

Effect of N-Alkylation on the Affinities of Analogues of Spiperone for Dopamine D₂ and Serotonin 5-HT₂ Receptors

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Two series of N-substituted spiperone analogues were prepared and evaluated in vitro to measure their affinities for dopamine D₂ and serotonin 5-HT₂ receptors. Substitution of the amide nitrogen with an alkyl group of five carbon units or less resulted in analogues displaying a low selectivity for D₂ compared to 5-HT₂ receptors. However, a moderate improvement in selectivity for D₂ receptors was observed with N-benzylspiperone. Substitution at either the ortho or para position of the benzyl group resulted in a further reduction in affinity for 5-HT₂ receptors and improvement in the selectivity ratio. Examination of N-substituted analogues of spiperone may provide insights into the topography of the antagonist binding region of the 5-HT₂ receptor. The results also suggest that an ¹⁸F-labeled analogue of N-(4-nitrobenzyl)spiperone (4p) may be a suitable tracer for studying D₂ receptors with positron emission tomography since this compound displays a high selectivity for D₂ receptors relative to that of spiperone and N-methylspiperone.

The use of positron emission tomography (PET) to evaluate alterations in neurotransmitter receptors permits the longitudinal study of changes associated with neurological and neuropsychiatric disorders antemortem. A primary limitation of this technique is the availability of only a limited number of positron-emitting radiotracers displaying characteristics appropriate for conducting quantitative PET imaging studies. The desired properties of a suitable PET-based radiotracer include the ability to cross the blood-brain barrier, nanomolar affinity and selectivity for the receptor, and low nonspecific binding.¹ The formation of radiolabeled metabolites should be minimal, and those formed should not cross the blood-brain barrier. The neurotransmitter receptor that has been most extensively studied by PET is the dopamine D₂ receptor. This is due to the availability of a number of potent antipsychotics that are high-affinity antagonists for the D₂ receptor. Several of these drugs have been used as "lead" compounds for the development of PET-based radiotracers. A potential problem with many of the radiotracers currently used is that they have nanomolar affinity for more than one class of receptor. Examples of such compounds are [¹⁸F]spiperone² and [¹⁸F]- or [¹¹C]-N-methylspiperone,³ which display a high affinity for both D₂ and serotonin 5-HT₂ receptors;⁴ [¹⁸F]haloperidol,^{2b,c} which binds potently to both D₂ and σ receptors;⁵ and [¹¹C]SCH 23390, which has nanomolar affinity for dopamine D₁ and serotonin 5-HT₂ and 5-HT_{1C} receptors.^{4c,6}

The lack of dopamine D₂ receptor specificity of spiperone and N-methylspiperone is well established.^{3a,4} Approximately 15-25% of the saturable in vitro binding of these radiotracers to receptors in rat caudate is to 5-HT₂ sites.^{4b,e} The properties of these ligands under in vivo

conditions are less clear. In vivo studies using the 5-HT₂ antagonist ketanserin to block binding of the radiotracer indicated that the specific binding of [³H]spiperone,⁷

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[⁷⁶Br]bromospiperone,⁸ and [¹¹C]-*N*-methylspiperone⁹ in the striatum of rodents occurs predominantly at D₂ sites. However, other investigators using a higher dose of ketanserin observed a reduction in the binding of [³H]spiperone¹⁰ and [³H]-*N*-methylspiperone¹¹ to 70–80% of controls.

A second series of compounds that display high affinity and selectivity for the D₂ receptor are the benzamides, sulpiride and raclopride.¹² [¹¹C]Raclopride has been prepared,¹³ and a number of quantitative PET imaging studies have been carried out using this tracer.¹⁴ Concerns with the use of [¹¹C]raclopride for in vivo imaging comes from the discrepancy between K_d values observed under in vitro and in vivo conditions^{2b,15} and the ability of endogenous dopamine to compete with the radiotracer for the D₂ receptor.¹⁶ These concerns decrease the reliability of estimates of the density of receptors with PET. A number of ¹⁸F-labeled benzamide analogues that display a higher affinity than raclopride for the D₂ receptor have recently been described.¹⁷ It is not yet clear if these analogues satisfy all of the criteria necessary for use in quantitative PET imaging studies.

In contrast to raclopride, endogenous dopamine has been shown not to reduce the in vivo binding of [³H]-*N*-methylspiperone to D₂ receptors.¹⁶ As part of our ongoing research on the development of ¹⁸F-labeled radiotracers for PET, we explored the possibility of modifying the dopamine D₂ and serotonin 5-HT₂ binding affinities of spiperone-based analogues by substituting the amide nitrogen with alkyl groups of increasing steric demand. An analogue of spiperone possessing a reduced affinity for 5-HT₂ receptors while retaining a high affinity for D₂ re-

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ceptors may be more suitable for studying D₂ receptors in vivo with PET as compared with radiolabeled analogues of spiperone or *N*-methylspiperone. A number of analogues containing *N*-alkyl and *N*-haloalkyl groups have been reported to exhibit a high affinity for the D₂ receptor.¹⁸ The work reported previously was initiated to address problems resulting from the multistep synthesis and the concomitant low yields associated with the production of [¹⁸F]-*N*-methylspiperone and [¹⁸F]spiperone for clinical PET imaging studies.^{2b,19} Although the affinities of the reported analogues for the 5-HT₂ receptor were not measured, results of recent studies have shown that [¹⁸F]-*N*-(fluoroethyl)spiperone labels 5-HT₂ receptors in the frontal cortex of primates with an apparent K_d of 0.5 nM,^{18f} indicating that no change in selectivity for D₂ compared to 5-HT₂ receptors was achieved via this substitution. Similarly, both the *E* and *Z* isomers of *N*-([¹²⁵I]iodoallyl)spiperone have a high affinity for the 5-HT₂ receptor in rat frontal cortex.²⁰ In the present study, several *N*-alkyl and *N*-benzyl analogues of spiperone were prepared and evaluated to determine their affinities for both D₂ and 5-HT₂ receptors. Structural features resulting in an increase in the selectivity of *N*-alkylspiperone analogues for D₂ receptors were identified.

Chemistry

The synthesis of the target compounds is outlined in Scheme I. Alkylation of 4-phenyl-1,3,8-triazaspiro[4.5]decan-4-one, 1, with 4-chloro-1,1-(ethylenedioxy)-1-(4-fluorophenyl)butane, 2,²¹ gave the ethylenedioxy analogue of spiperone, 3.^{18b} Acid hydrolysis of 3 followed by alkylation of spiperone with the appropriate alkyl halide or tosyloxy derivative (method A) afforded the desired *N*-alkyl analogue in moderate yield (50–60%). Alternatively, the desired compounds were prepared by alkylation of 3 followed by acid hydrolysis of the ketal (method B). However, no improvement in overall yield was observed with this modification (Table I). Compound 4p was synthesized via the sequence of reactions outlined in Scheme II.

