis</i> -CH <sub>3</sub> ); 1.68 (s, 6H, <i>trans</i> -CH <sub>3</sub> ); 2.0 (m, 6H, CH <sub>2</sub> C=C); 2.4 (dt, 2H, $J_d = 2.0$ , $J_t = 7.5$ , CH <sub>2</sub> C=O); 5.07 (m, 2H, HC=C); 9.70 (t, 1H, $J = 2.0$ , CHO)
<b>10<sup>d</sup></b>	63	178-180/ 0.018 <sup>c</sup>	C <sub>21</sub> H <sub>28</sub> O <sub>4</sub> (344.4)	3090, 3070, 3040, 2980, 2940, 2890, 2860, 2720, 1690, 1640, 1500, 1460, 1405, 1380, 1360, 1310, 1260, 1210, 1140, 1070, 1030, 940, 905, 880, 810, 735, 700	1.68 (s, 3H, CH <sub>3</sub> ); 1.7 (m, 2H, CH <sub>2</sub> ); 2.35 (m, 6H, CH <sub>2</sub> C=C); 3.9 (m, 4H, CH <sub>2</sub> O); 4.01 (d, 2H, $J = 6.7$ , CH <sub>2</sub> OBn); 4.50 (s, 2H, CH <sub>2</sub> Ph); 4.81 (t, 1H, $J = 4.5$ , OCHO); 5.50 (t, 1H, $J = 6.7$ , H-7); 6.42 (t, 1H, $J = 6.7$ , H-3); 7.3 (m, 5H <sub>arom</sub> ); 9.30 (s, 1H, CHO)
<b>11</b>	90	165-170/ 0.018 <sup>c</sup>	C <sub>21</sub> H <sub>30</sub> O <sub>4</sub> (346.5)	3650, 3050, 2970, 2940, 2870, 1460, 1410, 1360, 1210, 1140, 1070, 1030, 950, 900, 740, 700	1.75 (m, 2H, CH <sub>2</sub> ); 1.76 (s, 3H, CH <sub>3</sub> ); 2.15 (m, 6H, CH <sub>2</sub> C=C); 3.9 (m, 4H, CH <sub>2</sub> O); 3.98 (d, 2H, $J = 6.0$ , CH <sub>2</sub> OBn); 4.00 (s, 2H, CH <sub>2</sub> O); 4.50 (s, 2H, CH <sub>2</sub> Ph); 4.83 (t, 1H, $J = 5.0$ , OCHO); 5.37 (t, 1H, $J = 6.0$ , H-9); 5.43 (t, 1H, $J = 6.7$ , H-5); 7.3 (m, 5H <sub>arom</sub> )
<b>12</b>	69	148-150/ 0.018 <sup>c</sup>	C <sub>21</sub> H <sub>30</sub> O <sub>3</sub> (330.5)	3040, 2970, 2950, 2870, 1670, 1460, 1410, 1380, 1360, 1140, 1100, 1070, 1030, 950, 900, 740, 700	1.70 (s, 3H, CH <sub>3</sub> ); 1.75 (m, 2H, CH <sub>2</sub> ); 1.77 (s, 3H, CH <sub>3</sub> ); 2.10 (m, 6H, CH <sub>2</sub> ); 3.9 (m, 4H, CH <sub>2</sub> O); 4.05 (d, 2H, $J = 7.0$ , CH <sub>2</sub> OBn); 4.50 (s, 2H, CH <sub>2</sub> Ph); 4.83 (t, 1H, $J = 4.8$ , OCHO); 5.13 (t, 1H, $J = 7.0$ , H-5); 5.43 (t, 1H, $J = 7.0$ , H-9); 7.3 (m, 5H <sub>arom</sub> )
<b>13<sup>e</sup></b>	45	153-154/ 0.02 <sup>c</sup>	C <sub>20</sub> H <sub>32</sub> O <sub>3</sub> (320.4)	2960, 2920, 2870, 2710, 1690, 1640, 1450, 1410, 1380, 1135, 1045, 1030, 940, 830, 760, 710	1.59 (s, 3H, <i>cis</i> -CH <sub>3</sub> ); 1.63 (m, 4H, CH <sub>2</sub> ); 1.68 (s, 6H, <i>trans</i> -CH <sub>3</sub> ); 2.0 (m, 6H, CH <sub>2</sub> C=C); 2.25 (t, 2H, $J = 7.5$ , H-3); 2.40 (dt, 2H, $J_d = J_t = 7.5$ , H-2); 3.9 (m, 4H, CH <sub>2</sub> O); 4.87 (t, 1H, $J = 4.8$ , OCHO); 5.08 (t, 2H, $J = 7.5$ , HC=C); 6.43 (t, 1H, $J = 7.5$ , H-1'); 9.34 (s, 1H, CHO)
<b>14</b>	90	165-170/ 0.02 <sup>c</sup>	C <sub>20</sub> H <sub>34</sub> O <sub>3</sub> (322.5)	3620, 2960, 2920, 2880, 2860, 1670, 1450, 1410, 1380, 1195, 1140, 1030, 945, 890, 840, 705	1.52 (m, 2H, CH <sub>2</sub> ); 1.61 (s, 3H, <i>cis</i> -CH <sub>3</sub> ); 1.65 (m, 2H, CH <sub>2</sub> ); 1.68 (s, 6H, <i>trans</i> -CH <sub>3</sub> ); 2.1 (m, 10H, CH <sub>2</sub> C=C); 3.9 (m, 4H, CH <sub>2</sub> O); 4.03 (s, 2H, CH <sub>2</sub> O); 4.85 (t, 1H, $J = 4.8$ , OCHO); 5.13 (m, 2H, HC=C); 5.43 (t, 1H, $J = 6.8$ , H-5)
<b>15</b>	72	135-138/ 0.037 <sup>c</sup>	C <sub>20</sub> H <sub>34</sub> O <sub>2</sub> (306.5)	2960, 2920, 2880, 2860, 1730, 1660, 1450, 1410, 1375, 1310, 1140, 1040, 940, 875, 835, 735, 710	1.46 (m, 2H, CH <sub>2</sub> ); 1.60 (s, 3H, <i>cis</i> -CH <sub>3</sub> ); 1.64 (m, 2H, CH <sub>2</sub> ); 1.68 (s, 9H, <i>trans</i> -CH <sub>3</sub> ); 2.16 (m, 10H, CH <sub>2</sub> C=C); 3.9 (m, 4H, CH <sub>2</sub> O); 4.83 (t, 1H, $J = 4.5$ , OCHO); 5.1 (m, 3H, HC=C)
<b>16</b>	67	126-130/ 0.04 <sup>c</sup>	C <sub>18</sub> H <sub>30</sub> O (262.4)	2970, 2930, 2860, 2720, 1730, 1670, 1450, 1410, 1380, 1110, 1070, 835, 740	1.61 (s, 3H, <i>cis</i> -CH <sub>3</sub> ); 1.65 (m, 2H, CH <sub>2</sub> ); 1.68 (s, 9H, <i>trans</i> -CH <sub>3</sub> ); 2.0 (m, 10H, CH <sub>2</sub> C=C); 2.41 (dt, 2H, $J_d = 2.0$ , $J_t = 7.5$ , CH <sub>2</sub> C=O); 5.1 (m, 3H, HC=C); 9.75 (t, 1H, $J = 2.0$ , CHO)

