

Ring Enlargement of Enantiopure 1,2-Oxazines to 1,2-Oxazepine Derivatives and Their Palladium-Catalyzed Couplings

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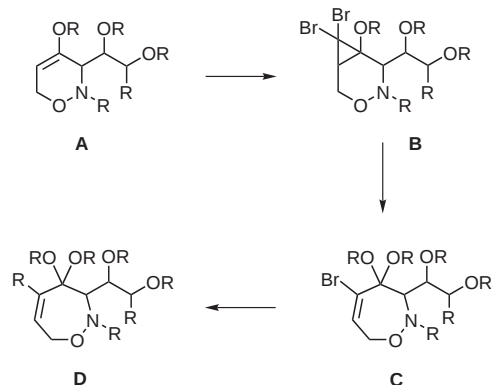
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Received 5 July 2005

Abstract: Phase-transfer-catalyzed cyclopropanation of enantiopure 1,2-oxazine derivatives *anti*-**1a,b** or *syn*-**1** followed by solvolysis of the resulting geminal dibromocyclopropane intermediates afforded the expected ring-expanded products, namely 1,2-oxazepines *anti*-**3a,b** or *syn*-**3**. These heterocycles could be further substituted by use of their bromoalkenyl group employing palladium-catalyzed coupling reactions, which smoothly led to new enantiopure 1,2-oxazepine derivatives *anti*-**4**–*anti*-**8**.

Key words: dibromocyclopropanes, ring enlargement, 1,2-oxazines, 1,2-oxazepines, palladium catalysis

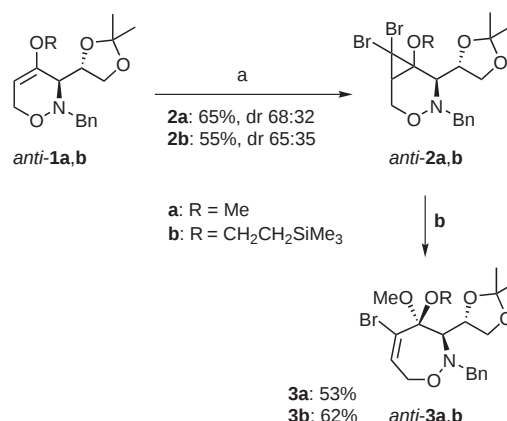
We have recently demonstrated that readily available 3,6-dihydro-2*H*-1,2-oxazines **A**¹ are versatile intermediates for syntheses of several classes of enantiopure compounds.² Here we enclose that homologous compounds **C** are smoothly available via intermediate geminal dibromocyclopropanes **B**. The palladium-catalyzed coupling reactions of bromoalkenes **C** led to new 1,2-oxazepines **D** of considerable diversity (Scheme 1).



Scheme 1 Transformation of 3,6-dihydro-2*H*-1,2-oxazines **A** into enantiopure 1,2-oxazepine derivatives **C** and **D** via dibromocyclopropanes **B**.

Enantiopure 1,2-oxazines *anti*-**1a,b** were first treated with bromoform, potassium fluoride, and sodium hydroxide in the presence of a phase transfer catalyst, which furnished the desired dibromocyclopropane derivatives *anti*-**2a,b** in moderate yields with low diastereoselectivities (Scheme 2). It is noteworthy that experiments which were

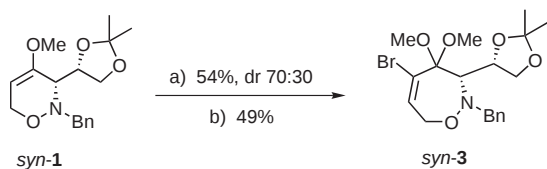
carried out using bromoform and base but without addition of potassium fluoride did not lead to *anti*-**2a,b**. Instead, the starting material underwent fast decomposition. This problem has, however, been circumvented by employing a large excess of potassium fluoride. Similar observations have been described in literature for related reactions.³ Subsequent solvolysis of the dibromocyclopropane derivatives in refluxing methanol with potassium carbonate³ gave the expected 1,2-oxazepine derivatives *anti*-**3a,b** in moderate yields. The isolation of 1,2-oxazepine derivative *anti*-**3b** as a single diastereomer can be attributed to the fact that the intermediate allylic cation probably accepts the nucleophile methanol with high preference *trans* to the bulky dioxolanyl group.



Scheme 2 Reagents and conditions: a) CHBr₃, 50% NaOH (aq), KF, Et₃BnNCl, r.t., 2 d; b) K₂CO₃, MeOH, reflux, 20 h.