Pharmacology

In vitro radioligand binding competition experiments were carried out to determine the affinities of the synthesized compounds for 5-HT₂ and D₂ receptors. The affinities of the compounds for 5-HT₂ receptors were determined using the pituitary tumor cell line, P11,²² which has been shown to express 5-HT₂ receptors in the absence of D₂ receptors. 5-HT₂ receptors were labeled with [¹²⁵I]I-LSD. Inhibition of the binding of [³H]spiperone to rat striatal tissue made it possible to determine the affinity of the spiperone analogues for D₂ receptors. Data from competition experiments were analyzed using a mathematical modeling program to determine the concentration of unlabeled analogue required to inhibit 50% of the binding of the radioligand (IC₅₀). Dissociation constants (K_i) were calculated from IC₅₀ values using the method of Cheng and Prusoff.²³ The selectivity of the analogues for D₂ and 5-HT₂ receptors is expressed as the ratio of the K_i values for 5-HT₂ and D₂ receptors (K_{i-5HT₂}/K_{i-D₂}). A higher ratio corresponds to a greater selectivity for D₂ receptors.

Results and Discussion

The goal of this study was to determine the structural features required to improve the D₂ vs 5-HT₂ selectivity of spiperone-based analogues. The approach chosen involved the preparation of a series of analogues in which the amide nitrogen of spiperone was substituted with alkyl groups of increasing steric demand. Previous studies have shown this to be a region of bulk tolerance for the binding of spiperone to the D₂ receptor.^{18b-f,20,24} The effect of this substitution on the affinity of spiperone analogues for the 5-HT₂ receptor has not been thoroughly investigated. The

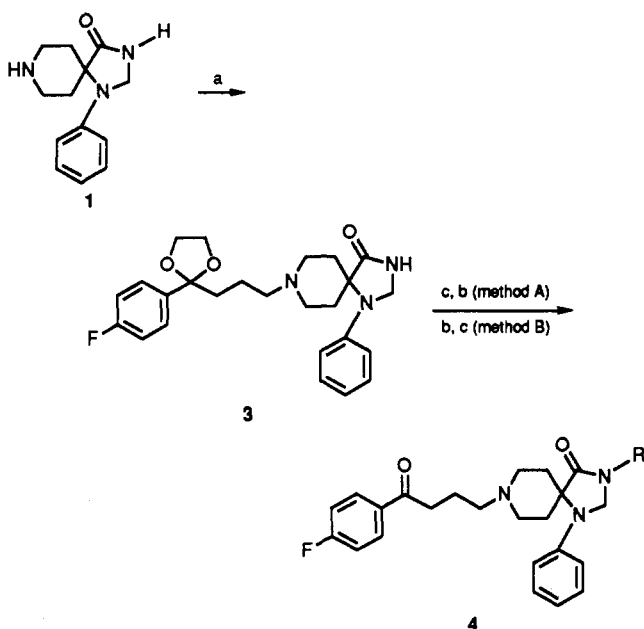
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Table I. Physical Properties, Yields, and Elemental Analyses

no.	method	% yield	mp (°C)	formula	analytical data	recrystallization solvent
4a	A	10	256–259 dec	C ₂₄ H ₂₆ N ₃ O ₂ FCl	C, H, N	ethanol
4b	B	28	224.5–226.5	C ₂₅ H ₃₁ N ₃ O ₂ FCl	C, H, N ^a	acetone
4c	B	56	210.5–211	C ₂₆ H ₃₃ N ₃ O ₂ FCl	C, H, N	ethanol
4d	B	53	215.5–217	C ₂₇ H ₃₅ N ₃ O ₂ FCl	C, H, N	acetone
4e	B	15	216.5–219	C ₂₇ H ₃₅ N ₃ O ₂ FCl	C, H, N	ethanol
4f	B	67	215–215.5	C ₂₈ H ₃₇ N ₃ O ₂ FCl	C, H, N	ethanol
4g	B	65	226–227	C ₂₈ H ₃₇ N ₃ O ₂ FCl	C, H, N	acetone
4h	B	44	217–219	C ₃₀ H ₃₉ N ₃ O ₂ FCl	C, H, N ^b	ethanol
4i	B	45	175–176	C ₃₁ H ₃₆ N ₃ O ₂ FCl	C, H, N	ethanol-ether
4j	B	40	209–209.5	C ₃₀ H ₃₂ N ₃ O ₂ F ₂ Cl	C, H, N	acetone
4k	B	26	199.5–200	C ₃₀ H ₃₂ N ₃ O ₂ F ₂ Cl	C, H, N	acetone
4l	A	40	218–219.5 dec	C ₃₀ H ₃₂ N ₃ O ₂ F ₂ Cl	C, H, N	ethanol-ether
4m	A	59	230–231.5	C ₃₀ H ₃₂ N ₃ O ₂ FCl	C, H, N	ethanol-acetone
4n	A	51	238–240	C ₃₀ H ₃₂ N ₃ O ₂ FCl	C, H, N	ethanol-acetone
4o	A	46	203–204	C ₃₀ H ₃₂ N ₄ O ₂ FCl	C, H, N	acetone
4p	A	45	246–250 dec	C ₃₀ H ₃₂ N ₄ O ₂ FCl	C, H, N	ethanol-ether
4q	A	28	220–220.5	C ₃₁ H ₃₅ N ₃ O ₂ FCl	C, H, N	acetone
4r	A	43	203–203.5	C ₃₁ H ₃₅ N ₃ O ₂ FCl	C, H, N	acetone
4s	A	33	215–216.5	C ₃₁ H ₃₅ N ₃ O ₃ FCl	C, H, N	ethanol-ether
4t	A	43	207–207.5	C ₃₁ H ₃₅ N ₃ O ₃ FCl	C, H, N	ethanol

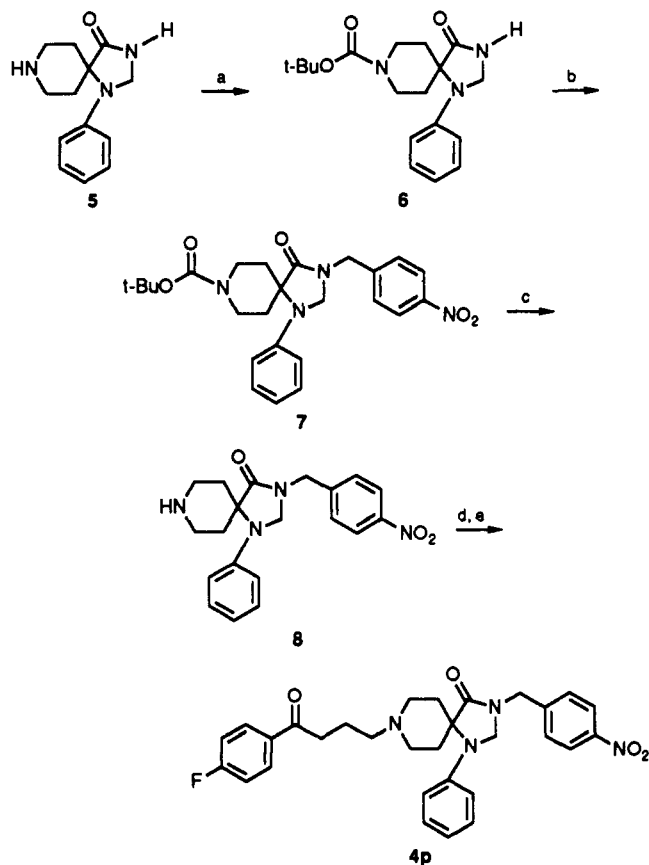
^a C: calcd, 65.28; found, 63.59. ^b C: calcd, 69.02; found, 68.24.