<sup>a</sup> Yield of isolated, pure product.<sup>b</sup> Satisfactory microanalyses obtained (exception: unstable **2a,b** and **3c**): C  $\pm$  0.22, H  $\pm$  0.15, N  $\pm$  0.3, S  $\pm$  0.1.<sup>c</sup> In bath.<sup>d</sup> UV:  $\lambda_{\max} = 230$  nm, lg  $\epsilon = 4.21$ .<sup>e</sup> UV:  $\lambda_{\max} = 234$  nm, lg  $\epsilon = 4.14$ .

and dried ( $\text{MgSO}_4$ ). The solvent is removed *in vacuo*, and the residue (12.0 g) is distilled to give **3a** as a colorless liquid (Table); yield: 9.1 g (70%);  $n_D^{20}$  1.4469.

#### 5,5-Ethylenedioxy-2-trimethylsilylpentanal *tert*-Butylimine (**3c**):

A solution of **3b** (16.5 g, 83 mmol) in THF (20 mL) is added dropwise (40 min) to a stirred solution of LDA (90 mmol) in hexane (60 mL) and THF (30 mL) at  $-5^\circ\text{C}$ . The mixture is stirred for 40 min and cooled at  $-70^\circ\text{C}$ . A solution of trimethylsilyl chloride (9.22 g, 85 mmol) in THF (15 mL) is then added during 10 min, and after an additional 10 min, stirring is continued at  $-40^\circ\text{C}$  for 4 h. The reaction mixture is then treated at  $-5^\circ\text{C}$  with ice-cold  $\text{H}_2\text{O}$  (50 mL), and the aqueous phase is separated and extracted with  $\text{Et}_2\text{O}$  ( $3 \times 70$  mL). The combined organic layer is washed with sat. aq. NaCl ( $2 \times 40$  mL) and dried ( $\text{Na}_2\text{SO}_4$ ). The solvent is removed *in vacuo*, and the residue (21.8 g) is distilled to give **3c** as a colorless oil (Table); yield: 15.3 g (68%).

