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ndisin and veraguensin isoxazole derivatives

SI= 36-161



1	Design and synthesis of a new series of 3,5-disubstituted isoxazoles active
2	against Trypanosoma cruzi and Leishmania amazonensis
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12	
13	Abstract: Chagas disease and leishmaniasis are neglected tropical diseases (NTDs)
14	endemic in developing countries. Although there are drugs available for their treatment,
15	efforts on finding new efficacious therapies are continuous. The natural lignans
16	grandisin (1) and veraguensin (2) show activity against trypomastigote T. cruzi and
17	their scaffold has been used as inspiration to design new derivatives with improved
18	potency and chemical properties. We describe here the planning and microwave-
19	irradiated synthesis of 26 isoxazole derivatives based on the structure of the lignans 1
20	and 2. In addition, the in vitro evaluation against culture trypomastigotes and
21	intracellular amastigotes of T. cruzi and intracellular amastigotes of L. amazonensis
22	and L. infantum is reported. Among the synthesized derivatives, compounds 17 (IC_{50} =
23	5.26 μ M for <i>T. cruzi</i>), 29 (IC ₅₀ = 1.74 μ M for <i>T. cruzi</i>) and 31 (IC ₅₀ = 1.13 μ M for <i>T. cruzi</i>
24	and IC ₅₀ = 5.08 μ M for <i>L. amazonensis</i>) were the most active and were also evaluated
25	against recombinant trypanothione reductase of T. cruzi in a preliminary study of their
26	mechanism of action.

Keywords: natural product analogues; drug design; isoxazole; *Trypanosoma cruzi*;
 Leishmania amazonensis; trypanothione reductase.

Abbreviations: %GI, percentage of growth inhibition; AB, amphotericin B; BZN,
benznidazol; logP, partition coefficient octanol/water; MW, microwave; ND, not
determined due to absence of trypanocidal or leishmanicidal activity; NTD, neglected
tropical disease; SI, selectivity index; (r)TR, (recombinant) trypanothione reductase.

1 **1. Introduction**

2 Chagas disease and leishmaniasis are neglected tropical diseases that affect 3 more than 15 million people worldwide, representing a major public health issue especially in Latin American countries [1–3]. Benznidazole and nifurtimox are the only 4 5 drugs available for treatment of Chagas disease, while pentavalent antimonials, amphotericin B, miltefosine, paromomycin and pentamidine are drugs for leishmaniasis 6 7 treatment. There are drawbacks associated with these drugs, such as high cost, 8 toxicity and limited efficacy resulting in high levels of treatment interruption. Resistant 9 strains of *T. cruzi* and *Leishmania* spp. are also emerging and might compromise even 10 more the limited therapeutic options [4-8].

11 Through the centuries, biodiversity has been a useful source of medicinal 12 therapies. There are reports of traditional use of plants dating from more than two thousand years ago and, currently, about 39% of the drugs on the market are natural 13 14 products or have a chemical structure directly based on natural products [9-12]. Several natural products as flavonoids, terpenes, quinones and lignans have been 15 16 tested against trypanosomatid parasites [13-15]. In spite of some identified hits, there are still limitations associated to them, especially regarding the isolation of larger 17 quantities to conduct further biological investigations [16]. On the other hand, 18 secondary metabolites obtained from natural sources represent a library of diverse 19 20 chemical features that can give insights on the development of new chemical entities to 21 treat parasitic diseases [16,17].

The development of efficient drugs for treatment of Chagas disease and leishmaniasis depends on the identification of new lead compounds [18–20]. Grandisin (1) and veraguensin (2) are two tetrahydrofuran lignans isolated from *Virola surinamensis* twigs and *Piper solmsianum* inflorescences that have been described as having promising trypanocidal activity (IC_{50} = 3.7 µM and 2.3 µM, respectively) against the *T. cruzi* trypomastigote stage (Figure 1) [21,22].

28 Attempts on the total synthesis of lignans 1 and 2 were successful and exemplify 29 the importance of organic synthesis to the field of natural products [23-26]. Likewise, the design of analogues applying distinct molecular modification strategies is a 30 valuable tool, and can be used to overcome problems in isolation, physicochemical 31 32 properties or to optimize the bioactivity of known natural product hits [27-31]. Our research group has been working on different compounds based on the grandisin and 33 veraguensin scaffold. Between them, non-rigid 1,4-diaryl-1,4-diol analogues were 34 35 synthesized with the same methoxylation pattern of compounds 1 and 2, and displayed IC₅₀ values as low as 10 µM on *T. cruzi* trypomastigotes [32]. In another work, cyclic 36

derivatives obtained through Michael addition-carbocyclization were planned aiming to
increase water solubility and also yielded compounds with good trypanocidal activity
[33]. These results confirm the potential of the tetrahydrofuran lignans scaffold and
encourage us to continue our research for bioactive compounds related to their
structure.

6 In this work we describe the planning, synthesis and a preliminary evaluation of 7 the structure-activity relationship of a set of isoxazole derivatives structurally related to the lignans 1 and 2 (Figure 1). The choice of the isoxazole ring as a bioisosteric 8 replacement is justified by its higher water solubility, what can improve biological and 9 chemical properties. The design of the compounds also relied on molecular 10 simplification, once the chiral tetrahydrofuran ring in compounds 1 and 2 gives place to 11 the aromatic isoxazole ring. Both strategies have been widely used in medicinal 12 chemistry on the study of molecules, affording new hits. 13

As the synthesis of compounds containing different substituents linked to the isoxazole ring is feasible and straightforward, we prepared compounds substituted with one and two aromatic rings bearing different R groups via a microwave-assisted methodology.

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Fig 1. Structure of trypanocidal tetrahydrofuran lignans grandisin (1), veraguensin (2)
 and proposed derivatives.

Thereafter, the derivatives were assessed against *T. cruzi* and *L. amazonensis* intracellular amastigotes. The chemical variability present in the first synthesized series allowed us to identify sites for punctual modifications that enhanced bioactivity and selectivity, providing hints on the structure-activity relationship. The inhibitory potential of the compounds on recombinant trypanothione reductase (rTR) was also evaluated, due to its relevance as a target for anti-trypanosomatid drug design and its presence in both genera *Trypanosoma* and *Leishmania*.

2. Results and Discussion 1

2.1. Chemistry 2

3 Over the last years, microwave irradiation has been proved a powerful tool in 4 synthetic organic chemistry, allowing the preparation of compounds more efficiently in comparison to traditional methods. Although there is a range of different applications 5 6 for its use, one of the main advantages of microwave irradiation is the possibility of 7 preparing potentially bioactive molecules within minutes [34-37].

8 Based on the methodology described by Himo et al. (2005) [38], we focused on synthesizing 3,5-disubstituted isoxazole derivatives via 1,3-dipolar cycloaddition, under 9 10 microwave irradiation in order to increase the speed of the synthesis (Scheme 1).





Scheme 1. Reagents and conditions: i) NCS, DMF, MW: 30 °C, 150 W, 1 min; ii) 13 CuSO₄, sodium ascorbate, NaHCO₃, MW: 30 °C, 150 W, 10 min. 14

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Our initial efforts synthesizing 5-hydroxymethyl-3-phenylisoxazole (6) have 16 shown that after 1 minute of irradiation (at 30 °C) it was possible to visualize by TLC 17 18 analysis the total consumption of the starting material, leading to a more apolar product suggested to be the imidoyl chloride (Scheme 1, intermediate 4). Subsequently, when 19 20 the imidoyl chloride was irradiated with an alkyne and the final product was purified, we could observe that the yield depends on the total time of irradiation (Table 1). 21

Table 1. Time-based yield optimization for 5-hydroxymethyl-3-phenylisoxazole (6) 22 23 synthesis.

Entry	R ¹	R ²	Time ^a (min)	Yield (%)
i	-H	-CH ₂ OH	1	37
ii	-		3	35
iii	-		10	77
iv	-		15	74

24 a. After addition of the alkyne. All the experiments were performed at 30 °C and the maximum potency was set to 150 25 W.

