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Raman scattering studies of the Ge–In sulfide glasses

Haizheng Tao *, Shun Mao, Guoping Dong, Haiyan Xiao, Xiujian Zhao *

Key Laboratory of Silicate Materials Science and Engineering, Wuhan University of Technology, Ministry of Education, 122 Luoshi Road, Hongshan-Qu, Wuhan, Hubei 430070, People's Republic of China

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Abstract

Room temperature Raman scattering measurements of samples of the $(1-x)GeS_2-xIn_2S_3$ (x=0.00-0.35) system have been conducted together with the FTIR transmission spectra of partial samples. Based on the Raman scattering and FTIR transmission spectra of the prepared defect spinel polycrystalline In₂S₃, the additional peaks at 306 and 245 cm⁻¹ were ascribed to the local symmetric stretching vibration of InS₄ tetrahedra and InS₆ octahedra, respectively. According to the Raman scattering spectral evolution of the Ge–In sulfide glasses, the microstructure of the studied glasses was considered to be that S₃Ge–GeS₃ and S₃In–InS₃ ethane-like units originated from the sulfur deficient with the addition of In₂S₃ are homogeneously dispersed in a disordered polymer network formed by GeS₄, InS₄ tetrahedra and a small quantity of InS₆ octahedra interconnected by sulfur bridges.

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1. Introduction

Recently, Ga_2S_3 -based chalcogenide glasses have attracted much attention because of their possible applications such as ultra-fast all optical switching, rare-earth doped 1.3 µm fiber amplifier and, etc. [1–4]. In addition, based on the similarity of chemical properties with Ga and In, it can be anticipated that In_2S_3 -based chalcogenide glasses also be the leading candidates in the above-mentioned fields. However, to our best knowledge, there are rare reports about the microstructure of In_2S_3 -based chalcogenide glasses. Considering the intimate relations among the above-mentioned applications, microstructure and basic properties of materials, understanding and utilization of the basic studying findings are worthwhile from theoretical and practical viewpoints.

The understanding and utilization of the microstructure of the Ge–In sulphide glasses is a key clue to further elucidating the short range order of the In_2S_3 -based chalcogenide glasses. The main objective of this work was to study the

microstructure of the (1-x)GeS₂-xIn₂S₃ pseudo-binary glasses through the Raman scattering and FTIR technique.

2. Experimental procedure

The samples were prepared from particular high-purity elements (at least 99.9%) using the 10 mm inner-diameter silica tube by conventional melt-quenching techniques. For details see the Refs. [5,6].

Chemical compositions of the samples were analyzed using an energy-dispersive XRF analyzer. Homogeneity and amorphous characteristics of the prepared materials were confirmed by optical and electron microscopy and X-ray diffraction (XRD) with Cu K_{α} radiation.

Raman scattering (RS) and Fourier-transform infrared (FTIR) transmission spectra were recorded at room temperature. The Raman measurements were performed in a 180° backscattering geometry using an inVia Laser Confocal Raman Microscope made by Renishaw PLC. The 632.8 nm line of a He–Ne laser was used and the spectral resolution was of the order of 1 cm⁻¹. For the avoidance of local laser damage that could easily occur under the microscope and could locally crystallize the amorphous samples, a laser power not exceeding an approximate level of 2 mW was used. FTIR transmission spectra were recorded using a Nicolet Nexus Fourier-transform

^{*} Corresponding authors. Tel./fax: +86 27 87669729.

E-mail addresses: taohaizheng@hotmail.com (H.Z. Tao), opluse@mail. whut.edu.cn (X.J. Zhao).



Fig. 1. Several sorts of typical Raman spectra of the sample 10 $(0.65 GeS_2 - 0.35 In_2S_3)$ under the microscope.

infrared spectrometer under the N₂ gas atmosphere. Each FTIR spectrum is the average of 64 scans at a spectral resolution of 4 cm^{-1} . The samples were ground, then dispersed in CsI, and pressed at 5 kPa into pellets to obtain far-infrared spectra in the range 200–600 cm⁻¹.

3. Results

Samples with x up to 0.3 were confirmed to be optically homogeneous and transparent glasses to the eye and according to the methods given above. And these $(1-x)GeS_2-xIn_2S_3$ (x=0.00-0.30) glasses changed from pale yellow through tint red to deep red in color with the addition of In₂S₃. Based on the XRF analysis, the final glasses preserved their initial nominal compositions except for negligible losses in sulfur concentration. So, hereinafter, the glass composition was expressed by the batch composition. Glass formation seemed to be limited to the range of x=0.00-0.30 and several attempts made to prepare a glass with higher In₂S₃ content failed. According to the XRD pattern, only the polycrystalline In₂S₃ separates out within the sample 10 (x=0.35). In addition, this sample is very inhomogeneous to the eye and several types of Raman spectra were obtained under the assistance of the microscope (Fig. 1).

