# Chemical Synthesis and Pharmacology of 6- and 7-Hydroxylated 2-Carbomethoxy-3-(p-tolyl)tropanes: Antagonism of Cocaine's Locomotor **Stimulant Effects**

Lianyun Zhao,<sup>†</sup> Kenneth M. Johnson,<sup>‡</sup> Mei Zhang,<sup>‡</sup> Judith Flippen-Anderson,<sup>§</sup> and Alan P. Kozikowski<sup>\*,†</sup>

Drug Discovery Program, Department of Neurology, Georgetown University Medical Center, 3900 Reservoir Road, NW, Washington, D.C. 20007-2197, Department of Pharmacology and Toxicology, University of Texas Medical Branch, Galveston, Texas 77555-1031, and Laboratory for the Structure of Matter, Code 6030, Naval Research Laboratory, 4555 Overlook Avenue, SW, Washington, D.C. 20375-5000

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In our efforts to identify molecules that might act as cocaine antagonists or cocaine partial agonists, efforts were made to further capitalize on our earlier finding regarding the ability of a 7-methoxylated pseudococaine analogue to act as a weak cocaine functional antagonist. Herein, a series of the 6- and 7-hydroxylated WIN analogues possessing a boat or chair conformation of the tropane ring were prepared and tested for their ability to displace [<sup>3</sup>H]mazindol binding and to inhibit high-affinity monoamine uptake into rat brain nerve endings. These 6- and 7-hydroxylated WIN analogues were readily prepared by use of a classical Willstätter synthesis to construct an appropriately functionalized tropane ring followed by use of a Suzuki coupling reaction to introduce the aryl group at position 3. Reduction of the resulting tropene by use of SmI<sub>2</sub> or by catalytic hydrogenation followed by deprotection delivered the final target compounds. Some of these compounds were found to retain considerable affinity as inhibitors of the dopamine transporter (DAT) and the norepinephrine transporter (NET), but they were less potent inhibitors of the serotonin transporter (SERT). None of the compounds of the present series revealed any substantial potency difference in [3H]mazindol binding versus [3H]DA uptake, and failed to show "cocaine antagonism" when tested for their ability to prevent cocaine's inhibition of DA transport. However, one of these hydroxylated WIN analogues, namely 12b, which possesses nanomolar potency at the DAT and NET and micromolar potency at the SERT, when tested in vivo, was found capable of attenuating cocaine's locomotor activity ( $AD_{50} = 94$ mg/kg). Taken together, this work provides further support for our hypothesis that drugs that lack the ability to inhibit transport by all three monoaminergic transporters may exhibit "partial" cocaine-like properties, but act as cocaine antagonists. Consequently, it may prove valuable to examine the behavioral activity of other 6- and 7-substituted tropanes in animal behavioral paradigms in the search for a cocaine medication.

## Introduction

Immediate therapies are needed for the treatment of cocaine abuse worldwide.<sup>1</sup> It has been estimated, for example, that there are at present over 1.5 million cocaine abusers in the United States alone, and that approximately one-third of these users are consistent or hard core users. Undoubtedly, cocaine abuse has a tremendous negative impact on our society, for it has been estimated that the health and crime costs associated with cocaine abuse amount to about 100 billion dollars per year. Moreover, it has been reported that approximately one-half million Americans ended up in hospital emergency rooms last year. Cocaine use figured into about 28% of these emergency visits, up by 15% from 1993. At present there are no effective medications available to treat cocaine addiction, although selegeline is in phase III trials.<sup>1</sup>

To develop agents that might find use in the treatment of cocaine abuse, the search for both cocaine antagonists and partial agonists is being pursued. While antagonists are more likely to find use in situations of cocaine overdose, the partial agonist approach may prove more useful in maintenance programs. Compounds that possess the ability to mimic partially the effects of cocaine may help to maintain individuals in treatment programs in a manner analogous to methadone, a drug widely used in the treatment of opiate abuse. Alternatively, or in addition, such "partial agonists" may be able to prevent the effects of cocaine.

In pursuit of possible medications we have focused much of our chemistry efforts on the modification of cocaine itself, in the belief that appropriate structural alterations may lead to the desired type of partial agonist or antagonist activity we seek. In fact, some time ago we reported the ability of a 7-methoxylated analogue of pseudococaine to act as weak antagonist of cocaine, in that it was able to inhibit cocaine's ability to block dopamine reuptake.<sup>2</sup> Based upon this interesting finding, it became attractive to investigate the activity of other 6- and 7-substituted cocaine analogues. In particular, we chose to explore these modifications in the higher affinity WIN series, in which the 3-position of

<sup>\*</sup> To whom correspondence should be addressed. Phone: 202-687-0686. Fax: 202-687-5065. E-mail: kozikowa@giccs.georgetown.edu. <sup>†</sup> Georgetown University Medical Center.

<sup>&</sup>lt;sup>‡</sup> University of Texas Medical Branch.

<sup>§</sup> Naval Research Laboratory.

Scheme 1<sup>a</sup>



<sup>a</sup> Reagents and conditions: (a) 3 N HCl, rt, 12 h; (b) neutralization with 6 N NaOH; (c) NaOAc,  $CH_3NH_2$ ·HCl, rt, 2 d; (d) TBDMSCl, imidazole, DMF, rt, 12 h; (e) LDA, NC–COOMe, THF, –78 °C, 1 h.

the tropane ring bears phenyl in place of cocaine's benzoate group. While we have reported to date on the synthesis of a variety of such 6- and 7-substituted tropanes through use of an oxidopyridinium-based cycloaddition strategy, we found this method to be cumbersome in achieving our present objective, which was the construction of all possible 6- and 7-substituted tropanes of both chair and boat conformation.<sup>3</sup> Herein, we report on the chemical synthesis and pharmacological characterization of a series of hydoxylated WIN derivatives, and in one case we also present data showing the ability of one of these derivatives to antagonize cocaine's stimulant effects in a rat locomotor assay. As will become apparent, the present work provides further support for our hypothesis that drugs that lack the ability to inhibit transport by all three monoaminergic transporters may exhibit only "partial" cocaine-like properties.<sup>4</sup> That is, drugs that are selective for SERT and NET, or DAT and NET, or SERT and DAT may have properties that could act either alone or in combination with an "antagonist" to reduce cocaine selfadministration.

**Chemistry.** In terms of developing a chemical approach to the 6- and 7-substituted WIN analogues, we initially attempted to use cocaine itself, investigating the possibility to bring about the remote functionalization of this molecule by tethering appropriate functionality to the nitrogen atom. Unfortunately, such efforts did not meet with success. We then modifed our approach by adapting the route devised by Willstätter nearly eight decades ago for the synthesis of cocaine itself.<sup>5</sup> The chemical route used to generate the required 6- and 7-hydroxylated tropinones is shown in Scheme 1, and it resembles closely the approach we first used to prepare the methoxylated pseudococaine derivative metioned in the Introduction.<sup>6</sup>

Accordingly, commercially available 2,5-dimethoxy-2,5-dihydrofuran (1) was stirred in 3 N HCl overnight at room temperature and then neutralized by the addition of 6 N NaOH. This mixture was then added to a solution of acetonedicarboxylic acid, methylamine hydrochloride, and sodium acetate in water (pH approximately 4.3). The resulting mixture was then stirred for 2 days at room temperature. The crude product thus obtained was recrystallized from 2-propanol to afford  $6\beta$ -hydroxytropinone (5) as a white solid in 42% yield.

The hydroxytropinone **5** was protected in turn as its tert-butyldimethylsilyl ether 6 using tert-butyldimethylsilyl chloride and imidazole in DMF at room temperature for 12 h. This intermediate then was deprotonated with lithium diisopropylamine (LDA) in the same manner as described by Majewski for tropinone.<sup>7</sup> Ketone 6 was treated with LDA in THF at -78 °C, and the resulting enolate was reacted with methyl cyanoformate to afford the corresponding racemic carbomethoxylated derivatives 7 and 8 in 64% yield and in a ratio of 9:7. The two isomers can be separated by careful flash column chromatography, and the structure of compound 7, which exists as its enol after removal of the TBDMS group, was assigned by X-ray analysis.<sup>8</sup> The keto esters 7 and 8 served as key intermediates in the synthesis of 6- and 7-hydroxylated WIN type analogues of cocaine.

From this point, a Suzuki coupling protocol was used to install the tolyl group following a protocol much like that first reported by Carroll in his synthesis of other cocaine analogues.<sup>9</sup> Hence, the enol triflate **9** was readily obtained in 73% yield by the addition of *N*phenyltrifluoromethanesulfonimide to a THF solution of ketone **7** containing sodium bis(trimethylsilyl)amide at -78 °C (Scheme 2). Reaction of **9** with 4-methylphenylboronic acid, which was prepared from triisopropyl borate,<sup>10</sup> in 1,2-dimethoxyethane using tris(dibenzylideneacetone)dipalladium(0) as catalyst together with sodium carbonate and lithium chloride at reflux gave **10** in 98% yield.

Reduction of 10 with samarium iodide in THF at -78°C using methanol as the proton source followed by quenching with acetic acid provided a mixture of the desired tropanes 11a (22.5%), 12a (32.5%), 13a (17.5%), and 14a (12.5%) in a total yield of 85%. These products are readily separable by flash column chromatography. On the other hand, hydrogenation of 10 over 10% Pd/C in methanol gives rise to a single product in 80% yield to which we assign  $2\alpha$ ,  $3\alpha$ -stereochemistry (compound **14a**) in accordance with literature precedent.<sup>11</sup> Under basic conditions (NaOMe, MeOH, reflux, 24 h; 83%), 14a is converted to the more stable  $2\beta$ ,  $3\alpha$ -isomer **12a**. Compounds 11a and 13a are therefore the other two isomers. Since **11a** epimerizes to **13a** with NaOMe in methanol, **13a** must be the  $2\alpha$ ,  $3\beta$ -isomer, and **11a** the  $2\beta$ ,  $3\beta$ -isomer.

Finally, the TBDMS group of each of the compounds **11a**, **12a**, and **13a** was removed in high yield with *n*-Bu<sub>4</sub>NF in THF at room temperature to give the corresponding products **11b**–**13b** (87, 99, and 85%, respectively). However, when compound **14a** was treated with *n*-Bu<sub>4</sub>NF in THF, only **12b** was obtained as the product. To avoid this epimerization, which was obviously a consequence of the basicity of the fluoride anion, compound **14a** was instead deprotected with 48% HF to afford  $7\beta$ -hydroxy- $3\alpha$ -(*p*-tolyl)tropane- $2\alpha$ -carboxylic acid methyl ester (**14b**) in 79% yield. The structure of **12b** was confirmed by X-ray analysis (Figure 1).

To access the  $7\alpha$ -epimers, the TBDMS group of compound **10** was removed with *n*-Bu<sub>4</sub>NF in THF at room temperature. The resulting alcohol **15** (91% yield) was subjected to a Swern oxidation,<sup>12</sup> which delivered the ketone **16** in 97% yield. Reduction of **16** with NaBH<sub>4</sub> in methanol gave a 15:1 mixture of the alcohols **17** and **15** in 91% yield. These two compounds were readily

#### Scheme 2<sup>a</sup>



<sup>*a*</sup> Reagents and conditions: (a) NaN(TMS)<sub>2</sub>, PhNTf<sub>2</sub>, THF, -78 °C to rt, overnight; (b) *p*-tolylboronic acid, Pd<sub>2</sub>dba<sub>3</sub>, Na<sub>2</sub>CO<sub>3</sub>, LiCl, DME, 1 h; (c) SmI<sub>2</sub>, MeOH, THF, -78 °C, 1 h; (d) H<sub>2</sub> (40 psi), 10% Pd/C, MeOH, 12 h; (e) *n*-Bu<sub>4</sub>NF, THF, rt, 12 h; (f) 48% HF, CH<sub>3</sub>CN, rt, overnight.







