

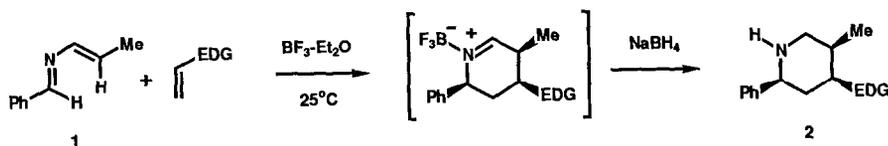
ACID CATALYZED INTRAMOLECULAR 2-AZA-1,3-DIENE DIELS-ALDER CYCLIZATION PROCESSES

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Abstract. Preparation and acid-catalyzed Diels-Alder cyclizations of a 1-alkenyl-2-aza-1,3-diene have been explored.

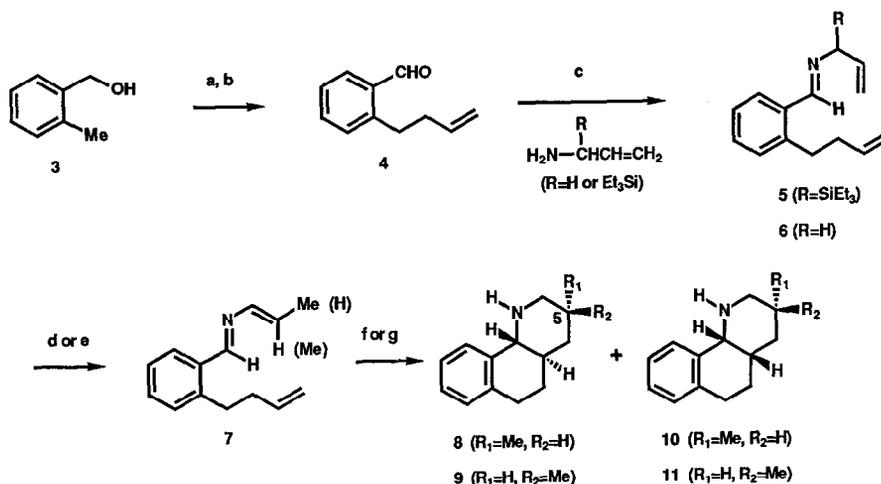
Hetero-Diels-Alder cycloaddition processes have recently received increased attention owing to their obvious use in the area of heterocycle synthesis. The comprehensive review by Boger and Weinreb¹ contains numerous examples demonstrating the broad application of 4+2-cycloadditions with hetero-atom containing dienes and dienophiles. Our interest in this area is concerned with Diels-Alder cycloadditions of 2-aza-1,3-dienes. In an earlier report,² we described our initial studies which focused on the development of a mild, protodesilylation-based method for 2-azadiene preparation. In an investigation which followed, we demonstrated that the model azadiene **1** undergoes intermolecular cycloaddition to a variety of electron rich dienophiles under mild (25°C), Lewis-acid (BF₃) catalyzed conditions to produce after NaBH₄ reductive work-up the substituted piperidines **2**.³ The highly regioselective, stereoselective and stereospecific nature of these processes suggested that they occur by concerted mechanisms.⁴ While these intermolecular processes show promise as methods for substituted piperidine synthesis, their application would be limited by low yields (25 - 45%), the lack of reactivity of all but electron rich dienophiles (enol ethers, enamines) and of azadienes which can not readily adopt low energy *s-cis* conformations (e.g. *E,Z*-isomer of **1**), and competitive azadiene dimerizations which occur under the BF₃ reaction conditions.



In order to probe further the synthetic potential of 2-azadiene cycloadditions and to determine if the limitations noted above for the intermolecular reactions can be circumvented, we have initiated efforts to explore intramolecular versions of this process.⁵ Initial phases of our studies have focused on the acid catalyzed cyclization chemistry of azatriene **7**. This substance, as a mixture of C-C double bond isomers, was prepared by the sequences shown in Scheme 1 which feature base-catalyzed isomerization⁶ and fluoride-induced protodesilylation² methods for azadiene formation. The silylallyl-imine **5** can be transformed to a 7:1 *E,E,E,Z*

isomeric mixture of **7** by treatment with CsF in THF containing 0.4 equiv. of 18-crown-6. Alternatively, base-catalyzed (KOtBu/THF) isomerization of allyl-imine **6** in the presence or absence of 18-crown-6 provides **7** as respective 8:1 or 1.1:1 mixtures of E,E,Z-isomers. The stereoselectivities (or lack thereof) displayed in these processes are discussed in detail elsewhere.^{2b}

Scheme 1.



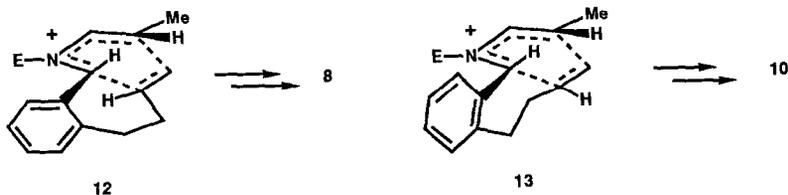
(a) nBuLi, THF/CH₂=CH-CH₂Br (73%); (b) PCC, CH₂Cl₂ (92%); (c) C₆H₆, 3A mol. sieves (ca. 95%); (d) for **5** → **7** CsF, THF, 0.4 eq. 18-crown-6, 25°C 8h (ca. 90%); (e) for **6** → **7** KOtBu, THF, w/wo 18-crown-6, -30°C, 3 min (99%); (f) BF₃-Et₂O (1.1 equiv), C₆H₆ followed by NaBH₄, EtOH; (g) HBF₄ (1.1 equiv.), C₆H₆/ NaBH₄.