Fig. 2 represents the Raman spectra of the $(1-x)GeS_2-xIn_2S_3$ glasses (sample 0-9). Five spectral changes can be extracted following the addition of In₂S₃ into the GeS₂ glass. First, in the region 100–200 cm⁻¹, the little prominence at 150 cm⁻¹ is gradually enhanced with the increased content of In₂S₃. Secondly, in the region $200-270 \text{ cm}^{-1}$, a clear spectral evolution about the location of the dominating peak from 255 to 245 cm^{-1} can be seen going with the gradual increasing about the intensity of the shoulder at about 225 cm^{-1} . The third change is about the increasingly enhancement of the peak at 306 cm^{-1} in the 270–320 cm⁻¹ range. Fourthly, the strongest peak at 340 cm^{-1} is broadened little by little towards two sides with the increased content of In₂S₃. And the descending scattering intensity by inchmeal of the shoulder at 370 cm^{-1} is the final change originated from the addition of In_2S_3 . To further attain the micro-structural information, the FTIR transmission spectra of the glassy samples (x=0.00-0.25) were presented in Fig. 3. However, no distinct spectral transformations are found except for the gradual descending of the characteristic absorption band of the GeS₂ glass at 377 cm⁻¹.

Finally, polycrystalline In_2S_3 was synthesized by direct reaction of a stoichiometric mixture of the elements using the melt-quenching slowly method. The reaction led to a reddish powder that X-ray powder diffraction showed to be the pure defect spinel phase In_2S_3 (JCPDF No: 84-1385). The Raman



Fig. 2. Raman spectra of the (1-x)GeS₂-xIn₂S₃ glasses. 0, x=0.00; 1, x=0.01; 2, x=0.025; 3, x=0.04; 4, x=0.07; 5, x=0.10; 6, x=0.15; 7, x=0.20; 8, x=0.25; 9, x=0.30.



Fig. 3. FTIR spectra of the (1-x)GeS₂-xIn₂S₃ glasses. 1, x = 0.00; 2, x = 0.05; 3, x = 0.10; 4, x = 0.15; 5, x = 0.20; 6, x = 0.25.

and FTIR spectra of this polycrystalline sample were shown in Fig. 4. Group theoretical treatment of the spinel structure yields four infrared active modes of species F_{1u} and five Raman active ones, one each species A_{1g} and E_g , three of species F_{2g} [7]. According to the research of Lutz et al. [7,8], the Raman peak at 366 cm⁻¹ of the prepared polycrystalline In₂S₃ can be ascribed to the A_{1g} mode, the peak at 262 cm⁻¹ to the E_g mode, and the peaks at 117,179,322 cm⁻¹ to the F_{2g} mode. As to the additional peaks at 306 and 245 cm⁻¹, which are stronger than the Raman active modes originated from the spinel structure, it can be rationally presumed that these two additional peaks are so-called clustering modes caused by high density of states of distinct phonon branches at $q \neq 0$, which is Raman allowed

because of the breakdown of the translation symmetry. And this can be confirmed by the random distribution of indium on the octahedral and tetrahedral sites according to the JCPDF card together with the corresponding absorption bands of the FTIR spectrum.

In terms of our previous study [9], the A_{1g} breathing mode of GaS₄ tetrahedra locates at about 342 cm⁻¹ in the GeS₂– Ga₂S₃ pseudo-binary glassy system. Based on the similarity of chemical properties for Ga and In located in the same main family, it is reasonable to presume the same force constants among the GaS₄ and InS₄ tetrahedra. According to the theory of molecular vibration [10], then, the A_{1g} breathing mode of InS₄ tetrahedra should locate at about 315 cm⁻¹ considering



Fig. 4. Raman and FTIR transmission (CsI plate) spectra of the polycrystal In₂S₃ sample.

the bigger weight of In compared with Ga. Furthermore, considering the less force constant of In-S bonds compared with Ga–S ones, the A_{1g} mode of InS_4 tetrahedra should locate at the lower wave number than 315 cm⁻¹. So it is rational to attribute the additional peak at 306 cm⁻¹ to the symmetric stretching vibration of InS_4 tetrahedra.

According to the analysis about the Raman spectra of inverse and partially inverse indium sulfide spinels, Lutz et al. [7] ascribed the bands in the range 240–255 to the A_{1g} breathing mode based on the facts that the intensity of this band increases with increase in the degree of inversion of the spinels. So it is reasonable to ascribed Raman peak at 245 cm⁻¹ to the symmetric stretching vibrating of InS₆ octohedra.

