

The Effect of Lipophilic Substituents on the H₂-Histaminergic Activity of Some Close Analogues of Impromidine^{+)**)}

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The cimetidine-like moiety of the potent H₂-agonist impromidine (**9a**) and three closely related guanidines (**10a**, **11a**, and **12a**) which are modified in the imidazolylpropyl portion, has been replaced by 2-[(2-pyridyl)methylthio]ethyl, 2-(benzylthio)ethyl and 3,3-diphenylpropyl substituents. Guanidines **10-12** were obtained from acidic hydrolysis of corresponding *N*-benzoyl guanidines **7**, **8**, and **15**, accessible by successive aminolysis of diphenyl *N*-benzoyl carbonimidate (**2**) according to known methods. Compared with leads **10a** and **11a** lipophilic substitution affords almost equipotent H₂-agonists **10b-d** and **11b-d**, while substituents with increasing lipophilicity enhance both intrinsic activity and potency of the weak partial agonist **12a** at guinea-pig atrial H₂-receptors. Guanidines **10-12** are weak H₁-antagonists on the isolated guinea-pig ileum.

Der Einfluß lipophiler Substituenten auf die H₂-histaminerge Aktivität einiger Impromidin-Analoga

Der cimetidinähnliche Substituent des potenten H₂-Agonisten Impromidin (**9a**) sowie dreier nahe verwandter Guanidine (**10a**, **11a** und **12a**), die eine Modifikation in der Imidazolylpropyl-Gruppierung aufweisen, wurde durch 2-[(2-Pyridyl)methylthio]ethyl-, 2-(Benzylthio)ethyl- und 3,3-Diphenylpropyl-Reste ersetzt. Die Guanidine **10-12** wurden durch saure Hydrolyse aus den entspr. *N*-Benzoylguanidinen **7**, **8** und **15** erhalten; letztere waren nach bekannter Methodik durch schrittweise Aminolyse von Diphenyl-*N*-benzoyl-imidocarbonat (**2**) zugänglich. Verglichen mit den Leitstrukturen **10a** und **11a** ergibt eine lipophile Substitution jeweils nahezu äquipotente H₂-Agonisten **10b-d** und **11b-d**, wohingegen zunehmend lipophile Reste sowohl die intrinsische Aktivität als auch die Potenz des schwachen partiellen Agonisten **12a** am H₂-Rezeptor des Meerschweinchen-Atriums deutlich verstärken. Die Guanidine **10-12** sind schwache H₁-Antagonisten am isolierten Meerschweinchen-Ileum.

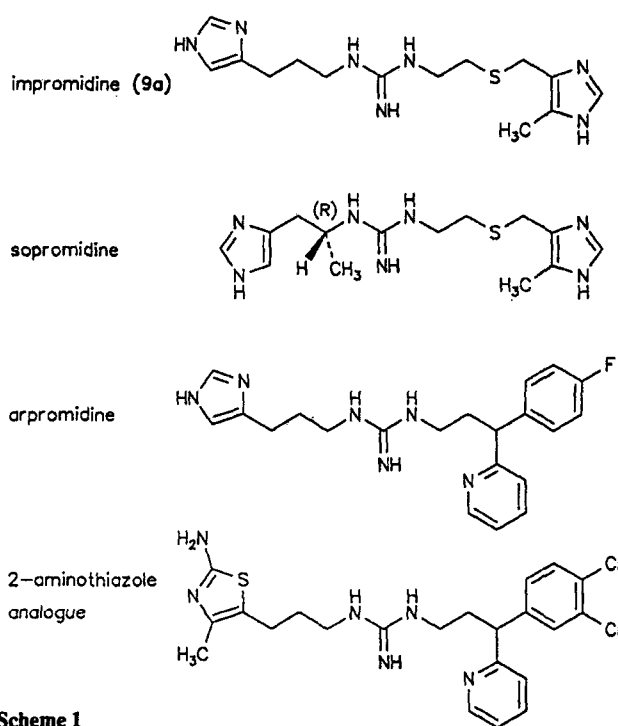
Since the introduction of impromidine (**9a**) as a highly potent selective H₂-agonist¹⁾ numerous publications dealing with structure-activity relationships of impromidine-type H₂-agonists have appeared. The results have clearly demonstrated the major importance of the guanidine^{2,3)} and 3-(1*H*-

