

Synthesis and Reactions of 2(5)-[Benzyl or Cyanomethyl]-1,3,4-oxadiazoles

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A series of five membered heterocycles, namely 5-(benzyl or cyanomethyl)-3-acetyl-2,2-disubstituted-1,3,4-oxadiazolines (3a-e), 2,5-disubstituted-1,3,4-oxadiazoles (4a, b), 2-hydroxy-5-(benzyl or cyanomethyl)-1,3,4-oxadiazoles (5a, b), 1,2,5-trisubstituted-1,3,4-triazoles (6a, b), and 2-(benzyl or cyanomethyl)-1,3,4-oxadiazole-5-thiols (10a, b), were synthesized. Also 5-chloro-2-benzyl-1,3,4-oxadiazole 7 was prepared. Reaction of amines, hydrazines, and sodium azide with 7 gave the corresponding 2-arylamino (8a, b), 5-hydrazino or phenylhydrazino (8c, d) and 2-azido derivatives 9, respectively. Mannich bases (11a, b) were prepared by the reaction of sec. amines with 9a. 5-Carboxymethylthio-1,3,4-oxadiazoles (12a, b) and their ethyl esters (13a, b) were also prepared. The ester 13a reacted with amines and hydrazines to give the corresponding amides (14a-d).

Synthese und Reaktionen von 2(5)-[Benzyl- oder Cyanomethyl]-1,3,4-oxadiazolen

Eine Serie fünfgliedriger Heterocylen, nämlich 5-(Benzyl- oder Cyanomethyl)-3-acetyl-2,2-disubstituierte-1,3,4-oxadiazoline (3a-e), 2,5-disubstituierte 1,3,4-Oxadiazole (4a, b), 2-Hydroxy-5-(benzyl- oder cyanomethyl)-1,3,4-oxadiazole (5a, b), 1,2,5-trisubstituierte 1,3,4-Triazole (6a, b) und 2-(Benzyl- oder Cyanomethyl)-1,3,4-oxadiazol-5-thiole (10a, b) wurden synthetisiert. Auch 5-Chlor-2-benzyl-1,3,4-oxadiazol (7) wurde hergestellt. Reaktionen von Aminen, Hydrazinen und Natriumazid mit 7 führten zu den entspr. 2-Arylamino-(8a, b), 5-Hydrazino- oder Phenylhydrazino (8c, d) und 2-Azido-Derivaten (9). Sek. Amine reagieren mit 9a zu den Mannich-Basen 11a, b. Auch die 5-Carboxymethylthio-1,3,4-oxadiazole (12a, b) wurden hergestellt, desgleichen die entspr. Ethylester 13a, b. Der Ester 13a reagierte mit Aminen und Hydrazinen zu den entspr. Amiden 14a-d.

A considerable number of 1,3,4-oxadiazole derivatives were found to have bactericidal, analgesic, muscle relaxant and tranquilizing properties¹⁻⁵. 5-Aryl-2-thiono-1,3,4-oxadiazoles have analgesic, antipyretic, antiphlogistic properties^{6,7} and show antitubercular activity⁸. This manuscript describes the synthesis and reactions of some 1,3,4-oxadiazoles.

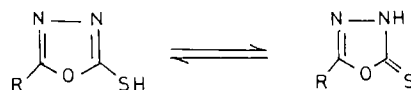
Thus, 2-cyano (or phenyl) ethanoic acid hydrazides 1a, b were condensed with selected carbonyl compounds to give the hydrazones 2a, f, which upon treatment with acetic anhydride afforded the corresponding 5-benzyl- (or cyanomethyl)-3-acetyl-2,2-disubstituted-1,3,4-oxadiazolines 3a, e.

Interaction of the hydrazide 1a or b with aliphatic or aromatic acid chlorides in pyridine yielded the corresponding N,N'-diacylhydrazines as isolated intermediates, which upon reflux with POCl₃⁹ or acetic anhydride gave the expected 2,5-disubstituted-1,3,4-oxadiazoles 4a, b. Prolonged heating of 4a with the appropriate aliphatic primary amine in xylene yielded the corresponding triazole derivatives 6a, b.

