- [20] a) P. K. Freeman, L. L. Hutchinson, J. Org. Chem. 1980, 45, 1924– 1930; b) R. E. Ireland, M. G. Smith, J. Am. Chem. Soc. 1988, 110, 854– 860.
- [21] B. E. Rossiter, T. R. Verhoeven, K. B. Sharpless, *Tetrahedron Lett.* 1979, 20, 4733–4736.
- [22] D. H. Barton, W. B. Motherwell, A. Stange, *Synthesis* 1981, 743–745.
  [23] The stereochemistry of the newly formed C12 stereocenter of
- tetrahydrofuran 20b was assigned by using 2D NMR spectroscopy.
  [24] J. W. Gillard, R. Fortin, H. E. Morton, C. Yoakim, C. A. Quesnelle, S. Daignault, Y. Guidan, J. Org. Chem. 1988, 53, 2602–2608. This unusual protecting group offered the necessary acid stability during the synthesis, as well as the required fluoride ion lability for the final global deprotection with tris(dimethyamino)sulfur(trimethylsilyl) difluoride (See: K. A. Scheidt, H. Chen, B. C. Follows, S. R. Chemler, D. S. Coffey, W. R. Roush, J. Org. Chem. 1998, 63, 6436–6437). Under these conditions, the time required for C11-OTBS deprotection was prohibitively long. The use of a more base-labile TES ether did not afford the required acid stability during the synthesis.
- [25] S. V. Ley, L. R. Cox, J. Chem. Soc. Perkin Trans. 1 1997, 3315-3324.
- [26] L. D.-L Lu, R. A. Johnson, M. G. Finn, K. B. Sharpless, J. Org. Chem. 1984, 49, 728-731. Sharpless has noted (See: Y. Gao, R. M. Hanson, J. M. Klunder, S. Y. Ko, H. Masamune, K. B. Sharpless, J. Am. Chem. Soc. 1987, 109, 5765-5780, and references therein) that 1,1-disubstitued olefins are problematic substrates for epoxidation under stoichiometric [Ti(OiPr)<sub>4</sub>] conditions due to epoxide opening by isopropyl alcohol. The use of [Ti(OtBu)<sub>4</sub>] conveniently avoided any such side reactions.

## Asymmetric Syntheses of Pectenotoxins-4 and -8, Part II: Synthesis of the C20–C30 and C31– C40 Subunits and Fragment Assembly\*\*

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In the preceding communication,<sup>[1]</sup> the proposed synthesis plan identified the two principal pectenotoxin-4 subunits **II** and **III** (Figure 1). It was our intention to couple these fragments through the alkylation of the metalloenamine derived from hydrazone **III**, readily available from the coupling of advanced intermediates **IV** and **V** (transform  $T_2$ ), by epoxide **II**. However, this investigation revealed that the above bond construction was not feasible due to the decomposition of metalloeneamine **III** under the reaction conditions.<sup>[2]</sup> Accordingly, the objective in the present communication is the synthesis of the subunits **IV** and **V**, and the completion of the syntheses of pectenotoxin-4 (1) and pectenotoxin-8 by a revised fragment coupling strategy, where epoxide alkylation (transform  $T_1$ ) precedes diene formation (transform  $T_2$ ).

The plan for the construction of the F-ring tetrahydrofuran **IV** was to involve a C37 hydroxy-directed epoxidation of olefin **VI** with a subsequent ring closure by the C32 hydroxy moiety (transform  $T_3$ ). Finally, the stereoselective formation of the E-ring tetrahydrofuran **V** from its acyclic precursor **VII** was based on an iodoetherification precedent provided by Bartlett and Rychnovsky (transform  $T_4$ ).<sup>[3]</sup>

