



Convergent synthesis of the F–K ring segment of brevetoxin B

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Abstract—A convergent synthesis of the F–K ring segment of brevetoxin B has been achieved via the intramolecular allylation of an α -chloroacetoxy ether and subsequent ring-closing metathesis.

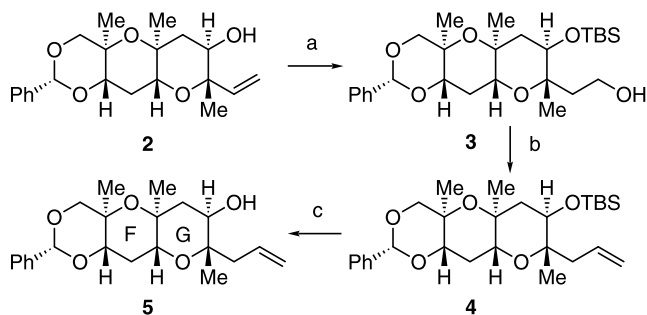
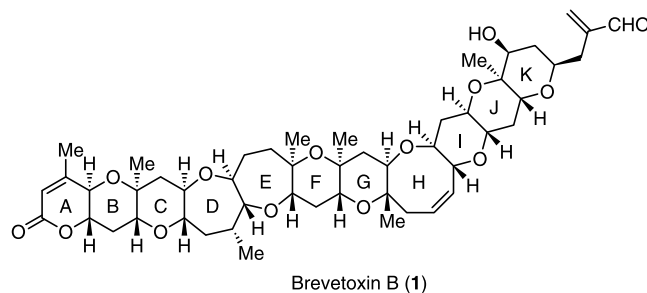
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Brevetoxin B (**1**), a potent neurotoxin was isolated from the red tide organism *Gymnodinium breve* Davis in 1981 as the first example of a marine polycyclic ether.¹ The unique structural features and biological activity of this molecule have attracted the attention of synthetic chemists.² In 1995, the first total synthesis of **1** was achieved by Nicolaou and co-workers using a hydroxy dithioacetal cyclization for the key segment connection constructing the H ring moiety.³ Nakata, et al. reported the second total synthesis based on the same segment coupling strategy in 2002.⁴ Recently, we developed an efficient method for the convergent synthesis of polycyclic ethers via the intramolecular allylation of an α -acetoxy ether and subsequent ring-closing metathesis.⁵ In this paper, we wish to report a new approach to the convergent synthesis of the F–K ring segment of **1** based on our own methodology.

Scheme 1 shows the synthesis of the FG ring segment. The known olefin **2**^{6d} was converted to the alcohol **3** via TBS protection and hydroboration in 86% yield. Swern oxidation of **3** followed by Wittig reaction gave **4** in 87% yield. Treatment of the silyl ether **4** with TBAF gave the FG ring segment **5** in 90% yield.

The synthesis of the JK ring segment is illustrated in Scheme 2. Ozonolysis of the known olefin **6**^{2d} afforded the corresponding aldehyde, which was subjected to the

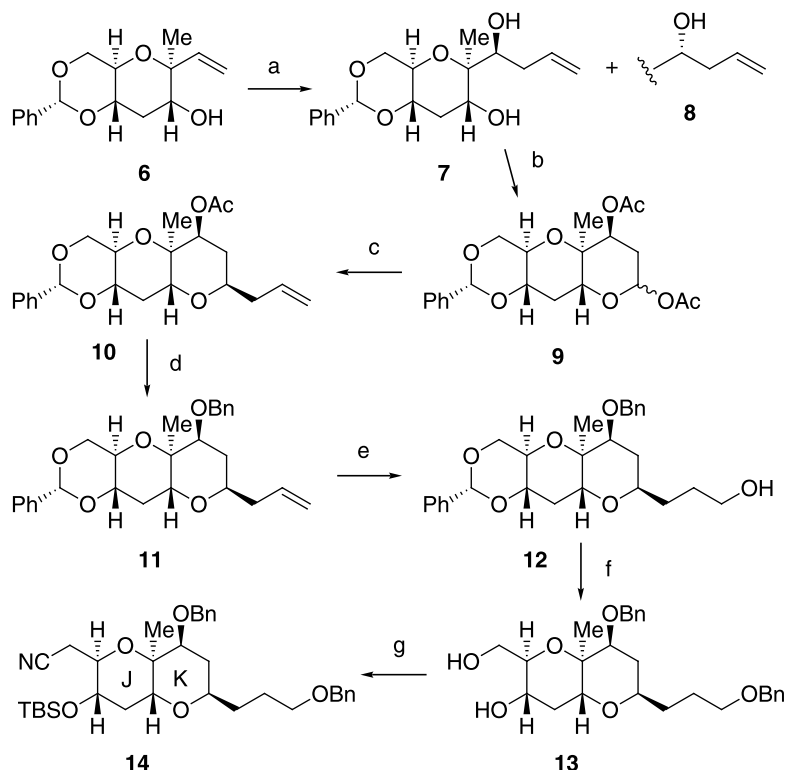
Barbier type allylation using allyl bromide and Zn powder in the presence of satd NH₄Cl to give a 2:1 mixture of the desired homoallylic alcohol **7** and its stereoisomer **8** in 93% combined yield.^{6,7} Ozonolysis of



