

View Article Online View Journal

# ChemComm

## Accepted Manuscript

This article can be cited before page numbers have been issued, to do this please use: G. C. Senadi, B. Guo, W. Hu and J. Wang, *Chem. Commun.*, 2016, DOI: 10.1039/C6CC05138C.



This is an *Accepted Manuscript*, which has been through the Royal Society of Chemistry peer review process and has been accepted for publication.

Accepted Manuscripts are published online shortly after acceptance, before technical editing, formatting and proof reading. Using this free service, authors can make their results available to the community, in citable form, before we publish the edited article. We will replace this Accepted Manuscript with the edited and formatted Advance Article as soon as it is available.

You can find more information about *Accepted Manuscripts* in the **Information for Authors**.

Please note that technical editing may introduce minor changes to the text and/or graphics, which may alter content. The journal's standard <u>Terms & Conditions</u> and the <u>Ethical guidelines</u> still apply. In no event shall the Royal Society of Chemistry be held responsible for any errors or omissions in this *Accepted Manuscript* or any consequences arising from the use of any information it contains.



www.rsc.org/chemcomm



#### Journal Name

### COMMUNICATION

# Iodine-promoted cyclization of *N*-propynyl amides and *N*-allyl amides via sulfonylation and sulfenylation

Received 00th January 20xx, Accepted 00th January 20xx

Gopal Chandru Senadi,<sup>†,a</sup> Bing-Chun Guo,<sup>†,a</sup> Wan-Ping Hu,<sup>b</sup> and Jeh-Jeng Wang<sup>\*a</sup>

DOI: 10.1039/x0xx00000x

www.rsc.org/

An iodine-promoted regioselective cyclization of *N*-propynyl/allyl amides with sulfonyl hydrazides has been developed for the synthesis of 5-methyl-arylsulfonyloxazoles and 5-methylarylthiooxazolines via sulfonylation and sulfenylation reactions. The notable features of this reaction are the formation of new C-S and C-O bonds, the broad functional group tolerance, and its applicability to alkyl sulfonyl hydrazides as well as internal alkynes.

Sulfones and thioethers are important classes of functional groups with diverse applications as bioactive substituents.<sup>1</sup> Likewise, oxazoles and oxazolines are pivotal classes of oxaza heterocycles because of their widespread use in chemistry and biological applications.<sup>2</sup> Therefore, sulfur-containing oxaza heterocyles are of potential biological interest (Figure 1).<sup>3</sup>



Figure 1. Bioactive sulfur-containing oxazole and oxazoline heterocycles.

In the last decade, *N*-propargyl amides have gained significant popularity in the synthesis of oxazoles and oxazolines;<sup>4,5</sup> however, difficulties in controlling exo vs endo selectivity<sup>6</sup> and low feasibility with internal alkynes<sup>7</sup> are major drawbacks. Recently, an alternative strategy was used to overcome this issue by in situ installation of functional groups



Recently, sulfonyl hydrazides have been used to install sulfonyl or thioether functional groups in organic molecules, depending on reaction conditions.<sup>13</sup>,<sup>14</sup> As part of our ongoing research in metal-free reactions and oxaza heterocycle synthesis,<sup>15,5a</sup> we wish to extend the utilization of sulfonyl hydrazides to generate  $\beta$ -iodovinylsulfones from *N*-propargyl amides with I<sub>2</sub>/TBHP, followed by intramolecular cyclization in the presence of 1,8-Diazabicycloundec-7-ene (DBU) in a sequential one-pot method (Scheme 1b), for the synthesis of 2,5-disubstituted oxazoles. In addition, we disclose the synthesis of oxazolines from *N*-allylamides through electrophilic addition to an in situ generated sulfenyl iodide (Scheme 1c).





