



Tetrahedron Letters 44 (2003) 8857-8859

TETRAHEDRON LETTERS

Total synthesis of macrosphelide A by way of palladium-catalyzed carbonylative esterification

Shin-ichi Kusaka, Suguru Dohi, Takayuki Doi and Takashi Takahashi*

Department of Applied Chemistry, Tokyo Institute of Technology, 2-12-1, Ookayama, Meguro, Tokyo 152-8552, Japan Received 31 July 2003; revised 3 September 2003; accepted 22 September 2003

Abstract—We achieved the total synthesis of macrosphelide A, as part of a combinatorial library of its analogues. The key intermediate, the *seco*-acid derivative, was prepared from the corresponding vinyl iodide using sequential carbonylative esterification.

© 2003 Elsevier Ltd. All rights reserved.

Macrosphelide A–L are a family of compounds isolated from two different culture medium of *Macrospaeropsis* sp. FO-5050 and/or *pericania byssoides* by the Omura¹ and Numata² groups. Macrosphelides are 16-membered macrolides including three ester linkages consisting of one β -hydroxybutyric acid and two 5-hydroxy-2hexenoic acid units and strongly inhibit the adhesion of human leukemia HL-60 cells to human-umbilical-vein endothelial cell (HUVEC) in dose-dependent fashion.^{1–4} In order to prepare a variety of their analogues, a simple synthetic method is required.^{3,5–7} As a part of the synthesis of a macrosphelide library, we wish to report the total synthesis macrosphelide A via sequential carbonylative esterification.

An outline of the synthetic strategy is described in Scheme 1. The target molecule 1 would be synthesized through macrolactonization of *seco*-acid derivative **2**. Compound **2** can be prepared from **3** and **4**^{6c} by applying alkoxycarbonylation^{8,9} as the key reaction. Since palladium-catalyzed carbonylative esterification of vinyl halides can be regarded as the synthetic equivalent of the formation of an α , β -unsaturated ester, we designed the vinyl iodide **3** as a masked activated ester for this synthesis which can be repeatedly utilized in the formation of remaining two ester linkages in **1**.

The (*E*)-vinyl iodide **3** was prepared from commercially available methyl (*S*)-lactate **5**, which was converted into the ynone **6** in three steps. Chelate-controlled reduction with dimethylaluminium chloride¹⁰ provided the *ery*-*thro*-alcohol **7** with the required stereochemistry (>95%). Deprotection of the TMS group and subsequent protection as the methoxyethoxymethyl ether gave



Scheme 1. Retrosynthetic analysis of macrosphelide A (1). MEM = methoxyethoxymethyl, TBS = tert-butyldimethylsilyl, Tce = 2,2,2-trichloroethyl.

Keywords: macrolactone; palladium catalyst; carbonylation; natural product synthesis.

^{*} Corresponding author. Tel.: +81-3-5734-2120; fax: +81-3-5734-2884; e-mail: ttak@apc.titech.ac.jp

^{0040-4039/\$ -} see front matter @ 2003 Elsevier Ltd. All rights reserved. doi:10.1016/j.tetlet.2003.09.186



Scheme 2. Synthesis of (*E*)-vinyl iodide 3. (a) TBSCl, imidazole, CH_2Cl_2 , rt, 6 h; (b) Me(OMe)NH·HCl, *i*-PrMgCl, THF, 0°C, 30 min; (c) trimethylsilyl acetylene, BuLi, THF, -78°C, 30 min; (d) Me₂AlCl, Bu₃SnH, CH₂Cl₂, -78°C, 1 h; (e) MEMCl, *i*-Pr₂NEt, rt, 1.5 h; (f) K₂CO₃, MeOH–H₂O, rt, 24 h; (g) PdCl₂(PPh₃)₂, Bu₃SnH, THF, 0°C, 30 min; (h) NIS, THF, 0°C to rt, 10 min.

alkyne 8. Treatment of 8 with tributyltinhydride in the presence of a catalytic amount $PdCl_2(PPh_3)_2$ afforded the vinyl stannane¹¹ which was converted to the corresponding vinyl iodide with *N*-iodosuccinimide to afford vinyl iodide 3 in 55% overall yield (Scheme 2).

