Synthesis of a simplified analogue of eleutherobin *via* a Claisen rearrangement and ring closing metathesis strategy[†]

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The enantioselective synthesis of a simplified eleutherobin analogue 7 by ring closing metathesis (RCM) of the 2,9divinyl-substituted tetrahydro-oxonin 5 is described; the analogue 7 and an advanced intermediate 15 revealed microtubule stabilising properties in the micromolar range.

Eleutherobin 1 (Fig. 1) has emerged as an exciting cytotoxic compound and was shown to possess microtubule stabilising properties similar to that of paclitaxel.^{1,2} Its novel molecular architecture and its scarce availability from natural sources have already prompted the total synthesis of eleutherobin by both the Nicolaou³ and Danishefsky groups.⁴

We demonstrate that a nine-membered medium ring lactone **3** available from our Claisen ring expansion methodology (Scheme $1\ddagger)^{5,6}$ can be elaborated to a simplified analogue of eleutherobin **7** that shows microtubule stabilising activity.

The design of our target was based upon comprehensive SAR studies on the sarcodictyins,⁷ a class of compounds structurally related to eleutherobin. Thus, we elected to keep the crucial urocanic acid side chain attached to the core bicycle (see structure 7). A key intermediate in the synthesis would be the aldehyde 4,⁶ which can be readily elaborated to a 2,9-divinyl derivative for ring closing metathesis^{8,9} to the core bicycle **6**.

The synthesis of eleutherobin analogue 7 is depicted in Scheme 2.‡ From aldehyde 4, Wittig methylenation afforded the alkene 8 (90%), which was then treated with buffered pyridinium



† Electronic supplementary information (ESI) available: experimental procedure for the conversion of 5 into 6 and 9 and spectroscopic data for 6 and 7. See http://www.rsc.org/suppdata/cc/b4/b413426e/ *abh1@cam.ac.uk aholmes@unimelb.edu.au



Scheme 1‡

hydrofluoride to effect silyl group removal (89%). Our initial attempts used TBAF as the reagent for silyl deprotection, but this was found to be non-chemoselective and gave a diol. IBX oxidation then yielded an aldehyde (94%), which was subsequently methylenated to afford the triene **5** (93%). RCM of **5** using the first generation Grubbs catalyst⁸ **10** yielded two compounds, the required fused bicyclic medium-ring ether **6** (69%) and the isomeric bis-cyclopentene **9** (22%).

To investigate whether the bis-cyclopentene was formed directly from the triene **5** or from the bicyclic compound **6**, the latter was resubjected to the RCM conditions. After 3 days, ¹H NMR analysis of the reaction mixture showed a 20% conversion to **9**. This result suggests that equilibration of **6** to the thermodynamically more stable **9** was possible, but the slow conversion suggested that **9** was formed directly from the triene **5** and not through **6** as an intermediate.

An alternative route to the synthesis of 6 was explored in order to avoid the formation of the bis-cyclopentene 9. The strategy was to protect the interfering endocyclic ring double bond with an



Scheme 2 Synthesis of 6. *Reagents and conditions*: (i) CH₃PPh₃Br, *n*BuLi, THF, -78 °C, 20 min, 90%; (ii) HF·py, THF, pyridine, 16 h, 89%; (iii) IBX, DMSO, rt, 16 h, 94%; (iv) CH₃PPh₃Br, *n*BuLi, THF, -78 °C, 1 h, 93%; (v) [(PCy₃)₂Cl₂Ru(=CHPh)] **10**, CH₂Cl₂, 18 h, 69% **6**, 22% **9**; (vi) [(PCy₃)₂Cl₂Ru(=CHPh)] **10**, CH₂Cl₂, 72 h, 20% conv. IBX = *o*-iodoxybenzoic acid.[‡]

epoxide group, then perform the RCM reaction and then deoxygenate the epoxide back to the alkene 6 (Scheme 3⁺₄).



