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Cyclocarbopalladation of Alkynes: A Stereoselective Method for Preparing Dibenzoxapine Containing Tetrasubstituted Exocyclic Alkenes

Hannah Yu,*,† Rachel N. Richey,† Matthew W. Carson,‡ and Michael J. Coghlan‡

Chemical Product Research and Development, Lilly Research Laboratories, Eli Lilly and Company, Indianapolis, Indiana 46285, and Discovery Chemistry Research, Lilly Research Laboratories, Eli Lilly and Company, Indianapolis, Indiana 46285

yu_hannah@lilly.com

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ABSTRACT

$$R_1 \longrightarrow R_2 \longrightarrow R_2 \longrightarrow R_1 \longrightarrow R_2 \longrightarrow R_2 \longrightarrow R_1 \longrightarrow R_2 \longrightarrow R_2 \longrightarrow R_1 \longrightarrow R_2 \longrightarrow R_2 \longrightarrow R_2 \longrightarrow R_1 \longrightarrow R_2 \longrightarrow R_2$$

A palladium-catalyzed cascade carbometalation-cross coupling of alkyne route was developed for the preparation of tetrasubstituted exocyclic alkenes with high stereo- and regiocontrol. The effectiveness of this novel methodology was demonstrated by the synthesis of a number of dibenzoxapines in sufficient quantities to support their further development.

Dibenzoxapine and dibenzosuberane derivatives have been an interest of the pharmaceutical industry for a number of years. Recently, a novel series of these tricyclic compounds have been identified at Eli Lilly and Company as nuclear hormone receptor modulators.¹ To support their further development, quantities of tetrasubstituted dibenzoxapine derivatives, such as **1a**-**d** (Figure 1), were needed.

The dibenzoxapine derivatives containing tetrasubstituted alkene moieties (1a-d) represent a novel structure and also pose a unique synthetic challenge. Although these tetrasubstituted dibenzoxapine derivatives could be prepared by methods such as the Wittig reaction² or simply dehydration of a carbinol, the synthetic approaches are lengthy and often

limited in scope. In addition, a mixture of the (*E*)- and (*Z*)isomers is often produced. Since each geometric isomer may
exhibit different toxicological and pharmacological activities,
stereoselective syntheses of these targets are advantageous.
Our need to develop a practical stereoselective route led to
an investigation of the palladium-catalyzed tandem cyclization of 3a-d. The synthetic strategy was to incorporate a
key intramolecular carbometalation of alkynes to establish
the desired geometric relationship. The convergent approach
involved preparation of cyclization precursors 3a-d, which
could be derived from benzyl alcohols 4a-c and phenols
5a or 5b through either a one-step Mitsunobu reaction³ or a
two-step alkylation procedure (Figure 1). Benzyl alcohols
4a-c could be easily synthesized from the corresponding
aryl halides by using Sonogashira coupling.⁴

^{*} To whom correspondence should be addressed.

[†] Chemical Product Research and Development.

[‡] Discovery Chemistry Research.

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Figure 1. Retrosynthetic approach to 1a-d.

The carbometalation of alkynes by organometallic reagents is now widely used for the stereospecific synthesis of di-, tri-, and even tetrasubstituted alkenes.⁵ The major advantage of these reactions lies in their high stereo- and regioselectivity, which enables them to add to the triple bond of various alkynes. Another advantage of this approach is the generation of a new organometallic species, a vinylic metal, which can be further transformed in such a way that the overall result is the one-pot positioning of two new substituents in a cis or trans manner, depending on the organometallic reagent used.⁵ In particular, intramolecular carbopalladation of alkynes (cyclocarbopalladation) is a versatile approach to various tetrasubstituted alkenes.⁶ The reaction generally involves an initial cyclocarbopalladation followed by crosscoupling of the vinylpalladium species with boron, tin, or zinc organometallic species. The termination of the vinylpalladium species effectively converts the Pd(II) to Pd(0), thus completing the catalytic cycle. The initial carbopalladation generally follows the 5-exo-dig, 6-exo-dig, or 7-exodig process. However, to the best of our knowledge, there

are still no reports on the preparation of seven-membered rings through a palladium-catalyzed carbometalation-cross coupling cascade with use of organometallics as the terminating trapping species.⁷

The commercially available 3-nitrophenylboronic acid was first used as the terminating species. Reaction between alkyne **3a** and 3-nitrophenylboronic acid was attempted to optimize the reaction conditions (Table 1). Modifying a literature