Protection of methyl (S)-3-hydroxybutyrate 9 as its TBS ether, transformation of the methyl ester to the 2,2,2-trichloroester, followed by desilylation generated the second building block 4. With the desired vinyl iodide 3 and alcohol 4 in hand, we next focused our attention to the palladium-catalyzed carbonylative esterification. The alkoxy carbonylation between 3 and 4 under the standard reaction conditions $(Pd(PPh_3)_4,$ CO 30 atm, NEt₃, DMAP, DMF) provided moderate conversion and produced the homo dimer of 3 and the α,β -unsaturated carboxylic acid as a β -elimination product via the desired ester 10 (Entry 1). Consequently, the carbonylation reaction was extensively optimized and the results are summarized in Table 1. The best result was obtained using PdCl₂(MeCN)₂, as a palladium catalyst, giving the desired ester 10 as the sole product in 87% yield (Entry 6).¹²

Desilylation of **10** produced alcohol **11** in 82% yield. Palladium-catalyzed carbonylative esterification of **11** was accomplished under the above conditions except using excess amount of vinyl iodide **3** (5 equiv) to provide **12** in 78% yield. Deprotection of the TBS and Tce groups furnished the key intermediate **13** (49%). Macrolactonization of **13** under various conditions (i.e. Yamaguchi,^{13a} Keck,^{13b} and Mitsunobu^{13c} method) yielded recovered starting material or the undesired 12-membered lactone that was formed via β-elimination



Scheme 3. Total synthesis of macrosphelide A (1) by carbonylation and macrolactonization. (a) TBSCl, imidazole, CH₂Cl₂, rt, 3 h; (b) 1 M NaOH in MeOH, rt, 12 h; (c) 2,2,2-trichloroethanol, DIC, DMAP, CH₂Cl₂, rt, 5 h; (d) HF·pyridine, CH₃CN, rt, 12 h; (e) TBAF, AcOH, CH₂Cl₂, rt, 24 h; (f) CO(30 atm), PdCl₂(MeCN)₂, Et₃N, DMAP, DMF, rt, 12 h; (g) AcOH–H₂O–THF (1:1:3), rt, 48 h; (h) Zn dust, NH₄OAc, THF–H₂O (3:1), rt, 2 h; (i) 2,2'-dipyridyl disulfide, PPh₃, toluene, rt, 1 h; AgOTf, rt, 30 min; (j) TFA, CH₂Cl₂. 6 h, rt; DIC=*N*,*N*-diisopropylcarbodiimide, DMAP=4-(dimethylamino)pyridine, DMF = *N*,*N*-dimethylformamide, Tf = trifluoroacetic acid.

60

72

81

87 37

60

60

rt

rt

rt

Entry	Catalyst	3:4	DMAP (equiv.)	Temp. (°C)	Yield of 10^{b} (%)
1	$Pd(PPh_3)_4$	1:3	1.0	60	57
2	$Pd(PPh_3)_4$	1:3	None	60	0

1.0

1.0

1.0

0.1

0.1

Table 1. Palladium-catalyzed carbonylation between vinyl iodide 3 and alcohol 4^a

1:3

1:3

1:3

1:3

3:1

^a Reaction was performed on a 0.5 mmol scale in DMF using 10 mol% Pd-catalyst in the presence of triethylamine for 6 h under 30 atm of carbon monoxide.

