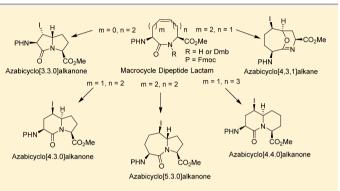
Insight into Transannular Cyclization Reactions To Synthesize Azabicyclo[X.Y.Z]alkanone Amino Acid Derivatives from 8-, 9-, and 10-Membered Macrocyclic Dipeptide Lactams

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Supporting Information

ABSTRACT: An efficient method for synthesizing different functionalized azabicyclo[*X.Y.0*]alkanone amino acid derivatives has been developed employing electrophilic transannular cyclizations of 8-, 9-, and 10-membered unsaturated macrocycles to form 5,5-, 6,5-, 7,5-, and 6,6-fused bicylic amino acids, respectively. Macrocycles were obtained by a sequence featuring peptide coupling of vinyl-, allyl-, homoallyl-, and homohomoallylglycine building blocks followed by ring-closing metathesis. X-ray crystallographic analyses of the 8-, 9-, and 10-membered macrocyclic lactam starting materials as well as certain bicyclic amino acid products provided insight into their conformational preferences as well as the mechanism



for the diastereoselective formation of specific azabicycloalkanone amino acids by way of transannular iodolactamization reactions.

INTRODUCTION

Azabicyclo [X.Y.0] alkane structures, such as pyrrolizidines, indolizidines, and qunolizidines, are found in numerous biologically active alkaloid natural products.^{1,2} The related azabicyclo [X.Y.0] alkanone amino acid counterparts are rigid dipeptide surrogates that have been employed to study conformation-activity relationships of biologically active peptides.^{3,4} Inherent to the syntheses of both classes of heterocycles has been the stereocontrolled construction of the bicyclic ring system in a way that provides for effective introduction of ring functionality. Among various synthetic strategies,⁵ transannular reactions featuring amine nucleophiles reacting on macrocyclic olefin and epoxide precursors have exhibited considerable utility for the stereoselective construction of the bicyclic amines,⁶ albeit few examples of the related cyclizations of macrocyclic lactams have been reported.8 In both cases, regioselective formation of specific bicycles has often been observed to occur with high diastereoselectivity. Limited mechanistic understanding exists however to predict the outcome of such transannular cyclization reactions, because they have been typically performed on a restricted number of ring sizes with substituents that have varied greatly both in functional group and in location. Generally, in cycloalkanes of 8-11 ring atoms, unfavorable interactions of ring substituents can cause significant strain, so-called "Prelog strain".9,10 Moreover, nitrogen lone pair electrons have been reported to prefer an antiorientation with respect to the olefin π -bond to minimize repulsive interactions.¹¹ Such steric and electronic interactions serve likely in part as driving forces for preferred macrocycle conformations and favored regioselective cyclizations.

In the interest of developing a general strategy for synthesizing azabicyclo[X.Y.0] alkanone amino acids of varying ring sizes, we have pursued electrophilic transannular lactamizations of unsaturated macrocyclic dipeptides.^{8a} This strategy is appealing, because a variety of macrocycles may be generated by coupling various ω -unsaturated amino acids, followed by ring-closing metathesis (RCM; Figure 1). Moreover, after transannular iodolactamization, the resulting iodide has served as a handle for introducing various side chain functional groups onto the bicyclic system.¹²

Considering that both bicycle and macrocycle constraints serve to rigidify the peptide backbone,^{5c,13} efforts have focused on studying the synthesis and conformational analysis of a series of ring systems in which the same dipeptide motif is maintained as the ring size and olefin orientation are modulated. Employing X-ray crystallography and NMR spectroscopy, insight has now been gained with respect to the conformational preferences of the macrocyclic and bicyclic dipeptide motifs, as well as the regioselective nature of the transannular lactamization. Exploring five different macrocycles, we have found conditions to prepare regioselectively and stereoselectively five distinct bicyclic dipeptide mimics, validating the effectiveness of this approach for peptidomimetic synthesis.

