

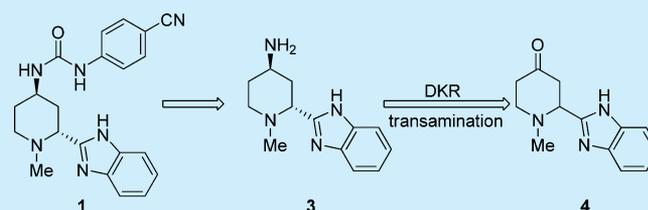
# Development of a Concise, Asymmetric Synthesis of a Smoothened Receptor (SMO) Inhibitor: Enzymatic Transamination of a 4-Piperidinone with Dynamic Kinetic Resolution

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## S Supporting Information

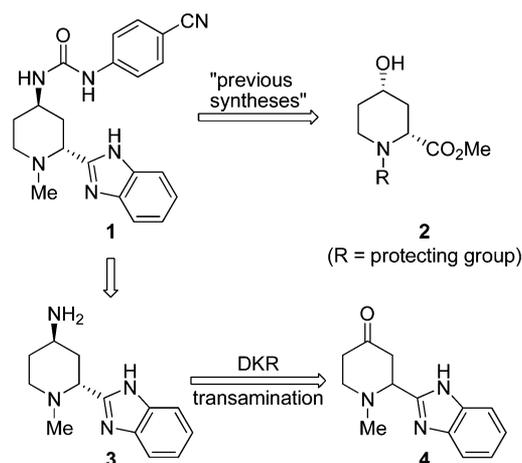
**ABSTRACT:** A concise, asymmetric synthesis of a smoothened receptor inhibitor (**1**) is described. The synthesis features an enzymatic transamination with concurrent dynamic kinetic resolution (DKR) of a 4-piperidone (**4**) to establish the two stereogenic centers required in a single step. This efficient reaction affords the desired *anti* amine (**3**) in >10:1 dr and >99% ee. The title compound is prepared in only five steps with 40% overall yield.



The smoothened receptor (SMO), a component of the hedgehog (Hh) signaling pathway, represents a novel therapeutic target in a broad range of human cancers.<sup>1</sup> Compound **1**, a potent and orally bioavailable inhibitor of SMO recently reported by Pfizer,<sup>2</sup> has shown early signs of efficacy and a good safety profile in phase 1 clinical trials involving patients with various blood-related cancers, including acute myeloid leukemia (AML).

To support the ongoing drug development program, a practical and scalable synthesis of **1** was required. The key synthetic challenge for the preparation of **1** was the effective and practical establishment of the *anti* stereogenic centers on the 2,4-positions of the piperidine ring. Previous syntheses<sup>2,3</sup> employed (2*R*,4*S*)-4-hydroxypipercolic acid derivatives **2** (Scheme 1), which were not readily available on scale. The syntheses were also tedious, and the overall yields were low.<sup>3</sup> Furthermore, installation of the C-4 amino group required three steps (alcohol activation, *S<sub>N</sub>2* displacement with azide, and reduction) and posed safety concerns in large scale preparation. Over the past few decades, elegant methodologies have been developed for the synthesis of chiral amines.<sup>4</sup> However, reports of direct enantioselective reductive amination of prochiral ketones in the absence of extra functional group interconversion steps are rare.<sup>5</sup> In this regard, enzymatic transamination reactions provide an advantage over traditional chemical synthesis.<sup>6</sup> The increased availability of transaminases and the advancement of enzyme engineering have made enzymatic transamination an increasingly attractive and viable manufacturing option for chiral amine synthesis in pharmaceuticals.<sup>7</sup> We envisioned that an enzymatic transamination of 4-piperidone **4** would not only generate the C-4 chiral amino group required in the penultimate **3** but also establish the *anti* C-2 stereocenter at the same time through a dynamic kinetic resolution (DKR) process.<sup>8</sup> It should be noted that the facile

## Scheme 1. Retrosynthetic Analysis



racemization and a crystallization-driven DKR of 2-aryl-4-piperidones have been reported in the literature previously.<sup>9</sup>