The synthesis of the ring-E synthon V began with the known aldol adduct adduct 2 (Scheme 1).<sup>[4]</sup> Reduction of 2 (LiBH<sub>4</sub>, THF, 0°C), and selective protection of the primary alcohol (TBSCl, Im, CH<sub>2</sub>Cl<sub>2</sub>, 100% over two steps) afforded allylic alcohol 3.<sup>[5]</sup> Acylation of 3 with the PMB-protected lactic acid 4<sup>[6]</sup> (DCC, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, 52%), followed by carbonyl olefination of **5a** with Tebbe reagent<sup>[7]</sup> afforded the 1,5-diene 5b. Claisen rearrangement of 5b in refluxing toluene gave the desired rearrangement product 6 in 82% yield for the two steps. Chelate-controlled reduction of the resulting ketone (Zn(BH<sub>4</sub>)<sub>2</sub>, Et<sub>2</sub>O, -78°C, 86%, d.r. 86:14) provided the precursor for the key iodoetherification reaction. In spite of the modest selectivity that was observed for the formation of the desired tetrahydrofuran 7 (NIS, CH<sub>3</sub>CN, -40°C, 89%, d.r. 72:28), this outcome proved sufficient to pursue the planned route.

Successive radical dehalogenation of **7** (Bu<sub>3</sub>SnH, AIBN, toluene, 100%) and deprotection of the primary TBS ether (TBAF, THF, 95%) afforded alcohol **8**. Oxidation with Dess-Martin reagent<sup>[8]</sup> (py, CH<sub>2</sub>Cl<sub>2</sub>, 99%), Wittig homologation (EtOC(O)CC(CH<sub>3</sub>)PPh<sub>3</sub>, THF, 65°C; 100% *E:Z* > 95:5), and ester reduction (LiAlH<sub>4</sub>, Et<sub>2</sub>O, 0°C, 92%) completed the carbon assembly of the E-ring fragment. Benzyl protection (NaH, BnBr, TBAI, THF/DMF, 94%) followed by PMB deprotection (DDQ, CH<sub>2</sub>Cl<sub>2</sub>/pH 7 buffer, 95%) gave alcohol **10**. Oxidation to the methyl ketone<sup>[8]</sup> (Dess-Martin periodinane, py, CH<sub>2</sub>Cl<sub>2</sub>, 93%), and hydrazone formation (TMSCl, CH<sub>2</sub>Cl<sub>2</sub>/Me<sub>2</sub>NNH<sub>2</sub>, 100%) completed the synthesis of hydrazone **11**.

As summarized in Figure 1, the first stage of the synthesis of the ring-F fragment **IV** will be simplified to the construction of the C31–C35 phosphonium salt, the C36–C40 aldehyde, and their union through a Wittig coupling to afford the *Z*-olefin **VI**.

The synthesis of the C31–C35 phosphonium salt began with the known triol derivative **12** (Scheme 2).<sup>[9]</sup> Protection of the hydroxy group at C33 of **12** as a PMB ether (PMBBr, NaH, THF/DMF, 95%) followed by reductive ozonolysis (O<sub>3</sub>, EtOH, then DMS, then NaBH<sub>4</sub>, 95%) afforded alcohol **13**. Transformation of **13** to the corresponding iodide (I<sub>2</sub>, Im, Ph<sub>3</sub>P, CH<sub>2</sub>Cl<sub>2</sub>, 0°C, 89%) proceeded smoothly, but careful control of the temperature was required to access phosphonium salt **14** (Ph<sub>3</sub>P, CH<sub>3</sub>CN, 55°C, 89%).<sup>[10]</sup>

The synthesis of the aldehyde partner **17** began with protection of the hydroxy group at C37 of aldol adduct **15**<sup>[11]</sup> as a base-sensitive triphenylsilyl ether (TPSCl, Im, DMAP, DMF, 0°C, 98%; Scheme 2). Half reduction of the *S*-phenyl thioester<sup>[12]</sup> (Pd/C, Et<sub>3</sub>SiH, acetone, 95%), and olefination under modified Lombardo conditions<sup>[13]</sup> ([Cp<sub>2</sub>ZrCl<sub>2</sub>], Zn dust, CH<sub>2</sub>I<sub>2</sub>, THF, 0°C, 84%) afforded olefin **16**. Rhodium-

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## COMMUNICATIONS



Figure 1. The major disconnections.



