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Two new glycoalkaloids from Stephania succifera

Shi-Ting Wang^a, Wen-Qi Qian^a, Peng He^b, Mei-Qing Feng^b, Yun Kang^a, Ya-Qin Wang^a, Jian-Ming Huang^a,

^a Department of Pharmacognosy, School of Pharmacy, Fudan University, Shanghai, 201203, China
^b Department of Microbiological and Biochemical Pharmacy, School of Pharmacy, Fudan University, Shanghai, 201203, China

ARTICLE INFO	A B S T R A C T
Keywords: Stephania succifera Alkaloid Antibacterial activity	Two new glycoalkaloids, manshurienine C (1) and 8-oxotetrahy-corydalmine-1- O - β -D-glucopyranoside (2), along with eleven known alkaloids (3–13), were isolated from the tubers of <i>Stephania succifera</i> . Their structures were elucidated by spectroscopic methods, including NMR and MS spectra. Compounds 5, 6, 8, and 10 showed <i>in vitro</i> antibacterial activity against methicillin-resistant <i>Staphylococcus aureus</i> (MRSA).

1. Introduction

Stephania succifera H. S. Lo et Y. Tsoong (Menispermaceae) is mainly distributed in southern China. The tubers are used in Chinese folk medicine for acesodyne, sedation, and detoxication (The Chinese Academy of Science Flora of China Edition Board, 1996). Several classes of alkaloids have been isolated from S. succifera, including aporphine, proaporphine, and protoberberine alkaloids (Chen et al., 1989; He et al., 2017; Semwal et al., 2010). Most of these alkaloids exhibited significant pharmacological effects such as cytotoxic, antibacterial, and analgaesic activities (Yang et al., 2010; Zeng et al., 2017). The continuous search for novel and bioactive compounds from S. succifera has led to the isolation of two new glycoalkaloids (1 and 2) (Fig. 1) and 11 known alkaloids (3-13) (Supplementary Fig. S1). Compound 3 is found in nature for the first time, and compounds 4–5, 7, 8, 11, and 13 are isolated from this plant for the first time. The isolates were assayed for their antibacterial activities against methicillin-resistant Staphylococcus aureus (MRSA). Compounds 5, 6, 8, and 10 showed antibacterial activities.

2. Results and discussion

Compound 1 was obtained as a yellow amorphous powder. Its molecular formula was identified as $C_{22}H_{19}NO_9$ by HRESIMS with an $[M + H]^+$ ion at m/z 442.1114 (calcd 442.1133). The IR spectrum exhibited absorption bands for lactam (1668 cm⁻¹) and aromatic groups (1557 and 1417 cm⁻¹). The UV absorptions at 381, 323, 289, 277, 267, and 231 nm resembled those of aristolactam derivatives (Yang et al., 2010; Zhang and Jiang, 2006). The ¹³C NMR data of 1 (Table 1) showed

