

Molecular modulation of muscarinic antagonists. Synthesis and affinity profile of 2,2-diphenyl-2-ethylthio-acetic acid esters designed to probe the binding site cavity

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Abstract

The synthesis and preliminary pharmacological profile of a new series of muscarinic antagonists, derived from previously studied 2,2-diphenyl-2-ethylthio-acetic acid esters, are reported. The parent molecules were decorated with linkers of different length, carrying an amino group to catch a putative anionic function outside the recognition site of the receptor. It was hoped that the interception of this function would give molecules with higher potency and selectivity. The attempt has not been successful, but a new series of compounds with a peculiar pharmacological profile has been identified.

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1. Introduction

Muscarinic receptors are found in the central and peripheral nervous system. They play an important role in several vital functions such as control of movement, cognition, smooth muscle tone, and glandular secretion [1]. As a consequence of this vast array of actions, muscarinic receptor agonists and antagonists have potential for therapeutic use in a variety of pathologies such as smooth muscle hyperactivity and neurodegenerative diseases [2–4]. However, non selective compounds, because of the large distribution of the receptors in the body and the existence of several subtypes, may produce several, severe, side effects.

Five distinct subtypes of muscarinic receptors have been cloned (M_1 – M_5) but only four have been characterized pharmacologically (M_1 – M_4) [5]. Fundamental information on muscarinic receptor subtypes has been collected from a variety of more or less selective synthetic ligands [6,7], trans-

genic mice lacking genes encoding each of the muscarinic receptor subtypes [8] and muscarinic toxins from snake venoms [9]. As a matter of fact, several synthetic ligands (mainly antagonists) do show some subtype selectivity [2], but none is so selective as to display at least 100-fold selectivity for one subtype over all the others. This can be due to the high degree of sequence identity in the binding sites of the five subtypes [10,11], even if it has to be considered that a single amino acid difference in GPCR subtypes may yield significant differences in ligand binding affinity [12]. As a consequence, the number of muscarinic agonists and antagonists that have been introduced in therapy are relatively small [2].

Given the situation described above, it is not surprising that research on muscarinic ligands is still very active [6,7] in particular on antagonists [13–21]. Indeed, selective compounds are badly needed both for further characterization of subtypes and for therapeutic applications.

In the past decade, we have been engaged in the synthesis and study of a series of muscarinic antagonists that are amino esters of 2,2-diphenyl-2-ethylthio-acetic acid such as **1** and **2**

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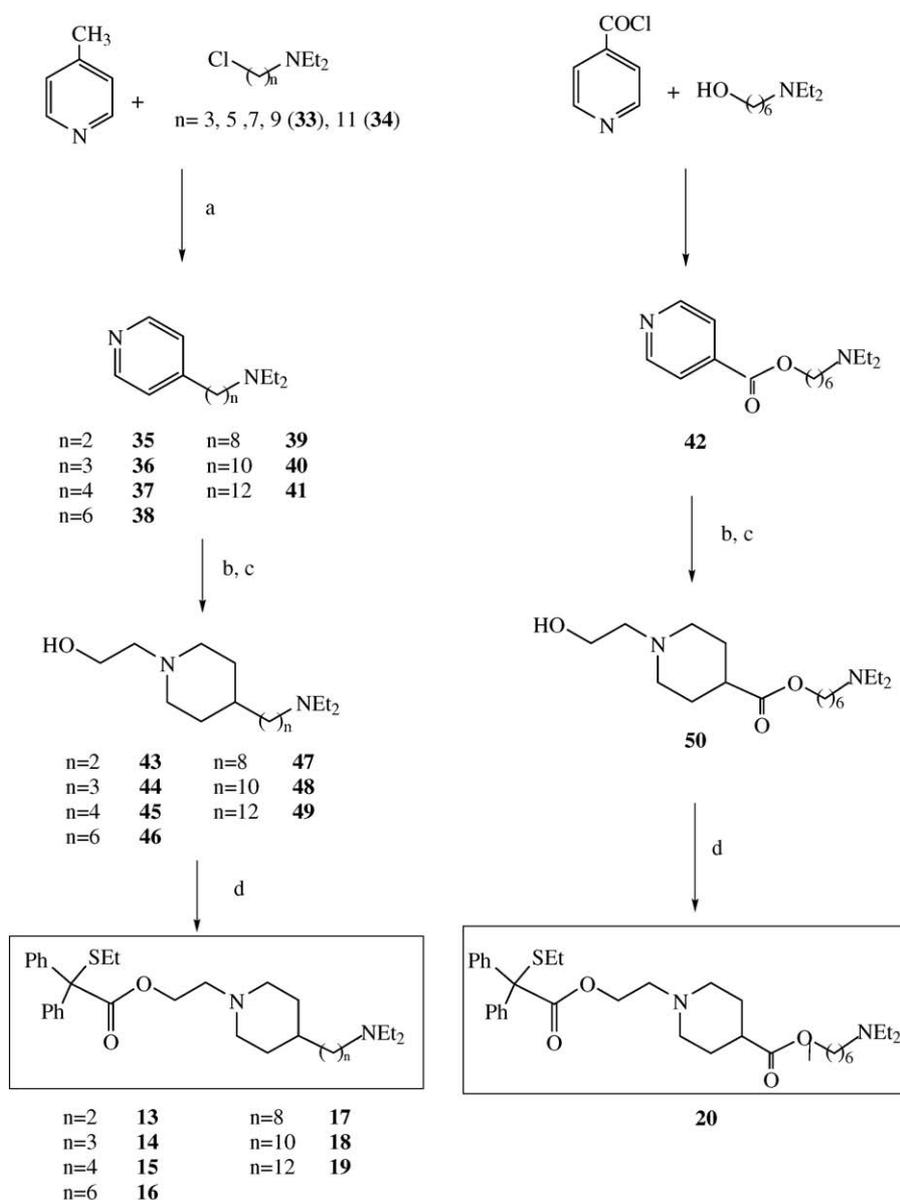
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mines. 2-Chloroethylamine is commercially available, while other chloramines ($n = 3$ [28], 4 [29], 8 [30], 10 (23)) were obtained by treating the corresponding amino alcohols with SOCl_2 in CHCl_3 stabilized with amylene. In turn, the amino alcohols ($n = 3$ [31], 4 [32], 8 [30], 10 [33]) were synthesized from the corresponding commercially available chloro or bromoalcohols and diethylamine in anhydrous CH_3CN in presence of K_2CO_3 according to a previously reported procedure [34]. Compounds 6 and 9 were obtained by an alternative pathway treating 2-hydroxyethylpiperazine with the proper chloramines ($n = 6$ [35], 12 (24)) and then coupling the obtained compounds 28 and 29 with 2,2-diphenyl-2-ethylthio-acetyl chloride.

