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## Potent and selective, sulfamide-based human β<sub>3</sub>-adrenergic receptor agonists

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**Abstract**—A series of sulfamide-based analogs related to L-796568 were prepared and evaluated for their biological activity at the human  $\beta_3$ -adrenergic receptor (AR). This modification allows for a significant reduction in molecular weight, while maintaining single-digit nanomolar potencies at the  $\beta_3$ -AR and high selectivities versus the  $\beta_1$ - or  $\beta_2$ -AR. © 2004 Elsevier Ltd. All rights reserved.

It has been fifty-five years since the receptor systems that are activated by the endogenous catecholamines, adrenaline, and noradrenaline, were classified into  $\alpha$ - and  $\beta$ -subtypes.<sup>1</sup> Twenty years later the  $\beta$ -adrenoceptor ( $\beta$ -AR) classes were further subdivided into the  $\beta_1$ - and  $\beta_2$ -subtypes.<sup>2</sup> In this classification system the lipolytic response (i.e., hydrolysis of fatty acid stores), induced by  $\beta$ -AR agonists in adipocytes, was initially defined as a  $\beta_1$ -AR-mediated response.<sup>2</sup> With the advent of selective  $\beta$ -AR antagonists by the early 1980s it became clear that the lipolytic response was not wholly mediated via the  $\beta_1$ -subtype, leading to the proposal of a third member of the beta-adrenoceptor family  $(\beta_3-AR)$ .<sup>3</sup> Drug discovery efforts directed toward optimization of the pharmacological properties associated with the  $\beta_3$ -AR, such as thermogenesis (energy expenditure) in brown adipose tissue and lipolysis in white adipose tissue, lead to the identification of several potent chemotypes (see Fig. 1).<sup>4</sup> Based on the tissue expression of the  $\beta_3$ -AR there is therapeutic potential for these agents in the treatment of obesity, type-II diabetes, and urinary incontinence.

A number of agents were advanced into human clinical trials during the late 1980s and early 1990s, however

none of these compounds produced significant responses in the disease endpoints of type-II diabetes or obesity.<sup>4</sup> It was during this period that the human  $\beta$ 3-AR was cloned and expressed.<sup>5</sup> Subsequent investigation revealed that the rat  $\beta_3$ -AR pharmacology was significantly different from the human receptor, with the former being significantly more responsive to agonists.<sup>6</sup> Since optimization of the early agonist structures 1 were carried out utilizing the rat receptor, research efforts shifted to identification of agents that were more effective against the human  $\beta$ 3-AR.

One of the series to arise from efforts to make more potent agonists of the human  $\beta_3$ -AR was a series of compounds in which the acid moiety of 1 is replaced by a sulfonamide functionality.<sup>7</sup> These efforts ultimately led to the discovery of L-796568 (2), which was advanced to human clinical trials. After a single-dose of L-796568 in healthy, overweight to obese male subjects modest increases in energy expenditure (8%) were observed at a dose of 1000 mg.<sup>8</sup> However, upon chronic treatment (28 days) there were no statistically significant differences in energy expenditure or body composition.9 Pharmacokinetic evaluation of L-796568 revealed that it has low oral bioavailability in preclinical species, with poor absorption being the major factor.<sup>10</sup> This poor absorption is most likely due to the high molecular weight (MW = 625) and lipophilicity (cLogP = 5.3) of L-796568, attributes that have been associated with this pharmacokinetic issue.<sup>11</sup> Unfortunately, the large

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Figure 1.

Scheme 1. Reagents and conditions: (a) 4-nitrophenethylamine, N,O-bis(trimethylsilylamide), DMSO, 75 °C; (b) carbonyldiimidazole,  $CH_2Cl_2$ , 57% overall; (c) stannous chloride, EtOH, reflux, 100%; (d) RR'NSO<sub>2</sub>Cl, pyridine, 1,2-dichloroethane -35 to 60 °C, 69% for **4d**; (e) KOH, EtOH, 80 °C, 40% for **5d**.

sulfonamide functionalities found in this structural class are necessary for the  $\beta_3$ -AR binding affinity and selectivity. Based on these results a series of sulfamide-containing analogs were investigated to determine where this pharmacophore modification would potentially enhance  $\beta_3$ -AR binding affinity/selectivity, allowing for reductions in molecular weight of the resulting agonists.