26 The highest yield was obtained when the reaction was further irradiated for 10 27 minutes (Table 1, entry 6iii). As irradiation for 15 minutes did not enhance the yield of

the desired product, 10 minutes was standardized for the synthesis of a series of 1 2 isoxazole derivatives 6-20 (Table 2). Furthermore, we compared the yield of the products obtained via the microwave-assisted reaction (20-77%) with those obtained 3 via the reaction under conventional conditions [38] (18-72%) and it was not possible to 4 5 observe much difference besides a significant decrease in reaction time.

6

Table 2. Isoxazole derivatives initially synthetized and their yields (%)

			R ¹		
Compound	R ¹	R ²	R ³	Yield (%) ^a	Yield MW ^b (%)
6	-H	-H	-CH ₂ OH	72	77
7	-OCH ₃	-H	-CH ₂ OH	52	56
8	-OCH ₃	-OCH ₃	-CH ₂ OH	50	55
9	-F	-H	-CH ₂ OH	47	38
10	-NO ₂	-H	-CH ₂ OH	36	32
11	-OCH	H ₂ O-	-CH ₂ OH	63	56
12	-H	-H	-C ₆ H ₄ -4-CH ₂ OH	47	52
13	-OCH ₃	-H	-C ₆ H ₄ -4-CH ₂ OH	45	52
14	-OCH ₃	-OCH ₃	-C ₆ H ₄ -4-CH ₂ OH	51	51
15	-F	-Н	-C ₆ H ₄ -4-CH ₂ OH	42	40
16	-NO ₂	-H	-C ₆ H ₄ -4-CH ₂ OH	22	20
17	-OCH	H ₂ O-	-C ₆ H ₄ -4-CH ₂ OH	18	22
18	-H	-н	$-C_6H_4$ -4-NH ₂	49	50
19	-Н	-Н	(2,3,4,6-tetracetyl-α-D-	40	54
			glucopyranose-1-yl)methyl		
20	-0CH	H ₂ O-	(2,3,4,6-tetracetyl-α-D-	46	48
			glucopyranose-1-yl)methyl		

N[∙]O ⊥ → B³

7 a. Yield obtained in a non-irradiated reaction; i) NCS, DFM, rt.; ii) CuSO₄, sodium ascorbate, NaHCO₃, alkyne, 8 H₂O/tBuOH, rt.

9 b. All microwave experiments were performed at 30 °C and the maximum potency was set to 150 W.

10

Additionally, compound 6 was oxidized with Jones reagent, yielding compound 11 21 (Scheme 2), and the derivative 22 was synthesized from previously obtained 12 compound 7 through protection with tosyl chloride (compound 7a) followed by reaction 13 with sodium azide. Also, compounds 19 and 20 were synthesized from a glycosidic 14 alkyne, and further deprotected leading to 23 and 24 (Scheme 3) [39]. These further 15

modifications were planned to increase the diversity related to functional groups and 1 physicochemical properties. The ionized moieties in compounds 21 and 22 might be 2 subject to ionic interactions with molecular targets while the monosaccharide unit 3 present in derivatives 23 and 24 led to the most hydro soluble compounds in the series. 4 Carbohydrates are also known for acting on cellular signaling in both parasites and 5 humans [40-43], and glycosides for being responsible for different biological activities, 6 7 such as antiviral [44,45], antiparasitic [46,47], antibacterial [48-50] and cytotoxic 8 [51,52].



Scheme 2. Reagents and conditions: i) Jones reagent 8N, acetone, 30 min, 0 °C; ii)
 TsCl, Et₃N, K₂CO₃, H₂O/CH₂Cl₂, 2 h, rt.; iii) NaN₃, DMF, MW, 7 min, 70 °C.

11 TsCl, Et₃N, K₂CO₃, H₂O/CH₂Cl₂, 2 h, rt.; iii) NaN₃, DMF, MW, 7 min, 70 °C. 12



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Scheme 3. Reagents and conditions: i) NaOMe, MeOH, pH= 9, 15 min, rt.

All synthetized compounds were purified by column chromatography and characterized by nuclear magnetic resonance (¹H, ¹³C NMR) and mass spectrometry (ESI-TOF). In addition, IR spectroscopy was employed in the characterization when necessary. The spectral data is available as supplementary material.

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1 **2.2.** Biologic investigation and structural optimization.

2 2.2.1. Trypanocidal evaluation

T. cruzi presents a complex life cycle and a wide range of mammalian hosts. Its trypomastigote stage is the infective form both for triatomine vectors and mammals, while its intracellular amastigote stage is the main form responsible for the disease [53].

Considering the importance of the amastigote stage for the pathogenesis of Chagas disease, compounds **6-24** were firstly screened (at 100 μ M) against this life stage of the parasite (Tulahuen strain). The compounds that led to a percentage of growth inhibition (%GI) >50% were assayed in different concentrations to determine the half maximal inhibitory concentration (IC₅₀). The cytotoxicity to THP-1 cells (CC₅₀) was also assessed to calculate the selectivity index (SI).

The results showed that most of the compounds were active against intracellular amastigotes to some extent, except for compounds **8** and **21** (Table 3). These results allowed us to set the synthetic direction for obtaining new improved derivatives that will be discussed in the session 2.2.2.

16	be discussed in the session 2.2.2.
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Compound	%GI (100 μM)	IC ₅₀ (μΜ)	СС ₅₀ (μМ)	SI
6	7.57 (±0.06)	>100	ND	-
7	9.32 (±0.49)	>100	ND	-
8	0.00 (±0.00)	ND	ND	-
9	8.27 (±0.54)	>100	ND	-
10	12.17 (±0.86)	>100	ND	-
11	1.95 (±0.06)	>100	ND	-
12	7.01 (±1.62)	>100	ND	-
13	5.77 (±0.06)	>100	ND	-
14	14.15 (±1,02)	>100	ND	-
15	10.57 (±1.41)	>100	ND	-
16	15.86 (±1.49)	>100	ND	-
17	65.04(±4.55)	5.26 (±0.96)	70.18 (±14.94)	13.3
18	72.97 (±2.58)	90.39 (±14.49)	407.2 (±88.2)	4.5
19	1.16 (±1.00)	>100	ND	-
20	3.15 (±1.13)	>100	ND	-
21	0.00 (±0.00)	ND	ND	-
22	60.35 (±4.64)	ND	<15.6	-
23	7.30 (±0.40)	>100	ND	-
24	2.93 (±0.32)	>100	ND	-
BZN 20 μΜ	85.94 (±1.29)	10.18 (±0.3)	> 500	> 49.

Table 3. Trypanocidal activity of compounds **6-24** against intracellular amastigotes

The results are averages ± SD of triplicates. Compounds 17-18 were tested in triplicate in two independent
 experiments.

In the series substituted with only one aromatic ring (ring A, compounds 6-11), a para substitution on the aromatic ring led to a slight increase in activity when compared with the unsubstituted compound 6. On the other hand, a 3,4-substitution on the A ring led to a loss of activity (compound 8) or to a 3-fold decrease in activity (compound 11) in this series.

In the series substituted with two aromatic rings (rings A and B, compounds 1218), except for compound 13, a substituent group on the A ring seems to contribute to
an increase in biological activity in comparison with compound 12. When the
methylenehydroxy group in the B ring is replaced by a primary amine (compound 18),
the activity also increases. These findings are summarized in Figure 2. Compounds 17

- and 18 were the most promising identified among compounds 6-24, with IC_{50} values of
- 2 5.26 μ M and 90.39 μ M, respectively.

3



4 5 Fig 2. Preliminary SAR for derivatives 6-24. 6 7 Regarding compound 17, the methylenedioxy group linked to the aromatic ring A seems to be important for the activity. It was the A ring substituent that led to a higher 8 increase in activity in comparison with other substituents in compounds 12-16. Also, 9 10 the replacement of the methylenedioxy group in compound 17 by two methoxy groups (compound 14), which maintains similar chemical properties, modifies the conformation 11 of the molecule due to a steric clash between the aromatic substituents, and this might 12 be the cause of the lower activity of compound 14 (Figure 3). 13



Fig 3. 3D shape of compounds 17 (a) and 14 (b) generated using Marvin Suite (version
 16.4.25.0, 2016, http://chemaxon.com/).