Scheme 1. Reagents and conditions: (a) (i) TBSOTf, 2,6-lutidine, CH₂Cl₂, rt, 97%; (ii) 9-BBN, THF, rt, then 3N NaOH, 30% H₂O₂, 0°C, 89%; (b) (i) (COCl)₂, DMSO, CH₂Cl₂, –78°C, then Et₃N, –78°C to rt; (ii) Ph₃P⁺CH₃Br[–], NaHMDS, THF, 0°C, 87% (two steps); (c) TBAF, THF, rt, 90%.

Keywords: brevetoxin B; polycyclic ethers; convergent synthesis.

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Scheme 2. Reagents and conditions: (a) (i) O_3 , MeOH, -78°C , then Me_2S ; (ii) allyl bromide, Zn powder, satd NH_4Cl , THF, 0°C , 93% (two steps, **7**:**8** = 2:1); (b) (i) O_3 , CH_2Cl_2 , -78°C , then PPh_3 ; (ii) Ac_2O , pyridine, DMAP, CH_2Cl_2 , reflux, 97% (two steps); (c) allyltrimethylsilane, $\text{BF}_3\cdot\text{OEt}_2$, CH_3CN , 0°C , 65%; (d) (i) NaOMe , MeOH, rt, 98%; (ii) BnBr , KH, THF, rt, 100%; (e) 9-BBN, THF, rt, then 30% H_2O_2 , 3N NaOH , 0°C , 96%; (f) (i) BnBr , KH, THF, rt; (ii) CSA, MeOH, rt, 95% (two steps); (g) (i) TsCl , Et_3N , CH_2Cl_2 , reflux, 89%; (ii) NaCN , DMSO, 50°C , 100%; (iii) TBSOTf , 2,6-lutidine, CH_2Cl_2 , 0°C , 100%.

7 followed by acetylation of the resulting hemiacetal gave **9** in 97% yield. Treatment of **9** with allyltrimethylsilane and $\text{BF}_3\cdot\text{OEt}_2$ gave **10** as the sole product in 65% yield. Removal of the acetyl group followed by benzyl protection of the resulting alcohol afforded **11** in quantitative yield. Hydroboration of **11** gave the alcohol **12** in 96% yield. Benzyl protection followed by hydrolysis of the benzylidene acetal provided the diol **13** in 95% yield. Selective tosylation of the primary alcohol, treatment with sodium cyanide, and TBS protection of the remaining secondary alcohol gave the JK ring segment **14** in 89% yield.