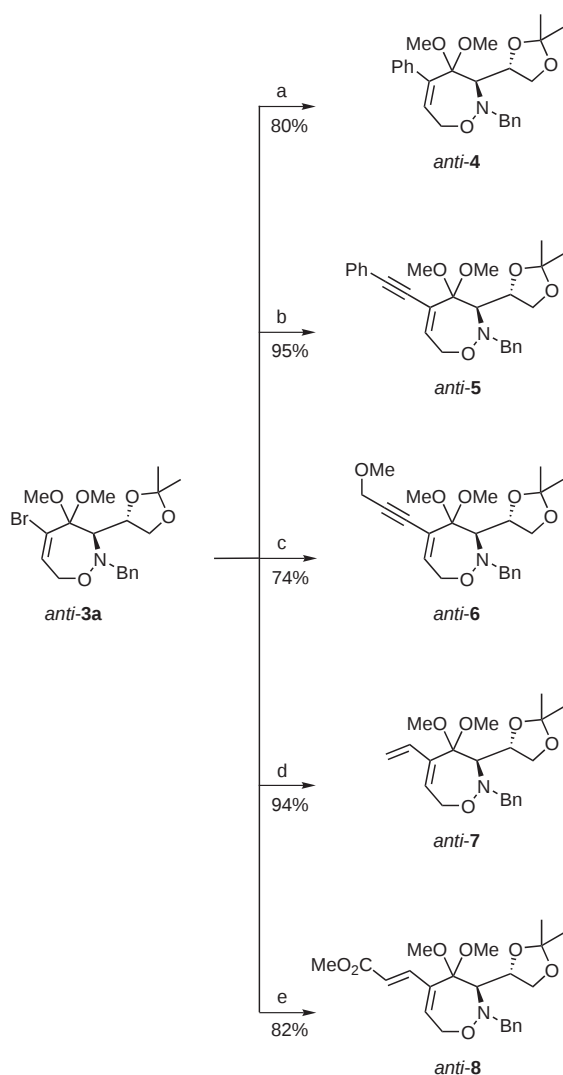
The diastereomeric enantiopure 3,6-dihydro-2*H*-1,2-oxazine *syn*-**1** was analogously converted into the ring-expanded heterocycle *syn*-**3** in moderate overall yield (Scheme 3).⁴ The geminal dibromocyclopropane derivatives **2** could be isolated (and the diastereomers may be separated) but in general it is more efficient to use unseparated intermediates **2** for the solvolysis reactions leading to 1,2-oxazepines **3**.

Enantiopure seven-membered 1,2-oxazepines⁵ *anti*-**3a,b** and *syn*-**3** allow an approach to many interesting product classes, e.g. amino-substituted polyols (heptanose derivatives) or stereodefined enantiopure piperidines. As well, the presence of a bromoalkenyl moiety makes these 1,2-oxazepine derivatives ideal candidates for palladium-catalyzed coupling reactions. Scheme 4 summarizes our preliminary results employing *anti*-**3a**, which demon-



Scheme 3 Reagents and conditions: a) CHBr_3 , 50% NaOH (aq), KF , Et_3BNHCl , r.t., 2 d; b) K_2CO_3 , MeOH , reflux, 20 h.

strate that Suzuki couplings⁶ or Sonogashira reactions⁷ provide compounds such as *anti*-4, *anti*-5, and *anti*-6, respectively, in excellent yields. Gratifyingly, Stille reactions⁸ and Heck couplings⁹ of 1,2-oxazepine *anti*-3a also occurred smoothly delivering 1,3-dienes *anti*-7 and *anti*-8 with very good efficacy. These latter coupling products should be excellent partners in Diels–Alder reactions, which may lead to interesting enantiopure skeletons incorporating a 1,2-oxazepine ring.



Scheme 4 Reagents and conditions: a) $\text{PhB}(\text{OH})_2$, $\text{Pd}(\text{OAc})_2$, PPh_3 , K_2CO_3 , DMF , 70°C , 10 h; b) phenylacetylene, $\text{Pd}(\text{OAc})_2$, PPh_3 , CuI , $i\text{-Pr}_2\text{NH}$, DMF , r.t., 10 h; c) 3-methoxypropyne, $\text{Pd}(\text{OAc})_2$, PPh_3 , CuI , $i\text{-Pr}_2\text{NH}$, DMF , r.t., 10 h; d) tributylvinylstannane, $\text{Pd}(\text{OAc})_2$, PPh_3 , DMF , 65°C , 3 h; e) methyl acrylate, $\text{Pd}(\text{OAc})_2$, LiCl , Et_3N , DMF , 70°C , 10 h.

