Accepted Manuscript

Title: Transformations of 3-acyl-4*H*-polyfluorochromen-4-ones under the action of amino acids and biogenic amines

Authors: Konstantin V. Shcherbakov, Mariya A. Artemyeva, Yanina V. Burgart, Natalya P. Evstigneeva, Natalya A. Gerasimova, Natalia V. Zilberberg, Nicolai V. Kungurov, Victor I. Saloutin, Oleg N. Chupakhin



PII: DOI: Article Number:	S0022-1139(19)30195-2 https://doi.org/10.1016/j.jfluchem.2019.109354 109354
Reference:	FLUOR 109354
To appear in:	FLUOR
Received date:	7 June 2019
Revised date:	30 July 2019
Accepted date:	30 July 2019

Please cite this article as: Shcherbakov KV, Artemyeva MA, Burgart YV, Evstigneeva NP, Gerasimova NA, Zilberberg NV, Kungurov NV, Saloutin VI, Chupakhin ON, Transformations of 3-acyl-4*H*-polyfluorochromen-4-ones under the action of amino acids and biogenic amines, *Journal of Fluorine Chemistry* (2019), https://doi.org/10.1016/j.jfluchem.2019.109354

This is a PDF file of an unedited manuscript that has been accepted for publication. As a service to our customers we are providing this early version of the manuscript. The manuscript will undergo copyediting, typesetting, and review of the resulting proof before it is published in its final form. Please note that during the production process errors may be discovered which could affect the content, and all legal disclaimers that apply to the journal pertain.

Transformations of 3-acyl-4*H*-polyfluorochromen-4-ones under the action of amino acids and biogenic amines

Konstantin V. Shcherbakov^{a, *}, Mariya A. Artemyeva^{a, b}, Yanina V. Burgart^{a, b}, Natalya P. Evstigneeva^c, Natalya A. Gerasimova^c, Natalia V. Zilberberg^c, Nicolai V. Kungurov^c, Victor I. Saloutin^{a,b} and Oleg N. Chupakhin^{a, b}

^a Postovsky Institute of Organic Synthesis UD RAS, 22 Kovalevskoy str., Yekaterinburg 620990, Russian Federation

^b Ural Federal University, 19 Mira str., Yekaterinburg 620002, Russian Federation

^c Ural Research Institute of Dermatovenereology and Immunopathology, 8 Shcherbakova str., Yekaterinburg 620076, Russian Federation

*Corresponding author. Tel.: +7-343-362-3229; e-mail: kvshcherbakov@gmail.com

Graphical abstract



Highlights

- Synthesis of the new 3-benzoylpolyfluoroflavones
- The alternative routes for modification of 3-acylpolyfluorochromones with amino acids and biogenic amines
- New functionalized polyfluorinated chromones, flavones and aminoenketones
- *N*-substituted 3-amino-1-(2-hydroxypolyfluorophenyl)prop-2-en-1-ones as potential antimycotic agents

Abstract

For the first time 6,7,8-trifluoro- and 5,6,7,8-tetrafluorinated 3-benzoylflavones have been obtained. Their reactions with amino acids and biogenic amines were studied in comparison with polyfluorinated 3-acetyl-2-methylchromones. For chromones, reactions at the C-2 are preferred, which lead to the pyrone ring opening to form *N*-substituted aminoenketones. Whereas in the case of flavones the main route is the nucleophilic aromatic substitution of the fluorine atom at the C-7. Flavones and chromones react in the same way both with dopamine to give aminoenketones, and with proline to form 7-amino-substituted chromen-4-ones. All the reactions of chromen-4-ones are accompanied by deacylation, except ones of flavones with proline. Among the synthesized aminoenketones, compounds with high antimycotic and antibacterial action were found.

Keywords

3-acyl-4*H*-polyfluorochromen-4-ones; amines; nucleophilic aromatic substitution; pyrone ring opening; deacylation; antimycotic and antibacterial activity

1. Introduction

Chromen-4-ones (1,4-benzopyrones) are important scaffold in medicinal chemistry,^[1] as there are numerous examples of natural and synthetic derivatives with diverse biological activities (antithrombotic,^[2] antimicrobial^[3] antitumor^[4] vasodilating activity^[5] *etc.*), some of which are successfully used as modern drugs (*flacoside*, *alvocidib*, *intal*, *quercetin*, *etc.*). In addition, because of their multifunctional nature chromen-4-ones are widely used in organic synthesis as block-synthons to form new open-chain and heterocyclic compounds.^[6] Current studies show wide possibilities for the modification of chromone-based structures by amino acids that allow obtaining a new type of hybrid multifunctional compounds, which combine biological activity with high stability and bioavailability.^[7]

Transformations of fluorine-containing compounds is of growing interest. It is well known that the presence of electron-withdrawing fluorine atoms in the organic molecules not only significantly changes their physical and chemical properties, offering new synthetic possibilities, but also leads to the spectrum variety of biological activity.^[8] Polyfluorinated 3-acylchromen-4-ones are the promising multifunctional compounds, having several nonequivalent electrophilic centers, that make them convenient substrates for modification in reactions with nucleophilic reagents. Previously, we have found^[9] that the main reaction route of 3-acetylpolyfluorochromones with highly basic aliphatic amines and esters of α -amino acids, similar to non-fluorinated substrates,^[10] is pyrone ring opening, accompanied by deacylation, leading to formation of substituted aminovinyl ketones. contrast, the corresponding In 3-ethoxycarbonyl-substituted polyfluoroflavones are able not only to the pyrone ring opening, but also to formation of S_N^{Ar}-substitution products depending on the nature of the amine reagent.^[11]

In this work, the interaction features of 3-acylpolyfluorochromen-4-ones with amino acids (β -alanine, γ aminobutyric acid (GABA) and proline) and biogenic amines (dopamine and pyrrolidine) were studied. A wide range of their biological effect determined the aim of modifying of chromones and flavones with such amines. Based on amino acids the effective antihypertensive, hormonal drugs were designed. Dopamine is known to play an important role in the activity of the central and peripheral nervous system. The pyrrolidine fragment is a part of the natural alkaloids, for example, nicotine and hygrin, in drugs, such as *procyclidine* and *bepridil*.^[12]

2. Results and discussion

Trifluoro- and tetrafluoro-substituted 3-acetylchromones 4a,b and 3-benzoylflavones 5a,b were selected as objects for the modification. For their synthesis the previously proposed method was used^[13] based on acylation of 1,3-diketones 2, 3, activated with magnesium alkoxides, with polyfluorobenzoyl chlorides 1a,b. It should be noted that 6,7,8-trifluoro- and 5,6,7,8-tetrafluoro-substituted 3-benzoyl-2-phenyl-4*H*-chromen-4-ones 5a,b were synthesized for the first time using dibenzoylmethane 3 as a diketone component (Scheme 1). The formation of polyfluorinated chromones 4a,b and flavones 5a,b is caused by spontaneous cyclization of the intermediate triketones **B**, obtained by acid hydrolysis of their magnesium salts **A**. The cyclization proceeds through the intramolecular nucleophilic substitution of the *ortho*-fluorine atom of the polyfluorophenyl ring by hydroxyl group of the enolized diketone fragment.

However, to improve the yields of the compounds **4a**,**b** and **5a**,**b**, the previously used reaction conditions^[13] were optimized. Necessity for the optimization was caused by the formation of asymmetric polyfluorinated 1,3-diketones **6a**, **7a** in reactions of acetylacetone **2** and dibenzoylmethane **3** with tetrafluorobenzoyl chloride **1a** as a result of deacylation of intermediates **B**, as by-products in addition to the target chromene-4-ones **4a**, **5a** (Scheme 1). Only 1,3-diketone **6a** was isolated, while its phenyl analogue **7a** was obtained only in a mixture with flavone **5a**. According to ¹H, ¹⁹F NMR spectra, compound **6a** exists in (CD₃)₂SO as a mixture of keto-enol tautomers in a ratio of 1÷3.

It was found that for more efficient synthesis of polyfluorinated chromen-4-ones **4a**,**b**, **5a**,**b**, it is necessary to remove excess of alcohol from the reaction mixture immediately after the activation of 1,3-diketones **2**, **3** with magnesium methoxide to prevent the esterification of polyfluorobenzoyl chlorides **1a**,**b**. Subsequent neutralization of magnesium salts **A** preferably should be proceeded by using of 0.5M HCl at room temperature, instead of 10% aqueous H_2SO_4 at heating, to prevent the deacylation of the intermediate triketones **B**. For the cyclization of triketones **B** to target chromen-4-ones **4a**,**b**, **5a**,**b** *N*,*N*diisopropylethylamine (DIPEA) was used. As a result of the synthetic method optimization, the yields of products **4a**,**b**, **5a**,**b** were increased up to 80–84% (Scheme 1).



Further, it was found that the result of the interaction of polyfluorinated chromones **4a**,**b** and flavones **5a**,**b** with aliphatic amino acids depends on the reaction conditions. Amino acids are well known to exist predominantly in a zwitter-ionic form in an acidic or neutral media in low polar solvents.^[14] In this regard, at first the reactions of compounds **4a**,**b**, **5a**,**b** with β -alanine and GABA were carried out in ethanol in the presence of AcONa and DIPEA at room temperature. However, under these conditions deacylated chromones **8a**,**b**, **9a**,**b** were obtained as the main products (50–56%) while *N*-substituted aminoenketones **10a**,**b**–**13a**,**b** were formed as minor products (11–20%), resulted from the subsequent pyrone ring opening of compounds **8a**,**b**, **9a**,**b** under the action of amino acids (Scheme 2). To increase the yields of open-chain products **10a**,**b**–**13a**,**b**, reactions of chromen-4-ones **4a**,**b**, **5a**,**b** with β -alanine and GABA were performed in a mixture of ethanol and 0.5M carbonate buffer (Na₂CO₃/NaHCO₃, pH 9.4) at 80°C in a microwave reaction module at discrete power, limited up to 20 W. As a result, the yield of the target aminoenketones **10a**,**b**, **11a**,**b** was significantly increased up to 55% (Scheme 2).

In contrast to chromones **4a,b**, reactions of polyfluoroflavones **5a,b** with β -alanine and GABA proceeded an alternative pathway under these conditions. Trifluoroflavone **5a** reacted with β -alanine and GABA to give flavones **14a**, **15a**, having the amino acid fragment, as a result of the S_N^{Ar} mono-substitution of the fluorine atom at the C-7. Under similar conditions, tetrafluoroflavone **5b** formed a mixture of *N*-substituted 7-amino-5,6,8-trifluoroflavones **14b**, **15b** and 7-amino-6,8-difluoro-5-hydroxyflavones **14c**, **15c** (Scheme 2) in a ratio of 2÷1 (for **14b**, **14c**) and 3÷1 (for **15b**, **15c**). In the case of GABA, by ramping the reaction temperature and pressure (130°C, 0.9–1.0 MPa) it was succeeded to synthesize the mono-substituted flavone **15b** almost selectively, since its content in a mixture increased up to 10÷1, according to ¹⁹F NMR spectroscopy data. However, we did not manage to find conditions for the selective synthesis of flavones **14b**, **14c**, and **15b**, **15c**, as well as to isolate them individually because of their joint crystallization and elution. The structure of compounds **14b,c**, **15b,c** was confirmed by ¹H, ¹⁹F NMR and HRMS data of their mixtures.



Polyfluorochromen-4-ones **4a**,**b**, **5a**,**b** reacted with dopamine in ethanol at room temperature forming *N*-substituted aminoenketones **16a**,**b**, **17a**,**b** as a result of the pyrone ring opening and deacylation (Scheme 3). The transformations of flavones **5a**,**b** with dopamine were accompanied by a significant resinification of the reaction mass; therefore these reactions were carried out in absolute ethanol under argon atmosphere.



According to ¹H, ¹⁹F NMR spectra, aminobutenones **10a,b**, **11a,b**, **16a,b** exist as mixtures of (*E*,*Z*)-isomers in (CD₃)₂SO with a predominance of (*Z*)-form. While ¹H, ¹⁹F NMR spectra of phenyl analogues **12a,b**, **13a,b**, **17a,b** contain the only one set of signals of the (*Z*)-isomer. (*Z*)-Aminobutenones **10a,b**, **11a,b**, **16a,b** are characterized by the low-field shift of a proton signal of the amino group (δ 10.83–11.08 ppm) that determined by intramolecular *H*-bonding with the carbonyl group, in contrast to the analogous signal of (*E*)isomer (δ 7.68–8.48 ppm). Moreover, methylene proton of the (*Z*)-isomer, existing in pseudo-aromatic enol form, was registered in the low-field region, compared to the CH-signal of the (*E*)-isomer (see Supplementary material for full NMR data).