imidazol-4-yl)propyl moiety²⁻⁵⁾ for an effective full agonism at atrial H₂-receptors. Only a few attempts to replace the efficacy-causing imidazolylpropyl substituent have been successful so far, leading to the chiral guanidine sopromidine⁶⁾ and a series of 3-(2-aminothiazol-5-yl)propyl analogues⁷⁾ of both **9a** and arpromidine (Scheme 1). On the other hand the cimetidine-like partial structure in **9a** can be replaced by either alternative moieties known from H₂-antagonists^{4,8-10)} or by H₂-nonspecific lipophilic groups¹⁰⁻²¹⁾. Among these compounds arpromidine (Scheme 1) and halogenated analogues^{11,12)} justify special mention as promising positive inotropic agents more potent than impromidine *in vivo* and *in vitro*. Additionally a more beneficial profile of action with respect to lower stimulating effects on heart rate and reduced arrhythmogenic potential *in vivo* has been reported²²⁻²⁴⁾.

In the present study the effect of a lipophilic H₂-nonspecific substitution on the H₂-agonistic activity of three closer analogues of the classical H₂-agonist fragment [3-(1*H*-imidazol-4-yl)propyl]guanidine was investigated. The leads selected for lipophilic modification, *i.e.* **10a**, **11a**, and **12a** (Scheme 2), have recently been reported to possess moderate (**10a**, **11a**) or very weak (**12a**) H₂-agonistic activity as compared with impromidine⁴⁾.

Chemistry

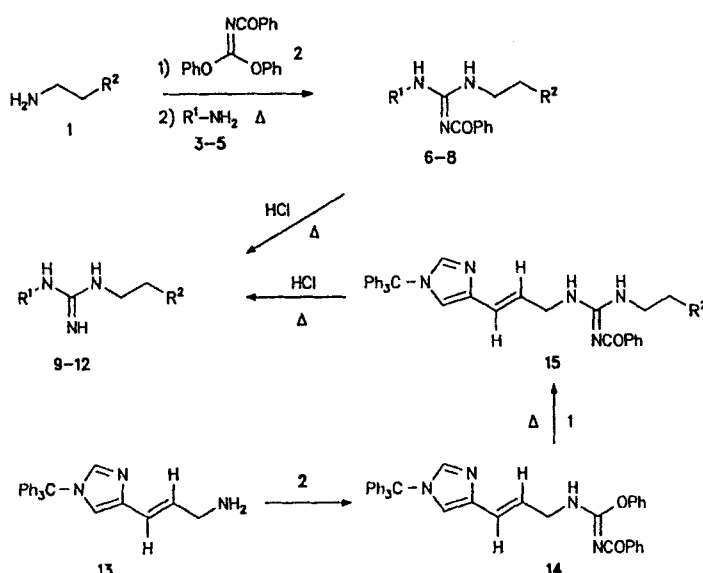
The guanidines **10b-d**, **11b-d**, and **12b-d** were prepared according to a known method¹⁴⁾ using diphenyl *N*-benzoyl carbonimidate (**2**)²⁵⁾ as a source of the central guanidine unit (Scheme 2). Thus the amines **1b-d** were first treated with **2** in CH₂Cl₂, followed by aminolysis of the resulting



Scheme 1

⁺⁾ Presented in part at the Xth International Symposium on Medicinal Chemistry, August 15-19, 1988, Budapest (Hungary)

^{**)} Dedicated to Prof. Dr. Dr. E. Mutschler, Frankfurt am Main, on the occasion of his 60th birthday



Scheme 2

N-benzoyl isoureas with 3-(5-methyl-1*H*-imidazol-4-yl)propanamine (4) or racemic 3-(5-methyl-1*H*-imidazol-4-yl)-2-methylpropanamine (5)²⁶, respectively, and subsequent chromatographic purification of the resulting benzoyl guanidines 7b-d and 8b-d. Acidic hydrolysis afforded the corresponding guanidines 11b-d and 12b-d in almost quantitative yield. For the synthesis of 10b-d the *N*^t-protected (*E*)-configured 3-(1*H*-imidazol-4-yl)allylamine 13²⁶ was treated with 2, followed by isolation of the crystalline isourea 14 and subsequent aminolysis with 1b-d, chromatographic purification of 15b-d and final hydrolysis. Yields, physical and spectroscopic data are given in tab. 1 and 2.