Scheme 1. Previous studies and this work towards the in situ functionalization of N-propynyl or N-allyl amide

Our preliminary investigation began with propargyl amide **1a** and tosyl hydrazide **2a** as model substrates, as shown in Table 1. To our surprise, N-((E)-2-iodo-3-tosyl-allyl)benzamide **3aa** was obtained in 67% when these substrates were treated

#### This journal is © The Royal Society of Chemistry 20xx

<sup>&</sup>lt;sup>a.</sup> Department of Medicinal and Applied Chemistry, Kaohsiung Medical University, No. 100, Shiquan 1st Rd, Sanmin District, Kaohsiung City, 807, Taiwan.

<sup>&</sup>lt;sup>b.</sup> Department of Biotechnology, Kaohsiung Medical University, No. 100, Shiquan 1st Rd, Sanmin District, Kaohsiung City, 807, Taiwan.

<sup>&</sup>lt;sup>†</sup> These two authors contributed equally.

Electronic Supplementary Information (ESI) available: [details of any supplementary information available should be included here]. See DOI: 10.1039/x0xx00000x

H 1a	1. [1] 0 xid: + TsNHNH <sub>2</sub> _ <u>solv</u> ; 90 % 2a 2. Di THF	ant (2.0 equiv) ent (2.0 mL) C, 2 h BU (2.0 equiv) (1.0 mL), rt, 2 h	N O S S 4aa		Ts J J J J J J J J
entry	l source (x	oxidant (y	aaluant	yield (%)	
	equiv)	equiv)	solvent	3aa <sup>b</sup>	4aa <sup>c</sup>
1	NIS (50)	TBHP (2)	CH₃CN	67	62
2	TBAI (50)	TBHP (2)	CH₃CN	25	20
3	KI (50)	TBHP (2)	CH₃CN	26	21
4	I <sub>2</sub> (50)	TBHP (2)	CH₃CN	75	70
5	I <sub>2</sub> (25)	TBHP (2)	CH₃CN	50	45
6	I <sub>2</sub> (50)	TBHP (1.5)	CH₃CN	60	54
7 <sup><i>d</i></sup>	I <sub>2</sub> (50)	TBHP (2)	CH₃CN	65	60
8	I <sub>2</sub> (50)	CHP (2)	CH₃CN	64	59
9 <sup>e</sup>	I <sub>2</sub> (50)	TBHP(2)	CH₃CN	40	35
10	I <sub>2</sub> (50)	DTBP (2)	CH₃CN	<10	-
11 <sup>f</sup>	I <sub>2</sub> (50)	H <sub>2</sub> O <sub>2</sub> (2)	CH₃CN	<10	-
12	I <sub>2</sub> (50)	K <sub>2</sub> S <sub>2</sub> O <sub>8</sub> (2)	CH₃CN	<10	-
13	I <sub>2</sub> (50)	TBHP (2)	1,4-dioxane	15	-
14	I <sub>2</sub> (50)	TBHP (2)	THF	<5	-
15	I <sub>2</sub> (50)	TBHP (2)	DMSO	0	-
16	I <sub>2</sub> (50)	TBHP (2)	DMF	0	-
17	I <sub>2</sub> (50)	TBHP (2)	toluene	<10	-
18 <sup>g</sup>	I <sub>2</sub> (50)	TBHP (2)	CH₃CN	75	21
19 <sup><i>h</i></sup>	I <sub>2</sub> (50)	TBHP (2)	CH₃CN	75	30
20 <sup>i</sup>	I <sub>2</sub> (50)	TBHP (2)	CH₃CN	75	45
<b>21</b> <sup>j</sup>	I <sub>2</sub> (50)	TBHP (2)	CH₃CN	75	63
22 <sup>k</sup>	I <sub>2</sub> (50)	TBHP (2)	CH₃CN	75	64
23		TBHP (2)	CH <sub>2</sub> CN	-	-