RESULTS AND DISCUSSION

Iodoazabicyclo[4.3.0]- and iodoazabicyclo[5.3.0]alkanones 1 and 2 were previously made by the diastereoselective

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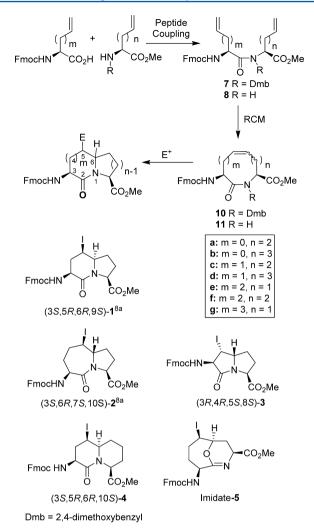


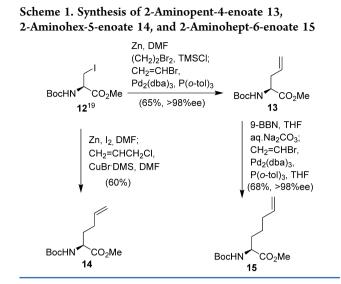
Figure 1. General synthetic plan for making azabicyclo[*X*.*Y*.0] alkanones and representative examples of previously prepared bicycles **1** and **2**, as well as new examples, **3**–**5**, prepared herein.

iodolactamization of 9- and 10-membered macrocycles (*Z*)-10c and (*E*)-11f, respectively (Figure 1).^{8a,12,14} Macrocycles (*Z*)-10c and (*E*)-11f had been made from allyl- and homoallylglycine building blocks.¹⁴ Herein, vinyl- and homohomoallylglycines were added to provide a set of four ω -unsaturated amino acids for RCM/electrophilic transannular cyclizations that have now given access to the new 5,5- and 6,6-fused bicyclic ring systems 3 and 4, as well as anti-Bredt imidate 5 (Figure 1). The latter bridgehead imidate represents a rare substituted heterocycle example that challenges Bredt's rules.¹⁵ From all five of these related transannular cyclizations, only a single heterocyclic system was typically formed diastereoselectively.

The enantiomerically pure ω -unsaturated amino acid starting materials with lengths of 4–7 carbons were synthesized efficiently by atom economical routes. Many synthetic approaches exist for making these building blocks;^{14,16,17} however, they often require long reaction sequences, especially in the cases of 2-aminopent-4-enoate (allylglycine) and 2-aminohept-6-enoate (homohomoallylglycine).

Inspired by the relatively efficient synthesis of methyl 2-(N-Boc-amino)hex-5-enoate (14),¹⁸ which featured the coppercatalyzed cross-coupling of allyl chloride and the zincate derived from iodoalanine 12 (Scheme 1),¹⁹ related transition-metal crosscoupling methods were developed to make the higher and lower





amino acid homologues. Methyl 2-(N-Boc-amino)pent-4-enoate (13) and methyl 2-(N-Boc-amino)hept-6-enoate (15) were thus respectively prepared by using vinyl bromide in the Pd-catalyzed reactions on the zincate derived from iodoalanine 12 and the boronate derived from 13.

Starting from L-serine as a chiral educt, protected enantiomerically pure ω -unsaturated amino acids with chain lengths of 5–7 carbons were prepared by a common sequence. Methyl *N*-Boc-iodoalanine (**12**; Scheme 1) was assembled in three steps from serine and used in the synthesis of the five- and sixcarbon analogues.¹⁹ On treatment with zinc, iodoalanine **12** was converted to the corresponding zincate, which was coupled with vinyl bromide and Pd₂(dba)₃²⁰ or allyl chloride and CuBr·DMS,¹⁸ respectively, to give pentenoate **13** and hexenoate **14** in 65% and 60% yields. Heptenoate **15** was synthesized from pentenoate **13** in 68% yield by hydroboration of the terminal alkene using 9-borabicyclo[3.3.1]nonane (9-BBN) and crosscoupling to vinyl bromide using Pd₂(dba)₃.

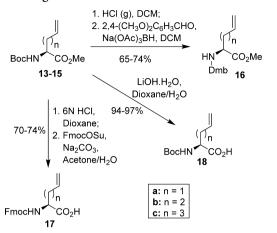
After the syntheses of pentenoate **13** and heptenoate **15**, their enantiomeric purities were respectively evaluated by conversion to diastereomeric dipeptides on removal of the Boc group with HCl gas in dichloromethane and coupling to L- and D-N-Bocalanine with HATU (Supporting Information). Examination of the diastereotopic methyl ester singlets by ¹H NMR spectroscopy in CD₃CN during incremental addition of the (*S*,*R*)diastereomer to the (*S*,*S*)-diastereomer demonstrated in both cases a >99:1 dr. Hence, pentenoate **13** and heptenoate **15** are both assumed to be of >98% enantiomeric purity.