Scheme I^a

^a Reagents: (a) 2/KI/CH₃CN; (b) NaH/RX or ROTs/THF; (c) aqueous HCl/EtOH.

desired analogues were prepared in moderate yield by N-alkylation of spiperone. Alkylation of the ethylenedioxy analogue of spiperone^{18h} followed by acid hydrolysis of the protecting group did not result in a significant improvement in overall yield of the N-alkyl product.

The affinities of compounds for the 5-HT₂ receptor were determined from competition experiments using membrane homogenates prepared from P11 cells. P11 cells are derived from a pituitary tumor that has been shown to express 5-HT₂ receptors, while no detectable binding to D₂ receptors has been observed. [¹²⁵I]-LSD was used to selectively label 5-HT₂ receptors since P11 cells do not express 5-HT_{1c} receptors.²² In vitro binding assays for the D₂ receptor were conducted using rat striatal tissue homogenates and [³H]spiperone. Based on results of previous studies on the binding of [³H]spiperone to membranes from rat striatum, the ratio of D₂ to 5-HT₂ receptors in the striatum is 73:27 (D₂:5-HT₂).⁴⁶ Considering the relative affinities of D₂ and 5-HT₂ receptors for [³H]spiperone and

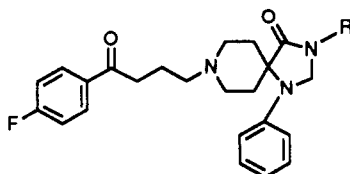
Scheme II^a

^a Reagents: (a) O[CO₂C(CH₃)₃]₂/CH₂Cl₂; (b) NaH/THF/4-NO₂C₆H₄CH₂Br; (c) CF₃COOH; (d) 2/KI/Et₃N/CH₃CN; (e) 3 N HCl/EtOH.

the conditions of our assay system, less than 15% of the binding of [³H]spiperone to striatal membranes was due to binding to 5-HT₂ receptors.⁴⁶ This estimate appears to be validated by the fact that the K_i value calculated for the competition of [³H]spiperone (Table II: 0.058 nM) is in good agreement with the K_d values (0.02–0.05 nM) obtained from Scatchard transformations of direct radio-ligand binding experiments.

The structures of the N-alkyl analogues of spiperone, their physicochemical parameters, and the results of in

Table II. Structures, Physicochemical Properties, and in Vitro Binding Data of Spiperone and Analogues 4a-i

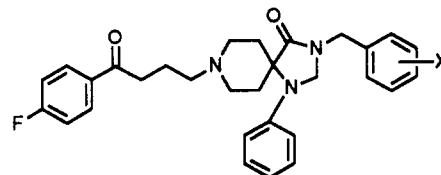


no.	R	π	MR	$\log P^a$	K_i		
					D ₂ (nM) ^b [n]	5-HT ₂ (nM) ^c [n]	5-HT ₂ /D ₂ ^d
-	H	0.00	1.03	2.67 ^e	0.058 ± 0.022 [4]	0.45 ± 0.08 [5]	7.8
4a	CH ₃	0.56	5.70	3.23	0.118 ± 0.045 [4]	0.55 ± 0.16 [5]	4.7
4b	CH ₂ CH ₃	1.02	10.3	3.69	0.057 ± 0.014 [4]	0.31 ± 0.04 [4]	5.4
4c	(CH ₂) ₂ CH ₃	1.55	14.9	4.22	0.029 ± 0.013 [4]	0.096 ± 0.039 [4]	3.3
4d	(CH ₂) ₃ CH ₃	2.05	19.5	4.72	0.036 ± 0.017 [4]	0.41 ± 0.16 [4]	11
4e	CH ₂ CH(CH ₃) ₂	2.05	19.5	4.72	0.039 ± 0.015 [4]	0.13 ± 0.02 [4]	3.3
4f	(CH ₂) ₄ CH ₃	2.55	24.2	5.22	0.062 ± 0.032 [4]	1.1 ± 0.1 [4]	18
4g	CH ₂ CH ₂ CH(CH ₃) ₂	2.55	24.2	5.22	0.063 ± 0.031 [4]	1.3 ± 0.5 [4]	21
4h	CH ₂ Ph	2.01	30.0	4.68	0.023 ± 0.008 [4]	3.2 ± 1.1 [4]	140 ^f
4i	CH ₂ CH ₂ Ph	2.66	34.6	5.33	0.035 ± 0.012 [4]	1.0 ± 0.5 [4]	29

^a Calculated value. ^b Mean K_i value for inhibiting [³H]spiperone binding to D₂ sites ± SE. ^c Mean K_i value for inhibiting [¹²⁵I]-LSD binding to 5-HT₂ sites ± SE. ^d Ratio of K_i values. ^e Octanol-water partition coefficient reported in ref 18a. ^f Indicates *N*-benzylspiperone has a 140-fold higher selectivity for the D₂ receptor vs the 5-HT₂ receptor.