<sup>*o*</sup> Reaction condition: **1a** (1.0 mmol), **2a** (2.0 mmol), iodine source (x equiv), oxidant (y equiv), solvent (2.0 mL) and stirred for 2 h at 90 °C, then cooled to rt and diluted with THF (1.0 mL) followed by addition of DBU (2.0 equiv) and stirred at rt for another 2h unless otherwise noted. <sup>*b*</sup> Determined by GC-mass analysis. <sup>*c*</sup> Isolated yields. <sup>*d*</sup> 1.5 mmol of **2a** was used <sup>*e*</sup> TBHP (5-6 M in decane). <sup>*f*</sup> 30% H<sub>2</sub>O<sub>2</sub>. <sup>*g*</sup> DABCO (2.0 equiv). <sup>*h*</sup> K<sub>2</sub>CO<sub>3</sub> (2.0 equiv). <sup>*i*</sup> Cs<sub>2</sub>CO<sub>3</sub> (2.0 equiv). <sup>*j*</sup> 1.0 equiv DBU was used. <sup>*k*</sup> DBU (2.0 equiv) was added without THF.

with NIS (50 mol %) and TBHP (2.0 equiv) in acetonitrile for 2 h at 90 °C. The resultant compound was further treated with 2.0 equiv of DBU and THF (1.0 mL) to yield 2-phenyl-5-(tosylmethyl)oxazole 4aa in 62% yield in a sequential one-pot reaction (Table 1, entry 1). The structure of compound 4aa was confirmed using X-ray analysis.<sup>16</sup> Subsequent work (Table 1, entries 2-4) revealed that 50 mol% I<sub>2</sub> followed by treatment with DBU gave 4aa at a 70% yield. Lowering the quantities of I2, TBHP, and tosyl hydrazide, or screening other oxidants such as CHP, TBHP (5-6 M decane), DTBP, 30%  $H_2O_2$ , and  $K_2S_2O_8$ , resulted in lower yields (Table 1, entries 5-12). The effect of solvents was tested using dioxane, THF, DMSO, DMF, and toluene (Table 1, entries 13-17). However, no solvents resulted in improved yields. Bases besides DBU, such as DABCO, K<sub>2</sub>CO<sub>3</sub> and Cs<sub>2</sub>CO<sub>3</sub>, led to lower conversion (Table 1, entries 18-20). The yield of 4aa was reduced when only 1.0 equiv of DBU was used or when THF was omitted (Table 1, entries 21-22). The reaction gave a complex mixture in the absence of iodine (Table 1, entry 23). Thus, of the reaction conditions screened in Table 1, entry 4 was chosen as the optimum conditions.

The scope of this methodology was investigated by screening N-propynylamides 1(a-m) and aryl sulfonyl hydrazides 2(a-h) under the standard conditions, as shown in Table 2. The reaction worked well with benzene sulfonyl hydrazide (2b) as well as with a series of substituents of aryl functionalities such as p-MeO (2c), p-F (2d), p-Cl (2e), p-Br (2f), and p-CF<sub>3</sub> (2g) to afford oxazole derivatives 4(ab-ag) in 50-95% yield. Surprisingly, the reaction worked with mesyl hydrazide (2h) to produce the corresponding alkyl derivative 4ah, albeit in low yield. Substituted phenyl groups were tested at R<sup>1</sup>, including electron-donating groups such as p-Me (1b), p-MeO (1c) and *m*-MeO (1d), as well as electron-withdrawing groups such as p-F (1e), p-Cl (1f), p-Br (1g), p-I (1h), p-CF<sub>3</sub> (1i) and p-NO<sub>2</sub> (1j). The reaction proceeded well to give compounds 4(ba-ja) at 40-74% yields. Cumene hydroperoxide (CHP) was required for p-MeO (1c) and p-I (1h) due to low yields with TBHP. The reaction underwent smooth conversion when the aryl group was replaced with alkyl (1k) and heteroaryl (1m) groups to generate 4(ka-la) in moderate yields. Finally, the reaction also worked well with internal alkyne 1m to produce 4ma at a 68% yield. The structures of compounds 4ab, 4ac,