the presence of 22 carbons, including 14 aromatic carbons ($\delta_{\rm C}$ 105.3 – 148.2), a carbonyl ($\delta_{\rm C}$ 167.2), a methylenedioxy ($\delta_{\rm C}$ 103.2), and a glucopyranosyl group ($\delta_{\rm C}$ 60.9, 70.0, 73.7, 76.2, 76.9, and 104.4). The ¹H NMR data (Table 1) presented a methylenedioxy group at $\delta_{\rm H}$ 6.49 (1H, d, J = 1.0 Hz) and 6.50 (1H, d, J = 1.0 Hz); an aromatic AA'BB' coupling system at $\delta_{\rm H}$ 7.66 (1H, dd, J = 7.5, 2.0 Hz, H-6), 7.67 (1H, dd, *J* = 7.5, 2.0 Hz, H-7), 8.59 (1H, dd, *J* = 7.5, 2.0 Hz, H-5), and 8.62 (1H, dd, J = 7.5, 2.0 Hz, H-8); a typical amide proton at $\delta_{\rm H}$ 10.10 (s, 1 H); and a singlet aromatic proton at $\delta_{\rm H}$ 7.68 (1H, s, H-2). Further examination of the ¹H-NMR spectrum of **1** revealed the presence of the glucopyranosyl moiety, which should be in a β -configuration based on the coupling constant (8.0 Hz) of the anomeric proton (H-1'). The absolute configuration was assigned as D-form by comparing the TLC and optical rotation value between the enzymatic hydrolysis product of 1 and the authentic standards (D-glucose and L-glucose). Therefore, it was concluded that this compound was an aristolactam derivative with a β -D-glucopyranose. When comparing the ¹H- and ¹³C-NMR data of **1** with those of manshurienine A and manshurienine B (Zhang and Jiang, 2006), the only difference was that the glucose of compound 1 was attached at C-9 ($\delta_{\rm C}$ 132.1), which was confirmed by the HMBC correlation (Fig. 2) from H-1' ($\delta_{\rm H}$ 4.82) to the aromatic carbon C-9. A combination of the HMBC and HSOC experiments permitted the assignment of all protonated carbons. Thus, compound 1 was elucidated as aristolactam II-9-O- β -D-glucopyranoside and given the trivial name manshurienine C.

Compound **2** was obtained as a yellow amorphous solid. The molecular formula was assigned as $C_{26}H_{31}NO_{11}$ by HRESIMS (m/z 534.1950 [M+H]⁺, calcd 534.1970). The IR spectrum exhibited absorption bands for lactam (1630 cm⁻¹) and aromatic groups (1618 and

* Corresponding author.

E-mail address: jmhuang@shmu.edu.cn (J.-M. Huang).

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Fig. 1. Structures of compounds 1 and 2 from Stephania succifera.

Table 1	
¹ H and ¹³ C NMR spectroscopic data of 1 and 2 (δ i	in ppm).

No.	1		No.	2	
	$\delta_{\rm C}^{\ a}$	$\delta_{ m H}^{\ a}$ (<i>J</i> in Hz)		$\delta_{\rm C}{}^{ m b}$	$\delta_{ m H}{}^{ m b}$ (<i>J</i> in Hz)
1	118.6		1	148.6	
2	105.3	7.68, s	2	141.8	
3	148.2		3	153.9	
4	146.8		4	107.8	6.79, s
4a	108.8		4a	133.2	
4b	125.1		4b	124.0	
5	126.1	8.59, dd (7.5, 2.0)	5	25.1	α 2.57, ddd (16.1, 12.5, 4.5)
6	126.4	7.66, dd (7.5, 2.0)			β 3.36, overlapped
7	127.5	7.67, dd (7.5, 2.0)	6	39.4	α 2.83, td (12.5, 2.9)
8	123.7	8.62. dd (7.5, 2.0)			β 4.90, ddd (12.5, 4.5, 2.9)
8a	130.1		8	164.9	r,,,,
9	132.1		8a	123.0	
10	122.1		9	149.5	
10a	126.0		10	151.5	
11	167.2		11	120.9	7.00, d (8.1)
-OCH ₂ O-	103.2	6.50, d (1.0)	12	124.0	6.95, d (8.1)
2		6.49, d (1.0)			
N-H		10.10. s	12a	131.8	
1'	104.4	4.82, d (8.0)	13	39.7	α 3.16, dd (15.5, 3.0)
2′	73.7	3.46, m			β 2.74, t (15.5)
3′	76.2	3.36, m	13a	56.8	4.79. dd (13.3, 3.0)
4'	70.0	3.23. dd (8.0. 5.1)	1′	105.2	4.98. d (7.5)
5'	76.9	3.29. m	2'	75.7	3.47, overlapped
6′	60.9	3.82. dt (11.3. 4.2)	3′	78.0	3.44. overlapped
		3.51, dt (11.3, 6.0)	4'	71.6	3.39. m
Glc-OH		5.79. d (5.5)	5′	78.2	3.19. m
Glc-OH		5.22. d (4.9)	6'	62.6	3.77. dd (11.9. 5.3)
Glc-OH		5.13, d (5.4)	-		3.65. dd (11.9. 2.3)
6'-OH		5.16. dd (6.0. 4.2)	2-OCH ₂	61.7	3.85. s
		, (,,	3-OCH ₂	56.7	3.86. s
			9-OCH ₂	62.0	3.84. s
			,	02.0	5.0., 0

^a Measured at 600 MHz (¹H) and 150 MHz (¹³C) in DMSO- d_6 .