vitro binding studies for each compound are given in Table II. All of the *N*-alkyl analogues displayed a high affinity for the D₂ receptor, a result that is consistent with those of other studies in which the in vitro affinities of a series of *N*-alkyl and *N*-haloalkyl analogues of spiperone for D₂ receptors were reported.^{18a,b,d,h,20} There is no clear trend with respect to the effect of increasing steric demand (MR) or lipophilicity (π) on either the affinity of 5-HT₂ receptors or D₂ vs 5-HT₂ selectivity for this series of compounds. Substitution of the amide nitrogen with an alkyl group with up to four carbon units resulted in analogues that possessed subnanomolar affinity for the 5-HT₂ receptor and low D₂ selectivity (5-HT₂/D₂ ratio ≤ 11). Further extension of the alkyl group resulted in an analogue (4f) that had a lower affinity for the 5-HT₂ receptor (K_i ~ 1 nM). Branching of the alkyl group appears to have little effect on binding to 5-HT₂ receptors since compounds 4f and 4g had similar affinities for 5-HT₂ receptors. This is further exemplified by the isobutyl analogue, 4e, which had a K_i value similar to that of the *n*-propyl derivative, 4c, as opposed to the corresponding *n*-butyl analogue, 4d. ANOVA and Dunnett tests were used to compare the affinities of compounds 4a-i for 5-HT₂ receptors. The only analogue found to differ significantly ($p < 0.01$) with respect to its affinity for 5-HT₂ receptors was *N*-benzylspiperone, 4h. This analogue displayed the lowest affinity for 5-HT₂ receptors (K_i = 3.2 nM) and a relatively high selectivity for D₂ receptors (5-HT₂/D₂ ratio = 140). No significant differences in affinity for D₂ receptors were found for the compounds listed in Table II. An unexpected result was the relatively high affinity (K_i = 1.0 nM) and low D₂ selectivity (5-HT₂/D₂ ratio = 29) of the *N*-phenylethyl analogue, 4i. This suggests that the selectivity of this series of compounds for D₂ and 5-HT₂ receptors is not determined by the values of MR or π for the substituent since the phenylethyl group possessed the highest value for these two parameters.²⁵ The differences in the affinities of 4h and 4i for 5-HT₂ receptors may be attributed to the increased conformational flexibility of 4i due to the presence of the additional methylene group.

The results of the initial study indicate that an improvement in the dopaminergic/serotonergic selectivity of

Table III. In Vitro Binding Data for the Substituted *N*-Benzylspiperone Analogues

no.	X	K_i		
		D ₂ (nM) ^a [n]	5-HT ₂ (nM) ^b [n]	5-HT ₂ /D ₂ ^c
4h	H	0.023 ± 0.008 [3]	3.2 ± 1.1 [3]	140
4j	2'-F	0.043 ± 0.008 [3]	4.4 ± 0.7 [3]	100
4k	3'-F	0.023 ± 0.013 [3]	5.3 ± 2.5 [3]	230
4l	4'-F	0.069 ± 0.032 [4]	21 ± 6 [6]	300
4m	2'-I	0.063 ± 0.028 [3]	19 ± 4 [3]	300
4n	4'-I	0.213 ± 0.099 [3]	89 ± 19 [3]	420
4o	2'-NO ₂	0.120 ± 0.097 [3]	0.68 ± 0.29 [3]	5.7
4p	4'-NO ₂	0.119 ± 0.054 [3]	14 ± 3 [6]	120
4q	2'-CH ₃	0.033 ± 0.013 [3]	9.6 ± 4.7 [3]	290
4r	4'-CH ₃	0.067 ± 0.035 [3]	37 ± 12 [3]	550
4s	2'-OCH ₃	0.047 ± 0.027 [3]	1.9 ± 0.5 [3]	40
4t	4'-OCH ₃	0.087 ± 0.059 [3]	12 ± 2 [3]	140

^{a-c} Refer to Table II.

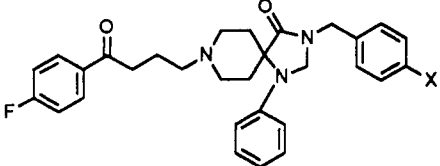
spiperone could be achieved by substitution of the amide nitrogen with a benzyl group. To determine whether there are stereoelectronic effects with respect to D₂ or 5-HT₂ binding, a series of substituted *N*-benzylspiperone analogues was prepared. The structures of the analogues and the results of in vitro binding studies are shown in Table III. Substituents were chosen on the basis of physicochemical parameters (σ , F , R , MR, π) as well as the availability of suitable starting materials. The binding of substituted *N*-benzyl derivatives to D₂ receptors was unaffected by the nature of the substituent since all representatives of this series of compounds had a subnanomolar affinity for the D₂ receptor. There was, however, a pronounced substituent effect with respect to binding to 5-HT₂ receptors. A comparison of fluorine-substituted analogues revealed that para substitution was preferred over the corresponding ortho and meta positions with respect to increasing the K_i value for the inhibition of the binding of [¹²⁵I]-LSD [4'-F (21 nM) > 3'-F (5.3 nM) ~ 2'-F (4.4 nM)]. Para-substituted analogues consistently exhibited higher K_i values for the inhibition of the binding

(25) Hansch, C.; Leo, A.; Unger, S. H.; Kim, K. H.; Nikaitani, D.; Lien, E. J. "Aromatic" Substituent Constants for Structure-Activity Correlations. *J. Med. Chem.* 1973, 16, 1207-1216.

of [125 I]-LSD compared to the corresponding ortho-substituted derivatives. No correlation was found between the electronic substituent parameters (σ , F , or R) or the molar refractivity (MR) and the affinity for 5-HT₂ receptors or the selectivity ratio. Analogues possessing a lipophilic substituent (+ π value) generally possessed a lower affinity for the 5-HT₂ receptor relative to the *N*-benzyl analogue, 4h. A linear relationship was observed between the Hansch lipophilicity constant π and the K_i value for inhibiting the binding of [125 I]-LSD to 5-HT₂ receptors for both the ortho- and para-substituted analogues (data not shown). However, the limited number of analogues evaluated precludes a detailed structure-activity relationship analysis. The possible correlation between π and K_i for inhibiting the binding of [125 I]-LSD suggests the presence of polar or charged amino acid residues in the spirodecanone binding region of the 5-HT₂ receptor. As with other members of the G protein-coupled receptor family, the 5-HT₂ receptor is thought to contain seven transmembrane regions that together form the ligand-binding domain of the receptor.²⁶ The results of our binding studies in concert with receptor model-building studies may provide insights into the orientation and mechanism of binding of spiperone to the ligand-binding region of the 5-HT₂ receptor.

The primary goal of the present work was to identify spiperone analogues that displayed an improved selectivity for D₂ vs 5-HT₂ receptors. The results of this study suggest that several 4'-substituted *N*-benzylspiperone analogues may possess this property. A potential problem with using these analogues for PET studies is the relatively high lipophilicity of the compounds, a feature that may interfere with the ability of an 18 F-labeled analogue to cross the blood-brain barrier. Previous studies with 18 F-labeled *N*-alkyl- and *N*-(fluoroalkyl)spiperone analogues revealed an optimal log *P* range of 2.67–4.00 for high initial brain extraction of this series of compounds.^{18a,b} The initial brain uptake of the *N*-substituted spiperone analogues began to decrease when the log *P* exceeded a value of 4.20.^{18b} However, a suitable brain uptake and striatum:cerebellum ratio was observed with analogues possessing a log *P* as high as 4.54.^{18b} A comparison of the calculated log *P* values of the 4'-substituted *N*-benzyl analogues (Table IV)²⁷ indicates that compound 4p may be appropriate for further investigation. This analogue displays a high D₂:5-HT₂ selectivity (5-HT₂/D₂ = 120) as compared with spiperone (5-HT₂/D₂ = 7.8) and *N*-methylspiperone (5-HT₂/D₂ = 4.7). It also has a calculated log *P* value (4.40) that is reasonably close to 4.20, the value at which a decline in brain uptake was observed with the *N*-alkylspiperone analogues.^{18a,b} Although 4n displays a high D₂:5-HT₂ selectivity (5-HT₂/D₂ = 420), the relatively high log *P* value of this analogue may preclude the use of the corresponding 123 I-labeled analogue as a single-photon emission computed tomography (SPECT) tracer. However, a 125 I-labeled analogue could prove to be useful as an in vitro ligand for studies of D₂ receptors.

