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SYNTHESIS OF NOVEL BENZOCHROMENE, BENZOQUINOLINE, BENZOCHROMENOPYRIMIDINE AND PYRIMIDOBENZOQUINOLINE DERIVATIVES

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SYNTHESIS OF NOVEL BENZOCHROMENE, BENZOQUINOLINE, BENZOCHROMENOPYRIMIDINE AND PYRIMIDOBENZOQUINOLINE DERIVATIVES

M. N. Jachak*, D. B. Kendre, A. B. Avhale, R. B. Toche and V. J. Medhane

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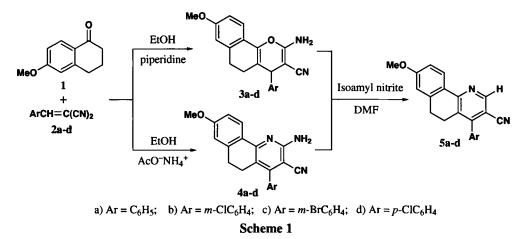
Benzochromene, benzoquinoline and their derivatives have a wide range of biological activities such as antimicrobial,^{1,2} antiasthmic,³ antimalarial,⁴ antiseptic,⁵ hypnotic, CNS and sedative activities.^{6,7} These observations made it of interest to synthesize new tricyclic heterocycles which may have potential medicinal applications. The search for new heterocyclic compounds and for novel methods of their synthesis are major areas in contemporary organic chemistry.^{8,9}

Otto and co-workers⁵ had reported the synthesis of benzochromene and benzoquinoline derivatives by the Michael addition of malononitrile to 2-arylidene-1-tetralone. As part of our ongoing program in this area,^{10,11} we now report the preparation of new tricyclic and tetracyclic heterocycles such as benzochromene, benzoquinoline, benzochromenopyrimidine and pyrimi-dobenzoquinoline derivatives from 6-methoxy-1-tetralone (1) by the Michael addition of benzylidenemalononitrile 2 to 6-methoxy-1-tetralone (1) in better yields and less time than by Otto's protocol.

Addition of 6-methoxy-1-tetralone (1) to benzylidenemalononitriles 2 in the presence of catalytic amount of piperidine in ethanol at reflux temperature yielded 2-amino-4-aryl-8-methoxy-5,6-dihydro-4*H*-benzo[*h*]chromene-3-carbonitriles 3 (*Scheme 1*) which were characterized by IR, PMR, ¹³C NMR and elemental analysis. For example, **3a** showed absorptions at 3465, 3311, 2191 cm⁻¹ in the IR spectrum which indicated the presence of NH₂ and CN groups respectively. The singlet at δ 4.06 and broad singlet at δ 4.53 confirmed the presence of C₄H and protons of NH₂ group respectively in ¹H NMR spectrum. The proposed structure of **3a** was further supported by ¹³C NMR in CDCl₃ which exhibited peaks at δ 24, 28, 41, 55, 115 for C₅, C₆, C₄, OCH₃ and carbon of nitrile respectively and aromatic carbons were observed between δ 121-161. However, addition of compound 1 to benzylidenemalononitriles 2 in presence of

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ammonium acetate and catalytic amount of acetic acid in ethanol at reflux temperature yielded 2-amino-4-aryl-8-methoxy-5,6-dihydrobenzo[*h*]quinoline-3-carbonitriles **4** in good yield (*Scheme 1*). Structures of compounds **4** were assigned on the basis of spectroscopic and analytical data. For example, ¹³C NMR in CDCl₃ of compound **4a** showed peaks at δ 24, 28, 56, 117 for C₅, C₆, OCH₃, and carbon of nitrile function respectively and aromatic carbons were observed between δ 126-159. Compounds **3** were used for the synthesis of benzochromenopyrimidines **7** as well as benzoquinolines **5** and **10**. Analogously, compounds **4** were used for the synthesis of benzo-quinolines **5** and pyrimidobenzoquinolines **9** (*Schemes 1-3*).

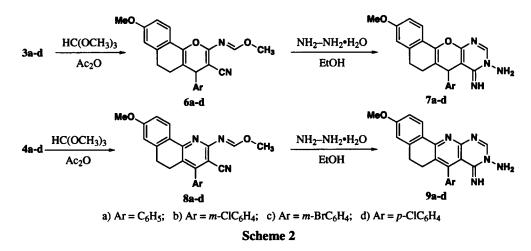


The deamination of compounds 4 with isoamyl nitrite¹² in *DMF* yielded the benzoquinoline derivatives 5 (*Scheme 1*). The absence of NH₂ absorption band in the IR spectrum of **5a** and appearance of singlet at δ 7.83 for C₂H proton in ¹H NMR clearly confirmed its structure which was also supported by the ¹³C NMR in CDCl₃, which showed peaks at δ 24, 28, 55, 119 for C₅, C₆, OCH₃ and carbon of the nitrile respectively and aromatic carbons were observed between δ 128-161. Similar reaction with compounds 3 unexpectedly yielded benzoquinolines 5 rather than the benzochromene. This may be due to opening of the pyran ring and cyclization to give the stable, fully aromatic benzoquinoline. Compounds 5 were characterized by IR, PMR, ¹³C NMR and elemental analysis. Melting points, spectroscopic and analytical data of compounds 5 obtained either from 3 or 4 were identical.

