This article was downloaded by: [Iowa State University] On: 29 September 2014, At: 14:14 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/gpss20

Synthesis of 4-Aryl-2imino-2H-selenazolines by a Reaction of a a-(Selenocyanato)acetophenones With Anilines

Anja Bodtke^a, Madeleine Kandt^a, Wolf-Diethard Pfeiffer^a & Peter Langer^{bc}

^a Institut für Biochemie, Universität Greifswald , Greifswald, Germany

^b Institut für Chemie, Universität Rostock , Rostock, Germany

^c Leibniz-Institut für Katalyse e. V. an der Universität Rostock , Rostock, Germany Published online: 01 Feb 2007.

To cite this article: Anja Bodtke , Madeleine Kandt , Wolf-Diethard Pfeiffer & Peter Langer (2007) Synthesis of 4-Aryl-2-imino-2H-selenazolines by a Reaction of a a-(Selenocyanato)acetophenones With Anilines, Phosphorus, Sulfur, and Silicon and the Related Elements, 182:1, 209-217, DOI: <u>10.1080/10426500600892685</u>

To link to this article: http://dx.doi.org/10.1080/10426500600892685

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness,

or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden. Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions



Synthesis of 4-Aryl-2-imino-2H-selenazolines by a Reaction of α -(Selenocyanato)acetophenones With Anilines

Anja Bodtke Madeleine Kandt Wolf-Diethard Pfeiffer Institut für Biochemie, Universität Greifswald, Greifswald, Germany

Peter Langer

Institut für Chemie, Universität Rostock, Rostock, Germany, and Leibniz-Institut für Katalyse e. V. an der Universität Rostock, Rostock, Germany

4-Aryl-2-imino-2H-selenazolines have been prepared by a reaction of α -(selenocyanato)acetophenones with anilines.

Keywords Cyclization; heterocycles; selenium

Selenium represents an essential element for higher organisms.¹ As selenium-containing enzymes—i.e., *Glutathioneperoxidase* and 5'-*Deiodase type 1*—are very important for the organism, various diseases can result from selenium deficiency.^{2,3} As a consequence, seleniumcontaining heterocycles are of considerable biochemical and pharmacological relevance. In this context, the antitumor and antiviral agent C-glycosyl selenazole selenazofurin represents a prominent example.⁴ Selenium heterocycles are often less stable than the corresponding sulfur analogues. In addition, methods and conditions available for the synthesis of sulfur compounds often can not be applied to selenium. Therefore, the development of new methods for the synthesis of small selenium-containing building blocks is of considerable current interest.^{5,6}

Received May 30, 2006; accepted June 13, 2006.

Financial support from the state of Mecklenburg-Vorpommern (scholarship for A. B. and Landesforschungsschwerpunkt "Neue Wirkstoffe und Screeningverfahren") is gratefully acknowledged.

Address correspondence to Peter Langer, Institut für Chemie, Universität Rostock, Albert-Einstein Str. 3a, Rostock, 18059 Germany. E-mail: peter.langer@uni-rostock.de

2	\mathbb{R}^1	\mathbb{R}^2	Yield $(\%)^a$
a	H	H	59
b	Cl	H	33
c	Br	H	41
d	H	Ph	46

TABLE I The Synthesis of 2a-d

^aYields of isolated products.

useful represent synthetic build- α -(selenocyanato)ketones ing blocks for the synthesis of selenium heterocycles.⁷ 1.3.4cycloaddition of selenadiazoles have been prepared by [3+2] α -(selenocyanato)acetophenones with diazonium salts.⁸ Aliphatic selenium heterocycles are available by [4+2] cycloaddition of α -(selenocyanato)ketones with 1,3-butadienes.⁹ In addition, sodiumhydride mediated cyclizations of α -(selenocyanato)acetophenones five-membered selenium heterocycles that give have been reported.¹⁰ Herein, we report the synthesis of 4-aryl-2-imino-2Hselenazolines that are, to the best of our knowledge, the first reactions of α -(selenocyanato)acetophenones with amines.¹¹ (selenocyanato)acetophenones are readily available by a reaction of phenacylbromides with potassium selenocyanate (KSeCN).^{9b,10,12}