Table IV. Structures and Physicochemical Parameters of the *N*-(4'-Substituted-benzyl)spiperone Analogues



no.	X	π_x	log <i>P</i> ^a	σ_p	<i>F</i> _x	<i>R</i> _x	MR _x
4h	H	0.00	4.68	0.00	0.00	0.00	1.03
4l	F	0.14	4.82	0.06	0.43	-0.34	0.92
4n	I	1.12	5.80	0.18	0.40	-0.19	13.9
4p	NO ₂	-0.28	4.40	0.78	0.67	0.16	7.36
4r	CH ₃	0.56	5.24	-0.17	-0.04	-0.13	5.65
4t	OCH ₃	-0.02	4.66	-0.27	0.26	-0.51	7.87

^a Calculated value.²⁷

In conclusion, two series of *N*-alkylspiperone and substituted *N*-benzylspiperone analogues were prepared, and their affinities for both D₂ and 5-HT₂ receptors were determined. Substitution of the amide nitrogen with a benzyl group resulted in an analogue displaying a moderate selectivity for the dopamine D₂ receptor. A further improvement in D₂:5-HT₂ selectivity was observed with compounds containing substitutions on the 4'-position of the benzyl group. The results indicate that 18 F-labeled 4p may be a suitable analogue for measuring D₂ receptor density in vivo with PET.

Experimental Section

Melting points were determined in an open capillary tube with a Mel-Temp melting point apparatus and are uncorrected. IR spectra were determined on a Perkin-Elmer 1600 Series FT-IR. 1 H NMR spectra were recorded on a Varian EM-360L NMR spectrometer. Microanalyses were performed by Atlantic Microlab, Inc., Norcross, GA, and were within $\pm 0.4\%$ of the theoretical value unless otherwise noted (Table I). Tetrahydrofuran was distilled from sodium metal immediately prior to use. All other reagents and solvents were used without further purification. The sample of spiperone used in the in vitro binding assay was purchased from Janssen Life Sciences Products, 40 Kingsbridge Rd, Piscataway, NJ 08854.

1-Phenyl-8-(*tert*-butyloxycarbonyl)-1,3,8-triazaspiro[4.5]decan-4-one (6). A solution of di-*tert*-butyl dicarbonate (10.21 g, 45.36 mmol) in dichloromethane (25 mL) was added in portions to a stirred suspension of 1-phenyl-1,3,8-triazaspiro[4.5]decan-4-one (5; 10 g, 43.2 mmol) in dichloromethane (50 mL), and the reaction mixture was stirred at ambient temperature for 20 h. Volatile components were removed in vacuo, and the residue was suspended in pentane and filtered to yield the crude product as a tan solid. Purification by silica gel column chromatography (dichloromethane-acetone, 3:1) gave 6 as fine white needles (12.17 g, 85%): mp 209–212 °C; NMR (CDCl₃/TMS) δ 1.45 (s, 9 H), 1.90–2.90 (complex m, 4 H), 3.10–4.20 (complex m, 4 H), 4.70 (s, 2 H), and 6.50–7.70 (complex m, 6 H); IR (film) 3200, 3120, 3080, 2990, 2950, 1720, 1700, 1610, 1520, 1470, 1430, 1390, 1370, 1290, 1250, 1190, 1155, 1110, 1100, 1050, 1040, 1000, 975, 950, 925, 870, 785, 775, and 750 cm⁻¹. Anal. (C₁₈H₂₅N₃O₃) C, H, N.

1-Phenyl-3-(*p*-nitrobenzyl)-8-(*tert*-butyloxycarbonyl)-1,3,8-triazaspiro[4.5]decan-4-one (7). Solid sodium hydride (60% dispersion in mineral oil; 0.46 g, 11.5 mmol) was added in portions to a stirred solution of 6 (4 g, 12.1 mmol) in tetrahydrofuran (15 mL), and the reaction mixture was stirred at ambient temperature for 5 min. A solution of *p*-nitrobenzyl bromide (2.61 g, 12.1 mmol) in tetrahydrofuran (5 mL) was added dropwise, and the reaction mixture stirred at ambient temperature for 4 h. Volatile components were removed in vacuo, the resultant residue was dissolved in saturated aqueous sodium bicarbonate (30 mL), and the mixture extracted with dichloromethane (2 \times 50 mL). The combined organic layers were dried (sodium sulfate) and concentrated in vacuo to give a yellow oil. Purification by

- (26) (a) Dohlman, H. G.; Caron, M. G.; Lefkowitz, R. J. A Family of Receptors Coupled to Guanine Nucleotide Regulatory Proteins. *Biochemistry* 1987, 26, 2657–2664. (b) Pritchett, D. B.; Bach, A. W. J.; Wozny, M.; Taleb, O.; Dal Toso, R.; Shih, J. C.; Seeburg, P. H. Structure and Functional Expression of Cloned Rat Serotonin 5HT-2 Receptor. *EMBO J.* 1988, 7, 4135–4140. (c) Julius, D.; Huang, K. N.; Livelli, T. J.; Axel, R.; Jessel, T. M. The 5HT₂ Receptor Defines a Family of Structurally Distinct but Functionally Conserved Serotonin Receptors. *Proc. Natl. Acad. Sci. U.S.A.* 1990, 87, 928–932. (27) Leo, A.; Hansch, C.; Elkins, D. Partition Coefficients and Their Uses. *Chem. Rev.* 1971, 71, 525–554.

silica gel column chromatography (dichloromethane–acetone, 9.5:0.5) yielded the product as a yellow foam (4.42 g, 78.4%): mp softens at 56–59 °C; NMR (CDCl₃/TMS) δ 1.50 (s, 9 H), 2.00–2.90 (complex m, 4 H), 3.10–4.25 (complex m, 4 H), 4.55 (s, 2 H), 4.65 (s, 2 H), 6.50–7.55 (complex m, 7 H), and 8.20 (d, J = 9 Hz, 2 H); IR (film) 2960, 2910, 2850, 1700, 1610, 1535, 1510, 1475, 1425, 1390, 1375, 1350, 1290, 1250, 1180, 1160, 1100, 1070, 1010, 1000, 970, 910, 850, 800, 750, and 740 cm⁻¹.