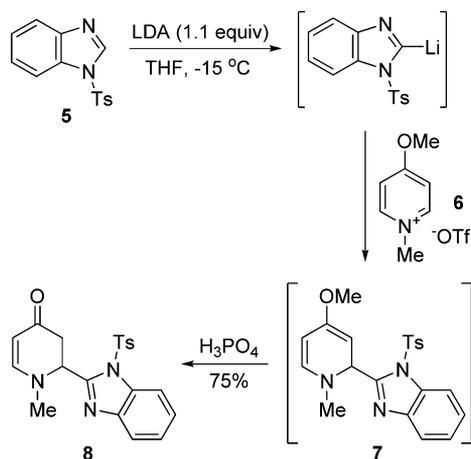
To assemble 4-piperidone **4**, we elected to explore the nucleophilic addition of benzimidazole to a pyridinium salt, which has proven to be a powerful method for the synthesis of substituted piperidines.<sup>10</sup> To the best of our knowledge, heterocyclic carbon nucleophiles such as benzimidazole have not been employed in this approach. We were pleased to find that the *N*-Ts benzimidazole **5**<sup>11</sup> was fully deprotonated with LDA at  $-15\text{ }^{\circ}\text{C}$  and underwent addition to 4-methoxy *N*-methyl pyridinium triflate **6**<sup>12</sup> to give the corresponding dihydropyridine **7**. Following mild acidic hydrolysis of the

Received: December 16, 2013

Published: January 22, 2014

enol ether the enone could be unmasked to furnish dihydropyridone **8** (Scheme 2). A competing reaction pathway

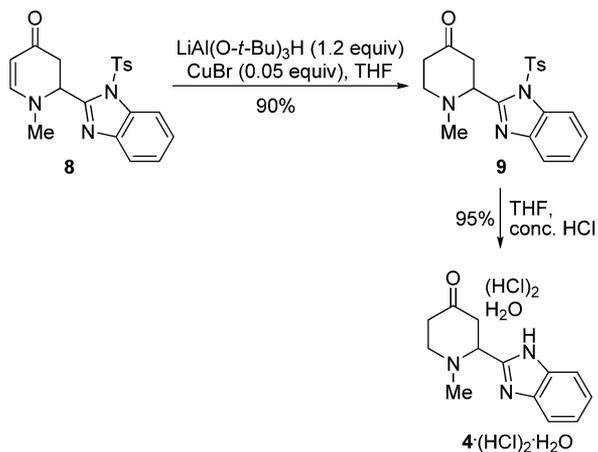
### Scheme 2. Preparation of Dihydropyridone **8**



was found to be the deprotonation of the  $\alpha$ -C–H of the pyridinium salt<sup>13</sup> by the lithiated benzimidazole and resulted in regeneration of 10–20% of *N*-Ts benzimidazole **5**. All attempts to improve selectivity through examination of reaction parameters (bases, temperature, concentration, addition rates, etc.) were unfruitful. Nevertheless, the product could be isolated in high purity (>98%) by crystallization from isopropyl acetate in 75% yield.

Reduction of enone **8** to the corresponding ketone and removal of the Ts group were required to prepare 4-piperidone **4**. The conjugate reduction of **8** proved to be challenging due to competitive over-reduction to the saturated alcohol. Following experimentation, we found that  $\text{LiAl}(\text{O}-t\text{-Bu})_3\text{H}$  in the presence of  $\text{CuBr}$ <sup>14</sup> gave excellent selectivity with less than 2% of the over-reduced alcohol observed (Scheme 3).

### Scheme 3. Preparation of 4-Piperidone **4**

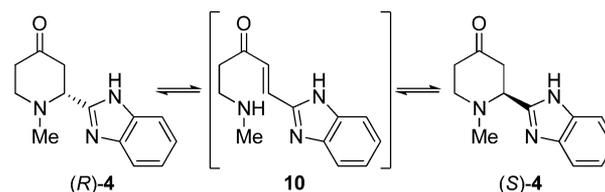


Optimized conditions were identified when a slight excess of  $\text{LiAl}(\text{O}-t\text{-Bu})_3\text{H}$  (1.2 equiv) and only 5 mol % of  $\text{CuBr}$  were employed.<sup>15</sup> Piperidone **9** could be isolated in 90% yield and with high purity (>99%) by crystallization from methanol– $\text{H}_2\text{O}$ . Finally, the *N*-Ts protecting group of **9** was found to be labile to both aqueous basic ( $\text{NaOH}$ ) and acidic ( $\text{HCl}$ ) conditions. While the basic conditions generated over 10% of the aldol dimer byproducts, the acidic conditions ( $\text{THF}$ , conc.

