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Catalaytic Isomerization of 1,5-Enynes to Bicyclo[3.1.0]hexenes

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Transition metal-catalyzed isomerization and rearrangement reactions of unsaturated systems provide access to structural motifs not accessible through their thermal counterparts. This is exemplified by the numerous applications of transition metal-catalyzed Alder-ene reactions of 1,6- and 1,7-enynes for the synthesis of cyclopentyl and cyclohexyl ring systems.¹ The corresponding skeletal rearrangements of simple 1,5-envnes are much less studied. Berson and co-workers conjectured that the thermal rearrangement of 1,5-enyne 1 proceeds via bicyclo[3.1.0]hexene 2 to afford toluene and triene 3 as the major constituents of a complex mixture.² Scattered reports of transition metal-catalyzed isomerizations of 1,5-enynes³ exist; however, these generally employ enol ethers as the ene component.^{4,5} While enols are expected to be excellent nucleophiles,⁶ we were intrigued by the possibility that metal-alkyne complexes could be electrophilic enough to react even with simple olefins and catalyze processes related to the thermal rearrangement.

$$\underset{Me}{\longrightarrow} \underset{Me}{\overset{Me}{\longrightarrow}} \left[\underset{2}{\overset{Me}{\longrightarrow}} \right] \underset{Me}{\longrightarrow} \underset{3}{\overset{Me}{\longrightarrow}} \underset{Me}{\overset{Me}{\longrightarrow}} \underset{1}{\overset{Me}{\longrightarrow}} \underset{1}{\overset{Me}{\overset{Me}{\longrightarrow}} \underset{1}{\overset{Me}{\longrightarrow}} \underset{1}{\overset{Me}{\overset{Me}{\longrightarrow}} \underset{1}{\overset{Me}{\overset{Me}{\longrightarrow}} \underset{1}{\overset{Me}$$

To this end, treatment of 1,5-enyne **4** with 1 mol % palladium-(II) or platinum(II) complexes returned mainly starting material (eq 2). Both silver(I) tetrafluoroborate and triphenylphosphinegold(I) chloride failed to catalyze the rearrangement of **4**. On the other hand, the combination of these two complexes⁷ rapidly (5 min) and cleanly produced bicyclo[3.1.0]hexene⁸ **5**, an olefin isomer of the proposed intermediate (**2**) in the thermal isomerization. In sharp contrast to the gold(I)-catalyzed cyclizations of ω -alkynyl β -ketoesters,⁹ none of the competing 5-*exo-dig* cyclization to afford an *exo*-methylene product was observed. Finally, gold(III) chloride also catalyzed this reaction, however, with significantly lower conversion. On the other hand, 5% AuCl₃ with 15% AgOTf gave complete conversion; however, this was accompanied by a substantial amount of decomposition.



A range of 1,5-enynes undergo the triphenylphosphinegold(I)catalyzed rearrangement (Table 1). The propargylic position can be unsubstituted (entry 6) or substituted with aryl (entries 1-5) or alkyl substituents (entries 8 and 9). Additionally, potentially nucleophilic aromatic groups (entry 2) do not interfere with the reaction.¹⁰ Introduction of an alkyl group at the allylic position is also tolerated, producing bicyclo[3.1.0]hexene **15** as a 10:1 mixture of diastereomers¹¹ (entry 6). The rearrangement proceeds when the 1,5-enyne (**16**) is unsubstituted at both the allylic and propargylic positions, albeit with slightly decreased efficiency (entry 7).

Table 1. Au(I)-Catalyzed Synthesis of Bicyclo[3.1.0]hexenes ^a					
entry	substrate		(mol%) catalyst	product	yield
1	Ar = Ph	(4)	1% (Ph ₃ P)AuPF ₆		(5) 99%
2	Ar = MeO	(6)	2% (Ph ₃ P)AuPF ₆	Ar	(7) 99%
3	Ar = O Br	(8)	1% (Ph ₃ P)AuSbF ₆		(9) 95%
4	Ph	(10)	1% (Ph ₃ P)AuSbF ₆	Ph	(11) 94%
5	Ph	(12)	3% (Ph ₃ P)AuSbF ₆	Ph	(13) 96%
6	Ph	(14)	3% (Ph ₃ P)AuSbF ₆	Ph H	(15) 82% (10:1 dr) ^b
7	OTIPS	(16)	2% (Ph ₃ P)AuSbF ₆		(17) ^{61%°} (>99:1 dr) ^b
8	Ph	(18)	1% (Ph ₃ P)AuSbF ₆	Ph	(19) 98% (>99:1 dr) ^b
9	Ph	(20)	1% (Ph ₃ P)AuSbF ₆	Ph	(21) 96% (97:3 dr) ^b
10	Me., OMe OMe	TIPS (22) ee, 98:2 dr	3% (Ph ₃ P)AuPF ₆	Me, H H H H OMe OMe	OTIPS (23) 99% 91% ee, >99:1 dr ^t