In the same scheme, the synthesis of compounds 10–12 is also reported. Halo esters 25–27 were obtained from com-

mercially available 4-chlorobutyryl chloride and 3-chloropropionyl chloride, respectively, with 3-diethylpropano-amine (25) and 12-diethyldodecanolamine (27)[33] and from the reaction of 5-bromopentanoyl chloride [36] and 2-diethyletanolamine (26). Their reaction with 2-hydroxyethylpiperazine gave 30–32, which, after reaction with 2,2-diphenyl-2-ethylthio-acetyl chloride, afforded compounds 10–12.

The key intermediates in the synthesis of the piperidine compounds 13–20 (Scheme 2) were pyridine amines 35–42. Compounds 35 [37], 36 [38] and 37 [38] have already been described, and compounds 38–41 were synthesized with the method reported in literature [38], using the proper chloramines ($n = 5$ [39], 7 [40], 9 (33), 11 (34)). Compound 42 was obtained from the reaction of isonicotinoyl chloride [41] with



a) LDA, THF; b) H_2/PtO_2 ; c) $\text{BrCH}_2\text{CH}_2\text{OH}$; d) 2,2-diphenyl-2-ethylthioacetyl chloride, CH_2Cl_2 .

Scheme 2.

the appropriate amino alcohol. The pyridine ring of compounds **35–42** was then hydrogenated at 48 psi for 48 h using PtO_2 as a catalyst and a mixture of MeOH and HCl conc. as a solvent. Reaction with bromoethanol in anhydrous CH_3CN gave the corresponding alcohols **43–50** that were then reacted with 2,2-diphenyl-2-ethylthio-acetyl chloride affording compounds **13–20**.

3. Results and discussion

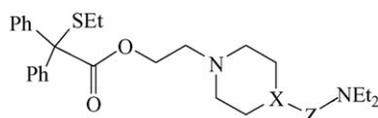
Binding affinity of the synthesized compounds **3–20** and of the parent compounds **1, 2** was measured on CHO-expressed human muscarinic receptor subtypes (hM_1 – hM_5) and the results are reported in Table 1. Compared to the parent compounds, none of the newly synthesized molecules showed the sharp increase in affinity expected in the case of a successful ionic interaction of the introduced amine function.

There is indeed a small increase of affinity for piperazinic compounds with $n = 8$ (**7**), 10 (**8**), 12 (**9**) and for piperidinic compounds with $n = 6$ (**16**), 8 (**17**) but, very likely, this increase is due to the contribution of the methylene groups rather than to new ionic interactions. This is somehow confirmed by the fact that the increase in affinity is similar in all subtypes and the new molecules completely lack selectivity.

Since, as discussed in the Introduction, the parent compounds **1** and **2** show a fairly good potency and selectivity in functional assays, we decided to check the functional activity of selected compounds **7** and **17**, which are among those showing an increase in affinity. The results are reported in Table 2. Unlike the parent compounds, **7** and **17** are only modest functional antagonists and do not show any subtype selectivity, confirming that our attempt was unsuccessful.

However, these two compounds do show some peculiar features that recall those of a related series of 2,2-diphenyl-2-ethylthio-acetic acid esters described in a previous work

Table 1
Binding affinity of compounds **1–20** on CHO expressed human muscarinic receptors ^a