$\text{HCl}$ ) cleanly afforded 4-piperidone **4** in 95% isolated yield by direct filtration as a crystalline di- $\text{HCl}$  salt hydrate.<sup>16</sup>

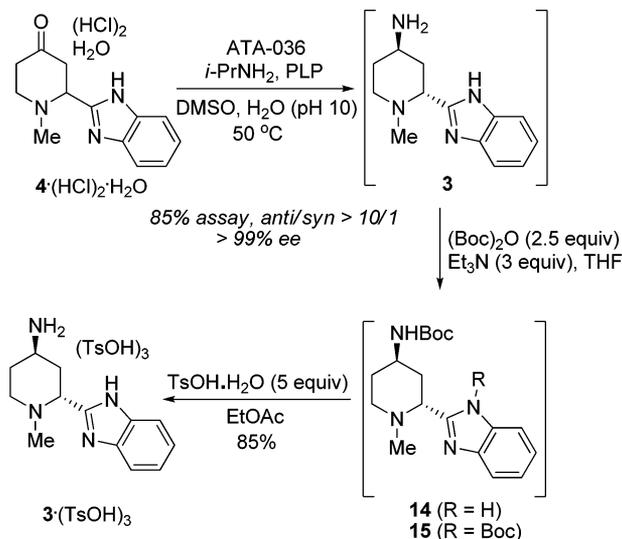
With 4-piperidone **4** in hand, the stage was set for the crucial enzymatic transamination with DKR. To investigate the feasibility of the racemization of **4**, we obtained enantiomerically pure material by chiral SFC separation of racemic **4**. Complete racemization of a single enantiomer was found to occur in less than 8 h at 40 °C in a 1/4 mixture of DMSO and aqueous pH 10 buffer, the medium for the enzymatic transamination. Although we did not have direct evidence of the corresponding ring-opened intermediate **10**, we suspected that retro-aza-Michael/aza-Michael was the mechanism by which the racemization of **4** occurred (Scheme 4).<sup>17</sup>

### Scheme 4. Racemization of 4-Piperidone **4**



Following the investigation of commercially available transaminases, we were delighted to find several enzymes that catalyzed the transamination of **4**.<sup>18,19</sup> Enzyme ATA-036 was identified as catalyzing highly efficient transamination with DKR to afford the desired (2*R*,4*R*)-amine **3** in high conversion and excellent selectivities. Moreover, ATA-036 demonstrated excellent thermo-stability and catalyzed the transamination of **4** in reactions with temperatures up to 75 °C. Unfortunately, 4-piperidone **4** was found to be unstable at elevated temperatures as shown by a control experiment, where in the absence of any enzyme approximately 56% of **4** decomposed after 23 h at 60 °C. Performing reactions at less than 60 °C minimized degradation of **4**, but required longer reaction times compared to reactions at higher temperatures. Consequently, a balance was obtained for the transamination reaction through heating the reaction at 50 °C for 50–60 h, generating amine **3** in 85% assay yield with an *anti*/*syn* ratio of >10:1 and >99% ee (Scheme 5).<sup>20</sup>

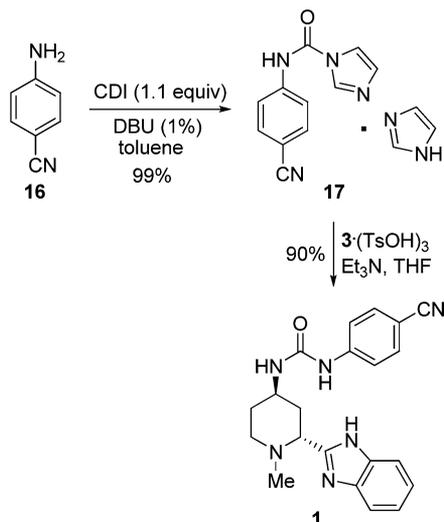
### Scheme 5. Transamination and DKR of 4-Piperidone **4**