Suitably protected 2-(*N*-Dmb-amino) ester 16 and 2-(*N*-Bocamino) and 2-(*N*-Fmoc-amino) acids 17 and 18 were prepared from 2-(*N*-Boc-amino) esters 13-15 using previously described protocols (Scheme 2).¹⁴ 2-(*N*-Fmoc-amino)but-3-enoic acid (vinylglycine 19) was synthesized as previously described.^{21,22}

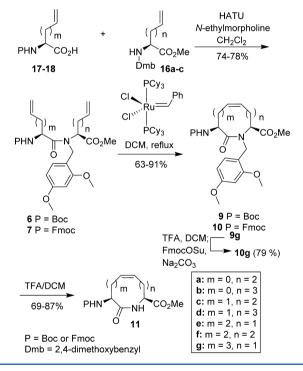
Dipeptides 6 and 7 were synthesized by coupling ester 16 and acids 17 and 18 using HATU and *N*-ethylmorpholine in 70–80% yield (Scheme 3).¹⁴ Double bond migration occurred during coupling of but-3-enoic acid 19 to 2-(*N*-Dmb-amino)hex-5-oate 16b using HATU and DIEA and gave dehydropeptide 20 (Scheme 4). Double bond migration was avoided by converting vinylglycine 19 to the corresponding acid chloride, which reacted with 16b and 16c in the absence of base to afford dipeptides 7a and 7b in 50–60% yield.

Macrocycles 9 and 10 were prepared from dipeptides 6 and 7 in 60-90% yields in solution using RCM employing the

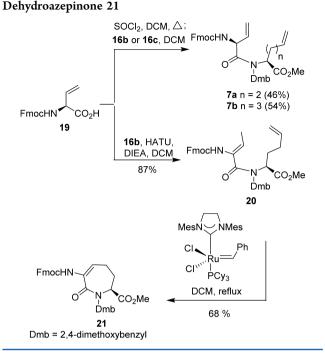
Scheme 2. Protecting Group Shuffle To Make Boc, Fmoc, and Dmb Building Blocks



Scheme 3. Syntheses of Dipeptides 6 and 7 and Macrocycles 9–11



Grubbs first-generation catalyst in dichloromethane at reflux (Scheme 3). 23,24 Dehydroamino acid analogue **20** failed to react using the first generation of the Grubbs catalyst, but could be converted to dehydroazepinone 21 with the Grubbs secondgeneration catalyst (Scheme 4).²⁵ In spite of the acid sensitivity of the Dmb group, the Boc group could be selectively removed from macrocycle 9g using 50% TFA in DCM for 2 h to afford the corresponding amine, which was converted to its Fmoc counterpart 10g using Fmoc-OSu and Na₂CO₃ (Scheme 3).¹⁴ To remove the Dmb group, N-Fmoc-amino lactams 10 were treated with 50% TFA/DCM for 18 h, which afforded macrocycles 11 in 70-80% yields. In the case of 10-membered N-Dmb-amino lactam 10d, starting material was however not consumed even after treatment for 3 days, when lactam 11d was afforded in 69% yield after chromatography. The olefin and amide isomer geometries of macrocycles 9-11 were established by NMR spectroscopy and X-ray analysis (vide infra).