Table 1. Optimization Studies^a

			temp	%	%	%	%
entry	catalyst	solvent	(° C)	$3\mathbf{a}^b$	6^{b}	7^{b}	$2\mathbf{a}^c$
1^d	Pd(OAc) ₂ /A ^e	$toluene^h$	90	51	4	9	
2	$Pd(OAc)_2/A^e$	toluene	90	84	0	0	
3	$Pd(OAc)_2$	dioxane	110	88	0	0	0
4	Pd(OAc) ₂ /B ^f	dioxane	90	76	2	3	
5^d	$Pd(OAc)_2/A^e$	$n ext{-BuOH}$	90	0	0	25	
6^d	$Pd(OAc)_2$	DMF	110	0	43	4	0
7^d	$Pd(OAc)_2/A^e$	DMF	90	0	1	24	32
8	$Pd(OAc)_2/A^e$	DMF/ H_2O^i	70	0	74	3	
9	$Pd(OAc)_2/A^e$	dioxane/H ₂ O ⁱ	70	0	76	5	
10	Pd(OAc) ₂ /B ^f	dioxane/ H ₂ O ⁱ	70	0	37	2	33
11	Pd(OAc) ₂ /C ^g	dioxane/ H ₂ O ⁱ	110	0	19	12	38
12	$Pd(OAc)_2 \\$	dioxane/ H_2O^i	70	0	3	6	89

^a All reactions were run with 2% catalyst, 1.2 equiv of 3-nitroboronic acid, and 3 equiv of Na₂CO₃ for 12 h with total concentration at 0.1 M. ^b Determined by reverse-phase HPLC. ^c Isolated yields. ^d Additional oligomers and polymers were formed. Some of the oligomers and polymers may not be HPLC detectable. ^e Ligand A: tri-o-tolyphosphine. ^f Ligand B: triphenylphosphine. ^g Ligand C: 2-(dicyclohexylphosphino)biphenyl. ^h 1 equiv of Bu₄NCl was added. ⁱ 4:1 v/v ratio.

procedure^{6f} for the carbopalladation reaction, **3a** and 3-nitrophenylboronic acid were heated in toluene at 90 °C for 12 h with 2 mol % Pd(OAc)₂, 4 mol % tri-*o*-tolylphosphine, 1 equiv of tetrabutylammonium chloride, and 3 equiv of Na₂-CO₃ (Table 1, entry 1). Significant amounts of starting materials **3a** and 3-nitrophenylboronic acid remained unreacted, along with the desired product **2a**, direct Suzuki coupling byproduct **6**,⁸ 3-nitroboronic acid dimerization byproduct **7**,⁹ and some oligomeric and polymeric byproducts which were formed presumably due to the intermolecular coupling between vinylic palladium intermediate and alkyne

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Table 2. Carbometalation Terminated with Different Boronic Acids^a

entry	boronic acid	product	% yield ^b	
1	\longrightarrow B(OH) ₂	8a	77	
2	B(OH) ₂	8b	91	
3	F—B(OH) ₂	8c	76	
4	F_3C $B(OH)_2$	8d	70	
5	$H_3CO \longrightarrow B(OH)_2$	8e	74	
6	CI——B(OH) ₂	8 f	78	
7	$B(OH)_2$ BO_2CH_3	1a	59	

^a All reactions were run with 1.2 equiv of boronic acid, 3.0 equiv of Na₂CO₃, and 1 mol % Pd(OAc)₂ in 4:1 dioxane/water (0.1 M) at 100 °C for 12 h. ^b Isolated yields after column chromatography.

starting material.¹⁰ Increasing the catalyst loading to 10 mol % Pd(OAc)₂ and 20 mol % tri-o-tolylphosphine and raising the temperature to 110 °C did consume all of the starting material 3a; however, a much lower yield of the desired product was observed. As anticipated, elimination of the tetrabutylammonium chloride resulted in a more incomplete reaction (entry 2). Next, we investigated several additional solvents, under both ligand and ligandless conditions (entries 3-7). Although none of these reactions gave a satisfactory result, this initial screening indicated that substrate 3a is more reactive in solvents with higher dielectric constants. With 2 mol % of catalyst, toluene and dioxane led to a significant amount of starting material 3a, indicating that the insertion of palladium to alkyne did not take place under the reaction conditions. n-BuOH and DMF led to a faster palladium insertion process and completed reactions. However, significant amounts of oligomers and polymers were observed. It is likely that under these conditions, the initial palladium insertion was activated, but the subsequent transmetalation between vinylic palladium and boronic acid did not occur at a reasonable rate.