Figure 1. ORTEP drawings of compounds 17, 19a, 12b, and 24a.

separated by flash chromatography, and the structure of **17** was established by X-ray analysis (Figure 1). Attempts at saturating the double bond in the free alcohol **17** with SmI<sub>2</sub> as described above were unsuccessful. Therefore, compound **17** was converted to its *tert*-butyldimethylsilyl ether **18**, and this intermediate was reacted with SmI<sub>2</sub> in THF/MeOH. Of the reduction products, only the major isomer **19a** was obtained in pure form in 66% yield while the minor stereoisomers formed an inseparable mixture. Upon treatment with NaOMe in MeOH at reflux, approximately 75% of this mixture was converted to compound **19a**. The remaining material exhibited a complex <sup>1</sup>H NMR spectrum and was not further investigated. If the mixture of minor products was deprotected with *n*-Bu<sub>4</sub>NF in THF, most of it was transformed into the lactone **20b**. Hydrogenation (40 psi) of **18** over 10% Pd/C in methanol gave the same lactone directly. We conclude that the predominant isomer among the minor products has  $2\alpha$ , $3\alpha$ configuration and that compound **19a** has  $2\beta$ , $3\alpha$ configuration. The structure of **19a** was subsequently confirmed by X-ray crystallographic analysis (Figure 1). Deprotection of **19a** with *n*-Bu<sub>4</sub>NF in THF provided the desired compound **19b** (Scheme 3).

In the same manner as described above, compound **8** was converted into its enol triflate **21** in 62% yield, which was coupled with *p*-tolylboronic acid to produce the aryl olefin **22** in high yield (Scheme 4). Reduction

Scheme 3<sup>a</sup>



<sup>*a*</sup> Reagents and conditions: (a) *n*-Bu<sub>4</sub>NF, THF, rt, 12 h; (b) oxalyl chloride, DMSO, -78 °C, 30 min; (c) MeOH, NaBH<sub>4</sub>, 0 °C–rt, overnight; (d) TBDMSCl, imidazole, DMF, rt, 12 h; (e) SmI<sub>2</sub>, MeOH, THF, -78 °C, 1 h; (f) H<sub>2</sub> (30 psi), 10% Pd/C, MeOH, 12 h; (g) *n*-Bu<sub>4</sub>NF, THF, rt, 12 h.

#### Scheme 4<sup>a</sup>



<sup>*a*</sup> Reagents and conditions: (a) NaN(TMS)<sub>2</sub>, PhNTf<sub>2</sub>, THF, -78 °C to rt, overnight; (b) *p*-tolylboronic acid, Pd<sub>2</sub>dba<sub>3</sub>, Na<sub>2</sub>CO<sub>3</sub>, LiCl, DME, 1 h; (c) SmI<sub>2</sub>, MeOH, THF, -78 °C, 1 h; (d) H<sub>2</sub> (30 psi), 10% Pd/C, MeOH, 12 h; (e) *n*-Bu<sub>4</sub>NF, THF, rt, 12 h; (f) 48% HF, CH<sub>3</sub>CN, rt, overnight.

of **22** with SmI<sub>2</sub> at -78 °C in THF/MeOH gave the saturated tropane analogues **23a** (38%), **24a** (48%), and **25a** (5%). Intermediate **23a** was converted to **25a** in high yield by treatment with NaOMe in MeOH at reflux. Hydrogenation of **22** over 10% Pd/C, on the other hand, provided the 2 $\alpha$ ,3 $\alpha$ -isomer **26a** exclusively. The structure of compound **24a** was confirmed by X-ray analysis (Figure 1).<sup>13</sup> Finally, the TBDMS group of each of the compounds **23a**, **24a**, and **25a** was removed in high yield with *n*-Bu<sub>4</sub>NF in THF to give the desired products **23b**, **24b**, and **25b**, respectively. The remaining isomer **26b** was obtained in 92% yield by treatment of **26a** with 48% HF.

Compounds of the  $6\alpha$ -series were prepared, like their  $7\alpha$ -isomers, by stereochemical inversion at the oxygenbearing carbon atom. Thus, compound **22** was deprotected with *n*-Bu<sub>4</sub>NF and the resulting alcohol **27** converted by Swern oxidation to ketone **28**, which upon NaBH<sub>4</sub> reduction afforded **27** and its  $6\alpha$ -isomer **29** in a ratio of 1:5.3. After hydroxyl protection as the silyl ether

**30**, the double bond was reduced with SmI<sub>2</sub> in THF/ MeOH to afford the saturated products **31a** (6%), **32a** (69%), and **33a** (17%). Hydrogenation of **30** gave the  $2\alpha$ , $3\alpha$ -isomer **34a** in 89% yield. Finally, all of these four intermediates were deprotected with *n*-Bu<sub>4</sub>NF or 48% HF to give the desired products **31b**-**34b** (Scheme 5).

**Biology.** All of the hydroxylated tropanes described herein were tested for their ability to displace [<sup>3</sup>H]mazindol binding. Mazindol has been shown to label the cocaine binding site on the dopamine transporter of rat striatal membranes.<sup>14</sup> This ligand binds with high affinity ( $K_d = 8.63 \pm 0.53$  nM) to a single, sodiumdependent site in striatal membranes, representing the dopamine carrier. Additionally, these compounds were tested for their ability to inhibit high-affinity uptake of DA, 5-HT, and NE into nerve endings (synaptosomes) which were prepared from various regions of the rat brain that are enriched with the particular transporter under study.<sup>15</sup> Comparison of the  $K_i$  values for inhibition of monoamine uptake and mazindol binding was facili-

#### Scheme 5<sup>a</sup>



<sup>a</sup> Reagents and conditions: (a) *n*-Bu<sub>4</sub>NF, THF, rt, 12 h; (b) oxalyl chloride, DMSO, -78 °C, 30 min; (c) NaBH<sub>4</sub>, MeOH 0 °C-rt, overnight; (d) TBDMSCl, imidazole, DMF, rt, 12 h; (e) SmI<sub>2</sub>, MeOH, THF, -78 °C, 1 h; (f) H<sub>2</sub> (30 psi), 10% Pd/C, MeOH, 12 h; (g) *n*-Bu<sub>4</sub>NF, THF, rt, 12 h; (h) 48% HF, CH<sub>3</sub>CN, rt, overnight.

**Table 1.**  $K_i$  Values for 6- and 7-Hydroxylated Tropane Analogues in the Inhibition of [<sup>3</sup>H]Mazindol Binding and in the Inhibition of Radiolabeled DA, 5-HT, and NE Uptake<sup>*a*</sup>

compound	[ <sup>3</sup> H]mazindol binding	[ <sup>3</sup> H]dopamine uptake	[ <sup>3</sup> H]5-HT uptake	[ <sup>3</sup> H]NE uptake
cocaine	$230\pm16~\mathrm{nM}$	$423\pm147~nM$	$155\pm0.4~\mathrm{nM}$	$108\pm4\;nM$
WIN 35065-2 <sup>b</sup>	89.4 nM	$49.8\pm2.2~\mathrm{nM}$	$173\pm13~\mathrm{nM}$	$37.2\pm5.2~\mathrm{nM}$
11b	$25.2\pm2.5~\mathrm{nM}$	$39.2\pm2.6~\mathrm{nM}$	$581\pm57~\mathrm{nM}$	$44.7\pm9.5~nM$
12b	$135\pm15~\mathrm{nM}$	$156\pm25~\mathrm{nM}$	$5.9\pm0.1\mu\mathrm{M}$	$60.9\pm3.5~\mathrm{nM}$
13b	$2.9\pm0.1\mu\mathrm{M}$	$5.6\pm0.2~\mu\mathrm{M}$	$\approx 3  \mu M$	$\approx 1  \mu M$
14b	$1.4\pm0.3\mu\mathrm{M}$	$2.9\pm0.3\mu\mathrm{M}$	$> 100 \ \mu M$	≈300 nM
19b	$540\pm53~\mathrm{nM}$	$676 \pm 43 \text{ nM}$	NT	NT
20b	$14.5\pm0.2~\mu\mathrm{M}$	$25\pm1\mu\mathrm{M}$	NT	NT
23b	$126\pm18~nM$	$292\pm80~\mathrm{nM}$	$6.7\pm0.4\mu\mathrm{M}$	$154\pm23~\mathrm{nM}$
24b	$659\pm39~\mathrm{nM}$	$996 \pm 11 \text{ nM}$	NT	NT
25b	$13.2\pm3.0\mu\mathrm{M}$	$3.1\pm0.3~\mu\mathrm{M}$	NT	NT
26b	$5.1 \pm 0.1 \mu \mathrm{M}$	$1.9\pm0.3~\mu\mathrm{M}$	NT	NT
31b	$1.8\pm0.1\mu\mathrm{M}$	$2.3\pm0.1~\mu{ m M}$	NT	NT
32b	$13.2\pm1.0\mu\mathrm{M}$	$11.7\pm1.7~\mu\mathrm{M}$	NT	NT
33b	>100 µM	$97.2\pm5.6\mu\mathrm{M}$	NT	NT
34b	$54\pm3\mu{ m M}$	$64\pm2\mu\mathrm{M}$	NT	NT
<b>35</b> <sup>c</sup>	$21.0\pm2.3~nM$	$51.8\pm2.1~nM$	$80.2\pm1.1~nM$	$6.3\pm1.6~nM$

<sup>*a*</sup>  $K_i$  values are mean ± SE from 2 to 4 independent experiments, each consisting of six drug concentrations (in triplicate) that were selected on the basis of preliminary screening experiments to bracket the approximate IC<sub>50</sub> value. <sup>*b*</sup> WIN 35065-2 is 2β-carbomethoxy-3β-phenyltropane. <sup>*c*</sup> Compound **35** is (1*R*,5*S*)-2α-*n*-propyl-3β-*p*-tolyltropane.

tated by the use of the same buffer in all assays. The binding and uptake data are provided in Table 1 along with comparison data for cocaine.

As is apparent from the accompanying table, of the 7-hydroxylated tropanes, only 11b and 12b, the former with the chair conformation and the latter with the boat conformation, were found to have nanomolar range potencies in binding and uptake. The chair tropane 11b is particularly potent, with a  $K_i$  of 25 nM in mazindol binding. The DAT and NET potencies are comparable (39 and 45 nM) and about 10-fold better than the SERT potency. On the other hand, the NET potency of **12b** is 2-fold better than its DAT potency, while its SERT potency is in the 6  $\mu$ M range. Both of the 7 $\beta$ -hydroxylated analogues **13b** and **14b**, which have an  $\alpha$ -oriented ester group, exhibit only micromolar potencies at all three transporters. The  $7\alpha$ -hydroxylated boat tropane **19b** is less active than cocaine in the DAT assay, with a  $K_i$  of about 0.7  $\mu$ M, with very little activity being shown by the lactone 20b. During the course of our work, Meltzer has also reported on the activity of some 7-hydroxylated WIN analogues, finding that a  $7\beta$ - hydroxy-3 $\alpha$ -(3,4-dichlorophenyl) substituted tropane has both high affinity and selectivity for DAT over the SERT (IC<sub>50</sub> = 1.2 nM vs 1390 nM). The chemical route used by Meltzer also makes use of the Willstätter approach first employed by us in combination with the Suzuki coupling chemistry.<sup>16</sup>

Of the 6-hydroxylated compounds, only the tropane **23b** of chair conformation had activity in the nanomolar range (290 nM) at the DAT, while all other 6-hydroxylated derivatives exhibited binding and uptake data in the micromolar range. The NET activity of **23b** is also good, while this compound shows poor activity at the SERT. Compound **23b** is thus DAT + NET selective.

