Sesquiterpene Lactones

Enantioselective Total Synthesis of the Highly Oxygenated 1,10-*seco*-Eudesmanolides Eriolanin and Eriolangin**

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Dedicated to Professor Johann Mulzer on the occasion of his 60th birthday

The highly oxygenated 1,10-seco-eudesmanolides eriolanin (1) and eriolangin (2, Scheme 1) isolated from the plant



Scheme 1. Representatives of the 1,10-seco-eudesmanolides.

Eriophyllum lanatum inhibit the growth of the human KB tumor cell line in vitro and additionally display a significant antileukemic activity in vivo in mice.^[1] While several syntheses for racemic $1^{[2,3]}$ and one for racemic 2 have been published,^[2] the absolute configuration of these sesquiterpene lactones was unknown prior to our work. Here we report an efficient enantioselective sultone route^[4] to 1 and 2 that also opens a synthetic access toward the less highly oxygenated, cytotoxic britannilactone derivatives 3 and 4.^[5–7] As the central intermediate, δ -sultone 10 (Scheme 2) was employed, the racemic mixture of which already enabled a short and highly diastereoselective synthesis of the 1,10-*seco*-eudesmanolide ivangulin (5).^[8]

Alcohol 9,^[9] required as the starting material for the enantiomerically pure sultone 10, was available on a multigram scale by catalytic enantioselective transfer hydrogenation^[10] of 2-bromo-1-(2-furyl)ethanone (6)^[11] to 7 (> 98.5 % *ee*

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according to capillary GC), mild basic treatment to give epoxide **8**, and subsequent ring opening with full regioselectivity and complete inversion of configuration^[12] (Scheme 2). By treatment of **9** with β -chloroethanesulfonic acid chlo-



Scheme 2. Sultone route to 1,3-diene 13. a) 0.2 mol% [Cp*RhCl((R, R)-tsdpen)], HCO₂H, Et₃N, 0°C; b) K₂CO₃, MeCN, RT; c) MeCu(CN)Li, Et₂O, -78°C \rightarrow RT, 50% over three steps; d) β -chloroethanesulfonic acid chloride, Et₃N, CH₂Cl₂, RT; e) cat. BHT, EtOAc, 120°C, micro-waves, 81% over two steps; f) 1. MeLi, THF, -78°C, 2. 12, -78°C \rightarrow -20°C, 3. ICH₂MgCl, THF, -78°C \rightarrow RT, 61%; g) LiDBB, THF, -78°C. BHT = 2,6-di-*tert*-butyl-4-methylphenol, Cp* = pentamethylcyclopenta-dienyl, LiDBB = lithium 4,4'-di-*tert*-butylbiphenylide, tsdpen = *N*-(4-tol-uenesulfonyl)-1,2-diphenylethylenediamine.

ride^[13] and triethylamine, a mixture of **10** and a further *exo* sultone isomer was formed in a domino process consisting of elimination, esterification, and intramolecular Diels–Alder reaction, from which pure **10** could be isolated in excellent yield after thermal equilibration.^[8] Conversion of **10** to methylenecyclohexene **13** succeeded by a sequential transformation consisting of elimination, alkoxide-directed 1,6-addition of lithiosilane **12**,^[14,15] and desulfurization with simultaneous methylenation in a one-pot procedure.^[16] In a single synthetic operation, the prefunctions for a γ -lactone were unfolded, activation for 1,4-dioxygenation was created by virtue of the 1,3-diene, and the primary hydroxy group was liberated for side-chain elongation.

For side-chain elongation, diol **13** was first bissilylated, and then the primary hydroxy group was selectively deprotected (Scheme 3). After conversion^[17] of the resulting alcohol **15** to iodide **16**, the required C₂ unit was attached by alkylation with dimethyl malonate in the presence of proazaphosphatrane **17**^[18,19] and demethoxycarbonylation^[8,20] of **18**. An intramolecular protocol was eventually decisive for the efficient generation of the enediol fragment of the target molecules. Carboxylic acid **20**, obtained after saponification of **19**, was treated successively with bis(*sym*-collidine)iodine(i) hexafluorophosphate^[21] followed by silver acetate in dimethylformamide^[22] in a one-pot procedure, whereupon the formyloxy ε -lactone **21**^[23] was isolated as the major product. Substitution of the formylation by a reduction^[24,25] of the intermediate allyl iodide should allow concise access to