The known α -(selenocyanato)acetophenones **2a–d** were prepared by a reaction of phenacylbromides **1a–d** with KSeCN (Scheme 1, Table I). The acid-mediated cyclization of α -(selenocyanato)acetophenones **2a– d** with anilines **3a–c** afforded 4-aryl-2-imino-2*H*-selenazolines **4a–g** (Scheme 2, Table II). The formation of the latter can be explained by an acid-catalyzed attack of the amine onto the selenocyanato group, an attack of the imino group thus formed onto the carbonyl group, and the subsequent elimination of water. The reaction of α -(selenocyanato)acetophenone **2b** with hydroxylamine hydrochloride afforded the selenazoline **4h** (Scheme 3). During optimization of reaction conditions, employment of amines in the form of their hydrochlorides proved to be important. In addition, the reaction time and the solvent played an important role. The relatively low yields can be explained by the unstable nature of the products and decomposition during the reaction.



SCHEME 1 The cyclization of α -(selenocyanato)acetophenones.



SCHEME 2 The cyclization of α -(selenocyanato)acetophenones with phenacyl bromides.

In summary, 4-aryl-2-imino-2*H*-selenazolines have been prepared by a reaction of α -(selenocyanato)acetophenones with anilines.

EXPERIMENTAL

General Comments

All solvents were dried by standard methods, and all reactions were carried out under an inert atmosphere. For 1 H and 13 C NMR, the deuterated solvents indicated were used. Mass spectrometric data (MS) were

4	\mathbb{R}^1	\mathbb{R}^2	\mathbb{R}^3	\mathbb{R}^4	\mathbb{R}^5	Yield (%) ^a
a	Н	Н	Н	Cl	Н	27
b	н	\mathbf{H}	Cl	\mathbf{H}	Cl	21
с	Cl	н	н	н	н	13
d	Cl	\mathbf{H}	н	Cl	\mathbf{H}	15
е	\mathbf{Br}	н	Cl	н	Cl	2
f	\mathbf{Br}	\mathbf{H}	н	Cl	\mathbf{H}	b
g	Η	Ph	Η	Cl	Η	10

TA	BLE	Π	The	Synt	hesis	of	4a-g
----	-----	---	-----	------	-------	----	------

^aYields of isolated products.

^bProduct contains 4-chloroaniline hydrochloride.

Downloaded by [Iowa State University] at 14:14 29 September 2014



SCHEME 3 The cyclization of α -(selenocyanato)acetophenones with phenacyl bromides.

obtained by electron ionization (70 eV), chemical ionization (CI, H_2O), or electrospray ionization (ESI). For preparative scale chromatography, silica gel (60–200 mesh) was used. Melting points are uncorrected.

Synthesis of Phenacyl Selenocyanates 2a-d

The synthesis of 2a-d by a reaction of potassium selenocyanate with phenacyl bromides has been previously reported.⁷⁻¹²

Phenacyl Selenocyanate (2a)

A mixture of potassium selenocyanate (2.88 g, 20.0 mmol) and phenacyl bromide (3.96 g, 20.0 mmol) in aceton (30 mL) was refluxed for 4 h. The hot mixture was filtrated, the solution concentrated in vacuo, and the resulting solid was isolated by filtration and recrystallization (EtOH). Yield: 2.63 g (59%), yellow prisms, m.p. 121°C (ethanol). IR (KBr): $\tilde{\nu} = 3002, 2945$ (w), 2155 (m), 1666 (s), 1594, 1578 (m), 1449, 1374 (s), 1324, 1310, 1287 (m), 1266, 1195, 996, 755, 688 (s) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 4.94$ (s, 2 H, CH₂), 7.51–7.71 (m, 3 H, Ar), 7.94–7.98 (m, 2 H, Ar). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 35.5$ (CH₂), 103.8 (C), 128.8, 128.9 (CH), 129.2, 134.1, 134.2, 194.6 (C). MS (EI, 70 eV): m/z (%) = 225 (3, [M+H]⁺ (⁸⁰Se)), 139 (4), 105 (100, [C₆H₅CO]⁺), 91 (8), 77 (35, [C₆H₅]⁺), 51 (14). Anal. calcd. for C₉H₇NOSe (224.12): C, 48.23; H, 3.15; N, 6.25. Found: C, 48.42; H, 3.65; N, 6.41.