1 Compounds 19-20 and 23-24 displayed only a small inhibition of growth of T. 2 cruzi. At the first experiment, compound 22 showed a percentage of growth inhibition similar to compound 17, but when it was assayed again one week later, the activity was 3 lost. Mass spectrometry analysis (HRMS ESI+/TOF) of the stock solution (Supporting 4 Information, Figure S1) revealed the absence of a peak relative to the cationized 5 molecule or other adduct, suggesting that compound 22 degraded in the test solution. 6 7 However, two relevant peaks are observed at m/z 247.1470 and m/z 188.0713. The 8 first one might be relative to the oxidized compound 22 (calculated for $C_{11}H_{11}N_4O_3^+$: 9 247.0826) and the second peak shows that oxidation is likely to have occurred at the 10 azide moiety, since it can be attributed fragment 4to the methoxyphenylisoxazolylmethylene (calculated for $C_{11}H_{10}NO_2^+$: 188.0706). 11

12 **2.2.2.** Synthesis of new derivatives

For a better understanding of the properties responsible for the activity of compounds **17** and **18**, derivatives **25-30** were synthetized using the methodology already described in session 2.1. Otherwise, compound **31** was obtained directly from **17** through reaction with thionyl chloride, under N_2 atmosphere, in excellent yield (94%).

Compounds **25-27** were planned to verify the importance of the amine group for compound **18** activity. Moreover, compounds **28-31** were synthesized to understand the role of the *para*-substituted ring B for activity. The structure of derivatives **25-31** together with results of the trypanocidal evaluation are shown in Table 4.

			R1	\mathbb{R}^2	R ⁴	Y	
Compound	R ¹	R ²	R^4	%GI (100 μM)	IC ₅₀ (μΜ)	СС ₅₀ (μМ)	SI
25	-H	-H	-H	29.40 (±2.21)	>100	ND	-
26	-H	-H	-NHAc	0.00 (±0.00)	ND	ND	-
27	-H	-H	-NHBoc	0.00 (±0.00)	ND	ND	-
28	-0C	H ₂ O-	-CH ₂ OAc	47.59 (±4.00)	23.21 (±3.19)	>500	>21.5
29	-0C	H ₂ O-	-CH ₃	81.65 (±3.53)	1.74 (±0.41)	81.32 (±9.74)	46.7
30	-0C	H ₂ O-	-H	64.77 (±3.54)	23.02 (±6.94)	>500	>21.7
31	-0C	H ₂ O-	-CH ₂ CI	95.75 (±0.31)	1.13 (±0.39)	181.8 (±25.9)	160.9
BZN 20 μΜ				85.94 (±1.29)	10.18 (±0.3)	> 500	> 49.1

 Table 4. Trypanocidal activity of compounds 25-31 against intracellular amastigotes.

The results are averages ± SD of two independent experiments run in triplicate.

1 Considering compound **18**, acetylation and carbamatization of the primary amine 2 lead to a loss of trypanocidal activity in compounds **26** and **27**, respectively. In both 3 derivatives, the nitrogen lone pair is delocalized into a carbonyl, resulting in a lack of 4 basicity. Besides, a reduction in activity is observed when the amine group is removed 5 (compound **25**), suggesting therefore that a basic group might play an important role 6 for the bioactivity of compound **18** in *T. cruzi* (Figure 4).

When compound **17** was acetylated (compound **28**) or had its methylenehydroxy 7 8 group removed (compound 30) the activity had a fourfold decrease. However, an 9 improvement of activity and selectivity was observed in derivatives where the hydroxyl 10 was removed (compound 29) or replaced by a chlorine atom (compound 31) (Figure 4). 11 The calculation of some chemical properties using Marvin (version 16.4.25.0, 2016, http://chemaxon.com/, Supporting Information, Table S1) showed that compounds 17, 12 29 and 31 have a similar surface area, while compound 28 has a larger and compound 13 14 **30** a smaller surface area. Previously, we observed that the presence of two benzene rings linked to the isoxazole ring favored the inhibition of T. cruzi growth, and now our 15 results indicate that there is an ideal range of molecular size in order to optimize 16 17 bioactivity in this class of compounds.



Replacement for -H or -CH₂OAc
 ↓ activity and ↑ selectivity
 Replacement for -CH₃ ou -CH₂Cl ↑ activity and selectivity

19 **Fig 4.** Compounds **25-30** contribution for SAR of 3,5-disubstituted isoxazoles.

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An important aspect of compounds **17**, **29** and **31** is their potency compared to the reference drug benznidazole (Figure 5). The amastigote stage of *T. cruzi* is predominately found during the chronic course of Chagas disease, which cannot be effectively treated by benznidazole or other drugs. Therefore, these hits will be further investigated aiming to design new efficacious and selective molecules to treat and control the mortality and morbidity of Chagas disease.



1 2



3 4

5 Finally, although 6 compounds had high activity against *T. cruzi* amastigotes (IC_{50} <100 6 μ M, compounds **17**, **18**, **28-31**), none of them were active against the trypomastigote 7 stage of the parasite (Tulahuen) at the tested concentration. It is not uncommon for 8 compounds to have different potencies against different *T. cruzi* life stages [54–56], 9 implying that morphological changes experienced by the parasite during its life cycle 10 could affect its interactions with micromolecules.

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2.2.3. Leishmanicidal evaluation

Alongside Chagas disease, leishmaniasis is also a NTD that affects millions of 12 people and lacks options for efficient treatment. In the New World, the Leishmania 13 14 mexicana and Leishmania braziliensis complexes comprise 7 different species, 15 whereas L. braziliensis and L. guyanensis are the main species responsible for 16 mucocutaneous leishmaniasis; L. mexicana, L. amazonensis, L. venezuelensis, L. panamensis and L. peruviana are causative agents for cutaneous leishmaniasis. Other 17 5 species are commonly found in the Old World, being L. donovani and L. infantum, 18 responsible for causing visceral leishmaniasis and L. major, L. tropica and L. aethiopica 19 20 etiological agents of cutaneous leishmaniasis [57]. As we obtained molecules with an 21 interesting bioactivity against T. cruzi, we found valuable to test compounds 6-31 also 22 against intracellular amastigotes of L. amazonensis (MHOM/BR/77/LTB0016). Similarly to the trypanocidal screening, the IC_{50} and CC_{50} (THP-1 cells) values were only 23 determined for compounds that led to a percentage of growth inhibition >50% (%GI). 24 25 The results are summarized in Table 5.

Table 5. Leishmanicidal activity of compounds 9, 10, 19, 20, 23-25, 28, 29 and 31
 against *L. amazonensis* intracellular amastigotes.

Compound	%GI (100 μM)	IC ₅₀ (μΜ)	CC ₅₀ (μM)	SI
9	3.35 (±0.61)	>100	ND	-
10	4.98 (±0.09)	>100	ND	-
19	2.06 (±0.02)	>100	ND	-
20	2.05 (±1.09)	>100	ND	
23	4.57 (±0.18)	>100	ND	-
24	3.82 (±0.51)	>100	ND	-
25	6.03 (±1.97)	>100	NĎ	-
28	15.65 (±8.29)	>100	ND	-
29	28.12 (±1.51)	>100	ND	-
31	57.67 (±1.85)	5.08 (±1.31)	181.8 (±25.9)	35.8
ΑΒ 2 μΜ	81.62 (±0.43)	0.14 (±0.02)	27.86 (± 0.98)	199

The results are averages ± SD of triplicates. Compound **31** was tested in triplicate in two independent
 experiments.

5 While most of the compounds inhibited the growth of T. cruzi to some degree, in this assay only derivatives 9, 10, 19, 20, 23-25, 28, 29 and 31 presented activity at 100 6 µM against L. amazonensis. Overall, it is not possible to observe a clear correlation 7 between their structure and leishmanicidal activity. However, regarding the derivatives 8 9 based on the structure of compound 17 (compounds 28, 29, 30 and 31), lipophilicity is 10 a factor that might be related with percentage of growth inhibition. While compound 17 does not show any activity against L. amazonensis, compounds 28, 29 and 31 are 11 increasingly more potent as their lipophilicity, indicated by a higher logP value, also 12 increases (Figure 6). Lipophilic compounds are more permeable to cellular 13 membranes, what could justify this high in vitro activity. Additionally, a para-substituted 14 B ring might also be important for activity, as compound **30** showed no growth inhibition 15 at 100 µM, even though its logP value is 3.56. From an electronic point of view, the 16 different substituents in compounds 17, 28, 29 and 31 modify the electron distribution 17 on the aromatic ring B and this characteristic, together with lipophilicity, could favor the 18 19 activity for these compounds.