Structure of aminoenketones **10b** and **17a** in the solid state was confirmed by XRD. It was turned out that crystals of aminobutenone **10b**, having an acidic fragment, were isolated as the sodium salt **10b**·Na. According to XRD data (Fig. 1, 2), compounds **10b**·Na, **17a** exist in (*Z*)-aminoenketone form, stabilized by the two pairs of intramolecular hydrogen bonds O2–H2…O1 and N1–H1…O1 (**17a**, Fig. 1), and O4–H4…O1

and N1–H1...O1 (10b·Na, Fig. 2) (see Supplementary material for parameters of the hydrogen bond of compounds 10b·Na and 17a). Due to the stabilization, atoms of the aminoenketone fragment of the compounds 17a ({N1,C15,C16,C17,O1}, Fig. 1) and 10b·Na ({N1,C5,C6,C7,O1}, Fig. 2) are located in the plane of the fluoroaromatic ring. The molecules of compound 17a are bound into dimers by intermolecular contacts between the amino and carbonyl groups of the aminoenketone fragment of one molecule and the hydroxyl group of the dopamine moiety of another one (Fig. 3). Zig zag layers with an interplanar distance of 3.333 Å are formed in the molecular packing of compound 17a due to the π - π interaction of the fluoroaromatic fragment and the σ - π interaction of the dopamine moiety of the molecules (Fig. 3). The crystals of 10b·Na are polymer structures, in which sodium ions are linked with each another through the two molecules of water; at the same time, each sodium ion binds another water molecule. Water molecules link dimers into polymer chains in direction to the *b*-axis due to the formation of intermolecular contacts, in which oxygen atoms of the carboxylate fragment are involved (Fig. 4).



Figure 1. Molecular structure of compound 17a with atoms represented as thermal ellipsoids of thermal vibrations with a 50% probability



Figure 2. Molecular structure of compound **10b Na** with atoms represented as thermal ellipsoids of thermal vibrations with a 50% probability



Figure 3. Molecular packaging of compound 17a

Figure 4. Molecular packaging of compound 10b·Na along the *b*-axis

It was found that transformations of polyfluorinated chromen-4-ones **4a,b**, **5a,b** with the secondary heterocyclic amines (such as pyrrolidine and its 2-carboxylated analogue proline) under similar reaction conditions depend on the nature of fluorinated substrates and nucleophiles used (Schemes 4, 5). In ethanol (ε 24.3)^[15] under the action of pyrrolidine, chromones **4a,b** undergo pyrone ring opening and deacylation, forming 3-(pyrrolidin-1-yl)-1-(polyfluorophenyl)but-2-en-1-ones **18a,b**. In more polar dimethylsulfoxide (ε 49)^[15] these transformations are accompanied by substitution of a fluorine atom to form compounds **19a,b**, having two pyrrolidine fragments (Scheme 4). However, interaction of chromones **4a,b** with proline in the same reaction conditions had no significant conversion, therefore a mixture of EtOH and 0.5M carbonate buffer was used. Only 7-[2-(carboxy)pyrrolidin-1-yl]-substituted 4*H*-chromen-4-ones **20a,b** as products of S_N^{Ar} substitution of fluorine atoms were isolated in an individual form (Scheme 4).



Scheme 4. Reactions of 3-acetyl-4H-chromen-4-ones with pyrrolidine and proline

According to XRD data, 3-(pyrrolidinyl)buthenone **18b** is in (*E*)-aminoenketone form in the solid state, in contrast to aminoenketones **10a**,**b**–**13a**,**b**, **16a**,**b**, **17a**,**b** having an NH-group, which stabilizes the (*Z*)-isomer (Fig. 5). The molecules of **18b** are characterized by one intramolecular hydrogen bond O5–H5A[…]O1. Besides, the only one set of signals was registered in the ¹H, ¹⁹F NMR spectra of compounds **18a**,**b**, **19a**,**b** in (CD₃)₂SO, corresponding to the (*E*)-form, obviously.



Figure 5. Molecular structure of compound 18b with atoms represented as thermal ellipsoids of thermal vibrations with a 50% probability

Only 7-substituted 3-benzoylflavones **21a,b**, **22a,b** were isolated from the reaction of polyfluoroflavones **5a,b** with pyrrolidine and proline as products of nucleophilic aromatic substitution of fluorine atoms (Scheme 5). It should be noted that the isolated products **21a,b**, **22a,b** have retained benzoyl function in contrast to chromones **4a,b**.



We have studied antimycotic and antibacterial activities of some synthesized compounds. Antimycotic evaluations were carried out against eight fungal strains (Table 1). *Terbinafine* was used as a positive control. *N*-Substituted aminoenketones showed an antimycotic effect from moderate (compounds **10b**, **11b**, **12a**, **16a**,**b**) to high values (compounds **13b**, **17b**).

Tetrafluorinated 3-phenylpropenones **13b**, **17b** revealed the highest activity of the tested compounds. 3-Phenylpropenone **17b** having dopamine moiety showed the elevated antimycotic activity against *T. rubrum* and *M. canis* (MIC 1.56 μ g/mL), *T. tonsurans* and *T. violaceum* (MIC 3.12 μ g/mL) along with the moderate activity against *E. floccosum* and *C. albicans* (MIC 12.5 μ g/mL). 3-Phenylpropenone **13b**, involving 4-aminobutanoic acid fragment, had the same activity against *T. rubrum* and *M. canis* (MIC 1.56 μ g/mL). In contrast to tetrafluorinated compounds **13b**, **17b**, their trifluorinated analogues **13a**, **17a** exhibited the low antimycotic action. Change of phenyl substituent in the aminoenketones **13b**, **17b** to methyl group decreased the antimycotic avtivity of compounds **11a,b**, **16a,b**. We did not observe the clear correlation between fungicidal action and the number of the fluorine atoms in the case of compounds **11a,b**, **16a,b**. In this series trifluorinated butenone **16a** containing dopamine moiety was more effective, suppressing *T. violaceum* strains at MIC of 12.5 μ g/mL. In contrast to *N*-substituted aminoenketones, chromone derivatives **14a**, **15a**, **20a**, **21b**, **22a**, as well as diaminosubstituted aminoenketone **19b**, did not show antimycotic activity. Most antimicrobial drugs act by a non-enzymatic mechanism, accompanied by a destruction of the microbial cell membrane.^[16] It is possible that chromones

14a, 15a, 20a, 21b, 22a possess the less adsorption on the surface of microbial cells, unlike openchain products 13b, 16a, 17b, having aminoenediketone fragment.

		Minimum inhibitory concentration (MIC), µg/mL									
Entry	Compound	T. rubrum	T. gypseum	T. tonsurans	T. violaceum	T. interdigitale	E. floccosum	M. canis	C. albicans		
1	10a	50	100	50	50	100	100	50	100		
2	10b	25	50	50	50	50	50	25	100		
3	11a	50	100	100	100	100	100	50	100		
4	11b	>200	>200	>200	>200	50	25	25	>200		
5	12a	25	200	200	>200	>200	>200	25	>200		
6	12b	12.5	200	200	>200	200	200	200	>200		
7	1 3 a	25	50	50	>200	200	200	200	>200		
8	13b	1.56	25	50	200	200	50	1.56	>200		
9	14a	>200	>200	>200	>200	>200	>200	>200	>200		
10	15a	>200	>200	>200	>200	>200	>200	>200	>200		
11	16a	25	100	50	12.5	100	50	50	100		
12	16b	25	25	25	25	100	100	100	>200		
13	17a	50	100	100	>200	>200	>200	100	>200		
14	17b	1.56	25	3.12	3.12	200	12.5	1.56	12.5		
15	19b	>200	>200	>200	>200	>200	>200	>200	>200		
16	20a	>200	>200	>200	200	>200	>200	>200	>200		
17	21b	>200	>200	>200	>200	>200	>200	>200	>200		
18	22a	200	>200	>200	100	>200	>200	200	>200		
Terbinafine		0.002	0.001	_	0.01	0.01	0.01	0.02	6.25		

Table 1. Antimycotic activity of synthesized compounds.

Testing of synthesized compounds against eight strains of conditionally pathogenic microorganisms and *N. gonorrhoeae* revealed a higher activity of *N*-substituted aminoenketones **12a**, **16b** against *St. aureus* and *St. aureus MRSA* (MIC 15.6 mg/mL), compared to *Spectinomycin*.

Table 2. Antibacterial activity of synthesized compound	ds.
---	-----

		Minimum inhibitory concentration, MIC, µg/mL									
Entry Compound		N. gonorrhoeae	E. coli	K. pneumoniae	C. braakii	S. marcescens	S. flexneri	P. auruginosa	St. aureus	St. aureus MRSA	
1	10a	125	125	250	125	250	250	>250	250	250	
2	10b	62.5	125	>250	125	250	250	>250	62.5	62.5	
3	11a	125	250	250	250	250	250	250	250	250	
4	11b	n.d.	125	250	250	250	125	>250	62.5	62.5	

5	12a	62.5	>250	>250	>250	>250	>250	>250	15.6	15.6
6	12b	62.5	>250	>250	>250	>250	>250	>250	>250	>250
7	1 3 a	62.5	>250	>250	>250	>250	>250	>250	>250	>250
8	13b	62.5	>250	>250	>250	>250	>250	>250	>250	31.2
9	14a	31.2	>250	>250	>250	>250	>250	>250	>250	>250
10	15a	250	>250	>250	>250	>250	>250	>250	>250	>250
11	16a	125	>250	250	>250	>250	>250	>250	62.5	>250
12	16b	n.d.	>250	250	>250	>250	250	>250	15.6	15.6
13	17a	62.5	>250	>250	>250	>250	>250	>250	>250	>250
14	17b	250	>250	>250	>250	>250	>250	>250	>250	>250
15	19b	>250	>250	>250	>250	>250	>250	>250	>250	>250
16	20a	62.5	>250	>250	>250	>250	>250	>250	>250	>250
17	21b	31.2	>250	>250	>250	>250	>250	>250	>250	>250
18	22a	250	>250	>250	>250	>250	>250	>250	>250	125
Spectin	omycin	15.6	15.6	15.6	31.25	15.6	7.8	250	31.25	n.a.

n.d. – not determined, n.a. – data not available

3. Conclusion

The present study has demonstrated that multifunctional character of polyfluorinated 3acylchromen-4-ones 4a,b, 5a,b determines the different routes for their interaction with biogenic amines and amino acids, depending on the structure of the fluorine-containing substrate and nucleophilic reagent as well as reaction conditions. Chromones 4a,b react with the primary amines (amino acids and dopamine) to give N-substituted aminoenketones as a result of nucleophilic attack at the C-2 with the subsequent pyron ring opening. In addition, flavones 5a,b can react with amino acids with retention of the pyrone ring, forming products of the fluorine atom substitution. The S_N^{Ar} process becomes the main route for the reactions of flavones **5a**,**b** with secondary amines and an alternative for the similar transformations of chromones 4a,b. At the same time, all these transformations are accompanied by deacylation, even under mild conditions, except for the interaction of flavones 5a,b with secondary amines. Deacylation of acyl-functionalized derivatives is known to easily carry out under the basic conditions.^[10,17] Differences in the reactivity of chromones 4a,b and flavones 5a,b when interacting with primary and secondary amines can be explained by the steric factors of an amine attack on the C-2 of the chromenone cycle. In the case of a difficult attack on C-2, an alternative path of transformations along the activated C-7 position of the aromatic cycle is realized. The presence of fluorine atoms in the aromatic part of the studied chromenones 4a,b, 5a,b provides the additional opportunities for their transformations. The case of simultaneous S_N^{Ar} substitution, pyrone ring opening and deacylation in the reaction of chromone 4a,b with pyrrolidine was found. Among the synthesized aminoenketones, compounds with high antimycotic and antibacterial action were found. Moreover, 7-substituted flavones may reveal angioprotective activity, as well as GABA-modified aminoenketones may be of interest as neurotransmitters. Chromenes and aminoenketones, having dopamine fragment, can be the promising effectors of the corresponding dopamine receptors.

Acknowledgements

The work was supported by the State project AAAA-A19-119012290115-2.