Scheme 3 Epoxidation strategy and completion of 7. *Reagents and conditions*: (i) *m*CPBA, THF, 0 °C \rightarrow rt, 80% (major 63%); (ii) IBX, DMSO, rt, 20 h, 73%; (iii) CH₃PPh₃Br, *n*BuLi, THF, -78 °C, 20 min, 98%; (iv) [(PCy₃)₂Cl₂Ru(=CHPh)] **10**, CH₂Cl₂, 18 h, rt, 86%; (v) WCl₆, *n*BuLi (2 equiv.), -78 °C \rightarrow rt, 53%; (vi) TBAF, THF, 36 h, 0 °C \rightarrow rt, 61%; (x) **16**, DMAP, NEt₃, THF, 20 h, 50%. IBX = *o*-iodoxybenzoic acid.‡

Starting from the alcohol **11**, epoxidation furnished a 3 : 1 mixture of epoxides (80%), which were separated by flash chromatography. Only the major isomer was carried forward in the synthesis. IBX oxidation (73%) gave aldehyde **12**, which was followed by a Wittig methylenation (98%) to give the epoxide protected RCM precursor **13**. RCM of **13** with the first generation Grubbs catalyst⁸ **10** efficiently gave the desired ring-closed product **14** (86%). Epoxide removal was accomplished by treatment with WCl₆–*n*BuLi¹⁰ to give the previously isolated bicyclic compound **6**, albeit in only a moderate yield (53%). Although the epoxide protection route was less efficient than our original route, the sequence served to confirm independently our assignments of the major and minor products of the RCM of **5**.

Completion of the synthesis of the eleutherobin analogue **7** was accomplished by silyl deprotection (61%) to give the crystalline alcohol **15** {61%, mp 85–86 °C (from ether)}, followed by coupling to the mixed anhydride **16**³ (50%). The X-ray crystal structure (Fig. 2) of the bicycle **15**^{11,12} provides insight into the conformation.

The eleutherobin analogue 7 and the advanced intermediate 15 were investigated for tubulin stabilising properties.⁹ The results show that the analogue 7 exhibits microtubule stabilising properties, but with less potency ($ED_{90} = 6 \pm 1.2 \mu M$) than paclitaxel ($ED_{90} < 0.5 \pm 1.2 \mu M$). Nevertheless, this is an important result since the cyclohexene ring of eleutherobin is believed to be an important determinant for its antimitotic activity⁷ and that fragment is not present in the analogue 7. In addition, even the alcohol 15 was more potent ($ED_{90} = 3 \pm 1.2 \mu M$) than 7. This result could imply that it is possible to design further eleutherobin analogues without the urocanic ester linkage and still maintain microtubule stabilising properties.¹³

In summary, the synthesis of a simplified eleutherobin analogue 7 has been accomplished. Biological data indicated that 7 exhibits microtubule stabilising properties in the micromolar range. In addition, compound **15**, which lacks the crucial urocanic ester side chain shows greater potency than 7. Efforts to synthesise more potent analogues are currently under way.

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Fig. 2 X-Ray crystal structure of 15.

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Notes and references

‡ All new compounds exhibited satisfactory spectroscopic and analytical and/or exact mass data.

§ CCDC 249051. See http://www.rsc.org/suppdata/cc/b4/b413426e/ for crystallographic data in .cif or other electronic format.

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- 11 Crystal data for **15**: C₁₀H₁₄O₂, M = 166.21, tetragonal, space group $P4_{3}2_{1}2$, a = b = 8.2007(4), c = 26.5896(9) Å, $\alpha = \beta = \gamma = 90$ °, U = 1788.19(14) Å³, Z = 8, μ (Mo–K α) = 0.084 mm⁻¹, 4922 reflections measured at 180(2) K using an Oxford Cryosystems Cryostream cooling apparatus, 1997 unique ($R_{int} = 0.066$); $R_1 = 0.0532$, $wR_2 = 0.118$ [$I > 2\sigma(I)$]; goodness-of-fit on F^3 , S = 1.187. The structure was solved with SHELXL-97¹².§
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