1

Leishmanicidal activity

Fig 6. Positive correlation between logP and leishmanicidal activity for compounds 17,
28, 29 and 31.

In view of the promising leishmanicidal property of compound **31**, we also tested its activity against intracellular amastigotes of *L. infantum* (MHOM/BR/74/PP75) to gain an insight into the spectrum of action of the compound. Despite the high activity observed against *L. amazonensis*, compound **31** had only a weak activity against *L. infantum* (%GI= 9.01 ± 0.75 at 50 μ M).

9 In conclusion, compound **31** was found to be the most active one against 10 intracellular amastigotes of both *T. cruzi* and *L. amazonensis*.

11

2.2.4. Enzymatic assay

Since the essential role of trypanothione reductase (TR) to the redox metabolism of trypanosomatids was established, it has become one of the main exploited targets in *T. cruzi* and *Leishmania* spp [58–61]. Different inhibitors have been described in literature, but up to date, none of them proceeded to the further step of drug development [62–66].

In order to investigate potential molecular targets for the most promising 17 synthesized molecules (IC₅₀< 10 μ M), the rTR inhibitory activity of compounds 17, 29 18 19 and **31** was assessed. All compounds evaluated have IC₅₀ values >100 μ M suggesting that TR is not the main target involved on their antiparasitic property. Consequently, 20 additional studies are necessary to elucidate the mechanism of action of the 21 compounds presented herein. Once a mechanism of cell death or molecular target is 22 identified, further chemical modifications might be planned to obtain isoxazole 23 24 derivatives with an optimized activity.

1 3. Concluding remarks

By using bioisosterism and molecular simplification strategies, 26 isoxazole 2 derivatives analogues of grandisin and veraguensin were designed and synthesized. 3 4 The screening of their trypanocidal, leishmanicidal (intracellular amastigotes) and 5 cytotoxic activity (THP-1 cells) afforded 6 non-cytotoxic derivatives highly active against 6 T. cruzi (compounds 17, 18, 28, 29, 30 and 31) and one non-cytotoxic derivative highly 7 active against L. amazonensis (compound 31).

Different criteria for hit identification are described in the literature and include, for 8 9 example, IC₅₀ values <1 μg/mL and SI>50 for Chagas disease [18,20,67]. Among our 10 compounds, three have IC₅₀ values lower than the IC₅₀ for benznidazole (compounds) 11 17, 29 and 31), and two have $IC_{50} < 1 \mu g/mL$ (compounds 29 and 31). Concerning 12 selectivity, compound **31** has a SI=160.9 for cells infected by *T. cruzi* and a SI=35.8 for cells infected by L. amazonensis, making it the most promising compound described 13 herein for satisfying all the hit selection criteria. Compound 31, as well as other 14 15 compounds in this study, also fits drug-likeness requirements related to Lipinski's rule of five (Supporting Information, Table S1) suggesting potential for further development 16 and biological investigations, especially due to its action on both parasites [68]. 17

- 18
- 19 4. Supporting Information

- Representative copies of ¹H and ¹³C spectra are available online. 20
- 5. Experimental section 21
- 5.1. Chemistry 22

Melting points were measured in a melting point apparatus MQAPF- 301 and are 23 reported uncorrected. All ¹H and ¹³C NMR spectra were obtained in Nuclear Bruker 24 25 Advance DPX 400 MHz, Bruker Fourier 300 MHz and Varian Oxford AS-400 using 26 TMS as internal standard, unless indicated otherwise. Mass spectra were performed in 27 ESI-TOF Bruker micrOTOF Q II and Waters Xevo G2-S QTOF. Reactions under microwave irradiation were conducted in a Discovery - CEM Explorer microwave 28 reactor with cooling, pressure and gas addition systems. Solvents and reagents, 29 purchased from Sigma Aldrich, were treated and purified when necessary, according to 30 literature [69]. Finally, thin layer chromatography was performed on silica G60 gel 31 layers SILICYCLE® with fluorescence indicator F-254 and column chromatography 32 was performed using silica gel with particle size 40-63 and 63-200 µm (Sigma Aldrich) 33 34 and hexane:ethyl acetate (Tedia) as eluent.

5.1.1. General procedure for preparation of isoxazoles 6-31 under microwave
 irradiation.

To a solution of an aldoxime (0.1 mmol) in DMF (0.3 mL) in a microwave tube, N-3 4 chloro succinimide (0.105 mmol) was slowly added to avoid overheating. The tube was sealed and submitted to microwave irradiation (30 °C, 150 W) for 1 min, and then TLC 5 6 analysis showed the complete consumption of the starting material. An alkyne (0.105 7 mmol), copper (II) sulphate (2 mol%), sodium ascorbate (10 mol%), sodium 8 bicarbonate (0.4 mmol) and water (0.3 mL) were added and the tube was further 9 irradiated for 10 min (30 °C, 150 W). The solution was diluted with 15 mL of brine, extracted with ethyl acetate (3 x 10 mL), dried (Na₂SO₄) and concentrated under 10 vacuum. The crude extract was purified by flash column chromatography on silica 11 (hexane:ethyl acetate 6:4) yielding the expected product. 12

13 **5.1.2.** General procedure for preparation of isoxazoles 6-20.

14 The compounds were synthesised based on procedure reported in the literature 15 [38]. For the preparation of the imidovl chloride, N-chloro succinimide (0.1 mmol) was slowly added to a solution of an aldoxime (0.105 mmol) in DMF (1 mL) and the reaction 16 was stirred until the starting material was not visible on the TLC analysis. After, the 17 reaction was diluted with brine (15 mL), extracted with ethyl ether (3x 10 mL), dried 18 over Na₂SO₄, concentrated under vacuum and utilized without any purification in the 19 next step. Following, propargylic alcohol (0.105 mmol), copper (II) sulphate (2 mol%), 20 sodium ascorbate (10 mol%), sodium bicarbonate (0.4 mmol) and 4 mL of H2O:t-BuOH 21 were added to the product obtained in the first part, and the reaction was further stirred 22 23 for 4 hours. Next, the reaction was diluted with brine (15 mL), extracted with ethyl acetate (3x 10 mL), dried over Na₂SO₄, concentrated under vacuum and the crude 24 extract was purified by flash column chromatography on silica (hexane:ethyl acetate 25 6:4) yielding the expected product. After purification, they were compared via TLC 26 27 analysis to the respecting compounds synthesized under microwave irradiation, when 28 this comparison was desired, and characterized by NMR and mass spectrometry.

- 29 **5.1.3.** Structural characterization of compounds 6-31
- 30 5.1.3.1. 5-hydroxymethyl-3-phenylisoxazole (6)

Yield: 72%. Yield MW: 77%. Mp: 48.8-49.5 °C. ¹H NMR (300 MHz, Methanol-d₄): $\overline{\delta}$ = 7.78-7.85 (m, 2H, Ar), 7.43-7.50 (m, 3H, Ar), 6.76 (s, 1H, Isoxazole), 4.71 (s, 2H, CH₂). ¹³C NMR (75 MHz, Methanol-d₄): $\overline{\delta}$ = 171.7, 162.5, 130.1, 128.9, 128.8, 126.8, 100.0, 56.7. HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₀H₁₀NO₂ 176.0712; Found 176.0699.

1 5.1.3.2. 5-hydroxymethyl-3-(4-methoxyphenyl)isoxazole (7)

2 Yield: 52%. Yield MW: 56%. Mp: 90.1-92.0 °C. ¹H NMR (400 MHz, CDCl₃): δ = 7.74 (d, 3 *J*= 9.1 Hz, 2H, Ar), 6.98 (d, *J*= 9.1 Hz, 2H, Ar), 6.52 (s, 1H, Isoxazole), 4.82 (s, 2H, 4 CH₂), 3.86 (s, 3H, OCH₃). ¹³C NMR (100 MHz, CDCl₃): δ = 171.4, 162.1, 161.0, 128.2, 5 121.3, 114.3, 99.8, 56.7, 55.4. HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₁H₁₂NO₃ 6 206.0817; Found 206.0815.