From the present study one may make the observation that the mazindol and DA recognition sites are more tolerable for the 6- and 7-hydroxylated tropanes that possess a  $\beta$ -oriented hydroxyl group. From a structural standpoint, this is reasonable, as the nitrogen atom and *N*-methyl group that make up the tropane ring already provide some steric bulk to this face of the seven-membered carbocycle, and the presence of the extra hydroxy group causes little further "steric dis-



Figure 2. Time course of effects of 12b on horizontal activity counts/10 min.

turbance". Also, for the three ligands showing 300 nM or better DAT potency, their SERT potency is 15- to 50-fold less. Thus, the 6- or 7-hydroxy group causes a reduction in activity at the SERT in comparison to the activity found for cocaine itself or for a 6/7-unsubsituted WIN analogue. As can be seen from Table 1, cocaine shows better activity at the SERT than the DAT. Data are also provided for the WIN analogue **35** bearing an  $\alpha$ -oriented *n*-propyl group at the 2-position. The DAT and SERT potencies are comparable in this case, whereas the compound is more active at the NET.

Unfortunately, based upon a comparison of binding and uptake data, it appeared that none of our hydroxylated WIN type compounds might act as a cocaine antagonist by binding to the mazindol site more avidly than the DA site on DAT. Even when tested as such, in contrast to what we found previously for the  $7\alpha$ methoxylated pseudococaine analogue which possesses low binding affinity, no functional antagonism was observed.<sup>2a</sup> Nonetheless, from our prior experience with a DAT/NET-selective piperidine-based ligand that shows no functional antagonism in vitro, but yet is able to antagonize the effects of cocaine in vivo in locomotor studies, we chose to scale-up two of the more active of these compounds, 11b and 12b, to allow for in vivo studies.<sup>2b</sup> Compound **12b** was chosen based both upon its DAT potency and its selectivity for the DAT over the SERT.

Locomotor Study of 11b and 12b. Tropanes 11b and 12b were tested in nonhabituated male Swiss-Webster mice in the locomotor assay using the standard operating procedure developed by the Medications Development Division of the National Institute on Drug Abuse. Not unexpectantly, the more DAT-potent ligand 11b had an efficacy similar to cocaine when it was tested alone at a dose range of 3-100 mg/kg, and it was found to stimulate locomotor activity with an ED<sub>50</sub> value of 14.03 mg/kg and a maximal effect/cocaine miximal effect ratio of 0.88. As a consequence of this activity profile, **11b** was not tested for its ability to attenuate cocaineinduced locomotor activity. On the other hand, the less DAT-active ligand **12b** proved to be more interesting. Figure 2 shows the average horizontal activity counts/ 10 min as a function of time, immediately following injection of **12b** or vehicle. Since **12b** did not affect spontaneous locomotor activity in the dose range finding experiment, a cocaine interaction study was conducted next. Figure 3 shows the average horizontal activity counts/10 min for the different treatment groups as a function of time, starting 20 min after pretreatment with **12b**. The period of 0-30 min was selected for



Figure 3. Interaction between 12b and cocaine on LMA: time course of effect.



**Figure 4.** Interaction between **12b** and cocaine on LMA: dose response during first 30 min after a 20 min pretreatment.

analysis of the dose-response data because this is the time period in which cocaine produced its maximal effects. Figure 4 shows the average horizontal activity counts/10 min over 30 min as a function of the dose condition, 20 min after pretreatment with **12b**. The bar above "vehicle" represents the effect of vehicle 20 min prior to saline injection. The bar above "coc" represents the effect of vehicle 20 min prior to injection of 20 mg/ kg cocaine. The bars above "3, 10, 30, and 100" represent the effects of **12b** at the designated doses 20 min prior to injection of 20 mg/kg cocaine. The mean average horizontal activity counts/10 min on the descending portion of the dose-effect curve (30 to 100 mg/kg dose range) for this 30 min period were fit to a linear function of log dose, and the AD<sub>50</sub> (dose attenuating cocaineinduced stimulation by 50%) was calculated to be 94 mg/ kg. [The ordinate value for the  $AD_{50}$  was calculated using the mean of the vehicle + 0.9% saline (vehicle) group as the minimum value, and the mean of the vehicle plus 20 mg/kg cocaine group as the maximum value.]

A one-way analysis of variance conducted on log horizontal activity counts for the 0–30 min time period indicated a significant overall effect of the treatment group F(5,42) = 7.90, p < 0.05; planned comparisons (a priori contrast) against the cocaine group showed significant differences for vehicle and the 100 mg/kg dose (all p values < 0.05 denoted in Figure 3 with an

asterisk). However, as the difference between the "no effect" dose of **12b** at 100 mg/kg alone and the antagonism result is only a difference of outcome at the 100 mg/kg dose, it is important that these studies be extended further. Nonetheless, the present results do suggest that these hydroxylated tropanes are worthy of further exploration.

## Conclusions

The present work details the synthesis of a series of the 6- and 7-hydroxlated WIN analogues possessing a boat or chair conformation of the tropane ring. Some of these compounds retain considerable DAT and NET potency, with diminished activity at the SERT. While previously we had found that a methoxylated pseudococaine analogue was able to antagonize cocaine's inhibition of dopamine reuptake, none of the compounds of the present series revealed any substantial difference in binding versus uptake, and they failed to show "cocaine antagonism" when tested for such. However, one of the hydroxylated WIN analogues, namely 12b, was tested in vivo and found capable of attenuating cocaine's locomotor activity. When the compound was tested alone at a dose range of 3-100 mg/kg, the compound was found to have no effect on locomotor activity. When tested in animals that had been treated with 20 mg/kg (i.p.), 12b was able to attenuate locomotor activity with an AD<sub>50</sub> of 94 mg/kg. In contrast, locomotor studies using the more DAT-active ligand 11b revealed this compound to be cocaine-like in its action (maximal effect/cocaine miximal effect ratio of 0.88).

The present finding suggests that it may prove valuable to examine the behavioral activity of other 6and 7-substituted tropanes in animal behavioral paradigms in the search for a cocaine medication. Of interest in this connection is our related finding concerning the action of  $7\alpha$ -fluoro- $3\alpha$ -(*p*-fluorophenyl)- $2\beta$ -propyltropane. This compound was evaluated in the brain stimulation reward paradigm, a behavioral screen that allows measuring the intrinsic hedonic properties of compounds separately from the concomitant changes in the rate of responding.<sup>17</sup> The fluorinated tropane was found to be unrewarding when administered alone, with only a slight increase in reward at 20 mg/kg (i.p.), the highest dose tested to date. However, when 5 mg/kg or 10 mg/kg i.p. of this tropane was combined with 10 mg/ kg cocaine (i.p.), a reduction in cocaine-induced rewardenhancement (by approximately 60%) was noted. No changes in motor/performance capacity were observed, either when the derivative was administered alone or in combination with cocaine. Like 12b, this fluorinated tropane also shows better activity at the DAT and NET versus the SERT (49, 52, and 560, respectively).<sup>18</sup>

Thus, taken together this work provides further support for our hypothesis that drugs that lack the ability to inhibit transport by all three monoaminergic transporters may exhibit only "partial" cocaine-like properties. It is our intention to further explore this hypothesis using other tropanes showing varying levels of transporter selectivity.

Last, we call attention to the fact that the hydroxlated tropanes reported herein have been made in racemic form. Based upon the encouraging behavioral results, it is our plan to prepare the more potent compounds in optically pure form using enzymatic methods (PLE) as previously described.  $^{19}\,$ 

### **Experimental Section**

Chemical Methods. General. Diethyl ether was distilled from phosphorus pentoxide. THF was freshly distilled under N<sub>2</sub> from sodium benzophenone. <sup>1</sup>H and <sup>13</sup>C NMR spectra were obtained in CDCl3 with a Varian Unity Inova instrument at nominal frequencies of 300 and 75 MHz, respectively. <sup>1</sup>H chemical shifts ( $\delta$ ) are reported in ppm downfield from internal TMS. <sup>13</sup>C chemical shifts are referenced to CDCl<sub>3</sub> (central peak,  $\delta$  = 77.0 ppm). Melting points were determined in Pyrex capillaries with a Thomas-Hoover Unimelt apparatus and are uncorrected. Mass spectra were measured in the EI mode at an ionization potential of 70 eV. X-ray data were collected on a computer controlled Bruker P4 automatic 4-circle diffractometer. The structures were solved by direct methods and refined, using all independent data, with full matrix leastsquares on F2 values using the SHELXTL program package. TLC was performed on Merck silica gel 60F<sub>254</sub> glass plates. Column chromatography (CC) was performed using Merck silica gel (60-200 mesh).

(±)-**6**β-**Hydroxytropan-3-one** (5). A solution of 2,5dimethoxy-2,5-dihydrofuran (2.6 g, 20 mmol) in 3 N HCl (36 mL) was stirred at room temperature for 16 h, neutralized with 6 N NaOH (about 18 mL) and added to a solution of sodium acetate (13.6 g, 168 mmol), methylamine hydrochloride (2.72 g, 40 mmol), and acetonedicarboxylic acid (5.84 g, 40 mmol) in water (320 mL). The mixture was left standing for 2 d, during which time its pH increased from 3.0 to 4.9. K<sub>2</sub>CO<sub>3</sub> and NaCl (100 g each) were added, and the solution was extracted with chloroform (4 × 150 mL). The combined chloroform extracts were dried over MgSO<sub>4</sub> and concentrated. The residue was crystallized from 2-propanol to give the product (1.30 g, 42%) as a white solid: mp 107–108 °C (lit. mp.<sup>5f</sup> 108–109 °C). <sup>1</sup>H NMR  $\delta$  4.14 (dd, 1H), 3.61 (m, 1H), 3.38 (d, 1H), 2.78–2.55 (m, 2H), 2.67 (s, 3H), 2.45 (br s, 1H), 2.28–1.92 (m, 4H).

(±)-6β-(*tert*-Butyldimethylsilyloxy)tropan-3-one (6). A solution of 5 (132 mg, 0.85 mmol), imidazole (68 mg, 1.0 mmol), and *tert*-butyldimethylsilyl chloride (150 mg, 1.0 mmol) in DMF (2 mL) was stirred under N<sub>2</sub> at room temperature for 32 h. The mixture was diluted with ether, and the solution was washed with saturated NH<sub>4</sub>Cl and brine, dried over MgSO<sub>4</sub>, and concentrated. CC of the residue (Et<sub>2</sub>O/Et<sub>3</sub>N 99:1–19:1) gave compound **6** (218 mg, 95%) as a clear oil: <sup>1</sup>H NMR δ 4.11 (dd, J = 3.6, 6.3 Hz, 1H), 3.59 (m, 1H), 3.25 (m, 1H), 2.67 (s, 3H), 2.72–2.60 (m, 2H), 2.30–2.00 (m, 4H), 0.88 (s, 9H), 0.23 (s, 6H).