Pharmacological Results and Discussion

Guanidines 10-12 were screened for H₂-agonistic activity on the isolated spontaneously beating right guinea-pig atrium as well as for H₁-antagonistic properties^{11,12} in the isolated guinea-pig ileum. Results together with lit. data for 9a-d and for hydrophilic leads 10a, 11a, and 12a are listed in tab. 3.

All new compounds display a significant decrease of H₂-agonistic activity compared with those analogues [9a-d] containing the classical 3-(1*H*-imidazol-4-yl)propyl substituent. This fact again emphasizes the outstanding efficacy-contributing properties of the "homohistamine"⁴ fragment (3). A view into each group of compounds 10, 11, or 12, respectively, reveals heterogeneous trends of structure-activity relationships. The activity of 11a - which is more potent than 10a as well as 12a (potency: 24% relative to impromidine) - is not significantly altered by introduction of one of the three lipophilic substituents [11b-d]. Apparently the situation is similar for the analogues of 10a (potency: 7% relative to impromidine) with the exception of 10c. For all compounds mentioned so far full or nearly full agonism is observed (intrinsic activities 0.83-1.0). As far as the group of racemic 12a-d is concerned, lipophilic substitution can obviously enhance both intrinsic activity and potency com-

pared with 12a which itself is merely a weak partial H₂-agonist (i.a. = 0.34). Especially the 3,3-diphenylpropyl substituent seems to exert a beneficial effect and partially compensates the loss of H₂-agonism observed when the homohistamine part of 9a is formally methylated at C-5 of the imidazole nucleus and C-2 of the propyl chain (9a → 12a).

These data confirm previous observations that the nature of the affinity-contributing moiety can apparently influence the intrinsic activity of guanidine-type H₂-agonists in a positive sense^{4,5}. Taken together with the antagonistic activity at ileal H₁-receptors (-lg K_B ≈ 5 - 6.7) the new guanidines 10b-d, 11b-d, and 12b-d have to be regarded as weak or moderate H₂-agonists that do not display a pronounced preference for one of the histamine receptor subtypes concerning receptor subtype selectivity.

The expert technical assistance of Mrs. Monika Ewald-Feldt is gratefully acknowledged. This study was supported by the Verband der Chemischen Industrie, Fonds der Chemie.

Experimental Part

Chemistry

Melting points (uncorrected): Büchi apparatus. Elemental analyses (C, H, N): Analytical department of the Institute of Pharmacy, Freie Universität Berlin. Structures were confirmed by ¹H-NMR (Bruker WP 60 (60 MHz) or Bruker WM 250 (250 MHz), TMS) (see Tab. 2) and by mass spectrometry (EI: Finnigan MAT CH7A (170°C, 70 eV) or MAT 711 (200°C, 80 eV); FAB: Finnigan MAT CH5DF (⁺FAB, xenon, DMSO/glycerol) (see Tab. 1). Chromatographic separations: Chromatotron 7924T (Harrison Research), glass rotors with 4 mm layers of silica gel 60 PF₂₅₄ containing gypsum (Merck).

General procedure for the synthesis of *N*-benzoyl guanidines 7b-d and 8b-d

A mixture of 5 mmol of 1b-d and 5 mmol 2 is stirred in CH₂Cl₂ (20 ml) for 15 min at room temp. and smoothly evaporated to dryness. The crude isourea is dissolved in pyridine (30 ml), 5.5 mmol of 4 or 5 are added and the solution is refluxed for 1 h. After evaporation *in vacuo* the crude