Since the coupling of a vinylic palladium species with arylboronic acids is essentially a variant of the Suzuki crosscoupling reaction, which is known to proceed much faster in water when a hydroxide is present, we examined the possibility of using water as the cosolvent (entries 8 and

Table 3. Synthesis of Dibenzoxepin Derivatives **2a-e** from **3a-e** and **3**-Nitroboronic Acid^a

entry	alkyne	product	yield (%) ^b	
1	3a	NO ₂	89	
2	H ₃ CO 3b	H ₃ CO NO ₂	70	
3	O F	NO ₂	77	
4	F O F S Ad	F———FNO ₂	73	
5°	3e	O_2N	67	

 a All reactions were run with 1.2 equiv of boronic acid, 3.0 equiv of Na₂CO₃, and 1 mol % Pd(OAc)₂ in 4:1 dioxane/water (0.1 M) at 70 °C for 12 h unless otherwise indicated. b Isolated yields after column chromatography. c 4:1 DMF/water (0.1 M) was used as solvent.

9).^{7a,11,12} In fact, a significant enhancement in the reaction rate was observed when the reactions were conducted in DMF/water (4:1 v/v) and dioxane/water (4:1 v/v). Both reactions went to completion at 70 °C in less than 1 h accompanied by only trace amounts of polymerization. However, under these conditions the direct Suzuki coupling product 6 predominated.

In an effort to find conditions where the alkyne insertion would be faster than the direct Suzuki coupling, we continued our investigation by using dioxane/water (4:1) as the solvent with various ligands (entries 9–11). None of the ligands gave the desired **2a/6** ratios, but we were pleased to find that under the ligandless condition (entry 12), desired product **2a** and direct Suzuki coupling product **6** were obtained in an excellent ratio of 29:1. Therefore, under this optimized condition, 2 mol % Pd(OAc)₂, 1.2 equiv of 3-nitroboronic acid, and 3 equiv of Na₂CO₃¹³ at 70 °C for 12 h, the desired product **2a** was isolated in 89% yield.

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To understand the scope of this cyclization, alkyne **3a** was next reacted with different arylboronic acids bearing both electron-donating and electron-withdrawing groups. The results summarized in Table 2 indicated that the reaction was fairly general in scope as electron density in boronic acid had little effect on reaction yield. In all cases, excellent ratios of desired compounds (**8a-f**, **1a**) versus direct coupling products were obtained (>25:1). The excellent selectivity could be explained by assuming that the desired cascade reaction is sterically favored over the undesired direct coupling reaction. A moderate yield (59%) was obtained when *m*-methylsulfonylaminophenyl boronic acid was used (Table 2, entry 7). The reduced yield resulted from competitive deboronation, ¹⁴ which was observed only in this case where a sulfonamide group was present.

Similar reaction conditions used for the preparation of **2a** were applied to the formation of nuclear hormone modulators **2b**—**e**¹⁵ (Table 3). Both electron-donating groups (entry 3) and electron-withdrawing groups (entry 4) could be employed in the aromatic ring directly connected to the alkyne to afford the desired dibenzoxapine derivatives. The same strategy was also used in the synthesis of dibenzoxapine derivative with the opposite olefin isomer (entry 5). In all cases, the desired compounds were isolated as single stereoisomers. As anticipated, the eight-membered-ring product was not observed, indicating the regioselectivity was dictated by the 7-exo-dig nature of the process rather than the steric effect as in the

intermolecular reactions.¹⁶ It was also noteworthy that the reaction proceeded with high catalyst turnover rate. The reaction was more efficient with 0.5 mol % than 5 mol % Pd(OAc)₂.

With stereodefined tetrasubstituted exocyclic alkenes $2\mathbf{a} - \mathbf{d}$ in hand, the final targets $1\mathbf{a} - \mathbf{d}$ were prepared uneventfully via a nitro group reduction followed by a sulfonation procedure (79–89% yield). Because of steric hindrance, the tetrasubstituted alkenes were generally inert to various hydrogenation conditions.

In summary, the palladium-catalyzed intramolecular alkyne carbometalation reaction has proven to be a practical means for preparing dibenzoxapine containing tetrasubstituted exocyclic alkenes. High regio- and stereocontrol was obtained in all cases. Employing phosphine-free Pd(0) catalyst and water as the cosolvent afforded the target compounds in excellent yields. The practical utility of this novel method was demonstrated on the kilogram scale to afford the desired dibenzoxapine with over 70% yield.

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Supporting Information Available: Complete experimental details along with spectroscopic data for all new molecules synthesized. This material is available free of charge via the Internet at http://pubs.acs.org.

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⁽¹³⁾ Since all of the above optimization was carried out with Na_2CO_3 as base, we also investigated the effect of base on the reaction results. CsF, $NaHCO_3$, K_2CO_3 , and $KHCO_3$ were used to replace Na_2CO_3 under the optimized conditions. They all afforded good 2a/6 selectivity, but the yield is generally lower.

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