The equipment of the CCU "Spectroscopy and analysis of organic compounds" of the Postovsky Institute of Organic Synthesis UD RAS was used in this work. Microbiological assay of the synthesized compounds was performed at the Ural Research Institute of Dermatovenereology and Immunopathology.

4. Experimental section

4.1. General information

All the reagents except fluorine-containing benzoyl chlorides, chromones and flavones are commercially available and used without further purification. Solvents were prepared according to standard methods of purification. Synthesis under microwave irradiation was carried out in a CEM Discover SP Explorer Hybrid microwave reaction module in closed reaction vessels. The Nuclear magnetic resonance spectra (NMR) of the synthesized compounds were recorded on a Bruker DRX-400 and Bruker AVANCE III 500 spectrometers (¹H, 400 (DRX400) and 500 (AV500) MHz, ¹³C, 100.6 MHz, SiMe₄ as an internal standart, ¹⁹F, 376 (DRX400) and 470 (AV500) MHz, C₆F₆ as internal standart (chemical shifts were converted to CCl₃F)). IR spectra were recorded on a Perkin Elmer Spectrum One FT-IR spectrometer using a diffuse reflectance attachment (DRA) in the range of 4000–400 cm⁻¹. High-resolution mass spectra (HRMS) were recorded on a Bruker MaXis Impact HD mass spectrometer (ESI). Elemental (C, H, N) analysis was performed on a Perkin Elmer PE 2400 Series II CHNS-O EA 1108 elemental analyzer. Melting points were measured on a Stuart SMP3 in open capillaries and are uncorrected. The reaction progress was monitored by TLC on ALUGRAM SIL G/ UV₂₅₄ plates. Column chromatography was performed on Alfa Aesar silica gel 60 (0.063–0.2 mm).

4.2. General procedure of synthesis of 3-acyl-4H-polyfluorochromen-4-ones 4a,b, 5a,b.

In a three-necked flask, equipped with a reflux condenser and a drop funnel, several iodine crystals and 1.2 g (50 mmol) of magnesium turnings were added, heated with stirring until the vigorous sublimation of iodine, then 10 mL (250 mmol) of MeOH and a few drops of CCl₄ were added. While vigorous hydrogen evoluation, 25 mL of absolute toluene was added to the mixture. Then 50 mmol of corresponding diketone **2**, **3** was added in portions while stirring. The mixture was refluxed until the magnesium turnings were completely dissolved, and then cooled to room temperature. The solvent was removed *in vacuo* till the mass thickening, then 25 mL of absolute toluene was added to the mixture dropwise with stirring at the room temperature. The mixture was stirred for 6 h at room temperature, then poured into 150 mL 0.5M HCl. An organic layer was separated and dried over anhydrous Mg₂SO₄. To the mixture, separated from Mg₂SO₄, 0.6 g (5 mmol) of DIPEA was added, and then refluxed for 2 h. The solvent was removed *in vacuo*, and the residue was crystallized fractionally from toluene.

4.2.1. 3-Benzoyl-6, 7, 8-trifluoro-2-phenyl-4H-chromen-4-one (**5a**) was synthesized from 11.2 g (50 mmol) of dibenzoylmethane **3** and 10.6 g (50 mmol) of tetrafluorobenzoyl chloride **1a**. Yield 15.2 g (80%), white powder, mp 181–182°C. IR (DRA): v 1676 (Bz), 1632 (C=O), 1523, 1496, 1447 (C=C, C–H), 1006, 927 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO+CCl₄): δ 7.39–7.91 (10H, m, 2 Ph), 7.79 (1H, m, H, C-5). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO+CCl₄): δ -152.11 (1F, d, *J* = 18.8 Hz), -150.48, -136.65 (2F, both m). Anal. calcd. for C₂₂H₁₁F₃O₃: C, 69.48; H, 2.92; found: C, 69.57; H, 3.18.



4.2.2. 3-Benzoyl-5,6,7,8-tetrafluoro-2-phenyl-4H-chromen-4-one (5b) was synthesized from 11.2 g (50 mmol) of dibenzoylmethane 3 and 11.5 g (50 mmol) of pentafluorobenzoyl chloride 1b. Yield 16.7 g (84%), white powder, mp 178–180°C. IR (DRA): v 1674 (Bz), 1645 (C=O), 1522, 1502, 1448 (C=C, C–H), 994 (C–F) cm⁻¹. ¹H NMR (400.13 MHz,

 $(CD_3)_2SO+CCl_4$): δ 7.39–7.95 (10H, m, 2 Ph). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO+CCl₄): δ - 162.17, -159.21, -148.87, -144.13 (4F, all m). Anal. calcd. for C₂₂H₁₀F₄O₃: C, 66.34; H, 2.53; found: C, 66.00; H, 2.66.

4.2.3. 1-(2,3,4,5-*Tetrafluorophenyl*)*butane*-1,3-*dione* (*6a*) was synthesized from 1.0 g (10 mmol) of acetylacetone **2** and 2.1 g (10 mmol) of tetrafluorobenzoyl chloride **1a** under reaction conditions.^[13] Yield 585 mg (25%), orange powder, mp 82–85°C (CHCl₃/C₆H₁₄). ¹H NMR (500.13 MHz, (CD₃)₂SO) spectrum of keto-enol mixture, 3÷1, enol form: δ 2.23 (3H, s, CH₃), 6.34 (1H, s, CH), 7.79–7.84 (1H, m, H, C₆HF₄), 15.82 (1H, br.s, OH); keto form: δ 2.23 (3H, s, CH₃), 4.25 (2H, d, *J* = 3.1 Hz, CH₂), 7.78–7.84 (1H, H, C₆HF₄). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO) spectrum of keto-enol mixture, enol form: δ -154.96, -150.28, -138.71, -137.57 (4F, all m), keto form: δ -155.02, -148.42, -138.60, -137.70 (4F, all m). Anal. calcd. for C₁₀H₆F₄O₂: C, 51.30; H, 2.58; found: C, 51.19; H, 2.65.

4.2.4. 1-Phenyl-3-(2,3,4,5-tetrafluorophenyl)propane-1,3-dione (7a) in a mixture with **5a** was synthesized from 11.2 g (50 mmol) of dibenzoylmethane **3** and 10.6 g (50 mmol) of tetrafluorobenzoyl chloride **1a** under reaction conditions. ^[13] ¹H NMR (500.13 MHz, (CD₃)₂SO): δ 7.08, (1H, s, CH), 7.57–8.18 (5H, m, Ph), 7.67 (1H, m, H, C-5). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO): δ -154.99, -149.97, -138.85, -137.54 (4F, all m).

4.3. General procedures for the synthesis of compounds 8-22.

Method A. 3-Acyl-4*H*-polyfluorochromen-4-one **4a**,**b**, **5a**,**b** (1 mmol), amino acid (2 mmol) and sodium acetate trihydrate (2 mmol) were suspended in 10 mL of EtOH, then 65 mg (0.5 mmol) of DIPEA were added to a mixture. Reaction mixture was stirred at room temperature for 96 hrs. At the end of the reaction, the mixture was poured onto 50 mL of 0.5M HCl, extracted with AcOEt, 3×10 mL. Organic layer was dried with Mg₂SO₄. Then mixture was separated from Mg₂SO₄, the solvent was removed in vacuo. The residue was purified by crystallization from a suitable solvent or by column chromatography on silica gel, using suitable eluent.

Method B. 3-Acyl-4*H*-polyfluorochromen-4-one **4a,b**, **5a,b** (1 mmol), nucleophile reagent (2 mmol) were suspended in a mixture of 5 mL of EtOH and 5 mL of 0.5M carbonate buffer. The mixture was placed in a reaction vessel and irradiated in a microwave reactor at a discrete power (10–20 W) and a constant temperature (80 °C) for 4 hrs while stirring. At the end of the reaction, the mixture was poured onto 50 mL of 0.5M HCl, extracted with AcOEt, 3×10 mL. Organic layer was dried with Mg₂SO₄. Then the mixture was separated from Mg₂SO₄, the solvent was removed *in vacuo*. The residue was purified by crystallization from a suitable solvent or by column chromatography on silica gel, using suitable eluent.

Method C. 3-Acyl-4*H*-polyfluorochromen-4-one **4a**,**b**, **5a**,**b** (1 mmol) was suspended in 10 mL of solvent, an excess of amine (2 or 4 mmol) was added to the mixture. The mixture was stirred at room temperature for 8 hrs. At the end of the reaction, the mixture was poured onto 50 mL of 0.5M HCl, the precipitate was filtered off, washed with water, and crystallized from a suitable solvent.

4.3.1. 6,7,8-*Trifluoro-2-methyl-4H-chromen-4-one* (*8a*) was synthesized according to the *method A* from 256 mg (1 mmol) of chromone **4a** and 2 mmol of the corresponding amino acid (β-alanine, GABA). White powder, mp 107–109°C (EtOH). ¹H NMR (400.13 MHz, (CD₃)₂SO): δ 2.44 (3H, s, CH₃), 6.37 (1H, s, H, C-3), 7.80 (1H, d.d.d, $J_{HF} = 10.2$, 8.0, 2.3 Hz, H, C-5). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -152.14, -151.70, -137.74 (3F, all m). Anal. calcd. for C₁₀H₅F₃O₂: C, 56.09; H, 2.35; found: C, 56.44; H, 2.40.

4.3.2. 5,6,7,8-*Tetrafluoro-2-methyl-4H-chromen-4-one* (**8b**) was synthesized according to *method A* from 274 mg (1 mmol) of chromone **4b** and 2 mmol of the corresponding amino acid (β -alanine, GABA). White powder, mp 102–104°C (EtOH), (lit.^[13a] 102–104°C). ¹H NMR (500.13 MHz, (CD₃)₂SO): δ 2.41 (3H, s, CH₃), 6.30 (1H, s, H, C-3). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO): δ -172.90, -164.97, -154.99, -139.67 (4F, all m).

4.3.3. 6,7,8-*Trifluoro-2-phenyl-4H-chromen-4-one* (**9***a*) was synthesized according to *method A* from 380 mg (1 mmol) of flavone **5a** and 2 mmol of the corresponding amino acid (β-alanine, GABA). White powder, mp 203–206°C (EtOH). ¹H NMR (500.13 MHz, (CD₃)₂SO): δ 7.18 (1H, s, H, C-3), 7.61–8.09 (5H, m, Ph), 7.87 (1H, d.d.d, J_{HF} = 10.0, 8.0, 2.3 Hz, H, C-5). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO): δ -152.02, -151.31, -137.38 (3F, all m). Anal. calcd. for C₁₅H₇F₃O₂: C, 65.23; H, 2.55; found: C, 65.28; H, 2.57.

4.3.4. 5,6,7,8-*Tetrafluoro-2-phenyl-4H-chromen-4-one* (**9b**) was synthesized according to *method A* from 398 mg (1 mmol) of flavone **5b** and 2 mmol of the corresponding amino acid (β -alanine, GABA). White powder, mp 224-226°C (EtOH). ¹H NMR (400.13 MHz, (CD₃)₂SO): δ 7.13 (1H, s, H, C-3), 7.59–8.07 (5H, m, Ph). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -162.64, -159.10, -149.72, -145.42 (4F, all m). Anal. calcd. for C₁₅H₆F₄O₂: C, 61.24; H, 2.06; found: C, 61.72; H, 2.43.