 $(\pm)$ -7 $\beta$ -(*tert*-Butyldimethylsilyloxy)-2 $\beta$ -(methoxycarbonyl)tropan-3-one (7) and  $(\pm)$ -6 $\beta$ -(*tert*-Butyldimethylsilyloxy)-2β-(methoxycarbonyl)tropan-3-one (8). A solution of diisopropylamine (155  $\mu$ L, 1.1 mmol) in THF (5 mL) was cooled to 0 °C, and *n*-butyllithium (0.48 mL, 2.3 M solution in hexane, 1.1 mmol) was added. The solution was stirred for 25 min at 0 °C and then cooled to -78 °C. Ketone 6 (269 mg, 1.0 mmol) dissolved in THF (2 mL) was added dropwise over 15 min. The resulting mixture was stirred for 45 min at -78°C, and then methyl cyanoformate (0.1 mL, 1.3 mmol) was added dropwise at -78 °C. Stirring was continued at -78 °C for 20 min. Glacial acetic acid (1.0 mL) and then silver acetate (168 mg, 1.0 mmol) in glacial acid (1.0 mL) were added at -78°C. The mixture was warmed to room temperature and stirred for another 30 min, then concentrated NH<sub>4</sub>OH was added until the mixture became clear. The resulting solution was extracted with CHCl<sub>3</sub> (5  $\times$  30 mL). The combined organic extracts were washed with water and brine and dried over MgSO4. The solvent was evaporated, and the residue was subjected to CC (CH<sub>2</sub>Cl<sub>2</sub> saturated with concentrated NH<sub>4</sub>OH) to give the  $\beta$ -ketoesters 7 (117 mg, 36%) and 8<sup>20</sup> (93 mg, 28%). Compound 7: <sup>1</sup>H NMR  $\delta$  4.13 (dd, J = 1.5, 6.6 Hz, 1H), 3.78 (s, 3H), 3.67 (br s, 1H), 3.45 (m, 1H), 2.72-2.60 (m, 2H), 2.43 (s, 3H), 2.14-2.04 (m, 1H), 1.97 (dd, J = 6.9, 13.8 Hz, 1H), 1.74 (d, J = 18.6Hz, 1H), 0.89 (s, 9H), 0.09 (s, 3H), 0.08 (s, 3H). Compound 8:1H NMR  $\delta$  4.14 (dd, J = 4.2, 7.5 Hz, 1H), 3.85 (d, J = 5.7 Hz, 1H), 3.76 (s, 3H), 3.17 (d, J = 5.4 Hz, 1H), 2.76–2.60 (m, 2H), 2.42 (s, 3H), 2.45–2.20 (m, 1H), 2.12–1.96 (m, 1H), 1.91 (d, J = 20.7 Hz, 1H), 0.88 (s, 9H), 0.05 (s, 6H).

(±)-Methyl 7 $\beta$ -(tert-Butyldimethylsilyloxy)-8-methyl-3-(trifluoromethanesulfonyloxy)-8-azabicyclo[3.2.1]oct-2-ene-2-carboxylate (9). Sodium bis(trimethylsilyl)amide (1.0 M in THF, 1.7 mL) was added dropwise to a solution of 7 (462 mg, 1.41 mmol) in THF (5 mL) at -78 °C under N<sub>2</sub>. After 30 min of stirring, N-phenyltrifluoromethanesulfonimide (555 mg, 1.55 mmol) was added in one portion at -78 °C. The mixture was allowed to warm to room temperature and was then stirred overnight. The solvent was removed, the residue was dissolved in  $\breve{CH}_2Cl_2$  , and the solution was washed with water and brine and dried over MgSO<sub>4</sub>. The organic phase was concentrated, and the residue was purified by CC (hexane/ EtOAc 8:1-4:1) to afford the product (471 mg, 73%) as an oil: <sup>1</sup>H NMR  $\delta$  4.28 (dd, J = 2.1, 6.9 Hz, 1H), 3.83 (s, 3H), 3.53 (m, 1H), 2.78 (dd, J = 4.2, 18.6 Hz, 1H), 2.20-1.91 (m, 1H), 1.85 (d, J = 18.6 Hz, 1H), 0.89 (s, 9H), 0.10 (s, 6H).

(±)-Methyl 7 $\beta$ -(*tert*-Butyldimethylsilyloxy)-8-methyl-3-(p-tolyl)-8-azabicyclo[3.2.1]oct-2-ene-2-carboxylate (10). The triflate 9 (470 mg, 1.02 mmol), p-tolylboronic acid (184 mg, 1.35 mmol), LiCl (95 mg, 2.05 mmol), tris(dibenzylideneacetone)dipalladium(0) (41 mg, 45  $\mu$ mol), and Na<sub>2</sub>CO<sub>3</sub> (0.99 mL, 2.0 M aqueous solution) were combined in 1,2-dimethoxyethane (5 mL) under Ar and heated at reflux for 1 h. The mixture was cooled to room temperature, filtered through Celite, and washed with ether. The mixture was basified with concentrated NH<sub>4</sub>OH and washed with brine. The organic layer was dried (MgSO<sub>4</sub>) and concentrated to dryness. The residue was purified by CC (hexane/EtOAc 2:1) to afford 10 (405 mg, 98%) as a yellowish solid: mp 45–46 °C; <sup>1</sup>H NMR  $\delta$ 7.13 (d, J = 7.8 Hz, 2H), 7.02 (d, J = 7.8 Hz, 2H), 4.36 (dd, J= 1.8, 6.3 Hz, 1H), 3.77 (s, 1H), 3.55 (s, 3H), 3.51 (m, 1H), 2.72 (dd, J = 4.2, 18.9 Hz, 1H), 2.56 (s, 3H), 2.35 (s, 3H), 2.23-2.00 (m, 2H), 1.91 (d, J = 18.9 Hz, 1H), 0.91 (s, 9H), 0.11 (s, 3H), 0.10 (s, 3H); <sup>13</sup>C NMR  $\delta$  -5.1, -4.6, 18.4, 21.2, 26.0 (3C), 34.8, 35.5, 43.5, 51.3, 57.0, 68.3, 78.8, 126.2, 126.5 (2C), 128.7 (2C), 137.3, 137.8, 144.8, 168.3. Anal. (C23H35NO3Si) C, H, N.

(±)-Methyl 7 $\beta$ -(*tert*-Butyldimethylsilyloxy)-3 $\beta$ -(*p*-tolyl)tropane-2 $\beta$ -carboxylate (11a), (±)-Methyl 7 $\beta$ -(*tert*-Butyldimethylsilyloxy)- $3\alpha$ -(p-tolyl)tropane- $2\beta$ -carboxylate (12a), (±)-Methyl 7β-(tert-Butyldimethylsilyloxy)-3β-(p-tolyl)tropane-2 $\alpha$ -carboxylate (13a), and (±)-Methyl 7 $\beta$ -(*tert*-Butyldimethylsilyloxy)-3α-(p-tolyl)tropane-2α-carboxylate (14a). To a solution of the aryl olefin 10 (40 mg, 0.10 mmol) in THF (2 mL) at -78 °C under N<sub>2</sub> was added SmI<sub>2</sub> (4.5 mL, 0.1 M in THF, 0.45 mmol). After stirring for 30 min, MeOH (anhydrous, 0.5 mL) was added. The mixture was stirred at -78 °C for another 2 h, then the reaction was quenched with acetic acid (0.5 mL). The mixture was basified with NH<sub>4</sub>OH and extracted with ether, and the organic phase was dried over MgSO<sub>4</sub> and concentrated. The residue was purified by preparative TLC (hexane/EtOAc 2:1), and the products 11a - 14a were obtained in a total yield of 85%. Compound 11a (9 mg, 22.5%): white solid, mp 108-109 °C; <sup>1</sup>H NMR  $\delta$  7.14 (d, J = 7.8 Hz, 2H), 7.08 (d, J = 7.8 Hz, 2H), 4.42 (dd, J = 3.0, 6.9 Hz, 1H), 3.56 (m, 1H), 3.52 (s, 3H), 3.42 (m, 1H), 3.00 (m, 1H), 2.74-2.62 (m, 1H), 2.53 (s, 3H), 2.52-2.44 (m, 1H), 2.30 (s, 3H), 2.26-2.10 (m, 1H), 2.04 (dd, J = 7.2, 13.8 Hz, 1H), 1.59 (dt, J = 3.6 Hz (t), 12.3 Hz (d), 1H), 0.92 (s, 9H), 0.13 (s, 3H), 0.12 (s, 3H);  $^{13}\mathrm{C}$  NMR  $\delta$  –4.8, –4.7, 17.8, 21.0, 25.7 (3C), 33.0, 34.5, 38.8, 43.0, 50.2, 63.5, 77.6, 127.1 (2C), 128.7 (2C), 135.3, 139.6, 171.6. Anal. (C<sub>23</sub>H<sub>37</sub>NO<sub>3</sub>-Si) C, H, N. Compound 12a (13 mg, 32.5%): yellowish solid, mp 74–75 °C; <sup>1</sup>H NMR  $\delta$  7.08 (s, 4H), 4.31 (dd, J = 3.0, 6.9Hz, 1H), 3.50-3.28 (m, 2H), 3.20 (s, 1H), 2.58 (s, 3H), 2.56-2.49 (m, 1H), 2.45-2.33 (m, 1H), 2.30 (s, 3H), 2.18-2.06 (m, 1H), 1.98 (dd, J = 6.9, 13.8 Hz, 1H), 1.29 (m, 1H), 0.87 (s, 9H), 0.06 (s, 3H), 0.05 (s, 3H);  $^{13}$ C NMR  $\delta$  -4.9, -4.8, 17.9, 21.0, 25.8 (3C), 36.2, 37.9, 41.9, 42.1, 51.8, 52.3, 60.2, 72.2, 81.5, 127.5 (2C), 129.0 (2C), 135.7, 140.7, 174.7. Anal. (C<sub>23</sub>H<sub>37</sub>NO<sub>3</sub>-Si) C, H, N. Compound 13a (7 mg, 17.5%): white solid, mp 90-92 °C; <sup>1</sup>H NMR  $\delta$  7.13 (d, J = 8.1 Hz, 2H), 7.07 (d, J = 8.1 Hz, 2H), 4.57 (dd, J = 2.7, 12.0 Hz, 1H), 3.54 (s, 3H), 3.43 (m, 1H), 3.27 (d, J = 2.4 Hz, 1H), 3.04 (dd, J = 2.7, 12.0 Hz), 2.90 (dt, J = 5.7 Hz (d), 12.3 Hz (t), 1H), 2.65 (s, 3H), 2.30 (s, 3H), 2.22–2.10 (m, 2H), 1.82 (t, J = 12.6 Hz, 1H), 1.56–1.43 (m, 1H), 0.88 (s, 9H), 0.05 (s, 3H), 0.03 (s, 3H); <sup>13</sup>C NMR  $\delta$  –5.0, –4.9, 18.0, 21.0, 25.8 (3C), 36.4, 37.1, 39.6, 40.3, 48.2, 51.5, 61.6, 71.6, 73.5, 127.4 (2C), 129.1 (2C), 136.0, 140.6, 173.3. Compound **14a** (5 mg, 12.5%): clear oil; <sup>1</sup>H NMR  $\delta$  7.08 (d, J = 8.4 Hz, 2H), 7.04 (d, J = 8.4 Hz, 2H), 4.64 (dd, J = 2.7, 6.9 Hz, 1H), 3.63–3.56 (m, 1H), 3.51 (s, 3H), 2.04–1.90 (m, 2H), 1.82 (dd, J = 6.9, 12.3 Hz, 1H), 0.86 (s, 9H), 0.08 (s, 3H), 0.03 (s, 3H); <sup>13</sup>C NMR  $\delta$  –5.0, –4.8, 17.8, 20.8, 25.8 (3C), 35.0, 35.5, 40.3, 41.6, 47.5, 51.2, 60.8, 70.1, 75.5, 127.3 (2C), 128.5 (2C), 135.1, 140.2, 174.0 Anal. (C<sub>23</sub>H<sub>37</sub>NO<sub>3</sub>Si) C, H, N.