Compd. no.	R ¹	1, 6-12, 15	R ²
3, 6, 9		a	
10		b	
4, 7, 11		c	
5, 8, 12		d	

Tab. 1: Preparative, analytical and mass spectrometric data

Compd.	yield (%)	mp. (°C) (solvent)	formula (molecular mass)	analyses calc:			m/z ^{a)} (% relative intensity)
				C	H	N	
7b	70	94 (EtOAc/Et ₂ O)	C ₂₃ H ₂₈ N ₆ OS (436.6)	63.3 63.3	6.46 6.54	19.3 19.2	436 (M ⁺ , 20), 341 (10), 125 (9), 109 (30), 96 (45), 95 (53), 93 (70), 77 (50)
7c	60	106 (EtOAc/Et ₂ O)	C ₂₄ H ₂₉ N ₅ OS (435.6)	66.2 66.2	6.71 6.74	16.1 16.2	435 (M ⁺ , 4), 344 (16), 123 (40), 109 (27), 105 (100), 96 (15), 95 (22), 91 (73), 77 (67), 65 (14)
7d	65	amorphous	C ₃₀ H ₃₃ N ₅ O (479.6)	75.1 74.5	6.93 7.08	14.6 14.3	479 (M ⁺ , 12), 312 (8), 167 (12), 109 (27), 105 (100), 95 (17), 77 (45)
8b	68	amorphous	C ₂₄ H ₃₀ N ₆ OS (450.6)	64.0 63.7	6.71 6.92	18.7 18.2	450 (M ⁺ , 8), 358 (9), 282 (28), 267 (6), 222 (6), 204 (8), 191 (6), 137 (12), 123 (30), 105 (100), 96 (14), 95 (19), 93 (53), 82 (17), 77 (24)
8c	73	amorphous	C ₂₅ H ₃₁ N ₅ OS (449.6)	66.8 66.3	6.95 6.99	15.6 15.5	450 (M ⁺ , 9), 359 (40), 335 (12), 327 (8), 204 (18), 137 (31), 123 (30), 105 (100), 96 (13), 95 (20), 91 (52), 77 (39)
8d	75	amorphous	C ₃₁ H ₃₅ N ₅ O (493.6)	75.4 75.6	7.15 7.11	14.2 14.0	493 (M ⁺ , 7), 398 (11), 167 (17), 137 (12), 123 (28), 105 (100), 96 (22), 95 (27), 91 (8), 77 (60)
10b	>95	188-9 (H ₂ O)	C ₁₅ H ₂₀ N ₆ S·3C ₆ H ₃ N ₃ O ₇ (1003.7)	39.5 39.4	2.91 2.76	20.9 20.5	c)
10c	>95	110-13 (Me ₂ CO/Et ₂ O)	C ₁₆ H ₂₁ N ₅ S·2C ₆ H ₃ N ₃ O ₇ (773.6)	b)			316 ([M+H] ⁺ , 67), 210 (51), 151 (16), 107 (30), 91 (100), 80 (14)
10d	>95	133-6 (Et ₂ O)	C ₂₂ H ₂₅ N ₅ ·2C ₆ H ₃ N ₃ O ₇ (817.7)	b)			360 ([M+H] ⁺ , 39), 254 (100), 167 (32), 165 (15), 107 (37), 100 (26), 91 (51)
11b	>95	208-10 (441.8)	C ₁₆ H ₂₄ N ₆ S·3HCl (EtOH)	43.5 43.5	6.16 6.16	19.0 18.8	333 ([M+H] ⁺ , 100), 242 (13), 208 (5), 165 (7), 152 (4), 140 (9), 124 (44), 123 (67), 114 (10), 109 (18), 100 (13), 96 (23), 95 (22), 93 (48)
11c	>95	100-103 (H ₂ O)	C ₁₇ H ₂₅ N ₅ S·2C ₆ H ₃ N ₃ O ₇ ·H ₂ O (807.7)	43.1 43.1	4.12 4.07	19.1 18.7	332 ([M+H] ⁺ , 97), 242 (5), 165 (10), 151 (9), 123 (61), 109 (14), 96 (12), 95 (19), 91 (100)
11d	>95	amorphous	C ₂₃ H ₂₉ N ₅ ·2HCl (448.4)	b)			376 ([M+H] ⁺ , 100), 254 (14), 208 (4), 167 (26), 123 (76), 100 (30), 91 (28)
12b	>95	201-2 (EtOH/Et ₂ O)	C ₁₇ H ₂₆ N ₆ S·3HCl (455.9)	44.8 44.4	6.41 6.62	18.4 18.3	347 ([M+H] ⁺ , 69), 256 (20), 137 (40), 124 (25), 123 (20), 100 (14), 96 (22), 95 (26), 93 (100)
12c	>95	amorphous	C ₁₈ H ₂₇ N ₅ S·2HCl (418.4)	b)			346 ([M+H] ⁺ , 100), 137 (43), 123 (15), 100 (16), 96 (16), 95 (23), 91 (85)
12d	>95	214 (EtOH)	C ₂₄ H ₃₁ N ₅ ·2HCl (462.5)	62.3 62.0	7.19 7.41	15.1 15.2	390 ([M+H] ⁺ , 100), 294 (11), 254 (10), 167 (32), 165 (25), 154 (9), 152 (13), 137 (87), 123 (15), 117 (18), 115 (16), 104 (16), 100 (25), 96 (17), 95 (43), 91 (39)
14	85	108	C ₃₉ H ₃₂ N ₄ O ₂ (588.7)	79.6 79.5	5.48 5.56	9.52 9.52	c)
15b	71	136-7 (Et ₂ O)	C ₄₁ H ₃₈ N ₆ OS (662.8)	74.3 74.3	5.78 5.79	12.7 12.6	c)
15c	68	134-5 (Et ₂ O)	C ₄₂ H ₃₉ N ₅ OS (661.9)	76.2 76.1	5.94 5.94	10.6 10.6	c)
15d	75	184 (EtOH)	C ₄₈ H ₄₃ N ₅ O (705.9)	81.7 81.6	6.14 6.15	9.92 9.70	c)