N	X	Z	pK _i N	HM ₁	HM ₂	HM ₃	HM ₄	HM ₅
1	–	–	–	6.47 ± 0.08	6.25 ± 0.10	6.44 ± 0.08	5.95 ± 0.11	6.25 ± 0.14
2	–	–	–	6.56 ± 0.04	6.48 ± 0.06	6.43 ± 0.07	6.04 ± 0.16	6.28 ± 0.10
3	N	(CH ₂) _n	2	5.82 ± 0.39	5.79 ± 0.30	5.93 ± 0.62	5.51 ± 0.43	5.73 ± 0.39
4	N	(CH ₂) _n	3	6.03 ± 0.19	6.39 ± 0.14	5.72 ± 0.42	5.32 ± 0.82	5.39 ± 0.85
5	N	(CH ₂) _n	4	6.64 ± 0.06	6.68 ± 0.04	6.38 ± 0.08	6.14 ± 0.07	6.31 ± 0.09
6	N	(CH ₂) _n	6	6.07 ± 0.19	5.73 ± 0.57	5.95 ± 0.15	5.59 ± 0.35	5.80 ± 0.30
7	N	(CH ₂) _n	8	7.17 ± 0.01	6.59 ± 0.06	6.46 ± 0.04	6.84 ± 0.01	6.82 ± 0.03
8	N	(CH ₂) _n	10	7.30 ± 0.01	7.00 ± 0.02	6.46 ± 0.04	7.18 ± 0.01	6.77 ± 0.04
9	N	(CH ₂) _n	12	7.19 ± 0.02	7.07 ± 0.02	6.79 ± 0.04	7.29 ± 0.01	6.95 ± 0.03
10	N	(CH ₂) ₃ COO(CH ₂) ₃	–	6.23 ± 0.01	6.46 ± 0.10	5.67 ± 0.23	5.34 ± 0.68	5.71 ± 0.40
11	N	(CH ₂) ₄ COO(CH ₂) ₂	–	5.92 ± 0.20	5.65 ± 0.65	5.16 ± 1.10	5.36 ± 0.66	5.44 ± 0.77
12	N	(CH ₂) ₂ COO(CH ₂) ₁₂	–	6.61 ± 0.08	6.47 ± 0.07	6.63 ± 0.07	6.80 ± 0.04	6.58 ± 0.07
13	CH	(CH ₂) _n	2	6.10 ± 0.12	6.29 ± 0.09	5.82 ± 0.20	5.70 ± 0.26	5.90 ± 0.24
14	CH	(CH ₂) _n	3	6.13 ± 0.11	6.67 ± 0.05	5.90 ± 0.11	5.86 ± 0.17	5.98 ± 0.21
15	CH	(CH ₂) _n	4	6.49 ± 0.05	6.63 ± 0.07	6.39 ± 0.06	6.31 ± 0.06	6.54 ± 0.05
16	CH	(CH ₂) _n	6	7.07 ± 0.02	7.08 ± 0.02	6.46 ± 0.07	6.94 ± 0.01	6.69 ± 0.04
17	CH	(CH ₂) _n	8	7.00 ± 0.03	6.60 ± 0.07	6.12 ± 0.14	6.72 ± 0.03	6.31 ± 0.13
18	CH	(CH ₂) _n	10	6.85 ± 0.04	6.83 ± 0.03	6.38 ± 0.10	6.89 ± 0.02	6.41 ± 0.01
19	CH	(CH ₂) _n	12	6.43 ± 0.09	6.23 ± 0.13	6.40 ± 0.10	6.27 ± 0.09	6.35 ± 0.12
20	CH	COO(CH ₂) ₆	–	7.03 ± 0.02	6.58 ± 0.10	6.27 ± 0.07	6.53 ± 0.04	6.63 ± 0.05

^a 3–6 Experiments in duplicate.

Table 2
Functional assays of compounds **1, 2, 7, 17** on muscarinic receptor subtypes

N	pK _b ± SE (pA ₂ ± SE)			
	M ₁ Rabbit vas deferens	M ₂ Guinea pig atrium	M ₃ Guinea pig ileum	M ₄ Guinea pig lung
1	(7.79 ± 0.01)	(7.93 ± 0.03)	(5.88 ± 0.09)	–
2	(8.97 ± 0.06)	(9.12 ± 0.03)	(6.50 ± 0.07)	–
7^a	6.47 ± 0.09	7.17 ± 0.09	5.94 ± 0.01	5.58 ± 0.01
17^a	<5.52	6.04 ± 0.08	<5.52	5.40 ± 0.12

^a Compounds **7** and **17** behave also as putative muscarinic agonists on M₁ and M₂ receptors. See text for details.

[42]. In fact, compounds **7** and **17** behave also as putative muscarinic agonists on M_1 and M_2 receptors since, after 1 h incubation at 10^{-5} M, they strongly reduce the twitch induced by electrical stimulation of rabbit vas deferens and guinea pig heart, as classical muscarinic agonists do, even if they, like the compounds described previously, are characterized by large size structures typical of muscarinic antagonists. The reasons for such behavior are under investigation.

4. Experimental section

4.1. Chemistry

All melting points were taken on a Büchi apparatus and are uncorrected. Infrared spectra were recorded with a Perkin–Elmer 681 or a Perkin–Elmer Spectrum RX I FT-IR spectrophotometer in Nujol mull for solids and neat for liquids. Unless otherwise stated, NMR spectra were recorded on a Gemini 200 spectrometer, and chromatographic separations were performed on a silica gel column by gravity chromatography (Kieselgel 40, 0.063–0.200 mm; Merck) or flash chromatography (Kieselgel 40, 0.040–0.063 mm; Merck). Although the IR spectra data are not always included (because of lack of unusual features), they were obtained for all reported compounds and are consistent with the assigned structures. Yields are given after purification, unless otherwise stated. Where analyses are indicated by symbols, the analytical results are within $\pm 0.4\%$ of the theoretical values. Compounds were named following IUPAC rules as applied by Beilstein-Institut AutoNom (version 2.1), a software for systematic names in organic chemistry. When reactions were performed in anhydrous conditions, the mixtures were maintained under nitrogen.