^b Isolated yield after silica-gel column chromatography

PdCl₂(PPh₃)₂

PdCl₂(MeCN)₂

PdCl₂(MeCN)₂

PdCl₂(MeCN)₂

PdCl₂(MeCN)₂

3

4

5

6

7

of β-alkoxyester followed by cyclization. The Mukaiyama–Corey method¹⁴ (2,2'-dipyridyl disulfide, PPh₃), provided the desired lactone **14**, however in disappointingly low yield. Extensive optimization of the Mukaiyama method increased the yield of the desired lactone **14** up to 40%. Addition of Ag-salt was crucial to activate the pyridinium thioester intermediate.¹⁴ Finally, deprotection of the MEM group with TFA in CH₂Cl₂ furnished **1** in 92% yield. The synthetic **1** exhibited ¹H and ¹³C NMR spectral data as well as optical rotation identical to those published for the natural product^{1b} (Scheme 3).

In summary, the total synthesis of macrosphelide A has been achieved with a highly convergent and efficient strategy. Further refinement of the synthetic scheme toward the synthesis of combinatorial library of its analogues will be reported.

Acknowledgements

This work was supported by a Grant-In-Aid from the Ministry of Education, Culture, Sports, Science and Technology, Japan (No. 14103013). We also thank Professors Satoshi Omura and Toshiaki Sunazuka (Kitasato University, Japan) for fruitful discussion.

References

- (a) Hayashi, M.; Kim, Y.-P.; Hiraoka, H.; Natori, M.; Takamatsu, S.; Kawakubo, T.; Masuma, R.; Komiyama, K.; Omura, S. J. Antibiot. 1995, 48, 1435–1439; (b) Takamatsu, S.; Kim, Y.-P.; Hayashi, M.; Hiraoka, H.; Natori, M.; Komiyama, K.; Omura, S. J. Antibiot. 1996, 49, 95–98; (c) Takamatsu, S.; Hiraoka, H.; Kim, Y.-P.; Hayashi, M.; Natori, M.; Komiyama, K.; Omura, S. J. Antibiot. 1997, 50, 878–880; (d) Fukami, A.; Taniguchi, Y.; Nakamura, T.; Rho, M.-C.; Kawaguchi, K.; Hayashi, M.; Komiyama, K.; Omura, S. J. Antibiot. 1999, 52, 501–504.
- (a) Numata, A.; Iritani, M.; Yamada, T.; Minoura, K.; Matsumura, E.; Yamori, T.; Tsuruo, T. *Tetrahedron Lett.* 1997, 38, 8215–8218; (b) Yamada, T.; Iritani, M.; Doi, M.; Minoura, K.; Ito, T.; Numata, A. J. Chem. Soc., Perkin Trans. 1 2001, 3046–3053; (c) Yamada, T.; Iritani, M.; Minoura, K.; Numata, A.; Kobayashi, Y.; Wang, Y.-G. J. Antibiot. 2002, 55, 147–154.
- Sunazuka, T.; Hirose, T.; Harigaya, Y.; Takamatsu, S.; Hayashi, M.; Komiyama, K.; Omura, S.; Sprengeler, P. A.; Smith, A. B., III J. Am. Chem. Soc. 1997, 119, 10247–10248.
- Fukami, A.; Iijima, K.; Hayashi, M.; Komiyama, K.; Omura, S. Biochem. Biophys. Res. Commun. 2002, 291, 1065–1070.
- (a) Kobayashi, Y.; Kumar, G. B.; Kurachi, T. Tetrahedron Lett. 2000, 41, 1559–1563; (b) Kobayashi, Y.;

Kumar, G. B.; Kurachi, T.; Acharya, H. P.; Yamazaki, T.; Kitazume, T. J. Org. Chem. 2001, 66, 2011–2018; (c) Kobayashi, Y.; Achaarya, H. P. Tetrahedron Lett. 2001, 42, 2817–2820; (d) Kobayashi, Y.; Wang, Y.-G. Tetrahedron Lett. 2002, 43, 4381–4384.

