

## Synthesis of some 1-(3-hydroxy-4-methoxybenzyl)-2-alkyl-6,7-methylenedioxy-1,2,3,4-tetrahydroisoquinolines and their binding affinities to DA receptor subtypes

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**Summary** — 1-(3-Hydroxy-4-methoxybenzyl)-2-alkyl-6,7-methylenedioxytetrahydroisoquinolines (**3a–c**), derived from the D<sub>1</sub> selective antagonist *S*-bulbocapnine by cleavage of the bond between the 2 aromatic moieties, have been synthesized and their *in vitro* affinity towards D<sub>1</sub> and D<sub>2</sub> receptors evaluated. Of the compounds tested, the 2-methyl derivative **3a** while showing poor affinity towards D<sub>1</sub> receptors was able to inhibit the D<sub>2</sub> radioligand <sup>3</sup>H raclopride binding by 60% at 10<sup>–5</sup> M. Conformational analysis allows reasonable explanations of the loss of D<sub>1</sub>-affinity of **3a** with respect to the model.

**substituted 1-(3-hydroxy-4-methoxybenzyl)-6,7-methylenedioxytetrahydroisoquinoline / D<sub>1</sub> and D<sub>2</sub> receptor binding / bulbocapnine conformational analysis**

### Introduction

Several biochemical and pharmacological studies have led to the identification of different subpopulations of dopamine (DA) receptors in the mammalian central nervous system (CNS) [1]. Of these receptors, the D<sub>1</sub> are located postsynaptically and are positively linked to adenylate cyclase as its second messenger while the D<sub>2</sub> receptors are located both pre- and postsynaptically and are either not coupled or negatively coupled to adenylate cyclase [2–4]. While research has mainly been concentrated on the function of D<sub>2</sub> receptors in the CNS, little is known about the functional significance of D<sub>1</sub> receptors.

Only recently, studies on the D<sub>1</sub> selective antagonist SCH 23390 **1** [5] and *S*-bulbocapnine **2** [6, 7] evidenced a possible role of D<sub>1</sub> receptors in the pathophysiology of some mental diseases [8, 9]. Thus, SCH 23390 has been suggested as an useful drug in the treatment of schizophrenia and mania. It has been found to possess antipsychotic activity [10] without

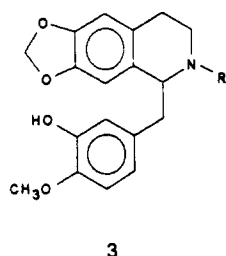
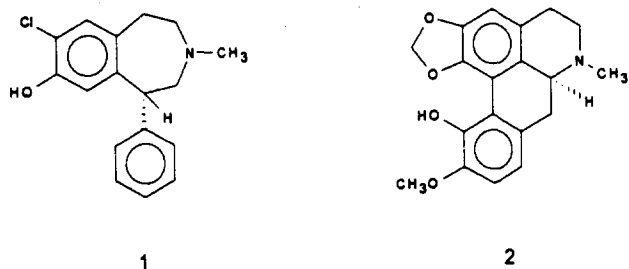
inducing extrapyramidal side-effects such as Parkinsonism and possible dyskinesia, which are induced by the D<sub>2</sub> antagonists currently used as neuroleptics [11, 12].

These considerations together with the evidence that bulbocapnine is quite a selective blocker of DA-sensitive adenylate cyclase prompted us to synthesize and test new tetrahydrobenzyl isoquinolines (**3a–c**) for D<sub>1</sub> and D<sub>2</sub> receptor affinity. They can either be regarded as open derivatives of *S*-bulbocapnine, or as ring-contracted analogs of SCH 23390.

### Chemistry

The synthesis of compounds **3a–c** is depicted in scheme 1. The key intermediate **4** [13] was prepared by condensing 3-benzyloxy-4-methoxy-phenylacetic acid [14, 15] and 3,4-methylenedioxy- $\beta$ -phenethylamine [10] in refluxing xylene with removal of water. Cyclization of the resulting amide **4** in boiling toluene with POCl<sub>3</sub> gave the hydrochloride of **5** in 77% yield. The free base was reacted with the requisite alkyl iodide to give excellent yields of the quaternary salts **6a–c** which were then reduced with sodium boro-

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hydride to the corresponding tetrahydroisoquinolines **7a–c**. Catalytic hydrogenation of the latter over palladium or  $\text{PdCl}_2$  on carbon led to the requisite **3a–c** in satisfactory yield.