In conclusion, the cyclopropanation–ring enlargement sequence of enantiopure 1,2-oxazines leads to homologous 1,2-oxazepine derivatives, which should have a high potential for diversity orientated chemistry.¹⁰

Acknowledgment

Generous support of this work by the Deutsche Forschungsgemeinschaft, the Deutscher Akademischer Austauschdienst (fellowship for A.H.), the Fonds der Chemischen Industrie and the Schering AG is most gratefully acknowledged. We thank Dr. W. Schade for preliminary experiments and M. Brasholz for help during the preparation of this manuscript.

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- Typical Procedure, conversion of *anti*-1a into *anti*-3a.** A solution of NaOH (0.60 g) and KF (4.10 g) in H_2O (4.10 mL) was added to a vigorously stirred solution of *anti*-1a (0.50 g, 1.64 mmol) in CHBr_3 (2.70 mL) containing benzytriethylammonium chloride (5.5 mg). The biphasic mixture was stirred for 2 d at r.t. and then diluted with H_2O (8 mL) and extracted with Et_2O . The combined ethereal extracts were washed with brine, dried (Na_2SO_4) and concentrated. The excess CHBr_3 was removed in vacuum. The residue was purified by column chromatography (silica gel, hexane– EtOAc , 4:1) to give **2a** as colorless liquid (511 mg, 65%, dr 68:32).
Dibromocyclopropane derivative **2a** was refluxed for 20 h in a solution of anhyd K_2CO_3 (0.87 g, 6.27 mmol) in MeOH (7 mL) under Ar atmosphere. The mixture was cooled to r.t., diluted with H_2O and extracted with CH_2Cl_2 . The combined organic extracts were dried (Na_2SO_4) and concentrated. The product was purified by column chromatography (silica gel, hexane– EtOAc , 9:1) to yield 241 mg (53%) of *anti*-3a as colorless oil.
Analytical data for (3*R*,4'*S*)-2-benzyl-5-bromo-3-(2',2'-dimethyl-1',3'-dioxolan-4'-yl)-4,4-dimethoxy-2,3,4,7-tetrahydro-1,2-oxazepine: $[\alpha]_{\text{D}}^{22}$ –84.2 (c 0.45, CHCl_3). ^1H NMR (500 MHz, CDCl_3): δ = 1.33, 1.36 (2 s, 3 H each, Me), 3.19, 3.29 (2 s, 3 H each, OMe), 3.28 (d, J = 5.6 Hz, 1 H, 3-H), 4.07 (dd, J = 8.4, 8.8 Hz, 1 H, 5'-H), 4.11 (dd, J = 8.0, 8.4 Hz, 1 H, 5'-H), 4.29, 4.34 (2 d, J = 14.0 Hz, 1 H each,

- NCH₂), 4.31 (m, 1 H, 4'-H), 4.37 (dd, $J = 1.1, 13.5$ Hz, 1 H, 7-H), 4.47 (dd, $J = 0.8, 13.5$ Hz, 1 H, 7-H), 6.70 (dd, $J = 0.8, 1.1$ Hz, 1 H, 6-H), 7.24–7.39 (m, 5 H, Ph) ppm. ¹³C NMR (126 MHz, CDCl₃): $\delta = 25.7, 26.3$ (2 q, CH₃), 49.0, 49.1 (2 q, OMe), 58.1 (t, NCH₂), 63.6 (t, C-7), 66.7 (d, C-3), 67.2 (t, C-5'), 74.9 (d, C-4'), 99.2 (d, C-2'), 108.5 (d, C-6), 108.9 (s, C-4), 127.1, 128.2, 128.6, 138.0 (3 d, s, Ph), 135.3 (s, C-5) ppm. IR (KBr): $\nu = 3090\text{--}3030$ (=C-H), 2985–2845 (C-H), 1635 (C=C) cm⁻¹. MS (EI, 80 eV, 100 °C): m/z (%) = 429 (1) [M]⁺, 427 (1) [M]⁺, 414 (1) [M - CH₃]⁺, 412 (1) [M - CH₃]⁺, 398 (0.4) [M - OCH₃]⁺, 396 (0.4) [M - OCH₃]⁺, 329 (6) [M - CH₅H₉O₂]⁺, 327 (6) [M - CH₅H₉O₂]⁺, 91 (100) [CH₂Ph]⁺. Anal. Calcd for C₁₉H₂₀BrNO₅ (428.3): C, 53.28; H, 6.12; N, 3.27. Found: C, 53.10; H, 5.91; N, 2.68. HRMS (EI, 80 eV, 100 °C): m/z calcd for C₁₉H₂₀⁷⁹BrNO₅: 427.09943; found: 427.09947; m/z calcd for C₁₉H₂₀⁸¹BrNO₅: 429.09744; found: 429.09738.
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