Scheme 1. Synthesis of C20–C30 E-ring fragment. a) LiBH<sub>4</sub>, THF, 0°C; b) TBSCl, Im, CH<sub>2</sub>Cl<sub>2</sub>; 100% (two steps); c) DCC, DMAP, CH<sub>2</sub>Cl<sub>2</sub>; 52%; d) [Cp<sub>2</sub>TiCl<sub>2</sub>], AlMe<sub>3</sub>, THF, then **5a**; e) toluene, 110°C; 82% (two steps); f) Zn(BH<sub>4</sub>)<sub>2</sub>, Et<sub>2</sub>O, -78°C; 74%; g) NIS, CH<sub>3</sub>CN, -40°C; 64%; h) Bu<sub>3</sub>SnH, AIBN, toluene; 100%; i) TBAF, THF; 95%; j) Dess–Martin periodinane, py, CH<sub>2</sub>Cl<sub>2</sub>; 99%; k) EtOC(O)CC(CH<sub>3</sub>)PPh<sub>3</sub>, THF, 65°C; 100%; l) LiAlH<sub>4</sub>, Et<sub>2</sub>O, 0°C; 92%; m) NaH, BnBr, TBAI, THF/DMF (3:1); 94%; n) DDQ, CH<sub>2</sub>Cl<sub>2</sub>/PH 7 buffer (8:1); 95%; o) Dess–Martin periodinane, py, CH<sub>2</sub>Cl<sub>2</sub>; 93%; p) TMSCl, CH<sub>2</sub>Cl<sub>2</sub>/Me<sub>2</sub>NNH<sub>2</sub> (1:1), 0°C; 100%. See reference [5] for abbreviations.

catalyzed hydroboration<sup>[14]</sup> ([(Ph<sub>3</sub>P)<sub>3</sub>RhCl], catecholborane, THF, 0°C, then H<sub>2</sub>O<sub>2</sub>, EtOH, pH 7 buffer, 88%), TIPS protection of the resulting primary alcohol (TIPSOTf, lut., CH<sub>2</sub>Cl<sub>2</sub>, -78°C, 100%), and half reduction of the ethyl ester (DIBAL-H, toluene, -78°C, 89%) completed the synthesis of aldehyde **17**.

Wittig coupling of phosphonium salt **14** and aldehyde **17** proceeded smoothly (LiHMDS, THF, -78 °C, 79 %, Z:E > 95:5) to afford **18**. Selective deprotection of the TPS ether under basic conditions (K<sub>2</sub>CO<sub>3</sub>, MeOH/CH<sub>2</sub>Cl<sub>2</sub>, 87 %), directed epoxidation with *m*-CPBA<sup>[15]</sup> (CH<sub>2</sub>Cl<sub>2</sub>, 0 °C, 95 %,

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d.r. > 95:5) and reprotection of the hydroxy group at C37 as a TBS ether (TBSOTf, lut.,  $CH_2Cl_2$ , -78 °C, 95%) afforded epoxide **19** as a single diastereomer. Selective deprotection of the benzyl ether group at C32 was possible under reducing conditions.<sup>[16]</sup> When the resulting epoxy alcohol was exposed to PPTS in MeOH/CH<sub>2</sub>Cl<sub>2</sub>, cyclization to form the F-ring tetrahydrofuran with simultaneous deprotection of the benzthiazole sulfide at C31 under Mitsunobo conditions (BtSH, DIAD, Ph<sub>3</sub>P, THF, 0°C), oxidation to the sulfone<sup>[18]</sup>