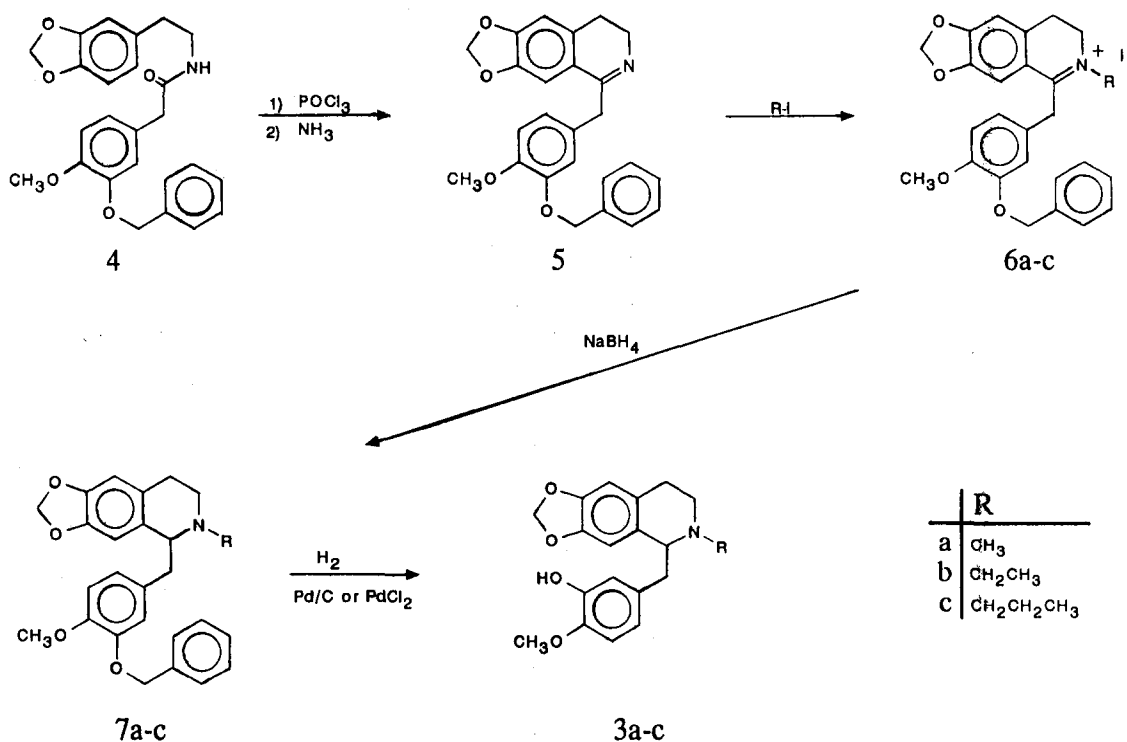
### Receptor binding

The potency and selectivity of *N*-substituted tetrahydroisoquinoline derivatives **3a–c** were evaluated by measuring their *in vitro* affinity for the  $\text{D}_1$  and  $\text{D}_2$  receptors in a competition binding assay, using membrane preparations from rat caudate nucleus.  $^3\text{H}$ -SCH 23390 [17] and  $^3\text{H}$ -raclopride [18] were used as selective ligands for  $\text{D}_1$  and  $\text{D}_2$  dopamine receptors, respectively.

### Molecular modelling

A conformational analysis of **3a** and bulbocapnine was performed using the MM2 force field developed by Allinger and included in the Macromodel [19] molecular software package. The dihedral angles involved in the side chain and in the tetrahydroisoquinolinic moiety were allowed to rotate with  $30^\circ$  increments. With bulbocapnine the global minimum at 36.69 kcal/mol corresponded to a neat planar structure in which the 2 aromatic rings formed an angle of  $28^\circ$ .

The global minimum of **3a** (**A-3a**) at 32.01 kcal/mol showed the benzyl group oriented towards the isoquinolinic nitrogen and did not permit superimposition with bulbocapnine (fig 1).