***N*-(4'-Nitrobenzyl)spiperone (4p).** A solution of 7 (4.42 g, 9.48 mmol) in trifluoroacetic acid (20 mL) was stirred at ambient temperature for 2 h. Volatile components were removed in vacuo to give a dark oil that was dissolved in saturated aqueous sodium bicarbonate (25 mL) adjusted to pH 10 by addition of 10% aqueous sodium hydroxide. The aqueous mixture was extracted with dichloromethane (3 \times 75 mL), and the combined organic layers were dried (sodium sulfate) and concentrated in vacuo to give 8 as a dark yellow foam (3.16 g, 91%).

A solution of 8 (1 g, 2.73 mmol), 2 (0.634 g, 2.59 mmol), KI (0.453 g, 2.73 mmol), and triethylamine (0.304 g, 3.00 mmol) in acetonitrile (5 mL) was stirred at reflux for 20 h. Volatile components were removed in vacuo to give a dark yellow residue that was dissolved in saturated aqueous sodium bicarbonate (20 mL) and extracted with dichloromethane (2 \times 25 mL). The combined organic layers were dried (sodium sulfate) and concentrated in vacuo to give a yellow foam that was dissolved in absolute ethanol (20 mL). A 3 N solution of aqueous HCl (4 mL) was added, and the reaction mixture was stirred at reflux for 15 min. Volatile components were removed in vacuo to give a yellow oil that was dissolved in saturated aqueous sodium bicarbonate (25 mL) and extracted with dichloromethane (2 \times 20 mL). The combined organic layers were dried (sodium sulfate) and concentrated in vacuo to give a dark yellow oil that was purified by silica gel column chromatography (dichloromethane–acetone–NH₄OH, 8:2:0.1) to afford 4p as a yellow foam. The product was converted to the hydrochloride salt by treatment with a saturated HCl–ethyl acetate solution; recrystallization from ethanol–ether afforded 4p as yellow prisms (0.83 g, 60%): NMR (free base, CDCl₃/TMS) δ 1.10 to 3.25 (complex m, 14 H), 4.50 (s, 2 H), 4.60 (s, 2 H), 6.50–7.55 (complex m, 9 H), and 7.65–8.30 (m, 4 H); IR (free base, neat) 3050, 3400, 3300, 1700, 1680, 1600, 1520, 1500, 1470, 1420, 1375, 1350, 1300, 1270, 1230, 1150, 1110, 1060, 1000, 980, 850, 810, 750, 740, 700, and 660 cm⁻¹.

General Methods for the Synthesis of the *N*-Alkyl- and *N*-Benzylspiperone Analogues. **Method A. *N*-(4'-Fluorobenzyl)spiperone (4l).** Solid sodium hydride (0.096 g of 60% NaH in oil, 2.40 mmol) was added to a stirred suspension of spiperone (1 g, 2.53 mmol) in tetrahydrofuran (5 mL), and the reaction mixture was stirred at ambient temperature for 5 min. A solution of *p*-fluorobenzyl bromide (0.454 g, 2.40 mmol) in tetrahydrofuran (1 mL) was added dropwise, and the reaction mixture was stirred at ambient temperature for 2 h. Volatile components were removed in vacuo to give a yellow oil that was dissolved in dichloromethane (25 mL) and washed with saturated aqueous sodium bicarbonate (2 \times 20 mL). The organic layer was dried (sodium sulfate) and concentrated in vacuo to give a yellow solid that was purified by silica gel column chromatography (dichloromethane–acetone–NH₄OH, 85:15:0.1) to afford 4l as a colorless foam (0.484 g, 40%), which was then treated with an HCl-saturated solution of ethyl acetate to give an analytical sample of 4l (HCl salt) as a fluffy white solid: NMR (free base, CDCl₃/TMS) δ 1.40–3.25 (complex m, 14 H), 4.50 (s, 4 H), 6.50–7.35 (complex m, 11 H), and 7.95 (dd, J = 5 Hz, 2 H); IR (free base, film) 3100, 3080, 2950, 2850, 2280, 1700, 1600, 1500, 1475, 1430, 1380, 1300, 1275, 1230, 1160, 1125, 1100, 1075, 1000, 925, 850, 780, 750, 740, 710, and 660 cm⁻¹.

***N*-Methylspiperone (4a):** NMR (HCl salt, CF₃COOH/TMS) δ 1.90–4.30 (complex m, 19 H) and 6.85–8.30 (complex m, 9 H); IR (HCl salt, KBr) 3072, 2964, 2936, 2663, 2568, 2491, 2415, 1702, 1678, 1599, 1506, 1485, 1458, 1443, 1405, 1379, 1326, 1288, 1227, 1156, 1098, 1084, 1054, 1002, 957, 891, 854, 752, 699, and 584 cm⁻¹.

***N*-(2'-Iodobenzyl)spiperone (4m):** NMR (HCl salt, CDCl₃/TMS) δ 1.35–2.45 (complex m, 5 H), 2.70–4.00 (complex m, 10 H), 4.38 (d, J = 7 Hz, 4 H), 6.10–7.15 (complex m, 10 H), and 7.20–7.90 (complex m, 3 H); IR (HCl salt, KBr) 3448, 3059, 2951, 2374, 2346, 1702, 1696, 1599, 1508, 1474, 1466, 1458, 1438, 1391, 1376, 1296, 1282, 1259, 1226, 1209, 1157, 1095, 826, 746, and

695 cm⁻¹.

***N*-(4'-Iodobenzyl)spiperone (4n):** NMR (HCl salt, CF₃COOH/TMS) δ 1.90–2.85 (complex m, 6 H), 2.90–3.45 (complex m, 4 H), 3.50–4.15 (complex m, 4 H), 4.45 (s, 2 H), 5.05 (s, 2 H), and 6.40–7.90 (complex m, 13 H); IR (HCl salt, KBr) 3448, 3059, 2966, 2799, 2363, 1702, 1688, 1598, 1508, 1560, 1542, 1508, 1499, 1475, 1451, 1400, 1389, 1375, 1345, 1292, 1221, 1211, 1160, 1098, 1007, 990, 816, 790, 746, 695, 670, and 600 cm⁻¹.

***N*-(2'-Nitrobenzyl)spiperone (4o):** NMR (HCl salt, CDCl₃/TMS) δ 1.30–2.50 (complex m, 6 H), 2.65–4.10 (complex m, 9 H), 4.35 (s, 2 H), 4.60 (s, 2 H), and 6.20–7.70 (complex m, 13 H); IR (HCl salt, KBr) 3448, 2472, 2378, 2346, 1713, 1692, 1599, 1535, 1508, 1473, 1391, 1369, 1346, 1295, 1280, 1253, 1211, 1158, 826, 748, 696, and 602 cm⁻¹.