 a Reaction conditions: 0.5 M 1,5-enyne in dichloromethane, rt. b Diastereomeric ratio determined by $^1{\rm H}$ NMR. c Starting material (19%) was recovered.

Terminal and internal alkynyl substrates (entry 4) react with equal facility, the latter producing an allylic quaternary center. Substrates containing either 1,1- (entry 5) or 1,2-disubstituted (entries 8-10) olefins cleanly undergo the gold(I)-catalyzed isomerization. For example, 3 mol % triphenylphosphinegold(I) hexafluoroantimonate smoothly catalyzes the formation of **13** by the rearrangement of 1,1-disubstituted olefin **12**.

To gain insight into the mechanism of this transformation, we studied the stereochemical course of the rearrangement. We found that the gold(I)-catalyzed reaction of substrates containing 1,2-disubstituted olefins is stereospecific. For example, (*E*)-olefin **18** selectively affords *trans*-cyclopropane **19** (entry 8), while (*Z*)-olefin **20** produces a 97:3 mixture of diastereomers in favor of *cis*-cyclopropane **21** (entry 9). Additionally, enantioenriched 1,5-enyne **22** is isomerized to **23** with excellent chirality transfer (entry 10).¹² Finally, deuterium-labeled 1,5-enyne **24** underwent gold(I)-catalyzed conversion to a bicyclo[3.1.0]hexene (**25**) in which the deuterium was selectively incorporated at the vinyl position (eq 3).



On the basis of these data, we propose the process detailed in Scheme 1 as the most likely mechanism for this transformation. Coordination of cationic gold(I) to the alkyne followed by nucleo-philic addition of the pendant olefin produces cyclopropylcarbinyl¹³ cation **27**, which may have some gold(I) carbene character (**28**). The bicyclo[3.1.0]hexene product is generated by a 1,2-hydrogen shift onto a cation or a gold(I) carbene. The stereoselectivity and stereospecificity of the reaction can be accounted for by considering half-chair transition states, with the large groups occupying pseudoequatorial positions, similar to those proposed for the acetylenic Cope rearrangement.¹⁴

The proposed mechanism suggests that cationic intermediate **27**/ **28** could potentially be trapped in the presence of a nucleophile. In accord with this hypothesis, cyclohexenyl methyl ether **30** was produced when the gold(I)-catalyzed reaction of enyne **29** was carried out in methanol (eq 4).¹⁵ Notably, for this reaction the presence of a quaternary carbon at the propargylic position is necessary to prevent competing formation of the bicyclo[3.1.0]hexene; however, in the absence of nucleophile, a 1,2-alkyl shift is observed. For example, 1,5-enynes **31a** and **31b** undergo a gold-(I)-catalyzed tandem cycloisomerization—ring enlargement process¹⁶ to afford tricyclic structure **32a** and **32b** in 72 and 66% yield, respectively (eq 5).



In conclusion, we have developed a transition metal-catalyzed rearrangement of 1,5-enynes that produces bicyclo[3.1.0]hexenyl products that are isomeric to those produced as intermediates in the thermal reaction. The gold(I)-catalyzed reaction can be comducted under "open-flask" conditions and as such can be combined with our rhenium-catalyzed propargylic allylation¹⁷ to provide a one-pot synthesis of bicyclo[3.1.0]hexenes from propargyl alcohols (eq 6). This carbon—carbon bond-forming reaction provides a stereospecific method for the synthesis of a variety of cyclopropane containing carbocycles, including tricyclic structures prepared by a tandem cycloisomerization—ring enlargement reaction. Development of gold(I)-catalyzed¹⁸ carbon—carbon bond-forming reactions, including an enantioselective version of this cycloisomerization, is ongoing and will be reported in due course.

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Supporting Information Available: Experimental procedures and compound characterization data (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

Scheme 1. Mechanistic Proposal for Au(I)-Catalyzed Cycloisomerization



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