N-[1-Methyl-3-oxo-3-(3,4,5-trifluoro-2-hydroxyphenyl)prop-1-enyl]-3-4.3.5. aminopropanoic acid, hydrate (10a) was synthesized from 256 mg (1 mmol) of chromone **4a** and 179 mg (2 mmol) of β -alanine. Yield 45 mg (14% method A), 179 mg (55% method B), yellow powder, mp 135–140°C (CHCl₃). IR (DRA): v 3545, 3466 (N–H), 2970, 2924 (O–H), 1713 (CO₂H), 1596, 1566 (C=O), 1509, 1437, 1364, 1258 (C-N, C=C, C-H), 1032, 1010 (C-F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO) spectrum of (*E*,*Z*)-isomer mixture, $10 \div 1$, (*Z*)-isomer: δ 2.18 $(3H, s, CH_3), 2.59 (2H, t, {}^{3}J_{HH} = 6.4 Hz, CH_2, C-2), 3.58 (2H, q, {}^{3}J_{HH} = 6.4 Hz, CH_2, C-3), 5.88$ $(1H, s, H, C-6), 7.79 (1H, d.d.d, J_{HF} = 11.6, 9.0, 2.0 Hz, H, C-9), 10.87 (1H, t, {}^{3}J_{HH} = 6.1 Hz, NH),$ 14.50 (1H, br.s, OH); (E)-isomer: δ 2.40 (3H, s, CH₃), 2.88 (2H, t, ³J_{HH} = 6.3 Hz, CH₂, C-2), 3.49 (2H, m, CH₂, C-3), 5.74 (1H, s, H, C-6), 7.98 (1H, d.d.d, J_{HF} = 11.5, 8.9, 2.1 Hz, H, C-9), 8.34 (1H, br.s, NH). ¹³C NMR (125.76 MHz, (CD₃)₂SO) spectrum of (Z)-isomer: δ 19.21 (s, C-4), 34.26 (s, C-2), 39.40 (m, C-3+(CD₃)₂SO), 90.80 (s, C-6); 109.33 (br.s, C-9), 115.97 (br.s, C-8), 139.84 (m, C-12), 141.52 (m, C-10), 141.96 (m, C-11), 148.30 (m, C-13), 168.40 (m, C-5), 172.64 (s, C-1), 184.84 (br.s, C-7). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO) spectrum of (*E*,*Z*)-isomer mixture, 10÷1, (Z)-isomer: δ -159.33, -154.81, -149.45 (3F, all m); (E)-isomer: δ -160.01, -154.91, -150.22 (3F, all m). Anal. calcd. for C₁₃H₁₂F₃NO₄·H₂O: C, 48.60; H, 4.39; N, 4.36; found: C, 48.60; H, 4.15; N, 3.73.



4.3.6. N-[1-Methyl-3-oxo-3-(2,3,4,5-tetrafluoro-6-hydroxyphenyl)prop-1-enyl]-3-aminopropanoic acid, hemihydrate (10b) was synthesized from 274 mg (1 mmol) chromone**4b** $and 178 mg (2 mmol) <math>\beta$ -alanine. Yield 66 mg (20% method A), 185 mg (54% method B), yellow powder, mp 239–240°C (CHCl₃). IR (DRA): v 3117, 3007 (N–H), 2959 (O–H), 1719 (CO₂H), 1595, 1578 (C=O), 1513, 1433, 1363, 1285 (C–N, C=C, C–H), 1005, 981 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO) spectrum of (*E*,*Z*)-isomer mixture, 12÷1, (*Z*)-isomer: δ 2.12 (3H, s, CH₃), 2.46 (2H, t, ³J_{HH} = 6.4 Hz, CH₂, C-2); 3.53 (2H, q, ³J_{HH} = 6.4 Hz, CH₂, C-3), 5.49 (1H, s, H, C-6), 11.06 (1H, br.s, NH); (*E*)-isomer: δ 2.09 (3H, s, CH₃), 2.53 3.26 (4H, both m, 2 CH₂, C-2,3), 5.55 (1H, s, H, C-6), 8.52 (1H, br.s, NH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO) spectrum (*E*,*Z*)-isomer mixture, 12÷1, (*Z*)-isomer: δ -177.10, -166.12, -157.65, -141.55 (4F, all m). Anal. calcd. for C₁₃H₁₁F₄NO₄+/₂H₂O: C, 47.28; H, 3.66; N, 4.24; found: C, 47.35; H, 3.51; N, 4.12.

4.3.7. N-[1-Methyl-3-oxo-3-(3,4,5-trifluoro-2-hydroxyphenyl)prop-1-enyl]-4aminobutanoic acid (**11a**) was synthesized from 256 mg (1 mmol) of chromone **4a** and 206 mg (2 mmol) of GABA. Yield 35 mg (11% *method A*), 171 mg (54% *method B*), yellow powder, mp 160–161°C (CHCl₃, EtOH_{aq}). IR (DRA): v 3066, 3011 (N–H), 2945 (O–H), 1704 (CO₂H), 1590, 1563 (C=O), 1506, 1445, 1349, 1247 (C–N, C=C, C–H), 1003 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO) spectrum of (E,Z)-isomer mixture, 5÷1, (Z)-isomer: δ 1.81 (2H, qnt, ${}^{3}J_{HH} = 7.2$ Hz, CH₂, C-3), 2.16 (3H, s, CH₃), 2.32 (2H, m, CH₂, C-2), 3.42 (2H, m, CH₂, C-4), 5.90 (1H, s, H, C-7), 7.79 (1H, d.d.d, $J_{HF} = 11.6$, 9.0, 2.0 Hz, H, C-10), 10.83 (1H, t, ${}^{3}J_{HH} = 6.1$ Hz, NH), 12.24, 14.69 (2H, both br.s, 2 OH); (*E*)-isomer: δ 1.81 (2H, m, CH₂, C-3), 2.42 (3H, s, CH₃), 2.37, 3.29 (4H, both m, CH₂, C-2, C-4), 5.76 (1H, s, H, C-7), 7.94 (1H, m, H, C-10), 8.32 (1H, m, NH), 12.24 (1H, br.s, OH). ¹³C NMR (125.76 MHz, (CD₃)₂SO) spectrum of (*Z*)-isomer: δ 19.13 (s, C-5), 24.74 (s, C-3), 30.69 (s, C-2), 42.59 (s, C-4), 90.79 (s, C-7), 109.37 (m, C-10), 115.96 (m, C-9), 139.66 (m, C-13), 141.72 (m, C-11), 141.99 (m, C-12), 148.19 (m, C-14), 168.73 (s, C-6), 173.74 (s, C-1), 184.85 (br.s, C-8). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO) spectrum of (*E*,*Z*)-isomer mixture, 5÷1, (*Z*)-isomer: δ -159.30, -154.80, -149.38 (3F, all m); (*E*)-isomer: δ -159.97, -154.93, -150.10 (3F, all m). Anal. calcd. for C₁₄H₁₄F₃NO₄: C, 53.00; H, 4.45; N, 4.41; found: C, 53.05; H, 4.38; N, 4.58.



4.3.8. N-[1-Methyl-3-oxo-3-(2,3,4,5-tetrafluoro-6-hydroxyphenyl)prop-1-enyl]-4-aminobutanoic acid (11b) was synthesized from 274 mg (1 mmol) chromone 4b and 206 mg (2 mmol) of GABA. Yield 60 mg (18% method A), 157 mg (47% method B), yellow powder, mp 161–163°C (CHCl₃). IR (DRA): v 3012 (N–H), 2944 (O–H), 1712 (CO₂H), 1602, 1567 (C=O), 1509, 1450, 1340, 1245 (C–N, C=C, C–H), 998, 969 (C–F) cm^{-1. 1}H NMR (400.13 MHz, (CD₃)₂SO) spectrum of (*E*,*Z*)-isomer mixture, 5÷1, (*Z* $)-isomer: <math>\delta$ 1.82 (2H, qnt, ³*J*_{HH} = 7.3 Hz, CH₂, C-3), 2.16 (3H, s, CH₃), 2.32 (2H, t, ³*J*_{HH} = 7.3 Hz, CH₂, C-2), 3.45 (2H, m, CH₂, C-4), 5.59 (1H, s, H, C-7), 11.05 (1H, br.s, NH), 12.44, 14.07 (2H, both br.s, 2 OH); (*E*)-isomer: δ 1.83, 2.33 (4H, both m, CH₂, C-3, C-2), 2.43 (3H, s, CH₃), 3.18 (2H, m, CH₂, C-4), 5.61 (1H, s, H, C-7), 8.48 (1H, m, NH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO) spectrum of (*E*,*Z*)-isomer mixture, 5÷1, (*Z*)-isomer: δ -173.53, -165.46, -155.20, -139.06 (4F, all m). Anal. calcd. for C₁₄H₁₃F₄NO₄: C, 50.16; H, 3.91; N, 4.18; found: C, 50.18; H, 4.09; N, 4.03.

4.3.9. *N*-[3-Oxo-1-phenyl-3-(3,4,5-trifluoro-2-hydroxyphenyl)prop-1-enyl]-3aminopropanoic acid, hydrochloride hydrate (**12a**) was synthesized from 380 mg (1 mmol) flavone **5a** and 178 mg (2 mmol) β-alanine. Yield 59 mg (14% method A), yellow powder, mp 198–200°C (CHCl₃). IR (DRA): v 3002, 2928 (O–H), 1698 (CO₂H), 1565 (C=O), 1512, 1447, 1360, 1263 (C–N, C=C, C–H), 1005, 998 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO): δ 2.57 (2H, t, ³J_{HH} = 6.2 Hz, CH₂, C-2), 3.42 (2H, m, CH₂, C-3), 5.86 (1H, s, H, C-5), 7.54 (5H, m, Ph), 7.92 (1H, m, H C-8), 11.09 (1H, t, ³J_{HH} = 5.2, NH), 14.18 (1H, br.s, OH). ¹³C NMR (125.76 MHz, (CD₃)₂SO): δ 34.34 (s, C-2), 40.91 (s, C-3), 91.97 (s, C-5), 110.15 (m, C-7), 116.25 (m, C-8), 127.73 (s, C-14), 128.65 (s, C-15), 130.08 (s, C-16), 134.23 (s, C-13), 139.72 (m, C-11), 141.83 (m, C-9), 142.35 (m, C-10), 148.03 (m, C-12), 168.59 (s, C-4), 172.59 (s, C-1), 186.30 (m, C-6). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO): δ -159.06, -154.09, -149.07 (3F, all m). Anal. calcd. for C₁₈H₁₄F₃NO₄+HCl·H₂O: C, 51.50; H, 4.08; N, 3.34; found: C, 51.27; H, 3.59; N, 3.61.



4.3.10. N-[3-Oxo-1-phenyl-3-(2,3,4,5-tetrafluoro-6-hydroxyphenyl)prop-1-enyl]-3-aminopropanoic acid (**12b**) was synthesized from 398 mg (1 mmol) flavone**5b** $and 178 mg (2 mmol) <math>\beta$ -alanine. Yield 45 mg (11% *method A*), yellow powder, mp 210–211°C (CHCl₃). IR (DRA): v 3964 (N–H), 2927 (O–H), 1699 (CO₂H), 1603, 1582 (C=O), 1516, 1451, 1355 (C–N, C=C, C–H), 996 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CDCl₃): δ 2.59 (2H, t, ³J_{HH} = 6.2 Hz, CH₂, C-2), 3.45 (2H, m, CH₂, C-3), 5.53 (1H, s, H C-5), 7.51–7.55 (5H, m, Ph), 11.17 (1H, br.s, NH), 12.51, 13.73 (2H, both br.s, 2 OH). ¹⁹F NMR (376.44 MHz, (CDCl₃): δ -172.35, -164.31, -154.42, -140.09 (4F, all m). Anal. calcd. for C₁₈H₁₃F₄NO₄: C, 56.40; H, 3.42; N, 3.65; found: C, 56.35; H, 3.61; N, 3.36.

4.3.11. N-[3-Oxo-1-phenyl-3-(3,4,5-trifluoro-2-hydroxyphenyl)prop-1-enyl]-4-aminobutanoic acid (**13a**) was synthesized from 380 mg (1 mmol) flavone**5a**and 206 mg (2 mmol) of GABA. Yield 76 mg (20%*method A* $), yellow powder, mp 175–178°C (CHCl₃, EtOH_{aq}). IR (DRA): v 2952 (O–H), 1704 (CO₂H), 1586, 1564 (C=O), 1511, 1448, 1368 (C–N, C=C, C–H), 1025, 1008 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO): <math>\delta$ 1.75 (2H, qnt, ³J_{HH} = 7.1 Hz, CH₂, C-3), 2.21 (2H, t, ³J_{HH} = 7.1 Hz, CH₂, C-2), 3.30 (2H, m, CH₂, C-4), 5.88 (1H, s, H, C-6), 7.51–7.55 (5H, m, Ph), 7.92 (1H, m, H, C-9), 11.00 (1H, t, ³J_{HH} = 6.1 Hz, NH), 12.13, 14.27 (2H, both br.s, 2 OH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -159.04, -154.13, -149.04 (3F, all m). Anal. calcd. for C₁₉H₁₆F₃NO4: C, 60.16; H, 4.25; N, 3.69; found: C, 60.32; H, 4.57; N, 3.45.