4.2. 4-[2-(2-Ethylsulfanyl-2,2-diphenylacetoxy)ethyl]piperazine-1-carboxylic acid tert-butyl ester (**21**)

A total of 1.92 g (9 mmol) of (*tert*-BOC)hydroxyethylpiperazine were reacted with 2.62 g (9 mmol) of 2,2-diphenyl-2-ethylthioacetylchloride [25] in anhydrous CH_2Cl_2 at r.t. for 2 h. The solution was then basified with 10% NaOH and extracted with CHCl_3 . The crude product was purified by flash chromatography using abs. EtOH: CH_2Cl_2 :Et₂O:petroleum ether = 180:360:360:900 as eluting system. The oily compound, 39% yield, was used as such in the next reaction.

¹H-NMR (CDCl_3) δ : 1.10 (*t*, $J = 6.67$ Hz, 3H, SCH_2CH_3); 1.45 (*s*, 9H, $\text{C}(\text{CH}_3)_3$); 2.16–2.26 (*m*, 4H, 4 *CH* piperazine); 2.34 (*q*, $J = 6.67$ Hz, 2H, SCH_2CH_3); 2.53 (*t*, $J = 3.33$ Hz, 2H, $\text{COOCH}_2\text{CH}_2\text{N}$); 3.21–3.35 (*m*, 4H, 4 *CH* piperazine); 4.28 (*t*, $J = 3.33$ Hz, 2H, COOCH_2); 7.15–7.45 (*m*, 10H, aromatics).

4.3. Ethylsulfanyldiphenylacetic acid 2-piperazin-1-ylethyl ester (**22**)

A total of 0.28 g (0.7 mmol) of **21** was dissolved in 10 ml of AcOEt added with 6 ml of 6N HCl. The reaction was

stirred at r.t. for 12 h; the mixture was then treated with 10% NaOH and the product extracted in CHCl_3 . The organic phase was anhydried over Na_2SO_4 , the solvent removed under vacuum and the title compound, obtained quantitatively as an oil, used as such for the next reaction.

¹H-NMR (CDCl_3) δ : 1.10 (*t*, $J = 3.33$ Hz, 3H, SCH_2CH_3); 2.25–2.45 (*m*, 6H, 4 *CH* piperazine; SCH_2CH_3); 2.55 (*t*, $J = 1.67$ Hz, 2H, $\text{COOCH}_2\text{CH}_2\text{N}$); 2.73–2.83 (*m*, 4H, 4 *CH* piperazine); 4.30 (*t*, $J = 1.67$ Hz, 2H, COOCH_2); 7.20–7.50 (*m*, 10H, aromatics).

4.4. 12-(Chlorododecyl)diethylamine (**24**)

One equivalent of 12-diethylaminododecanol [33] was dissolved in CHCl_3 stabilized with amylene, added with an excess of SOCl_2 , and refluxed for 2 h. The excess of SOCl_2 was eliminated under vacuum, the crude mixture basified with Na_2CO_3 and the chlorododecyl-diethylamine (**24**) extracted with CHCl_3 . The organic phase was dried over Na_2SO_4 and the solvent removed under vacuum. The title compound was obtained as oil. Yield 73%. ¹H-NMR (CDCl_3) δ : 1.02 (*t*, $J = 7.14$ Hz, 6H, $2\text{NCH}_2\text{CH}_3$); 1.22–1.50 (*m*, 18 H); 1.76 (*m*, 2H); 2.40 (*t*, $J = 7.69$ Hz, 2H, $\text{CH}_2\text{CH}_2\text{N}$); 2.68 (*q*, $J = 7.14$ Hz, 4H, $2\text{NCH}_2\text{CH}_3$); 3.53 (*t*, $J = 6.78$ Hz, 2H, CH_2Cl). Anal. ($\text{C}_{16}\text{H}_{34}\text{ClN}$) C,H,N.

Compounds **23**, **33** and **34** were obtained in the same way and their spectroscopic characteristics are consistent with the proposed structures.

4.5. 4-Chloro-butyrac acid 3-diethylamino-propyl ester (**25**)

One equivalent of 3-diethylaminopropanol was dissolved in anhydrous CH_2Cl_2 , cooled at 0 °C and added dropwise with 1 equivalent of 4-chlorobutirrylchloride; the solution was stirred at r.t. for 48 h. The mixture was basified with a 10% NaOH solution and extracted with CH_2Cl_2 . The organic phase was dried over Na_2SO_4 and the solvent removed under vacuum afforded compound **25** as oil. Yield 100%. ¹H-NMR (CDCl_3) δ : 0.99 (*t*, $J = 7.10$ Hz, 6H, NCH_2CH_3); 1.68–1.82 (*m*, 2H, $\text{ClCH}_2\text{CH}_2\text{CH}_2\text{CO}$); 2.00–2.14 (*m*, 2H, $\text{OCH}_2\text{CH}_2\text{CH}_2\text{N}$); 2.43–2.54 (*m*, 8H, CH_2COO , CH_2NEt_2 , $2\text{NCH}_2\text{CH}_3$); 3.58 (*t*, $J = 6.60$ Hz, 2H, ClCH_2); 4.11 (*t*, $J = 6.60$ Hz, 2H, COOCH_2). Anal. ($\text{C}_{11}\text{H}_{22}\text{ClNO}_2$) C,H,N.

Compounds **26** and **27** were obtained in the same way and their spectroscopic characteristics are consistent with the proposed structures.