***N*-(2'-Methylbenzyl)spiperone (4q):** NMR (HCl salt, CDCl₃/TMS) δ 1.10–2.50 (complex m, 8 H), 2.65–3.85 (complex m, 10 H), 4.28 (d, J = 6 Hz, 4 H), 6.20–7.10 (complex m, 11 H), and 7.25–7.75 (complex m, 2 H); IR (HCl salt, KBr) 3448, 3062, 2951, 2371, 2346, 1701, 1694, 1599, 1508, 1474, 1390, 1375, 1298, 1282, 1265, 1209, 1157, 826, 748, 696, 668, and 600 cm⁻¹.

***N*-(4'-Methylbenzyl)spiperone (4r):** NMR (HCl salt, CDCl₃/TMS) δ 1.20–2.60 (complex m, 8 H), 2.65–3.90 (complex m, 10 H), 4.33 (d, J = 2 Hz, 4 H), 6.35–7.10 (complex m, 11 H), and 7.25–7.80 (complex m, 2 H); IR (HCl salt, KBr) 3448, 2906, 2530, 2365, 2346, 1704, 1687, 1654, 1599, 1560, 1508, 1500, 1468, 1458, 1389, 1375, 1291, 1223, 1160, 1099, 1004, 990, 818, and 745 cm⁻¹.

***N*-(2'-Methoxybenzyl)spiperone (4s):** NMR (HCl salt, CDCl₃/TMS) δ 1.20–2.55 (complex m, 6 H), 2.60–3.90 (complex m, 12 H), 4.25 (s, 4 H), 6.15–7.00 (m, 11 H), and 7.10–7.60 (complex m, 2 H); IR (HCl salt, KBr) 3449, 3063, 2951, 2373, 2346, 1702, 1694, 1599, 1508, 1496, 1475, 1391, 1376, 1295, 1282, 1247, 1209, 1158, 1121, 1097, 1051, 1028, 826, 748, 696, 669, and 600 cm⁻¹.

***N*-(4'-Methoxybenzyl)spiperone (4t):** NMR (HCl salt, CDCl₃/TMS) δ 1.25–2.55 (complex m, 5 H), 2.60–3.95 (complex m, 13 H), 4.25 (s, 4 H), 6.15–7.05 (m, 11 H), and 7.10–7.80 (complex m, 2 H); IR (HCl salt, KBr) 3449, 3062, 2962, 2372, 2346, 1700, 1685, 1598, 1512, 1467, 1390, 1375, 1297, 1282, 1247, 1213, 1180, 1159, 1034, 1003, 990, 824, 745, 696, 656, and 601 cm⁻¹.

Method B. *N*-Ethylspiperone (4b). Solid sodium hydride (0.091 g of 60% suspension in oil, 2.27 mmol) was added to a stirred suspension of 3 (1 g, 2.27 mmol) in tetrahydrofuran (5 mL), and the reaction mixture was stirred at ambient temperature for 5 min. Iodoethane (0.551 g, 3.41 mmol) was added, and the reaction mixture was stirred at reflux for 18 h. Volatile components were removed in vacuo, and the residue was dissolved in absolute ethanol (10 mL) and treated with a 3 M aqueous solution of HCl (2 mL). The mixture was stirred at reflux for 15 min, volatile components were removed in vacuo, and the mixture was dissolved in saturated aqueous sodium bicarbonate (35 mL) and extracted with dichloromethane (2 \times 50 mL). The combined organic layers were dried (sodium sulfate) and concentrated in vacuo to give an oil that was purified by silica gel column chromatography (dichloromethane–acetone–NH₄OH, 4:2:0.1) to afford 4b as a viscous oil (0.612 g, 63.7%), which was then treated with an HCl-saturated solution of ethanol to give an analytical sample of 4b (HCl salt) as an amorphous, white solid: NMR (HCl salt, CF₃COOH/TMS) δ 1.34 (t, J = 7 Hz, 3 H), 1.8–4.15 (complex m, 18 H), and 6.35–7.75 (complex m, 9 H); IR (HCl salt, KBr) 3104, 3068, 2966, 2938, 2884, 2661, 2475, 2405, 1695, 1680, 1599, 1576, 1506, 1472, 1446, 1393, 1380, 1299, 1227, 1157, and 1098 cm⁻¹.

***N*-Propylspiperone (4c):** NMR (HCl salt, CF₃COOH/TMS) δ 0.95 (t, J = 6, 7 Hz, 3 H), 1.30–2.75 (complex m, 10 H), 2.80–4.00 (complex m, 11 H), 6.30–6.90 (complex m, 3 H), and 6.95–7.80 (complex m, 6 H); IR (HCl salt, KBr) 3066, 2960, 2934, 2876, 2470, 2417, 1699, 1684, 1599, 1507, 1476, 1392, 1372, 1301, 1273, 1225, 1191, 1158, 1061, 1012, 974, 888, 841, 824, 746, and 697 cm⁻¹.

***N*-Butylspiperone (4d):** NMR (HCl salt, DMSO-CDCl₃/TMS) δ 0.70–2.55 (complex m, 11 H), 2.60–4.20 (complex m, 14 H), 4.45 (s, 1 H), 6.10–7.05 (complex m, 7 H), 7.10–7.70 (complex m, 2 H); IR (HCl salt, KBr) 3102, 3074, 2960, 2932, 2653, 2478, 2418, 1693, 1687, 1598, 1507, 1479, 1371, 1303, 1278, 1228, 1215, 1158, and 1101 cm⁻¹.

***N*-(2-Methylpropyl)spiperone (4e):** NMR (HCl salt, CF₃COOH/TMS) δ 1.10 (d, J = 6 Hz, 6 H), 1.50–4.20 (complex m, 19 H), 6.50–7.00 (m, 2 H), 7.10–7.45 (m, 5 H), and 7.50–7.90

(complex m, 2 H); IR (HCl salt, KBr) 3070, 2962, 2874, 2468, 2413, 1698, 1682, 1599, 1507, 1476, 1394, 1373, 1296, 1222, 1158, 974, 840, 748, and 698 cm^{-1} .

N-Pentylspiperone (4f): NMR (HCl salt, $\text{CF}_3\text{COOH}/\text{TMS}$) δ 0.80–1.20 (m, 3 H), 1.25–3.15 (complex m, 13 H), 3.20–4.20 (complex m, 11 H), 6.50–7.10 (m, 2 H), 7.15–7.40 (m, 5 H), and 7.45–8.10 (m, 2 H); IR (HCl salt, KBr) 3067, 2958, 2930, 2862, 2813, 2655, 2476, 2407, 1691, 1599, 1508, 1480, 1444, 1372, 1302, 1276, 1229, 1215, 1159, 1101, 1056, 974, 749, and 697 cm^{-1} .