Scheme 2. C31–C40 FG fragment synthesis. a) PMBBr, NaH, THF/DMF (3:1), 0°C; 95%; b)  $O_3$ , EtOH, -78°C, then DMS, then NaBH<sub>4</sub>; 95%; c) I<sub>2</sub>, Im, Ph<sub>3</sub>P, CH<sub>2</sub>Cl<sub>2</sub>, 0°C; 89%; d) Ph<sub>3</sub>P, CH<sub>3</sub>CN, 55°C, 89%; e) TPSCl, Im, DMAP, DMF, 0°C; 98%; f) Pd/C, Et<sub>3</sub>SiH, acetone; 95%; g) [Cp<sub>2</sub>ZrCl<sub>2</sub>], Zn dust, CH<sub>2</sub>I<sub>2</sub>, THF, 0°C; 84%; h) [(Ph<sub>3</sub>P)<sub>3</sub>RhCl], catecholborane, THF, then pH 7 buffer, H<sub>2</sub>O<sub>2</sub>; 88%; i) TIPSOTf, lut., CH<sub>2</sub>Cl<sub>2</sub>, -78°C; 100%; j) DIBAL-H, toluene, -78°C; 89%; k) LiHMDS, THF, then aldehyde **17**, -78°C–0°C; 79%; l) K<sub>2</sub>CO<sub>3</sub>, MeOH/CH<sub>2</sub>Cl<sub>2</sub> (2:1); 87%; m) *m*-CPBA, CH<sub>2</sub>Cl<sub>2</sub>, 0°C; 95%; n) TBSOTf, lut., CH<sub>2</sub>Cl<sub>2</sub>, -78°C; 0) LiDBA, THF, -78°C; 79%; p) PPTS, MeOH/CH<sub>2</sub>Cl<sub>2</sub> (1:1); 73%; q) BtSH, Ph<sub>3</sub>P, DIAD, THF, 0°C; r) *m*-CPBA, CH<sub>2</sub>Cl<sub>2</sub>, 0°C; 70%; t) DDQ, CH<sub>2</sub>Cl<sub>2</sub>/pH 7 buffer (10:1); 97%; u) TMSCl, Im, CH<sub>2</sub>Cl<sub>2</sub>, 99%. See reference [5] for abbreviations.

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## COMMUNICATIONS

(*m*-CPBA, CH<sub>2</sub>Cl<sub>2</sub>, 0°C, 70% over two steps), and protection of the hydroxy group at C36 as a TES ether (TESCl, Im, CH<sub>2</sub>Cl<sub>2</sub>, 90%) afforded **21**. In preparation for the diene formation step, the C33-OPMB ether of sulfone **21** was exchanged for the corresponding TMS ether (DDQ, CH<sub>2</sub>Cl<sub>2</sub>/ pH 7 buffer, 97%; then TMSCl, Im, CH<sub>2</sub>Cl<sub>2</sub>, 0°C, 99%) to afford **22**.<sup>[19]</sup>

Completion of the synthesis is shown in Scheme 3. The C19–C20 bond construction joining advanced intermediates

11 and 23 was accomplished by the addition of the magnesium bromide activated epoxide complex 23-MgBr<sub>2</sub><sup>[20]</sup> to the metalloenamine derived from 11. The resulting unstable hydrazinyl lactol 24 was treated with acid under biphasic conditions (pentane/CH<sub>2</sub>Cl<sub>2</sub>/10% aq. NaHSO<sub>4</sub>) to access the corresponding lactol, which underwent desilylation at C16 and spontaneous bicyclic ketal formation to afford adduct 25 upon exposure to PPTS in MeOH/CH<sub>2</sub>Cl<sub>2</sub> (52% over three steps). Protection of the hydroxymethyl moiety at C18