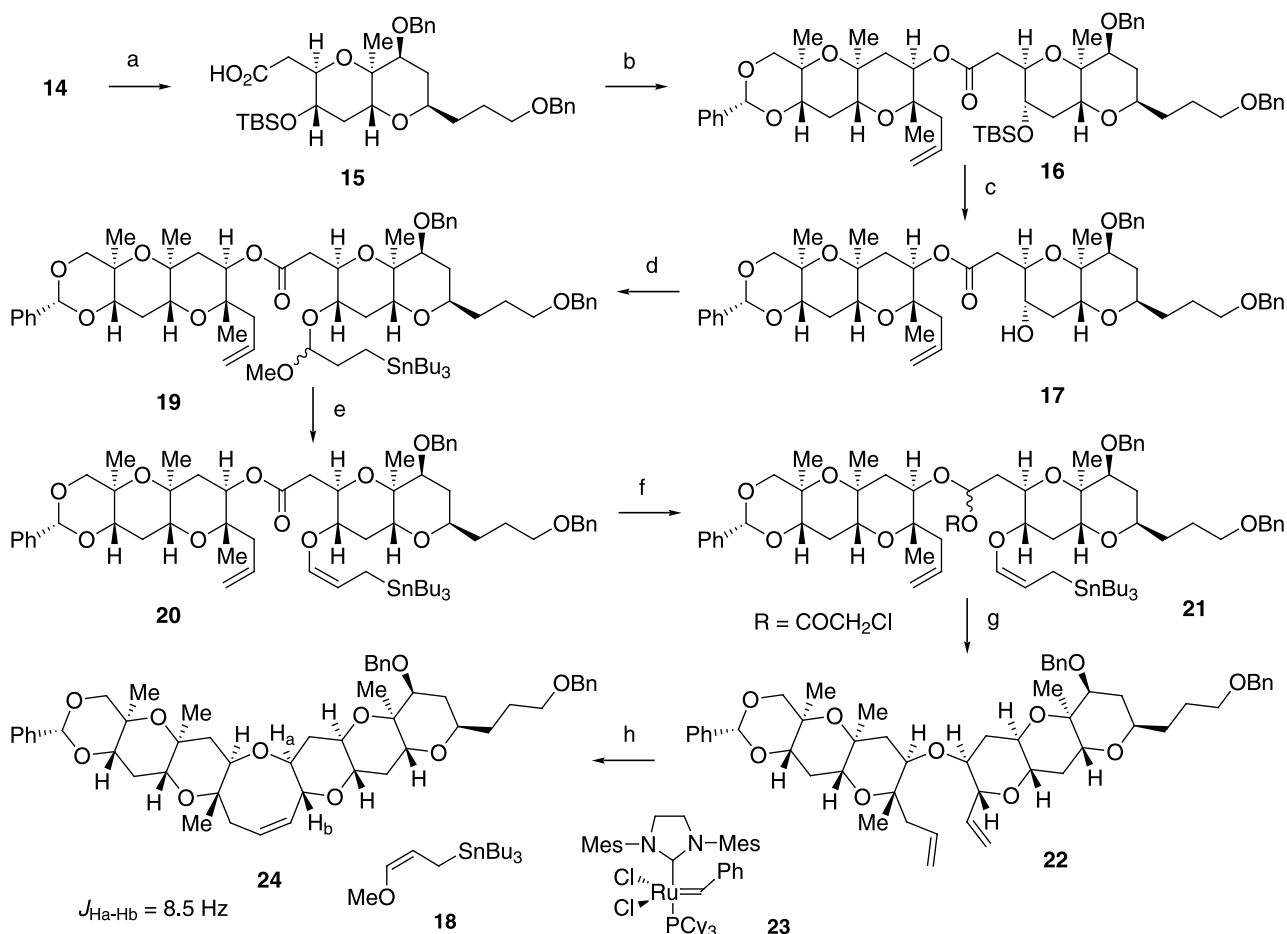
Scheme 3 describes the coupling of the FG and JK ring segments. DIBAL-H reduction of **14** followed by oxidation of the resulting aldehyde with NaClO_2 gave the carboxylic acid **15**, which was subjected to the Yamaguchi esterification⁸ with the alcohol **5** to give the ester **16** in 55% yield. Desilylation of the TBS group of **16** with TBAF, acid catalyzed reaction of **17** with **18** gave the mixed acetal **19**. Subsequent acetal cleavage with TMSI/HMDS furnished the allylic stannane **20** in 73% yield.⁹ Modified Rychnovsky acetylation including the partial reduction of **20** with DIBAL-H followed by trapping of the resulting aluminum hemiacetal with

chloroacetic anhydride gave the cyclization precursor **21** in quantitative yield.^{10,11} Intramolecular allylation of **21** was carried out using $\text{BF}_3\cdot\text{OEt}_2$ to afford **22** as a single stereoisomer in 75% yield. Finally, the diene **22** was subjected to ring-closing metathesis using the second generation Grubbs catalyst **23**¹² to furnish the F–K ring framework **24** in 91% yield. The *trans* relationship between Ha and Hb was confirmed by the large coupling constant, $J_{\text{Ha-Hb}} = 8.5$ Hz.

In conclusion, we have achieved the convergent synthesis of the F–K ring segment of brevetoxin B (**1**) via the intramolecular allylation of α -chloroacetoxy ether **21** and subsequent ring-closing metathesis. Further studies toward the total synthesis of **1** are now in progress in our laboratories.

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Scheme 3. Reagents and conditions: (a) (i) DIBAL-H, CH_2Cl_2 , -78°C , (ii) NaClO_2 , NaH_2PO_4 , 2-methyl-2-butene, $t\text{-BuOH/THF/H}_2\text{O}$, rt; (b) 2,4,6-trichlorobenzoyl chloride, Et_3N , THF, rt, then **5**, DMAP, toluene, rt, 55% (three steps); (c) TBAF, THF, rt, 96%; (d) **18**, CSA, CH_2Cl_2 , rt, 95%; (e) TMSI, HMDS, CH_2Cl_2 , 0°C , 83%; (f) DIBAL-H, CH_2Cl_2 , -78°C , then $(\text{CH}_2\text{ClCO})_2\text{O}$, pyridine, DMAP, -78°C , 100%; (g) $\text{BF}_3\cdot\text{OEt}_2$, $\text{CH}_3\text{CN-CH}_2\text{Cl}_2$ (20:1), -45 to 0°C , 75%; (h) **23**, CH_2Cl_2 , rt, 91%.