Synthetic routes to these sulfamide-based derivatives are detailed in Schemes 1–4. For the chlorophenyl analogs (Scheme 1), reaction of commercially available (R)-3-chlorophenylstyrene oxide with excess 4-nitropheneth-ylamine and reaction with carbonyldiimidazole afforded oxazolidinone **3**. Reduction of the nitro group and reaction with the appropriate sulfamoyl chloride,<sup>12</sup> followed by deprotection provided **5**. The N-methyl derivative **6** was prepared via Eq. 1.



For the corresponding analogs containing the 3-pyridyl 'head group' the ethanolamine core was constructed via alkylation of tosylate 7<sup>13</sup> with 4-nitrophenethylamine and protection of the secondary amine as the *tert*-butylcarbamate (Scheme 2). Conversion of the nitro group to the sulfamide functionality and deprotection of the ethanolamine core afforded analogs 10. Preparation of homologs of 10 in which the phenethylamine portion is replaced by phenoxyethylamine (13) is outlined in Scheme 3. Reaction of 7 with ethanolamine and protection of the amine afforded 11. Incorporation of the phenoxy-based sulfamide substituent proceeded via Mitsunobu coupling with 4-nitrophenol and reduction of the nitro group. Reaction with sulfamoyl chlorides and deprotection afforded targets 13.

Scheme 4 details the synthetic route to  $\beta$ , $\beta$ -dimethylphenethylamine-based derivatives 17. Condensation of hindered amine 14<sup>14</sup> with tosylate 7 could be achieved by heating at 100 °C for 48 h in DMSO. Attempted protection of the amino functionality as the *tert*-butylcar-bamate failed due to steric hindrance. This result suggested that selective reaction with the aniline might be possible, thus obviating the need for protection of the secondary amine. Reduction of the nitro functionality of 15, followed by reaction with sulfamoyl chlorides provided moderate to good yields of the monosulfamides



Scheme 2. Reagents and conditions: (a) 4-nitrophenethylamine, DMSO, 100 °C; (b) (Boc)<sub>2</sub>O, THF, 45% overall; (c) 10% Pd/carbon, H<sub>2</sub>, ethyl acetate, 77%; (d) RR'NSO<sub>2</sub>Cl, pyridine, 1,2-dichloroethane, -35 to 60 °C, 55% for **9d**; (e) tetrabutylammonium fluoride, THF; (f) 4 M HCl in dioxane 87% overall for **10d**.



Scheme 3. Reagents and conditions: (a) ethanolamine, diisopropylethylamine, DMSO,  $80 \,^{\circ}$ C; (b) (Boc)<sub>2</sub>O, THF, 87% overall; (c) 4-hydroxynitrophenol, diisopropylazo-dicarboxylate, triphenylphosphine, THF; (d) 10% Pd/C, ammonium formate, MeOH, 50% overall; (e) RR'NSO<sub>2</sub>Cl, pyridine, 1,2-dichloroethane -35 to  $60\,^{\circ}$ C; (f) tetrabutylammonium fluoride, THF; (g) 4 M HCl in dioxane, 70% overall for 13b.



Scheme 4. Reagents and conditions: (a) diisopropylethylamine, DMSO, 100 °C, 48 h, 63%; (b) 10% Pd/C, ammonium formate, MeOH, 93%; (c) RR'NSO<sub>2</sub>Cl, pyridine, 1,2-dichloroethane –35 to 60 °C, 40% for **16j**; (d) tetrabutylammonium fluoride, THF; (e) HCl, dioxane/MeOH 90% overall for **17j**.