Refluxing of equimolar amounts of 1a or b and ClCOOC₂H₅ produced 2-hydroxy-5-(benzyl or cyanomethyl)-1,3,4-oxadiazoles 5a, b. 5a reacted with SOCl₂¹⁰ to give 5-chloro-2-benzyl-1,3,4-oxadiazole (7).

When N-nucleophiles¹⁰ such as aniline, p-toluidene, hydrazine hydrate, and phenylhydrazine were reacted with 7, 2-arylamino, hydrazino, or phenylhydrazino-5-benzyl-1,3,4-oxadiazoles 8a-d were readily obtained. On the other hand when 7 was reacted with NaN₃ in boiling acetic acid, the 2-azido derivative 9 was obtained.

Condensation of hydrazide 1a or b with KOH and CS₂ in ethanol¹¹ gave the corresponding 2-(benzyl or cyanomethyl)-1,3,4-oxadiazol-5-thiols 10a, b. Evidently¹², these compounds exist as thiol-thione equilibrium:



Thus, compound 10a reacted with piperidine or morpholine in the presence of formaldehyde to give the expected Mannich bases¹³ 11a, b. On the other hand 2-(benzyl or cyanomethyl)-5-carboxymethylthio-1,3,4-oxadiazoles 12a, b and their esters 13a, b were obtained by condensation of 10a and b with chloroacetic acid in alkaline medium or with ethyl bromoacetate in the presence of NaOH¹⁴, respectively.

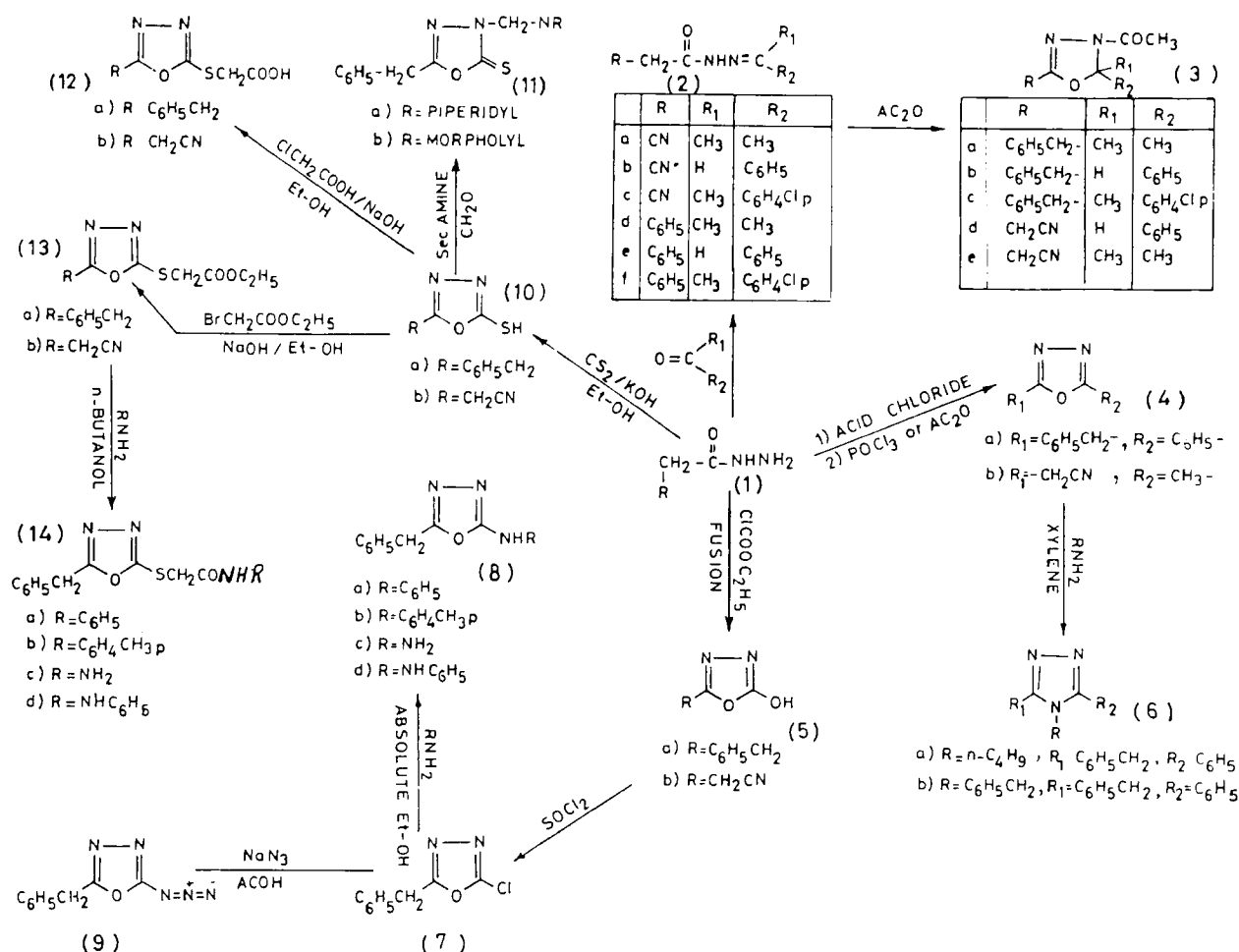
The ester 13a reacted readily with prim. aromatic amines and hydrazines, namely: aniline, p-toluidene, hydrazine hydrate, and phenylhydrazine giving the corresponding amides, hydrazide, and phenylhydrazide derivatives 14a-d, respectively.