a) ^aFAB-MS except for 7 and 8 (EI-MS). b) Not determined, very hygroscopic. c) Not determined.

product is dissolved in 5% aqueous HCl and phenol is thoroughly extracted with ether. The acidic layer is alkalinized with 10% aqueous NH₃ and crude 7b-d and 8b-d are extracted into CH₂Cl₂. After drying over Na₂SO₄ the solution is concentrated and the product is purified by Chromatotron (CHCl₃/MeOH 99/1 (v/v), NH₃ atmosphere). Compounds 7d, 8b-d are obtained as amorphous solids; 4b,c are recrystallized from EtOAc/Et₂O.

General procedure for the synthesis of the guanidines 10b-d, 11b-d, and 12b-d

1.2-1.5 mmol of the corresponding *N*-benzoyl guanidine 7, 8, or 15 is refluxed in 6 N-HCl (40 ml) under control by TLC. After completion of the hydrolysis the solution is cooled, extracted thoroughly with ether and repeatedly evaporated to dryness after addition of absol. ethanol to obtain an amor-

Tab. 2: ¹H-NMR data^{a)}

Compd.	δ (ppm); J (Hz)
7b	1.91 (m, 2H, -CH ₂ -CH ₂ -CH ₂ -), 2.13 (s, 3H, Im-5-CH ₃), 2.59 (m, 2H, CH ₂), 2.8 (m, 2H, CH ₂), 3.35 (br, 2H, -CH ₂ -NH-), 3.75 (br, 2H, -CH ₂ -NH-), 3.87 (s, 2H, Py-CH ₂ -S), 7.1-7.7 (m, 6H, arom.), 8.2 (m, 2H, arom.), 8.45 (m, 1H, arom.), 10.4 (br, 1H ^a), Imidazol-NH)
7c	1.89 (m, 2H, -CH ₂ -CH ₂ -CH ₂ -), 2.12 (s, 3H, Im-5-CH ₃), 2.6-2.8 (m, 4H, CH ₂), 3.34 (br, 2H, -CH ₂ -NH-), 3.75 (m, 4H, Ph-CH ₂ -S, -CH ₂ -NH-)
7d	1.83 (m, 2H, -CH ₂ -CH ₂ -CH ₂ -), 2.12 (s, 3H, Im-5-CH ₃), 2.2-2.8 (m, 4H, CH ₂), 3.5 (br, 4H, -CH ₂ -NH-), 4.2 (t, J=7, 1H, CH), 6.9-7.6 (m, 14H, arom.), 8.13 (m, 2H, arom.)
8b	0.96 (d, J=3.5, 3H, CH ₃), 2.1 (m, 4H, Im-5-CH ₃ , CH), 2.52 (m, 2H, Im-CH ₂ -), 2.8 (br, 2H, SCH ₂), 2.95-3.88 (m, 6H, Py-CH ₂ -S, -CH ₂ -NH-), 7.15-8.65 (m, 10H, arom.), 10.55 (br, 1H ^a), Imidazol-NH)
8c	0.95 (d, J=6, 3H, CH ₃), 2.1 (m, 4H, Im-5-CH ₃ , CH), 2.55 (m, 4H, Im-CH ₂ -), SCH ₂), 3.35 (m, 6H, -CH ₂ -NH-, Ph-CH ₂ -S), 7-8.2 (m, 11H, arom.), 10.4 (br, 1H ^a), Imidazol-NH)