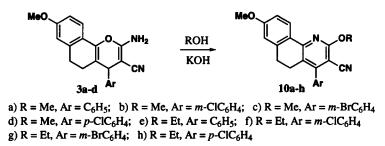
Treatment of benzochromenes **3** with trimethyl orthoformate in acetic anhydride at reflux yielded the benzo[*h*]chromen-2-ylimidoformates **6** which on cyclization with hydrazine hydrate in ethanol, led to benzochromenopyrimidines **7** in good yields. Compounds **6** and **7** were characterized by spectroscopic and analytical data. The NH₂ and NH stretching bands were observed at 3299, 3276, 3138 cm⁻¹ in IR spectrum of **7a** and broad singlet at δ 5.54, 6.35 for NH₂ and NH protons in ¹H NMR spectrum respectively. This structural assignment was further supported by ¹³C NMR in CDCl₃ which showed peaks at δ 24, 27, 42, 55 for C₆, C₇, C₅, OCH₃ respectively and aromatic carbons were observed between δ 127-159. A similar sequence of

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reactions was performed on the benzoquinoline derivatives 4, which on treatment with trimethyl orthoformate in acetic anhydride at reflux temperature, yielded benzo[h]quinolin-2-ylimidoformates 8; further treatment of 8 with hydrazine hydrate in ethanol at reflux temperature furnished pyrimidobenzoquinoline derivatives 9 (Scheme 2). Compounds 8 and 9 were characterized by IR, ¹H NMR and analytical data. The ¹³C NMR in CDCl₃ of 9a exhibit peaks at δ 24, 27, 53, for C₆, C₇, OCH₃ respectively and aromatic carbons between δ 128-160 also supported the proposed structure 9a.

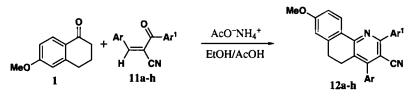


The benzochromenes 3 furnished the benzoquinoline derivatives 10a-d upon reflux with methanolic potassium hydroxide. The corresponding ethyl ethers 10e-h were obtained by treatment with ethanolic potassium hydroxide (*Scheme 3*). Formation of compounds 10 could arise from the attack of alkoxide ion on vinyl carbon, facilitated by the electron-withdrawing *o*-nitrile group and subsequent opening of the pyran ring followed by S_N^i attack of NH₂ to lead to the benzoquinoline derivatives.



The presence of C_2 -OCH₃ in **10a** is indicated by a singlet at δ 4.19 in ¹H NMR spectrum and supported by peak at δ 54 in ¹³C NMR spectrum. The presence of C_2 -OCH₂CH₃ group in **10e** was confirmed by a triplet at δ 1.53 and quartet at δ 4.53 in ¹H NMR spectrum and peaks at δ 14, 62 and 126-162 for CH₃, OCH₂ and aromatic carbons respectively in ¹³C NMR spectrum.

The condensation of 2-aroyl-3-arylacrylonitrile¹³ 11 with 6-methoxy-1-tetralone (1) in presence of ammonium acetate and acetic acid in ethanol at reflux temperature furnished the benzoquinoline derivatives 12 which were characterized by spectroscopic and analytical data.



a) Ar, $Ar_1 = C_6H_5$; b) Ar = *m*-BrC₆H₄, $Ar_1 = C_6H_5$; c) Ar = *p*-ClC₆H₄, $Ar_1 = C_6H_5$ d) Ar = C₆H₅, Ar₁ = *p*-BrC₆H₄; e) Ar = *m*-ClC₆H₄, Ar₁ = *p*-BrC₆H₄; f) Ar = *p*-ClC₆H₄, Ar₁ = *p*-BrC₆H₄; g) Ar = C₆H₅, Ar₁ = *p*-ClC₆H₄; h) Ar = *m*-BrC₆H₄, Ar₁ = *p*-ClC₆H₄

Scheme 3

The ¹³C NMR in CDCl₃ of **12a** exhibited peaks at δ 25, 27, 55,117 and 126-161 for C₅, C₆, OCH₃, CN and aromatic carbons respectively. The analytical and spectroscopic data for all above compounds are summarized in *Tables 1* and 2.