4ad, 4af, 4ag and 4ma were confirmed by X-ray analysis.<sup>16</sup>



<sup>*a*</sup> Reaction conditions: compound **1a-m** (1.0 mmol), sulfonyl hydrazide **2a-h** (2.0 mmol), I<sub>2</sub> (50 mol%), TBHP (2.0 equiv), CH<sub>3</sub>CN (2.0 mL), 90 °C, 2-5 h, then DBU (2.0 equiv), THF (1.0 mL), rt, 2-5 h. <sup>*b*</sup> isolated yields. <sup>*c*</sup> CHP was used instead of TBHP. <sup>*c*</sup> Without the addition of DBU the intermediate **3ja'** was isolated at a 45% yield; the structure was confirmed by X-ray. <sup>*d*</sup> After adding DBU (2.0 equiv) reaction was stirred for 24 h at 80 °C.

The synthetic feasibility of this procedure was extended to the reaction of  $\alpha$ , $\alpha'$ -disubstituted propynyl amides **1(n-o)** with tosyl hydrazide (**2a**) for the synthesis of vinyl sulfones as shown in Scheme 2 (eqs. 1 and 2). The reaction went smoothly to produce vinyl sulfones **4na-oa** in 50-53% yields. The structure of **4oa** was confirmed by X-ray analysis.<sup>16</sup>

DOI: 10.1039/C6CC05138C

Journal Name

Journal Name



Following the successful synthesis of the oxazole derivatives, we investigated the scope of the reaction between N-allyl benzamide (5a) and tosyl hydrazide (2a) using the standard conditions. However, the desired oxazoline was not obtained; instead, N-(2-hydroxy-3-tosylpropyl)benzamide 6aa' was isolated at low yields.<sup>17</sup> Interestingly, 5-((ptolylthio)methyl)-4,5-dihydro-2-phenyloxazole 6aa was isolated at 65% yields when the conditions were changed to I<sub>2</sub> (1.0 equiv) in toluene at 120 °C for 5 h (Table 3).18 Next, the scope the reaction between allyl amides and aryl hydrazides was evaluated using benzene sulfonyl hydrazide (2b) and 4bromo benzene sulfonyl hydrazide (2f). The reaction was successful in both the cases, affording oxazoline derivatives 6ab and 6af at 62% and 50% yields, respectively. The reaction also worked for mesyl hydrazide (2h) to give corresponding alkylthioethers 6ah and 6bh in moderate yields. Finally, the reaction also proceeded with the R<sup>1</sup> substituted aryls p-Me (5b) and *p*-F (5c), to produce 6(ba-ca) at 40-43% yields.



<sup>*a*</sup> Reaction conditions: compounds **5a-c** (1.0 mmol), sulfonyl hydrazide **2a**, **b**, **f**, **h** (1.5 mmol), I<sub>2</sub> (1.0 equiv), toluene (2.0 mL) at 120 °C for 5-7 h. <sup>*b*</sup> isolated yields. <sup>*c*</sup> In the absence of iodine, *S*-(*p*-tolyl) 4-methylbenzenesulfonothioate was isolated as major product and 1,2-di-*p*-tolyldisulfane was isolated as minor product with recovery of **5a**.

To understand the reaction mechanism, the control experiments shown in Scheme 3 were carried out. The reaction of *N*-propargyl amide **1a** with *S*-*p*-tolyl-4-methyl benzene sulfonothioate **2a'** under the standard condition without DBU gave trace amounts of *N*-((*E*)-2-iodo-3-tosylallyl)benzamide **3aa** (Scheme 3, eq. 3). This result suggests that **2a'** is not a key intermediate formed in the reaction.<sup>19a</sup> Next, the reaction of **1a** and **2a** under standard conditions from Table 2 in the presence of TEMPO gave trace or no product (Scheme 3, eq. 4). Finally, the reaction of **5a** and **2a** 

View Article Online DOI: 10.1039/C6CC05138C COMMUNICATION

under the standard conditions from Table 3 with TEMPO gave the compound 5aa at a 55% yield (Scheme 3, eq. 5). These results suggest that the sulfonylation reaction proceeds via radical reaction and that the sulfenylation proceeds through a non-radical pathway.