Experimental Part

Melting points: uncorrected. - IR spectra (KBr): Pye-Unicam SP 1000 (ν in cm⁻¹). - ¹H-NMR-spectra (CDCl₃): Varian EM-390-90 MHz, TMS as internal reference (chemical shift δ in ppm).

N-Arylidene-N'-(cyanoacetyl or phenylacetyl)-hydrazine 2a-f

The appropriate carbonyl compound (0.01 mol) was added to a hot solution of 1a or b (0.01 mol) in 50 ml ethanol and the mixture was refluxed for 1 h. The solid which separated on cooling was washed with ethanol and recrystallized from the proper solvent (Table 1).



5-(Benzyl or cyanomethyl)-3-acetyl-2,2-disubstituted-1,3,4-oxadiazolines 3a-e

The appropriate hydrazine 2a-f (2g) was heated under reflux with acetic anhydride (10 ml) for 30 min. After cooling, the mixture was poured onto 100 g crushed ice and stirred for 20 min. The separated solid was washed with water and recrystallized from the suitable solvent (Table 1). The solid was washed with water and recrystallized from the suitable solvent (Table 1). The IR-spectrum of 3a exhibited bands at 1700 (C=O), 1610 (C=N), and 1250 (C-O-C). The ¹H-NMR spectrum of 3d displayed signals at 8.2-7.4 (m, 6H; oxadiazoline-H and arom-H), 2.8 (s, 2H, CH₂-CN) and 2.3 (s, 3H; COCH₃).

2,5-Disubstituted-1,3,4-oxadiazoles 4a, b

Preparation of N,N'-diacylhydrazines

To a solution of 1a or b (0.01 mol) in 50 ml pyridine, the appropriate acid chloride (0.01 mol) was added dropwise and the mixture was heated under reflux for 20 min. On cooling, the mixture was poured onto 100 ml dil. HCl and the separated solid was washed and dried.

Cyclization of N,N'-diacylhydrazines: Formation of 4a, b

A mixture of the appropriate N,N'-diacylhydrazine (the solid separated as described above) (0.01 mol) and 10 ml of POCl₃ or acetic anhydride was refluxed for 6 h. The mixture was distilled under reduced pressure and the solid which separated was crystallized from the proper solvent (Table 1).

The IR-spectrum of 4b shows bands at 2250 (C≡N), 1630 (C=N), and 1210 cm⁻¹ (C-O-C). - ¹H-NMR Spectrum of 4a reveals signals at 8.6-7.8 (m, 10H; arom. H) and 2.9 (s, 2H; benzylic-H).