Cmpd.	Yield	mp.	Elem	ental Analysis. (Fo	und)
	(%)	(°Ĉ)	С	Н	N
3a	80	214-215	76.41 (76.22)	5.47 (5.25)	8.47 (8.30)
3b	82	220-222	69.13 (68.90)	4.69 (4.58)	7.67 (7.49)
3c	75	225-226	61.62 (61.40)	4.18 (4.00)	6.84 (6.90)
3d	72	190-192	69.13 (69.00)	4.69 (4.47)	7.67 (7.50)
4 a	85	218-219	77.04 (76.90)	5.23 (5.16)	12.83 (12.60)
4b	90	211-213	69.70 (69.50)	4.45 (4.33)	11.61 (11.45)
4c	86	228-229	62.08 (61.90)	3.96 (3.74)	10.34 (10.20)
4d	88	207-208	69.70 (69.48)	4.45 (4.26)	11.61 (11.50)
5a	76	185-186	80.74 (80.55)	5.16 (5.05)	8.96 (8.76)
5b	70	195-196	72.72 (72.50)	4.35 (4.25)	8.07 (7.90)
5c	68	205-206	64.46 (64.23)	3.86 (3.74)	7.16 (6.97)
5d	73	190-192	72.72 (72.40)	4.35 (4.20)	8.07 (8.10)
6a	77	197-198	74.17 (74.00)	5.41 (5.21)	7.52 (7.43)
6b	60	207-209	67.89 (67.66)	4.70 (4.53)	6.88 (6.69)
6с	81	217-218	61.21 (61.01)	4.24 (4.12)	6.20 (6.00)
6d	71	181-183	67.89 (67.56)	4.70 (4.48)	6.88 (6.70)
7a	66	230-232	70.95 (70.79)	5.41 (5.21)	15.04 (15.00)
7b	69	225-227	64.94 (64.75)	4.70 (4.51)	13.77 (13.60)

Table 1. Yield, mps and Elemental Analysis of Compounds 3-12

Cmpd.	Yield	mp.	Eleme	ental Analysis. (Fo	und)
	(%)	(°C)	С	H	N
7c	78	235-237	58.54 (58.32)	4.24 (4.12)	12.41 (12.30)
7d	85	240-242	64.94 (64.74)	4.70 (4.49)	13.77 (13.55)
8a	77	205-206	74.78 (74.56)	5.18 (5.16)	11.37 (11.20)
8b	74	203-205	68.38 (68.26)	4.49 (4.46)	10.40 (10.23)
8c	80	219-220	61.62 (61.41)	4.04 (4.01)	9.37 (9.18)
8d	90	192-194	68.38 (68.16)	4.49 (4.26)	10.40 (10.27)
9a	79	230-232	71.52 (71.30)	5.18 (5.05)	18.98 (18.76)
9b	70	232-234	65.42 (65.28)	4.49 (4.37)	17.34 (17.20)
9c	75	240-241	58.94 (58.72)	4.04 (4.00)	15.62 (15.50)
9d	77	235-237	65.42 (65.20)	4.49 (4.27)	17.34 (17.21)
10a	81	185-187	76.95 (76.87)	5.28 (5.17)	8.18 (8.02)
10b	69	177-179	70.12 (69.99)	4.54 (4.46)	7.43 (7.22)
10c	77	187-189	62.72 (62.53)	4.06 (3.94)	6.64 (6.42)
10d	88	191-192	62.72 (62.53)	4.06 (3.94)	7.43 (7.20)
10e	91	167-168	77.56 (77.34)	5.63 (5.41)	7.86 (7.66)
10 f	95	175-177	70.75 (70.53)	4.88 (4.53)	7.16 (7.00)
10g	96	188-186	63.54 (63.33)	4.38 (4.16)	6.43 (6.25)
10h	98	178-180	70.75 (70.50)	4.88 (4.66)	7.16 (7.02)
1 2 a	88	215-216	83.48 (83.37)	5.18 (5.00)	7.21 (7.17)
1 2 b	89	225-227	69.38 (69.27)	5.18 (4.21)	5.99 (5.80)
12c	78	201-203	76.68 (76.47)	5.18 (4.31)	6.62 (6.60)
12d	79	185-186	69.38 (69177)	4.00 (3.90)	5.99 (5.84)
12e	82	190-191	64.62 (64.50)	3.61 (3.38)	5.58 (5.51)
1 2f	86	198-200	64.62 (64.40)	3.61 (3.41)	5.58 (5.44)
12g	90	180-181	76.68 (76.47)	4.52 (4.30)	6.62 (6.60)
12h	94	191-19 2	64.62 (64.40)	3.61 (3.40)	5.58 (5.61)

Table 1. Continued...

a) All compounds are colorless solids

Ta	ble 2.	Spectr	oscopy	Data	of Con	npounds	3-12

Cmpd.	IR (cm ⁻¹)	¹ Η NMR (δ)
3a	3465, 3311, 2376, 2191, 1136,	2.02 (m, 2H, CH ₂), 2.66 (m, 2H, CH ₂), 3.81 (s, 3H, OCH ₃), 4.06 (s, 1H, C ₄ H), 4.53 (s, 2H, NH ₂), 6.68 (d, $J = 2.8$ Hz, C ₁₀ H), 6.74 (dd, $J = 8.3$, 2.8 Hz, C ₉ H), 7.28 (m, 5H, Ar-H), 7.37 (d, $J = 8.3$ Hz, C ₇ H)