The transamination product **3** was found to be highly soluble in DMSO–H<sub>2</sub>O and was very difficult to extract with organic solvents. Thus, we chose to employ an in situ *N*-Boc protection of the resulting amine with (Boc)<sub>2</sub>O. After the transamination was complete, the DMSO–H<sub>2</sub>O solution of **3** was treated with excess (Boc)<sub>2</sub>O and Et<sub>3</sub>N to give a mixture of mono- and bis-Boc compounds **14** and **15**, which could be extracted with EtOAc (Scheme 5). The crude mixture of *N*-Boc compounds in EtOAc was then treated with excess tosylic acid for removal of the Boc groups. Fortunately, the tritosylate salt of **3** was directly crystallized from the reaction mixture and was isolated in 85% yield and >99% purity with the undesired *syn* isomer observed at <0.1%.

With the key penultimate **3** prepared from the enzymatic transamination, the final step of the synthesis of **1** involved formation of the urea. Previously, the urea moiety was formed by coupling of the amine with 4-cyanophenyl isocyanate. The use of this isocyanate on scale, however, is undesirable due to cost and stability issues. The corresponding *N*-carbamoylimidazole reagent, which can be prepared from the desired aniline and *N,N*-carbonyldiimidazole (CDI), has been utilized as an efficient alternative to the isocyanate in urea formations.<sup>21</sup> Indeed, we found that *N*-carbamoylimidazole **17** was easily prepared by the reaction of 4-aminobenzonitrile **16** with a slight excess of CDI (1.1 equiv) in toluene at ambient temperature, and **17** was isolated via direct filtration as a 1:1 mixture with imidazole in almost quantitative yield (Scheme 6). We also

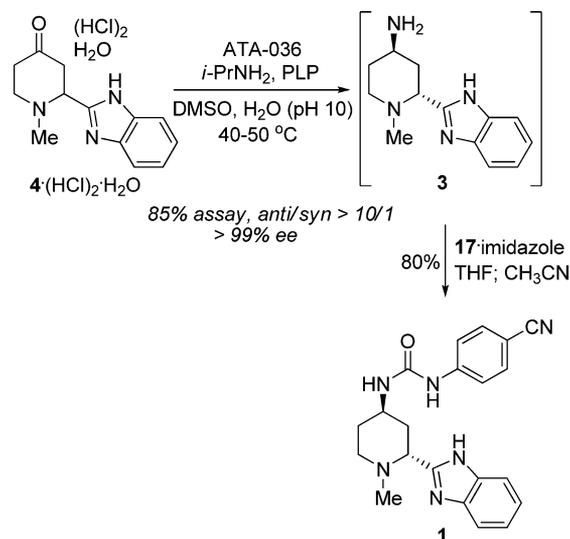
**Scheme 6. Preparation of *N*-Carbamoylimidazole **17** and Synthesis of **1** with Isolated **3**·(TsOH)<sub>3</sub>**



discovered that the addition of only 1 mol % DBU yielded shortened reaction times from over 12 h to 2–3 h.<sup>22,23</sup> The reaction of **17** (mixture with imidazole) with the tritosylate salt of **3** proceeded rapidly in the presence of Et<sub>3</sub>N in THF to furnish **1** in >95% assay yield. Following aqueous workup, **1** was isolated by crystallization from acetonitrile in 90% yield and >99% purity.