4.6. 2-[4-(6-Diethylaminoethyl)piperazin-1-yl]ethanol (**28**)

One equivalent of 6-chlorohexyldiethylamine dissolved in abs. EtOH was added with *N*-2-hydroxyethylpiperazine (1 equivalent) and with anhydrous NEt_3 (2 equivalents) and the mixture refluxed for 24 h. The solvent was removed under reduced pressure and the resulting thick oil was heated at 80 °C for 20 h; 20 ml of a saturated solution of Na_2CO_3 were

added and the resulting suspension extracted with CH_2Cl_2 . After anhydrication over Na_2SO_4 , the solvent was removed under vacuum and the mixture purified by flash chromatography using NH_4OH :abs. EtOH: CH_2Cl_2 :petroleum ether = 45:225:600:90 as eluting system. Oil; yield 16%. $^1\text{H-NMR}$ (CDCl_3) δ : 1.02 (*t*, $J = 6.96$ Hz, 6H, $2\text{NCH}_2\text{CH}_3$); 1.28–1.38 (*m*, 4H); 1.42–1.55 (*m*, 4H); 2.28–2.58 (*m*, 18H, $\text{HOCH}_2\text{CH}_2\text{N}$, CH_2NEt_2 , Npip-CH_2 , NCH_2CH_3 , 8H piperazine); 2.95 (*bs*, 1H, *OH*); 3.6 (*t*, $J = 5.49$ Hz, 2H, CH_2OH). Anal. ($\text{C}_{16}\text{H}_{35}\text{N}_3\text{O}$) C,H,N.

Compound **29** was obtained in the same way and its spectroscopic characteristics are consistent with the proposed structure.

4.7. 4-[4-(2-Hydroxyethyl)piperazin-1-yl]butyric acid 3-diethylaminopropyl ester (**30**)

Compound **25** (1 equivalent) was dissolved in toluene and 2-hydroxyethylpiperazine (1 equivalent) and an excess of K_2CO_3 were added. The mixture was heated at 100 °C under stirring for 2 days, then K_2CO_3 was filtered off, the solvent evaporated under reduced pressure and the oily product purified by flash chromatography using NH_4OH :abs. EtOH: CH_2Cl_2 :petroleum ether = 8:65:340:60 as eluting system. Compound **30** was obtained as oil in 100% yield. $^1\text{H-NMR}$ (CDCl_3) δ : 1.01 (*t*, $J = 7.10$ Hz, 6H, $2\text{NCH}_2\text{CH}_3$); 1.70–1.85 (*m*, 5H, $\text{NCH}_2\text{CH}_2\text{CH}_2\text{CO}+\text{OCH}_2\text{CH}_2\text{CH}_2\text{N}+\text{OH}$); 2.30–2.56 (*m*, 20H); 3.60 (*t*, $J = 5.30$ Hz, 2H, CH_2OH); 4.11 (*t*, $J = 6.20$ Hz, 2H, COOCH_2). Anal. ($\text{C}_{17}\text{H}_{35}\text{N}_3\text{O}_3$) C,H,N.

Compounds **31** and **32** were obtained in the same way and their spectroscopic characteristics are consistent with the proposed structures.

4.8. Ethylsulfanyldiphenylacetic acid 2-[4-(6-diethylaminohexyl)piperidin-1-yl]ethyl ester (**6**)

Compound **28** (0.04 g) was reacted with 2-ethylthio-2,2-diphenylacetyl chloride [25] (0.041 g 0.14 mmol) in 20 ml of anhydrous CH_2Cl_2 under N_2 at r.t. for 6 days. The reaction mixture was then treated with 10% NaOH solution and extracted with CH_2Cl_2 ; the organic phase was dried over Na_2SO_4 and CH_2Cl_2 evaporated under vacuum. The title compound was obtained as oil after flash chromatographic separation using NH_4OH :abs. EtOH: CH_2Cl_2 :petroleum ether = 8:65:340:60 as eluting system. Yield 19%. $^1\text{H-NMR}$ (CDCl_3) δ : 0.99–1.18 (*m*, 9H, SCH_2CH_3 , $2\text{NCH}_2\text{CH}_3$); 1.25–1.35 (*m*, 4H); 1.46–1.56 (*m*, 4H); 2.25–2.65 (*m*, 20H, 8CH piperazine, SCH_2CH_3 ; $2\text{NCH}_2\text{CH}_3$, CH_2NEt_2 , $\text{OCH}_2\text{CH}_2\text{N}$, Npip-CH_2); 4.32 (*t*, $J = 5.49$ Hz 2H, COOCH_2); 7.27–7.38 (*m*, 6H, aromatics); 7.38–7.50 (*m*, 4H, aromatics). Anal. ($\text{C}_{32}\text{H}_{49}\text{N}_3\text{O}_2\text{S}$) C,H,N.

Compounds **9–12** were obtained in the same way and their spectroscopic characteristics are consistent with the proposed structures.