	NRR′	$\beta_3$ -AR EC <sub>50</sub> $\mu$ M (IA)	$\beta_2$ -AR IA @ 30 $\mu$ M	$\beta_1$ -AR IA @ 30 $\mu$ M
2		0.003 (92)	1	1
5a	Dimethylamino	0.299 (75)	6	5
5b	Cyclohexylamino	0.464 (63)	7	10
5c	N-Cyclohexyl-N-methylamino	0.537 (62)	5	6
5d	Piperidinyl	0.617 (84)	5	4
6	Dimethylamino	9.9 (105)	1	2

**Table 1.**  $\beta$ -Adrenergic activities for series **5**<sup>15</sup>

**16**. Deprotection of the alcohol moiety then afforded the desired aminoalcohols **17**.

Early  $\beta_3$ -AR agonist programs identified the 3-chlorophenyl moiety to be a good pharmacophore replacement for the catechol group of the endogenous hormones.<sup>7</sup> As a baseline, sulfamide targets 5 were initially prepared and tested for  $\beta_3$ -AR agonist activity and selectivity versus  $\beta_1$ - and  $\beta_2$ -AR. These analogs would also serve the role of beginning to define the structure-activityrelationship (SAR) around the sulfamide functionality. As mentioned above, the focus of sulfamide substituent SAR was focused on analogs derived from low molecular weight amines. As can be seen from the examples in Table 1, both 2°- and 3°-sulfamide-based analogs containing the 3-chlorophenylethanolamine core were 100-200-fold less potent than 2. Alkylation of the aniline nitrogen of the sulfamide functionality leads to a further order of magnitude reduction in potency (6 vs 5a).

Next, we investigated incorporation of the 3-pyridylethanolamine core found in 2 (Table 2). While this modification does not produce enhanced B<sub>3</sub>-AR functional potency for the dimethylamino (10a) and cyclohexylamino (10b) sulfamide analogs, other 3°-sulfamides such as 10c and 10d exhibit 4–6-fold improvements in potency. Based on the in vitro profile of the piperidinyl analog 10d a series of functionalized piperidine derivatives were investigated. Incorporation of lipophilic substituents into either the 3-, 4-, or 5-positions (10e-g) of the piperidine ring result in further improvements in agonist potency.  $\beta_3$ -AR agonist potencies were now in the 20–30 nM range and selectivities versus  $\beta_1$ - and  $\beta_2$ -AR were >1000-fold. Before exploring further improvements in the in vitro profile of this series, through modifications of the sulfamide functionality, the SAR of the phenylethyl linker region of these compounds was evaluated.

Previous efforts (unpublished) in our laboratories had shown for certain series of  $\beta_3$ -AR agonists that

replacement of the two carbon linker between the secondary amine and phenyl group (attached to the sulfamide functionality in this case) with an ethoxy-based linker provided up to an order of magnitude improvement in potency. This modification was incorporated into the current series, utilizing both a relatively small sterically demanding sulfamide substituent (13a) and the more sterically demanding 4-methylpiperidine derivative 13b. Incorporation of the oxygen atom in the linker leads to a significant loss in potency relative to the ethyllinked homologs (Table 3).

Pitha and co-workers showed that incorporation of a quaternary carbon in the phenethyl linker next to the secondary amine of  $\beta$ -AR agonists can produce dramatic increases in potency.<sup>14</sup> This work was carried out in a rat reticulocyte binding assay (presumably assessing mostly  $\beta_1$ -AR binding) and does not address the potential for such a modification to impact  $\beta_3$ -AR binding. However, extension of this concept to the sulfamide series had a positive impact on  $\beta_3$ -AR agonist activity (Table 4). The better analogs, such as the *cis*-3,5-dimethylpiperdinyl-**17e** and anilino-based **17f** were single-digit nanomolar, full agonists of the human  $\beta_3$ -AR. This structural modification did not enhance  $\beta_1$ - or  $\beta_2$ -AR agonism, with selectivities of >5000-fold for the more potent analogs.

Analogs 17a-f did not significantly increase oxygen consumption (surrogate measure of energy expenditure) in rats when dosed orally. This is presumably the result of their high intrinsic clearances. For example, the rat

Table 3.  $\beta$ -Adrenergic activities for series 13<sup>15</sup>

	NRR′	β <sub>3</sub> -AR EC <sub>50</sub> μM (IA)	β <sub>2</sub> -AR EC <sub>50</sub> μM (IA)	β <sub>1</sub> -AR EC <sub>50</sub> μM (IA)
13a	Dimethylamino	0.943 (96)	4	3
13b	4-Methylpiperidinyl	1.36 (98)	7	5