Scheme 1.

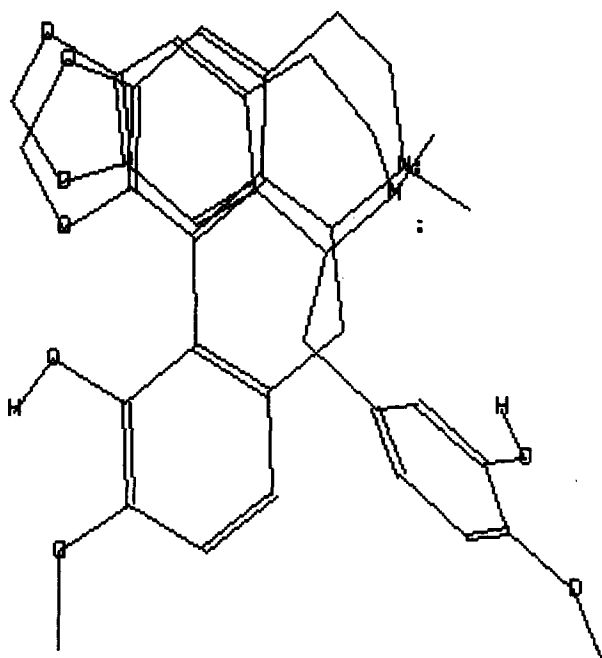


Fig 1. Superimposition of A-**3a** and bulbocapnine.

The conformer in which the benzyl group was oriented in the opposite direction (**B-3a**) was very close to the former (32.77 kcal/mol). In any case no minimum structure of **3a** could superimpose the benzyl group on the bulbocapnine moiety. In fact, even considering the conformers with the desired orientation of the side chain towards the dioxymethylenic moiety, the aromatic rings assumed different bond angles values, that were never close to the  $28^\circ$  angle shown in bulbocapnine (fig 2).

It is interesting to note that both in bulbocapnine and **3a**, the distance ( $> 4 \text{ \AA}$ ) between the phenolic hydrogen and the methylenedioxy oxygen did not allow the formation of an intramolecular hydrogen bond.

## Results and discussion

As illustrated in figure 3, bulbocapnine displaced  $^3\text{H-SCH 23390}$  binding in a concentration-dependent manner. This finding suggests that in the rat CNS this compound binds with higher affinity to  $D_1$  than to  $D_2$  receptors.

In fact, as shown in figure 4, bulbocapnine was much less potent in displacing  $^3\text{H-raclopride}$  from  $D_2$  sites than  $^3\text{H-SCH 23390}$  was from  $D_1$  sites (fig 3). On the contrary, the tetrahydroisoquinoline analog of bulbocapnine, **3a**, was able to maximally inhibit

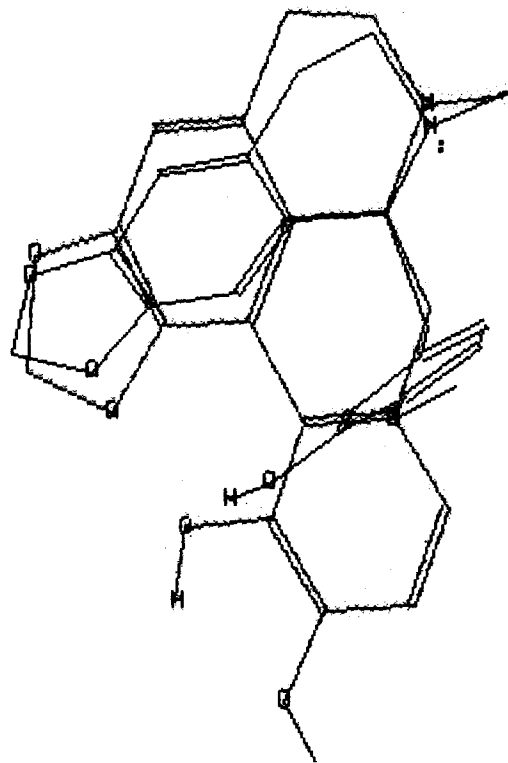
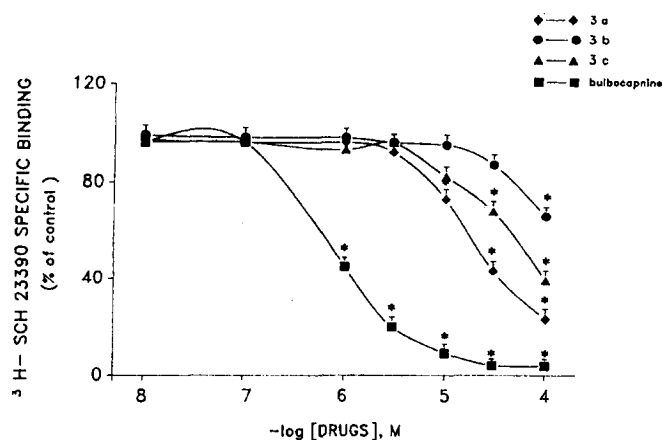