4.9. Ethylsulfanyldiphenylacetic acid 2-[4-(8-diethylaminoethyl)piperazin-1-yl]ethyl ester (**7**)

A total of 0.6 g (1.46 mmol) of **22** was dissolved in 15 ml of abs. EtOH, added with (8-chlorooctyl)diethyl amine [30] 0.37 g (1.46 mmol) and anhydrous NEt_3 (2 equivalents, 0.4 ml) and refluxed for 3 days. The solvent was removed and the residue treated with 10% NaOH solution and extracted with CHCl_3 ; the organic phase was dried over Na_2SO_4 and CHCl_3 evaporated under vacuum. The title compound was obtained as oil after two column chromatographic separations using NH_4OH :abs. EtOH: CH_2Cl_2 : Et_2O :petroleum ether = 9.9:180:360:360:900 and NH_4OH :abs. EtOH: CH_2Cl_2 : petroleum ether = 8:65:340:60, respectively, as eluting system. Yield 22%. $^1\text{H-NMR}$ (CDCl_3) δ : 0.94–1.56 (*m*, 9H, SCH_2CH_3 , $2\text{NCH}_2\text{CH}_3$); 1.19–1.56 (*m*, 14H); 1.65–1.87 (*m*, 2H); 2.18–2.66 (*m*, 14H, 8CH piperazine, SCH_2CH_3 , $2\text{NCH}_2\text{CH}_3$); 3.52 (*t*, $J = 1.67$ Hz, 2H, $\text{COOCH}_2\text{CH}_2\text{N}$); 4.32 (*t*, $J = 1.67$ Hz 2H, COOCH_2); 7.20–7.37 (*m*, 6H, aromatics); 7.37–7.54 (*m*, 4H, aromatics). Anal. ($\text{C}_{35}\text{H}_{53}\text{N}_3\text{O}_2\text{S}$) C,H,N.

Compounds **3–5** and **8** were obtained in the same way starting from compound **22** and the appropriate chloroalkyldiethylamine ($n = 2, 3, 4, 10$) and their spectroscopic characteristics are consistent with the proposed structures.

4.10. Diethyl(6-pyridin-4-ylhexyl)amine (**38**)

Butyllithium in hexane (3.2 ml of a 1.6 M solution; 5 mmol) was added to 0.5 g (5 mmol) of diisopropylamine dissolved in THF (15 ml), cooled at 0 °C and the mixture stirred for 30 min. 4-Methylpyridine (0.45 g, 5 mmol) was then added at 0 °C and the mixture stirred for 30 min; then the electrophile 5-chloropentyl-diethyl amine [39] was added (1.52 g, 5.5 mmol) and the mixture warmed up to r.t. and treated with a saturated solution of NH_4Cl . The solution was extracted with diethyl ether and the organic phase dried. The oily product was purified by flash chromatography using NH_4OH :abs.

EtOH: CHCl_3 : Et_2O :petroleum ether = 25:45:180:180:450 as eluting system. Yield 18%. $^1\text{H-NMR}$ (CDCl_3) δ : 0.84 (*t*, $J = 7.30$ Hz, 6H, $2\text{NCH}_2\text{CH}_3$); 1.15–1.22 (*m*, 4H); 1.38–1.50 (*m*, 2H); 1.60–1.80 (*m*, 2H); 2.23 (*t*, $J = 7.30$ Hz, 2H, CH_2NEt_2); 2.29–2.45 (*m*, 6H, $2\text{NCH}_2\text{CH}_3$, pyridine- CH_2); 6.92 (*d*, $J = 5.86$ Hz, 2H pyridine); 8.28 (*d*, $J = 5.86$ Hz, 2H pyridine). Anal. ($\text{C}_{15}\text{H}_{26}\text{N}_2$) C,H,N.

Compounds **37** [38], **39**, **40** and **41** were obtained in the same way, treating 4-methylpyridine with the appropriate electrophile. Their spectroscopic characteristics are consistent with the proposed structures.

4.11. Isonicotinic acid 6-diethylaminoethyl ester (**42**)

A total of 0.8 g (4.6 mmol) of 6-diethylaminohexanol was dissolved in CHCl_3 , and anhydrous NEt_3 (2 equivalent, 1.3 ml) was added; the reaction was cooled at 0 °C and

isonicotinoylchloride in CHCl_3 (0.76 g, 4.6 mmol) was added. The mixture was stirred for 36 h at r.t., treated with 10% NaOH solution, extracted with CHCl_3 and the organic phase dried. The oily product was then purified by flash chromatography using NH_4OH :abs. EtOH: CH_2Cl_2 :Et₂O:petroleum ether = 9.9:180:360:360:900 as eluting system. Yield 26%. ¹H-NMR (CDCl_3) δ : 1.00 (t, 6H, $2\text{NCH}_2\text{CH}_3$, $J = 7.10$ Hz); 1.30–1.55 (m, 6H); 1.74–1.81 (m, 2H); 2.41 (t, $J = 7.33$ Hz, 2H, CH_2NEt_2); 2.51 (q, $J = 7.10$ Hz, 4H, $2\text{NCH}_2\text{CH}_3$); 4.34 (t, $J = 6.59$ Hz, 2H, COOCH_2); 7.83 (d, $J = 4.76$ Hz, 2H pyridine); 8.76 (d, $J = 4.76$ Hz, 2H pyridine). Anal. ($\text{C}_{16}\text{H}_{26}\text{N}_2\text{O}_2$) C,H,N.

4.12. 2-[4-(6-Diethylaminoethyl)piperidin-1-yl]ethanol (46)

A total of 1.84 g (8 mmol) of **38** were dissolved into MeOH (50 ml) and conc. HCl (2 ml), added with 0.05 g of PtO_2 and hydrogenated in a Parr apparatus for 24 h at 48 psi. At the end of the reaction, the catalyst was filtered off and the solvent removed under vacuum. The residue was treated with 10% NaOH solution, extracted with Et₂O and the organic phase dried and evaporated under reduced pressure. The oily compound was used as such for the next reaction. Yield 42%. ¹H-NMR (CDCl_3) δ : 0.98 (t, $J = 7.10$ Hz, 6H, $2\text{NCH}_2\text{CH}_3$); 1.15–1.43 (m, 11H); 1.43–1.61 (m, 2H piperidine); 2.02–2.18 (m, 2H); 2.24–2.62 (m, 8H); 2.95–3.08 (m, 2H).