Table 2.  $\beta$ -Adrenergic activities for series  $10^{15}$ 

	-				
	NRR′	$\beta_3$ -AR EC <sub>50</sub> $\mu$ M (IA)	$\beta_2$ -AR IA @ 30 $\mu$ M	$\beta_1$ -AR IA @ 30 $\mu$ M	
2		0.003 (92)	1	1	_
10a	Dimethylamino	0.265 (84)	11	6	
10b	Cyclohexylamino	1.46 (71)	0	0	
10c	N-Cyclohexyl-N-methylamino	0.180 (67)	16	15	
10d	Piperidinyl	0.099 (77)	6	7	
10e	4-Methylpiperidinyl	0.036 (80)	8	6	
10f	4-Phenylpiperidinyl	0.023 (100)	2	2	
10g	cis-3,5-Dimethylpiperidinyl	0.024 (95)	17	11	

	NRR′	$\beta_3$ -AR EC <sub>50</sub> $\mu$ M (IA)	$\beta_2$ -AR IA @ 30 $\mu$ M	$\beta_1$ -AR IA @ 30 $\mu$ M
2		0.003 (92)	1	1
17a	Dimethylamino	0.038 (98)	7	10
17b	4-Methylpiperidinyl	0.010 (91)	6	12
17c	4-Phenylpiperidinyl	0.047 (106)	9	20
17d	Cyclohexylamino	0.025 (110)	19	17
17e	cis-3,5-Dimethylpiperidinyl	0.004 (89)	11	7
17f	Anilino	0.007 (107)	8	29
17g	2-Methoxyethyl-amino	0.165 (103)	24	28
17h	N-Methylpiperizinyl	0.041 (79)	4	7
17i	N-Benzylpiperizinyl	0.018 (105)	13	46
17j	cis-3,5-Dimethylmorpholinyl	0.008 (85)	8	10

**Table 4.**  $\beta$ -Adrenergic activities for series 17<sup>15</sup>

and human microsomal intrinsic clearances for 17e are 466 and 130 mL/min/kg, respectively. Since the in vitro clearances tended to track with the molecular weight/ lipophilicity of the sulfamide group, a series of analogs were prepared in which polar/ionic functionality was incorporated into the sulfamide. A few representative examples are provided in Table 4 (17g-j). Incorporation of small alkoxy (17g) or tertiary amine (17h) containing sulfamide functionalities lead to significantly reduced clearances, though they also suffered significant reductions in  $\beta_3$ -AR agonist activity. Adding additional lipophilicity (17i) to the tertiary amine of 17h enhances the functional response, but not surprisingly reacquires microsomal lability. Substitution of the 4-methylene substituent of the high potency 3,5-dimethylpiperidinyl analog 17e with an oxygen atom providing morpholine 17j, maintains single-digit nanomolar potency and excellent selectivity. This activity profile for 17j is further confirmed through  $\beta$ -AR binding studies that demonstrate high  $\beta_3$ -AR binding affinity (7 nM), with relatively weak binding to the  $\beta_1$ -AR (9750 nM) and  $\beta_2$ -AR  $(1400 \,\mathrm{nM}).$ 

More importantly, the rat and human microsomal intrinsic clearances of this compound are reduced to 165 and 11 mL/min/kg, respectively. With a molecular weight of 463 and a cLogP = 2.6 there is an expectation that this compound should also not have the oral absorption issues associated with **2**.

Increases in oxygen consumption in the rat has been utilized as a surrogate endpoint for determination of increased energy expenditure produced by  $\beta_3$ -AR agonists.<sup>16</sup> When dosed intraperitoneally at 30 mg/kg, **17j** produces a 38 ± 6% increase in oxygen consumption relative to the control animals.<sup>17</sup> This effect was dose responsive with 10 mg/kg resulting in a 26 ± 3% increase. When dosed orally in rats at 30 mg/kg, **17j** elicits a 35 ± 3% increase in oxygen consumption. That this oral response is similar to that seen from intraperitoneal administration supports the hypothesis that **17j** has good oral absorption.

In conclusion, this study has detailed the discovery of a novel series of highly potent and selective  $\beta_3$ -AR agonists. These compounds have physiochemical properties that are projected to lead to good oral absorption, which is supported by preliminary in vivo efficacy studies.