8d	0.93 (br, 3H, CH ₃), 2.09 (m, 4H, Im-5-CH ₃ , CH), 2.48 (m, 4H, Im-CH ₂ -), CH ₂), 2.95-3.65 (m, 4H, -CH ₂ -NH-), 4.05 (m, 1H, Diphenyl-CH), 7.15-7.55 (m, 14H, arom.), 8.12 (m, 2H, arom.), 10.6 (br, 1H ^a), Imidazol-NH)
10b	2.7 (m, 2H, SCH ₂), 3.5 (m, 2H, -CH ₂ -NH-), 4.07 (m, 2H, -CH ₂ -NH-), 4.26 (s, 2H, Py-CH ₂ -), 6.56 (m, 2H, -CH=CH-), 7.7-7.95 (m, 1H, 1H, 2H ^a), Py-5-H, Im-5-H, C=NH ₂ ⁺), 8.06 (m, 1H ^a), 1H, -NH-CH ₂ -), Py-3-H), 8.26 (m, 1H ^a), -NH-CH ₂ -), 8.43 (m, 1H, Py-4-H), 8.78 (m, 1H, Py-6-H), 9.14 (s, 1H, Im-2-H), 13.63 (br, 2H ^a), Imidazolium-NH)
10c	2.55 (m, 2H, SCH ₂), 3.43 (m, 2H, -CH ₂ -NH-), 3.82 (s, 2H, Ph-CH ₂ -S), 4.06 (m, 2H, -CH ₂ -NH-), 6.55 (m, 2H, -CH=CH-), 7.25-7.45 (m, 5H, arom.), 7.74 (s, 1H, Im-5-H), 7.77 (m, 2H ^a), C=NH ₂ ⁺), 8.0 (m, 1H ^a), -NH-CH ₂ -), 8.2 (m, 1H ^a), -NH-CH ₂ -), 9.13 (s, 1H, Im-2-H), 14.98 (br, 2H ^a), Imidazolium-NH)
10d	2.3 (m, 2H, CH ₂), 3.08 (m, 2H, -CH ₂ -NH-), 4.03 (m, 2H, CH ₂ -NH-), 4.13 (t, J=8.5, CH), 6.5 (m, 2H, CH=CH-), 7.15-7.35 (m, 10H, arom.), 7.65 (s, 2H ^a), C=NH ₂ ⁺), 7.73 (s, 1H, Im-5-H), 8.12 (m, 2H ^a), -NH-CH ₂ -), 9.13 (s, 1H, Im-2-H), 14.9 (br, 2H ^a), Imidazolium-NH)
11b	1.83 (m, 2H, -CH ₂ -CH ₂ -CH ₂ -), 2.25 (s, 3H, Im-5-CH ₃), 2.69 (m, 4H, SCH ₂ , Im-CH ₂ -), 3.17 (m, 2H, -CH ₂ -NH-), 3.46 (m, 2H, -CH ₂ -NH-), 4.31 (s, 2H, Py-CH ₂ -), 7.75 (s, 2H ^a), C=NH ₂ ⁺), 8.84 (d, J=5.5, 1H, Py-6-H), 8.52 (m, 1H, Py-4-H), 7.89-8.0 (m, 2H, Py-5-H, NH), 8.15 (m, 1H, 1H ^a), Py-3-H, NH), 8.97 (s, 1H, Im-2-H), 14.6 (br, 2H ^a), Imidazolium-NH)
11c	1.84 (m, 2H, -CH ₂ -CH ₂ -CH ₂ -), 2.24 (s, 3H, Im-5-CH ₃), 2.54 (m, 2H, Im-CH ₂ -), 2.7 (t, J=7.5, 2H, SCH ₂), 3.19 (m, 2H, -CH ₂ -NH-), 3.45 (m, 2H, CH ₂), 3.83 (s, 2H, Ph-CH ₂ -), 7.2-7.4 (m, 5H, arom.), 7.73 (s, 2H ^a), C=NH ₂ ⁺), 7.96 (m, 1H ^a), -NH-CH ₂ -), 8.22 (m, 1H ^a), -NH-CH ₂ -), 9.02 (s, 1H, Im-2-H)
11d	1.8 (m, 2H, -CH ₂ -CH ₂ -CH ₂ -), 2.23 (s, 3H, Im-5-CH ₃), 2.34-2.77 (m, 4H, CH ₂), 3.07 (m, 4H, -CH ₂ -NH-), 4.1 (t, J=5, 1H, CH), 7.3 (m, 10H, arom.), 7.5 (s, 2H ^a), C=NH ₂ ⁺), 8.09 (m, 2H ^a), -NH-CH ₂ -), 8.92 (s, 1H, Im-2-H),