4.3.12. N-[3-Oxo-1-phenyl-3-(2,3,4,5-tetrafluoro-6-hydroxyphenyl)prop-1-enyl]-4-aminobutanoic acid, hydrate (13b) was synthesized from 398 mg (1 mmol) of flavone**5b**and 206 mg (2 mmol) of GABA. Yield 67 mg (17%*method A* $), yellow powder, mp 159–161°C (CHCl₃). IR (DRA): v 3172 (N–H), 2955 (O–H), 1737, 1715 (CO₂H), 1652, 1583 (C=O), 1515, 1452, 1349, 1288 (C–N, C=C, C–H), 997 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO): <math>\delta$ 1.76 (2H, qnt, ³J_{HH} = 7.2 Hz, CH₂, C-3), 2.22 (2H, t, ³J_{HH} = 7.2 Hz, CH₂, C-2), 3.32 (2H, m, CH₂, C-4), 5.54 (1H, s, H, C-6), 7.48–7.55 (5H, m, Ph), 11.08 (1H, t, ³J_{HH} = 5.4 Hz, NH), 12.14, 13.65 (2H, both br.s, 2 OH). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO): δ -172.36, -164.30, -154.54, -140.16 (4F, all m). Anal. calcd. for C₁₉H₁₅F₄NO₄·H₂O: C, 54.94; H, 4.13; N, 3.37; found: C, 55.59; H, 3.95; N, 3.56.

4.3.13. N-(6, 8-Difluoro-4-oxo-2-phenyl-4H-chromen-7-yl)-3-aminopropanoic acid (14a) was synthesized according to *method B* from 380 mg (1 mmol) flavone **5a** and 178 mg (2 mmol) β -alanine. Yield 197 mg (57%), yellow powder, mp 235–238°C (CHCl₃, EtOH_{aq}). IR (DRA): v 3292 (N–H), 2965, 2921 (O–H), 1712 (CO₂H), 1641, 1632 (C=O), 1547, 1471, 1409, 1370 (C–N, C=C, C–H), 1017 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO): δ 2.59 (2H, t, ³J_{HH} = 7.1 Hz, CH₂, C-10), 3.67 (2H, q, ³J_{HH} = 7.1 Hz, CH₂, C-11), 6.43 (1H, br.s, NH), 6.95 (1H, s, H, C-3), 7.44 (1H, d.d, J_{HF} = 12.0, 1.5 Hz, H, C-5), 7.58–8.04 (5H, m, Ph), 12.31 (1H, br.s, OH). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO): δ -154.65, -130.48 (2F, both m). Anal. calcd. for C₁₈H₁₃F₂NO₄: C, 62.61; H, 3.79; N, 4.06; found: C, 62.83; H, 3.59; N, 4.00.



4.3.14. N-(5,6,8-Trifluoro-4-oxo-2-phenyl-4H-chromen-7-yl)-3aminopropanoic acid (**14b**) in a mixture with **14c** was synthesized according to method B from 398 mg (1 mmol) of flavone **5b** and 356 mg (4 mmol) β -alanine. ¹H NMR (400.13 MHz, (CD₃)₂SO): δ 2.58, 3.67 (4H, both m, 2 CH₂, C-10, C-11), 6.77 (1H, br.s, NH), 6.86 (1H, s, H, C-3), 7.56–8.03 (5H, m, Ph), 12.32 (1H, br.s, OH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -160.00, -159.68, -148.52 (3F, all m). HRMS (ESI), m/z: calcd. for [C₁₈H₁₁F₃NO₄]⁻ 362.0646; found 362.0639.

4.3.15. N-(6,8-Difluoro-5-hydroxy-4-oxo-2-phenyl-4H-chromen-7-yl)-3aminopropanoic acid (**14c**) in a mixture with **14b** was synthesized according to method B from 398 mg (1 mmol) of flavone **5b** and 356 mg (4 mmol) β-alanine. ¹H NMR (400.13 MHz, (CD₃)₂SO): δ 2.66, 3.68 (4H, both m, 2 CH₂, C-10, C-11), 6.62 (1H, br.s, NH), 6.99 (1H, s, H, C-3), 7.56–8.03 (5H, m, Ph), 12.41 (1H, br.s, OH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -167.35, -161.84 (2F, both m). HRMS (ESI), m/z: calcd. for [C₁₈H₁₂F₂NO₅]⁻: 360.0689; found 360.0684.

4.3.16. N-(6,8-Difluoro-4-oxo-2-phenyl-4H-chromen-7-yl)-4-aminobutanoic acid, hemihydrate (15a) was synthesized according to method B from 380 mg (1 mmol) flavone**5a** $and 206 mg (2 mmol) GABA. Yield 182 mg (50%), yellow powder, mp 204–207°C (CHCl₃, EtOH_{aq}). IR (DRA): v 3296 (N–H), 2977, 2941 (O–H), 1721 (CO₂H), 1641, 1622 (C=O), 1545, 1480, 1370 (C–N, C=C, C–H), 1022 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO): <math>\delta$ 1.83 (2H, qnt, ³J_{HH} = 7.1 Hz, CH₂, C-11), 2.32 (2H, t, ³J_{HH} = 7.1 Hz, CH₂, C-10), 3.45 (2H, q, ³J_{HH} = 6.1 Hz, CH₂, C-12), 6.54 (1H, br.s, NH), 6.93 (1H, s, H, C-3), 7.40 (1H, d.d, J_{HF} = 11.9, 1.6 Hz, H, C-5), 7.58–8.02 (5H, m, Ph), 12.11 (1H, s, OH). ¹³C NMR (125.76 MHz, (CD₃)₂SO): δ 25.65 (s, C-11), 30.87 (s, C-10), 43.60 (m, C-12), 104.18 (m, C-5), 106.33 (s, C-3), 111.34 (m, C-4a), 125.99 (s, C-14), 129.12 (s, C-15), 131.00 (s, C-13), 131.41 (m, C-7), 131.63 (s, C-16), 137.86 (m, C-8a), 143.06 (m, C-8), 149.02 (m, C-6), 161.31 (s, C-2), 174.22 (s, C-9), 174.93 (m, C-4). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO): δ -154.96, -130.44 (2F, both m). Anal. calcd. for C₁₉H₁₅F₂NO₄⁴/₂H₂O: C, 61.96; H, 4.38; N, 3.80; found: C, 62.41; H, 4.27; N, 3.95.



4.3.17. N-(5,6,8-Trifluoro-4-oxo-2-phenyl-4H-chromen-7-yl)-4-aminobutanoic acid (**15b**) in a mixture with **15c** was synthesized according to *method B* from 398 mg (1 mmol) of flavone **5b** and 412 mg (4 mmol) of GABA. ¹H NMR (400.13 MHz, (CD₃)₂SO): δ 1.83, 2.31, 3.45 (6H, all m, 3 CH₂, C-10, C-11, C-12), 6.85 (1H, s, H, C-3), 6.92 (1H, br.s, NH), 7.55–8.04 (5H, m, Ph), 12.32 (1H, br.s, OH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -160.25, -159.81, -148.58 (3F, all m). HRMS (ESI), *m/z*: calcd. for [C₁₉H₁₃F₃NO₄]⁻: 376.0802; found 376.0798.

4.3.18. N-(6,8-Difluoro-5-hydroxy-4-oxo-2-phenyl-4H-chromen-7-yl)-4-aminobutanoic acid (15c) in a mixture with 15b was synthesized according to*method B* $from 398 mg (1 mmol) of flavone 5b and 412 mg (4 mmol) of GABA. ¹H NMR (400.13 MHz, (CD₃)₂SO): <math>\delta$ 1.83, 2.28, 3.42 (2H, m, CH₂, C-10, C-11, C-12), 6.78 (1H, br.s, NH), 6.98 (1H, s, H, C-3), 7.55–8.04 (5H, m, Ph). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -167.46, -161.85 (2F, both m). HRMS (ESI), *m/z*: calcd. for [C₁₉H₁₄F₂NO₅]⁻: 374.0846; found 374.0840.

4.3.19. 3-{[2-(3,4-Dihydroxyphenyl)ethyl]amino}-1-(3,4,5-trifluoro-2hydroxyphenyl)but-2-en-1-one (**16a**) was synthesized according to *method* A from 256 mg (1 mmol) of chromone 4a, 379 mg (2 mmol) of dopamine hydrochloride and 168 mg (2 mmol) NaHCO₃. Yield 180 mg (49%), yellow powder, mp 154–156°C (CHCl₃/C₆H₁₄). IR (DRA): v 3458 (N–H), 2930, 2870 (O–H), 1595 (C=O), 1509, 1437, 1370 (C–N, C=C, C–H), 1001 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO+CCl₄) spectrum of (*E*,*Z*)-isomer mixture, $17 \div 1$, (*Z*)-isomer: δ 2.04 (3H, s, CH₃), 2.74 (2H, t, ${}^{3}J_{HH} = 7.0$ Hz, CH₂, C-12), 3.56 (2H, q, ${}^{3}J_{HH} = 6.9$ Hz, CH₂, C-11), 5.72 (1H, s, H, C-2), 6.47–6.64 (3H, m, C₆H₃), 7.59 (1H, m, H, C-6), 8.51, 8.55 (2H, both s, 2 OH), 10.85 (1H, t, ${}^{3}J_{HH} = 5.1$ Hz, NH), 14.45 (1H, s, OH); (*E*)-isomer: δ 2.66, 3.42 (4H, m, 2 CH₂, C-11, C-12), 5.60 (1H, s, H, C-2), 6.47–6.64 (3H, m, C₆H₃), 7.73 (1H, m, H, C-6), 8.07 (1H, m, NH), 8.46, 8.52 (2H, both s, 2 OH), 15.49 (1H, s, OH). ¹³C NMR (125.76 MHz, (CD₃)₂SO+CCl₄) spectrum of (Z)-isomer: δ 19.16 (s, C-4), 34.65 (s, C-12), 45.11 (s, C-11), 90.61 (s, C-2), 109.32 (m, C-6), 115.51 (s, C-14), 115.93 (m, C-5), 116.18 (s, C-17), 119.44 (s, C-18), 128.98 (s, C-13), 139.70 (m, C-9), 141.63 (m, C-7), 141.95 (m, C-8), 143.79 (s, C-16), 145.16 (s, C-15), 148.27 (m, C-10), 168.55 (s, C-3), 184.71 (s, C-1). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO+CCl₄) spectrum of (E,Z)-isomer mixture, 17÷1, (Z)-isomer: δ -159.54 (1F, d, J = 19.5 Hz), -155.34, -149.96 (2F, both m); (E)-isomer: δ -160.19, -155.52, -150.75 (3F, all m). Anal. calcd. for C₁₈H₁₆F₃NO₄: C, 58.86; H, 4.39; N, 3.81; found: C, 58.24; H, 4.59; N, 3.91.



4.3.20. $3 - \{[2 - (3, 4 - Dihydroxyphenyl)ethyl]amino\} - 1 - (2, 3, 4, 5 - tetrafluoro - 6-hydroxyphenyl)but - 2 - en - 1 - one ($ **16b**) was synthesized according to*method*A from 274 mg (1 mmol) of chromone**4b**, 379 mg (2 mmol) of dopamine hydrochloride and 168 mg (2 mmol) NaHCO₃. Yield 193 mg (50%), yellow powder, mp 177–179°C (CHCl₃/C₆H₁₄). IR (DRA): v 3405 (N–H), 2954, 2925 (O–H), 1656, 1595 (C=O), 1511, 1446, 1340 (C–N, C=C, C–H), 1004 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO+CCl₄) spectrum of (*E*,*Z*)-isomer mixture, 14÷1, (*Z* $)-isomer: <math>\delta$ 2.07 (3H, s, CH₃), 2.75 (2H, t, ³J_{HH} = 7.2 Hz, CH₂, C-12), 3.59 (2H, q, ³J_{HH} = 7.0 Hz, CH₂, C-11), 5.58 (1H, s, H, C-2), 6.47–6.64 (3H, m, C₆H₃), 8.54, 8.57 (2H, both s, 2 OH), 11.08 (1H, t, ³J_{HH} = 5.5 Hz, NH), 14.60 (1H, s, OH); (*E*)-isomer: δ 2.44 (3H, s, CH₃), 2.72, 3.30 (4H, m, 2 CH₂, C-11, C-12), 5.69 (1H, s, H, C-2), 6.45–6.59 (3H, m, C₆H₃), 7.68 (1H, m, NH), 8.47 (2H, br.s, 2 OH), 15.50 (1H, s, OH). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO+CCl₄) spectrum of (*E*,*Z*)-isomer mixture, 14÷1, (*Z*)-isomer: δ -173.91, -165.71, -155.01, -139.27 (4F, all m); (*E*)-isomer: δ -174.50, -166.12, -155.41, -138.94 (4F, all m). Anal. calcd. for C₁₈H₁₅F₄NO₄: C, 56.11; H, 3.92; N, 3.64; found: C, 56.04; H, 4.05; N, 3.80.