A plausible mechanism for the formation of oxazoles and oxazolines based on control studies and previously reported literatures<sup>13,14e,f</sup> is shown in Scheme 4. The reaction is proposed to involve a sulfonyl radical **E**, which is generated by the reaction of I<sub>2</sub> with TBHP through the elimination of N<sub>2</sub>. The attack of the resulting sulfonyl radical **E** to *N*-propynylamide **1** gives the vinyl radical **F**. The vinyl radical is trapped by I<sub>2</sub> to afford β-iodovinylsulfone **G**.<sup>13e</sup> Then, the reaction of the β-iodovinylsulfone with DBU gives the 5-exo-trig product **H**. The elimination of HI from intermediate **H** followed by the isomerization gave the oxazole derivative **4**<sup>19b</sup> (Scheme 4A).



On the other hand, upon reaction with iodine, sulfonyl hydrazide **2** undergoes sequential eliminations of HI and HOI to form sulfenyl iodide **N** via arylthiodiazonium salt **M**.<sup>14e,f</sup> The electrophilic addition of sulfenyl iodide **N** to *N*-allylamide **5** gives the sulfonium intermediate **O**. The 5-exo-tet-cyclization of intermediate **O** gives the intermediate **P**, which is deprotonated to give the oxazoline derivative **6**.

In conclusion, we have developed a metal-free approach for the synthesis of 5-methyl-arylsulfonyloxazoles by iodinepromoted sulfonylation of *N*-propynyl amides with sulfonyl hydrazides followed by DBU-mediated cyclization. Similarly, the oxysulfenylation of *N*-allyl amides, which proceeded via the electrophilic addition of in situ generated sulfenyl iodide, produced 5-methyl-arylthiooxazolines. The key features of these reactions are broad functional group tolerance, mild conditions, the formation of two new bonds (C-S and C-O) and applicability to alkyl sulfonyl hydrazides as well as to internal alkynes.

The authors gratefully acknowledge funding from the ministry of science and technology (MOST), Taiwan, and the Centre for Research and Development of Kaohsiung Medical University for 400 MHz NMR analyses, LC-MS and GC-MS analysis.

#### Notes and references

Published on 25 August 2016. Downloaded by Northern Illinois University on 25/08/2016 18:42:05.