12b	0.85 (d, J=6.5, 3H, CH ₃), 2.09 (m, 1H, CH), 2.24 (s, 3H, Im-5-CH ₃), 2.45-2.85 (m, 4H, CH ₂), 3.05-3.55 (m, 4H, -CH ₂ -NH-), 4.27 (s, 2H, Py-CH ₂ -), 7.75 (s, 2H ^a), C=NH ₂ ⁺), 7.86 (m, 1H, Py-5-H), 8.09 (m, 1H, 1H ^a), Py-3-H, NH), 8.21 (s, 1H ^a), -NH-CH ₂ -), 8.45 (m, 1H, Py-4-H), 8.8 (d, J=5.5, Py-6-H), 8.96 (s, 1H, Im-2-H), 14.54 (br, 2H ^a), Imidazolium-NH)
12c	0.85 (d, J=6.5, 3H, CH ₃), 2.09 (m, 1H, CH), 2.23 (s, 3H, Im-5-CH ₃), 2.45-2.85 (m, 4H, CH ₂), 3.05-3.25 (m, 2H, -CH ₂ -NH-), 3.43 (m, 2H, -CH ₂ -NH-), 3.83 (s, 2H, Ph-CH ₂ -), 7.24-7.39 (m, 5H, arom.), 7.72 (s, 2H ^a), C=NH ₂ ⁺), 7.96 (m, 1H ^a), -NH-CH ₂ -), 8.18 (m, 1H ^a), -NH-CH ₂ -), 8.96 (s, 1H, Im-2-H), 14.5 (br, 2H ^a), Imidazolium-NH)
12d	0.84 (d, J=6.5, CH ₃), 2.06 (m, 1H, CH), 2.23 (s, 3H, Im-5-CH ₃), 2.29 (m, 2H, CH ₂), 2.45-2.85 (m, 2H, CH ₂), 3.02-3.26 (m, 4H, -CH ₂ -NH-), 4.19 (t, J=8.5, Diphenyl-CH), 7.18-7.32 (m, 12H, arom.), C=NH ₂ ⁺), 7.62 (s, 1H ^a), -NH-CH ₂ -), 8.11 (s, 1H ^a), -NH-CH ₂ -), 14.5 (br, 2H ^a), Imidazolium-NH)
14	4.29 (m, 2H, CH ₂), 6.47-6.49 (m, 2H, -CH=CH-), 6.79 (s, 1H, Im-5-H), 7.13-7.92 (m, 18H, arom.), 10.26 (m, 1H ^a), NH)
15b	2.75 (t, J=6.5, 2H, SCH ₂), 3.67 (br, 2H, -CH ₂ -NH-), 3.81 (s, 2H, Py-CH ₂ -S), 4.1 (br, 2H, -CH ₂ -NH-), 6.35-6.53 (m, 2H, -CH=CH-), 6.77 (s, 1H, Im-5-H), 7.1-7.5 (m, 18H, arom.), 7.62 (m, 1H, arom.), 8.22 (m, 2H, arom.), 8.5 (m, 1H, arom.)
15c	2.66 (t, J=6.5, 2H, SCH ₂), 3.6 (br, 2H, -CH ₂ -NH-), 3.66 (s, 2H, Ph-CH ₂ -S), 4.0 (br, 2H, -CH ₂ -NH-), 6.38 (dt, J=16/5, 1H, =CH-CH ₂ -), 6.5 (d, J=16, 1H, Im-CH=CH-), 6.77 (s, 1H, Im-5-H), 7.1-7.45 (m, 19H, arom.), 8.18 (m, 2H, arom.), 10.5 (br, 1H ^a), Imidazol-NH)
15d	2.4 (m, 2H, CH ₂), 3.4 (br, 2H, -CH ₂ -NH), 4.0 (br, 3H, CH, -CH ₂ -NH-), 6.33 (dt, J=16/5, 1H, =CH-CH ₂ -), 6.44 (d, J=16, 1H, Im-CH=CH-), 6.77 (s, 1H, Im-5-H), 7.1-7.45 (m, 14H, arom.), 8.17 (m, 2H, arom.)