4. Discussion

The Raman spectrum of the compound GeS₂ glass in Fig. 1 (0) is similar to those reported previously [11-13]. Four bands can be distinctly observed at 115, 340 (A₁), 370 (A_1^c) and 431 cm^{-1} . The broad band at 115 cm^{-1} , superimposed on the high frequency tail of the boson peak, was ascribed ordinarily to the bond bending motions of S atoms in GeS₄ tetrahedra, while Kotsalas, etc. [11] suggested that considerable contribution to this band may arise from intermolecular displacements of whole GeS₄ structural units. The bands at 340 and 370 cm^{-1} are strongly polarized and the former is attributed undoubtedly to the A₁ symmetric stretching vibrations of GeS₄ tetrahedra, while the latter is due to the symmetric stretch vibrations of S atoms in bridges of edge-sharing $(GeS_{1/2})_4$ tetrahedra. The shoulder at 431 cm^{-1} is relatively depolarized (to the A₁ and A^c₁ bands) and generally recognized to be due to the S-S and/or multi-sulfur bonds although there were some controversies [14]. And this gives a probable existence of small S–S chains in g-GeS₂ either as separate molecular units (inhomogeneities) or as interconnecting units between GeS₄ units. In addition, the very weak Raman band at 255 cm^{-1} (can be attributed to the vibrational mode of the ethane-like units S₃Ge–GeS₃; and only can be seen clearly through local magnification of Raman spectra) give another proof that the microstructure of the stoichiometric GeS₂ glass does not consist only of GeS₄ tetrahedra.

According to the Fig. 1, the spectrum 1 exactly corresponds to the Raman spectra of the prepared polycrystalline In_2S_3 which further verified the result of the XRD analysis. As to the spectrum 4, the characteristic bands of the $S_3In-InS_3$ ethanelike units can be clearly distinguished based on the distinct broad peak at about 153 cm⁻¹ together with the small shoulders at about 226 and 321 cm⁻¹ [15]. In addition, the main peak at 340 cm⁻¹ is due to the symmetric stretching mode of GeS₄ and the small prominence at about 255 cm⁻¹ can be ascribed to the vibrating of the S_3Ge -GeS₃ ethane-like units. Furthermore, the bands at 306 and 244 cm⁻¹ can be attributed to the symmetric stretching vibrations of InS₄ and InS₆ units, respectively, based on the above-mentioned ascription. Lastly, the spectra 2 and 3 are intervenient between the spectrum 1 and 4.

Now let us see the spectral evolution of the (1-x) GeS₂ xIn_2S_3 glasses (x=0-0.3). Firstly, the clear evolution of the region 270–320 cm^{-1} from a small shoulder to a distinct peak at 306 cm^{-1} gives the direct proof about the existence of InS_4 tetrahedra within the Ge-In sulfide glassy net. In addition, following the addition of In₂S₃, the similar transformation of the region 225–255 cm^{-1} from a small prominence to a dominating peak at 245 cm^{-1} within the region $200-270 \text{ cm}^{-1}$ indicates another coordination surroundings of In atoms, viz. InS₆ octahedra. And the weaker scattering strength of the prominence at 245 cm⁻¹ compared with that of another one at 306 cm⁻¹ manifests itself that the added indium exists mainly in the form of InS₄ tetrahedra. Secondly, the incremental quantity of the ethanelike units S₃In–InS₃ with the addition of In₂S₃ within the glassy net can be confirmed by the gradual enhancement of the characteristic bands at about 150 and 225 cm⁻¹ together with the slowly broadening of the main peak at 340 cm^{-1} toward lower wave number aspect originated from another diagnostic band at 321 cm⁻¹ of the S₃In–InS₃ units. Thirdly, the far weaker intensity of the prominence of the band at about 225 cm^{-1} compared with that of another one at 255 cm^{-1} gives us the information that the ethane-like units S₃Ge–GeS₃ comes into being more preferentially than another similar units S₃In-InS₃ when sulfur is deficient due to the addition of In₂S₃. And the gradually widening of the main peak at 340 cm^{-1} toward the higher wave number aspect from sample 0-9 can be explained from the gradual enhancement of another characteristic band at about 360 cm⁻¹ [12] of the ethane-like units S₃Ge–GeS₃. Lastly, the descending scattering intensity by inchmeal of the shoulder at 370 cm^{-1} is due to the descending amount of bridged units Ge₂S₆ originated from the increasingly enhanced quantity of other units (e.g. InS_4 and InS_6 , etc.) following the addition of In_2S_3 .

5. Conclusion

The Raman scattering and FTIR transmission spectra of the prepared polycrystalline In_2S_3 that belongs to the defect spineltype structure was reported. The additional peaks at 306 and 245 cm⁻¹ were attributed to the local symmetric stretching vibrating of InS_4 and InS_6 microstructural units respectively.

The inhomogeneous nature of sample 10 $(0.65 \text{GeS}_2 - 0.35 \text{In}_2 \text{S}_3)$ was clearly confirmed by the Raman scattering measurement. Based on the spectral evolution of the (1-x) GeS₂-xIn₂S₃ glasses, the following micro-structural information can be deduced:

- (1) The added indium exists mainly in the form of InS_4 tetrahedra together with a small quantity of InS_6 octahedra within the glassy network.
- (2) With the addition of In₂S₃, the ethane-like units S₃Ge–GeS₃ are more preferentially formed than the similar S₃In–InS₃ units.

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