^b Measured at 600 MHz (¹H) and 150 MHz (¹³C) in methanol-d₄.

1497 cm⁻¹). The ¹³C NMR data of **2** (Table 1) showed the presence of 26 carbons, including 12 aromatic carbons ($\delta_{\rm C}$ 107.8 – 153.9), a carbonyl ($\delta_{\rm C}$ 164.9), three methoxy groups ($\delta_{\rm C}$ 56.7, 61.7, and 62.0), three methylenes ($\delta_{\rm C}$ 39.7, 39.4, and 25.1), a methine ($\delta_{\rm C}$ 56.8), and a glucopyranosyl group ($\delta_{\rm C}$ 62.6, 71.6, 75.7, 78.0, 78.2, and 105.2). The ¹H NMR spectrum revealed the presence of two ortho-coupled aromatic protons at $\delta_{\rm H}$ 6.95 (1H, d, J = 8.1 Hz, H-12) and 7.00 (1H, d, J = 8.1 Hz, H-11); one singlet aromatic proton at $\delta_{\rm H}$ 6.79 (1H, s, H-4); three methoxy groups at $\delta_{\rm H}$ 3.86, 3.85, and 3.84 (each 3H, s). The glucosyl unit with its anomeric proton appearing at $\delta_{\rm H}$ 4.98 (1H, d, J = 7.5 Hz, H-1') was speculated as a β -configuration. Therefore, an oxoprotoberberine with a glucosyl moiety was proposed. A comparison of the ¹H- and ¹³C-NMR data with those of (-)-8-oxo-10-hydroxy-2,3,9trimethoxyberberine revealed that compound 2 had an additional glucosyl moiety attached at C-1. The location of the β -glucosyl moiety was supported by the HMBC correlation (Fig. 3) of H-1' ($\delta_{\rm H}$ 4.98) with C-1 ($\delta_{\rm C}$ 148.6). Enzymatic hydrolysis of compound **2** afforded its sugar and aglycone moieties. The sugar moiety was assigned as D-glucose by

comparing the TLC and optical rotation data with those of the authentic standards (D-glucose and L-glucose). The absolute configuration at C-13a was confirmed as *S* (α -orientation) by comparing the negative optical rotation ([α]25D –71.1°) of the aglycone moiety with the literature value for this aglycone (Lee et al., 2009). A typical ¹H NMR signal at $\delta_{\rm H}$ 4.79 (1H, dd, J = 13.3, 3.0 Hz, H-13a) also supported the *S* configuration at C-13a (Lee et al., 2009). Thus, compound **2** was deduced as (-)-8-oxo-10-hydroxy-2,3,9-trimethoxyberberine-1-*O*- β -D-glucopyranoside.

The known compounds (3–13) (Supplementary Fig. S1) were identified as secorebanine (3) (Wang et al., 2016), norcrebanine (4) (Rayanil et al., 2016), 8-hydroxy-9-methoxy-1,2-methylenedioxyaporphine (5) (Thuy et al., 2012), crebanine (6) (Makarasen et al., 2011), *N*-formyl-asimilobine-2-*O*- β -D-glucopyranoside (7) (Choudhary et al., 2010), asimilobine-2-*O*- β -D-glucopyranoside (8) (Likhitwitayawuid et al., 1993), *N*-methylasimilobine-2-*O*- β -D-glucopyranoside (9) (Kashiwaba et al., 2000), asimilobine (10) (Costa



Fig. 2. Selected $^1\text{H}{-}^1\text{H}$ COSY (H—H) and HMBC (H→C) correlations of compound 1.