N-(3-Methylbutyl)spiperone (4g): NMR (HCl salt, $\text{CF}_3\text{COOH}/\text{TMS}$) δ 1.10 (d, $J = 5$ Hz, 6 H), 1.35–3.10 (complex m, 10 H), 3.15–4.20 (complex m, 11 H), 6.60–7.05 (m, 2 H), 7.10–7.40 (m, 5 H), and 7.45–8.00 (m, 2 H); IR (HCl salt, KBr) 3103, 3065, 2961, 2873, 2811, 2652, 2478, 2418, 1691, 1599, 1507, 1479, 1371, 1303, 1280, 1228, 1214, 1158, 1101, 1000, 973, 888, 845, 820, 750, 698, 587, and 567 cm^{-1} .

N-Benzylspiperone (4h): NMR (HCl salt, CDCl_3/TMS) δ 1.30–3.90 (complex m, 15 H), 4.30 (s, 4 H), and 5.95–7.60 (complex m, 14 H); IR (HCl salt, KBr) 3063, 3032, 2956, 2883, 2642, 2470, 2364, 2335, 1701, 1694, 1599, 1506, 1473, 1390, 1374, 1295, 1281, 1267, 1210, 1158, 1002, 985, 826, 748, and 704 cm^{-1} .

N-Phenethylspiperone (4i): NMR (HCl salt, CDCl_3/TMS) δ 1.10–3.70 (complex m, 19 H), 4.20 (s, 2 H), 5.80–7.00 (complex m, 12 H), and 7.10–7.60 (complex m, 2 H); IR (HCl salt, KBr) 3063, 3026, 2952, 2647, 2488, 2418, 1697, 1689, 1600, 1506, 1472, 1378, 1300, 1274, 1230, 1157, 972, 827, 745, and 702 cm^{-1} .

N-(2'-Fluorobenzyl)spiperone (4j): NMR (HCl salt, CDCl_3/TMS) δ 1.40–2.55 (complex m, 5 H), 2.65–4.20 (complex m, 10 H), 4.40 (s, 4 H), and 6.00–7.70 (complex m, 13 H); IR (HCl salt, KBr) 3455, 3067, 2957, 2471, 2374, 2472, 2375, 2339, 1710, 1692, 1600, 1506, 1494, 1474, 1454, 1391, 1375, 1293, 1279, 1231, 1210, 1159, 1109, 1098, 1003, 984, 826, 761, and 748 cm^{-1} .

N-(3'-Fluorobenzyl)spiperone (4k): NMR (HCl salt, CDCl_3/TMS) δ 1.20–2.55 (complex m, 5 H), 2.60–4.00 (complex m, 10 H), 4.35 (s, 4 H), and 5.80–7.65 (complex m, 13 H); IR (HCl salt, KBr) 3457, 2362, 1706, 1693, 1599, 1507, 1474, 1457, 1391, 1374, 1293, 1210, 1158, 1136, 826, 795, 749, and 696 cm^{-1} .

In Vitro Binding Assays. In vitro binding assays for rat D_2 receptors were performed using rat striatal homogenates suspended in 50 mM Tris-HCl/150 mM NaCl/10 mM EDTA buffer (pH 7.4). The assay volume was 1 mL, and the concentration of [^3H]spiperone was approximately 1.0 nM. Assays were incubated at 37 °C for 45 min, and samples were then filtered through Whatman GF/B filters. The filters were washed with 10 mL of 10 mM Tris-HCl/150 mM NaCl (pH 7.4). (+)-Butaclamol (2 μM) was used to define nonspecific binding. Filters were incubated

overnight in 3.0 mL of Econolite scintillation fluid, and radioactivity was determined by scintillation spectroscopy at an efficiency for tritium of 30%.

The affinities of compounds for the 5-HT₂ receptor were determined in competition experiments with 0.5 nM [^{125}I]I-LSD in 50 mM Tris-HCl (pH 7.4) using membranes from P11 cells.²² The assay volume was 100 μL , and 1 mM ketanserin was used to define nonspecific binding. Samples were incubated at 37 °C for 60 min and were filtered through Schleicher and Schuell filters coated with 3% polyethylenimine. Filters were washed with 10 mL of 50 mM Tris-HCl (pH 7.4), and radioactivity was determined using a Beckman 4000 gamma counter.

Data were analyzed using the mathematical modeling program FITCOMP available through the National Institutes of Health-sponsored PROPHET computer system. The K_i of each compound was calculated using the following equation:²³ $K_i = \text{IC}_{50}/(1 - [L]/K_d)$, where IC_{50} = the concentration of the unlabeled analogue required to inhibit 50% of radioligand binding, $[L]$ = the concentration of the radioligand used in the assay, and K_d = the dissociation constant of the radioligand (0.05 nM for [^3H]spiperone, D_2 sites, and 1.6 nM for [^{125}I]I-LSD, 5-HT₂ sites). The $D_2/5\text{-HT}_2$ selectivity (5-HT₂/ D_2 ratio) of each analogue is expressed as the ratio of the K_i values at 5-HT₂ and D_2 receptors ($K_{i,5\text{-HT}_2}/K_{i,D_2}$).

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Registry No. 1, 1021-25-6; 2, 3308-94-9; 3, 54080-21-6; 4a, 3725-68-6; 4a base, 87539-19-3; 4b, 102504-73-4; 4b base, 104066-91-3; 4c, 138091-55-1; 4c base, 104066-92-4; 4d, 138091-56-2; 4d base, 114115-90-1; 4e, 138091-57-3; 4e base, 138091-58-4; 4f, 138091-59-5; 4f base, 138091-60-8; 4g, 138091-61-9; 4g base, 138091-62-0; 4h, 138091-63-1; 4h base, 138091-64-2; 4i, 138091-65-3; 4i base, 138091-66-4; 4j, 138091-67-5; 4j base, 138091-68-6; 4k, 138091-69-7; 4k base, 138091-70-0; 4l, 138091-71-1; 4l base, 138091-72-2; 4m, 138091-73-3; 4m base, 138091-74-4; 4n, 138091-75-5; 4n base, 138091-76-6; 4o, 138091-77-7; 4o base, 138091-78-8; 4p, 138091-79-9; 4p base, 138091-80-2; 4q, 138091-81-3; 4q base, 138091-82-4; 4r, 138091-83-5; 4r base, 138091-84-6; 4s, 138091-85-7; 4s base, 138091-86-8; 4t, 138091-87-9; 4t base, 138091-88-0; 6, 138091-52-8; 7, 138091-53-9; 8, 138091-54-0; di-*tert*-butyl dicarbonate, 24424-99-5; *p*-nitrobenzyl bromide, 100-11-8; *p*-fluorobenzyl bromide, 459-46-1.