#### (*Z*)-6,10-Dimethyl-5,9-undecadienal *tert*-Butylimine (**2a**):

(*Z*)-6,10-Dimethyl-4-phenylthio-5,9-undecadienal Ethylene Acetal (**7**): To a stirred solution of neryl phenyl sulfide<sup>9</sup> (**6**; 16.97 g, 69 mmol) in THF (0.3 L) at  $-70^\circ\text{C}$ , 1.5 M BuLi in hexane (50 mL, 75 mmol) is added during 30 min. The reaction mixture is kept at  $-70^\circ\text{C}$  for 3.5, treated at this temperature with a solution of **5**<sup>7</sup> (19.9 g, 110 mmol) in THF (50 mL) for 20 min, stirred further for 1.5 h, and then quenched at  $-70^\circ\text{C}$  with  $\text{MeOH}/\text{Et}_2\text{O}$  (1:1, 20 mL). The mixture is poured with stirring into ice-cold  $\text{H}_2\text{O}/\text{Et}_2\text{O}$  (1:1, 0.2 L); the aqueous phase is separated and extracted with  $\text{Et}_2\text{O}$  ( $3 \times 50$  mL). The combined organic layer is washed with aq. NaCl ( $3 \times 50$  mL) and dried ( $\text{Na}_2\text{SO}_4$ ). The solvent and excess of **5** are removed *in vacuo*, and the residue (22.7 g) is chromatographed on silica gel (250 g) using gradient elution from hexane to hexane/ $\text{Et}_2\text{O}$  (85:15) to give the sulfide **7** as a colorless oil (Table); yield: 19.22 g (83%).

(*Z*)-6,10-Dimethyl-5,9-undecadienal Ethylene Acetal (**8**): To a stirred solution of Li (0.8 g, 114 mmol) in  $\text{NH}_3$  (150 mL) at  $-50^\circ\text{C}$ , a solution of **7** (8.5 g, 24.5 mmol) in  $\text{Et}_2\text{O}$  (15 mL) is added during 10 min. The reaction mixture is kept at  $-40^\circ\text{C}$  for 4 h, then quenched with an excess of solid  $\text{NH}_4\text{Cl}$ . The  $\text{NH}_3$  is evaporated, and the residue is treated with  $\text{H}_2\text{O}$  (50 mL) and extracted with  $\text{Et}_2\text{O}$  ( $5 \times 25$  mL). The extract is washed with sat. aq. NaCl ( $3 \times 25$  mL) and dried ( $\text{Na}_2\text{SO}_4$ ). The solvent is removed *in vacuo*, and the residue (5.8 g) is chromatographed on silica gel (100 g) using gradient elution from hexane to hexane/ $\text{Et}_2\text{O}$  (9:1) to give the acetal **8** as a colorless oil (Table); yield: 5.2 g (89%).

#### (*Z*)-6,10-dimethyl-5,9-undecadienal (**9**):

A solution of **8** (5.2 g, 21.8 mmol) and conc.  $\text{H}_2\text{SO}_4$  (0.1 mL) in 80% aq. acetone (0.5 L) is refluxed for 4 h, then neutralized with 10% aq.  $\text{NaHCO}_3$ , and concentrated *in vacuo*. The residue is extracted with  $\text{Et}_2\text{O}$  ( $5 \times 80$  mL). The extract is washed with sat. aq. NaCl ( $3 \times 40$  mL) and dried ( $\text{Na}_2\text{SO}_4$ ). The solvent is removed, and the residue is distilled to give the aldehyde **9** as a colorless oil (Table); yield: 4.88 g (68%).

(*Z*)-6,10-Dimethyl-5,9-undecadienal *tert*-butylimine (**2a**): To a stirred solution of **9** (2.72 g, 14 mmol) in  $\text{Et}_2\text{O}$  (70 mL) at  $-10^\circ\text{C}$ , a solution of *t*-BuNH<sub>2</sub> (5.4 g, 73 mmol) in  $\text{Et}_2\text{O}$  (10 mL) is added during 10 min. The reaction mixture is warmed to  $25^\circ\text{C}$ , kept at this temperature for 1.5 hours, and after addition of dried KOH powder (1 g), is vigorously stirred for a further 20 min. The organic phase is separated and dried ( $\text{K}_2\text{CO}_3$ ). The solvent is removed *in vacuo*, and the residue is kept at  $25^\circ\text{C}$  and 1 Torr for 6 h to give the imine **2a** as a colorless oil (Table); yield: 3.38 g (~100%).