To 0.07 g (0.3 mmol) of diethyl(6-piperidin-4-yl)hexylamine obtained as described above, dissolved in abs. EtOH, anhydrous NEt_3 (0.084 ml, 2 equivalent) and 2-bromoethanol (0.021 ml, 1 equivalent) were added and the resulting mixture was refluxed for 24 h. The solvent was removed and the residue treated with water and extracted with CHCl_3 . The organic phase was dried, the solvent removed and the crude product purified by column chromatography using NH_4OH :abs. EtOH: CH_2Cl_2 :petroleum ether = 45:225:600:90 as eluting system. Oil, yield 35%. ¹H-NMR (CDCl_3) δ : 1.02 (t, $J = 7.10$ Hz, 6H, $2\text{NCH}_2\text{CH}_3$); 1.20–1.41 (m, 12H); 1.43–1.62 (m, 2H); 1.64–1.68 (m, 2H); 2.01 (t, $J = 7.69$ Hz, 2H, CH_2NEt_2); 2.46–2.58 (m, 8H, $2(\text{NCH}_2\text{CH}_3)$, $\text{NCH}_2\text{CH}_2\text{OH}$, piperidine- CH_2); 2.82–2.91 (m, 2H piperidine); 3.59 (t, $J = 5.50$ Hz, 2H, CH_2OH). Anal. ($\text{C}_{17}\text{H}_{36}\text{N}_2\text{O}$) C,H,N.

Compounds **43–45** and **47–50** were obtained in the same way and their spectroscopic characteristics are consistent with the proposed structures.

4.13. Ethylsulfanyldiphenylacetic acid

2-[4-(6-diethylaminoethyl)piperidin-1-yl]ethyl ester (16)

Compound **46** (0.03 g, 0.11 mmol) was refluxed for 48 h under N_2 with 2-ethylthio-2,2-diphenylacetylchloride [25] (0.031 g, 0.11 mmol) in 20 ml of anhydrous CH_2Cl_2 . The reaction mixture was then treated with 10% NaOH solution and extracted with CH_2Cl_2 ; the organic phase was dried over Na_2SO_4 and CH_2Cl_2 evaporated under vacuum. The title

compound was obtained as oil after chromatographic separation using NH_4OH :abs. EtOH: CH_2Cl_2 :petroleum ether = 8:65:340:60 as eluting system. Yield 68%. ¹H-NMR (CDCl_3) δ : 1.03–1.18 (m, 9H, SCH_2CH_3 , $2\text{NCH}_2\text{CH}_3$); 1.20–1.28 (m, 10H); 1.42–1.65 (m, 4H); 1.81–1.96 (m, 2H); 2.38 (q, $J = 7.33$ Hz, 2H, SCH_2CH_3); 2.55–2.85 (m, 11H); 4.32 (t, $J = 5.50$ Hz, 2H, COOCH_2); 7.27–7.35 (m, 6H, aromatics); 7.35–7.45 (m, 4H, aromatics). Anal. ($\text{C}_{33}\text{H}_{50}\text{N}_2\text{O}_2\text{S}$) C,H,N.

Compounds **13–15** and **17–20** were obtained in the same way and their spectroscopic characteristics are consistent with the proposed structures.

5. Pharmacology

5.1. Binding

5.1.1. Cell culture and membrane preparation

Chinese hamster ovary cells stably expressing cDNA encoding human muscarinic M_1 – M_5 receptors were generously provided by Prof. R. Maggio (Department of Neuroscience, University of Pisa, Italy). Growth medium consisted in Dulbecco' modified Eagle's medium supplemented with 10% fetal calf serum (Gibco, Grand Island, N.Y.), 100 units/ml each of penicillin G and streptomycin, 4 mM glutamine (Sigma Aldrich, Milano, Italy) and non essential amino acids (Sigma Aldrich, Milano, Italy) and 50 $\mu\text{g}/\text{ml}$ of geneticin (Gibco, Grand Island, N.Y.) in a humidified atmosphere consisting of 5% CO_2 and 95% air.

Confluent CHO cell lines were scraped, washed with buffer (25 mM sodium phosphate containing 5 mM MgCl_2 at pH 7.4) and homogenized for 30 s using an Ultra-Turrax (setting 5). The pellet was sedimented $17000 \times g$ for 15 min at 4 °C and the membranes were resuspended in the same buffer, rehomogenized with Ultra-Turrax and stored at –80 °C [43]. An aliquot was taken for the assessment of protein content according to the method of Bradford [44] using the Bio-Rad protein assay reagent (Bio-Rad Laboratories, Munchen, West Germany) and bovine serum albumin was used as the standard.

5.1.2. Binding assay

The radioligand binding assay was run in polypropylene 96-well plates (Sarstedt, Verona, Italy) and performed for 120 min at r.t. In a final volume of 0.25 ml in 25 mM sodium phosphate buffer containing 5 μM MgCl_2 at pH 7.4. Final membrane protein concentrations were 30 $\mu\text{g}/\text{ml}$ (M_1), 70 $\mu\text{g}/\text{ml}$ (M_2), 25 $\mu\text{g}/\text{ml}$ (M_3), 50 $\mu\text{g}/\text{ml}$ (M_4) and 25 $\mu\text{g}/\text{ml}$ (M_5).