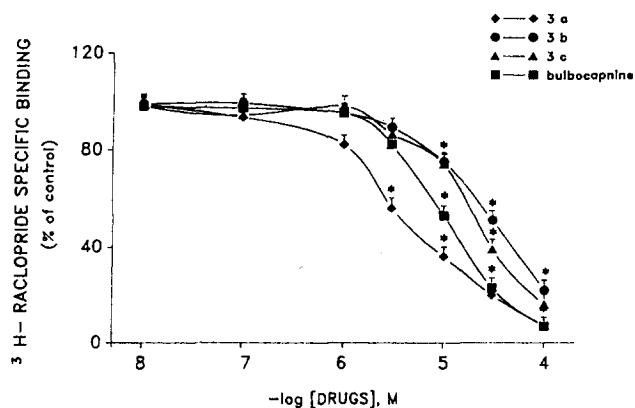
Fig 2. Superimposition of B-**3a** and bulbocapnine.

( $\sim 78\%$ )  $^3\text{H-SCH 23390}$  binding at  $10^{-4} \text{ M}$ , thus evidencing a significant decrease in  $D_1$  receptor affinity with respect to the reference compound.

Substitution of the nitrogen with alkyl groups larger than methyl, as in **3b**, **3c**, led to compounds with lower affinity for the  $D_1$  receptor (inhibition at  $10^{-4} \text{ M}$  being 58 and 32%, respectively) (fig 3). Moreover, **3a** was the most potent compound at inhibiting  $^3\text{H-raclopride}$  binding at  $D_2$  sites. In fact, at  $10^{-5} \text{ M}$  **3a** displaced  $^3\text{H-raclopride}$  binding by 60%, while the homologous **3b** and **3c** at the same concentration showed only  $\approx 20\%$  displacement; by increasing the concentration to  $10^{-4} \text{ M}$ , the binding inhibition was 93, 77 and 83%, respectively. By assuming that interaction of bulbocapnine with  $D_1$  receptors involves fitting of the 2 aromatic rings on properly oriented flat surfaces, the poor affinity of **3a** could reside in the exclusion of its 3-hydroxy-4-methoxy phenyl group from fitting on the flat area of the receptor. On the contrary, the comparable affinity of **3a** and bulbocapnine towards  $D_2$  receptors could be explained by a possible prominent role of the isoquinoline moiety in the interaction with these receptors.



**Fig 3.** Effect of *N*-substituted tetrahydroisoquinoline derivatives and bulbocapnine on  $^3\text{H}$ -SCH 23390 binding in membrane preparations of caudate nucleus from rat brain.  $^3\text{H}$ -SCH 23390 binding was performed as described in the *Experimental protocols* using a concentration of 1 nM of the ligand. Data are mean  $\pm$  SEM of 5 experiments. \* $p < 0.05$  vs solvent (ANOVA followed by Sheffe's test).



**Fig 4.** Effect of *N*-substituted tetrahydroisoquinoline derivatives and bulbocapnine on  $^3\text{H}$ -raclopride binding in membrane preparations of caudate nucleus from rat brain.  $^3\text{H}$ -Raclopride binding was performed as described in the *Experimental protocols* using a concentration of 2 nM of the ligand. Data are mean  $\pm$  SEM of 5 experiments. \* $p < 0.05$  vs solvent (ANOVA followed by Sheffe's test).