1,2,5-Trisubstituted-1,3,4-triazoles 6a, b

To a solution of 4a (0.01 mol) in 40 ml xylene, the appropriate primary aliphatic amine (0.02 mol) was added and the mixture was refluxed for 12 h. On cooling a solid precipitated which was washed with xylene and crystallized from the proper solvent (Table 1).

The IR-spectrum of 6b shows bands at 1650 (C=N) and at 1000 cm⁻¹ (C-N). - The ¹H-NMR spectrum of the same compound displays signals at 7.9-7.3 (m, 15H; arom. H), and 2.7 (s, 4H; benzylic-H).

2-Hydroxy-5-(benzyl or cyanomethyl)-1,3,4-oxadiazoles 5a, b

After refluxing 1a or b (0.01 mol) with ethyl chloroformate (0.01 mol) in an oilbath for 3 h the separated solid was recrystallized from the suitable solvent, forming 5a, b.

The IR-spectrum of 5a shows bands at 1700 (C=O), and 3250 cm⁻¹ (N-H) (these bands confirm the assumption that 5a exists as the keto tautomer), and 1620 cm⁻¹ (C=N). - The ¹H-NMR spectrum of 5b exhibits signals at 2.6 (s, 2H; CH₂-CN) and 2.2 (broad s, 1H; O-H or N-H).

5-Chloro-2-benzyl-1,3,4-oxadiazole (7)

SOCl₂ (10 ml) was added dropwise to 5a (0.01 mol), the mixture was warmed on a water bath for 7 h, cooled and then poured onto 100 ml

Table 1: Analytical data of compounds 2 - 14.

Comp. No.	M.P. °C Solvent	Yield %	Molecular- Formula (Mol. Wt)	Analysis Calcd/Found		
				C	H	N
2a	149-150	85	C ₆ H ₉ N ₃ O	51.8	6.52	30.2
	Et-OH		(139.2)	51.6	6.44	30.1
2b	183	82	C ₁₀ H ₉ N ₃ O	64.2	4.85	22.
	Et-OH		(187.2)	64.1	4.72	22.5
2c	190	75	C ₁₁ H ₁₀ N ₃ ClO	56.1	4.28	17.8
	Et-OH		(235.7)	55.9	4.15	17.7
2d	109-111	80	C ₁₁ H ₁₄ N ₂ O	69.5	7.42	14.7
	Et-OH		(190.3)	69.3	7.29	14.6
2e	146-148	87	C ₁₅ H ₁₄ N ₂ O	75.6	5.92	11.8
	n-Pr-OH		(238.3)	75.5	5.90	11.6
2f	164-166	79	C ₁₆ H ₁₅ N ₂ ClO	67.0	5.27	9.8
	n-Pr-OH		(286.8)	66.8	5.10	9.7
3a	160	48	C ₁₂ H ₁₆ N ₂ O ₂	67.2	6.94	12.1
	n-Pr-OH		(232.3)	67.0	6.84	12.0
3b	220	52	C ₁₇ H ₁₆ N ₂ O ₂	72.8	5.75	10.0
	Me-OH		(280.3)	72.6	5.66	9.8
3c	213-215	45	C ₁₈ H ₁₇ N ₂ ClO ₂	65.8	5.21	8.5
	Me-OH		(328.8)	65.5	5.20	8.4
3d	150	55	C ₁₂ H ₁₁ N ₃ O ₂	62.9	4.84	18.3
	dil		(229.2)	62.6	4.7	18.2
3e	ACOH	43	C ₉ H ₁₁ N ₃ O ₂	53.0	6.12	23.2
	>280		(181.2)	52.9	6.1	22.9
4a	104-105	65	C ₁₅ H ₁₂ N ₂ O	76.3	5.12	11.9
	Et-OH		(236.3)	76.1	4.8	11.6
4b	85.86	54	C ₅ H ₅ N ₃ O	48.8	4.09	34.1
	Et-OH		(123.1)	48.5	3.8	34.1
5a	225	68	C ₉ H ₈ N ₂ O ₂	61.4	4.58	15.9
	Dioxane		(176.2)	61.2	4.4	15.6
5b	203	60	C ₄ H ₃ N ₃ O ₂	38.4	2.42	33.6
	Dioxane		(125.1)	38.3	2.3	33.4
6a	144	38	C ₁₉ H ₂₁ N ₃	78.3	7.26	14.4
	Benzene		(291.4)	78.1	7.2	14.3
6b	162	32	C ₂₂ H ₁₉ N ₃	81.2	5.89	12.9
	Et-OH		(325.4)	80.9	5.6	12.7
7	140	75	C ₉ H ₇ N ₂ ClO	55.5	3.63	14.4
	Me-OH		(194.6)	55.3	3.4	14.1

petroleum ether (60-80°C). The precipitate was crystallized from the suitable solvent (Table 1).