References

- Lin, Y.-Y.; Risk, M.; Ray, S. M.; Van Engen, D.; Clardy, J.; Golik, J.; James, J. C.; Nakanishi, K. *J. Am. Chem. Soc.* **1981**, *103*, 6773–6775.
- For the synthetic studies of brevetoxin B, see: (a) Nicolaou, K. C.; Duggan, M. E.; Hwang, C.-K. *J. Am. Chem. Soc.* **1989**, *111*, 6666–6675; (b) Nicolaou, K. C.; Duggan, M. E.; Hwang, C.-K. *J. Am. Chem. Soc.* **1989**, *111*, 6676–6682; (c) Nicolaou, K. C.; Hwang, C.-K.; Duggan, M. E. *J. Am. Chem. Soc.* **1989**, *111*, 6682–6690; (d) Nicolaou, K. C.; Nugiel, D. A.; Couladouros, E.; Hwang, C.-K. *Tetrahedron* **1990**, *46*, 4517–4552; (e) Matsuo, G.; Matsukura, H.; Hori, N.; Nakata, T. *Tetrahedron Lett.* **2000**, *41*, 7673–7676; (f) Matsuo, G.; Hori, N.; Matsukura, H.; Nakata, T. *Tetrahedron Lett.* **2000**, *41*, 7677–7680; (g) Matsukura, H.; Hori, N.; Matsuo, G.; Nakata, T. *Tetrahedron Lett.* **2000**, *41*, 7681–7684.
- (a) Nicolaou, K. C.; Theodorakis, E. A.; Rutjes, F. P. J. T.; Tiebes, J.; Sato, M.; Untersteller, E.; Xiao, X.-Y. *J. Am. Chem. Soc.* **1995**, *117*, 1171–1172; (b) Nicolaou, K. C.; Rutjes, F. P. J. T.; Theodorakis, E. A.; Tiebes, J.; Sato, M.; Untersteller, E. *J. Am. Chem. Soc.* **1995**, *117*, 1173–1174; (c) Nicolaou, K. C.; Hwang, C.-K.; Duggan, M. E.; Nugiel, D. A.; Abe, Y.; Bal Reddy, K.; Defrees, S. A.; Reddy, D. R.; Awartani, R. A.; Conley, S. R.; Rutjes, F. P. J. T.; Theodorakis, E. A. *J. Am. Chem. Soc.* **1995**, *117*, 10227–10238; (d) Nicolaou, K. C.; Theodorakis, E. A.; Rutjes, F. P. J. T.; Sato, M.; Tiebes, J.; Xiao, X.-Y.; Hwang, C.-K.; Duggan, M. E.; Yang, Z.; Couladouros, E. A.; Sato, F.; Shin, J.; He, H.-M.; Bleckman, T. *J. Am. Chem. Soc.* **1995**, *117*, 10239–10251; (e) Nicolaou, K. C.; Rutjes, F. P. J. T.; Theodorakis, E. A.; Tiebes, J.; Sato, M.; Untersteller, E. *J. Am. Chem. Soc.* **1995**, *117*, 10252–10263.
- Matsuo, G.; Kawamura, K.; Hori, N.; Matsukura, H.; Nakata, T. *Tennen Yuki Kagobutsu Toronkai Koen Yoshishu* **2002**, 163–168.
- Kadota, I.; Ohno, A.; Matsuda, K.; Yamamoto, Y. *J. Am. Chem. Soc.* **2002**, *124*, 3562–3566.
- Pétier, C.; Luche, J.-L. *J. Org. Chem.* **1985**, *50*, 910–912.
- The Grignard reaction of the hydroxy aldehyde gave poor results.

8. Inanaga, J.; Hirata, K.; Saeki, H.; Katsuki, T.; Yamaguchi, M. *Bull. Chem. Soc. Jpn.* **1979**, *52*, 1989–1993.
9. Kadota, I.; Sakaihara, T.; Yamamoto, Y. *Tetrahedron Lett.* **1996**, *37*, 3195–3198.
10. Kadota, I.; Takamura, H.; Sato, K.; Ohno, A.; Matsuda, K.; Yamamoto, Y. *J. Am. Chem. Soc.* **2003**, *125*, 46–47.
11. For the original conditions, see: (a) Dahanukar, V. H.; Rychnovsky, S. D. *J. Org. Chem.* **1996**, *61*, 8317–8320; (b) Kopecky, D. J.; Rychnovsky, S. D. *J. Org. Chem.* **2000**, *65*, 191–198; (c) Kopecky, D. J.; Rychnovsky, S. D. *Org. Synth.* **2003**, *80*, 177–183.
12. Scholl, M.; Ding, S.; Lee, C. W.; Grubbs, R. H. *Org. Lett.* **1999**, *1*, 953–956.