**Compound 14a by Hydrogenation of 10.** Compound **10** (10 mg, 25  $\mu$ mol) was dissolved in absolute MeOH (2 mL) and hydrogenated over 10% Pd/C at 40 psi for 8 h. TLC showed almost complete conversion. The catalyst was filtered off over Celite, and the solvent was evaporated. The residue was purified by CC (hexane/EtOAc 2:1–1:1) to give **14a** (8 mg, 80%) as a clear oil exhibiting the same spectroscopic data as above.

**General Procedure for the Deprotection of** *tert***-Butyldimethylsilyl Ethers.** To a solution of the *tert*-butyldimethylsilyl ether in THF was added *n*-Bu<sub>4</sub>NF (1.2 equiv, 1 M in THF) under N<sub>2</sub>. The mixture was stirred at room temperature overnight and then concentrated, and the residue was dissolved in EtOAc followed by washing with water and brine. The organic phase was dried over MgSO<sub>4</sub> and concentrated, and the residue was purified by CC (CH<sub>2</sub>Cl<sub>2</sub> saturated with concentrated NH<sub>4</sub>OH) to give the free alcohol.

(±)-Methyl 7 $\beta$ -Hydroxy-3 $\beta$ -(p-tolyl)tropane-2 $\beta$ -carboxylate (11b). From the silyl ether 11a (21 mg, 52  $\mu$ mol), 11b (13 mg, 87%) was obtained as a white solid: mp 148–149 °C; <sup>1</sup>H NMR  $\delta$  7.13 (d, J = 7.8 Hz, 2H), 7.08 (d, J = 7.8 Hz, 2H), 4.57 (dd, J = 3.9, 6.3 Hz, 1H), 3.64–3.59 (m, 1H), 3.51 (s, 3H), 3.57–3.49 (m, 1H), 3.04 (t, J = 3.9 Hz, 1H), 2.78–2.66 (m, 1H), 2.57 (s, 3H), 2.51 (dd, J = 2.7, 12.6 Hz, 1H), 2.30 (s, 3H), 2.22–2.12 (m, 2H), 1.93 (br s, 1H), 1.60 (dt, J = 3.6 Hz (t), 8.4 Hz (d), 1H); <sup>13</sup>C NMR  $\delta$  21.0, 32.1, 34.5, 38.3, 42.3, 59.8, 51.2, 63.2, 73.2, 127.0 (2C), 128.7 (2C), 135.4, 139.5, 171.6; MS mz 289 (M<sup>+</sup>), 245, 230, 186, 172, 158, 129, 115, 98, 82, 72, 58, 43. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl 7β-Hydroxy-3α-(p-tolyl)tropane-2β-carboxylate (12b). From the silyl ether 12 (18 mg, 45 μmol), 12b (11 mg, 85%) was obtained as a white solid: mp 118–119 °C; <sup>1</sup>H NMR δ 7.08 (s, 4H), 4.28 (dd, J = 3.0, 5.4 Hz, 1H), 3.60 (s, 3H), 3.54–3.35 (m, 2H), 3.27 (s, 1H), 2.70 (s, 3H), 2.58–2.40 (m, 1H), 2.43–2.33 (m, 1H), 2.31 (s, 3H), 2.13–2.04 (m, 2H), 1.36 (dd, J = 9.3, 14.1 Hz, 1H); <sup>13</sup>C NMR δ 21.0, 35.3, 35.6, 39.4, 41.9, 50.7, 51.9, 57.2, 70.3, 79.1, 127.4 (2C), 129.1 (2C), 135.8, 141.3, 175.2; MS *m*/*z* 289 (M<sup>+</sup>), 245, 230, 186, 170, 158, 145, 129, 112, 98, 82, 72, 58, 42. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl 7 $\beta$ -Hydroxy-3 $\beta$ -(p-tolyl)tropane-2 $\alpha$ -carboxylate (13b). From the silyl ether 13 (17 mg, 42  $\mu$ mol), 13b (12 mg, 99%) was obtained as a white solid: mp 187–189 °C; <sup>1</sup>H NMR  $\delta$  7.12 (d, J = 8.4 Hz, 2H), 7.08 (d, J = 8.4 Hz, 2H), 4.55 (dd, J = 3.6, 6.3 Hz, 1H), 3.54 (s, 3H), 3.46–3.40 (m, 1H), 3.31 (d, J = 2.7 Hz, 1H), 3.13 (dd, J = 3.0, 12.0 Hz, 1H), 2.98 (dt, J = 6.0 Hz (d), 12.3 Hz (t), 1H), 2.68 (s, 3H), 2.30 (s, 3H), 2.45–2.20 (m, 2H), 2.16–1.80 (m, 2H), 1.44–1.34 (m, 1H); <sup>13</sup>C NMR  $\delta$  21.0, 32.0, 35.5, 37.1, 40.9, 44.2, 51.7, 59.5, 70.1, 72.4, 127.2 (2C), 129.2 (2C), 136.1, 140.5, 173.6; MS *m*/*z* 289 (M<sup>+</sup>), 245, 230, 186, 172, 158, 145, 129, 115, 98, 72, 57, 42. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl 7β-Hydroxy-3α-(p-tolyl)tropane-2α-carboxylate (14b). A THF solution (1.0 mL) of silyl ether 14a (16 mg, 39 μmol) was added to 48% aqueous HF (1.0 mL). After stirring at room temperature overnight, the mixture was poured into NH<sub>4</sub>OH (6.0 mL) and extracted with CHCl<sub>3</sub>. The combined extracts were dried, and the solvent was removed. Purification of the crude product by CC (CH<sub>2</sub>Cl<sub>2</sub> saturated with concentrated NH<sub>4</sub>OH) gave 14b (9 mg, 79%) as a white solid: mp 103–104 °C; <sup>1</sup>H NMR δ 7.09 (d, J = 8.4 Hz, 2H), 7.04 (d, J=8.4 Hz, 2H), 4.73 (dd,  $J=3.0,\,6.9$  Hz, 1H), 3.72–3.62 (m, 1H), 3.53 (s, 3H), 3.44–3.36 (m, 2H), 3.36–3.29 (m, 1H), 2.62 (s, 3H), 2.52–2.40 (m, 1H), 2.30 (s, 3H), 1.99–1.78 (m, 4H); ^{13}C NMR  $\delta$  20.8, 32.1, 35.4, 37.6, 39.2, 42.0, 51.4, 58.9, 68.8, 74.0, 127.2 (2C), 128.7 (2C), 135.2, 140.8, 174.4; MS m/z 289, 245, 230, 186, 171, 158, 145, 129, 115, 98, 84, 58. Anal. (C $_{17}H_{23}$ -NO<sub>3</sub>) C, H, N.

(±)-Methyl7β-Hydroxy-8-methyl-3-(*p*-tolyl)-8-azabicyclo-[3.2.1]oct-2-ene-2-carboxylate (15). Compound 10 (40 mg, 0.10 mmol) was deprotected according to the general procedure to give product 15 (26 mg, 91%) as a white solid: mp 124–126 °C; <sup>1</sup>H NMR δ 7.13 (d, J = 7.8 Hz, 2H), 7.01 (d, J = 7.8 H, 2H), 4.29 (dd, J = 3.6, 4.2 Hz, 1H), 3.74 (s, 1H), 3.53 (s, 3H), 3.44–3.36 (m, 1H), 2.66 (dd, J = 4.2, 19.2 Hz, 1H), 2.49 (s, 3H), 2.34 (s, 3H), 2.17 (s, 1H), 2.13–2.05 (m, 2H), 1.88 (d, J = 18.9 Hz, 1H); <sup>13</sup>C NMR δ 21.2, 33.4, 33.9, 43.1, 51.4, 55.5, 67.7, 77.7, 125.1, 126.5 (2C), 128.8 (2C), 137.4, 137.6, 144.5, 168.4. Anal. (C<sub>17</sub>H<sub>21</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl 8-Methyl-7-oxo-3-(p-tolyl)-8-azabicyclo[3.2.1]oct-2-ene-2-carboxylate (16). Oxalyl chloride (122 µL, 1.40 mmol) was dissolved in 5 mL of anhydrous methylene chloride, and the solution was cooled to -78 °C. Dimethyl sulfoxide (0.40 mL, 2.8 mmol) was added. After 5 min, compound 15 (200 mg, 0.70 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (2.0 mL) was added, and stirring was continued for another 30 min. The reaction was quenched by adding Et<sub>3</sub>N (2.0 mL), then the solution was warmed to room temperature and washed with water, and the aqueous phase was extracted several times with CH<sub>2</sub>Cl<sub>2</sub>. The combined organic phases were washed with saturated NH<sub>4</sub>Cl, dried over MgSO<sub>4</sub>, and concentrated under reduced pressure. Purification of the crude product by CC (Et<sub>2</sub>O/Et<sub>3</sub>N 99:1-9:1) gave ketone **16** (193 mg, 97%) as a clear oil: <sup>1</sup>H NMR  $\delta$  7.10 (d, J = 7.8Hz, 2H), 6.98 (d, J = 7.8 Hz, 2H), 4.10 (s, 1H), 3.50 (s, 3H), 3.68-3.44 (m, 2H), 2.96-2.69 (m, 2H), 2.41 (s, 3H), 2.30 (s, 3H), 2.18 (d, J = 18.3 Hz, 1H), 2.12 (d, J = 19.2 Hz, 1H); <sup>13</sup>C NMR  $\delta$  21.2, 34.2, 35.7, 42.8, 51.7, 55.0, 65.9, 121.9, 126.5 (2C), 128.8 (2C), 137.3, 137.9, 147.7, 166.6, 207.5; MS m/z 257 (M+-CO), 242, 229, 210, 198, 182, 167, 157, 141, 128, 115, 105, 91, 42. Anal. (C<sub>17</sub>H<sub>19</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl7a-Hydroxy-8-methyl-3-(*p*-tolyl)-8-azabicyclo-[3.2.1]oct-2-ene-2-carboxylate (17). NaBH<sub>4</sub> (47 mg, 1.33 mmol) was added cautiously with stirring to a solution of ketone 16 (190 mg, 0.67 mmol) in MeOH at 0 °C. The mixture was allowed to warm to room temperature overnight and then concentrated. The residue was dissolved in EtOAc (50 mL), and the solution was washed with water and brine. After drying over MgSO<sub>4</sub> and concentration, the residue was purified by CC (CH<sub>2</sub>Cl<sub>2</sub> saturated with concentrated NH<sub>4</sub>OH) to give the 7 $\alpha$ -alcohol 17 (162 mg, 85%) as a white solid together with compound **15** (11 mg, 6%). Compound **17**: mp 113-114 °C; <sup>1</sup>H NMR  $\delta$  7.13 (d, J = 7.8 Hz, 2H), 7.06 (d, J = 7.8 Hz, 2H), 4.65 (m, 1H), 3.88 (d, J = 5.4 Hz, 1H), 3.48 (s, 3H), 3.31–3.22 (m, 1H), 3.02 (br s, 1H), 2.80 (dd, J = 3.9, 18.9 Hz, 1H), 2.72-2.62 (m, 1H), 2.44 (s, 3H), 2.35 (s, 3H), 2.08 (d, J = 19.2 Hz, 1H), 1.40 (dd, J = 4.5, 13.2 Hz, 1H); <sup>13</sup>C NMR (CDCl<sub>3</sub>)  $\delta$  21.1, 34.7, 35.1, 40.4, 51.5, 56.5, 64.1, 77.1, 125.3, 126.6 (2C), 128.7 (2C), 137.4, 137.8, 145.0, 170.3. Anal. (C<sub>17</sub>H<sub>21</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl 7α-(*tert*-Butyldimethylsilyloxy)-8-methyl-3-(*p*-tolyl)-8-azabicyclo[3.2.1]oct-2-ene-2-carboxylate (18). Compound 17 (160 mg, 0.56 mmol) was silylated in the same manner as described for compound 5 and purified by CC (hexane/EtOAc 2:1–1:1) to furnish the product 18 (89%) as a clear oil: <sup>1</sup>H NMR  $\delta$  7.14 (d, J = 7.8 Hz, 2H), 7.06 (d, J = 7.8 Hz, 2H), 4.65–4.66 (m, 1H), 4.07 (d, J = 5.4 Hz, 1H), 3.43 (s, 3H), 3.25 (dd, J = 4.8, 6.9 Hz, 1H), 2.68–2.50 (m, 2H), 2.40 (s, 3H), 2.36 (s, 3H), 2.12 (d, J = 18.9 Hz, 1H), 1.36 (dd, J = 4.8, 12.9 Hz, 1H), 0.83 (s, 3H), 0.04 (s, 3H), 0.01 (s, 3H); <sup>13</sup>C NMR  $\delta$  –5.2, –5.0, 18.0, 21.2, 25.7 (3C), 34.6, 35.7, 41.0, 50.9, 56.6, 63.6, 77.6, 126.2, 126.9 (2C), 128.6 (2C), 137.0, 138.7, 144.4, 168.9.