Notably, we were delighted to find that *N*-carbamoylimidazole **17** reacted with **3** directly in the aqueous medium employed in the transamination reaction (Scheme 7), which eliminated the need for isolation of **3** through the *N*-Boc/de-Boc sequence.<sup>24</sup> A slight excess of **17** (1.2 equiv) was required to scavenge the remaining isopropylamine from the trans-

**Scheme 7. Synthesis of **1** with Crude **3****



amination reaction. Product **1** was formed in >90% assay yield and was isolated similarly by crystallization from acetonitrile in 80% yield and >98% purity.<sup>25</sup>

In summary, an efficient enzymatic transamination and DKR of 4-piperidone **4** enabled the development of a concise, asymmetric synthesis of the SMO inhibitor **1**. The entire synthesis required only five linear steps and produced **1** in 40% overall yield without the need for chromatography, thus, rendering this synthesis suitable for large scale commercial supply of **1**.

## ■ ASSOCIATED CONTENT

### Supporting Information

Experimental procedures, compound characterization, and copies of <sup>1</sup>H and <sup>13</sup>C NMR spectra. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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### Notes

The authors declare no competing financial interest.

## ■ ACKNOWLEDGMENTS

The authors would like to thank Pfizer colleagues Brian Jones for HRMS analyses; Robert Singer and Frank Busch for helpful discussions; and Nick Thomson and Stéphane Caron for management support.

## ■ REFERENCES

- (1) (a) Heretsch, P.; Tzagkaroulaki, L.; Giannis, A. *Bioorg. Med. Chem.* **2010**, *18*, 6613–6624. (b) Rubin, L. L.; de Sauvage, F. *Nat. Rev. Drug Discovery* **2006**, *5*, 1026–1033.
- (2) Munchhof, M. J.; Li, Q.; Shavnya, A.; Borzillo, G. V.; Boyden, T. L.; Jones, C. S.; LaGreca, S. D.; Martinez-Alsina, L.; Patel, N.; Pelletier, K.; Reiter, L. A.; Robbins, M. D.; Tkalcovic, G. T. *ACS Med. Chem. Lett.* **2012**, *3*, 106–111.
- (3) The current scale-up route was enabled from the original discovery route (ref 2), which produced tens of kilograms of **1** in 14 linear steps with an overall yield of only 4%. In this enabling route, (2*R*, 4*S*)-4-hydroxypiperidone **2** was prepared with a modified literature procedure: Gillard, J.; Abraham, A.; Anderson, P. C.;

Beaulieu, P. L.; Bogri, T.; Bousquet, Y.; Grenier, L.; Guse, I.; Lavallée, P. *J. Org. Chem.* **1996**, *61*, 2226–2231.

(4) Nugent, T. C.; El-Shazly, M. *Adv. Synth. Catal.* **2010**, *352*, 753–819.

(5) Matsumura, K.; Zhang, X.; Hori, K.; Murayama, T.; Ohmiya, T.; Shimizu, H.; Saito, T.; Sayo, N. *Org. Process Res. Dev.* **2011**, *15*, 1130–1137.

(6) Koszelewski, D.; Tauber, K.; Faber, K.; Kroutil, W. *Trends Biotechnol.* **2010**, *28*, 324–332.

(7) Savile, C. K.; Janey, J. M.; Mundorff, E. C.; Moore, J. C.; Tam, S.; Jarvis, W. R.; Colbeck, J. C.; Krebber, A.; Fleitz, F. J.; Brands, J.; Devine, P. N.; Huisman, G. W.; Hughes, G. J. *Science* **2010**, *329*, 305–309.

(8) The transaminase DKR approach has been reported with the chiral center  $\alpha$  to the carbonyl groups; see: Chung, C. K.; Bulger, P. G.; Kosjek, B.; Belyk, K. M.; Rivera, N.; Scott, M. E.; Humphrey, G. R.; Limanto, J.; Bachert, D. C.; Emerson, K. M. *Org. Process Res. Dev.* **2014**, *18*, 215–227.

(9) (a) Bravo, F.; Cimarosti, Z.; Tinazzi, F.; Smith, G. E.; Castoldi, D.; Provera, S.; Westerduin, P. *Org. Process Res. Dev.* **2010**, *14*, 1162–1168. (b) Lee, S. K.; Tambar, U. K.; Perl, N. R.; Leighton, J. L. *Tetrahedron* **2010**, *66*, 4769–4774.