Scheme 3. Fragment assembly and completion of synthesis. a) **23**, EtMgBr, HMPA, THF,  $-67 \,^{\circ}C \rightarrow 0^{\circ}C$ ; b) **11**, LDA, THF,  $-67 \,^{\circ}C \rightarrow 0^{\circ}C$ ; c) add **23**-MgBr<sub>2</sub> to **11**-Li, +10 \,^{\circ}C; d) pentane/CH<sub>2</sub>Cl<sub>2</sub>/10 % NaHSO<sub>4</sub> (4.5:1.5:4); e) PPTS, CH<sub>2</sub>Cl<sub>2</sub>/MeOH (1:1); 52 % (three steps); f) TBDPSCl, Im, DMAP, CH<sub>2</sub>Cl<sub>2</sub>; 90 %; g) DDQ, CH<sub>2</sub>Cl<sub>2</sub>/PH 7 buffer (6:1); 90 %; h) TESCl, Im, CH<sub>2</sub>Cl<sub>2</sub>; 100 %; i) LiDBB, THF,  $-78 \,^{\circ}C$ ; 85 %; j) SO<sub>3</sub>·Py, Et<sub>3</sub>N, CH<sub>2</sub>Cl<sub>2</sub>/DMSO (1:1); k) sulfone **22**, LiHMDS, THF,  $-78 \,^{\circ}C \rightarrow 0^{\circ}C$ ; 72 % (two steps); l) Boc<sub>2</sub>O, DMAP, CH<sub>3</sub>CN; 91 %; m) LiOH, H<sub>2</sub>O<sub>2</sub>, THF/H<sub>2</sub>O (3:1); n) trichorobenzoyl chloride, *i*Pr<sub>2</sub>NEt, then DMAP, toluene; RT; o) PPTS, CH<sub>2</sub>Cl<sub>2</sub>/MeOH (1:1); 35 % (three steps); p) Dess–Martin periodinane, py, CH<sub>2</sub>Cl<sub>2</sub>; 72 %; q) TAS-F, H<sub>2</sub>O, DMF; 85 %. See reference [5] for abbreviations.

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## COMMUNICATIONS

(TBDPSCl, Im, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, 90%), exchange of the C14 hydroxy protecting group (DDQ, CH<sub>2</sub>Cl<sub>2</sub>/pH 7 buffer, 90%; TESCl, Im, CH<sub>2</sub>Cl<sub>2</sub>, 100%), and benzyl deprotection<sup>[16]</sup> (LiDBB, THF, -78°C) afforded **27**, the fully elaborated precursor needed for the diene formation step.

On the basis of convergency considerations the C1-C30 ABCDE fragment 27 was employed as the electrophilic partner in the Julia olefination with the  $\beta$ -alkoxy sulfone 22.<sup>[21]</sup> Oxidation<sup>[22]</sup> of alcohol **27** to aldehyde **28** (SO<sub>3</sub>·Py, Et<sub>3</sub>N, DMSO/CH<sub>2</sub>Cl<sub>2</sub>) was followed by the addition of LiHMDS to a pre-mixed solution of 28 and sulfone 22 in THF at -78 °C to provide the desired diene 29 as a 88:12 mixture of C32 epimers (72% over two steps, E:Z > 95:5).<sup>[23]</sup> Pursuant to revealing the terminal carboxyl residue, N-phenylamide 29 was activated through its N-Boc imide, and hydrolysis to the acid (LiOH, H<sub>2</sub>O<sub>2</sub>, THF/H<sub>2</sub>O) proceeded with with concomitant C33-OTMS deprotection to give pectenotoxin-4 seco acid.<sup>[24]</sup> Macrocyclization under Yamagichi conditions<sup>[25]</sup> (2,4,6-trichlorobenzoyl chloride, *i*Pr<sub>2</sub>NEt, toluene, then DMAP, toluene) at room temperature provided adduct 30. Selective deprotection of the C14 and C36 OTES ethers with PPTS in MeOH/CH<sub>2</sub>Cl<sub>2</sub> (35% over three steps), oxidation to the diketone<sup>[8]</sup> (Dess-Martin periodinane, py, CH<sub>2</sub>Cl<sub>2</sub>, 72%) and global deprotection<sup>[26]</sup> (TAS-F, H<sub>2</sub>O, DMF, 85%) afforded pectenotoxin-4 in 36 steps (longest linear sequence) and 0.3% overall yield. The synthetic material was identical by <sup>1</sup>H NMR spectroscopy and optical rotation to natural pectenotoxin-4.[27] Further proof of structure was obtained by isomerizing synthetic pectenotoxin-4 to pectenotoxin-8[28] (1% TFA in CH<sub>3</sub>CN/H<sub>2</sub>O, 40%), and this material was identical to natural pectenotoxin-8 as judged by <sup>1</sup>H NMR, HPLC, TLC R<sub>f</sub>, and UV spectroscopy.