4-Chlorophenacyl Selenocyanate (2b)

Starting with potassium selenocyanate (2.88 g, 20.0 mmol) and 4chlorophenacyl bromide (4.64 g, 20.0 mmol) in acetone (40 mL), product **2b** was isolated by filtration and recrystallization (EtOH). Yield: 1.72 g (33%), orange needles, m.p. 137°C (ethanol). IR (KBr): $\tilde{\nu} = 2989$, 2935 (w), 2154 (m), 1659, 1590 (s), 1571 (m), 1490, 1405, 1386 (w), 1313 (m), 1294, 1181, 1094 (s), 1015 (w), 999, 819 (s) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 4.88$ (s, 2 H, CH₂), 7.49–7.54 (m, 2 H, Ar), 7.88–7.92 (m, 2 H, Ar). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 35.0$ (CH₂), 103.8 (C), 129.0, 130.7 (CH), 132.9, 139.1, 193.9 (C). MS (EI, 70 eV): m/z (%) = 259 (4, [M+H]⁺ (⁸⁰Se)), 139 (100), 110 (32). Anal. calcd. for C₉H₆ClNOSe (258.56): C, 41.81; H, 2.34; N, 5.42. Found: C, 43.02; H, 2.54; N, 5.60.

4-Bromophenacyl Selenocyanate (2c)

Starting with potassium selenocyanate (2.88 g, 20.0 mmol) and 4bromophenacyl bromide (5.52 g, 20.0 mmol) in acetone (40 mL), **2c** was isolated (2.46 g, 41%) by filtration and recrystallization (EtOH) as yellow prisms, m.p. 124°C (ethanol). IR (KBr): $\tilde{\nu} = 2988$, 2934 (w), 2154 (m), 1658, 1585 (s), 1567 (m), 1486 (s), 1402, 1313 (w), 1289, 1180, 1071, 997, 811 (m) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 4.88$ (s, 2 H, CH₂), 7.66–7.71 (m, 2 H, Ar), 7.80–7.84 (m, 2 H, Ar). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 35.0$ (CH₂), 103.8 (C), 128.3, 130.8 (CH), 132.0, 133.3, 194.5 (C). MS (EI, 70 eV): m/z (%) = 302 (5, [M + H]⁺ (⁸⁰Se)), 234 (22), 183 (100), 155 (21), 77 (18). Anal. calcd. for C₉H₆BrNOSe (303.02): C, 35.78; H, 2.00; N, 4.64. Found: C, 35.89; H, 2.26; N, 3.80.

Desyl Selenocyanate (2d)

Starting with potassium selenocyanate (2.88 g, 20.0 mmol) and desyl bromide (5.5 g, 20.0 mmol) in acetone (50 mL), **2d** was isolated (2.75 g, 46%) by filtration and recrystallization (EtOH) as red needles, m.p. 124°C (ethanol). IR (KBr): $\tilde{\nu} = 3051, 2959$ (w), 2150 (m), 1666 (s), 1595, 1450 (m), 1275 (s), 1143, 1004, 764 (m), 697, 620 (s) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 7.23$ (s, 1 H, CH), 7.29–7.51 (m, 7 H, Ar), 7.59–7.65 (m, 1 H, Ar), 7.96–7.98 (m, 2 H, Ar). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 58.1$ (CH), 104.0 (C), 128.6, 128.9, 129.3, 129.5, 129.6, 129.7 (CH), 133.0, 134.5, 136.6, 195.5 (C). MS (EI, 70 eV): m/z (%) = 301 (6, [M + H]⁺ (⁸⁰Se)), 195 (74), 167 (88), 151 (63), 105 (100, [C₆H₅CO]⁺), 90 (66), 77 (84, [C₆H₅]⁺). Anal. calcd. for C₁₅H₁₁NOSe (300.22): C, 60.01; H, 3.69; N, 4.67. Found: C, 60.03; H, 4.38; N, 4.37.

3-(4-Chlorophenyl)-4-phenyl-3*H*-(selenazol-2-ylidene)amine Hydrochloride (4a)