Communications



Scheme 3. Synthesis of the completely oxygenated basic skeleton 28. a) TBSCl, imidazole, DMAP, DMF, RT, 99%; b) TBAF, THF, 0°C, 81% 15 + 17% 13; c) I_2 , Ph₃P, imidazole, THF, MeCN, $-20^{\circ}C \rightarrow RT$, 84%; d) 17, dimethyl malonate, MeCN, RT, 91%; e) PhSH, K_2CO_3 , DMF, 90°C, 89%; f) KOH, MeOH, H₂O, reflux, 100%; g) 1. I(col)₂PF₆, PhMe, 0°C, 2. AgOAc, DMF, PhMe, RT, 67% 21 + 15% 22; h) zinc dust, HOAc, H₂O, THF, 0°C $\rightarrow RT$, 86%; i) LiAlH₄, Et₂O, 0°C; j) LiBH₄, Et₂O, $-10^{\circ}C$, 91% over two steps; k) TrCl, DMAP, pyridine, CH₂Cl₂, RT, 91%; l) Dess–Martin periodinane, pyridine, CH₂Cl₂, RT, 99%; m) TBAF, HOAc, THF, RT, 96%; n) Red-Al, CH₂Cl₂, PhMe, $-20^{\circ}C \rightarrow RT$, 90%; o) 1. TBAF, MS 4 Å, THF, reflux, 2. KF, H₂O₂, NAHCO₃, THF, MeOH, reflux, 99%. col = *sym*-collidine, DMAP = 4-(*N*,*N*-dimethylamino)pyridine, MS = molecular sieves, Red-Al = sodium bis (2-methoxyethoxy)aluminum hydride, TBAF = tetrabutylammonium fluoride, TBS = *tert*-butyldimethylsilyl, Tr = triphenylmethyl.

britannilactones 3 and 4. Iodolactone 22, which was formed in small amounts next to **21**, can be reductively eliminated^[26] to return 20. The undesired configuration at C6 (eudesmane numbering) in 21 set up in a completely stereoselective fashion during the sequential iodolactonization/allyl formiate generation from 20 was subsequently corrected by an oxidation/reduction strategy.^[27] Reduction of diester 21 with lithium aluminum hydride to give a hydroxy lactol^[28] and further reduction with lithium borohydride afforded triol 23. Chemoselective tritylation of the two primary hydroxy groups $(\rightarrow 24)$, Dess-Martin oxidation^[29] $(\rightarrow 25)$, and mild desilylation^[30] led to β -hydroxy ketone **26**. Hydroxy-directed^[31] reduction of the latter with the sodium aluminum dihydride Red-Al furnished the desired 6α allyl alcohol 27 with excellent diastereoselectivity.^[32] After Tamao-Fleming oxidation,^[33,34] the completely oxygenated skeleton of the target molecules with correct configuration at all stereogenic centers was finally available in the form of triol 28.

The final stage of the synthesis was initiated with a chemoselective oxidation^[35] of triol **28** to give hydroxy γ -lactone **29** (Scheme 4). After protection of the secondary hydroxy group, a one-step α -methylenation of lactone **30** succeeded with sodium hydride and paraformaldehyde,^[36] and

following desilylation, lactone 31 was isolated in good overall vield. Preparation^[2] of methacrylate **32** as well as detritylation to give 1 proceeded uneventfully and delivered (-)-eriolanin, which proved to be identical to the natural product by comparison of optical rotation data.^[37] Thus, our synthesis of 1 also clarifies the previously unknown absolute configuration of this sesquiterpene lactone, since the absolute configuration of 9 was unambiguously established.^[9] In addition, an X-ray diffraction analysis of our synthetic product 1 provided a further independent proof of the absolute configuration by anomalous X-ray scattering.^[38] Using a modified Yamaguchi esterification,^[39] **31** could also be transformed smoothly to angelate 33 without Z/E isomerization. Deblocking to give 2 delivered (-)-eriolangin, which also turned out to be identical to the natural product by comparison of optical rotation data.^[40]

Due to the sultone strategy applied, our enantioselective route to the 1,10-seco-eudesmanolides **1** and **2** requires only 26 steps from 2-bromo-1-(2-furyl)ethanone (**6**). Average yields of 87% for **1** and 86% for **2** highlight the efficacy of the route reported. Moreover, the selective manipulation of the diverse hydroxy groups on the 1,10-seco-eudesmanolide framework possible here offers great flexibility with respect