2-Arylamino-5-benzyl-1,3,4-oxadiazoles **8a, b**

To a solution of **7** (0.01 mol) in 30 ml ethanol, the appropriate amine (0.01 mol) was added and the reaction mixture was refluxed for 4 h. The mixture was concentrated and the precipitate was recrystallized from the proper solvent. The IR-spectrum of **8a** shows bands at 3340 (N-H), 1630 (C=N), and 1240 cm⁻¹ (C-O-C). -The ¹H-NMR spectrum of the same compound displays signals at 8.2-7.7 (m, 10H; arom-H), 2.7 (s, 2H; benzylic-H), and 2.1 (broad s, 1H; N-H).

5-Hydrazino or phenylhydrazino-2-benzyl-1,3,4-oxadiazoles **8c, d**

To a solution of **7** (0.005 mol) in 50 ml ethanol, hydrazine hydrate (1 ml; 100%) or phenylhydrazine hydrochloride (0.005 mol), and 5 ml KOH (10%) were added. The mixture was left over night. The products which were obtained were recrystallized from the suitable solvent (Table 1).

2-Azido-5-benzyl-1,3,4-oxadiazole (**9**)

A mixture of **7** (0.01 mol), NaN₃ (0.01 mol), acetic acid (30 ml), and water (5 ml) was refluxed for 5 h. The product obtained after concentration was crystallized from the suitable solvent. The IR-spectrum of **9** reveals bands at 2150 (azide) and 1640 (C=N).

2-(Benzyl or cyanomethyl)-1,3,4-oxadiazole-5-thiols **10a, b**

A mixture of **1a** or **b** (0.1 mol), KOH (0.1 mol), and 20 ml CS₂ in 50 ml ethanol was heated under reflux and stirring on a water bath until the evolution of H₂S ceased (about 6 h). The excess solvent was removed by distillation and the residue was stirred with cold water, filtered and the filtrate was acidified with dil. HCl. The precipitated solid was washed with water and crystallized from the proper solvent (Table 1). The IR-Spectrum of **9a** shows bands at 3370 (N-H), 1220 (C-O-C), and 1280-1290 (C=S), the latter band confirms the assumption that the compound exists as the thione tautomer.

Table 1: Cont.

