Organic Chemistry

Synthesis of alkynylbenzo-15-crown-5 ethers*

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Reactions of 4'-iodobenzo-15-crown-5 ether with ethynylarenes or 4'-ethynylbenzo-15crown-5 ether with haloarenes in the presence of catalytic amounts of Pd^{II} complex salts, CuI, and Et₃N gave 4'-(arylethynyl)benzo-15-crown-5 ethers in 55–80% yields.

Key words: benzocrown ethers, alkynes, cross-coupling, homogeneous catalysis.

Crown ethers have found wide use in organic, analytical, and coordination chemistry, as well as in biology.² In recent years, crown ethers containing alkynyl substituents have attracted the attention of researchers. In particular, a series of studies³⁻⁵ has been devoted to the investigation of structure—property relationships (thermal and spectral properties and selectivity of cation binding) for liquid crystals based on ethynylbenzocrown ethers in solution and in the nematic phase. Compounds of this type were also found⁶ to be convenient precursors for construction of assemblies containing two benzocrown ether fragments.

Currently, the sole method for the synthesis of ethynylbenzocrown ethers is the Heck—Sonogashira cross-coupling of aryl(hetaryl) halides with terminal alkynes,^{7,8} which is widely applied even to low-reactivity aryl halides (with reduced nucleofugicity of a halogen atom) and ethynylarenes (with reduced CH acidity).⁹ However, the small number of the synthesized alkynyl-

benzocrown ethers does not cover the whole pattern of cross-coupling applicability, especially for both halogen and ethynyl derivatives of crown ethers of low reactivity due to the presence of deactivating alkoxy groups (+M-effect) in the aromatic part of their molecules.

In connection with this, a comparative study of "direct" and "inverse" cross-coupling reactions was performed. In the former reaction (method A), an iodobenzocrown ether was used as a halogen-containing component and an ethynylarene as an alkyne component; in the latter reaction (method B), an iodo- or bromoarene was used in cross-coupling with an ethynylbenzocrown ether (Scheme 1).

Results and Discussion

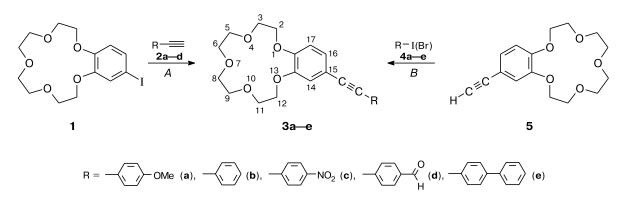
The yields of alkynes 3a-e obtained by methods A and B are close (55-80%). The reaction time varies from one to four hours; as expected, electron-withdrawing substituents both in halo- (method B) and ethynylarenes (method A) only slightly accelerate the reaction:

* For the preliminary communication, see Ref. 1.

Published in Russian in Izvestiya Akademii Nauk. Seriya Khimicheskaya, No. 9, pp. 1579-1581, September, 2002.

1066-5285/02/5109-1720 \$27.00 © 2002 Plenum Publishing Corporation

Scheme 1



Reagents and conditions: PdCl₂(PPh₃)₂ or Pd(OAc)₂, PPh₃; CuI, Et₃N, C₆H₆, 70–75 °C, an Ar atmosphere.

OMe (4 h, *A* and *B*) > H (2 h, *A* and *B*) > Ph (2.5 h, *B*) > CHO (1.5 h, *A* and *B*) > NO₂ (1 h, *A* and *B*).

Cross-coupling was carried out under standard conditions (boiling benzene, Et_2NH , $Pd(PPh_3)_2Cl_2$ (2 mol %), and CuI (4 mol %)).⁶ However, the use of Et_3N instead of Et_2NH reduced the reaction time from 24 to 1–4 h. For carbonyl-containing substrates, the reaction was catalyzed by $Pd(OAc)_2$ (4 mol %) (in the presence of other catalysts, the reaction mixture undergoes strong resinification¹⁰), PPh₃ (11 mol %), and CuI (3 mol %).

Thus, we developed two alternative methods for the synthesis of various arylalkynylbenzo-15-crown-5 ethers by the Heck—Sonogashira cross-coupling of the corresponding halides with alkynyl components.