## Experimental protocols

### Chemical section

Melting points were determined on a Kofler apparatus or in open capillaries with a Büchi apparatus and are uncorrected. IR spectra were measured on a Perkin-Elmer model (7.81) infrared spectrometer in nujol mulls or as film. Ultraviolet spectra were determined in 95% ethanol on a Perkin-Elmer Lambda 5

model. Nuclear magnetic resonance spectra were determined on a Varian XL-200 instrument with absorptions recorded in ppm downfield from internal  $(\text{Me})_4\text{Si}$ . Elemental analyses were performed by the Padua University microanalysis laboratory and are within  $\pm 0.4\%$  calculated values. TLC was carried out on 2-mm thick E Merck 60F-254 plates. Flash chromatography was performed with Merck 9385 silica gel. Chemical and spectroscopic data of the new differently substituted analogs synthesized have been reported in tables I and II.

### *N*-(3,4-Methylenedioxyphenethyl)-3-benzyloxy-4-methoxyphenylacetic acid amide **4**

A slurry of 5.95 g (0.022 mol) of 3-benzyloxy-4-methoxyphenylacetic acid [14, 15], 3.6 g (0.022 mol) and  $\beta$ -(3,4-methylenedioxyphenyl) ethylamine [16] in 7.5 ml xylene was refluxed under  $\text{N}_2$  for 18 h with azeotropic removal of water. On cooling, 7.1 g (78.02%) of **4** crystallized as a yellow solid, mp: 130–131°C (lit [13] 131–132°C).

### 1-(3-Benzyloxy-4-methoxybenzyl)-6,7-methylenedioxy-3,4-dihydroisoquinoline **5**

A solution of 3 g (0.0075 mol) of the amide **4** and 3 ml  $\text{POCl}_3$  in 210 ml dry toluene (dissolution at 50°C) was refluxed for

**Table I.** Chemical data of 3,4-dihydroisoquinolinium **6** and 1,2,3,4-tetrahydroisoquinoline (**7** and **3**) derivatives.

Compd	Yield (%)	Mp (°C) (solvent)
<b>6a</b>	91	189–192 <sup>a</sup> (EtOH)
<b>6b</b>	81.5	170–172 (EtOH)
<b>6c</b>	93.5	196 (EtOH)
<b>7a</b>	92	Oil 92–94 (HCl) <sup>b</sup>
<b>7b</b>	72.1	Oil 164–167 (HCl) (Acetone)
<b>7c</b>	89	Oil 188–190 (HCl) (Acetone)
<b>3a</b>	53.9	123–125 (EtOH)
<b>3b</b>	58	94–96 (IPA) <sup>c</sup>
<b>3c</b>	65	78–80 (IPA/H <sub>2</sub> O)

<sup>a</sup>See [13], mp: 197°C (ethanol); <sup>b</sup>after trituration in ether;

<sup>c</sup>isopropyl alcohol.