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Cmpd.	$IR (cm^{-1})$	¹ Η NMR (δ)
3b	3465, 3211, 2370, 2231, 1253,	1.98 (m, 2H, CH ₂), 2.65 (m, 2H, CH ₂), 3.81 (s, 3H, OCH ₃), 4.05 (s, 1H, C ₄ H), 4.53 (s, 2H, NH ₂), 6.69 (d, $J = 2.8$ Hz, C ₁₀ H), 6.75 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.18 (m, 4H, Ar-H), 7.37 (d, $J = 8.4$ Hz, C ₇ H)
3с	3365, 3211, 2241,1510, 1130,	2.01 (m, 2H, CH ₂), 2.70 (m, 2H, CH ₂), 3.84 (s, 3H, OCH ₃), 4.64 (s, 1H, C ₄ H), 4.53 (s, 2H, NH ₂), 6.71 (d, $J = 2.7$ Hz, C ₁₀ H), 6.72 (dd, $J = 8.3$, 2.7 Hz, C ₉ H), 7.19 (m, 4H, Ar-H), 7.40 (d, $J = 8.3$ Hz, C ₇ H)
3d	3398, 3223, 2237,1552, 1230,	2.03 (m, 2H, CH ₂), 2.77 (m, 2H,CH ₂), 3.86 (s, 3H, OCH ₃), 4.12 (s, 1H,C ₄ H), 4.62 (s, 2H, NH ₂), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 6.89 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.25 (m, 4H, Ar-H), 8.23 (d, $J = 8.4$ Hz, C ₇ H)
4 a	3455, 3321, 2213, 1542, 1163,	2.59 (m, 2H, CH ₂), 2.74 (m, 2H, CH ₂), 3.85 (s, 3H, OCH ₃), 5.40 (s, 2H, NH ₂), 6.71 (d, $J = 2.8$ Hz, C ₁₀ H), 6.89 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.29 (m, 5H, Ar-H), 8.24 (d, $J = 8.4$ Hz, C ₇ H)
4b	3265, 3315, 2244, 1534, 1133,	2.59 (m, 2H, CH ₂), 2.74 (m, 2H,CH ₂), 3.86 (s, 3H, OCH ₃), 5.15 (s, 2H, NH ₂), 6.71 (d, $J = 2.8$ Hz, C ₁₀ H), 6.86 (dd, $J = 8.4$, 2.7 Hz, C ₉ H), 7.30 (m, 4H, Ar-H), 8.19 (d, $J = 8.4$ Hz, C ₇ H)
4c	3345, 3316, 2214, 1512, 1153,	2.59 (m, 2H, CH ₂), 2.74 (m, 2H,CH ₂), 3.85 (s, 3H, OCH ₃), 5.15 (s, 2H, NH ₂), 6.72 (d, $J = 2.8$ Hz, C ₁₀ H), 6.85 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.28 (m, 4H, Ar-H), 8.19 (d, $J = 8.4$ Hz, C ₇ H)
4d	3265, 3315, 2244, 1591, 1128,	2.59 (m, 2H, CH ₂), 2.74 (m, 2H, CH ₂), 3.85 (s, 3H, OCH ₃), 5.15 (s, 2H, NH ₂), 6.72 (d, $J = 2.8$ Hz, C ₁₀ H), 6.85 (dd, $J = 8.4$, 2.8 Hz, C ₃ H), 7.28 (m, 4H, Ar-H), 8.19 (d, $J = 8.4$ Hz, C ₇ H)
5a	2929, 2233, 1552, 1163,	2.90 (m, 2H, CH ₂), 3.10 (m, 2H,CH ₂), 3.88 (s, 3H, OCH ₃), 6.70 (d, $J = 2.8$ Hz C ₁₀ H), 6.86 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.33 (m, 5H, Ar-H), 7.83 (s, 1H, C ₂ H) 8.24 (d, $J = 8.4$ Hz, C ₇ H)
5b	2925, 2228, 1546, 1145	2.91 (m, 2H, CH ₂), 3.02 (m, 2H,CH ₂), 3.84 (s, 3H, OCH ₃), 6.70 (d, $J = 2.7$ Hz, C ₁₀ H), 6.86 (dd, $J = 8.4$, 2.7 Hz, C ₉ H), 7.23 (m, 4H, Ar-H), 7.81 (s, 1H, C ₂ H) 8.10 (d, $J = 8.4$ Hz, C ₇ H)
5c	2976, 2248, 1536, 1120	2.88 (m, 2H, CH ₂), 2.90 (m, 2H,CH ₂), 3.82 (s, 3H, OCH ₃), 6.71 (d, $J = 2.8$ Hz, C ₁₀ H), 6.83 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.26 (m, 4H, Ar-H), 7.80 (s, 1H, C ₂ H) 8.10 (d, $J = 8.4$ Hz, C ₇ H)
5d	2915, 2228, 1526, 1256	2.91 (m, 2H, CH ₂), 3.05 (m, 2H,CH ₂), 3.81 (s, 3H, OCH ₃), 6.70 (d, $J = 2.8$ Hz, C ₁₀ H), 6.83 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.28 (m, 4H, Ar-H), 7.81 (s, 1H, C ₂ H) 8.10 (d, $J = 8.4$ Hz, C ₇ H)