4.3.21. $3 - \{[2-(3,4-Dihydroxyphenyl)ethyl]amino\} - 3-phenyl-1-(3,4,5-trifluoro-2-hydroxyphenyl)prop-2-en-1-one (17a) was synthesized according to method B in inert atmosphere of argon, using EtOH as solvent from 380 mg (1 mmol) of flavone$ **5a** $379 mg (2 mmol) of dopamine hydrochloride and 202 mg (2 mmol) of Et₃N. The reaction vessel was filled with argon, dopamine hydrochloride 379 mg (2 mmol) was added to a mixture. Yield 159 mg (37%), yellow powder, mp 159–161°C (column chromatography on silica gel, using CHCl₃). IR (DRA): v 3535, 3223 (N–H), 2949, 2935 (O–H), 1590 (C=O), 1511, 1441, 1269 (C–N, C=C, C–H), 1021, 1003 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO): <math>\delta$ 2.64 (2H, t, ³J_{HH} = 7.0 Hz, CH₂, C-15), 3.39 (2H, q, ³J_{HH} = 6.7 Hz, CH₂, C-14), 5.83 (1H, s, H, C-2), 6.35–6.62 (3H, m, C₆H₃), 7.40–7.54 (5H, m, Ph), 7.89 (1H, m, H, C-9), 8.72, 8.75 (2H, both s, 2 OH), 10.99 (1H, t, ³J_{HH} = 6.0 Hz, NH), 14.35 (1H, s, OH). ¹³C NMR (125.76 MHz, (CD₃)₂SO): δ 35.31 (s, C-15), 46.75 (s, C-14), 91.68 (s, C-2), 110.02 (m, C-8), 115.47 (s, C-17), 116.05 (s, C-21), 116.21 (m, C-9), 119.32 (s, C-22), 127.53 (s, C-5), 128.49 (s, C-6), 128.71 (s, C-7), 129.94 (s, C-16), 134.25 (s, C-4),

139.37 (m, C-12), 141.74 (m, C-10), 142.42 (m, C-11), 143.80 (s, C-20), 145.13 (s, C-19), 148.02 (m, C-13), 168.56 (s, C-3), 186.09 (m, C-1). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -159.17, -154.22, -149.12 (3F, all m). Anal. calcd. for C₂₃H₁₈F₃NO₄: C, 64.34; H, 4.23; N, 3.26; found: C, 64.52; H, 4.57; N, 3.44.



4.3.22. $3-\{[2-(3,4-Dihydroxyphenyl)ethyl]amino\}-3-phenyl-1-(2,3,4,5-tetrafluoro-6-hydroxyphenyl)prop-2-en-1-one ($ **17b**) was synthesized according to*method B*in inert atmosphere of argon, using EtOH as solvent from 398 mg (1 mmol) of flavone**5b** $379 mg (2 mmol) of dopamine hydrochloride and 202 mg (2 mmol) of Et₃N. The reaction vessel was filled with argon, dopamine hydrochloride 379 mg (2 mmol) was added to a mixture. Yield 170 mg (38%), yellow powder, mp 163–166°C (column chromatography on silica gel, using CHCl₃). IR (DRA): v 3443, 3378 (N–H), 2966, 2922 (O–H), 1653 (C=O), 1515, 1463, 1347 (C-N, C=C, C–H), 1002 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO): <math>\delta$ 2.66 (2H, t, ³J_{HH} = 7.0 Hz, CH₂, C-15), 3.41 (2H, q, ³J_{HH} = 6.8 Hz, CH₂, C-14), 5.51 (1H, s, H, C-2), 6.35–6.63 (3H, m, C₆H₃), 7.40–7.55 (5H, m, Ph), 8.73, 8.76 (2H, both s, 2 OH), 11.07 (1H, t, ³J_{HH} = 5.8 Hz, NH), 13.88 (1H, br.s, OH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -172.46, -164.49, -154.45, -139.97 (4F, all m). Anal. calcd. for C₂₃H₁₇F₄NO₄: C, 61.75; H, 3.83; N, 3.13; found: C, 61.75; H, 3.75; N, 3.03.

4.3.23. 3-(*Pyrrolidin-1-yl*)-1-(3,4,5-trifluoro-2-hydroxyphenyl)but-2-en-1-one (**18a**) was synthesized according to *method C* from 256 mg (1 mmol) of chromone **4a** and 142 mg (2 mmol) of pyrrolidine, using EtOH as solvent. Yield 165 mg (58%), yellow powder, mp 174–177°C (EtOH). IR (DRA): v 2984, 2878 (O–H), 1641, 1539 (C=O), 1445, 1341, 1227 (C–N, C=C, C–H), 1022, 983 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO+CCl₄): δ 2.00 (4H, m, (CH₂)₂, C-6), 2.59 (3H, s, CH₃), 3.56 (4H, m, (CH₂)₂, C-5), 5.51 (1H, s, H, C-2), 7.67 (1H, d.d.d, *J*_{HF} = 11.6, 9.0, 2.2 Hz, H, C-8), 15.47 (1H, s, OH). ¹³C NMR (125.76 MHz, (CD₃)₂SO): δ 18.54 (s, C-4), 24.25, 24.49 (both s, C-6), 48.90, 49.45 (both s, C-5), 89.66 (s, C-2), 109.69 (m, C-8), 117.08 (m, C-7), 139.90 (m, C-11), 141.44 (m, C-9), 142.40 (m, C-10), 149.42 (m, C-12), 165.21 (s, C-3), 185.30 (s, C-1). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO+CCl₄): δ -160.10, -155.65, -150.81 (3F, all m). Anal. calcd. for C₁₄H₁₄F₃NO₂: C, 58.95; H, 4.95; N, 4.91; found: C, 58.83; H, 5.12; N, 4.90.



4.3.24. 3-(*Pyrrolidin-1-yl*)-1-(2,3,4,5-tetrafluoro-6-hydroxyphenyl)but-2-en-1one (**18b**) was synthesized according to method *C* from 274 mg (1 mmol) chromone **4b** and 142 mg (2 mmol) of pyrrolidine, using EtOH as solvent. Yield 200 mg (66%), yellow powder, mp 144–145°C (EtOH). IR (DRA): v 2984, 2881 (O–H), 1650 (C=O), 1449, 1342, 1267 (C–N, C=C, C–H), 1000, 981 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO): δ 1.94 (4H, m, (CH₂)₂, C-6), 2.61 (3H, s, CH₃), 3.50 (4H, m, (CH₂)₂, C-5), 5.52 (1H, s H, C-2), 15.31 (1H, br.s, OH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -173.84, -165.61, -155.58, -139.22 (4F, all m). Anal. calcd. for C₁₄H₁₃F₄NO₂: C, 55.45; H, 4.32; N, 4.62; found: C, 55.67; H, 4.56; N, 4.52. 4.3.25. 1-[3, 5-Difluoro-2-hydroxy-4-(pyrrolidin-1-yl)phenyl]-3-(pyrrolidin-1-yl)but-2-en-1-one (19a) was synthesized according to *method C* from 256 mg (1 mmol) of chromone 4a and 284 mg (4 mmol) of pyrrolidine, using Me₂SO as solvent. Yield 296 mg (88%), yellow powder, mp 196–198°C (EtOH). IR (DRA): v 2972, 2890 (O–H), 1628 (C=O), 1442, 1355, 1216 (C–N, C=C, C–H), 1030, 1010 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO+CCl₄): δ 1.94 (8H, m, (CH₂)₄, C-6), 2.56 (3H, s, CH₃), 3.53 (8H, m, (CH₂)₄, C-5), 5.40 (1H, s, H, C-2), 7.17 (1H, d.d, *J*_{HF} = 15.5, 1.5 Hz, H, C-8), 14.86 (1H, br.s, OH). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO+CCl₄): δ -152.61, -136.65 (2F, both m). Anal. calcd. for C₁₈H₂₂F₂N₂O₂: C, 64.27; H, 6.59; N, 8.33; found: C, 64.53; H, 6.60; N, 8.22.

4.3.26. 3-(Pyrrolidin-1-yl)-1-[2,3,5-trifluoro-6-hydroxy-4-(pyrrolidin-1-yl)phenyl]but-2-en-1-one (**19b**) was synthesized according to *method C* from 274 mg (1 mmol) of chromone **4b** and 284 mg (4 mmol) of pyrrolidine, using Me₂SO as solvent. Yield 209 mg (59%), yellow powder, mp 178–180°C (EtOH). IR (DRA): v 2979, 2885 (O–H), 1641, 1618 (C=O), 1441, 1334, 1248 (C–N, C=C, C–H), 1027, 990 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO+CCl₄): δ 1.95 (8H, m, (CH₂)₄, C-6), 2.59 (3H, s, CH₃), 3.52 (8H, m, (CH₂)₄, C-5), 5.57 (1H, s H, C-2), 15.25 (1H, s, OH). ¹³C NMR (125.76 MHz, (CD₃)₂SO): δ 18.87 (s, C-4), 24.32, 24.52 (both s, C-6), 25.14 (s, C-6'), 48.62, 48.97 (both s, C-5), 50.99 (t, *J* = 6.3 Hz, C-5'), 94.62 (d, *J* = 18.5 Hz, C-2), 100.76 (d, *J* = 11.8 Hz, C-7), 130.13 (m, C-11), 133.73 (m, C-9), 136.64 (m, C-8), 146.51 (m, C-10), 148.61 (d.d, *J* = 12.6, 7.6 Hz, C-12), 164.25 (s, C-3), 183.22 (m, C-1). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO+CCl₄): δ -165.80, -159.33, -140.89 (3F, all m). Anal. calcd. for C₁₈H₂₁F₃N₂O₂: C, 61.01; H, 5.97; N, 7.91; found: C, 61.11; H, 6.04; N, 7.87.

4.3.27. 1-(6,8-Difluoro-2-methyl-4-oxo-4H-chromen-7-yl)proline (**20a**) was synthesized according to *method B* from 256 mg (1 mmol) of chromone **4a** and 230 mg (2 mmol) of proline. Yield 105 mg (34%), white powder, mp 221–223°C (CHCl₃, EtOH_{aq}). IR (DRA): v 2964, 2892 (O–H), 1742 (CO₂H), 1639, 1616 (C=O), 1524, 1471, 1354, 1314 (C–N, C=C, C–H), 1035 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO): δ 1.92 (4H, m, (CH₂)₂, C-10, C-11), 2.37 (3H, s, CH₃), 3.75 (2H, m, CH₂, C-12), 4.70 (1H, m, H, C-9), 6.16 (1H, s, H, C-3); 7.36 (1H, d.d, J_{HF} = 13.9, 2.0 Hz, H, C-5), 12.70 (1H, br.s, OH). ¹³C NMR (125.76 MHz, (CD₃)₂SO): δ 19.80 (s, C-14), 24.10 (s, C-11), 30.41(s, C-10), 52.20 (t, J = 6.4 Hz, C-12), 62.45 (t, J = 5.6 Hz, C-9), 104.94 (m, C-5), 109.30 (s, C-3), 112.93 (d, J = 8.1 Hz, C-4a), 130.31 (d.d, J = 14.6, 8.1 Hz, C-7), 139.92 (m, C-8), 112.93 (d, J = 11.2 Hz, C-8a), 150.13 (m, C-6), 165.87 (s, C-2), 173.61 (s, C-4), 174.55 (br.s, C-13). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -147.05, -124.19 (2F, both m). Anal. calcd. for C₁₅H₁₃F₂NO₄: C, 58.26; H, 4.24; N, 4.53; found: C, 58.10; H, 4.25; N, 4.50.



4.3.28. 1-(5,6,8-Trifluoro-2-methyl-4-oxo-4H-chromen-7-yl)proline (**20b**) was synthesized according to method C from 274 mg (1 mmol) of chromone **4b** and 230 mg (2 mmol) of proline. Yield 128 mg (39%), yellow powder, mp 230–232°C (CHCl₃, EtOH_{aq}). IR (DRA): v 3061 (O–H), 1693 (CO₂H), 1625 (C=O), 1521, 1480, 1353 (C-N, C=C, C–H), 1036, 1024 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO): δ 1.93 (4H, m, (CH₂)₂, C-10, C-11), 2.33 (3H, s, CH₃), 3.78 (2H, m, CH₂, C-12), 4.71 (1H, m, H, C-9), 6.07 (1H, s, H, C-3), 12.80 (1H, br.s, OH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -154.91, -153.41, -148.08 (3F, all m). Anal. calcd. for C₁₅H₁₂F₃NO₄: C, 55.05; H, 3.70; N, 4.28; found: C, 55.54; H, 4.05; N, 4.01.