^{a)} Bruker WM 250 (250 MHz) except data for 7d and 11d (Bruker WP 60 (60 MHz)). Solvent: CDCl₃ (7, 8, 14, 15) or [D₆]DMSO (10, 11, 12).

Abbreviations: br = broad, Im = imidazole, Ph = phenyl, Py = pyridyl. a) Exchangeable with D₂O.

phous solid or dry foam. Non stable or very hygroscopic hydrochlorides are converted into di- or tri-picrates for analysis. Compounds that gave even very hygroscopic picrates were characterized only by spectral methods.

N-Benzoyl-*N'*-[3-(1-triphenylmethyl-1*H*-imidazol-4-yl)-2-propen-1-yl]-*O*-phenylisourea (14)

A mixture of 30 mmol 13 and 30 mmol 2 is stirred in CH₂Cl₂ (70 ml) for 15 min at room temp. and smoothly evaporated to dryness. The residue is crystallized from ethanol.

General procedure for the synthesis of N-benzoyl guanidines 15b-d

A solution of 5 mmol 14 and 5.5 mmol 1b-d is refluxed in pyridine (40 ml) for 1 h. After evaporation of the solvent 15d is crystallized directly from ethanolic solution. For the isolation of 15b,c the residue is suspended in 5% aqueous HCl and extracted with ether. The acidic layer is alkalinized with 25% aqueous NH₃ and crude 15b,c are extracted into CH₂Cl₂. After drying over Na₂SO₄ the solution is evaporated and the product is purified by Chromatotron (CHCl₃/MeOH 99/1 (v/v), NH₃ atmosphere). Finally 15b,c are crystallized from ether.

Tab. 3: Biological data

compound	atrium (H ₂)			ileum (H ₁)	
	i. a. ^{a)} (±0.05)	pD ₂ ^{b)} (±0.10)	rel. pot. ^{c)} (%)	pA ₂ ^{d)} (±0.10)	Ref.
histamine	1.0	6.00	100	-	
arpromidine	1.0	8.01	10230	7.7	[12]
9a (impromidine)	1.0	7.68	4810	5.5	[1]
9b	1.0	7.30	2000	5.4	[11]
9c	1.0		1260	5.3	[20]
9d	1.0	7.20	1580	6.3	[11]
10a	0.91	6.54	350	4.7	[4]
10b	0.83	6.16	140	5.0	
10c	0.86	5.48	30	5.6	
10d	0.85	6.23	170	6.6	
11a	0.90	7.06	1150	-	[4]
11a	0.85		810	-	[3]
11b	0.86	6.88	760	5.2	
11c	0.89	6.58	380	5.7	
11d	1.0	6.77	590	6.7	
12a	0.34	-	-	5.8	[4]
12b	0.35	-	-	5.9	
12c	0.60	5.09	10	6.7	
12d	0.90	6.21	160	6.6	

a) Intrinsic activity; mean of 3-4 experiments (± s.e.m.). b) Mean of 3-4 experiments (± s.e.m.).

c) Relative potency compared to histamine (=100%). d) Mean of 4 experiments (± s.e.m.).

Pharmacology

H₂-Agonism on the isolated guinea-pig atrium²⁷⁾

Male guinea-pigs (350-500 g) were sacrificed by a blow on the head and exsanguinated. The right atrium was rapidly removed, cleared of connective tissue, attached to a tissue holder in a 20 ml organ bath (32.5°C) containing McEvans solution²⁷⁾ and aerated with carbogen. The pD₂ values and intrinsic activities were determined from isometrically recorded cumulative concentration-response curves²⁸⁾ as described by Lennartz et al.²⁹⁾, using histamine as the reference agonist (0.1-10 μM). Potential agonists were used as hydrochlorides. The maximum effect of the agonist studied was completely abolished by 10-30 μM cimetidine. H₂-receptor selectivity was verified by some experiments in the presence of 1 μM metoprolol.

H₁-antagonistic activity on the isolated guinea-pig ileum

The H₁-antagonistic activity was determined from isotonicity recorded (preload 0.5 g) cumulative concentration-response curves, as described²⁹⁾, using 20 ml organ baths containing Tyrode solution aerated with carbogen at 37°C. Hydrochlorides were tested at concentrations of 0.3-30 μM versus histamine after 10 min of incubation. Apparent -lg K_B values were calculated from the rightward shift of the histamine curve in the presence and absence of antagonist²⁸⁾.

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