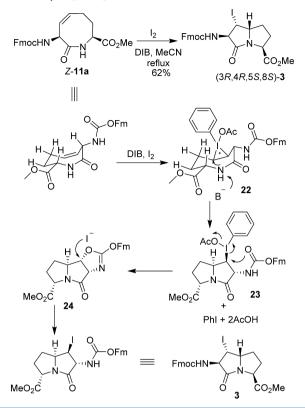


Scheme 4. Coupling with Vinylglycine 19 and Synthesis of

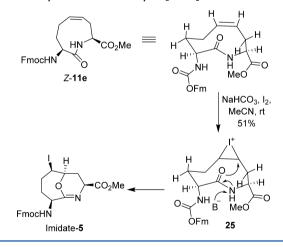
Transannular cyclization of 8-membered macrocyclic lactam (Z)-10a was initially studied by treatment with iodine in THF and MeCN at room temperature to reflux; however, only loss of the Dmb group was observed, and macrocycle (Z)-11a was isolated. 8-Membered lactam (Z)-11a proved resistant to transannular iodoamidation using I2 in THF buffered with $NaHCO_3$, as well as using I₂ in acetonitrile at reflux. Transannular cyclization was achieved by treating (Z)-11a with (diacetoxyiodo)benzene (DIB; 1.5 equiv) and 4 equiv of iodine in acetonitrile at reflux for 30 min (Scheme 5). Although DIB has been used in oxidative cyclizations of o-hydroxystyrenes to benzofurans, to the best of our knowledge, this is the first use of this reagent in a transannular iodolactamization. (3R,4R,5S,8S)-Pyrrolizidinone 3 was isolated as a single diastereomer in 62% yield after chromatography (Scheme 5). In the mechanism for the formation of 3, hypervalent iodine may activate the double bond as a 3-membered iodonium species, 22,²⁶ which undergoes intramolecular reaction with the amide nitrogen to form intermediate 23. Displacement of the hypervalent iodine by the neighboring carbamate may occur by way of an oxazole intermediate 24, which is opened by iodide to give iodopyrrolizidinone 3 with retention of configuration. The structure of 3 was confirmed by NMR spectroscopy and X-ray diffractometry as described below.

Although the related double bond isomer (Z)-10c (m = 1, n = 2) underwent iodolactamization to provide iodoindolizidinone 1 in 86% yield using I₂ in THF at 80 °C (Figure 1),^{8a} treatment of unsaturated macrocyclic lactam (Z)-10e (m = 2, n = 1) under analogous conditions only caused removal of the Dmb group to provide lactam (Z)-11e. Moreover, attempts to induce transannular cyclization of 9-membered lactam (Z)-11e using alternative conditions (I₂ in MeCN;¹² DIB and I₂ in MeCN, rt to reflux; Lewis base-catalyzed²⁷ iodolactamization using Ph₃P=S, NIS/DCM) produced a relatively less polar spot on TLC compared to the iodobicycles. The reaction proceeded cleanly, and the starting material was all consumed within 1 h as indicated by TLC. After the usual aqueous workup with Na₂S₂O₃, and purification by silica gel chromatography in the dark, imidate

Scheme 5. Proposed Mechanism for the Synthesis of Iodoazabicyclo[3.3.0]alkanone Amino Ester 3

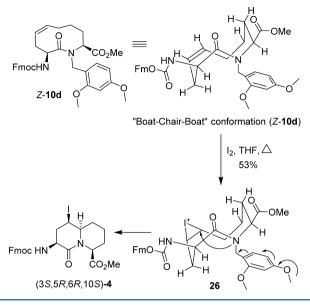


Scheme 6. Synthesis of Azabicyclo [4.3.1] alkane 5



5 was characterized using NMR spectroscopy and X-ray crystallography (Scheme 6). Imidate formation has been observed commonly in halocyclizations of linear unsaturated amides, and avoided using methods employing *N*,*O*-bis-silylation, *N*-tosyl and *N*-alkoxycarbonyl substitution, and strong bases.²⁸ On the other hand, transannular iodolactamization has favored azabicycloalkanone.^{8b,12} Albeit light-sensitive, imidate **5** is a unique anti-Bredt heterocycle formed by attack of the lactam oxygen on iodonium intermediate **25**.^{29,30}

Previously, iodoazabicyclo[5.3.0]alkanone 2 (Figure 1) was prepared diastereoselectively from 10-membered lactam (*E*)-11f (m = 2, n = 2) using 4 equiv of I₂ in THF at reflux.^{8a} Iodolactamization of 10-membered *N*-Dmb-lactam (*Z*)-10d (m = 1, n = 3) has now given diastereoselective access to the first example of a substituted quinolizidinone amino acid, Scheme 7. Synthesis of Iodoazabicyclo[4.4.0]alkanone Amino Ester 4



(3*S*,*SR*,*6R*,10*S*)-5-iodoazabicyclo[4.4.0]alkanone 4, in 53% yield using I₂ in THF at reflux (Scheme 7).