Experimental

¹H NMR spectra were recorded on a Bruker Avance 300 spectrometer (300 MHz) in CDCl₃ with Me₄Si as a standard. IR spectra were recorded on a Bruker IFS-66 spectrometer (KBr). Mass spectra were recorded on a Finnigan SSQ-710 instrument (EI, direct inlet, ionization chamber temperature 220-270 °C, ionizing voltage 70 eV). Column chromatography was carried out on KSK silica gel ($60/200 \mu m$). The course of the reaction was monitored and the purity of the products obtained was checked by TLC on Silufol UV-254 plates. 4'-Iodobenzo-15-crown-5 ether (1),¹¹ *p*-ethynylbenzaldehyde (2d),¹⁰ 4-iodobiphenyl (4e), iodobenzene (4b), p-iodonitrobenzene (4c),¹² and *p*-nitrophenylacetylene $(2c)^{13}$ were prepared according to the known procedures. 4'-Ethynylbenzo-15-crown-5 ether $(5)^{1,14}$ was synthesized by the reverse Favorsky reaction from a corresponding tertiary alkynol. p-Methoxyphenylacetylene (2a) was prepared as described earlier,⁹ b.p. 82-87 °C (10 Torr), m.p. 27-28 °C (cf. Ref. 15: b.p. 86 °C (9 Torr), m.p. 29 °C).

Copper(1) iodide, PPh₃, Pd(OAc)₂, Pd(PPh₃)₂Cl₂, phenylacetylene (**2b**), *p*-iodoanisole (**4a**), and *p*-bromobenzaldehyde (**4d**) (Lancaster Co.) were used without additional purification; the other reagents and organic solvents were purified according to standard procedures.¹⁶ 15-[2-(4-Methoxyphenyl)ethyn-1-yl]-2,3,5,6,8,9,11,12octahydro-1,4,7,10,13-pentaoxabenzocyclopentadecine (3a). Method A. A solution of iodide 1 (0.8 g, 2 mmol), alkyne 2a (0.3 g, 2.4 mmol), Pd(PPh₃)₂Cl₂ (30 mg, 0.04 mmol), and CuI (15 mg, 0.08 mmol) in a mixture of Et₃N (2 mL) and benzene (8 mL) was stirred in an atmosphere of Ar at 70–75 °C for 4 h. Then the reaction mixture was cooled and filtered through an SiO₂ layer (2.5×2 cm). The solvents were removed *in vacuo*, and the residue was recrystallized from benzene—hexane. The yield of crown ether 3a was 0.43 g (54%).

Method B. A solution of iodide **4a** (0.51 g, 2.2 mmol), alkyne **5** (0.58 g, 2 mmol), Pd(PPh₃)₂Cl₂ (30 mg, 0.04 mmol), and CuI (15 mg, 0.08 mmol) in a mixture of Et₃N (2 mL) and benzene (8 mL) was stirred in an atmosphere of Ar at 70–75 °C for 3.5 h. Then the reaction mixture was cooled and filtered through an SiO₂ layer (2.5×2 cm). The solvents were removed *in vacuo*, and the residue was recrystallized from benzene—hexane. The yield of crown ether **3a** was 0.45 g (57%).

Crown ether **3a**, m.p. 109–110 °C. ¹H NMR, δ : 3.74–3.81 (m, 8 H, OCH₂CH₂O); 3.84 (s, 3 H, OMe); 3.93–4.17 (m, 8 H, Ar–OCH₂CH₂O); 6.84 (d, 1 H, H(17), J = 9 Hz); 6.92 (d, 2 H, H(3), H(5), J = 8.5 Hz); 7.03 (s, 1 H, H(14)); 7.11 (d, 1 H, H(16), J = 9 Hz); 7.46 (d, 2 H, H(2), H(6), J = 8.5 Hz). IR, v/cm⁻¹: 2225 (C=C). MS, m/z (I_{rel} (%)): 398.1 [M]⁺ (25.68), 350.1 (21.28), 262.1 (11.07), 204.0 (14.89), 203.0 (100.00), 159.1 (3.81), 175.0 (13.04), 161.1 (11.80), 148.6 (48.18), 114.1 (15.07). Found: M = 398.1734 [M]⁺. C₂₃H₂₆O₆. Calculated: M = 398.17292.