7 5.1.3.3. 5-hydroxymethyl-3-(3,4-dimethoxyphenyl)isoxazole (8)

Yield: 50%. Yield MW: 55%. Mp: <25 °C. ¹H NMR (400 MHz, Methanol-d₄): δ = 7.42 (d, J = 2.0 Hz, 1H, Ar), 7.39 (dd, J= 2.0, 8.3 Hz, 1H, Ar), 7.04 (d, J= 8.3 Hz, 1H, Ar), 6.73 (t, J= 0.8 Hz, 1H, Isoxazole), 4.70 (d, J= 0.8 Hz, 2H, CH₂), 3.89 (s, 3H, OCH₃), 3.88 (s, 3H, OCH₃). ¹³C NMR (100 MHz, Methanol-d₄): δ = 174.4, 163.6, 152.3, 150.9, 122.9, 121.3, 112.9, 110.9, 100.8, 56.5 (x2), 56.4. HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₂H₁₄NO₄ 236.0923; Found 236.0922.

14 5.1.3.4. 3-(4-fluorophenyl)-5-hydroxymethylisoxazole (9)

15 Yield: 47%. Yield MW: 38%. Mp: 78.8-81.8 °C. ¹H NMR (300 MHz, CDCl₃): $\bar{\delta}$ = 7.79 16 (dd, *J*= 5.3, 8.6 Hz, 2H, Ar), 7.15 (t, *J*= 8.6 Hz, 2H, Ar), 6.54 (s, 1H, Isoxazole), 4.83 (s, 17 2H, CH₂). ¹³C NMR (100 MHz, CDCl₃): $\bar{\delta}$ = 171.9, 163.8 (d, ¹*J_{CF}*= 249 Hz), 161.6, 128.7 18 (d, ³*J_{CF}*= 8 Hz), 125.0, 116.1 (d, ²*J_{CF}*= 22 Hz), 99.9, 56.6. HRMS (ESI-TOF) m/z: [M+H]⁺ 19 Calcd for C₁₀H₉FNO₂ 194.0617; Found 194.0600.

20 5.1.3.5. 5-hydroxymethyl-3-(4-nitrophenyl)isoxazole (10)

21 Yield: 36%. Yield MW: 32%. Mp: 156.0-157.5 °C. ¹H NMR (400 MHz, CDCl₃): $\overline{\delta}$ = 8.00 22 (d, *J*= 9.0 Hz, 2H, Ar), 8.33 (d, *J*= 9.0 Hz, 2H, Ar), 6.67 (s, 1H, Isoxazole), 4.88 (s, 2H, 23 CH₂). ¹³C NMR (100 MHz, CDCl₃): $\overline{\delta}$ = 172.9, 160.7, 148.7, 134.9, 127.7, 124.3, 100.1, 24 56.6. HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₀H₉N₂O₄ 221.0562; Found 221.0555.

25 5.1.3.6. 3-(3,4-benzodioxole)-5-hydroxymethylisoxazole (11)

26 Yield: 63%. Yield MW: 56%. Mp: 76.0-77.7 °C. ¹H NMR (300 MHz, CDCl₃): δ = 7.28 (d, 27 J= 1.6 Hz, 1H, Ar), 7.24 (dd, J= 8.0, 1.6 Hz, 1H, Ar), 6.86 (d, J= 8.0 Hz, 1H, Ar), 6.47 28 (s, 1H, Isoxazole), 6.01 (s, 2H, OCH₂O), 4.78 (s, 2H, CH₂). ¹³C NMR (75 MHz, CDCl₃): 29 δ = 171.8, 162.1, 149.2, 148.2, 122.7, 121.2, 108.6, 106.9, 101.5, 99.9, 56.5. HRMS 30 (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₁H₁₀NO₄ 220.0610; Found 220.0598.

1 5.1.3.7. 5-(4-hydroxymethylphenyl)-3-phenylisoxazole (12)

2 Yield: 47%. Yield MW: 52%. Mp: 168.8-169.3 °C. ¹H NMR (400 MHz, CDCl₃): δ = 7.83-3 7.89 (m, 4H, Ar), 7.45-7.52 (m, 5H, Ar), 6.84 (s, 1H, Isoxazole), 4.78 (s, 2H, CH₂). ¹³C 4 NMR (100 MHz, CDCl₃): δ = 170.1, 163.0, 143.1, 130.0, 129.1, 128.9, 127.4, 126.8, 5 126.7, 126.0, 97.5, 64.8. HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₆H₁₄NO₂ 252.1025; 6 Found 252.0987.

7 5.1.3.8. 5-(4-hydroxymethylphenyl)-3-(4-methoxyphenyl)isoxazole (13)

Yield: 45%. Yield MW: 52%. Mp: 157.0-157.9 °C. ¹H NMR (300 MHz, CDCl₃): δ = 7.82
(d, J= 6.6 Hz, 2H, Ar), 7.80 (d, J= 6.6 Hz, 2H, Ar), 7.49 (d, J= 8.4 Hz, 2H, Ar), 7.00 (d, J= 8.4 Hz, 2H, Ar), 6.78 (s, 1H, Isoxazole), 4.78 (s, 2H, CH₂), 3.87 (s, 3H, OCH₃). ¹³C
NMR (75 MHz, CDCl₃): δ = 169.9, 162.6, 161.0, 143.0, 128.2, 127.3, 126.8, 126.0, 121.6, 114.3, 97.2, 64.8, 55.4. HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₇H₁₆NO₃
282.1130; Found 282.1139.

14 5.1.3.9. 5-(4-hydroxymethylphenyl)-3-(3,4-dimethoxyphenyl)isoxazole (14)

Yield: 51%. Yield MW: 51%. Mp: 142.6-145.5 °C. ¹H NMR (400 MHz, CDCl₃): δ = 7.83 (d, *J*= 8.4 Hz, 2H, Ar), 7.48 (d, *J*= 8.4 Hz, 2H, Ar), 7.47 (d, *J*= 2.0 Hz, 1H, Ar), 7.36 (dd, *J*= 2.0, 8.4 Hz, 1H, Ar), 6.95 (d, *J*= 8.4 Hz, 1H, Ar), 6.79 (s, 1H, Isoxazole), 4.77 (s, 2H, CH₂), 3.98 (s, 3H, OCH₃), 3.95 (s, 3H, OCH₃). ¹³C NMR (100 MHz, CDCl₃): δ = 170.0, 162.7, 150.6, 149.3, 143.1, 127.3, 126.7, 126.0, 121.8, 119.9, 111.0, 109.3, 97.3, 64.8, 56.0 (x2). HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₈H₁₈NO₄ 312.1236; Found 312.1236.

22 5.1.3.10. 3-(4-fluorophenyl)-5-(4-hydroxymethylphenyl)isoxazole (15)

23 Yield: 42%. Yield MW: 40%. Mp: 179.0-179.6 °C. ¹H NMR (400 MHz, CDCl₃): δ = 7.82-24 7.89 (m, 4H, Ar), 7.51 (d, *J*= 8.6 Hz, 2H, Ar), 7.18 (t, *J*= 8.6 Hz, 2H, Ar), 6.80 (s, 1H, 25 Isoxazole), 4.79 (s, 2H, CH₂). ¹³C NMR (100 MHz, CDCl₃): δ = 170.4, 162.1, 143.2, 26 128.7 (d, ³*J*_{CF}= 8 Hz), 127.4, 126.1, 116.1 (d, ²*J*_{CF}= 22 Hz), 97.3, 64.8. HRMS (ESI-27 TOF) m/z: [M+H]⁺ Calcd for C₁₆H₁₃FNO₂ 270.0930; Found 270.0922.