#### (4*Z*,8*Z*)-10-Benzyloxy-4,8-dimethyl-4,8-decadienal (**1b**):

(2*E*,6*Z*)-8-Benzyloxy-2-(3,3-ethylenedioxypropyl)-2,6-octadienal (**10**): A solution of **3c** (13.0 g, 48 mmol) in  $\text{Et}_2\text{O}$  (15 mL) is added dropwise (15 min) to a stirred solution of LDA (52 mmol) in hexane (45 mL) and  $\text{Et}_2\text{O}$  (225 mL) at  $-10^\circ\text{C}$ . The mixture is stirred for 40 min and cooled at  $-70^\circ\text{C}$ . A solution of (*Z*)-6-benzyloxy-4-methyl-4-decanal<sup>10</sup> (**1a**; 8.88 g, 40 mmol) in  $\text{Et}_2\text{O}$  (15 mL) is then added during 20 min, and after stirring for 2.5 h, the mixture is warmed to  $-10^\circ\text{C}$  and kept at that temperature for 1.5 h. The reaction mixture is then quenched at  $-70^\circ\text{C}$  with a solution of AcOH (2.88 g, 48 mmol) in  $\text{Et}_2\text{O}$  (10 mL) and poured into the vigorously stirred solution of oxalic acid dihydrate

dissolved in dry  $\text{CHCl}_3$  (90 mL). The solution is left at  $25^\circ\text{C}$  under argon atmosphere for 1 day and then evaporated. The residue is kept at  $25^\circ\text{C}$  and 1 Torr for 5 h to give the aldehyde **10** as a colorless oil (Table); yield: 8.68 g (63%).

(4*E*,8*Z*)-10-Benzyloxy-4-hydroxymethyl-8-methyl-4,8-decadienal Ethylene Acetal (**11**): To a stirred solution of **10** (8.6 g, 25 mmol) in EtOH (0.2 L) at  $0^\circ\text{C}$   $\text{NaBH}_4$  (1.22 g, 32 mmol) is added during 20 min. The reaction mixture is kept at  $25^\circ\text{C}$  for 1.5 h, then quenched at  $0^\circ\text{C}$  with AcOH (0.9 g, 15 mmol), stirred for 10 min, neutralized with sat. aq.  $\text{NaHCO}_3$  and evaporated *in vacuo*. The residue is dissolved in  $\text{Et}_2\text{O}$  (50 mL), washed with sat. aq. NaCl ( $3 \times 15$  mL), dried ( $\text{Na}_2\text{SO}_4$ ) and evaporated *in vacuo*. Chromatography of the crude product (8 g) on silica gel (250 g) using gradient elution from hexane to hexane/ $\text{Et}_2\text{O}$  (7:3) gives the alcohol **11** as a colorless oil (Table); yield: 7.8 g (90%).

(4*Z*,8*Z*)-10-Benzyloxy-4,8-dimethyl-4,8-decadienal Ethylene Acetal (**12**): Sulfur trioxide-pyridine complex<sup>13</sup> (5.28 g, 33 mmol) is added during 20 min to a vigorously stirred solution of **11** (7.8 g, 22.5 mmol) in THF (0.3 L) at  $0^\circ\text{C}$ . The mixture is maintained at this temperature for 2 h and then is treated dropwise (40 min) at  $-30^\circ\text{C}$  with a 1 M solution of  $\text{LiAlH}_4$  in THF (180 mL, 180 mmol). The reaction mixture is stirred at  $25^\circ\text{C}$  for 2 days, is then treated at  $0-2^\circ\text{C}$  successively with  $\text{H}_2\text{O}$  (5 mL), 15% aq. NaOH (5 mL) and  $\text{H}_2\text{O}$  (15 mL). The solid formed is filtered and washed with  $\text{Et}_2\text{O}$  (150 mL). The aqueous phase is separated and extracted with  $\text{Et}_2\text{O}$  ( $2 \times 10$  mL). The combined organic layers are washed with sat. aq. NaCl ( $3 \times 50$  mL) and dried ( $\text{Na}_2\text{SO}_4$ ). The solvent is removed *in vacuo* and the residue is dissolved in hexane (30 mL). The filtered hexane solution is evaporated *in vacuo* to give the oily product (6.2 g), which is chromatographed on silica gel (250 g). Gradient elution from hexane to hexane/ $\text{Et}_2\text{O}$  (4:1) leads to the acetal **12** as a colorless oil (Table); yield: 5.1 g (69%).