15-(2-Phenylethyn-1-yl)-2,3,5,6,8,9,11,12-octahydro-1,4,7,10,13-pentaoxabenzocyclopentadecine (3b). Method A. A solution of iodide 1 (0.98 g, 2.5 mol), alkyne 2b (0.3 g, 2.7 mol), Pd(PPh₃)₂Cl₂ (40 mg, 0.05 mmol), and CuI (20 mg, 0.1 mmol) in a mixture of Et₃N (2 mL) and benzene (8 mL) was stirred in an atmosphere of Ar at 70–75 °C for 2 h. Then the reaction mixture was cooled and filtered through an SiO₂ layer (2.5×2 cm). The solvents were removed *in vacuo*, and the residue was recrystallized from benzene—hexane. The yield of compound 3b was 0.66 g (72%).

Method *B*. Compound **3b** was obtained from alkyne **5** (0.58 g, 2 mmol) and iodide **4b** (0.45 g, 2.2 mmol) as described above for crown ether **3a**. The yield of compound **3b** was 0.55 g (75%); the reaction time was 2 h.

Crown ether **3b**, m.p. 112–113 °C (benzene–hexane). ¹H NMR, δ : 3.75–4.26 (m, 16 H, OCH₂CH₂O); 6.76 (d, 1 H, H(17), J = 8 Hz); 6.95 (s, 1 H, H(14)); 7.03 (d, 1 H, H(16), J = 8 Hz); 7.21–7.26 (m, 3 H, H(3), H(4), H(5)); 7.46–7.49 (m, 2 H, H(2), H(6)). IR, v/cm^{-1} : 2200 (C=C). MS, m/z (I_{rel} (%)): 368.1 [M]⁺ (51.08), 280.2 (10.86), 237.1 (18.73), 236.1 (100.00), 221.1 (26.29), 210.1 (10.63), 180.1 (47.39), 163.1 (13.79), 152.1 (30.26), 110.6 (17.05). Found: M = 368.1625 [M]⁺. C₂₂H₂₄O₅. Calculated: M = 368.16236.

15-[2-(4-Nitrophenyl)ethyn-1-yl]-2,3,5,6,8,9,11,12octahydro-1,4,7,10,13-pentaoxabenzocyclopentadecine (3c). Method A. Compound 3c was obtained from iodide 1 (0.98 g, 2.5 mmol) and alkyne 2c (0.4 g, 2.7 mmol) by analogy with 3b. The yield of crown ether 3c was 0.78 g (76%); the reaction time was 1 h.

Method *B*. Compound 3c was obtained from alkyne 5 (0.58 g, 2 mmol) and iodide 4c (0.55 g, 2.2 mmol) by analogy with 3a. The yield of crown ether 3c was 0.61 g (74%); the reaction time was 1 h.

Crown ether **3c**, m.p. 108–110 °C (benzene–hexane). ¹H NMR, δ : 3.75–4.17 (m, 16 H, OCH₂CH₂O); 6.83 (d, 1 H, H(17), J = 8 Hz); 7.05 (s, 1 H, H(14)); 7.12 (d, 1 H, H(16), J = 8 Hz); 7.06 (d, 2 H, H(2), H(6), J = 9 Hz); 8.16 (d, 2 H, H(3), H(5), J = 9 Hz). IR, v/cm⁻¹: 2210 (C=C). MS, m/z (I_{rel} (%)): 412.9 [M]⁺ (44.22), 325.0 (8.68), 292.0 (10.96), 282.0 (18.41), 280.9 (100.00), 265.9 (15.97), 235.0 (11.14), 224.9 (22.31), 163.0 (12.53), 151.1 (9.74). Found: M = 413.1476 [M]⁺. C₂₂H₂₃O₇N. Calculated: M = 413.14744.

4-[2-(2,3,5,6,8,9,11,12-Octahydro-1,4,7,10,13-pentaoxabenzocyclopentadecin-15-yl)ethyn-1-yl]benzaldehyde (3d). Method A. A solution of iodide 1 (1.2 g, 3 mmol), alkyne 2d (0.43 g, 3.3 mmol), $Pd(OAc)_2$ (30 mg, 0.13 mmol), PPh_3 (90 mg, 0.34 mmol), and CuI (15 mg, 0.08 mmol) in a mixture of Et₃N (2 mL) and benzene (8 mL) was stirred in an atmosphere of Ar at 70-75 °C for 1.5 h. Then the reaction mixture was cooled and filtered through an SiO₂ layer (2.5×2 cm). The solvents were removed *in vacuo*, and the residue was recrystallized from benzene—hexane. The yield of compound 3d was 0.97 g (82%).