**Table II.** Spectroscopic data of compounds **6**, **7** and **3**.

Compd	UV $\lambda$ max (log $\epsilon$ )	IR $\nu$ (cm <sup>-1</sup> )	<sup>1</sup> H-NMR ( $\delta$ , ppm) <sup>a</sup>
<b>6a</b>	205.5 (4.92); 254.0 (4.27); 281.0 (3.90); 307.0 (3.83); 368.3 (3.94)	1640, 1590	3.29 (t, 2H, C-4 CH <sub>2</sub> ); 3.67 (s, 3H, N-CH <sub>3</sub> ); 3.88 (s, 3H, O-CH <sub>3</sub> ); 4.02 (t, 2H, C-3 CH <sub>2</sub> ); 4.47 (s, 2H, CH <sub>2</sub> -Ph); 5.16 (s, 2H, O-CH <sub>2</sub> -Ph); 6.13 (s, 2H, O-CH <sub>2</sub> -O); 6.50–7.40 (m, 10H, Ar-H)
<b>6b</b>	205.5 (4.70); 254.0 (4.20); 286.0 (3.87); 304.9 (3.89); 3.63.3 (4.03)	1630, 1590	1.60 (t, 3H, CH <sub>2</sub> -CH <sub>3</sub> ); 3.23 (t, 2H, C-4 CH <sub>2</sub> ); 3.87 (s, 3H, O-CH <sub>3</sub> ); 3.96–4.15 (m, 4H, CH <sub>2</sub> -CH <sub>3</sub> and C-3 CH <sub>2</sub> ); 4.51 (s, 2H, CH <sub>2</sub> -Ph); 5.17 (s, 2H, O-CH <sub>2</sub> -Ph); 6.13 (s, 2H, O-CH <sub>2</sub> -O); 6.58–7.40 (m, 10H, Ar-H)
<b>6c</b>	205.3 (4.75); 287.0 (3.91); 301.7 (3.92); 364.4 (3.90)	1630, 1590	0.82 (t, 3H, CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>3</sub> ); 1.49–1.62 (m, 2H, CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>3</sub> ); 3.24 (t, 2H, C-4 CH <sub>2</sub> ); 3.88 (s, 3H, O-CH <sub>3</sub> ); 3.94–4.10 (m, 4H, CH <sub>2</sub> -CH <sub>2</sub> -CH <sub>3</sub> and C-3 CH <sub>2</sub> ); 4.49 (s, 2H, CH <sub>2</sub> -Ph); 5.16 (s, 2H, O-CH <sub>2</sub> -Ph); 6.13 (s, 2H, O-CH <sub>2</sub> -O); 6.50–7.41 (m, 10H, Ar-H)
<b>7a</b>	205.3 (4.75); sh 229.0 (4.34); 282.7 (3.83)	1610, 1590	2.38 (s, 3H, N-CH <sub>3</sub> ); 2.40–3.18 (m, 6H, CH <sub>2</sub> x 3); 3.65 (t, 1H, C-1 CH); 3.77 (s, 3H, O-CH <sub>3</sub> ); 5.01 (s, 2H, O-CH <sub>2</sub> -Ph); 5.88 (s, 2H, O-CH <sub>2</sub> -O); 6.46–7.45 (m, 10H, Ar-H)
<b>7b</b>	204.2 (4.63); 229.0 (4.11); 282.6 (3.60)	1610, 1590	1.01 (t, 3H, CH <sub>2</sub> -CH <sub>3</sub> ); 2.32–3.20 (m, 8H, CH <sub>2</sub> x 4); 3.69 (t, 1H, C-1 CH); 3.87 (s, 3H, O-CH <sub>3</sub> ); 5.08 (s, 2H, O-CH <sub>2</sub> -Ph); 5.84 (s, 2H, O-CH <sub>2</sub> -O); 6.10–7.50 (m, 10H, Ar-H)
<b>7c</b>	205.0 (4.67); sh 232.0 (4.08); 286.8 (3.74)	1610, 1590	0.79 (t, 3H, CH <sub>3</sub> ); 1.38 (m, 2H, CH <sub>2</sub> CH <sub>3</sub> ); 2.40–3.20 (m, 8H, CH <sub>2</sub> x 4); 3.65 (t, 1H, C-1 CH); 3.87 (s, 3H, O-CH <sub>3</sub> ); 5.09 (s, 2H, O-CH <sub>2</sub> -O); 6.15–7.45 (m, 10H, Ar-H)
<b>3a</b>	206.1 (4.74); 234.0 (4.10); 287.8 (3.91)	3400, 1594	2.46 (s, 3H, CH <sub>3</sub> ); 2.48–3.22 (m, 6H, CH <sub>2</sub> x 3); 3.68 (t, 1H, C-1 CH); 3.84 (s, 3H, O-CH <sub>3</sub> ); 5.83–5.86 (m, 2H, O-CH <sub>2</sub> -O); 6.20–6.80 (m, 5H, Ar-H)
<b>3b</b>	205.8 (4.62); 232.0 (3.98); 288.1 (3.76)	3450, 1590	1.05 (t, 3H, CH <sub>3</sub> ); 2.40–3.25 (m, 8H, CH <sub>2</sub> x 4); 3.79 (t, 1H, C-1, CH); 3.87 (s, 3H, O-CH <sub>3</sub> ); 5.83–5.87 (m, 2H, O-CH <sub>2</sub> -O); 6.17–6.78 (m, 5H, Ar-H)
<b>3c</b>	206.1 (4.64); 232.5 (4.00); 288.0 (3.80)	3380, 1600	0.79 (t, 3H, CH <sub>3</sub> ); 1.38–1.49 (m, 2H, CH <sub>2</sub> ); 2.40–3.30 (m, 10H, CH <sub>2</sub> x 5); 3.70 (t, 1H, C-1 CH); 3.83 (s, 3H, O-CH <sub>3</sub> ); 5.82–5.86 (m, 2H, O-CH <sub>2</sub> -O); 6.10–6.80 (m, 5H, Ar-H); 8.25 (br s, 1H, OH, exchange with D <sub>2</sub> O)

