Tetrahedron Letters,Vol.29,No.23,pp 2779-2782,1988 0040-4039/88 \$3.00 + .00 Printed in Great Britain Pergamon Press plc

AN EFFICIENT SYNTHESIS OF 4-BENZOYLOXYAZETIDINONE: AN IMPORTANT CARBAPENEM INTERMEDIATE

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Summary: A short, stereoselective synthesis of the key carbapenem intermediate (3R,4R)-3~[(1R)-1-hydroxyethyl]-4-benzoyloxyazetidin-2-one is described.

Recently there has been considerable interest in the development of new synthetic approaches to β -lactam antibiotics.^{1,2} In searching for an efficient and versatile route to thienamycin³ and related β -methyl carbapenems⁴ we have explored the use of β -hydroxybutyric acid as a readily available, chiral starting material. This reagent has previously been used in enolate condensations with imines to generate β -lactam intermediates; however, these approaches typically provide modest yields of condensation products and require extensive functional group manipulation in order to liberate the desired C-4 acetoxy azetidinone species.² Our strategy has focused on the hydroxybutyrate derived ketene-imine cycloaddition for direct generation of the useful C-4 keto-substituted azetidinone. Although ketene-imine cycloadditions have been widely used for the formation of β -lactams, these reactions usually provide moderate yields of azetidinones which lack the functionality necessary for rapid conversion to useful carbapenems.⁵

We wish to report here an efficient, diastereoselective conversion of methyl 3(S)-hydroxybutyrate to (3S,4S)-3-[(1S)-1-hydroxyethyl]-4benzoylazetidin-2-one (8) and its subsequent transformation to the versatile carbapenem precurser <u>11</u>.

3(s)-Triisopropylsilyloxybutyric acid (2) was treated with oxalyl chloride at room temperature to produce acid chloride 3 in quantitative yield. Ketene 4 was generated in situ using diisopropylethylamine⁶ and reacted with imine 5⁷ (Ar=p-methoxyphenyl) to afford a 7:1 mixture⁸ of cis β -lactams 6 and 7 in 90% overall yield.^{9,10}





It is important to note that the major isomer 6 posseses the C-3 stereochemistry required for thienamycin.¹¹ Also, the use of the keto-imine 5 is crucial as it provides direct access to a useful C-4 keto-substituted β -lactam¹² which is not readily available via standard enolate chemistry.² Treatment of <u>6</u> with tetrabutylammonium fluoride in THF led to desilylation as well as epimerization of C-4 to form trans β -lactam <u>8</u> in 78% yield. Under the standard Mitsunobu conditions, ¹³ (S) hydroxylethyl <u>8</u> inverted to the desired was (R) hydroxyethyl β -lactam <u>9</u> in 74% yield.



Removal of the p-methoxyphenyl group was achieved by cerric ammonium nitrate oxidation to provide unprotected β -lactam <u>10</u> in 67% yield.¹⁴ Baeyer-Villiger oxidation with <u>m</u>-chloroperoxybenzoic acid¹² at room temperature led to benzoyloxy azetidinone <u>11</u> which was subsequently chain extended under the conditions of Reider and Grabowski¹⁵ to form the known carbapenem intermediate <u>12</u>.¹⁶



It is also interesting to note that an intermediate such as <u>11</u> has been utilized for the stereoselective synthesis of $1-\beta$ -Me-carbapenems.¹⁷

Thus we have demonstrated a short and efficient synthesis of important carbapenem intermediates using the β -hydroxybutyrate derived ketene-imine cycloaddition.

Acknowledgement: We thank Mr. R. Reamer for valuable NMR assistance.

References and Notes

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- FTIR studies indicate that ketene $\underline{4}$ is indeed the reactive species. 6. Manuscript in preparation.
- 7. Imine 5 was produced by condensation of anisidine with phenyl glyoxal.
- 8. The best ratio of 6:7 was achieved using DMF as the solvent.
- Our studies show that the ratio of 6:7 improves as the steric bulk 9. of the hydroxyethyl protecting group increased. Conrotatory cyclization as shown below for the iminium ion \underline{A} , in its preferred conformation, would be expected to minimize steric interaction of the bulky hydroxyethyl group and the incipient C-4 substituent. See also reference 18.



- 10. The use of (R)hydroxybutyrate also led to a mixture of cis β -lactams. Unfortunately the major product possessed the undesired C-3 stereochemistry.
- Data for selected intermediates: <u>6</u>: ¹H NMR (CDCl₃, 300 MHz) δ 0.90 11. (18H,s), 0.90(3H,m) 1.34(3H,d,J=6.4 Hz), 3.77(3H,s), 3.93(1H,dd, J=3.8,6.3 Hz), 4.31(1H,m), 5.40(1H,d,J=6.3 Hz), 6.83(2H,m), 7.24 (2H_m), 7.52(2H,m), 7.63(1H,m), 8.01(2H,m). ^LH NMR (CDCl₃,300 MHz) δ 1.39(3H,d,J=6.3 MHz), 3.23(1H,dd,J=2.4, 9: 6.5 Hz), 4.38(IH,m), 5.52(1H,d,J=2.4 Hz), 6.82(2H,m), 7.20(2H,m), 7.55(2H,m), 7.67(1H,m), 8.20(2H,m). 10: H NMR (CD₃OD,300 MHz) δ 1.27(3H,d,J=6.4 Hz), 3.10(1H,dd,J=2.4, 6.9 Hz), 4.20(1H,m), 5.14(1H,d,J=2.3 Hz), 7.52(2H,m), 7.65(1H,m), 8.15(2H,m). <u>11</u>: H NMR (CD₃CN, 300 MHz) δ 1.27(3H,d,J=6.4 Hz), 3.34(1H,dd,J=1.3, 5.8 Hz), 4.11(1H,m), 6.05(1H,d,J=2.0 Hz), 7.42(1H,bs), 7.51(2H,m), 7.66(1H,m), 8.40(2H,m). Hanessian, S.; Bedeschi, A.; Battistini, C.; Mongelli, N.; J. Am. 12. Chem. Soc. 1985, 107, 1438.
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(Received in USA 25 March 1988)