28 5.1.3.11. 5-(4-hydroxymethylphenyl)-3-(4-nitrophenyl)isoxazole (16)

29 Yield: 22%. Yield MW: 20%. Mp: 189.3-191.2 °C. ¹H NMR (400 MHz, CDCl₃): δ = 8.28 30 (d, *J*= 9.0 Hz, 2H, Ar), 7.99 (d, *J*= 9.0 Hz, 2H, Ar), 7.77 (d, *J*= 8.4 Hz, 2H, Ar), 7.45 (d, 31 *J*= 8.4 Hz, 2H, Ar), 6.89 (s, 1H, Isoxazole), 4.65 (s, 2H, CH₂). ¹³C NMR (100 MHz,

1	$CDCI_3$): $\delta = 171.5$, 161.4, 148.8, 144.3, 135.3, 127.8, 127.5, 126.1, 125.9, 124.4, 97.6,
2	64.0. HRMS (ESI-TOF) m/z: $[M+H]^+$ Calcd for $C_{16}H_{13}N_2O_4$ 297.0875; Found 297.0900.
3	5.1.3.12. 3-(3,4-benzodioxole)-5-(4-hydroxymethylphenyl)isoxazole (17)
4	Yield: 18%. Yield MW: 22%. Mp: 180.1-181.7 °C. ¹ H NMR (400 MHz, CDCl ₃): δ = 7.83
5	(d, J= 8.5 Hz, 2H, Ar), 7.50 (d, J= 8.5 Hz, 2H, Ar), 7.39 (d, J= 1.8 Hz, 1H, Ar), 7.34 (dd,
6	J= 1.8, 8.1 Hz, 1H, Ar), 6.91 (d, J= 8.1, 1H, Ar), 6.75 (s, 1H, Isoxazole), 6.04 (s, 2H,
7	OCH ₂ O), 4.78 (s, 2H, CH ₂). ¹³ C NMR (100 MHz, Methanol-d ₄): \bar{o} = 171.5, 164.1, 150.8,
8	149.7, 145.4, 128.4, 127.5, 126.8, 124.0, 122.4, 109.6, 107.6, 102.9, 98.6, 64.6. HRMS
9	(ESI-TOF) m/z: $[M+H]^+$ Calcd for $C_{17}H_{14}NO_4$ 296.0923; Found 296.0891.
10	5.1.3.13. 5-(4-aminophenyl)-3-phenylisoxazole (18)
11	Yield: 49%. Yield MW: 50%. Mp: 130.6-132.4 °C. ¹ H NMR (400 MHz, CDCl ₃): δ = 7.80-
12	7.87 (m, 2H, Ar), 7.65 (d, J= 8.7 Hz, 2H, Ar), 7.74-7.50 (m, 3H, Ar), 6.75 (d, J= 8.7 Hz,
13	2H, Ar), 6.64 (s, 1H, Isoxazole), 3.95 (bs, 2H, NH ₂). ¹³ C NMR (100 MHz, CDCl ₃): δ =
14	170.9, 162.8, 148.3, 129.8, 129.4, 128.9, 127.4, 126.8, 117.9, 114.9, 95.1. HRMS (ESI-
15	TOF) m/z: $[M+H]^+$ Calcd for $C_{15}H_{13}N_2O$ 237.1028; Found 237.0993.
16	5.1.3.14. 5-[(2,3,4,6-tetraacetyl-α-D-glucopyranose-1-yl)methyl]-3-
17	phenylisoxazole (19)
18	Yield: 40%. Yield MW: 54%. Mp: 111.1-113.6 °C. ¹ H NMR (400 MHz, Methanol-d4): δ =
19	7.81-7.86 (m, 2H, Ar), 7.46-7.51 (m, 3H, Ar), 6.85 (s, 1H, Isoxazole), 5.28 (t, <i>J</i> = 9.6 Hz,

20 1H, H3'), 5.05 (t, *J*= 9.6 Hz, 1H, H4'), 4.95 (t, *J*= 8.4 Hz, 1H, H2'), 4.94 (d, *J*= 13.9 Hz, 21 1H, CH), 4.87 (d, *J*= 13.9 Hz, 1H, CH), 4.86 (t, *J*= 8.4 Hz, 1H, H1'), 4.29 (dd, *J*= 4.6, 22 12.4 Hz, 1H, H6'), 4.16 (dd, *J*= 2.5, 12.4 Hz, 1H, H6'), 3.92 (m, 1H, H5'), 2.04 (s, 3H, 23 CH₃), 2.01 (s, 3H, CH₃), 2.00 (s, 3H, CH₃), 1.96 (s, 3H, CH₃). ¹³C NMR (100 MHz, 24 Methanol-d4): δ = 170.6, 170.2, 169.4, 168.3, 130.2, 129.0, 126.8, 101.6, 99.9, 72.6, 25 72.1, 71.0, 68.2, 61.74, 61.72, 29.7, 20.73, 20.71, 20.6. HRMS (ESI-TOF) m/z: [M+H]⁺ 26 Calcd for C₂₄H₂₈NO₁₁ 506.1618; Found 506.1655.

5.1.3.15. 5-[(2,3,4,6-tetraacetyl-α-D-glucopyranose-1-yl)methyl]-3-(3,4benzodioxole)isoxazole (20)

29 Yield: 46%. Yield MW: 48%. Mp: 110.4-112.0 °C. ¹H NMR (400 MHz, Methanol-d4): δ = 30 7.37-7.31 (m, 2H, Ar), 6.93 (d, *J*= 8.3 Hz, 1H, Ar), 6.77 (s, 1H, Isoxazole), 6.03 (s, 2H, 31 OCH₂O), 5.28 (t, *J*= 9.3 Hz, 1H, H3'), 5.05 (t, *J*= 9.3 Hz, 1H, H4'), 4.94 (t, *J*= 9.1 Hz, 32 1H, H2'), 4.93 (t, *J*= 13.4 Hz, 1H, CH), 4.84 (d, *J*= 9.1 Hz, 1H, H1'), 4.83 (d, *J*= 13.4 Hz,

1 H, CH), 4.29 (dd, J= 4.6, 12.4 Hz, 1H, H6'), 4.16 (dd, J= 2.3, 12.4 Hz, 1H, H6'), 3.92 (m, 1H, H5'), 2.05 (s, 3H, CH₃), 2.01 (s, 3H, CH₃), 2.00 (s, 3H, CH₃), 1.96 (s, 3H, CH₃). ¹³C NMR (100 MHz, Methanol-d4): δ = 172.4, 171.2, 171.6, 171.3, 170.3, 163.5, 150.9, 149.8, 123.9, 122.4, 109.7, 107.6, 103.0, 102.9, 101.3, 74.1, 73.0, 72.7, 69.7, 63.0, 62.7, 20.6, 20.59, 20.56, 20.54. HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₂₅H₂₈NO₁₃ 550.1561; Found 550.1522.

7 5.1.3.16. (3,5-diphenyl)isoxazole (**25**)

Yield: 42%. Mp: 141.2-143.8 °C. ¹H NMR (300 MHz, CDCl₃): δ = 7.83-7.89 (m, 4H, Ar),
7.47-7.51 (m, 6H, Ar), 6.84 (s, 1H, Isoxazole). ¹³C NMR (100 MHz, CDCl₃): δ = 170.3,
162.9, 130.2, 130.0, 129.0, 128.9, 127.4, 126.8, 125.8, 97.4. HRMS (ESI-TOF) m/z:
[M+H]⁺ Calcd for C₁₅H₁₂NO 222.0913; Found 222.0912.

12 5.1.3.17. 5-(4-acetamidephenyl)-3-phenylisoxazole (26)

13 Yield: 19%. Mp: 179.0-182.0 °C. ¹H NMR (300 MHz, Methanol-d₄): δ = 7.85-7.91 (m, 14 2H, Ar), 7.84 (d, *J*= 8.8 Hz, 2H, Ar), 7.73 (d, *J*= 8.8 Hz, 2H, Ar), 7.46-7.52 (m, 3H, Ar), 15 7.10 (s, 1H, Isoxazole), 2.17 (s, 3H, CH₃). ¹³C NMR (75 MHz, Methanol-d₄): δ = 170.4, 16 170.1, 163.0, 140.5, 129.9, 128.7, 126.3, 126.1, 122.6, 119.6, 96.7, 23.3. HRMS (ESI-17 TOF) m/z: [M+H]⁺ Calcd for C₁₇H₁₅N₂O₂ 279.1134; Found 279.1139.

18 5.1.3.18. 5-(tert-butylcarbamatephenyl)-3-phenylisoxazole (27)

Yield: 48%. Mp: 188.3-192.9 °C. ¹H NMR (300 MHz, CDCl₃): δ = 7.84-7.90 (m, 2H, Ar),
7.77 (d, *J*= 8.9 Hz, 2H, Ar), 7.50 (d, *J*= 8.9 Hz, 2H, Ar), 7.45-7.50 (m, 3H, Ar), 6.76 (s,
1H, Isoxazole), 6.65 (bs, 1H, NH), 1.54 (s, 9H, CH₃). ¹³C NMR (75 MHz, CDCl₃): δ =
170.0, 162.6, 140.2, 130.2, 130.0, 128.9, 126.8, 122.5, 118.4, 96.6, 53.5, 28.3. HRMS
(ESI-TOF) m/z: [M+H]⁺ Calcd for C₂₀H₂₁N₂O₃ 337.1552; Found 337.1559.