Diels-Alder cyclization of **7** (8:1 E,E,Z) occurs efficiently under mild (25°C) conditions in the presence of Lewis (BF₃-Et₂O) or protic (HBF₄) acid catalysts. The hydrophenanthridines **8** - **11** are isolated in the yields shown in Table 1 after NaBH₄ workup and silica gel chromatographic separation. The distributions of stereoisomeric hydrophenanthridines for both the BF₃ and HBF₄ catalyzed processes vary in a regular fashion with time and temperature with the trans-isomer **8** being the major product for short time, lower temperature reactions. Additionally, under these conditions the **8** + **10** / **9** + **11** product ratios are nearly equal to the starting azadiene E,E/Z ratio (8:1). These results suggest that both stereoisomers of **7** undergo kinetically controlled, stereospecific cyclization (**7E,E** → **8** + **10** and **7E,Z** → **9** + **11**). Moreover, cyclization of **7E,E** is stereoselective, occurring preferentially via an exo-transition state, **12**. This preference appears to result from arene-azadiene conjugation which is maintained to a greater extent in **12** compared to the endo-transition state **13**.⁷ Factors similar to this are known to govern the stereochemistry of other intramolecular Diels-Alder cyclizations.⁸

Hydrophenanthridine stereoisomer distributions from reactions of **7** at higher temperatures and longer times favor the *trans*, *cis*-isomer **9**. This appears to be due to thermodynamic factors related to the relative energies of iminium cation precursors of **8** - **11**. Thus, transformation of the tricyclic iminium cation **14**, formed initially by exocyclization of protonated or BF₃-complexed **7E,E** to its β-Me epimer **16** (the precursor of **9**) should be an exothermic process. Evidence has been gained to support assignment of iminium cation-enamine interconversions^{9a} (**14** -> **15**-> **16**) as the major pathway responsible for equilibration of these systems.^{9b} Specifically, cyclization of **7** catalyzed by DBF₄ followed by NaBH₄ work-up provides hydrophenanthridines **8** and **9**, both of which contain mono-deuterium substitution at the piperidine C-ring, C-5 position. Under the

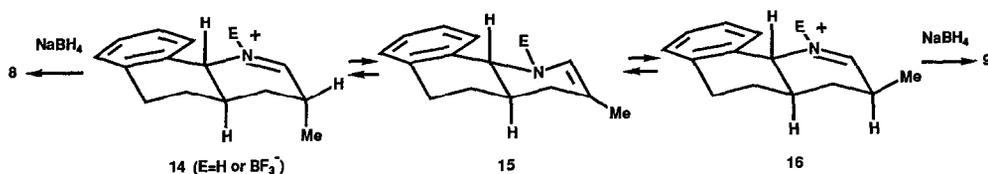
Table1. Representative Results of Azatriene **7** Cyclizations Catalyzed by BF₃ - Et₂O and HBF₄.

		BF ₃ - Et ₂ O						HBF ₄			
Time (h)	Temp. (°C)	Yields				Time (h)	Temp. (°C)	Yields			
		8	9	10	11			8	9	10	11
1	25	42	2	8	4	1	5	52	—	3	6
2	25	47	3	7	3	2	25	56	6	5	—
24	25	14	26	13	12	48	25	10	45	—	8
1	45	10	18	9	6	2	45	52	14	4	—
48	45	13	25	5	11	48	45	13	40	—	7



reaction conditions, deuterium is not incorporated into unreacted **7**.¹⁰ This result in conjunction with the observation that stereoisomeric azadienes related to **7** are only slowly isomerized under the BF₃ - Et₂O and HBF₄ cyclization reaction conditions,^{3,11} allows us to rule out an alternate route as the major pathway for the **14** -> **16** interconversion involving reversible cycloaddition and **7E,E** -> **7E,Z** isomerization.

The above results demonstrate that intramolecular Diels-Alder cyclizations can serve as efficient methods for stereocontrolled synthesis of N-heteroatom containing, fused polycyclic systems. The yields of these processes, catalyzed by proton or Lewis acids, are modestly high even in this case where a non-electron donating substituted dienophile moiety participates. The stereochemistry of the cyclization reactions appears to be controlled in a



predictable way by kinetic (transition state preferences) and thermodynamic (equilibration via iminium cation \rightarrow enamine) factors. Finally, the product iminium cation functions in the products of these cyclizations (*e.g.* 14 and 16) can potentially serve as a site for the introduction of additional functionality.¹² Thus, the use of this chemistry in alkaloid synthesis appears promising.

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References.

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- (7) Owing to the small amounts of 9 and 11 generated, it is difficult to accurately determine the *exo* vs. *endo* transition state preferences for kinetically controlled cyclization of 7E,Z.
- (8) See for example Stork, G.; Morgans, D.J. *J. Am. Chem. Soc.*, **1979**, 101, 7110.
- (9) (a) For an example of a related yet slower iminium cation epimerization process see Evans, D.A.; Mitch, C.H.; Thomas, R.C.; Zimmerman, D.M.; Robey, R.L. *J. Am. Chem. Soc.*, **1980**, 102, 5955; (b) the increase in yields of 10 and 11 in the BF_3 -catalyzed reaction of 7 suggests that another equilibration pathway involving reversible Diels-Alder cyclization is also operating.
- (10) ^1H NMR analysis of the N-arylmethyl-N-propyl amine NaBH_4 reduction product of unreacted 7 from reaction catalyzed by DBF_4 (1 equiv) at 25°C for 1h showed no D-incorporation.
- (11) Isomerization of azadiene 1 or its E,Z-isomer with HBF_4 (C_6H_6 , 25°C) is slow.
- (12) Work up of the reaction mixture from BF_3 -catalyzed cyclization of 7 with NaCN (2 equiv.) gives α -cyanoamine analogs of 8-11.

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