A mixture of 2a (1.12 g, 5.0 mmol), and 4-chloroaniline hydrochloride (0.82 g, 5.0 mmol) in ethanol (50 mL) was refluxed for 5 h. The precipitated product was isolated by filtration, washed (ether), and dried in vacuo. Yield: 0.50 g (27%), green to grey prisms, m.p. 245–252°C (decomp., EtOH). IR (KBr): $\tilde{\nu} = 3110$ (m), 3051, 3009, 1615, 1528, 1490 (s), 1087, 885, 744, 724, 696 (m) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 7.17-7.31$ (m, 6 H, 5-H, Ar-H), 7.48–7.57 (m, J = 9.0 Hz, 4 H, Ar-H), 8.98 (bs, 1 H, NH), 10.52 (bs, 1 H, NH). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 108.1$ (C-5), 128.2, 129.2, 129.6, 130.3 (CH), 130.5 (C), 130.7 (CH), 133.5, 135.0, 140.4, 173.8 (C). MS (EI, 70 eV): m/z (%) = 336 (40), 335 (47), 334 (94, [M–HCl]+ (³⁵Cl, ⁸⁰Se)), 333 (83), 332 (52), 331 (52), 330 (24), 214 (27), 102 (100). Anal. calcd. for C₁₅H₁₂Cl₂N₂Se (370.14): C, 48.68; H, 3.27; N, 7.57. Found: C, 48.37; H, 3.70; N, 7.69.

3-(3,5-Dichlorophenyl)-4-phenyl-3*H*-(selenazol-2ylidene)amine Hydrochloride (4b)

A mixture of **2a** (1.12 g, 5.0 mmol), 3,5-dichloroaniline (0.81 g, 5.0 mmol), and 3 drops of conc. hydrochloric acid in ethanol (30 mL) was refluxed for 10 h. The precipitated product was isolated by filtration, washed (ether), and dried in vacuo. Yield: 0.42 g (21%), grey prisms, m.p. 263–269°C (decomp., EtOH). IR (KBr): $\tilde{\nu} = 3121$, 3045, 2987 (m), 1627 (s), 1577, 1529, 1433 (m), 806, 744, 711, 696, 665 (w) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 7.20-7.32$ (m, 6 H, 5-H, Ar), 7.72–7.75 (m, 3 H, Ar), 9.23 (bs, 1 H, NH), 10.61 (bs, 1 H, NH). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 108.3$ (C-5), 128.1, 128.2, 129.2, 129.5, 130.2 (CH), 134.9, 136.5, 139.7, 173.7 (C). MS (EI, 70 eV): m/z (%) = 372 (13), 371 (22), 370 (55), 369 (69), 368 (88, [M-HCl]⁺ (³⁵Cl, ⁸⁰Se)), 367 (100), 366 (50), 365 (53), 364 (24), 363 (13), 261 (11), 260 (35), 252 (10), 250 (53), 248 (82), 182 (19), 146 (18), 145 (27), 102 (59). C₁₅H₁₁Cl₃N₂Se (404.58).

3-Phenyl-4-(4-chlorophenyl)-3*H*-(selenazol-2-ylidene)amine Hydrochloride (4c)

A mixture of **2b** (1.30 g, 5.0 mmol) and aniline hydrochloride (0.65 g, 5.0 mmol) in ethanol (60 mL) was refluxed for 10 h. The precipitated product was isolated by filtration, washed (ether), and dried in vacuo. Yield: 0.24 g (13%), grey prisms, m.p. 268–271°C (decomp., EtOH). IR (KBr): $\tilde{\nu} = 3023$, 2961, 2937 (m), 1614 (m), 1527 (w), 1494 (s), 1090 (w), 744 (m) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 7.19-7.20$ (m, 2 H, Ar), 7.30–7.32 (m, 2 H, Ar), 7.34 (s, 1 H, 5-H), 7.41–7.50 (m, 5 H, Ar), 9.60 (bs, NH). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 108.7$ (C-5), 127.7, 128.1 (CH), 129.2 (C), 129.8, 130.0, 130.8 (CH), 133.6, 134.1, 138.9, 173.4 (C). MS (EI, 70 eV): m/z (%) = 335 (14), 334 (36, [M–HC1]⁺ (³⁵Cl, ⁸⁰Se)), 333 (51), 332 (20), 331 (27), 214 (62), 77 (100). C₁₅H₁₂Cl₂N₂Se (370.14).