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- D. A. Evans, H. A. Rajapakse, D. Stenkamp, Angew. Chem. 2002, 114, 4751; Angew. Chem. Int. Ed. 2002, 41, 4569.
- [2] A full account of this work will be published in due course.
- [3] P. A. Bartlett, S. D. Rychnovsky, J. Am. Chem. Soc. 1981, 103, 3964– 3966.
- [4] D. A. Evans, D. M. Fitch, J. Org. Chem. 1997, 62, 454-455.
- [5] Abbreviations: TBS = tert-butyldimethylsilyl; DCC = dicyclohexyl carbodiimide; DMAP = 4-(N,N-dimethylamino)pyridine; Cp = cyclopentadienyl, d.r. = diastereomeric ratio; NIS = N-iodosuccinimide;AIBN = 2,2'-azobisisobutyronitrile; TBAF = tetra-(n-butyl)ammonium fluoride; THF = tetrahydrofuran; py = pyridine; Bn = benzyl; TBAI = tetra-(n-butyl)ammonium iodide; DMF = dimethylformamide: DDO = 2.3-dichloro-5.6-dicvano-1.4-benzoquinone: TMS = trimethylsilyl; DMS = dimethyl sulfide; Im = imidazole; TPS = triphenylsilyl; TIPS = triisopropylsilyl; lut. = 2,6-lutidine; DIBAL-H = diisobutylaluminum hydride; LiHMDS = lithium hexamethyldisilazide; m-CPBA = meta-chloroperoxybenzoic acid; Bt = benzthiazole; TES = triethylsilyl:PPTS = pyridinium para-toluenesulfonate; TBDPS = tert-butyldiphenylsilyl; LiDBB = lithium di-tert-butylbiphenylide:  $Boc_2O = di$ -tert-butyl dicarbonate: LDA = lithium diisopropylamide; HMPA = hexamethylphosphoramide; TBODPS = tert-butoxydiphenylsilyl; RT = room temperature; TAS-F = tris(dimethyamino)sulfur(trimethylsilyl)difluoride.
- [6] A. Chen, A. Nelson, N. Tanikkul, E. J. Thomas, *Tetrahedron Lett.* 2001, 42, 1251–1254.
- [7] S. H. Pine, R. Zahler, D. A. Evans, R. H. Grubbs, J. Am. Chem. Soc. 1980, 102, 3270.
- [8] D. B. Dess, J. C. Martin, J. Am. Chem. Soc. 1991, 113, 7277-7287.
- [9] M. T. Reetz, K. Kesseler, J. Org. Chem. 1985, 50, 5434-5436.