Fig. 3. Selected ${}^{1}H-{}^{1}H$ COSY (H—H), HMBC (H→C), and ROESY (H+→H) correlations of compound 2.

et al., 2015), liriodenine (11) (Fujita et al., 2010), 4-hydroxycrebanine (12) (Kunitomo et al., 1985), and *trans*-feruloyl-(3-O-methyl)-dopamine-4-O- β -D-glucopyranoside (13) (Yim et al., 2012) by comparison of experimental and reported spectroscopic data. Compound **3** is found in nature for the first time, and compounds **4**, **5**, **7**, **8**, **11**, and **13** are isolated from *S. succifera* for the first time.

All isolated compounds were tested for their *in vitro* antibacterial activity against MRSA by the broth microdilution method. Vancomycin was used as a positive control. As shown in Table 2, compounds **5**, **6**, **8**, and **10** showed antimicrobial activity against MRSA. The lowest minimum inhibitory concentration (MIC) was observed for compound **6** (128 μ g/mL) followed by compounds **5**, **8**, and **10** (256 μ g/mL).

Table 2 Anti-MRSA activities of the compounds isolated from S. succifera.

Compounds	MIC(µg/mL)
manshurienine C (1)	> 256
8-oxotetrahy-corvdalmine-1- O - β -D-glucopyranoside (2)	> 256
secocrebanine (3)	> 256
norcrebanine (4)	> 256
8-hydroxy-9-methoxy-1,2-methylenedioxyaporphine (5)	256
crebanine (6)	128
<i>N</i> -formyl-asimilobine-2- O - β -D-glucopyranoside (7) asimilobine-2- O - β -D-glucopyranoside (8)	> 256 256
<i>N</i> -methylasimilobine-2- <i>O</i> -β-D-glucopyranoside (9)	> 256
asimilobine (10)	256
liriodenine (11)	> 256
4-hydroxycrebanine (12)	> 256
trans-feruloyl-(3-O-methyl)-dopamine- 4-O- β -D-glucopyranoside (13) vancomycin ^a	> 256 0.5

^a Vancomycin was used as a positive control.

3. Material and methods

3.1. General experimental procedures

Optical rotations were recorded using a Rudolph Autopol IV-T autometic polarimeter (Rudolph Research Analytical, USA). UV spectra were measured on a U-2900 spectrophotometer (Hitachi, Japan). IR spectra were obtained on a Nicolet i5 spectrometer (Thermo Fisher, USA) with KBr pellets. HRESIMS spectra were recorded on a TripleTOF 5600 + mass spectrometer (Applied Biosystems Sciex, USA). NMR spectra were recorded on a Bruker Ascend[™] 600 MHz spectrometer with TMS as internal standard. Column chromatography (CC) was performed using silica gel (200-300 and 300-400 mesh. Oingdao Haivang Chemical Co., Ltd., China), HZ818 macroporous resin (HuaZhen Technology, China), C18 (40-60 µm, Agela Technologies, China), and Sephadex LH-20 (Amersham Biosciences, Sweden). Fractions were monitored by TLC using precoated GF254 silica gel plates (Qingdao Haiyang Chemical Co., Ltd., China) and RP-18 F₂₅₄S silica gel plates (Merck KGaA, Germany). Semipreparative HPLC was carried out on an LC-15C (Shimadzu, Japan) with a Gemini C₁₈ column $(10 \text{ mm} \times 150 \text{ mm}, 5 \mu\text{m}, \text{Phenomenex}, \text{Germany}).$

3.2. Plant material

The tubers of *S. succifera* were collected at Wuzhishan, Hainan Province, China, in 2014, and authenticated by Dr. Yun Kang, Fudan University. A voucher specimen (2014057 G) has been deposited in the Department of Pharmacognosy, Fudan University, China.