24 5.1.3.19. 5-(4-acetoxymethylphenyl)-3-(3,4-benzodioxole)isoxazole (28)

25 Yield: 28%. Mp: 164.2-168.0 °C. ¹H NMR (300 MHz, CDCl₃): δ = 7.83 (d, *J*= 8.3 Hz, 26 2H, Ar), 7.48 (d, *J*= 8.3 Hz, 2H, Ar), 7.38 (d, *J*= 1.5 Hz, 1H, Ar), 7.35 (dd, *J*= 1.5, 8.1 27 Hz, 1H, Ar), 6.91 (d, *J*= 8.1 Hz, 1H, Ar), 6.76 (s, 1H, Isoxazole), 6.04 (s, 2H, CH₂), 2.14 28 (s, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃): δ = 170.8, 169.8, 162.6, 149.2, 148.3, 138.1, 29 128.7, 127.3, 126.0, 123.0, 121.2, 108.7, 107.0, 101.5, 97.6, 65.7, 21.0. HRMS (ESI-30 TOF) m/z: [M+H]⁺ Calcd for C₁₉H₁₆NO₅ 338.1028; Found 338.1028.

1 5.1.3.20. 3-(3,4-benzodioxole)-5-(4-methylphenyl)isoxazole (29)

2 Yield: 26 %. Mp: 141.9-144.9 °C. ¹H NMR (300 MHz, CDCl₃): δ = 7.71 (d, *J*= 8.2 Hz, 3 2H, Ar), 7.37 (d, *J*= 1.7 Hz, 1H, Ar), 7.33 (dd, *J*= 1.7, 8.0 Hz, 1H, Ar), 7.28 (d, *J*= 8.2 4 Hz, 2H, Ar), 6.91 (d, *J*= 8.0 Hz, 1H, Ar), 6.69 (s, 1H, Isoxazole), 6.03 (s, 2H, CH₂), 2.41 5 (s, 3H, CH₃). ¹³C NMR (75 MHz, CDCl₃): δ = 170.4, 162.5, 149.1, 148.2, 140.5, 129.7, 125.7, 124.8, 123.2, 121.1, 106.6, 107.0, 101.5, 96.7, 21.5. HRMS (ESI-TOF) m/z: 7 [M+H]⁺ Calcd for C₁₇H₁₄NO₃ 280.0974; Found 280.0975.

8 5.1.3.21. 3-(3,4-benzodioxole)-5-phenylisoxazole (**30**)

9 Yield: 29%. Mp: 125.2-127.9 °C. ¹H NMR (300 MHz, CDCl₃): δ = 7.83 (dd, *J*= 1.9, 7.7 10 Hz, 2H, Ar), 7.52 – 7.44 (m, 2H, Ar), 7.39 (d, *J*= 1.70 Hz, 1H, Ar), 7.34 (dd, *J*= 1.7, 8.0 11 Hz, 1H, Ar), 6.90 (d, *J*= 8.0 Hz, 1H, Ar), 6.75 (s, 1H, Isoxazole), 6.04 (s, 2H, CH₂). ¹³C 12 NMR (75 MHz, CDCl₃): δ = 170.2, 162.6, 149.1, 148.2, 130.2, 129.0, 125.8, 123.1, 13 121.2, 108.6, 107.0, 101.5, 97.3. HRMS (ESI-TOF) m/z: [M+H]⁺ Calcd for C₁₆H₁₂NO₃ 14 266.0817; Found 266.0831.

15 **5.1.4.** General procedure for tosylation

16 The compounds were synthesized from primary alcohols following a methodology 17 described in literature [70]. Purification were done by column chromatography on silica 18 (hexane:ethyl acetate 8:2) yielding the desired products.

19 20 5.1.4.1. 3-(4-methoxyphenyl)-5-[(4-

methylbenzenesulfonate)methylphenyl)isoxazole (7a)

21 Yield: 76%. Mp: 81-82 °C. ¹H NMR (300 MHz, CDCl₃): δ = 7.81 (d, *J*= 8.3 Hz, 2H, Ar), 22 7.67 (d, *J*= 8,8 Hz, 2H, Ar), 6.98 (d, *J*= 8,3 Hz, 2H, Ar), 6.98 (d, *J*= 8,8 Hz, 2H, Ar), 6.50 23 (s, 1H, Isoxazole), 5.17 (s, 2H, CH₂), 3.85 (s, 3H, OCH₃), 2.41 (s, 3H, CH₃). ¹³C NMR 24 (75 MHz, CDCl₃): δ = 164.5, 162.1, 161.2, 145.5, 132.5, 130.0, 128.2, 128.0, 120.8, 25 114.4, 102.9, 61.2, 55.4, 21.7.

26 **5.1.5.** 3-phenylisoxazole-5-carboxylic acid (21)

Jones reagent 8 N (0.3 mmol) was added dropwise to a solution of compound **7** (0.1 mmol) in acetone (4 mL) in an ice bath. The solution was stirred for 30 minutes when TLC analysis showed the total consumption of the starting material. The solution was diluted in acetone (15 mL) filtered through celite and evapored under vacuum. The crude extract was purified by flash column chromatography on silica (hexane:ethyl acetate 3:7) affording compound **23** as a white solid. Yield: 69%. Mp: 155.0-57.4 °C.

1 ¹H NMR (300 MHz, CDCl₃): δ = 7.78-7.90 (m, 2H, Ar), 7.45-7.55 (m, 3H, Ar), 7.25 (s, 2 1H, Isoxazole). ¹³C NMR (75 MHz, CDCl₃): δ = 166.9, 165.5, 162.3, 134.4, 133.0, 3 132.8, 131.9, 130.7, 111.3. HRMS (ESI-TOF) m/z: [M-H]⁻ Calcd for C₁₀H₆NO₃ 4 188.0353; Found 188.0321.

5 **5.1.6.** 5-(azidomethyl)-3-(4-methoxyphenyl)isoxazole (**22**)

6 DMF (0.3 mL), sodium azide (0.15 mmol) and compound 8a (0.1 mmol) were 7 added to a microwave tube and submitted to irradiation for 7 minutes (150 W, 70 °C). At the end, the solution was diluted with 10 mL of brine, extracted with ethyl acetate (3 8 x 10 mL), dried (NaSO₄) and concentrated under vacuum. The product (compound 22) 9 was obtained guantitatively (20 mg, 100%). Mp: 51.9-53.0 °C. ¹H NMR (300 MHz, 10 11 $CDCl_3$): $\delta = 7.75$ (d, J= 9.0 Hz, 2H, Ar), 6.99 (d, J= 9.0 Hz, 2H, Ar), 6.55 (s, 1H, Isoxazole), 4.50 (s, 2H, CH₂), 3.87 (s, 3H, OCH₃). ¹³C NMR (75 MHz, CDCl₃): $\delta =$ 12 13 166.7, 161.2, 128.2, 121.0, 114.4, 101.1, 55.4, 45.5. IR (KBr): 2100 (N₃) cm⁻¹. HRMS $(ESI-TOF) m/z: [M+H]^+ Calcd for C_{11}H_{11}N_4O_2 231.0882; Found 231.0872.$ 14

5.1.7. General procedure for deacetylation of 2,3,4,6-tetracetyl-α-D glucopyranosydes (19 and 20)

The protected glucoside (compound **19** or **20**, 0.1 mmol) was solubilized in methanol (1 mL) and NaOMe (1 M) was added until the pH of the solution reached 9. The solution was stirred for 15 min, cooled to 0 °C and then the resin Dowex 50WX-200 was added until pH dropped to 7. The solution was filtered and the product (**23** or **24**) was obtained in good yields [39].

22 23 5.1.7.1. 5-[(2,3,4,6-tetrahydroxy-α-D-glucopyranose-1-yl)methyl]-3phenylisoxazole (**23**)

24 Yield: 95%. Mp: 149.0-152.0°C. HRMS (ESI-TOF) m/z: $[M+Na]^+$ Calcd for 25 $C_{16}H_{19}NO_7Na$ 360.1054; Found 360.1056.