Method B. A solution of bromide **4d** (0.57 g, 3.1 mmol), alkyne **5** (0.87 g, 3 mmol), $Pd(OAc)_{2,}$ (30 mg, 0.13 mmol), PPh₃ (90 mg, 0.34 mmol), and CuI (15 mg, 0.08 mmol) in a mixture of Et₃N (2 mL) and benzene (8 mL) was stirred in an atmosphere of Ar at 70–75 °C for 1.5 h. Then the reaction mixture was cooled and filtered through an SiO₂ layer (2.5×2 cm). The solvents were removed *in vacuo*, and the residue was recrystallized from benzene—hexane. The yield of compound **3d** was 0.93 g (78%).

Crown ether **3d**, m.p. 140–141 °C (benzene–hexane). Found (%): C, 69.72; H, 6.19. $C_{23}H_{24}O_6$. Calculated (%): C, 69.68; H, 6.10. ¹H NMR, 8: 3.38–4.01 (m, 16 H, OCH₂CH₂O); 6.41 (d, 1 H, H(17), J = 9 Hz); 7.02 (s, 1 H, H(14)); 7.19 (d, 1 H, H(16), J = 9 Hz); 7.39 (d, 2 H, H(2), H(6), J = 8 Hz); 7.44 (d, 2 H, H(3), H(5), J = 8 Hz); 9.56 (s, 1 H, CHO). IR, v/cm⁻¹: 2207 (C=C), 1701 (C=O).

15-[2-(Biphenyl-4-yl)ethyn-1-yl]-2,3,5,6,8,9,11,12-octahydro-1,4,7,10,13-pentaoxabenzocyclopentadecine (3e) was obtained as described for crown ether 3a (method *B*). Condensation of alkyne 5 (0.75 g, 2.4 mmol) with iodide 4e (0.84 g, 2.6 mmol) was carried out for 2.5 h. The yield of compound 3e was 0.78 g (73%), m.p. 167–168 °C (benzene). ¹H NMR, δ : 3.51–4.09 (m, 16 H, OCH₂CH₂); 6.92 (d, 1 H, H(17),
$$\begin{split} J &= 9 \text{ Hz}; \ 7.06-7.1 \ (m, \ 2 \ H, \ H(14), \ H(16)); \ 7.29 \ (t, \ 1 \ H, \\ H(4'), \ J &= 9 \ Hz); \ 7.41 \ (t, \ 2 \ H, \ H(3'), \ H(5'), \ J &= 9 \ Hz); \ 7.55 \\ (d, \ 2 \ H, \ H(3), \ H(5), \ J &= 9 \ Hz); \ 7.62-7.71 \ (m, \ 4 \ H, \ H(2), \ H(6)), \\ H(2'), \ H(6')). \ IR, \ \nu/cm^{-1}: \ 2211 \ (C=C). \ MS, \ m/z \ (I_{rel} \ (\%)): \\ 444.2 \ [M]^+ \ (69.18), \ 356.1 \ (15.23), \ 313.1 \ (24.64), \ 312.1 \ (100.00), \\ 256.1 \ (29.55), \ 156.1 \ (33.09), \ 148.6 \ (48.18), \ 114.1 \ (15.07), \ 101.1 \\ (9.60), \ 78.1 \ (24.23). \ Found: \ M &= \ 444.1923 \ [M]^+. \ C_{28}H_{28}O_5. \\ Calculated: \ M &= \ 444.19366. \end{split}$$

The ¹H NMR, IR, and mass spectra were recorded at the ATsKP of the Siberian Branch of the Russian Academy of Sciences with financial support from the Russian Foundation for Basic Research (Project No. 33-03-40135).

This work was financially supported by the Russian Foundation for Basic Research (Project No. 99-03-32877) and the CRDF (Award No. Rec-008).

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Received June 4, 2001; in revised form May 7, 2002