3-(4-Chlorophenyl)-4-(4-chlorophenyl)-3*H*-(selenazol-2ylidene)amine Hydrochloride (4d)

A mixture of **2b** (1.30 g, 5.0 mmol) and of 4-chloroaniline hydrochloride (0.82 g, 5.0 mmol) in ethanol (20 mL) was refluxed for 5 h. The precipitated product was isolated by filtration, washed (ether), and dried in vacuo. Yield: 0.31 g (15%), brown prisms, m.p. 255–275°C (decomp., EtOH). IR (KBr): $\tilde{\nu} = 3111$, 2986, 1618, 1529, 1489 (s), 1093, 841 (m) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 7.19-7.23$ (m, 2 H, Ar-H), 7.33 (s+d, ²*J*(¹H,⁷⁷Se) = 41.3 Hz, 1 H, 5-H), 7.34–7.37 (m, 2 H, Ar), 7.49–7.59 (m, 4 H, Ar), 8.95 (bs, 1 H, NH), 10.38 (bs, 1 H, NH). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 108.8$ (C-5), 128.3 (CH), 129.4 (C), 130.4, 130.7, 131.4 (CH), 133.4, 134.1, 135.1, 139.2, 173.7 (C). MS (EI, 70 eV): m/z (%) = 372 (2), 371 (3), 370 (11), 369 (11), 368 (20, [M-HC1]⁺ (³⁵Cl, ⁸⁰Se)), 367 (16), 366 (10), 295 (17), 248 (23), 103 (100), 75 (57). C₁₅H₁₁Cl₃N₂Se (404.58).

3-(3,5-Dichlorophenyl)-4-(4-bromophenyl)-3*H*-(selenazol-2ylidene)amine Hydrochloride (4e)

A mixture of **2c** (1.51 g, 5.0 mmol), 3,5-dichloroaniline (0.81 g, 5.0 mmol), and 3 drops conc. hydrochloric acid in ethanol (30 mL) was refluxed for 10 h. The precipitated product was isolated by filtration, washed (ether), and dried in vacuo. Yield: 40 mg (2%), yellow prisms, m.p. 155– 170°C (decomp., EtOH). IR (KBr): $\tilde{\nu} = 3073$, 3053 (w), 1604, 1577 (s), 1485, 1432, 1346, 1114, 1058, 831, 802, 748, 744 (m) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 6.64$ (s, 1 H, 5-H), 7.11–7.13 (m, 2 H, Ar), 7.25 (d, J = 1.8 Hz, 2 H, Ar), 7.45–7.48 (m, 3 H, Ar), 8.88 (bs, 1 H, NH). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 100.1$ (C-5), 121.5 (C), 126.8, 128.1, 130.1, 131.3 (CH), 132.0, 133.5, 137.8, 140.7, 160.2 (C). MS (EI, 70 eV): m/z (%) = 450 (26), 449 (35), 448 (60), 447 (69, [M–HCI]⁺ (⁷⁹Br, ³⁵Cl, ⁸⁰Se)), 446 (59), 445 (61), 444 (30), 443 (23), 340 (28), 338 (17), 330 (29), 328 (65), 326 (39), 146 (43), 145 (68), 102 (23). C₁₅H₁₀BrCl₃N₂Se (483.48).

3-(4-Chlorophenyl)-4-(4-bromophenyl)-3*H*-(selenazol-2ylidene)amine Hydrochloride (4f)

A mixture of **2c** (1.51 g, 5.0 mmol) and of 4-chloroaniline hydrochloride (0.82 g, 5.0 mmol) in ethanol (20 mL) was refluxed for 5 h. The precipitated product was isolated by filtration, washed (ether), and dried in vacuo. Yield: 0.25 g (4-chloroanilinium chloride could not be separated, **4f**/4-ClC₆H₄NH₃Cl = 1.3:1). IR (KBr): $\tilde{\nu} = 3112$, 3000, 1615, 1527 (m), 1489 (s), 1094 (m), 840, 822 (w) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz):

$$\begin{split} &\delta=7.11-7.15~(\text{m}, 2~\text{H}, \text{Ar}),~7.32~(\text{s}, 1~\text{H}, 5\text{-H}),~7.43-7.58~(\text{m}, 6~\text{H}, \text{Ar}),\\ &8.98~(\text{bs}, 1~\text{H}, \text{NH}),~10.81~(\text{bs}, 1~\text{H}, \text{NH}).~^{13}\text{C}~\text{NMR}~(\text{DMSO-d}_6,~75~\text{MHz});\\ &\delta=109.1~(\text{C-5}),~122.9~(\text{C}),~123.4,~129.5~(\text{CH}),~129.8,~129.9~(\text{C}),~130.4,\\ &130.8,~131.3,~131.7~(\text{CH}),~133.4,~134.2,~135.2,~139.1,~173.9~(\text{C}).~\text{MS}~(\text{EI},~70~\text{eV});~\text{m/z}~(\%)=417~(27),~416~(35),~415~(87),~414~(79),~413~(100,~[\text{M-Cl}]^+\\ &(^{79}\text{Br},~^{35}\text{Cl},~^{80}\text{Se})),~411~(80),~410~(51),~409~(37),~408~(16),~296~(22),~294\\ &(97),~292~(71).~\text{C}_{15}\text{H}_{11}\text{BrCl}_2\text{N}_2\text{Se}~(449.04). \end{split}$$