26 5.1.7.2. 3-(3,4-benzodioxole)-5-[(2,3,4,6-tetrahydroxy-α-D-glucopyranose-1-27 yl)methyl]isoxazole (**24**)

28 Yield: 80%. Mp: 186.3-188.5 °C. HRMS (ESI-TOF) m/z: $[M+Na]^+$ Calcd for 29 $C_{17}H_{19}NO_9Na$ 404.0952; Found 404.0941.

1 **5.1.8.** 3-(3,4-benzodioxole)-5-(4-chloromethylphenyl)isoxazole (**31**)

2 A solution of compound 17 (0.1 mmol) in thionyl chloride (100 µl) was stirred under a nitrogen atmosphere for 1 hour when TLC analysis showed total consumption 3 4 of the starting material. After, the reaction was cooled down to 0 °C, quenched with water and extracted with CH₂Cl₂ (3x 10 mL). The combined organic layers were then 5 washed with brine and dried under vacuum affording compound 31 as a pale yellow 6 powder (30 mg, 94%). Mp: 162.1-163.2 °C. ¹H NMR (300 MHz, CDCl₃): δ = 7.82 (d, J= 7 8 8.4 Hz, 2H, Ar), 7.51 (d, J= 8.4 Hz, 2H, Ar), 7.38 (d, J= 1.7 Hz, 1H, Ar), 7.34 (dd, J= 8.0, 1.7 Hz, 1H, Ar), 6.90 (d, J= 8.0 Hz, 1H, Ar), 6.76 (s, 1H, Isoxazole), 6.04 (s, 2H, 9 CH₂), 4.63 (s, 2H, CH₂). ¹³C NMR (75 MHz, CDCl₃): δ = 169.6, 163.2, 149.2, 148.2, 10 139.4, 129.2, 127.4, 122.9, 121.2, 106.6, 101.5, 97.7, 46.6. HRMS (ESI-TOF) m/z: 11 12 $[M+H]^+$ Calcd for C₁₇H₁₃CINO₃ 314.0584; Found 314.0576.

13

5.2. Biologic assays

14 **5.2.1.** Trypanocidal and Leishmanicidal screening in intracellular amastigotes.

THP-1 cells (ATCC TIB202) were grown in RPMI-1640 without phenol red
(Sigma-Aldrich, CO. St. Louis, MO, USA) supplemented with 10% FBS (Life
Technologies, USA), 12.5 mM HEPES, penicillin (100 U/ml), streptomycin (100 μg/ml)
and Glutamax (2 mM) (37 °C, 5% CO₂).

L. amazonensis MHOM/BR/77/LTB0016 and *L. infantum* MHOM/BR/74/PP75
 promastigotes, expressing β-galactosidase, were grown in Schneider's insect medium
 (Sigma Chemical Co., St. Louis, MO, USA) supplemented with 5% of heat inactivated
 FBS and 2% of human urine (26 °C, 5% CO₂).

T. cruzi Tulahuen trypomastigotes (raised from infected L929 cell line),
 expressing β-galactosidase, were provided by the Laboratory of Cellular and Molecular
 Parasitology, Centro de Pesquisas René Rachou, FIOCRUZ, Belo Horizonte [71].

Afterwards, THP-1 cells were cultivated in 96 well plates $(4.0 \times 10^4 \text{ cells/well})$ in supplemented RPMI-1640 and treated with 100 ng/ml of phorbol 12-myristate 13acetate (PMA) for 72 h (37 °C, 5% CO₂) to allow cell differentiation into non-dividing macrophages [72].

Four days culture promastigotes of *L. amazonensis* $(4.0 \times 10^6 \text{ parasites/ml})$ and *L. infantum* $(4.0 \times 10^6 \text{ parasites/ml})$ were washed with phosphate buffered saline (PBS, pH 7.4) and incubated in RPMI-1640 supplemented with 10% of heat-inactivated human B+ serum for 1 h (34 °C) to parasite opsonization. THP-1 cells were then incubated with a parasite/cell ratio of 10:1 for 3 h (34 °C, 5% CO₂) for *L. amazonensis* and *L. infantum*, and with a parasite/cell ratio of 2:1 overnight (37 °C, 5% CO₂) for *T.*

1 *cruzi.* After that, non-adherent parasites were removed by washing with PBS and 2 infected cells were incubated with 180 μ l of full supplemented RPMI-1640 medium for 3 another 24 h (34 °C and 37 °C, 5% CO₂) to allow the transformation of 4 promastigotes/trypomastigotes into intracellular amastigotes.

The infected cells were treated with 20 µl of each compound (2 µM of DMSO-5 diluted stock solution in 18 μ M of RPMI-1640) in triplicate, followed by incubation for 48 6 h (34 °C and 37 °C, 5% CO₂). Subsequently, cells were carefully washed with PBS and 7 8 incubated for 16 h (34 °C, 5% CO₂) with 250 µl of chlorophenolred-ß-dgalactopyranoside (Sigma-Aldrich Co., St. Louis, MO, USA) (CPRG) at 100 µM and 9 10 Nonidet P-40 (Amresco Inc, Solon, Ohio, USA) (NP-40) 0.1%. Optical density was read at 570/630 nm in an Infinite M200 TECAN, Austria. Amphotericin B (Bristol-Myers, 11 12 Squibb) and benznidazole (Sigma Aldrich) were used as positive control and DMSO 13 1% as negative control.

14 15

5.2.2. Trypanocidal screening in trypomastigotes

16

Culture T. cruzi Tulahuen trypomastigotes (raised from infected L929 cell line), 17 were cultivated in 96 well plates $(1.5 \times 10^6 \text{ parasites/well})$ and treated with the 18 compounds (2 µM of DMSO-diluted stock solution in 18 µM of RPMI-1640) serially 19 20 diluted in concentrations ranging from 250 to 3.9 µM. After incubation for 72 h (37°C, 5% CO₂), cell viability was assessed by the MTT assay, which consists in the 21 colorimetric measurement of the metabolization of 3-(4,5-dimethylthiazol-2-yl)-2,5-22 diphenyltetrazolium bromide (MTT) to formazan by viable cells. DMSO 1% was used 23 24 as negative control, and benznidazole (Sigma Aldrich) as positive control. The optical density was read at 540 nm in a Infinite M200 TECAN microplate reader immediately 25 after the dissolution of formazan crystals with DMSO [73,74]. 26

27 **5.2.3.** Cytotoxicity assay (MTT)

THP-1 cells were grown and cultivated in 96 well plates (4.0×10^6 cells/well), as described in 5.2.1., treated with the compounds (2 µM of DMSO-diluted stock solution in 18 µM of RPMI-1640) serially diluted in concentrations ranging from 15.6 µM to 500 µM and incubated for 72 h (37 °C, 5% CO₂). Cell viability was assessed as described in 5.2.2.

1 **5.2.4.** Trypanothione reductase enzyme assay

Recombinant trypanothione reductase from T. cruzi (TcTR), was expressed in 2 Escherichia coli BL21DE3 and purified by affinity chromatography. TR assays were 3 4 performed as described by Hamilton et al. (2003) [75]. In 96 well micro plates (final volume= 240 µL), TcTR (1 m-unit), HEPES (40 mM, pH 7.5), NADPH (0.15 mM), 5 6 DTNB (25 μ M) and EDTA (1 mM) were incubated for 5 min (27 °C) before T(S)₂ (1 μ M) 7 and the tested compound (diluted in DMSO) were added. Compounds and controls 8 were pre-incubated at 27 °C for 30 min and 10 μL of DTNB was added to the reaction 9 mixture. Absorbance at 412 nm was measured for 30 min to determine the enzymatic activity. DMSO 1% was used as negative control and clomipramine as positive control. 10

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Design and synthesis of a new series of 3,5-disubstituted isoxazoles active against *Trypanosoma cruzi* and *Leishmania amazonensis*

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Highlights

- NP-based analogues were planned using bioisosterism and simplification strategies.
- 26 isoxazole derivatives were synthesized.
- Their trypanocidal and leishmanicidal activities were evaluated.
- 22 compounds were active against *T. cruzi* and 10 against *L. amazonensis*.