Scheme 4. Final steps of the synthesis of (-)-eriolanin (1) and (-)-eriolangin (2). a) BAIB, cat. TEMPO, CH_2CI_2 , RT, 75%; b) TMSCI, imidazole, CH_2CI_2 , RT, 96%; c) NaH, paraformaldehyde, THF, 100°C (sealed tube); d) TBAF, THF, 0°C, 61% over two steps; e) methacrylic acid anhydride, Et₃N, DMAP, THF, 0°C \rightarrow RT, 85%; f) 1. angelic acid, 2,4,6-trichlorobenzoyl chloride, Et₃N, PhMe, RT, 2. **31**, 100°C, 60%; g) cat. *p*-TsOH, MeOH, RT, 97% **1** from **32**, 85% **2** from **33**. BAIB = bisacetoxyiodobenzene, TEMPO = 2,2,6,6-tetramethylpiperidin-1-oxyl (free radical), TMS = trimethylsilyl, *p*-TsOH = *p*-toluenesulfonic acid.

to the assembly of synthetic analogues. A synthesis of the britannilactone derivatives **3** and **4** is in preparation.

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- S. M. Kupchan, R. L. Baxter, C.-K. Chiang, C. J. Gilmore, R. F. Bryan, J. Chem. Soc. Chem. Commun. 1973, 842–843.
- [2] P. A. Grieco, T. Oguri, S. Gilman, J. Am. Chem. Soc. 1980, 102, 5886–5891.
- [3] a) M. R. Roberts, R. H. Schlessinger, J. Am. Chem. Soc. 1981, 103, 724-725; b) T. Wakamatsu, N. Miyachi, F. Ozaki, M. Shibasaki, Y. Ban, Tetrahedron Lett. 1988, 29, 3829-3832.
- [4] Review on sultone chemistry: P. Metz, J. Prakt. Chem. 1998, 340, 1-10.
- [5] Isolation: a) B.-N. Zhou, N.-S. Bai, L.-Z. Ling, G. A. Cordell, *Phytochemistry* **1993**, *34*, 249–252; b) F. Jeske, S. Huneck, J. Jakupovic, *Phytochemistry* **1993**, *34*, 1647–1649.
- [6] Compounds 3 and 4 effect phosphorylation of the antiapoptosis protein Bcl-2 and induce apoptosis in several cancer cell lines; for 4 cell-cycle arrest at the G2/M phase as well as polymerization of microtubules was proven: C.-T. Ho, M. Rafi, R. S. Dipaola, G. Ghai, R. T. Rosen, N. Bai, US 6,627,623 B2, 2003.
- [7] For the relative configuration of 3 and 4 shown in Scheme 1, see ref. [5b] as well as the crystal structures of 3 depicted in: a) A.-R.

Han, W. Mar, E.-K. Seo, *Nat. Prod. Sci.* **2003**, *9*, 28–30; b) S. Liu, H. Liu, W. Yan, L. Zhang, N. Bai, C.-T. Ho, *Bioorg. Med. Chem. Lett.* **2004**, *14*, 1101–1104.