3-(4-Chlorophenyl)-4,5-diphenyl-3*H*-(selenazol-2-ylidene)amine Hydrochloride (4g)

A mixture of **2d** (1.50 g, 5.0 mmol) and 4-chloroaniline hydrochloride (0.82 g, 5.0 mmol) in ethanol (25 mL) was refluxed for 10 h. The precipitated product was isolated by filtration, washed (ether), and dried in vacuo. Yield: 0.22 g (10%), colourless prisms, m.p. 255–270°C (decomp., EtOH). IR (KBr): $\tilde{\nu} = 3191$ (m), 3023 (w), 1614, 1598, 1575, 1490, 1402, 1341 (m), 1090 (w), 732, 697 (m) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 7.12-7.29$ (m, 10 H, Ar), 7.49–7.57 (m, 4 H, Ar), 9.18 (bs, 1 H, NH), 10.87 (bs, 1 H, NH). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 122.0$ (C-5), 128.3, 128.6, 128.7, 129.3, 129.7, 130.0, 130.8, 131.1 (Ar-H), 133.7, 134.9, 135.6, 170.8 (C). MS (EI, 70 eV): m/z (%) = 414 (12), 413 (43), 412 (45), 411 (100), 410 (70, [M–HCl]⁺ (³⁵Cl, ⁸⁰Se)), 408 (55), 407 (42), 406 (25), 405 (10), 216 (33), 215 (14), 214 (100), 178 (38), 165 (19). Anal. calcd. for C₂₁H₁₆Cl₂N₂Se (446.24): C, 56.52; H, 3.61; N, 6.28. Found: C, 56.25; H, 3.81; N, 6.07.

4-(4-Chlorophenyl)-2-iminoselenazol-3-ol Hydrochloride (4h)

A mixture of **2b** (1.30 g, 5.0 mmol) and hydroxyl amine hydrochloride (0.35 g, 5.0 mmol) in ethanol (50 mL) was refluxed for 10 h. The precipitated product was isolated by filtration, washed (ether), and dried in vacuo. Yield: 0.27 g (17%), colourless prisms, m.p. 215–223°C (decomp., EtOH). IR (KBr): $\tilde{\nu} = 3416$, 3284, 3244, 3176 (m), 3079 (s), 2679, 2622 (m), 1606 (s), 1545, 1485, 1092, 839, 740 (m) cm⁻¹. ¹H NMR (DMSO-d₆, 300 MHz): $\delta = 7.25$ (s+d, ²J(¹H,⁷⁷Se) = 40.7 Hz, 1 H, 5-H), 7.54–7.58 (m, J = 8.6 Hz, 2 H, Ar), 7.68–7.72 (m, J = 8.6 Hz, 2 H, Ar), 9.88 (bs, 2 H, 2 NH), 12.81 (bs, 1 H, OH). ¹³C NMR (DMSO-d₆, 75 MHz): $\delta = 104.6$ (¹J(¹³C,⁷⁷Se) = 108.6 Hz, C-5), 128.0 (C), 128.4, 130.4 (Ar), 134.2, 137.4 (C), 167.1 (¹J(¹³C,⁷⁷Se) = 151.2 Hz, CN). C₉H₈Cl₂N₂OSe (310.04).