Table 2	Continued
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Cmpd.	IR (cm ⁻¹)	¹ Η NMR (δ)
68	2325, 2214, 1569, 1160	1.73 (m, 2H, CH ₂), 2.54 (m, 2H, CH ₂), 3.70 (s, 3H, OCH ₃), 3.83 (s, 3H, OCH ₃) 4.27 (s, 1H, C ₄ H), 6.72 (s,1H, Ar-H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.25 (m, 5H, Ar-H), 7.47 (d, $J = 2.8$ Hz, C ₉ H), 8.69 (s, 1H, N = CH)
6b	2356, 2230, 1453, 1520, 1166	1.73 (m, 2H, CH ₂), 2.54 (m, 2H, CH ₂), 3.70 (s, 3H, OCH ₃), 3.83 (s, 3H, OCH ₃) 4.33 (s, 1H, C ₄ H), 6.71 (s,1H, Ar-H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.29 (m, 4H, Ar-H), 7.47 (d, $J = 2.7$ Hz, C ₉ H), 8.69 (s, 1H, N = CH)
6с	2371, 2232, 1512, 1157	1.73 (m, 2H, CH ₂), 2.55 (m, 2H, CH ₂), 3.71 (s, 3H, OCH ₃), 3.84 (s, 3H, OCH ₃) 4.32 (s, 1H, C ₄ H), 6.72 (s, 1H, Ar-H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.28 (m, 4H, Ar-H), 7.47 (d, $J = 2.8$ Hz, C ₉ H), 8.69 (s, 1H, N = CH)
6d	2344, 2230, 1596, 1145	1.73 (m, 2H, CH ₂), 2.54 (m, 2H,CH ₂), 3.70 (s, 3H, OCH ₃), 3.84 (s, 3H,OCH ₃), 6.70 (s, 1H, Ar-H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.29 (m, 4H, Ar-H), 7.47 (d, $J = 2.7$ Hz, C ₉ H), 8.69 (s, 1H, N = CH)
7a	3299, 3276, 3138, 1554, 1132	1.81 (m, 2H, CH ₂), 2.63 (m, 2H, CH ₂), 3.67 (s, 3H, OCH ₃), 4.36 (s, H, C ₅ H), 5.54 (s, 2H, NH ₂), 6.35 (bs, 1H, NH), 6.68 (d, $J = 2.8$ Hz, C ₁₁ H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.11 (m, 6H, Ar-H), 7.97 (s, 1H, C ₂ H)
7b	3298, 3270, 3136, 1544, 1132	1.81 (m, 2H, CH ₂), 2.61 (m, 2H, CH ₂), 3.63 (s, 3H, OCH ₃), 4.37 (s, H, C ₅ H), 5.52 (s, 2H, NH ₂), 6.32 (bs, 1H, NH), 6.68 (d, $J = 2.8$ Hz, C ₁₁ H), 6.72 (d, $J = 2.7$ Hz, C ₁₀ H), 7.27 (m, 5H, Ar-H), 7.95 (s, 1H, C ₂ H)
7c	3299, 3276, 3144, 1542	1.81 (m, 2H, CH ₂), 2.61 (m, 2H,CH ₂), 3.63 (s, 3H, OCH ₃), 4.37 (s, H, C ₅ H), 5.52 (s, 2H, NH ₂), 6.32 (bs, 1H, NH), 6.68 (d, $J = 2.8$ Hz, C ₁₁ H), 6.72 (d, $J = 2.8$ Hz, C ₁₀ H), 7.27 (m, 5H, Ar-H), 7.95 (s, 1H, C ₂ H)
7d	3299, 3276, 3138, 1554	1.81 (m, 2H, CH ₂), 2.62 (m, 2H, CH ₂), 3.64 (s, 3H, OCH ₃), 4.39 (s, H, C ₅ H), 5.53 (s, 2H, NH ₂), 6.33 (bs, 1H, NH), 6.68 (d, $J = 2.8$ Hz, C ₁₁ H), 6.72 (d, $J = 2.8$ Hz, C ₁₀ H), 7.26 (m, 5H, Ar-H), 7.95 (s, 1H, C ₅ H)
8a	2344, 2224, 1520, 1176	1.72 (m, 2H, CH ₂), 2.52 (m, 2H,CH ₂), 3.71 (s, 3H, OCH ₃), 3.83 (s, 3H,OCH ₃) 6.72 (s, 1H, Ar-H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.25 (m, 5H, Ar-H), 7.47 (d, $J = 2.8$ Hz, C ₉ H), 8.69 (s, 1H, N = CH)
8b	2412, 2231, 1596, 1144	1.70 (m, 2H, CH ₂), 2.50 (m, 2H,CH ₂), 3.74 (s, 3H, OCH ₃), 3.83 (s, 3H, OCH ₃) 6.72 (s, 1H, Ar-H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.22 (m, 4H, Ar-H), 7.47 (d, $J = 2.8$ Hz, C ₉ H), 8.66 (s, 1H, N = CH)
8c	2355, 2237, 1605, 1145	1.74 (m, 2H, CH ₂), 2.48 (m, 2H,CH ₂), 3.69 (s, 3H, OCH ₃), 3.81 (s, 3H, OCH ₃) 6.71 (s, 1H, Ar-H), 6.76 (d, $J = 2.8$ Hz, C ₁₀ H), 7.22 (m, 4H, Ar-H), 7.47 (d, $J = 2.8$ Hz, C ₉ H), 8.67 (s, 1H, N = CH)