- [8] P. Metz, J. Stölting, M. Läge, B. Krebs, Angew. Chem. 1994, 106, 2275–2276; Angew. Chem. Int. Ed. Engl. 1994, 33, 2195–2197.
- [9] A. Bierstedt, J. Stölting, R. Fröhlich, P. Metz, *Tetrahedron: Asymmetry* 2001, 12, 3399–3407.
- [10] T. Hamada, T. Torii, K. Izawa, R. Noyori, T. Ikariya, Org. Lett. 2002, 4, 4373-4376.
- [11] J. Dubac, A. Gaset, M. Maraval, Synth. Commun. 1991, 21, 11– 16.
- [12] This unusual regioselectivity of epoxide opening was already reported for racemic 8: B. Alcaide, P. Areces, E. Borredon, C. Biurrun, J. P. Castells, J. Plumet, *Heterocycles* 1990, 31, 1997– 2002. We found that the stereochemical course is strongly dependent on the nature of the methyl nucleophile. With methylmagnesium bromide, again regioisomer 9 is preferentially formed; however, the reaction proceeds with extensive racemization.
- [13] A. A. Goldberg, J. Chem. Soc. 1945, 464-467.
- [14] Thioether 11 was prepared by radical addition of PhSH to Me₂PhSiCH=CH₂ at 100°C (87%); see also: C.-N. Hsiao, H. Shechter, *Tetrahedron Lett.* 1982, 23, 1963–1966.
- [15] T. Cohen, M. Bhupathy, Acc. Chem. Res. 1989, 22, 152-161.
- [16] B. Plietker, D. Seng, R. Fröhlich, P. Metz, Eur. J. Org. Chem. 2001, 3669–3676.
- [17] P. J. Garegg, B. Samuelsson, J. Chem. Soc. Chem. Commun. 1979, 978–980.
- [18] Review: J. G. Verkade, P. B. Kisanga, *Tetrahedron* 2003, 59, 7819–7858.
- [19] S. Arumugam, D. McLeod, J. G. Verkade, J. Org. Chem. 1998, 63, 3677–3679.
- [20] E. Keinan, D. Eren, J. Org. Chem. 1986, 51, 3165-3169.
- [21] B. Simonot, G. Rousseau, J. Org. Chem. 1994, 59, 5912-5919.
- [22] A. G. Martinez, A. C. Villalobos, M. O. Ruiz, Synthesis 1988, 58– 60.
- [23] Apparently the intermediate allylcarbenium ion is trapped by the solvent DMF as the nucleophile to generate an immonium ion, which is converted to the formate during aqueous workup.
- [24] F. Homsi, G. Rousseau, J. Org. Chem. 1998, 63, 5255-5258.
- [25] W. P. Neumann, *Synthesis* **1987**, 665–683.
- [26] B. Deguin, J.-C. Florent, C. Monneret, J. Org. Chem. 1991, 56, 405-411.
- [27] Experiments toward Mitsunobu inversion at C6 in **24** were not successful.
- [28] Increasing the reaction temperature in order to achieve complete reduction caused desilylation of the secondary silyl ether.
- [29] D. B. Dess, J. C. Martin, J. Am. Chem. Soc. 1991, 113, 7277-7287.
- [30] a) J. S. Debenham, R. Rodebaugh, B. Fraser-Reid, J. Org. Chem.
 1997, 62, 4591 4600; b) D. L. Boger, R. M. Borzilleri, S. Nukui, R. T. Beresis, J. Org. Chem. 1997, 62, 4721 – 4736.
- [31] A. H. Hoveyda, D. A. Evans, G. C. Fu, Chem. Rev. 1993, 93, 1307-1370.
- [32] In addition to **27** the C6-epimeric alcohol was isolated in 6% yield.
- [33] Review: I. Fleming, Chemtracts: Org. Chem. 1996, 9, 1-64.
- [34] a) H.-J. Knölker, G. Wanzl, Synlett 1995, 378-382; b) Addition of molecular sieves 4 Å in the first step caused a significant increase in yield.
- [35] T. M. Hansen, G. J. Florence, P. Lugo-Mas, J. Chen, J. N. Abrams, C. J. Forsyth, *Tetrahedron Lett.* **2003**, *44*, 57–59.
- [36] B. Noya, M. D. Paredes, L. Ozores, R. Alonso, J. Org. Chem. 2000, 65, 5960-5968.
- [37] Synthetic 1: $[a]_{D}^{25} = -88.6$ (c = 1.0 in CHCl₃); natural 1:^[1] $[a]_{D}^{25} = -93$ (CHCl₃).

Communications

- [38] Crystal dimensions $0.25 \times 0.20 \times 0.20$ mm³, orthorhombic, space group $P2_12_12_1$ (No. 19), a = 8.251(1), b = 10.895(1), c = 20.845(2) Å, V = 1873.9(3) Å³, $\rho_{calcd} = 1.242$ g cm⁻³, Cu_{Ka} radiation, $\lambda = 1.54178$ Å, $\omega/2\theta$ scans, T = 233 K, 4186 reflections measured, 3823 independent ($R_{int} = 0.031$), of which 3369 were observed [$I \ge 2\sigma(I)$], $\mu = 7.58$ cm⁻¹, empirical absorption correction (0.833 $\le T \le 0.863$), Z = 4, 230 refined parameters, hydrogen atoms calculated and refined as riding atoms, R = 0.041, $wR^2 = 0.106$, largest difference peak and hole 0.15/-0.20 eÅ⁻³, Flack parameter 0.04(19). Programs used: EXPRESS, MolEN, SHELXS-97, SHELXL-97, SCHAKAL. CCDC-239240 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/ conts/retrieving.html (or from the Cambridge Crystallographic Data Centre, 12, Union Road, Cambridge CB21EZ, UK; fax: (+44)1223-336-033; or deposit@ccdc.cam.ac.uk).
- [39] B. Hartmann, A. M. Kanazawa, J. P. Deprés, A. E. Greene, *Tetrahedron Lett.* **1991**, *32*, 5077–5080.
- [40] Synthetic **2**: $[a]_{D}^{25} = -87.5$ (c = 1.05 in CHCl₃); natural **2**:^[1] $[a]_{D}^{25} = -91$ (CHCl₃).