(±)-Methyl 7 $\alpha$ -(*tert*-Butyldimethylsilyloxy)-3 $\alpha$ -(*p*-tolyl)tropane-2 $\beta$ -carboxylate (19a). The aryl olefin 18 (40 mg, 0.10 mmol) was treated with SmI<sub>2</sub> as described for compound 10. After purification, compound 19a was obtained by CC

(ether/Et<sub>3</sub>N 99:1) as a white solid in a yield of 66%. The mother liquor contained predominantly compound **20a** together with a minor component in a ratio of 3:1 (<sup>1</sup>H NMR). To a solution of the evaporated mother liquor (20 mg) in dry MeOH (2.0 mL) was added under N<sub>2</sub> 30% NaOMe/MeOH (0.1 mL). The mixture was refluxed for 24 h, cooled to room temperature, and concentrated. CH<sub>2</sub>Cl<sub>2</sub> and saturated NH<sub>4</sub>Cl were added, the phases were separated, and the organic layer was washed with brine and concentrated. Purification of the residue by TLC (ether/Et<sub>3</sub>N 99:1) afforded compound 19a (15 mg, 100%): mp 113–115 °C; <sup>1</sup>H NMR  $\delta$  7.16 (d, J = 8.1 Hz, 2H), 7.08 (d, J =8.1 Hz, 2H), 4.63 (m, 1H), 3.55 (s, 3H), 3.57-3.46 (m, 1H), 3.42-3.17 (m, 3H), 2.73-2.60 (m, 1H), 2.56-2.47 (m, 1H), 2.47 (s, 3H), 2.31 (s, 3H), 1.54-1.41 (m, 2H), 0.95 (s, 9H), 0.13 (s, 3H), 0.10 (s, 3H);  ${}^{13}$ C NMR  $\delta$  -5.1, -4.8, 18.0, 21.0, 25.8 (3C), 35.2, 39.1, 40.0, 41.0, 44.1, 51.4, 57.8, 66.1, 71.2, 128.0 (2C), 129.0 (2C), 135.4, 142.7, 176.6. Anal. (C23H37NO3Si) C, H, N.

(±)-Methyl 7α-Hydroxy-3α-(*p*-tolyl)tropane-2β-carboxylate (19b) was obtained (15 mg, 77%) as an oil in a similar manner as for the prepation of **11b** after purification by CC (CH<sub>2</sub>Cl<sub>2</sub> saturated with concentrated NH<sub>4</sub>OH): <sup>1</sup>H NMR δ 7.17 (d, J = 8.1 Hz, 2H), 7.07 (d, J = 8.1 Hz, 2H), 4.80–4.68 (m, 1H), 3.56 (s, 3H), 3.52–3.34 (m, 3H), 3.22 (t, J = 7.8 Hz, 1H), 2.80–2.64 (m, 1H), 2.58–2.47 (m, 1H), 2.45 (s, 3H), 1.82 (br s, 1H), 1.45–1.30 (m, 2H); <sup>13</sup>C NMR δ 21.0, 35.3, 39.3, 39.9, 40.2, 44.2, 51.8, 58.0, 66.0, 70.8, 128.0 (2C), 129.1 (2C), 135.7, 141.9, 176.2. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N.

(±)-9-Methyl-3α-(*p*-tolyl)-6-oxa-9-azatricyclo[5.2.1.0<sup>4,8</sup>]decan-5-one (20b) was obtained from 18 in 78% yield (9 mg) by catalytic hydrogenation using a procedure similar to that for the conversion of 10 to 14, or by SmI<sub>2</sub> reduction of 18 followed by treatment of the mixture of minor product stereoisomers with *n*-Bu<sub>4</sub>NF: <sup>1</sup>H NMR  $\delta$  7.12 (d, J = 8.1 Hz, 2H), 7.08 (d, J = 8.1 Hz, 2H), 5.18 (t, J = 7.2 Hz, 1H), 4.02 (m, 1H), 3.60–3.38 (m, 2H), 3.00 (dd, J = 6.9, 8.7 Hz, 1H), 2.58– 2.30 (m, 2H), 2.36 (s, 3H), 2.32 (s, 3H), 1.79 (d, J = 14.4 Hz, 1H), 1.66 (t, J = 13.8 Hz, 1H), 1.59 (br s, 1H); <sup>13</sup>C NMR  $\delta$  21.1, 33.6, 36.0, 38.0, 41.1, 45.4, 59.6, 66.9, 77.2, 81.7, 128.2 (2C), 128.9 (2C), 136.2, 136.5, 176.4; MS *m*/z 257 (M<sup>+</sup>), 242, 228, 213, 198, 184, 172, 156, 139, 115, 81, 57, 42.

(±)-Methyl 6β-(*tert*-Butyldimethylsilyloxy)-8-methyl-3-(trifluoromethanesulfonyloxy)-8-azabicyclo[3.2.1]oct-2-ene-2-carboxylate (21). Using a procedure similar to that for the preparation of 9, triflate 21 (470 mg, 62%) was obtained after CC (hexane/EtOAc 9:1) as an oil: <sup>1</sup>H NMR δ 4.13 (dd, *J* = 3.9, 7.5 Hz, 1H), 4.01 (d, *J* = 6.0 Hz, 1H), 3.81 (s, 3H), 3.26 (d, *J* = 5.4 Hz, 1H), 2.82 (dd, *J* = 5.4, 18.9 Hz, 1H), 2.50–2.38 (m, 4H), 2.16–2.04 (m, 1H), 1.99 (d, *J* = 18.9 Hz, 1H), 0.88 (s, 9H), 0.07 (s, 6H); <sup>13</sup>C NMR δ –4.9, –4.8, 18.2, 25.8 (3C), 29.3, 34.9, 47.5, 52.2, 60.4, 66.4, 78.1, 118.2 (q, *J* = 318 Hz), 125.8, 147.4, 163.6.

(±)-Methyl 6 $\beta$ -(*tert*-Butyldimethylsilyloxy)-8-methyl-3-(*p*-tolyl)-8-azabicyclo[3.2.1]oct-2-ene-2-carboxylate (22). Using a procedure similar to that for the preparation of 10, aryl olefin 22 (400 mg, 97%) was obtained after CC (silica gel, hexane/EtOAc 2:1) as an oil: <sup>1</sup>H NMR  $\delta$  7.13 (d, J = 7.8 Hz, 2H), 7.03 (d, J = 7.8 Hz, 2H), 4.22 (dd, J = 3.6, 7.2 Hz, 1H), 3.92 (d, J = 6.0 Hz, 1H), 3.51 (s, 3H), 3.19 (d, J = 5.1 Hz, 1H), 2.72 (dd, J = 5.4, 19.5 Hz, 1H), 2.51 (s, 3H), 2.47 (dd, J = 4.5, 12.6 Hz, 1H), 2.35 (s, 3H), 2.16–2.08 (m, 1H), 2.04 (d, J =19.2 Hz, 1H), 0.90 (s, 9H), 0.08 (s, 6H); <sup>13</sup>C NMR  $\delta$  –4.7 (2C), 18.3, 21.2, 25.9 (3C), 33.0, 35.7, 47.2, 51.3, 60.9, 66.6, 79.0, 126.6 (2C), 128.7 (2C), 130.5, 137.4, 137.7, 142.2, 168.5.

(±)-Methyl 6 $\beta$ -(*tert*-Butyldimethylsilyloxy)-3 $\beta$ -(p-tolyl)tropane-2 $\beta$ -carboxylate (23a), (±)-Methyl 6 $\beta$ -(*tert*-Butyldimethylsilyloxy)-3 $\alpha$ -(p-tolyl)tropane-2 $\beta$ -carboxylate (24a), and (±)-Methyl 6 $\beta$ -(*tert*-Butyldimethylsilyloxy)-3 $\beta$ -(ptolyl)tropane-2 $\alpha$ -carboxylate (25a) were obtained from 22 using a procedure similar to that for the preparation of 11a-14a. Compound 23a (38 mg, 38%): mp 68–70 °C; <sup>1</sup>H NMR  $\delta$ 7.13 (d, 8.1 Hz, 2H), 7.08 (d, J = 8.1 Hz, 2H), 4.33 (dd, J =3.0, 6.9 Hz, 1H), 3.77 (d, J = 5.4 Hz, 1H), 3.50 (s, 3H), 3.23 (br s, 1H), 2.86–2.80 (m, 1H), 2.80–2.68 (m, 1H), 2.55 (s, 3H), 2.55–2.44 (m, 1H), 2.30 (s, 3H), 2.25 (dd, J = 3.0, 6.9 Hz, 1H),

2.17 (dd, J = 6.9, 12.6 Hz, 1H), 1.76 (dt, J = 3.9 Hz (t), 12.6 Hz (d), 1H), 0.89 (s, 9H), 0.06 (s, 6H);  $^{13}\mathrm{C}$  NMR  $\delta$  –4.9 (2C), 17.8, 20.9, 25.7 (3C), 31.2, 34.2, 39.2, 43.1, 51.1, 51.8, 66.4, 70.9, 77.4, 126.9 (2C), 128.7 (2C), 135.3, 139.5, 172.0. Anal. (C<sub>23</sub>H<sub>37</sub>NO<sub>3</sub>Si) C, H, N. Compound **24a** (48 mg, 48%): mp 96-97 °C; <sup>1</sup>H NMR  $\delta$  7.08 (s, 4H), 4.21 (dd, J = 2.7, 6.6 Hz, 1H), 3.59 (s, 3H), 3.58-3.48 (m, 1H), 3.35 (dd, J = 9.6, 18.0 Hz, 1H), 3.14 (d, J = 9.3 Hz, 1H), 2.30 (s, 3H), 2.26–2.14 (m, 1H), 2.08 (dd, J = 6.9, 13.8 Hz, 1H), 1.42–1.23 (m, 1H), 0.87 (s, 9H), 0.05 (s, 3H), 0.04 (s, 3H);  $^{13}$ C NMR  $\delta$  –4.9 (2C), 17.8, 20.9, 25.7 (3C), 34.9, 36.4, 42.1, 42.3, 51.7, 55.3, 63.8, 68.3, 81.5, 127.4 (2C), 129.0 (2C), 135.7, 140.6, 175.0. Compound 25a (5.2 mg, 5.2%): mp 195–198 °C; <sup>1</sup>H NMR  $\delta$  7.14 (d, J = 8.1 Hz, 2H), 7.08 (d, J = 8.1 Hz, 2H), 4.35 (dd, J = 2.7, 7.5 Hz, 1H), 3.65-3.56 (m, 1H), 3.51 (s, 3H), 3.09 (m, 1H), 3.04 (dd, J =2.7, 11.7 Hz, 1H), 2.85 (dt, J = 6.3 Hz (d), 12.0 Hz (t), 1H), 2.66 (s, 3H), 2.39 (dd, J = 7.5, 14.1 Hz, 1H), 2.30 (s, 3H), 1.97 (ddd, J = 2.4, 6.9, 13.8 Hz, 1H), 1.79 (dt, J = 3.0 Hz (d), 13.2 Hz (t), 1H), 1.61 (ddd, J = 2.7, 6.0, 13.5 Hz, 1H), 0.90 (s, 9H), 0.09 (s, 6H);  ${}^{13}$ C NMR  $\delta$  -4.8, -4.7, 18.0, 21.0, 28.8 (3C), 34.3, 36.9, 37.2, 39.4, 48.9, 51.5, 64.2, 69.4, 76.7, 127.4 (2C), 129.1 (2C), 136.0, 140.5, 173.4.