- (a) S. Y. Woo, J. H. Kim, M. K. Moon, S. -H. Han, S. K. Yeon, J. W. Choi, B. K. Jang, H. J. Song, Y. G. Kang, J. W. Kim, J. Lee, D. J. Kim, O. Hwang and K. D. Park, *J. Med. Chem.*, 2014, 57, 1473; (b) A. Cohen, M. D. Crozet, P. Rathelot, N. Azas and P. Vanelle, *Molecules*, 2013, 18, 97; (c) K. A. M. Walker, D. J. Kertesz, D. M. Rotstein, D. C. Swinney, P. W. Berry, O. -Y. So, A. -S. Webb, D. M. Watson, A. Y. Mak, P. M. Burton, B. Mills-Dunlap, M. Y. Chiou, L. G. Tokes, L. J. Kurz, J. R. Kern, K. W. Chan, A. Salari and G. R. Mendizabal, *J. Med. Chem.*, 1993,36, 2236; (d) I. A. Leneva, R. J. Russell, Y. S. Boriskin and A. J. Hay, *Antiviral Res.*, 2009, 81, 132.
- 2 For reviews on oxazoles and oxazolines, see: (a) I. J. Turchi and M. J. S. Dewar, *Chem. Rev.*, 1975, **75**, 389; (b) H. H. Wasserman, K. E. McCarthy and K. S. Prowse, *Chem. Rev.*, 1986, **86**, 845; (c) I. J. Turchi, *Ind. Eng. Chem. Prod. Res. Dev.*, 1981, **20**, 32; (d) J. A. Frump, *Chem. Rev.*, 1971, **71**, 483; (e) A. I. Meyers and E. D. Mihelich, *Angew. Chem. Int. Ed.*, 1976, **15**, 250; (f) G. Desimoni and G. F. P. Quadrelli, *Chem. Rev.*, 2003, **103**, 3119.
- (a) WIPO., WO2010142934A1, 2010; (b) US pat., US6767426B1, 2004; (c) E. R. Freiter, A. H. Abdallah and S. J. Strycker, J. Med. Chem., 1973, 16, 510.
- For recent review on *N*-propynylamides, see: (a) Y. Hu, X. Xin,
   B. Wan, *Tetrahedron Lett*, 2015, 56, 32 and references cited therein.
- 5 For selected examples on synthesis of oxazoles and oxazolines from *N*-propynylamides, see: (a) G. C. Senadi, W. -P. Hu, J. -S. Hsiao, J. K. Vandavasi, C. -Y. Chen, and J. -J. Wang, *Org. Lett.*, 2012, **14**, 4478 and references cited therein; (b) J. P. Weyrauch, A. S. K.Hashmi, A. M. Schuster, T. Hengst, S. Schetter, A. Littmann, M. Rudolph, M. Hamzic, J. Visus, F. Rominger, W. Frey, J. W. Bats, *Chem. — Eur. J.*, 2010, **16**, 956; (c) V. H. L. Wong, A. J. P. White, T. S. Hor and K. K. Hii, *Adv. Synth. Catal.*, 2015, **357**, 3943 and references cited therein; (d) Y. Wang, M. Jiang and J. -T. Liu, *Org. Chem. Front.*, 2015, **2**, 542. (d) A. Alhalib and W. J. Moran, *Org. Biomol. Chem.*, 2014, **12**, 795.
- 6 (a) X. Meng and S. Kim, Org. Biomol. Chem., 2011, 9, 4429;
  (b) A. S. K. Hashmi, A. M. Schuster, M. Schmuck, and F. Rominger, Eur. J. Org. Chem., 2011, 4595.
- 7 (a) P. Wipf, Y. Aoyama, and T. E. Benedum, Org. Lett., 2004,
   6, 3593; (b) E. Merkul and T. J. J. Muller, Chem. Commun.,
   2006, 4817; (c) G. Bartoli, C. Cimarelli, R. Cipolletti, S.

DOI: 10.1039/C6CC05138C Journal Name

Diomedi, R. Giovannini, M. Mari, L. Marsili and E. Marcantoni, *Eur. J. Org. Chem.* 2012, 630.