3.3. Extraction and isolation

The dried and powdered tubers (5.5 kg) of S. succifera were extracted with 95% EtOH (50 L) at room temperature. After concentration under reduced pressure, the crude extract (798 g) were dissolved in water (2L) to get a homogeneous dispersion and then partitioned successively with petroleum ether (PE) and dichloromethane (CH₂Cl₂) to afford a PE extract (85 g) and a CH₂Cl₂ extract (196 g). The remaining aqueous solution was applied to an HZ818 macroporous resin column eluted with EtOH-H2O (95:5, v/v). The EtOH-H2O (95:5) eluate (99 g) was chromatographed on a silica gel column (200-300 mesh) with a gradient CH₂Cl₂-MeOH (15:1, 10:1, 5:1, 2:1, 0:1, v/v) as eluent to give nine fractions (Fr. A-I). Fr.G was loaded on another HZ818 macroporous resin column eluted with a gradient MeOH-H₂O (5:5, 6:4, 7:3, 1:0, v/v) to yield subfractions G1-G5. Fr.G3 was subjected to C₁₈ CC eluted with MeOH-H₂O (3:7, 4:6, 5:5, 6:4, 1:0, v/v) to give Fr.G3.1 to Fr.G3.5. Fr.G3.2 was further separated on a Sephadex LH-20 CC (MeOH) followed by semipreparative HPLC (MeOH-H₂O, 45:55, v/v, flow rate 3 mL/min) to yield 2 (5.4 mg) and 7 (10.0 mg). Fr.G3.3 was purified by Sephadex LH-20 CC (MeOH) and semipreparative HPLC (MeOH-H₂O, 35:65, v/v, flow rate 3 mL/min) to yield 13 (2.7 mg). Fr.G4 was subjected to silica gel CC (300-400 mesh) eluted with a gradient CH₂Cl₂-MeOH (15:1, 10:1, 8:1. 5:1, v/v) and further purified by semipreparative HPLC (MeOH-0.05% Et₂NH aqueous solution, 40:60, v/v, flow rate 3 mL/min) to give 10 (12.9 mg). Fr.G5 was purified by Sephadex LH-20 CC (MeOH) and semipreparative HPLC (MeOH- H_2O , 55:45, v/v, flow rate 3 mL/min) to yield 1 (2.4 mg). Fr.H was subjected to semipreparative HPLC (MeOH-0.02% Et₂NH aqueous solution, 30:70, v/v, flow rate 3 mL/min) to yield 8 (6.8 mg). Compound 9 (5.1 mg) was obtained from Fr.I using the same procedure as compound 8. The CH₂Cl₂ extract was fractionated on a silica gel CC (200-300 mesh) using a gradient CH₂Cl₂-MeOH (1:0, 50:1, 20:1, 10:1, 1:1, 0:1, v/v) as eluent to give eight fractions (Fr.J-Q). Fr.L was loaded on another silica gel CC (300-400 mesh) eluted with a gradient $\rm CH_2Cl_2–MeOH$ (30:1, 20:1, 10:1, v/v) to give subfractions L1–L6. Fr.L3 was purified by Sephadex LH-20 CC (MeOH) followed by semipreparative HPLC (MeOH-0.02% Et₂NH aqueous solution, 88:12, v/v,

flow rate 3 mL/min) to yield **6** (18.7 mg). Fr.L4 was subjected to C_{18} CC eluted with MeOH–H₂O (1:4, 1:1, 7:3, 9:1, 1:0, v/v) to give Fr.L4.1 to Fr.L4.5. Fr.L4.3 was separated by semipreparative HPLC (flow rate 3 mL/min) eluted with MeOH–0.02% Et₂NH aqueous solution (90:10, v/v), MeOH–0.02% Et₂NH aqueous solution (85:15, v/v), and CH₃CN–0.02% Et₂NH aqueous solution (6:4, v/v) to afford **4** (8 mg), **5** (4.9 mg), and **12** (33 mg), respectively. Fr.L4.4 was purified by Sephadex LH-20 CC (MeOH) followed by semipreparative HPLC (MeOH–0.02% Et₂NH aqueous solution, 60:40 and 95:5 for the purification of **3** and **11**, respectively, v/v, flow rate 3 mL/min) to yield **3** (2.1 mg) and **11** (1.8 mg).