(±)-Methyl 6β-(*tert*-Butyldimethylsilyloxy)-3α-(*p*-tolyl)tropane-2α-carboxylate (26a). Using a hydrogenation reaction similar to that described for the preparation of **14a**, **26a** (15 mg, 76%) was obtained from **22** after CC (hexane/EtOAc 1:1) as an oil: <sup>1</sup>H NMR δ 7.10 (d, J = 8.1 Hz, 2H), 7.05 (d, J = 8.1 Hz, 2H), 4.03 (dd, J = 3.3, 7.2 Hz, 1H), 3.78–3.66 (m, 1H), 3.60–3.50 (m, 1H), 3.50–3.38 (m, 4H), 3.09 (d, J = 5.7Hz, 1H), 2.62 (s, 3H), 2.29 (s, 3H), 2.42–2.29 (m, 2H), 2.08– 1.96 (m, 1H), 0.80 (s, 9H), -0.06 (s, 3H), -0.07 (s, 3H); <sup>13</sup>C NMR δ -5.0, -4.9, 17.9, 20.8, 25.7 (3C), 32.5, 34.8, 37.0, 41.1, 48.3, 51.1, 62.1, 68.9, 79.0, 127.3 (2C), 128.5 (2C), 135.1, 139.9, 174.0. Anal. (C<sub>23</sub>H<sub>37</sub>NO<sub>3</sub>Si) C, H, N.

( $\pm$ )-Methyl 6 $\beta$ -Hydroxy-3 $\beta$ -(*p*-tolyl)tropane-2 $\beta$ -carboxylate (23b), ( $\pm$ )-Methyl 6 $\beta$ -Hydroxy-3 $\beta$ -(*p*-tolyl)tropane- $2\alpha$ -carboxylate (24b), and (±)-Methyl 6 $\beta$ -Hydroxy- $3\alpha$ -(ptolyl)tropane-2β-carboxylate (25b) were from 23a, 24a, and 25a, respectively, by the General Procedure for desilylation with *n*-Bu<sub>4</sub>NF. Compound **23b** (13 mg, 91%): mp 169-171 °C; <sup>1</sup>H NMR  $\delta$  7.12 (d, J = 8.4 Hz, 2H), 7.08 (d, J = 8.4 Hz, 2H), 4.47 (dd, J = 3.9, 6.0 Hz, 1H), 3.83 (br s, 1H), 3.50 (s, 3H), 3.31 (br s, 1H), 2.83 (m, 1H), 2.75 (m, 1H), 2.58 (s, 3H), 2.51 (dd, J = 3.3, 12.6 Hz, 1H), 2.30 (s, 3H), 2.27 (m, 2H), 1.86 (br s, 1H), 1.78 (m, 1H); <sup>13</sup>C NMR & 21.0, 30.7, 38.7, 42.5, 51.2, 51.4, 66.2, 70.6, 77.3, 126.9 (2C), 128.7 (2C), 135.4, 139.5, 172.0. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N. Compound **24b** (12 mg, 93%): mp 102–104 °C; <sup>1</sup>H NMR  $\delta$  7.09 (br s, 4H), 4.25 (br s, 1H), 3.64 (m, 1H), 3.61 (s, 3H), 3.45 (m, 1H), 3.20 (d, J = 9.0 Hz, 1H), 2.68 (s, 3H), 2.47 (d, J = 9.0 Hz, 1H), 2.35 (m, 1H), 2.18 (m, 2H), 1.98 (br s, 1H), 1.39 (m, 1H); <sup>13</sup>C NMR δ 20.9, 33.0, 35.7, 40.6, 42.2, 51.8, 54.3, 62.0, 67.3, 79.9, 127.4 (2C), 129.1 (2C), 135.8, 141.0, 175.1. Anal. (C17H23NO3) C, H, N. Compound 25b (11 mg, 97%): mp 222–223 °C; <sup>1</sup>H NMR  $\delta$  7.13 (d, J = 8.1 Hz, 2H), 7.09 (d, J = 8.1 Hz, 2H), 4.34 (br d, J = 6.0 Hz, 1H), 3.58 (br d, J = 6.9 Hz, 1H), 3.52 (s, 3H), 3.14 (m, 2H), 2.93 (ddd, J= 6.0, 12.0, 12.6 Hz, 1H), 2.70 (s, 3H), 2.52 (dd, J = 7.2, 14.1Hz, 1H), 2.30 (s, 3H), 2.18 (br s, 1H), 1.87 (m, 2H), 1.55 (ddd, J = 2.1, 6.0, 13.8 Hz, 1H); <sup>13</sup>C NMR  $\delta$  21.0, 30.0, 35.4, 37.2, 37.6, 44.6, 51.6, 62.3, 67.8, 75.5, 127.3 (2C), 129.2 (2C), 136.1, 140.6, 173.7. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl 6 $\beta$ -Hydroxy-3 $\alpha$ -(*p*-tolyl)tropane-2 $\alpha$ -carboxylate (26b) was obtained (13 mg, 90%) from 26a by desilylation with HF as described for the preparation of 14b: mp 107–109 °C; <sup>1</sup>H NMR  $\delta$  7.09 (d, J = 8.4 Hz, 2H), 7.05 (d, J = 8.4 Hz, 2H), 4.09 (dd, J = 3.0, 7.2 Hz, 1H), 3.74 (m, 1H), 3.63 (m, 1H), 3.48 (s, 3H), 3.46 (m, 1H), 3.17 (m, 1H), 2.68 (s, 3H), 3.63 (dd, J = 6.9, 14.4 Hz, 1H), 2.46 (m, 1H), 2.32 (br s, 1H), 2.00 (m, 2H); <sup>13</sup>C NMR  $\delta$  20.8, 30.4, 34.8, 37.1, 38.5, 44.6, 51.3, 60.8, 68.1, 77.2, 127.4 (2C), 128.7 (2C), 135.4, 139.9, 173.9. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl6β-Hydroxy-8-methyl-3-(p-tolyl)-8-azabicyclo-[3.2.1]oct-2-ene-2-carboxylate (27). Compound 27 (279 mg) was obtained from 22 in 97% yield by desilylation with *n*-Bu<sub>4</sub>NF: mp 144–146 °C; <sup>1</sup>H NMR  $\delta$  7.13 (d, J = 7.8 Hz, 2H), 7.02 (d, J = 7.8 Hz, 2H), 4.23–4.10 (m, 1H), 3.93 (d, J = 6.6 Hz, 1H), 3.50 (s, 3H), 3.21 (d, J = 5.4 Hz, 1H), 2.67 (dd, J = 5.1, 19.5 Hz, 1H), 2.54 (dd, J = 7.5, 13.2 Hz, 1H), 2.47 (s, 3H), 2.34 (s, 3H), 2.09 (d, J = 19.5 Hz, 1H), 2.06–1.98 (m, 1H); <sup>13</sup>C NMR  $\delta$  21.2, 31.5, 34.3, 51.3, 59.9, 66.0, 77.9, 126.5 (2C), 128.7 (2C), 129.5, 137.4 (2C), 142.4, 168.4.

(±)-Methyl 8-Methyl-6-oxo-3-(*p*-tolyl)-8-azabicyclo[3.2.1]oct-2-ene-2-carboxylate (28) was prepared from 27 in 97% (268 mg) yield using a procedure similar to that for the preparation of 16: <sup>1</sup>H NMR  $\delta$  7.12 (d, J = 7.8 Hz, 2H), 7.01 (d, J = 7.8 Hz, 2H), 4.17 (d, J = 6.3 Hz, 1H), 3.50 (s, 3H), 3.45 (d, J = 5.4 Hz, 1H), 2.83 (dd, J = 6.3, 17.4 Hz, 1H), 2.70 (dd, J = 5.7, 19.5 Hz, 1H), 2.53 (s, 3H), 2.52–2.42 (m, 1H), 2.34 (s, 3H), 2.28 (d, J = 19.5 Hz, 1H); <sup>13</sup>C NMR  $\delta$  21.1, 29.8, 34.9, 47.7, 51.4, 58.6, 64.5, 126.5 (2C), 128.7 (2C), 129.4, 136.8, 137.7, 144.2, 167.8, 215.5. Anal. (C<sub>17</sub>H<sub>19</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl6a-Hydroxy-8-methyl-3-(*p*-tolyl)-8-azabicyclo-[3.2.1]oct-2-ene-2-carboxylate (29) was obtained from 28 in 77% yield (202 mg) besides 15% of 27 by NaBH<sub>4</sub> reduction as described in the preparation of compound 17: <sup>1</sup>H NMR  $\delta$  7.12 (d, *J* = 7.8 Hz, 2H), 7.07 (d, *J* = 7.8 Hz, 2H), 4.66–4.54 (m, 1H), 3.75 (d, *J* = 6.6 Hz, 1H), 3.54 (m, 1H), 3.48 (s, 1H), 3.38 (m, 1H), 2.70 (d, *J* = 19.5 Hz, 1H), 2.65–2.53 (m, 1H), 2.46 (s, 3H), 2.34 (s, 3H), 2.46–2.37 (m, 1H), 1.65 (dd, *J* = 1.8, 13.2 Hz, 1H); <sup>13</sup>C NMR  $\delta$  21.1, 26.7, 34.7, 43.5, 51.2, 59.1, 61.6, 72.1, 126.5 (2C), 128.5 (2C), 129.0, 137.2, 137.7, 145.0, 168.6. Anal. (C<sub>17</sub>H<sub>21</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl 6α-(*tert*-Butyldimethylsilyloxy)-8-methyl-3-(*p*-tolyl)-8-azabicyclo[3.2.1]oct-2-ene-2-carboxylate (30) was obtained in 85% yield (119 mg) by silylation of **29** in a similar fashion as described in the preparation of **18**: <sup>1</sup>H NMR δ 7.14 (d, J = 8.1 Hz, 2H), 7.05 (d, J = 8.1 Hz, 2H), 4.64–4.54 (m, 1H), 3.76 (d, J = 6.6 Hz, 1H), 3.50 (s, 3H), 3.33 (dd, J =5.4, 5.7 Hz, 1H), 2.67 (d, J = 19.2 Hz, 1H), 2.61–2.50 (m, 1H), 2.48 (s, 3H), 2.36 (s, 3H), 1.64 (dd, J = 5.4, 12.6 Hz, 1H), 0.90 (s, 9H), 0.07 (s, 3H), 0.06 (s, 3H); <sup>13</sup>C NMR δ –5.0, -4.8, 18.0, 21.2, 25.9 (3C), 27.5, 35.0, 44.1, 51.2, 59.2, 62.0, 73.2, 126.5 (2C), 128.6 (2C), 129.0, 136.9, 138.6, 146.1, 168.4.