Stereochemical Assignment Using NMR Spectroscopy. The assignment of the structure and stereochemistry of bicycles **3–5** was accomplished by 2D NMR spectroscopy and X-ray crystallography (Supporting Information). Typically, a COSY spectrum was used to assign the ring proton through-bond connectivities starting from the downfield carbamate NH proton. The observation of magnetization transfer between the peptide backbone protons and ring protons in the NOESY and ROESY spectra was then used to assign the relative stereochemistry at the ring-fusion and iodide-bearing carbons (Figure 2). For example,

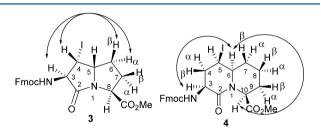


Figure 2. NOESY and ROESY correlations used to assign the relative configurations of bicycles 3 and 4.

(4*R*,5*S*)-4-iodopyrrolizidinone *N*-Fmoc-amino ester **3** exhibited a transfer of magnetization between the backbone C3 α and the ring C6 α protons in C₆D₆. A second long-range NOE between the ring C4 and C6 β -protons established the relative stereochemistry of the iodide-bearing carbon. Long-range transfer of magnetization was observed between the ring-fusion C6 and backbone C3 and C10 protons in the ROESY spectrum of (3*S*,5*R*,6*R*,10*S*)-5-iodoquinolizidinone **4**. The stereochemistry of the ring-fusion carbon of pyrrolizidinone **3** and the iodidebearing carbon of **4** were subsequently inferred on the basis of mechanistic considerations that placed the protons from the *cis*double bond of the macrocycle on the same face of the bicycle.

Although through-bond couplings of azabicyclo[4,3,1]alkane 5 could be used to assign all of the ring protons, no clear long-range NOEs were observed to assign the relative stereochemistry.

In the infrared spectrum of **5**, a C==N stretching band at 1657 cm⁻¹ was observed, indicative of an imidate structure. Examination of the coupling pattern (ddd, J = 2.7, 5.4, 11.0 Hz) for the proton on the iodide-bearing carbon, before and after saturation of the neighboring bridgehead proton, and application of the Karplus equation³¹ gave torsion angels that were consistent with those observed in the crystal structure.

Crystal Structures of 8-, 9-, and 10-Membered Macrocyclic Dipeptide Lactams. Lactams (*Z*)-11a, (*Z*)-11e, and (*Z*)-10d were crystallized by a common method (Supporting Information). In all three cases, the (*Z*)-double bond geometry was observed. In 8-membered lactam (*Z*)-11a, the *cis*-(*E*)-amide isomer was observed, as was previously described in the X-ray structure of the related olefin regioisomer 8-membered macrocycle 27 (Table 1).³² On the other hand, the *trans*-(*Z*)-amide

 Table 1. Comparison of Dihedral Angles of Macrocyclic and

 Bicyclic Peptidomimetics with Ideal Secondary Structures

FmocHN R = H or Dmb	BocHN	-N H NF 27	⊃ HMe	Ö	CO ₂ Me	
type of β -turn ³³	ϕ_1 , deg	ψ_1 , de	g ϕ_2	, deg	ψ_2 , deg	
Ι	-60	-30		-90	0	
II	-60	120	1	80	0	
II'	60	-120		-80	0	
VIb	-135	135	-	-75	160	
antiparallel eta -sheet 38 parallel eta -sheet 38	$\phi = -140$ $\phi = -120$			$\psi = 135$ $\psi = 115$		
Peptidomimetic Dihedral Angles						
		$\phi^{\scriptscriptstyle i+1}$, deg	ψ^{i+1} , deg	$\phi^{\scriptscriptstyle i+2}$, deg	ψ^{i+2} , deg	
8-membered lactam (Z) -11a		-142	156	-158	172	
8-membered lactam 2 7 ³²		-93	177	-148	90	
9-membered lactam (Z)-11e		-109	-27	-145	-28	
10-membered lactam (Z)-10d		-161	171	63	23	
10-membered lactam (E)- 28 ³⁹						
chair-chair		-107	-1	-130	23	
chair—boat		120	-64	-131	-43	
4-iodoazabicyclo[3.3.0]alk	-120	-140	-109	173		
6-hydroxyazabicyclo[3.3.0]	14	-141	-40	133		
iodoazabicyclo[5.3.0]alkanone 2 ^{8a}		-138	-64	-49	136	
azabicyclo[4,3,1]alkane 5	-166	-1	175	74		

isomer was detected in the crystal structures of 9- and 10membered lactams (Z)-11e and (Z)-10d.