4.3.29. 3-Benzoyl-6,8-difluoro-2-phenyl-7-(pyrrolidin-1-yl)-4H-chromen-4-one (21a) was synthesized according to method C from 380 mg (1 mmol) of flavone **5a** and 142 mg (2 mmol) of pyrrolidine, using EtOH as solvent. Yield 169 mg (39%), yellow powder, mp 210–

214°C (EtOH). IR (DRA): v 2978, 2883 (O–H), 1668, 1615 (C=O), 1479, 1447, 1349 (C–N, C=C, C–H), 1036, 1025 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO): δ 1.92 (4H, m, (CH₂)₂, C-10), 3.72 (4H, m, (CH₂)₂, C-9), 7.62 (1H, m, H, C-5), 7.42–7.92 (10H, m, 2 Ph). ¹³C NMR (125.76 MHz, (CD₃)₂SO): δ 25.32 (s, C-10), 51.34 (s, C-9), 105.35 (m, C-5), 111.39 (m, C-4a), 121.35 (s, C-3), 128.17 (s, C-12), 128.85 (s, C-17), 128.94 (s, C-13), 129.13 (s, C-14), 131.30 (s, C-11), 131.48 (s, C-14), 134.08 (s, C-19), 136.46 (s, C-16), 138.34–154.40 (all m, C-6, C-7, C-8, C-8a), 160.56 (br.s, C-2), 173.39 (br.s, C-4), 192.97 (s, C-15). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO): δ - 149.19, -123.75 (2F, both m). Anal. calcd. for C₂₆H₁₉F₂NO₃: C, 72.38; H, 4.44; N, 3.25; found: C, 72.96; H, 4.45; N, 3.35.



4.3.30. 3-Benzoyl-5,6,8-trifluoro-2-phenyl-7-pyrrolidin-1-yl-4H-chromen-4one (21b) was synthesized according to method C from 398 mg (1 mmol) of flavone **5b** and 142 mg (2 mmol) of pyrrolidine, using EtOH as solvent. Yield 153 mg (34%), yellow powder, mp 224–227°C (EtOH). IR (DRA): v 2980, 2886 (O–H), 1669, 1648 (C=O), 1494, 1448, 1352 (C–N, C=C, C–H), 1003 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO+CCl₄): δ 1.98 (4H, m, (CH₂)₂, C-10), 3.77 (4H, m, (CH₂)₂, C-9), 7.37–7.89 (10H, m, 2 Ph). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO+CCl₄): δ -155.68, -155.06 (2F, all m), -146.99 (1F, d.d, *J* = 17.7, 9.2 Hz). Anal. calcd. for C₂₆H₁₈F₃NO₃: C, 69.49; H, 4.04; N, 3.12; found: C, 69.71; H, 4.25; N, 3.02.

4.3.31. 1-(3-Benzoyl-6, 8-difluoro-4-oxo-2-phenyl-4H-chromen-7-yl)proline(22a) was synthesized according to *method B* from 380 mg (1 mmol) of flavone **5a** and 230 mg (2 mmol) of proline. Yield 171 mg (35%), yellow powder, mp 230–232°C (CHCl₃). IR (DRA): v 2979 (O–H), 1736 (CO₂H), 1674 (C=O), 1522, 1471, 1394 (C–N, C=C, C–H), 1038, 1019 (C–F) cm⁻¹. ¹H NMR (500.13 MHz, (CD₃)₂SO): δ 2.08 (4H, m, (CH₂)₂, C-10, C-11), 3.82 (2H, m, CH₂, C-12), 4.78 (1H, m, H, C-9), 7.42–7.94 (10H, m, 2 Ph), 7.62 (1H, m, H, C-5), 12.79 (1H, br.s, OH). ¹⁹F NMR (470.52 MHz, (CD₃)₂SO): δ -147.65 (1F, d, *J* = 11.9 Hz), -123.14 (1F, m). Anal. calcd. for C₂₇H₁₉F₂NO₅: C, 68.21; H, 4.03; N, 2.95; found: C, 68.30; H, 4.28; N, 2.98.

4.3.32. 1 - (3 - Benzoyl - 5, 6, 8 - trifluoro - 4 - oxo - 2 - phenyl - 4H - chromen - 7 - yl)proline(22b) was synthesized according to *method B* from 398 mg (1 mmol) of flavone **5b** and 230 mg (2 mmol) of proline. Yield 199 mg (40%), yellow powder, mp 265–268°C (CHCl₃). IR (DRA): v 2983, 2884 (O–H), 1748 (CO₂H), 1674 (C=O), 1523, 1479, 1368 (C–N, C=C, C–H), 1003 (C–F) cm⁻¹. ¹H NMR (400.13 MHz, (CD₃)₂SO): δ 2.14 (4H, m, (CH₂)₂, C-10, C-11), 3.85 (2H, m, CH₂, C-12), 4.80 (1H, m, H, C-9), 7.43–7.98 (10H, m, 2 Ph), 12.90 (1H, br.s, OH). ¹⁹F NMR (376.44 MHz, (CD₃)₂SO): δ -154.11, -153.97 (2F, both m), -147.71 (1F, d.d, *J* = 18.3, 9.1 Hz). Anal. calcd. for C₂₇H₁₈F₃NO₅: C, 65.72; H, 3.68; N, 2.84; found: C, 65.91; H, 3.90; N, 3.24.

4.4. The X-ray structural analysis of compounds 10b·Na, 17a, 18b was performed on an Xcalibur 3 automatic diffractometer (Oxford Diffraction, UK) with a CCD detector (graphite monochromator, λ (MoK α) 0.71073 Å, ω -scan). Absorption was corrected analytically using a multi-facet crystal model as implemented in the program CrysAlis RED v.1.171.39.38a. The structures were solved with the direct method from differential Fourier syntheses and refined by full-matrix least-squares F2 using the software package SHELXTL.^[18] Refinement for non-hydrogen atoms was performed in the anisotropic approximation. Hydrogen atoms were placed in geometrically calculated positions and were included in the refinement according to the "rider" model in an isotropic approximation with thermal parameters dependent on the "parent" atoms.

4.4.1. Crystals of compound **10b**·Na (C₁₃H₁₈F₄NNaO₈, M 415.27) are monoclinic; spatial symmetry group P2(1)/c; a 21.106(2), b 5.1186(4), c 17.0736(19) Å; a 90.00, β 106.223(12), γ 90.00°; V 1771.1(3) Å³; Z 4; d_{calc} 1.557 g/cm³; μ 0.172 mm⁻¹. 4763 total reflections, 2109 independent reflections, *R*-factor 0.0555, 289 refinable parameters (measured at 295(2) K).

4.4.2. Crystals of compound 17a (C₂₃H₁₈F₃NO₄, M 429.38) are monoclinic; spatial symmetry group P2(1)/c; a 19.237(4), b 4.5309(8), c 23.426(5) Å; a 90.00, β 108.76(3), γ 90.00°; V 1933.4(7) Å³; Z 4; d_{calc} 1.475 g/cm³; μ 0.120 MM⁻¹. 4735 total reflections, 1736 independent reflections, *R*- factor 0.0716, 296 refinable parameters (measured at 295(2) K).

4.4.3. Crystals of compound **18b** (C₁₄H₁₃F₄NO₂, *M* 303.25) are triclinic; spatial symmetry group *P*1⁻; *a* 7.6785(16), *b* 13.215(2), *c* 13.531(3) Å; α 80.743(16), β 80.049(18), γ 75.127(17)°; V 1297.2(4) Å³; Z 2; d_{calc} 1.553 g/cm³; μ 0.142 мм⁻¹. 4394 total reflections, 1417 independent reflections, *R*- factor 0.0401, 382 refinable parameters (measured at 295(2) K).

The full set of X-ray structural data for compounds **10b**·Na, **17a**, **18b** were deposited at Cambridge Crystallographic Data Center (deposits CCDC 1901179, CCDC 1901173, CCDC 1901184).

4.5. Microbiological activity assay. Evaluation of antimicrobial properties (fungistatic and bactericidal) was performed by the serial dilution method recommended by the Institute of Clinical and Laboratory Standards (CLSI). Test strains of fungi: *Trichophyton rubrum* (RCPF F-1408), *Trichophyton mentagrophytes var. gypseum* (RCPF F-1425), *Trichophyton tonsurans* (RCPF F-1458), *Trichophiton violaceum* (RCPF F-1393/658), *Trichophyton mentagrophytes var. interdigitale* (RCPF F-1229), *Epidermophyton floccosum* (RCPF F-1174), *Microsporum canis* (RCPF F-1403), *Candida albicans* (RCPF Y-401 / NCTC 885-653) (Table 1), and test strains of clinically significant obligate and conditionally pathogenic bacteria: *Escherichia coli* (ATCC 8739), *Klebsiella pneumoniae* (ATCC 13880 / NCTC 10211), *Shigella flexneri* (1a8516), *Pseudomonas aeruginosa* (ATCC 9027 / NCTC 12924), *Staphylococcus aureus* (ATCC 25923 / NCTC 12981 (F-49)), *Staphylococcus aureus MRSA* (NCTC12493) (Table 2).

Fungi were cultivated at 27°C on Sabouraud Dextrose Agar (SDA) and MIC was determined by using Sabouraud Dextrose Broth (SDB). The tested compounds and standard were dissolved in DMSO, 1÷10 and applied in different concentrations (from 1000 till 0.19 µg/mL). DMSO was used as a negative control and antifungal (Terbinafine) as positive control. The broth dilution test was performed in test tubes. The inoculum suspension, which gave the final concentration of 1×10^8 CFU/mL, was prepared. A growth control tube and sterility control tube were used in each test. After 14 days incubation at 27°C, the MIC was determined visually as the lowest concentration that inhibits growth, evidenced by the absence of turbidity.

The Muller-Hinton medium was used for testing antibacterial activity. Adjust the density of the suspension to contain 1.5×10^8 CFU/mL by comparison with a 0.5 McFarland turbidity standard. For suspension, using colonies from an overnight (20 to 24-hour) agar plate incubated in 5% CO₂ $36\pm1^{\circ}$ C. Dilute this suspension $1\div100$ in Muller-Hinton to give 10^{6} CFU/mL. The test compounds concentrations: 1000, 500, 250, 125, 62.5, 31.25, 15.6, 7.8, 3.9, 1.9, 0.97 µg/mL (solvent – DMSO, diluent – H₂O and GC agar base).

References

[1] (a) J. Reis, A. Gaspar, N. Milhazes, F. Borges, Chromone as a privileged scaffold in drug discovery: recent advances, J. Med. Chem. 60 (2017) 7941–7957. https://doi.org/10.1021/acs.jmedchem.6b01720. (b) A. Gaspar, M. João Matos, J. Garrido, E. Uriarte, F. Borges, Chromone: a valid scaffold in medicinal chemistry, Chem. Rev. 114 (2014) 4960–4992. https://doi.org/10.1021/cr400265z. (c) R.S. Keri, S. Budagumpi, R.K. Pai, R.G. Balakrishna, Chromones as a privileged scaffold in drug discovery, Eur. J. Med. Chem. 78 (2014) 340–374. https://doi.org/10.1016/j.ejmech.2014.03.047. [2] (a) M. Correia-da-Silva, E. Sousa, B. Duarte, F. Marques, F. Carvalho, L.M. Cunha-Ribeiro, M.M.M. Pinto, Flavonoids with an oligopolysulfated moiety: a new class of anticoagulant agents, J. Med. Chem 54 (2011) 95–106. https://doi.org/10.1021/jm1013117. (b) G. Roma, M. di Braccio, A. Carrieri, G. Grossi, G. Leoncini, M.G. Signorello, A. Carotti, Coumarin, chromone, and 4(3*H*)-pyrimidinone novel bicyclic and tricyclic derivatives as antiplatelet agents: synthesis, biological evaluation, and comparative molecular field analysis, Bioorg. Med. Chem. 11 (2003) 123–138. https://doi.org/10.1016/S0968-0896(02)00307-3. (c) M. Mazzei, E. Sottofattori, M. di Braccio, A. Balbi, G. Leocini, E. Buzzi, M. Maresca, Synthesis and antiplatelet activity of 2-(diethylamino)-7-ethoxychromone and related compounds, Eur. J. Med. Chem. 25 (1990) 617–622. https://doi.org/10.1016/0223-5234(90)90187-8.