3.4. Spectroscopic data

3.4.1. manshurienine C (1)

Yellow amorphous powder; $[\alpha]25D - 155.0^{\circ}$ (*c* 0.10, MeOH); UV (MeOH) λ_{max} (log ε): 381 (3.59), 323 (3.75), 289 (4.17), 277 (4.22), 267 (4.21), 231 (4.26) nm; IR (KBr) ν_{max} 3734, 3703, 3627, 3595, 1668, 1557, 1417, 1288, 1071 cm⁻¹; ¹H and ¹³C NMR data see Table 1; HRESIMS *m*/*z* 442.1114 [M+H]⁺ (calcd for C₂₂H₁₉NO₉, 442.1133).

3.4.2. 8-oxotetrahy-corydalmine-1-O- β -D-glucopyranoside (2)

Yellow amorphous solid; [α]25D –179.0° (*c* 0.10, MeOH); UV (MeOH) λ_{max} (log ε): 278 (4.07) nm; IR (KBr) ν_{max} 3643, 2938, 2334, 1630, 1618, 1497, 1483, 1331, 1026, 666 cm⁻¹; ¹H and ¹³C NMR data see Table 1; HRESIMS: *m*/*z* 534.1950 [M+H] ⁺ (calcd for C₂₆H₃₁NO₁₁, 534.1970).

3.5. Enzymatic hydrolysis of compounds 1 and 2

The enzymatic hydrolysis was performed based on a previously reported method (Cong et al., 2014). Compounds 1 and 2 (1 mg) were separately dissolved in H₂O (500 µL) and mixed with β -cellulase (2 mg). The mixtures were kept at 37°C for 2 days and then extracted with EtOAc. The EtOAc layer of compound 2 was concentrated to dryness under reduced pressure and redissolved with methanol (1 mL) for optical rotation measurement of the aglycone moiety. The aqueous phases were used for optical rotation measurement and TLC analysis of the sugar moieties. The liberated sugars of 1 and 2 were confirmed as glucose because of the same R_f value (0.35) in TLC (EtOAc: MeOH: H2O: HOAc = 13:3:3:4) as the authentic standards (D-glucose and L-glucose), and were identified as D-form based on their nearly identical optical rotation values to the D-glucose standard ([α]25D +52.6°).

3.6. Antibacterial assay

The isolated compounds were tested for in vitro antibacterial activities against MRSA strains ATCC 43,300 (American Type Culture Collection (ATCC), USA) by the broth microdilution method (Luo et al., 2014). The compounds and vancomycin (positive control) were dissolved in methanol, respectively, diluted in cation-adjusted Mueller Hinton broth (CAMHB) to achieve nine twofold serially diluted concentrations, and then added to the wells of microtiter plates. For inoculum preparation, conidial suspensions were harvested from cultures incubated for 24 h at 37°C in Mueller Hinton Agar medium (MHA). The bacterial suspensions were adjusted to a working inoculum concentration of 1×10^6 CFU/mL. A 100 μ L aliquot of the inoculum was dispensed into the microdilution wells (except for negative control well). The microplates were incubated at 37°C for 20 h. If the compound inhibited bacterial growth, the incubation solution appeared clear. The minimum inhibitory concentration (MIC) was recorded as the lowest concentration at which no bacterial growth was observed.

Declaration of Competing Interest

The authors declare that they have no known competing financial interests or personal relationships that could have appeared to influence the work reported in this paper.

There are no conflicts of interest to declare.

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Appendix A. Supplementary data

Supplementary data associated with this article can be found, in the online version, at https://doi.org/10.1016/j.phytol.2019.10.001.

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