In homologous competition curves, [³H]-NMS was present at 0.2 nM in tubes containing an increasing concentration of unlabeled NMS (0.03–1000 nM) and at 0.075–0.2 nM in tubes without unlabeled ligand. In heterologous competition curves, fixed concentrations of the tracer (0.2 nM) were displaced by increasing concentrations of several unlabeled

ligands (0.01–1000 μM); all measurements were obtained in duplicate. At the end of the binding reaction, free radioligand was separated from bound ligand by rapid filtration through UniFilter GF/C plates (Perkin Elmer Life Science, Boston, MA) using a FilterMate Cell Harvester (Perkin Elmer Life Science, Boston, MA); after filtration, the filters were washed several times with ice cold buffer and allowed to dry overnight at r.t. under air flow, added of 25 μl of scintillation liquid (Microscint-20, Perkin Elmer Life Science, Boston, MA) and counted by TopCount NXT Microplate Scintillation Counter (Perkin Elmer Life Science, Boston, MA). The binding data were analyzed by the weighted least-squares iterative curve fitting LIGAND [44] to obtain the affinity constant (K_i) and the binding capacity (B_{max}).

5.2. Functional tests

5.2.1. General considerations

Male guinea pigs (200–300 g) and male New Zealand white rabbits (3.0–3.5 kg) were killed by cervical dislocation. The organs required were set up rapidly under 1 g of tension in 20 ml organ baths containing physiological salt solution (PSS) maintained at an appropriate temperature (see below) and aerated with 5% CO_2 95% O_2 . Dose–response curves were constructed by cumulative addition of the reference agonist. The concentration of agonist in the organ bath was increased approximately three-fold at each step, with each addition being made only after the response to the previous addition had attained a maximal level and remained steady. Following 30 min of washing, tissues were incubated with the antagonist for 60 min, and a new dose–response curve to the agonist was obtained. In the case of the agonist following 30 min washing, a cumulative dose–response curve to the agonist under study was constructed. Responses were expressed as a percentage of the maximal response obtained in the control curve. Contractions were recorded by means of a force displacement transducer connected to the MacLab system PowerLab/800. In addition, parallel experiments in which tissues did not receive any antagonist were run in order to check any variation in sensitivity.

In all cases, parallel experiments in which tissues received only the reference agonist were run in order to check any variation in sensitivity.

All animal testing was carried out according to European Communities Council Directive of 24 November, 1986 (86/609/EEC).

5.2.2. Guinea pig ileum

Two-centimeter-long portions of terminal ileum were taken at about 5 cm from the ileum–cecum junction and mounted in PSS, at 37 °C, of the following composition (mM): NaCl 118, NaHCO_3 23.8, KCl 4.7, $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ 1.18, KH_2PO_4 1.18, CaCl_2 2.52, and glucose 11.7. Tension changes were recorded isotonicity. Tissues were equilibrated for 30 min, and dose–response curves to arecaidine propargyl ester (APE) were obtained at 30 min intervals, the first one

being discarded and the second one being taken as the control.

5.2.3. Guinea pig stimulated left atria

The heart was rapidly removed, and the right and left atria were separately excised. Left atria were mounted in PSS (the same used for ileum) at 30 °C and stimulated through platinum electrodes by square-wave pulses (1 ms, 1 Hz, 5–10 V) (Tetra Stimulus, N. Zagnoni). Inotropic activity was recorded isometrically. Tissues were equilibrated for 2 h and a cumulative dose–response curve to APE was constructed.

5.2.4. Guinea pig lung strips

The lungs were rapidly removed and strips of peripheral lung tissue were cut either from the body of a lower lobe with the longitudinal axis of the strip parallel to the bronchus or from the peripheral margin of the lobe. The preparations were mounted, with a preload of 0.3 g, in PSS with the following composition (mM): NaCl (118.78), KCl (4.32), $\text{CaCl}_2 \cdot 2\text{H}_2\text{O}$ (2.52), $\text{MgSO}_4 \cdot 7\text{H}_2\text{O}$ (1.18), KH_2PO_4 (1.28), NaHCO_3 (25), glucose (5.55). Contractions were recorded isotonicity at 37 °C after tissues were equilibrated for 1 h, then two cumulative dose–response curves to APE (0.01, 0.1, 1, 10, 100 μM) were obtained at 45 min intervals, the first one being discarded and the second one being taken as the control.

5.2.5. Rabbit stimulated vas deferens

This preparation was set up according to Eltze [45]. Vasa deferentia were carefully dissected free of surrounding tissue and were divided into four segments, two prostatic portions of 1 cm and two epididymal portions of approximately 1.5 cm length. The four segments were mounted in PSS with the following composition (mM): NaCl (118.4), KCl (4.7), CaCl_2 (2.52), MgCl_2 (0.6), KH_2PO_4 (1.18), NaHCO_3 (25), glucose (11.1); 10^{-6} M yohimbine and 10^{-8} M triptamine were included to block α_2 -adrenoceptors and M_2 muscarinic receptors, respectively. The solution was maintained at 30 °C and tissues were stimulated through platinum electrodes by square-wave pulses (0.1 ms, 2 Hz, 10–15 V). Contractions were measured isometrically after tissues were equilibrated for 1 h, then a cumulative dose–response curve to pCl-McN-A-343 was constructed.

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