(4*Z*,8*Z*)-10-Benzyloxy-4,8-dimethyl-4,8-decadienal (**1b**): A solution of **12** (3.46 g, 10.5 mmol) in 80% aq. acetone (0.2 L) containing conc.  $\text{H}_2\text{SO}_4$  (0.08 mL) is refluxed for 4 hours. Work-up as for **9** above and distillation of the crude product gives the aldehyde **1b**<sup>2</sup> as a colorless oil; yield: 2.17 g (72%).

#### (5*Z*,9*Z*)-6,10,14-trimethyl-5,9,13-pentadecatrienal *tert*-Butylimine (**2b**):

(*Z*)-2-(*Z*-5,5-Ethylenedioxypropylidene)-6,10-dimethyl-5,9-undecadienal (**13**): A solution of **2a** (3.44 g, 13.8 mmol) in THF (8 mL) is added dropwise (10 min) to a stirred solution of LDA (16 mmol) in hexane (12 mL) and THF (40 mL) at  $-10^\circ\text{C}$ . The mixture is stirred for 40 min and cooled at  $-50^\circ\text{C}$ . A solution of trimethylsilyl chloride (1.5 g, 13.8 mmol) in THF (12 mL) is then added during 20 min and stirring is continued for 4 h. The reaction mixture is filtered, cooled to  $-70^\circ\text{C}$  and added dropwise (30 min) to a vigorously stirred solution of LDA (16 mmol) in hexane (12 mL) and  $\text{Et}_2\text{O}$  (60 mL) at  $-50^\circ\text{C}$ . The mixture is kept at this temperature for 40 min, warmed rapidly to  $-30^\circ\text{C}$ , stirred for an additional 15 min, then cooled to  $-70^\circ\text{C}$ , and treated during 10 min with a solution of **3a** (1.58 g, 11 mmol) in  $\text{Et}_2\text{O}$  (5 mL). After 2.5 h, the reaction mixture is warmed to  $-10^\circ\text{C}$  for 1.5 h, recooled to  $-70^\circ\text{C}$ , quenched at this temperature with a solution of AcOH (0.96 g, 16 mmol) in  $\text{Et}_2\text{O}$  (3 mL) and poured into the vigorously stirred solution of oxalic acid dihydrate (6.05 g, 48 mmol) in  $\text{H}_2\text{O}$  (0.1 L) at  $5^\circ\text{C}$ . The mixture is then worked up as for **10** above to give, after chromatography of the crude product (ca. 3 g) on silica gel (100 g) using gradient elution from hexane to hexane/ $\text{Et}_2\text{O}$  (4:1) and storage at  $25^\circ\text{C}$  of this purified sample in  $\text{CHCl}_3$  solution for 1 week, the aldehyde **13** as a colorless oil (Table); yield: 1.6 g (45%).

(5*E*,9*Z*)-6-Hydroxymethyl-10,14-dimethyl-5,9,13-pentadecatrienal Ethylene Acetal (**14**): As described above for **11**, treatment of **13** (2.5 g, 7.8 mmol) with  $\text{NaBH}_4$  (0.38 g, 10 mmol) in EtOH (70 mL), work-up of the reaction mixture, and chromatography of the crude product (2.4 g) on silica gel (80 g) using gradient elution from hexane to hexane/ $\text{Et}_2\text{O}$  (7:3) gives the alcohol **14** as a colorless oil (Table); yield: 2.25 g (90%).

(5*Z*,9*Z*)-6,10,14-Trimethyl-5,9,13-pentadecatrienal Ethylene Acetal (**15**): As described above for **12**, successive treatment of **14** (5.45 g, 16.9 mmol) with sulfur trioxide-pyridine complex (4.48 g, 28.4 mmol) in THF (0.2 L) followed by 1 M solution of  $\text{LiAlH}_4$  in THF (135 mL, 135 mmol), work-up of the reaction mixture, and chromatography of the

elution from hexane to hexane/Et<sub>2</sub>O (3:1) to give the aldehyde **16** as a colorless oil (Table); yield: 2.16 g (67%).

(5Z,9Z)-6,10,14-Trimethyl-5,9,13-pentadecatrienal *tert*-Butylimine (**2b**): As described above for **2a**, starting from **16** (1.82 g, 6.9 mmol) and *t*-BuNH<sub>2</sub> (3.8 g, 36.5 mmol) in Et<sub>2</sub>O (40 mL), the imine **2b** is obtained as a colorless oil (Table); yield: 2.13 g (97%).

Received: 6 January 1989

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