- 8 A. Arcadi, S. Cacchi, L. Cascia, G. Fabrizi, F. Marinelli, *Org. Lett.*, 2001, **3**, 2501.
- 9 A. S. K. Hashmi, M. C. B. Jaimes, A. M. Schuster, and F. Rominger, J. Org. Chem., 2012, 77, 6394.
- 10 H. Peng, N. G. Akhmedov, Y. -F. Liang, N. Jiao, and X. Shi, J. Am. Chem. Soc., 2015, 137, 8912.
- 11 B. Wang, Y. Chen, L. Zhou, J. Wang, C. -H. Tung, Z. Xu, J. Org. Chem., 2015, **80**, 12718.
- 12 (a) N. G. Moon and A. M. Harned, *Tetrahedron Lett.*, 2013, 54, 2960; (b) G. -Q. Liu, C. -H. Yang, Y. -M. Li, *J. Org. Chem.*, 2015, 80, 11339; (c) A. Alhalib, S. Kamouka, and W. J. Moran, *Org. Lett.*, 2015, 17, 1453; (d) L. Engman, *J. Org. Chem.*, 1991, 56, 3425; (e) S. Minakata, Y. Morino, Y. Oderaotoshi, and M. Komatsu, *Org. Lett.*, 2006, 8, 3335; (f) Z. K. M. A. E. Samii, M. I. A. Ashmawy and J. M. Mellor, *Tetrahedron Lett.*, 1987, 28, 1949; (g) J. Yu, H. Tian, C. Gao, H. Yang, Y. Jiang, H. Fu, *Synlett*, 2015, 26, 676.
- 13 For selected examples on sulfonyl insertion from sulfonylhydrazides, see: (a) (a) L. Zhang, S. Chen, Y. Gao, P. Zhang, Y. Wu, G. Tang, and Y. Zhao, *Org. Lett.*, 2016, **18**, 1286; (b) W. –J. Hao, Y. Du, D. Wang, B. Jiang, Q. Gao, S. –J. Tu and G. Li, *Org. Lett.*, 2016, **18**, 1884; (c) R. Singh, B. K. Allam, N. Singh, K. Kumari, S. K. Singh, and K. N. Singh, *Org. Lett.* 2015, **17**, 2656; (d) J. Zhang, Y. Shao, H. Wang, Q. Luo, J. Chen, D. Xu, and X. Wan, *Org. Lett.*, 2014, **16**, 3312; (e) X. Li, X. Xu and X. Shi, *Tetrahedron Lett.*, 2013, **54**, 3071; (f) W. Wei, C. Liu, D. Yang, J. Wen, J. You, Y. Suo and H. Wang, *Chem. Commun.*, 2013, **49**, 10239.
- 14 For recent examples of sulfenylation, see: (a) Y. Yang, S. Zhang, L. Tang, Y. Hu, Z. Zha and Z. Wang, *Green Chem.*, 2016, **18**, 2609; (b) X. Zhao, T. Li, L. Zhang and K. Lu, *Org. Biomol. Chem.*, 2016, **14**, 1131; (c) X. Li, Y. Xu, W. Wu, C. Jiang, C. Qi and H. Jiang, *Chem. Eur. J.*, 2014, **20**, 7911; (d) G. Kumaraswamy, and R. Raju, *Adv. Synth. Catal.*, 2014, **356**, 2591; (e) F. -L. Yang, F. -X. Wang, T. -T. Wang, Y. -J. Wang and S. -K. Tian, *Chem. Commun.*, 2014, **50**, 2111; (f) F. -L. Yang and S. -K. Tian, *Angew. Chem. Int. Ed.*, 2013, **52**, 4929.
- (a) G. C. Senadi, B. S. Gore, W. P. Hu, and J. J. Wang, Org. Lett., 2016, 18, 2890; (b) J. K. Vandavasi, W. P. Hu, G. C. Senadi, H. T. Chen, H. Y. Chen, K. C. Hsieh and J. J. Wang, Adv. Synth. Catal., 2015, 357, 2788; (c) G. C. Senadi, W. P. Hu, T. Y. Lu, A. M. Garkhedkar, J. K. Vandavasi and J. J. Wang, Org. Lett., 2015, 17, 1521; (d) S. S. K. Boominathan, W. P. Hu, G. C. Senadi, J. K. Vandavasi, J. J. Wang, Chem. Commun., 2014, 51, 6726; (e) W. C. Lee, H. C. Shen, W. P. Hu, W. S. Lo, C. Murali, J. K. Vandavasi and J. J. Wang, Adv. Synth. Catal., 2012, 354, 2218.
- 16 The CCDC number for the compounds **4aa** (1465160), **4ab** (1465161), **4ac** (1465166), **4ad** (1465163), **4af** (1465164), **4ag** (1465165), **4ma** (1465167), **4oa** (1465162), and **3ja'** (1473425).
- 17 When *N*-allylbenzamide **5a** and tosylhydrazides **2a** was treated with I<sub>2</sub>/TBHP, the desired oxazoline was not isolated instead *N*-(2-hydroxy-3-tosylpropyl)benzamide **6aa'** was isolated in less than 10% along with other unidentified compounds.
- 18 For complete optimization of reaction condition (see the ESI+).
- 19 (a) The sulfonyl radical does not form via the intermediate Sp-tolyl-4-methylbenzenesulfonothioate 2a' as proposed in ref 13c; (b) Intermediate H was proposed based on the acidity of the methylene proton and result obtained from Scheme 2.