REFERENCES

- [1] K. Schwar and C. M. Foltz, J. Am. Chem. Soc., 79, 3292 (1957).
- [2] G. C. Mills, J. Biol. Chem., 229, 189 (1957).
- [3] J. T. Rotruck, A. E. Pope, H. E. Ganther, A. B. Swanson, D. G. Hafemann, and W. G. Hoekstra, *Science*, **179**, 588 (1973).
- [4] B. M. Goldstein, S. D. Kennedy, and W. J. Hennen, J. Am. Chem. Soc., 112, 8265 (1990), and references cited therein.
- [5] (a) R. Larsen, In Comprehensive Heterocyclic Chemistry II, eds. I. Shinkai, A. R. Katritzky, C. W. Rees, and E. F. V. Scriven, eds., Vol. 3, pp. 493–510 (Elsevier Science, Oxford, 1996); (b) W. D. Pfeiffer, In Science of Synthesis, ed. E. Schaumann, Vol. 11, pp. 941–989 (Thieme Verlag, Stuttgart, New York, 2002); (c) I. Lalezari and M. Shafiee, In Comprehensive Heterocyclic Chemistry, eds. A. R. Katritzky, C. W. Rees, and K. T. Potts, Vol. 6, pp. 333–363 (Elsevier Science, Oxford, 1984); (d) T. Wirth, In Modern Developments in Organic Synthesis (Springer, Berlin, 2000); (e) M. Koketsu and H. Ishihara, Curr. Org. Chem., 7, 175 (2003).
- [6] For selenium heterocycles from our laboratory, see (a) K. Geisler, A. Jacobs, A. Künstler, M. Mattes, I. Girrleit, B. Zimmermann, E. Bulka, W.-D. Pfeiffer, and P. Langer, Synlett, 1983 (2002); (b) K. Geisler, W.-D. Pfeiffer, C. Müller, E. Nobst, E. Bulka, and P. Langer, Synthesis, 1215 (2003); (c) K. Geisler, A. Künstler, E. Bulka, W.-D. Pfeiffer, and P. Langer, Synlett., 1195 (2003); (d) K. Geisler, A. Künzler, E. Bulka, W.-D. Pfeiffer, and P. Langer, Synthesis, 97 (2004); (e) K. Geisler, W.-D. Pfeiffer, A. Künzler, E. Bulka, and P. Langer, Synthesis, 875 (2004); (f) H. Below, W.-D. Pfeiffer, K. Geisler, M. Lalk, and P. Langer, Eur. J. Org. Chem., 3637 (2005).
- [7] G. Hofmann, Justus Liebigs Ann. Chem., 307, 250 (1889).
- [8] (a) M. Takahashi and M. Kurosawa, *Bull. Chem. Soc. Jpn.*, **53**, 1185 (1980); (b) H. M. Hassaneen, A. O. Abdelhamid, A. Shetta, and A. S. Shawali, *Gazz. Chim. Ital.*, **112**, 545 (1982).
- [9] (a) P. T. Meinke and G. A. Krafft, *Tetrahedron Lett.*, 28, 5121 (1987); (b) G. W. Kirby and A. N. Trethewey, *J. Chem. Soc. Perkin Trans.* 1, 1913 (1988); (c) T. Kataoka, Y. Ohe, A. Umeda, T. Iwamura, M. Yoshimatsu, and H. Shimizu, *Chem. Pharm. Bull.*, 42, 811 (1994).
- [10] (a) J. Gramza, R. B. Mitchell, and D. C. Dittmer, J. Org. Chem., 49, 2057 (1984);
 (b) M. D. Otero, B. Batanero, and F. Barba, *Tetrahedron*, 60, 4609 (2004).
- [11] For related heterocycles, see (a) J. Liebscher and H. Hartmann, Z. Chem., 16, 18 (1976); (b) E. Bulka and K.-D. Ahlers, Z. Chem., 349 (1963); (c) S. Bilinski and M. Chmielewski, Ann. Univ. Mariae Curie-Sklodowska Sect. D, 32, 231 (1977); (d) A. Shafiee and I. Lalezari, J. Heterocycl. Chem., 12, 675 (1975); (e) M. Morvan, G. Nadler and R. G. Zimmermann, J. Heterocycl. Chem., 28, 1365 (1991); (f) K. Szulzewsky, W.-D. Pfeiffer, E. Bulka, H. Rossberg, and B. Schulz, Acta Chem. Scand., 47, 302 (1993); (g) Z. Casar, A. M.-L. Marechal, and D. Lorcy, New J. Chem., 27, 1622 (2003).
- [12] (a) P. T. Meinke and G. A. Krafft, J. Am. Chem. Soc., 110, 8671 (1988); (b) F. Asinger, and M. K. Schmitz, Monatsh. Chem., 113, 1191 (1982); (c) V. Nair, A. Augustine, and T. G. George, Eur. J. Org. Chem., 14, 2363 (2002).