[3] (a) V.S. Dofe, A.P. Sarkate, D.K. Lokwani, D.B. Shinde, S.H. Kathwate, C.H. Gill, Novel O-alkylated chromones as antimicrobial agents: ultrasound mediated synthesis, molecular docking ADME prediction, J. Heterocycl. Chem. 54 (2017)2678-2685. and https://doi.org/10.1002/jhet.2868. (b) G. Sagrera, A. Bertucci, A. Vazques, G. Seoane, Synthesis and antifungal activities of natural and synthetic bioflavonoids, Bioorg. Med. Chem. 19 (2011) 3060-3073. https://doi.org/10.1016/j.bmc.2011.04.010. (c) O. Prakash, R. Kumar, V. Prakash, antifungal activity of some new 3-hydroxy-2-(1-phenyl-3-aryl-4-**Synthesis** and pyrazolyl)chromones, Eur. J. Med. Chem. 43 (2008)435-440. https://doi.org/10.1016/j.ejmech.2007.04.004. (d) V.A. Bykov, V.A. Dubinskava, L.B. Rebrov, M.F. Mineeva, S.G. Skuridin, Yu.M. Evdokimov, Complex approach to investigation of the mechanisms of action of antimicrobial and antiviral drugs, Pharm. Chem. J. 42 (2008) 105-110. https://doi.org/10.1007/s11094-008-0081-2.

[4] (a) J.F. Zeidner, M.C. Foster, A.L. Blackford, M.R. Litzow, L.E. Morris, S.A. Strickland, J.E. Lancet, P. Bose, M.Y. Levy, R. Tibes, I. Gojo, C.D. Gocke, G.L. Rosner, R.F. Little, J.J. Wright, L.A. Doyle, B.D. Smith, J.E. Karp, Final results of a randomized multicenter phase II study of alvocidib, cytarabine, and mitoxantrone versus cytarabine and daunorubicin (7+3) in newly diagnosed high-risk acute myeloid leukemia, Leuk. Res. 72 (2018) 92-95. https://doi: 10.1016/j.leukres.2018.08.005. (b) H.L.T. Anh, N.T. Cuc, B.H. Tai, P.H. Yen, N.X. Nhiem, D.T. Thao, N.H. Nam, C.V. minh, P.V. Kiem, Y.H. Kim, Synthesis of chromonylthiazolidines and their cytotoxity to human cancer cell lines, Molecules, 20 (2015)1151-1160. https://doi.org/10.3390/molecules20011151. (c) A. Szulawska-Mroczek, M. Szumilak, M. Szczesio, A. Olczak, R.B. Nazarski, W. Lewgowd, M. Czyz, A. Stanczak, Synthesis and biological evaluation of new bischromone derivatives with antiproliferative activity, Arch. Pharm. 346 (2013) 34-43. https://doi.org/10.1002/ardp.201200220.

[5] (a) M. Torres-Piedra, M. Figueroa, O. Hernández-Abreu, M. Ibarra-Barajas, G. Navarrete-Vázquez, S. Estrada-Soto, Vasorelaxant effect of flavonoids through calmodulin inhibition: ex vivo, in vitro and in silico approaches, Bioorg. Med. Chem. 19 (2011) 542–546. https://doi.org/10.1016/j.bmc.2010.10.063. (b) J. Duarte, C. Lugnier, A.I. Torres, F. Pérez-Vizcaíno, A. Zarzuelo, J. Tamargo, Effects of visnagin on cyclic nucleotide phosphodiesterases and their role in its inhibitory effects on vascular smooth muscle contraction, Gen. Pharmacol. 32 (1999) 71–74. https://doi.org/10.1016/S0306-3623(98)00083-4. (c) J. Duarte, F. Pérez-Vizcaíno, A.I. Torres, A. Zarzuelo, J. Jiménez, J. Tamargo, Vasodilator effects of visnagin in isolated rat vascular smooth muscle, Eur. J. Pharmacol. 286 (1995) 115–122. https://doi.org/10.1016/0014-2999(95)00418-K.

[6] (a) D. Kang, K. Ahn, S. Hong, Site-selective C–H bond functionalization of chromones and coumarins, Asian J. Org. Chem. 7 (2018) 1136–1150. https://doi.org/10.1002/ajoc.201800128. (b) C.M.M. Santos, V.L.M. Silva, A.M.S. Silva, Synthesis of chromone-related pyrazole compounds, Molecules, 22 (2017) 1665. https://doi.org/10.3390/molecules22101665. (c) K.V. Shcherbakov, Ya.V. Burgart, V.I. Saloutin, O.N. Chupakhin, Polyfluorine-containing chromen-4-ones: synthesis and transformations, Russ. Chem. Bull. 65 (2016) 2151–2162. https://doi.org/10.1007/s11172-016-1563-0. (d) M.Yu. Kornev, V.Ya. Sosnovskikh, Synthesis and chemical properties of

chromone-3-carboxylic acid, Chem. Heterocycl. Compd. 52 (2016) 71–83. https://doi.org/10.1007/s10593-016-1834-6.

[7] (a) D. Pajtás, K. Kónya, A. Kiss-Szikszai, P. Džubák, Z. Pethő, Z. Varga, G. Panyi, T. Patonay, Optimization of the synthesis of flavone-amino acid and flavone-dipeptide hybrids via **Buchwald-Hartwig** (2017)reaction. J. Org. Chem. 82 4578-4587. https://doi.org/10.1021/acs.joc.7b00124. (b) M. Veselovskaya, M. Garazd, A. Ogorodniichuk, Ya. Garazd, V. Khilya, Synthesis of amino-acid derivatives of chrysin, Chem. Nat. Compd. 44 (2008) 704-711. https://doi.org/10.1007/s10600-009-9194-4. (c) X. Song, Y. Liu, J. Ma, J. He, X. Zheng, X. Lei, G. Jiang, L. Zhang, Synthesis of novel amino acid derivatives containing chrysin as antitumor agents against human gastric carcinoma MGC-803 cells, Med. Chem. Res. 24 (2015) 1789-1798. https://doi.org/10.1007/s00044-014-1267-7. (d) N. Yao, C.Y. Chen, C.Y. Wu, K. Motonishi, H. Kung, K.S. Lam, Novel flavonoids with antiproliferative activities against breast cancer cells, J. Med. Chem. 54 (2011) 4339-4349. https://doi.org/10.1021/jm101440r. (e) M. Ninomiya, K. Tanaka, Y. Tsuchida, Y. Muto, M. Koketsu, K. Watanabe, Increased bioavailability of tricinamino acid derivatives via a prodrug approach, J. Med. Chem. 54 (2011) 1529-1536. https://doi.org/10.1021/jm1015457.

[8] (a) K. Haranahalli, T. Honda, I. Ojima, Recent progress in the strategic incorporation of fluorine into medicinally active compounds, J. Fluorine Chem. 217 (2019) 29-40. https://doi.org/10.1016/j.jfluchem.2018.11.002. (b) E.P. Gillis, K.J. Eastman, M.D. Hill, D.J. Donnelly, N.A. Meanwell, Applications of fluorine in medicinal chemistry, J. Med. Chem. 58 (2015) 8315-8359. 10.1021/acs.jmedchem.5b00258. (c) C. Isanbor, D. O'Hagan, Fluorine in medicinal chemistry: a review of anti-cancer agents, J. Fluorine Chem. 127 (2006) 303-319. https://doi.org/10.1016/j.jfluchem.2006.01.011. (d) J.-P. Bégué, D. Bonnet-Delpon, Recent advances in fluorinated pharmaceuticals based on natural products, J. Fluorine Chem. 127 (2006) 992-1012. https://doi.org/10.1016/j.jfluchem.2006.05.006. (e) W.K. Hagman, The many roles of fluorine in medicinal chemistry, J. Med. Chem. 51 (2008)4359-4369. https://doi.org/10.1021/jm800219f; (f) D. O'Hagan, Fluorine in health care: organofluorine containing blockbuster drugs, J. Fluorine Chem. 131 (2010)1071-1081. https://doi.org/10.1016/j.jfluchem.2010.03.003.

[9] K.V. Shcherbakov, D.N. Bazhin, Ya.V. Burgart, V.I. Saloutin, Novel 3-acetyl-2-methyl-polyfluorochromones in reactions with amines and esters of amino acids, Chem. Heterocycl. Compd. 51 (2015) 961–968. https://doi.org/10.1007/s10593-016-1805-y.

[10] K.A. Myannik, V.N. Yarovenko, M.M. Krayushkin, K.S. Levchenko, Synthesis of 2-substituted 3-iodo-4*H*-chromen-4-ones, Russ. Chem. Bull. 63 (2014) 543–545. https://doi.org/10.1007/s11172-014-0466-1.

[11] K.V. Shcherbakov, Ya.V. Burgart, V.I. Saloutin, O.N. Chupakhin, Modification of polyfluoro-containing 3-(ethoxycarbonyl)flavones by biogenic amines and amino acids, Curr. Org. Synth. 15 (2018) 707–714. https://doi.org/10.2174/1570179415666180405120706.

[12] (a) M.D. Mashkovski, Drugs (in Russian), fourteenth ed., Novaya Volna, Moscow, 2002.
(b) V.M. Berestovitskaya, I.N. Tyurenkov, O.S. Vasilyeva, V.N. Perfilova, E.S. Ostroglyadov, V.V. Bagmetova, Racetams: Methods of Synthesis and Biological Activity (in Russian), Asterion, Saint-Petersburg, 2016.

[13] (a) N.N. Vorozhtsov Jr., V.A. Barhash, A.T. Prudchenko, T.I. Khomenko, G.S. Shchegoleva, Pentafluorobenzoylacetic ester, Zh. Obshch. Khim. 34 (1964) 1501. (b) S.P. Kisil', Ya.V. Burgart, V.I. Saloutin, O.N. Chupakhin, Fluoroaryl containing β , β '-dioxoesters in the synthesis of fluorobenzopyran -4(2)-ones, J. Fluorine Chem. 108 (2001) 125–131. https://doi.org/10.1016/S0022-1139(01)00346-3.

[14] H.-D. Jakubke, H. Jeschkeit, Aminosäuren, Peptide, Proteine, Akademie-Verlag, Berlin, 1982.

[15] A.J. Gordon, R.A. Ford, The Chemist's Companion, John Wiley & Sons, New York, 1972.

[16] (a) J. Cho, I.S. Hwang, H. Choi, J.H. Hwang, J.S. Hwang, D.G. Lee, The novel biological action of antinicrobal peptides via apoptosis induction, J. Microbiol. Biotechnol. 22 (2012) 1457–

1466. https://doi.org/10.4014/jmb.1205.05041. (b) M. Wilmes, B.P. Cammue, H.-G. Sahl, K. Thevissen, Antibiotic activities of host defence peptides: more to it than lipid bilayer perturbation, Nat. Prod. Rep. 28 (2011) 1350–1358. https://doi.org/10.1039/C1NP00022E.

[17] (a) D.Yu. Demin, K.A. Myannik, P.A. Ermolich, M.M. Krayushkin, V.N. Yarovenko, Synthesis of 3-(N-arylthiocarbamoyl)chromones from 2-hydroxyaryl amino enones and isothiocyanates, Mendeleev Commun. 28 (2018)485-486. https://doi.org/10.1016/j.mencom.2018.09.010. (b) W. Löwe, N. Matzanke, Synthesis of flavone-3-sulfonylureas, J. Heterocycl. Chem. 33 (1996)943-948. https://doi.org/10.1002/jhet.5570330364. (c) E.S.C. Wu, T.E. Cole, T.A. Davidson, M.A. Dailey, K.G. Doring, M. Fedorchuk, J.T. Loch, T.L. Thomas, J.C. Blosser, A.R. Borrelli, C.R. Kinsolving, R.B. Parker, J.C. Strand, B.E. Watkins, Flavones. 2. Synthesis and structure activity relationship of flavodilol and its analogs. A novel class of antihypertensive agents with vatecolamine depleting properties, J. Med. Chem. 32 (1989) 183-192. https://doi.org/10.1021/jm00121a034.

[18] G.M. Sheldrick, A short history of SHELX, Acta Crystallogr., Sect. A: Found. Crystallogr. A64 (2008) 112–122. https://doi.org/10.1107/S0108767307043930.