- 6. (a) Ono, M.; Nakamura, H.; Konno, F.; Akita, H. *Tetrahedron: Asymmetry* 2000, *11*, 2753–2764; (b) Nakamura, H.; Ono, M.; Shiba, Y.; Akita, H. *Tetrahedron: Asymmetry* 2002, *13*, 705–713; (c) Nakamura, H.; Ono, M.; Makino, M.; Akita, H. *Heterocycles* 2002, *57*, 327–336; (d) Akita, H.; Nakamura, H.; Ono, M. *Chirality* 2003, *15*, 352–359.
- Sharma, G. V. M.; Mouli, C. C. Tetrahedron Lett. 2002, 43, 9159–9161.
- (a) Schoenberg, A.; Bartoletti, I.; Heck, R. F. J. Org. Chem. 1974, 39, 3318–3326; (b) Schoenberg, A.; Heck, R. F. J. Org. Chem. 1974, 39, 3327–3331; (c) Hidai, M.; Hikata, T.; Wada, Y.; Fujikura, Y.; Uchida, Y. Bull. Chem. Soc. Jpn. 1975, 48, 2075–2077; (d) Ito, T.; Mori, K.; Mizoroki, T.; Ozaki, A. Bull. Chem. Soc. Jpn. 1975, 48, 2091–2094; (e) Stille, J. K.; Wong, P. K. J. Org. Chem. 1975, 40, 532–534.
- (a) Takahashi, T.; Nagashima, T.; Tsuji, J. Chem. Lett.
 1980, 369–372; (b) Takahashi, T.; Ikeda, H.; Tsuji, J. Tetrahedron Lett. 1980, 21, 3885–3888.
- (a) Evans, D. A.; Allison, B. D.; Yang, M. G.; Masse, C. E. J. Am. Chem. Soc. 2001, 123, 10840–10852; (b) Evans, D. A.; Allison, B. D.; Yang, M. G. Tetrahedron Lett. 1999, 40, 4457–4460; (c) Evans, D. A.; Halstead, D. P.; Allison, B. D. Tetrahedron Lett. 1999, 40, 4461–4462.
- (a) Zhang, H. X.; Guibe, F.; Balavoine, G. J. Org. Chem.
 1990, 55, 1857–1867; (b) Smith, N. D.; Mancuso, J.; Lautens, M. Chem. Rev. 2000, 100, 3257–3282.
- 12. Carbonylative esterification was carried out as follows: To a solution of vinyl iodide (0.50 mmol) and alcohol (1.50 mmol) in DMF (5.0 ml) was added Et₃N (0.55 mmol), DMAP (0.05 mmol), and PdCl₂(MeCN)₂ (0.05 mmol) under an argon. The vessel was placed in an autoclave, which was purged with CO three times before applying a pressure of 30 atm. After stirring at room temperature for 6 h, the catalyst was deposited through a short florisil plug. The filtrate was partitioned between EtOAc and 1 M HCI. The organic layer was washed with saturated aqueous NaHCO₃ and brine, dried over MgSO₄, and concentrated in vacuo. Column chromatography on silica gel afforded the product.
- (a) Inanaga, J.; Hirata, K.; Saeki, H.; Katsuki, T.; Yamaguchi, M. *Bull. Chem. Soc. Jpn.* **1979**, *52*, 1989– 1993; (b) Boden, E. P.; Keck, G. E. J. Org. Chem. **1985**, *50*, 2394–2395; (c) Mitsunobu, O. *Synthesis* **1981**, 1–28.
- 14. (a) Mukaiyama, T.; Usui, M.; Saigo, K. Chem. Lett.
 1976, 49–50; (b) Corey, E. J.; Brunelle, D. J.; Stork, P. J. Tetrahedron Lett. 1976, 17, 3405–3408; (c) Corey, E. J.; Nicolaou, K. C. J. Am. Chem. Soc. 1974, 96, 5614–5616; (d) Buszek, K. R.; Sato, N.; Jeong, Y. J. Am. Chem. Soc.
 1994, 116, 5511–5512.