- [10] At temperatures above 55°C, intramolecular alkylation of the C<sub>32</sub> benzyl ether to form the corresponding tetrahydrofuran was observed. For a similar side reaction, see: D. A. Evans, A. M. Ratz, B. E. Huff, G. S. Sheppard, *J. Am. Chem. Soc.* **1995**, *117*, 3448–3467.
- [11] D. A. Evans, D. W. C. MacMillan, K. R. Campos, J. Am. Chem. Soc. 1997, 119, 10859-10860.
- [12] T. Fukuyama, S.-L. Lin, L. Li, J. Am. Chem. Soc. 1990, 112, 7050– 7051; see also: D. A. Evans, B. W. Trotter, P. J. Coleman, B. Cote, L. Carlos Dias, H. A. Rajapakse, A. N. Tyler, *Tetrahedron* 1999, 55, 8671–8726.
- [13] Conditions adapted from: M. Hartman, E. Zibral, *Tetrahedron Lett.* 1990, 31, 2875-2878. Both Wittig and Petersen conditions resulted in aldehyde decomposition with no observed product.
- [14] D. A. Evans, G. C. Fu, A. H. Hoveyda, J. Am. Chem. Soc. 1992, 114, 6671-6674.
- [15] B. E. Rossiter, T. R. Verhoeven, K. B. Sharpless, *Tetrahedron Lett.* 1979, 20, 4733–4736.
- [16] a) P. K. Freeman, L. L. Hutchinson, J. Org. Chem. 1980, 45, 1924–1930; b) R. E. Ireland, M. G. Smith, J. Am. Chem. Soc. 1988, 110, 854–860.
- [17] No undesired tetrahydropyran byproducts were observed in this reaction.
- [18] J. B. Baudin, G. Hareau, S. A. Julia, O. Ruel, Bull. Soc. Chim. Fr. 1993, 130, 856–878.
- [19] PMB deprotection with DDQ was incompatible with the pectenotoxin diene, as exclusive allylic oxidation was observed. For *successful* deprotections of PMB ethers in the presence of allylic dienes see: a) N. Murakami, W. Wang, M. Aoki, Y. Tsutsui, M. Sugimoto, M. Kobayashi, *Tetrahedron Lett.* **1998**, *39*, 2349–2352; b) G. Pattenden, A. T. Plowright, J. T. Tornos, T. Ye, *Tetrahedron Lett.* **1998**, *39*, 6099– 6102.
- [20] a) J. A. Marshall, R. C. Andrews, J. Org. Chem. 1985, 50, 1602–1606;
  b) D. A. Evans, R. P. Polniaszek, K. M. DeVries, D. E. Guinn, D. J. Mathre, J. Am. Chem. Soc. 1991, 113, 7613–7630.
- [21] For examples of β-alkoxy sulfone couplings see: a) ref. [19b]; b) D. A. Evans, D. M. Fitch, T. E. Smith, V. J. Cee, J. Am. Chem. Soc. 2000, 122, 10033–10046; c) A. B. Smith, B. M. Brandt, Org. Lett. 2001, 3, 1685–1688.
- [22] J. R. Parikh, W. E. von Doering, J. Am. Chem. Soc. 1967, 89, 5505 5507.
- [23] Initial feasibility studies on a model system revealed a dramatic counterion effect for this Julia coupling reaction. The use of KHMDS as the base resulted in a 29:71 ratio of C32 epimers favoring the undesired product. We attribute this epimerization side reaction to be due to ring-F cleavage through  $\beta$ -alkoxy elimination and readdition to the intermediate unsaturated sulfone.
- [24] D. A. Evans, P. H. Carter, C. J. Dinsmore, J. C. Barrow, J. L. Katz, D. W. Kung, *Tetrahedron Lett.* **1997**, *38*, 4535–4538.
- [25] J. Inanaga, K. Hirata, H. Saeki, T. Katsuki, M. Yamaguchi, Bull. Chem. Soc. Jpn. 1979, 52, 1989–1993.
- [26] K. A. Scheidt, H. Chen, B. C. Follows, S. R. Chemler, D. S. Coffey, W. R. Roush, J. Org. Chem. 1998, 63, 6436-6437.a) For previous uses of TAS-F for silyl ether deprotection see: R. A. Holton, C. Somoza, H.-B. Kim, F. Liang, R. J. Biediger, P. D. Boatman, M. Shindo, C. C. Smith, S. Kim, H. Nadizadeh, Y. Suzuki, C. Tao, P. Vu, S. Tang, P. Zhang, K. K. Murthi, L. N. Gentile, J. H. Liu, J. Am. Chem. Soc. 1994, 116, 1597-1598; b) P. A. Wender, N. F. Badham, S. P. Conway, P. E. Floreancig, T. E. Glass, C. Grånicher, J. B. Houze, J. Jånichen, D. Lee, D. G. Marquess, P. L. McGrane, W. Meng, T. P. Mucciaro, M. Muhnlebach, M. G. Natchus, H. Paulsen, D. B. Rawlings, J. Satkofsky, A. J. Shuker, J. C. Sutton, R. E. Taylor, K. Tomooka, J. Am. Chem. Soc. 1997, 119, 2755-2756.
- [27] A natural sample of pectenotoxin-4 was unavailable for comparison. Professor M. Satake of Tohoku University is thanked for kindly providing copies of <sup>1</sup>H NMR spectra of pectenotoxin-1, -4, and -8, as well as samples of pectenotoxin-1 and -8.
- [28] K. Sasaki, J. L. C. Wright, T. Yasumoto, J. Org. Chem. 1998, 63, 2475 2480. We observed a 11:10:79 ratio of pectenotoxins-1:-4:-8.

0044-8249/02/4123-4576 \$ 20.00+.50/0 Angew. Chem. Int. Ed. 2002, 41, No. 23