## **References and notes**

- 1. Ahlquist, R. P. Am. J. Physiol. 1948, 153, 586.
- Lands, A. M.; Arnold, A.; McAuliff, J. P.; Luduena, F. P.; Brown, T. G. Nature 1967, 214, 597.
- 3. Tan, S.; Curtis-Prior, P. B. Int. J. Obes. 1982, 7, 409.
- For reviews of the subject see: (a) Dow, R. L. Exp. Opin. Invest. Drugs 1997, 6, 1811; (b) Weber, A. E. Ann. Rep. Med. Chem. 1998, 33, 193.
- Emorine, L. J.; Marullo, S.; Briend-Sutren, M.-M.; Patey, G.; Devalier-Klutchko, C.; Strosberg, A. D. Science 1989, 245, 1118.
- 6. Liggett, S. Mol. Pharmacol. 1992, 42, 634.
- Mathvink, R. J.; Tolman, J. S.; Chitty, D.; Candelore, M. R.; Cascieri, M. A.; Colwell, L. F., Jr.; Deng, L.; Feeney, W. P.; Forrest, M. J.; Hom, G. J.; MacIntyre, D. E.; Miller, R. R.; Stearns, R. A.; Tota, L.; Wyvratt, M. J.; Fisher, M. H.; Weber, A. E. J. Med. Chem. 2000, 43, 3832, and references cited therein.
- 8. van Baak, M. A.; Hul, G. B. J.; Toubro, S.; Astrup, A.; Gottesdiener, K. M.; DeSmet, M.; Saris, W. H. M. *Clin. Pharmacol. Ther.* **2002**, *71*, 272.
- Larsen, T. M.; Toubro, S.; van Baak, M. A.; Gottesdiener, K. M.; Larson, P.; Saris, W. H. M.; Astrup, A. Am. J. Clin. Nutr. 2002, 76, 780.
- Stearns, R. A.; Miller, R. R.; Tang, W.; Kwei, G. Y.; Tang, F. S.; Mathvink, R. J.; Naylor, E. M.; Chitty, D.; Colandrea, V. J.; Weber, A. E.; Colletti, A. E.; Strauss, J. R.; Keohane, C. A.; Feeney, W. P.; Iliff, S. A.; Chiu, S.-H. L. Drug Metab. Disp. 2002, 30, 771.
- Lipinski, C. A.; Lombardo, F.; Dominy, B. W.; Feeney, P. J. Adv. Drug Del. Rev. 1997, 23, 3.
- Sulfamoyl chlorides were prepared by treating the appropriate sodium sulfamate (synthesized via reported procedure: Wiley, R. A.; Pearson, D. A.; Schmidt, V.; Wesche, S. B.; Roxon, J. J. *J. Med. Chem.* 1983, 26, 1077) with of phosphorus oxychloride (2 equiv) at 80 °C in 1,2-dichloroethane for 18 h, cooling, adding an equal volume of hexanes, filtering, and concentrating the filtrate in vacuo.
- 13. Dow, R. L.; Schneider, S. R. European Patent Application 1138685, 2001; Chem. Abstr. 2001, 135, 288694.
- Milecki, J.; Baker, S. P.; Standifer, K. M.; Ishizu, T.; Chida, Y.; Kusiak, J. W.; Pitha, J. J. Med. Chem. 1987, 30, 1563.
- 15. Agonist activities (EC<sub>50</sub>) were assessed by measuring cAMP levels in CHO cells expressing cloned human  $\beta$ -adrenergic receptors. Intrinsic activities (IA) represent the percentage of the maximal response attained by isoproterenol. NT = Not tested.
- 16. Depocas, F.; Hart, J. S. J. Appl. Physiol. 1957, 10, 388.
- 17. Animals are removed from general housing first thing in the morning (7:00–7:30 am) and are deprived of food and

water for the length of the oxygen consumption measurements. Animals are weighed (310-350 g), marked, and placed into individual activity monitored chambers  $(17'' \times 17'' \times 5'')$ . The system is calibrated and the run started (8:00-8:30 am). Oxygen consumption measurements are made every 10 min for 3 h, at which time the animals are dosed with test compounds (17j dosed as the

monotosylate salt) or vehicle (n=4 each). Oxygen consumption measurements continue for 2 h. Oxygen consumption values associated with periods of high ambulatory activity (>100 counts/10 min) are excluded from all calculations, as well as the first 5 values of the run and the first value after dosing (settle-down periods).