<sup>a</sup>All spectra were run in CDCl<sub>3</sub> except for compound **7a** (CD<sub>3</sub>)<sub>2</sub>CO.

12 min and then rapidly cooled to room temperature. The solvent and excess POCl<sub>3</sub> were evaporated *in vacuo*, the residue washed with petroleum ether 3 times and then dissolved in 25 ml ethanol. Upon cooling to 0–5°C, 3 ml concentrated HCl was added followed by 75 ml ether. On standing in a refrigerator for 48–72 h, 2.52 g (77.3%) **5**·HCl crystallized as a yellow solid, mp: 117–120°C. R<sub>f</sub> (CHCl<sub>3</sub>/MeOH, 9:1) 0.71. IR  $\nu$  cm<sup>-1</sup>: 1660. UV  $\lambda$  max (log  $\epsilon$ ): 211.1 (4.35); 245.7 (4.19); 303 (3.71); 363.7 (3.95). <sup>1</sup>H-NMR (CDCl<sub>3</sub>/DMSO-d<sub>6</sub>)  $\delta$ : 2.94 (m, 2H, C-4 CH<sub>2</sub>); 3.54 (m, 2H, C-3 CH<sub>2</sub>); 3.79 (O-CH<sub>3</sub>); (s, 2H, CH<sub>2</sub>-Ph); 3.80 (s, 3H, O-CH<sub>3</sub>); 4.47 (s, 2H, H<sub>2</sub>O); 5.13 (s, 2H, O-CH<sub>2</sub>-Ph); 6.17 (s, 2H, O-CH<sub>2</sub>-O); 6.85–7.62 (m, 10H, Ar-H); 14.02 (br, s, 1H, N-HCl).

The free base **5** was liberated from its hydrochloride in ethanol with 10% ammonium hydroxide followed by dilution

with water and cooling, mp: 87–90°C. R<sub>f</sub> (CHCl<sub>3</sub>/MeOH, 9:1) 0.77. IR  $\nu$  cm<sup>-1</sup>: 1600, 1590. UV  $\lambda$  max (log  $\epsilon$ ): 202 (4.17); 273 (3.62); 320 (3.38). <sup>1</sup>H-NMR (CDCl<sub>3</sub>)  $\delta$ : 2.60 (t, 2H, C-3 CH<sub>2</sub>); 3.65 (t, 2H, C-4 CH<sub>2</sub>); 3.84 (s, 3H, O-CH<sub>3</sub>); 3.89 (s, 2H, CH<sub>2</sub>-Ph); 5.10 (s, 2H, O-CH<sub>2</sub>-Ph); 5.96 (s, 2H, O-CH<sub>2</sub>-O); 6.61–7.50 (m, 10H, Ar-H).

*1-(3-Benzyloxy-4-methoxybenzyl)-2-alkyl-6,7-methylenedioxy-3,4-dihydroisoquinolinium iodide 6a-c*

**General procedure.** A slurry of 0.00486 mol dihydroisoquinoline **5** in 25 ml of the appropriate alkyl iodide was refluxed for 3 h. After dissolution, a yellow solid was precipitated, which was collected after cooling.

*1-(3-Benzoyloxy-4-methoxybenzyl)-2-alkyl-6,7-methylenedioxy-1,2,3,4-tetrahydroisoquinoline 7a-c*

**General procedure.** To a stirred solution of 0.0026 mol alkyl iodide in 18 ml  $\text{CHCl}_3/\text{MeOH}$  (2:1), 0.4 g (0.0112 mol)  $\text{NaBH}_4$  was added in small portions over a 1-h period. Stirring was continued for 0.5 h, the solvent evaporated and the residue diluted with water and extracted with  $\text{CHCl}_3$ . The extract was washed with water, dried ( $\text{Na}_2\text{SO}_4$ ) and evaporated to give **7a-c** as oils which were purified by flash chromatography.