(10) Bull, J. A.; Mousseau, J. J.; Pelletier, G.; Charette, A. B. *Chem. Rev.* **2012**, *112*, 2642–2713.

(11) Other protecting groups on the nitrogen of benzimidazole gave a lower yield than the tosyl group.

(12) Other pyridinium salts such as iodide, tosylate, methyl sulfate, and hexafluorophosphate were much less soluble in THF and gave poor yields.

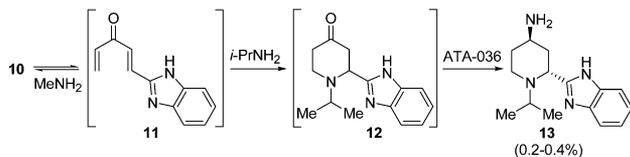
(13) The  $pK_a$  of the  $\alpha$ -C–H of the 4-methoxypyridinium salt was measured to be  $\sim 33$ ; see: Wong, F. M.; Capule, C. C.; Chen, D. X.; Gronert, S.; Wu, W. *Org. Lett.* **2008**, *10*, 2757–2760. And the  $D_2O$  quench of the reaction confirmed the deprotonation at the  $\alpha$ -carbon of the pyridinium salt.

(14) (a) Focken, T.; Charette, A. B. *Org. Lett.* **2006**, *8*, 2985–2988. (b) Semmelhack, M. F.; Stauffer, R. D.; Yamashita, A. *J. Org. Chem.* **1977**, *42*, 3180–3188.

(15) Previously reported procedures required more than 1 equiv of copper (I) salts; see ref 14.

(16) Alternatively, the free base of **4** could be isolated in 87% yield with a slightly different procedure; see Supporting Information for details.

(17) Indirect evidence for the racemization mechanism came from the *N*-isopropyl impurity **13** that was formed in the transamination reaction at 0.2–0.4% level. This impurity was generated by the transamination of the *N*-isopropyl piperidone **12**, which was produced by a second retro-aza-Michael of **10** with the elimination of methylamine followed by a double aza-Michael reaction with isopropylamine, the amine donor for the transamination reaction:



(18) In the initial screening (30 °C, 19 h), eight (*R*)-selective transaminases from Codexis gave >10% assay yields with various diastereoselectivities (assay, *anti/syn* ratio): ATA-015 (15%, 1.7), ATA-016 (21%, 1.1), ATA-024 (48%, 4.1), ATA-025 (41%, 10.3), ATA-033 (42%, 9.1), ATA-034 (42%, 5.9), ATA-035 (35%, 10.2), ATA-036 (61%, 8.3). Transaminase ATA-036 was selected for further evaluation based on activity and diastereoselectivity, and subsequent chiral analysis showed that reactions with this enzyme were highly enantioselective and only produced the desired (2*R*,4*R*)-amine. Many of the Codexis (*S*)-selective transaminases also showed activity to produce the undesired (2*S*,4*S*)-amine.

(19) Both the free base and di-HCl salt of **4** were used in the transamination reaction with similar results.

(20) There were no significant differences in both diastereo- and enantioselectivity with temperatures ranging from 40 to 70 °C.

(21) Rawling, T.; McDonagh, A. M.; Tattam, B.; Murray, M. *Tetrahedron* **2012**, *68*, 6065–6070 and references cited therein.

(22) Larrivée-Aboussafy, C.; Jones, B. P.; Price, K. E.; Hardink, M. A.; McLaughlin, R. W.; Lillie, B. M.; Hawkins, J. M.; Vaidyanathan, R. *Org. Lett.* **2010**, *12*, 324–327.

(23) The isolated solids consisting of **17** and imidazole showed no sign of degradation after several months at ambient temperature.

(24) For urea formation with *N*-carbamoylimidazole in water, see: Padiya, K. J.; Gavade, S.; Kardile, B.; Tiwari, M.; Bajare, S.; Mane, M.; Gaware, V.; Varghese, S.; Harel, D.; Kurhade, S. *Org. Lett.* **2012**, *14*, 2814–2817.

(25) A slururry of the isolated solids was needed to purge the undesired *syn* isomer to <1%.