Comp. No.	M.P. °C Solvent	Yield %	Molecular-Formula (Mol. Wt)	Analysis Calcd/Found		
				C	H	N
8a	179-180	71	C ₁₅ H ₁₃ N ₃ O	71.7	5.22	16.7
	Me-OH		(251.3)	71.6	5.12	16.6
8b	186	79	C ₁₆ H ₁₅ N ₃ O	72.4	5.70	15.8
	Me-OH		(265.3)	72.3	5.63	15.6
8c	201-202	62	C ₉ H ₁₀ N ₄ O	56.8	5.30	29.5
	Dioxane		(190.2)	56.6	5.25	29.3
8d	162	75	C ₁₅ H ₁₄ N ₄ O	67.7	5.30	21.0
	Me-OH		(266.3)	67.5	5.14	21.0
9	>280	55	C ₉ H ₇ N ₅ O	53.7	3.51	34.8
	D.M.F		(201.2)	53.6	3.45	34.7
10a	125	83	C ₉ H ₈ N ₂ OS	56.2	4.20	14.6
	Et-OH		(192.3)	56.1	4.10	14.4
10b	158-160	52	C ₄ H ₃ N ₂ OS	37.	2.38	22.0
	Et-OH		(127.2)	37.6	2.21	21.9
11a	139	43	C ₁₅ H ₁₉ N ₃ OS	62.3	6.62	14.5
	Me-OH		(289.4)	62.1	6.53	14.3
11b	150-151	60	C ₁₄ H ₁₇ N ₃ O ₂ S	57.7	5.88	14.4
	Me-OH		(291.4)	57.5	5.71	14.2
12a	112	47	C ₁₁ H ₁₀ N ₂ O ₃ S	52.8	4.03	11.9
	dil					
12b	Me-OH	42	(250.3)	52.5	3.8	11.7
	144		C ₆ H ₅ N ₃ O ₃ S	36.2	2.53	21.1
13a	dil	63				
	Et-OH		(199.2)	36.1	2.4	20.8
13b	98	58	C ₁₃ H ₁₄ N ₂ O ₃ S	56.1	5.07	10.1
	Benzene		(278.3)	55.9	4.8	10.1
14a	110	63	C ₈ H ₉ N ₃ O ₃ S	42.3	3.99	18.
	Benzene		(227.3)	42.1	3.7	18.2
14b	180-181	72	C ₁₇ H ₁₅ N ₃ O ₂ S	62.8	4.65	12.9
	Et-OH		(325.4)	62.4	4.5	12.8
14c	200-202	57	C ₁₈ H ₁₇ N ₃ O ₂ S	63.7	5.05	12.4
	Et-OH		(339.4)	63.6	4.9	12.2
14d	223	65	C ₁₁ H ₁₂ N ₄ O ₂ S	50.0	4.58	21.2
	Et-OH		(264.3)	49.7	4.4	21.1
14d	193	65	C ₁₇ H ₁₆ N ₄ O ₂ S	60.0	4.74	16.5
	Benzene		(340.4)	59.8	4.6	16.2

Mannich bases 11a, b

Formaldehyd (10 ml, 40%) was added to **10a** (0.01 mol) and the mixture was heated gently until the solution became clear. The appropriate amine (0.01 mol) in ethanol was added and the mixture was stirred at room temp. for 3 h. The mixture was evaporated gently to dryness and the residue was recrystallized from the proper solvent. The IR-spectrum of **11a** shows bands at 1640 (C=N), 1250-1270 (C=S), and 1210 (C-O-C).

2-(Benzyl or cyanomethyl)-5-carboxymethylthio-1,3,4-oxadiazoles 12a, b

To a solution of **10a** or **b** (0.01 mol) in aqueous NaOH (30 ml, 8%), chloroacetic acid (0.94 g, 0.01 mol) was added portionwise at room temp. with vigorous shaking. The mixture was then stirred for 4 h. The clear solution was acidified with dil. HCl and the separated product was washed with water and recrystallized from the suitable solvent (Table 1). The IR-spectrum of **12b** shows bands 3400-2600 (carboxylic-OH), 2230 (C≡N), 1710 (C=O), and 1630 (C=N). - The ¹H-NMR-spectrum of the same

compound displays signals at 11.3 (s, 1H; -COOH), 3.8 (s, 2H; -S-CH₂-COOH), and 2.4 (s, 2H; -CH₂CN).

2-(Benzyl or cyanomethyl)-5-carbethoxymethylthio-1,3,4-oxadiazoles 13a, b

A solution of **10a** or **b** (0.01 mol) in 50 ml alcoholic NaOH was treated with ethyl bromoacetate (0.015 mol). The mixture was then heated on a water bath for 1 h. The separated solid was recrystallized from the suitable solvent (Table 1). The IR-spectrum of **13a** shows bands at 1730 (C=O), 1640 (C=N), and 1300 (C-O-C).

Amides, hydrazide and phenylhydrazide derivatives 14a-d

A mixture of the ester **13a** (0.01 mol) and the appropriate amine or hydrazine (0.01 mol) in 50 ml n-butanol was refluxed for 6 h. The separated solid after cooling was crystallized from the proper solvent (Table 1). The IR-spectrum of **14a** shows bands at 3400 (N-H), 1670 (C=O), and 1620 cm⁻¹ (C=N). Also the IR-spectrum of **14c** shows absorptions at 3350 (N-H), 1660 (C=O), and 1630 cm⁻¹ (C=N).

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