*1-(3-Hydroxy-4-methoxybenzyl)-2-alkyl-6,7-methylenedioxy-1,2,3,4-tetrahydroisoquinoline 3a-c*

**General procedure.** A mixture of 0.00341 mol of **7** in an appropriate solvent (40 ml methanol for **7a**, 30 ml methanol with a few drops of acetic acid for **7b**, and 100 ml ethanol/ethyl acetate (2:1) for **7c**) was shaken with  $\text{H}_2$  (50 psi) at room temperature for a given period (3.5 h for **7a** and 45 h for **7b** and **7c**) in the presence of an appropriate catalyst (2.5 ml 2% aqueous  $\text{PdCl}_2$ , 0.5 g active carbon for **7a** and 0.23 g  $\text{Pd/C}$  (10% by wt) for **7b** and **7c**). The mixture was filtered and the solution evaporated to give a brown oily residue which was purified from the unreacted product by flash chromatography eluting with a solution of  $\text{CHCl}_3/\text{MeOH}$ , 95:5 to give **3a-c** as yellow oils which solidified on standing.

### Biochemical section

#### Receptor binding studies

**Tissue preparation.** Male Sprague-Dawley rats (Charles River, Como, Italy) weighing 150–200 g were used to obtain brain tissue. Rats were killed by decapitation and striata were rapidly dissected and stored frozen at  $-80^\circ\text{C}$  until the binding assay.

Striata were homogenized in 100 vol (w/v) of ice-cold 50 mM Tris-HCl buffer, pH 7.40 (buffer A) and centrifuged at 48 000 g for 10 min. The resultant pellet was washed twice by resuspension and recentrifugation in 100 vol of the same buffer. The final pellet was resuspended in buffer A containing 120 mM NaCl, 5 mM KCl, 2 mM  $\text{CaCl}_2$  and 1 mM  $\text{MgCl}_2$  (buffer B).

**$^3\text{H}$ -SCH 23390 binding assay.** A modification according to Porceddu *et al* [20] of the method of Billard *et al* [17] was used in these experiments. Radioligand binding was performed in 12 x 75-mm culture tubes at a total assay vol of 0.5 ml. Each tube contained 50  $\mu\text{l}$  competitor dissolved in 1% tartaric acid at various concentrations (from  $10^{-8}$  to  $10^{-4}$  M), 50  $\mu\text{l}$   $^3\text{H}$ -SCH 23390 (1 nM), 400  $\mu\text{l}$  tissue homogenate ( $\approx 0.150$  mg tissue per assay) prepared as described above. Non-specific binding was determined in the presence of  $10^{-6}$  M  $^3\text{H}$ -SCH 23390. Tubes were incubated at  $37^\circ\text{C}$  for 60 min and rapidly filtered

*in vacuo* through Whatman GF/B filters with 3 rinses of ice-cold 50 mM buffer A. The filters were placed in plastic mini-vials containing 4 ml scintillation fluid (Atomlight, New England Nuclear) and left overnight. Radioactivity was measured in a scintillation spectrophotometer with an efficiency of 40%.

**$^3\text{H}$ -Raclopride binding assay.** A modification of the method of Kohler *et al* [18] was used in these experiments. Briefly, incubation tubes contained striatal membrane suspension (equivalent to 150  $\mu\text{g}$  protein and  $^3\text{H}$ -raclopride (spec act 86.9 Ci/mmol; New England Nuclear) at a concentration of 2.0 nM in a final vol of 0.5 ml buffer B containing 0.01% (w/v) ascorbic acid. Non-specific binding was determined in the presence of 1  $\mu\text{M}$  (+)-butaclamol. Incubations were carried out for 60 min at  $23^\circ\text{C}$ . After incubation, samples were filtered and counted as described above. Protein was measured according to Lowry *et al* [21] using bovine serum albumin as a standard.

Data were analyzed using ANOVA followed by Scheffe's test.

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