Table 2.	Continued
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JACHAK, KENDRE, AVHALE, TOCHE AND MEDHANE

Cmpd.	IR (cm ⁻¹)	¹ Η NMR (δ)
8d	2376, 2241, 1526, 1114	1.70 (m, 2H, CH ₂), 2.50 (m, 2H, CH ₂), 3.74 (s, 3H, OCH ₃), 3.83 (s, 3H, OCH ₃), 6.72 (s, 1H, Ar-H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.22 (m, 4H, Ar-H), 7.47 (d, $J = 2.7$ Hz, C ₉ H), 8.66 (s, 1H, N = CH)
9a	3288, 3276, 3126, 1526, 1132	1.81 (m, 2H, CH ₂), 2.63 (m, 2H,CH ₂), 3.67 (s, 3H, OCH ₃), 5.54 (s, 2H, NH ₂) 6.35 (bs, 1H, NH), 6.68 (d, $J = 2.8$ Hz, C ₁₁ H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.11 (m, 6H, Ar- H), 7.97 (s, 1H, C ₂ H)
9b	3278, 3264, 3144, 1572	1.81 (m, 2H, CH ₂), 2.61 (m, 2H, CH ₂), 3.65 (s, 3H, OCH ₃), 5.54 (s, 2H, NH ₂), 6.35 (bs, 1H, NH), 6.67 (d, $J = 2.7$ Hz, C ₁₁ H), 6.74 (d, $J = 2.7$ Hz, C ₁₀ H), 7.11 (m, 5H, Ar- H), 7.97 (s, 1H, C ₂ H)
9c	3268, 3254, 3124, 1544	1.79 (m, 2H, CH ₂), 2.58 (m, 2H,CH ₂), 3.69 (s, 3H, OCH ₃), 5.51 (s, 2H, NH ₂), 6.35 (bs, 1H, NH), 6.67 (d, $J = 2.8$ Hz, C ₁₁ H), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 7.11 (m, 5H, Ar- H), 7.97 (s, 1H, C ₂ H)
9d	3278, 3264, 3138, 1522	1.84 (m, 2H, CH ₂), 2.64 (m, 2H,CH ₂), 3.65 (s, 3H, OCH ₃), 5.52 (s, 2H, NH ₂) 6.33 (bs, 1H, NH), 6.73 (d, $J = 2.8$ Hz, C ₁₁ H), 6.74 (d, $J = 2.7$ Hz, C ₁₀ H), 7.11 (m, 5H, Ar- H), 7.97 (s, 1H, C ₂ H)
10a	2929, 2860, 2221, 1555, 844	2.67 (m, 2H, CH ₂), 2.80 (m, 2H, CH ₂), 3.90 (s, 3H, OCH ₃), 419 (s, 3H, OCH ₃), 6.67 (d, $J = 2.8$ Hz, C ₁₀ H), 6.93 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.29 (m, 5H, Ar-H), 8.30 (d, $J = 8.3$ Hz, C ₇ H)
10Ь	2929, 2858, 2214, 1550,1157, 862	2.63 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.90 (s, 3H, OCH ₃), 420 (s, 3H, OCH ₃), 6.77 (d, $J = 2.7$ Hz, C ₁₀ H), 6.92 (dd, $J = 8.4$, 2.7 Hz, C ₉ H), 7.24 (m, 4H, Ar-H), 8.26 (d, $J = 8.3$ Hz, C ₇ H)
10c	2923, 2858, 2214, 1550, 1157, 862	2.63 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.90 (s, 3H, OCH ₃), 420 (s, 3H, OCH ₃), 6.77 (d, $J = 2.8$ Hz, C ₁₀ H), 6.92 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.24 (m, 4H, Ar-H), 8.26 (d, $J = 8.4$ Hz, C ₇ H)
10d	2925, 2860, 2221, 1556, 1155, 844	2.66 (m, 2H, CH ₂), 2.81 (m, 2H,CH ₂), 3.90 (s, 3H, OCH ₃), 419 (s, 3H, OCH ₃), 6.77 (d, $J = 2.7$ Hz, C ₁₀ H), 6.93 (dd, $J = 8.4$, 2.7 Hz, C ₉ H), 7.21 (m, 4H, Ar-H), 8.30 (d, $J = 8.4$ Hz, C ₇ H)
10e	2221, 1600, 1552, 1157, 908, 846	1.53 (t, 3H, CH ₃), 2.63 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.90 (s, 3H, OCH ₃), 4.63 (q, 2H, OCH ₂), 6.77 (d, $J = 2.8$ Hz, C ₁₀ H), 6.93 (dd, $J = 8.4$, 2.8 Hz, C ₃ H), 7.24 (m, 5H, Ar-H), 8.26 (d, $J = 8.4$ Hz, C ₃ H)
10f	2923, 2858, 2214, 1550, 1157, 862	1.51 (t, 3H, CH ₃), 2.63 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.90 (s, 3H, OCH ₃), 4.63 (q, 2H, OCH ₂), 6.77 (d, $J = 2.8$ Hz, C ₁₀ H), 6.92 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.24 (m, 4H,Ar-H), 8.26 (d, $J = 8.4$ Hz, C ₇ H)