(±)-Methyl  $6\alpha$ -(*tert*-Butyldimethylsilyloxy)-3 $\beta$ -(*p*-tolyl)tropane-2 $\beta$ -carboxylate (31a), (±)-Methyl 6 $\alpha$ -(*tert*-Butyldimethylsilyloxy)- $3\alpha$ -(*p*-tolyl)tropane- $2\beta$ -carboxylate (32a), and (±)-Methyl  $6\alpha$ -(*tert*-Butyldimethylsilyloxy)- $3\beta$ -(ptolyl)tropane-2α-carboxylate (33a) were obtained in 6, 69, and 17.5% yield, respectively, by SmI<sub>2</sub> reduction of **30** (297  $\mu$ mol) using a similar procedure as in the preparation of **11a**. The products were purified by CC (hexane/ÉtOAc 2:1-1:1). Compound **31a**: <sup>1</sup>H NMR  $\delta$  7.18 (d, J = 8.1 Hz, 2H), 7.09 (d, J = 8.1 Hz, 2H), 4.74 (m, 1H), 3.50 (s, 3H), 3.50–3.40 (m, 1H), 3.19 (m, 1H), 2.95 (m, 1H), 2.80-2.64 (m, 1H), 2.43 (dt, J= 3.3 Hz (d), 12.9 Hz (t), 1H), 2.35 (s, 3H), 2.30 (s, 3H), 2.06 (dt, J = 3.6 Hz (t), 12.3 Hz (d), 1H), 1.58 (dd, J = 3.3, 13.8 Hz, 1H), 0.85 (s, 9H), 0.08 (s, 3H), 0.05 (s, 3H);  $^{13}\mathrm{C}$  NMR  $\delta$  –5.0, -4.9, 17.9, 21.0, 25.8 (3C), 28.3, 33.6, 36.3, 42.0, 51.0, 52.4, 64.2, 64.9, 71.7, 127.1 (2C), 128.6 (2C), 134.8, 140.5, 172.6. Anal. (C<sub>23</sub>H<sub>37</sub>NO<sub>3</sub>Si) C, H, N. Compound **32a**: <sup>1</sup>H NMR & 7.16 (d, J = 8.1 Hz, 2H), 7.08 (d, J = 8.1 Hz, 2H), 4.59 (m, 1H), 3.56 (s, 3H), 3.36-3.14 (m, 3H), 2.80-2.66 (m, 1H), 2.52 (d, J = 11.1 Hz, 1H), 2.31 (s, 3H), 2.34–2.25 (m, 1H), 1.99 (dt, J =8.7 Hz (t), 14.1 Hz (d), 1H), 1.46 (dd, J = 3.9, 13.5 Hz, 1H), 0.93 (s, 9H), 0.11 (s, 3H), 0.07 (s, 3H);  $^{13}\mathrm{C}$  NMR  $\delta$  –5.0, –4.8, 18.0, 21.0, 25.8 (3C), 27.1, 36.0, 40.7, 40.8, 51.6, 57.1, 61.9, 62.1, 71.0, 127.9 (2C), 129.0 (2C), 135.6, 142.0, 175.4. Anal. (C<sub>23</sub>H<sub>37</sub>NO<sub>3</sub>Si) C, H, N. Compound **33a**: <sup>1</sup>H NMR  $\delta$  7.18 (d, J = 8.1 Hz, 2H), 7.10 (d, J = 8.1 Hz, 2H), 4.60 (m, 1H), 3.51 (s, 3H), 3.62-3.40 (m, 2H), 3.37-3.24 (m, 1H), 3.18 (m, 1H), 2.62 (s, 3H), 2.46-2.35 (m, 1H), 2.31 (s, 3H), 1.94-1.76 (m, 3H), 0.95 (s, 9H), 0.12 (s, 3H), 0.06 (s, 3H);  $^{13}\mathrm{C}$  NMR  $\delta$  –5.1, –4.8,  $18.0,\ 21.1,\ 29.9\ (3C),\ 28.7,\ 35.3,\ 36.1,\ 37.0,\ 45.8,\ 51.4,\ 61.8,$ 63.0, 71.7, 127.4 (2C), 129.2 (2C), 135.8, 141.6, 173.9.

(±)-Methyl 6α-(*tert*-Butyldimethylsilyloxy)-3α-(*p*-tolyl)tropane-2α-carboxylate (34a) was obtained (18 mg, 89%) from **30** by hydrogenation similarly as described for the preparation of **14a**: <sup>1</sup>H NMR  $\delta$  7.30 (d, J = 8.1 Hz, 2H), 7.03 (d, J = 8.1 Hz, 2H), 4.55 (m, 1H), 3.63–3.44 (m, 2H), 3.34 (dd, J = 6.6, 6.9 Hz, 1H), 3.28 (s, 3H), 3.09 (t, J = 6.9 Hz, 1H), 2.58–2.50 (m, 1H), 2.48 (s, 3H), 2.44–2.33 (m, 1H), 2.28 (s, 3H), 2.14–2.02 (m, 1H), 1.91 (dd, J = 5.4, 14.4 Hz, 1H), 0.91 (s, 9H), 0.07 (s, 3H), 0.05 (s, 3H); <sup>13</sup>C NMR  $\delta$  –5.0, –4.9, 18.0, 20.9, 23.2, 25.8 (3C), 33.1, 34.2, 38.7, 47.3, 50.6, 59.6, 61.9, 71.8, 128.4 (2C), 128.6 (2C), 135.0, 140.9, 173.3. Anal. (C<sub>23</sub>H<sub>37</sub>-NO<sub>3</sub>Si) C, H, N.

(±)-Methyl 6 $\alpha$ -Hydroxy-3 $\beta$ -(p-tolyl)tropane-2 $\beta$ -carboxylate (31b), (±)-Methyl 6α-Hydroxy-3α-(p-tolyl)tropane- $2\beta$ -carboxylate (32b), and (±)-Methyl  $6\alpha$ -Hydroxy- $3\beta$ -(ptolyl)tropane-2a-carboxylate (33b) were obtained from 31a, 32a, and 33a, respectively, by the General Procedure for desilylation with *n*-Bu<sub>4</sub>NF in yields of 85%, 88%, and 82%. Compound **31b** (7.3 mg): <sup>1</sup>H NMR  $\delta$  7.18 (d, J = 8.1 Hz, 2H), 7.08 (d, J = 8.1 Hz, 2H), 4.88 (m, 1H), 3.50 (s, 3H), 3.58–3.40 (m, 1H), 3.28 (m, 1H), 3.00 (m, 1H), 2.80 (ddd, J = 7.5, 10.5, 14.1 Hz, 1H), 2.52 (dd, J = 6.3, 12.6 Hz, 1H), 2.34 (s, 3H), 2.99 (s, 3H), 2.08 (dt, J = 3.9 Hz (t), 12.6 Hz (d), 1H), 1.69 (dd, J = 3.9, 14.1 Hz, 1H), 1.64 (s, 1H); <sup>13</sup>C NMR  $\delta$  21.0, 27.7, 33.5, 35.6, 42.0, 51.1, 52.4, 64.2, 64.4, 71.7, 127.0 (2C), 128.6 (2C), 135.0, 140.1, 172.5. Anal. (C17H23NO3) C, H, N. Compound 32b (14 mg): mp 123–124 °C; <sup>1</sup>H NMR  $\delta$  7.19 (d, J = 7.8 Hz, 2H), 7.11 (d, J = 7.8 Hz, 2H), 4.73 (m, 1H), 3.59 (s, 3H), 3.43-3.23 (m, 3H), 2.83 (ddd, J = 7.5, 9.6, 13.8 Hz, 1H), 2.60 (d, J =10.5 Hz, 1H), 2.46 (s, 3H), 2.33 (s, 3H), 2.28-2.16 (m, 1H), 2.14-2.00 (m, 1H), 1.71 (s, 1H), 1.53 (dd, J = 4.2, 13.8 Hz, 1H); <sup>13</sup>C NMR & 21.0, 27.2, 35.9, 39.8, 40.9, 51.7, 56.7, 61.8, 62.1, 70.8, 127.9, 129.1, 135.8, 141.4, 175.3. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N. Compound 33b (8.8 mg): mp 153-154 °C; <sup>1</sup>H NMR  $\delta$  7.20 (d, J = 8.1 Hz, 2H), 7.10 (d, J = 8.1 Hz, 2H), 4.70 (m, 1H), 3.52 (s, 3H), 3.57-3.44 (m, 1H), 3.40 (dd, J=2.4, 7.2 Hz, 1H), 2.25 (dd, J = 2.7, 12.0 Hz, 1H), 3.21 (dd, J = 3.0, 6.3 Hz, 1H), 2.63 (s, 3H), 2.47 (ddd, J = 7.5, 10.5, 13.5 Hz, 1H), 2.31 (s, 3H), 1.93–1.74 (m, 2H), 1.62 (br s, 1H);  $^{13}\mathrm{C}$  NMR  $\delta$  21.0, 28.3, 34.7, 36.2, 37.0, 46.0, 51.5, 61.8, 62.5, 71.7, 127.4 (2C), 129.1 (2C), 135.9, 141.2, 173.9. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N.

(±)-Methyl 6α-Hydroxy-3α-(*p*-tolyl)tropane-2α-carboxylate (34b) was obtained (11 mg, 77%) from 34a by desilylation with HF as described for the preparation of 14b: <sup>1</sup>H NMR δ 7.17 (d, J = 8.1 Hz, 2H), 7.08 (d, J = 8.1 Hz, 2H), 4.63 (m, 1H), 3.66–3.42 (m, 3H), 3.39 (s, 3H), 3.31 (dd, J = 7.8, 8.1 Hz, 1H), 3.01 (br s, 1H), 2.64–2.49 (m, 2H), 2.41 (s, 3H), 2.31 (s, 3H), 2.08 (dt, J = 8.1 Hz (t), 14.4 Hz (d), 1H), 1.52 (dd, J = 2.7, 15.0 Hz, 1H); <sup>13</sup>C NMR δ 20.8, 23.8, 34.0, 34.2, 40.5, 50.4, 51.5, 60.0, 62.2, 71.4, 128.0 (2C), 128.8 (2C), 135.7, 138.7, 175.5. Anal. (C<sub>17</sub>H<sub>23</sub>NO<sub>3</sub>) C, H, N.

**Locomotor Assay Method.** The dose range finding experiment for **12b** was conducted according to the MDD locomotor activity studies standard operating procedure (SOP, Nov. 13, 1992). The study was conducted using 16 Digiscan locomotor activity testing chambers ( $40.5 \times 40.5 \times 30.5$  cm). Panels of infrared beams (16 beams) and corresponding photodetectors were located in the horizontal direction along the sides of each activity chamber. Separate groups of eight nonhabituated male Swiss-Webster mice were injected via the intraperitoneal (i.p.) route with either vehicle (methylcellulose) or **12b** (10 or 100 mg/kg), immediately prior to locomotor activity testing. In all studies, horizontal activity (interruption of 1 photocell beam) was measured for 1 h within 10 min sample periods. Testing was conducted with one mouse per activity chamber.

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Supporting Information Available: X-ray structures of compounds 17, 19a, 12b, and 24a, tables of crystal data,

atomic coordinates, bond lengths and angles, anisotropic and isotropic displacement parameters, and hydrogen coordinates. This material is available free of charge via the Internet at http://pubs.acs.org.

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