Table 2.	Continued
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Cmpd.	IR (cm ⁻¹)	¹ Η NMR (δ)
10g	2922, 2858, 2216, 1550, 1157, 862	1.51 (t, 3H, CH ₃), 2.62 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.91 (s, 3H, OCH ₃), 4.61 (q, 2H, OCH ₂), 6.66 (d, $J = 2.7$ Hz, C ₁₀ H), 6.91 (dd, $J = 8.4$, 2.7 Hz, C ₉ H), 7.22 (m, 4H,Ar-H), 8.27 (d, $J = 8.4$ Hz, C ₇ H)
10h	2218, 1602, 1427, 1161, 1028, 877	1.51 (t, 3H, CH ₃), 2.64 (m, 2H, CH ₂), 2.80 (m, 2H, CH ₂), 3.90 (s, 3H, OCH ₃), 4.63 (q, 2H, OCH ₂), 6.67 (d, $J = 2.8$ Hz, C ₁₀ H), 6.92 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.28 (m, 4H,Ar-H), 8.26 (d, $J = 8.4$ Hz, C ₇ H)
1 2 a	2929, 2215, 1542, 1153, 950	2.73 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.88 (s, 3H, OCH ₃), 6.75 (d, $J = 2.8$ Hz, C ₁₀ H), 6.90 (dd, $J = 8.4$ 2.7 Hz, C ₉ H), 7.24 (m, 10H,Ar-H), 8.39 (d, $J = 8.4$ Hz, C ₇ H)
12b	2929, 2219, 1542, 1163, 950	2.73 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.88 (s, 3H, OCH ₃), 6.75 (d, $J = 2.8$ Hz, C ₁₀ H), 6.90 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.24 (m, 9H,Ar-H), 8.39 (d, $J = 8.4$ Hz, C ₇ H)
12c	2939, 2219, 1532, 1163, 950	2.73 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.88 (s, 3H, OCH ₃), 6.75 (d, $J = 2.8$ Hz, C ₁₀ H), 6.90 (dd, $J = 8.3$, 2.8 Hz, C ₉ H), 7.24 (m, 9H, Ar-H), 8.39 (d, $J = 8.3$ Hz, C ₇ H)
12d	2939, 2219, 1542, 1163, 950	2.73 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.88 (s, 3H, OCH ₃), 6.75 (d, $J = 2.7$ Hz, C ₁₀ H), 6.90 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.24 (m, 9H, Ar-H), 8.39 (d, $J = 8.4$ Hz, C ₇ H)
12e	2928, 2221, 1542, 1163, 950	2.73 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.88 (s, 3H, OCH ₃), 6.75 (d, $J = 2.8$ Hz, C ₁₀ H), 6.90 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.24 (m, 9H, Ar-H), 8.39 (d, $J = 8.4$ Hz, C ₇ H)
12f	2928, 2221, 1542, 1163, 950	2.76 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.87 (s, 3H, OCH ₃), 6.74 (d, $J = 2.8$ Hz, C ₁₀ H), 6.90 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.25 (m, 9H, Ar-H), 8.39 (d, $J = 8.4$ Hz, C ₇ H)
12g	2929, 2219, 1542, 1163, 950	2.73 (m, 2H, CH ₂), 2.81 (m, 2H, CH ₂), 3.88 (s, 3H, OCH ₃), 6.75 (d, $J = 2.8$ Hz, C ₁₀ H), 6.90 (dd, $J = 8.4$, 2.8 Hz, C ₉ H), 7.24 (m, 9H, Ar-H), 8.39 (d, $J = 8.4$ Hz, C ₇ H)
12h	2917, 2235, 1532, 1163, 950	2.73 (m, 2H, CH ₂), 2.81 (m, 2H,CH ₂), 3.85 (s, 3H, OCH ₃), 6.73 (d, $J = 2.8$ Hz, C ₁₀ H), 6.91 (dd, $J = 8.4$, 2.7 Hz, C ₉ H), 7.21 (m, 9H,Ar-H), 8.35 (d, $J = 8.4$ Hz, C ₇ H)

Table 2. Continued...

a) All compounds are colorless solids

EXPERIMENTAL SECTION

Melting points were determined on a Gallenkamp Melting point apparatus and are uncorrected. Elemental analysis was determined on a Hosli C, H Analyzer. ¹H NMR and ¹³C NMR spectra

were recorded on a Varian XL-300 MHz spectrometer using TMS as an internal standard. IR spectra were obtained as Nujol mulls on a Schimadzu IR 408 spectrometer.

2-Amino-8-methoxy-4-aryl-5,6-dihydro-4H-benzo[h]chromene-3-carbonitriles (3).

General Procedure.- A solution of 1 (1.76 g, 0.01 mole) and benzylidenemalononitrile 2 (1.54 g, 0.01 mole) in ethanol (15 mL) in presence of piperidine (0.5 mL) was refluxed for 5 hr. The solid formed on cooling was collected, washed with ethanol, dried and recrystallized from ethanol to yield **3a**. Compounds **3b-d** were prepared similarly and recrystallized from ethanol.

2-Amino-8-methoxy-4-aryl-5,6-dihydrobenzo[h]quinoline-3-carbonitriles (4).

General Procedure.- A solution of 1 (1.76 g, 0.01 mole), benzylidenemalononitrile 2 (1.54 g, 0.01 mole), ammonium acetate (1.54 g, 0.02 mole) and acetic acid (0.5 mL) in absolute ethanol (15 mL) was refluxed for 5 hr. The solid formed on cooling was collected, washed with ethanol, dried and recrystallized from ethanol to yield **4a**. Compounds **4b-d** were prepared similarly and were recrystallized from ethanol.

8-Methoxy-4-aryl-5, 6-dihydrobenzo[h]quinoline-3-carbonitriles (5).

General Procedure.- A solution of 3a (3.30 g, 0.01 mole) or 4a (3.27 g, 0.01 mole) and isoamyl nitrite (1.34 mL, 0.01 mole) in *DMF* (10 mL) was stirred at 90-95°C for 3 hr. The solvent was removed under reduced pressure, the solid obtained was collected and recrystallized from ethanol to yield 5a. Compounds 5b-d were synthesized similarly and were recrystallized from ethanol.

Methyl 3-cyano-8-methoxy-4-aryl-5,6-dihydro-4H-benzo[h]chromen-2-ylimidoformates (6).

General Procedure.- A solution of 3a (3.30 g, 0.01 mole) and trimethyl orthoformate (1.09 mL, 0.01 mole) in acetic anhydride (15 mL) was refluxed for 1 hr. Then the solvent was removed under reduced pressure, the solid obtained was collected, washed with ethanol, dried and recrystallized from ethanol to yield 6a. Compounds 6b-d were obtained similarly and recrystallized from ethanol.

3 (5*H*)-Amine-4-imino-5-aryl-6,7-dihydro-9-methoxy-4*H*-benzochromeno[2,3-*d*]pyrimidines (7).

General Procedure.- A solution of 6a (3.72 g, 0.01 mole) and hydrazine hydrate (0.48 mL, 0.01 mole) in absolute ethanol (15 mL) was refluxed for 1 hr. The solid formed was collected, washed with ethanol, dried and recrystallized from *DMF*-ethanol (1:2) to yield **7a**. Compounds **7b-d** were synthesized similarly and recrystallized from *DMF*-ethanol.

Methyl 3-cyano-8-methoxy-4-aryl-5,6-dihydrobenzo[h]quinolin-2-ylimidoformates (8).

General Procedure.- A solution of 4a (3.27 g, 0.01 mole), trimethyl orthoformate (1.09 mL, 0.01 mole) in acetic anhydride (15 mL) was refluxed for 1 hr. Then the solvent was removed under reduced pressure, the solid obtained was collected, washed with ethanol, dried and recrystallized from ethanol to yield 8a. Compounds 8b-d were synthesized similarly and recrystallized from ethanol.

3 (4H)-Amine-4-imino-5-aryl-6,7-dihydro-9-methoxypyrimido[4,5-b]benzoquinolines (9).

General Procedure.- A solution of 8a (3.69 g, 0.01 mole) and hydrazine hydrate (0.48 mL, 0.01 mole) in absolute ethanol (15 mL) was refluxed for 1 hr. The solid formed was collected, washed with ethanol, dried and recrystallized from *DMF*-ethanol (1:2) to yield 9a. Compounds 9b-d were obtained similarly and recrystallized from *DMF*-ethanol.

2, 8-Dimethoxy-4-aryl-5,6-dihydrobenzo[h]quinoline-3-carbonitriles (10).

General Procedure.- A solution of 3a (3.30 g, 0.01 mole) in dry methanol (15 mL) catalytic amount of potassium hydroxide was added and refluxed for 45 minutes. The solid formed on cooling was collected, washed with methanol, dried and recrystallized from methanol to yield 10a. Compounds 10b-d were synthesized similarly and were recrystallized from methanol. Compounds 10e-h were synthesized similarly by using ethanolic potassium hydroxide and recrystallized from ethanol.

2,4-Diarnyl-8-methoxy-5,6-dihydrobenzo[h]quinoline-3-carbonitriles (12).

General Procedure.- A solution of 1 (1.76 g, 0.01 mole) and 2-benzoyl-3-phenylacrylonitrile 11a (2.33 g, 0.01 mole) in absolute ethanol (15 mL), ammonium acetate (1.54 g, 0.02 mole) and acetic acid (1 mL) were added and refluxed for 3 hr. The solid formed on cooling was collected, washed with ethanol, dried and recrystallized from ethanol to yield 12a. Compounds 12b-h were synthesized similarly and recrystallized from ethanol.

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