Examination of the backbone dihedral angles found in the crystal structures of lactams (Z)-11a, (Z)-11e, and (Z)-10d may provide an understanding of their potential for mimicry of natural peptide structures, albeit ¹H NMR spectroscopy of macrocyclic lactams (Z)-11e and (Z)-10d did show doubling of some peaks, indicative of conformational isomers in CDCl₃. The torsion angles of lactams (Z)-11a, (Z)-11e, and (Z)-10d were compared with those for ideal turn and sheet structures (Table 1).³³ Notably, 8-membered lactam (Z)-11a possesses a dihedral angle geometry similar to that observed for the central residues of an ideal type VIB β -turn, similar to its olefin regioisomer 27. The torsion angle values of 9-membered lactam (Z)-11e are similar to those of an ideal type I β -turn. Previously, 10-membered lactam 28 possessing a *trans*-olefin geometry had

also been found to exhibit backbone dihedral angles similar to those of an ideal type I β -turn (Table 1). In the case of 10-membered lactam (Z)-10d, which possesses a *cis*-olefin at a different ring position, the backbone conformation was more similar to an ideal type II β -turn, albeit the dihedral angles of the *N*-terminal residue resembled an extended parallel β -pleated sheet. In summary, lactams (Z)-11a, (Z)-11e, and (Z)-10d appear to have potential for mimicry of three different β -turn conformations.

Crystal Structures of Azabicyclo[*X*.*Y*.*Z*]alkanes 3 and 5. The stereochemical assignments made for iodopyrrolizidinone 3 were confirmed by the X-ray structure (Figure 3). The dihedral angle values of the peptide backbone in the 5,5-fused bicycle 3 resembled those of an extended rather than a turn structure. In this respect, the peptide backbone geometry in azabicyclo[3.3.0]-alkanone 3 differed from those observed in X-ray analyses of larger azabicyclo[4.3.0]alkanone and iodoazabicyclo[5.3.0]-alkanone (2) amino ester counterparts, which were respectively similar to ideal type II' and type I β -turns.^{34,12} In addition, the X-ray structure of 4-iodopyrrolizidinone 3 exhibited dihedral angles different from those of the X-ray structure of the related 6-hydroxypyrrolizidinone, which closely resembled those of a type II' β -turn (Table 1).^{5a}

In the X-ray structure of cyclic imidate **5** (Figure 3), the (*Z*)-*anti*-periplanar isomer was observed. Although related structures of such anti-Bredt heterocycles were not found, the bond lengths of bridging imidate **5** (C–O, 1.39 Å; C=N, 1.24 Å) compared well with those from the crystal structures of cyclic imidates (C–O, 1.36 ± 0.2 Å; C=N, 1.25 ± 0.01 Å),^{35,36} indicating that little distortion was induced by the bridgehead geometry of **5**. In addition, the dihedral angle values of **5** resembled those of an ideal type II β -turn.

Mechanistic Considerations. Transannular iodolactamizations have been studied using a series of related macrocycles of 8-10 ring atoms. In spite of the structural similarities of the starting macrocyclic lactams, many factors may influence this reaction, including the ring conformation and cyclization conditions. In related iodolactamizations, regioselectivity has previously been considered to be a consequence of the macrocycle conformation.³⁷ In this light, the flexibility of the macrocycle ring may contribute to mixtures of bicyclic products, yet ring substituents that orient to avoid allylic strain with the olefin substituents and diaxial interactions will significantly reduce the number of cyclization pathways. Typically, a relaxed conformer may orient the lactam and olefin to give regio- and stereoselective pathways.^{37,8f,d} In such cases, the iodide substituent ends up on the lactam ring. In one case, a thermodynamically less stable 9-membered lactam has been found to give an alternative regio- and stereochemical indolizidinone featuring a 5-membered lactam and a piperidine ring bearing the iodide.³⁷ In addition, 9-membered lactam (Z)-11e gave imidate 5 instead of a fused 7,4-bicycle with the iodide on the 7-membered lactam or a 6,5-bicycle with the iodide on the 5-membered pyrrolidine.

Although electrophile-induced intramolecular cyclization reactions may proceed by formation of a dihalide that is subsequently displaced by the lactam to form the ring,^{8e} cyclizations of macrocycles (Z)-11a, (Z)-11e, and (Z)-10d are presumed to proceed by way of iodonium-like intermediates based on their stereochemical outcomes.⁴⁰ The ambidentate nucleophilic nature of the amide group has been observed to give imidates in iodocyclizations of simpler olefins,^{41,42} and such *O*-selectivity has been explained on the basis of hard–soft

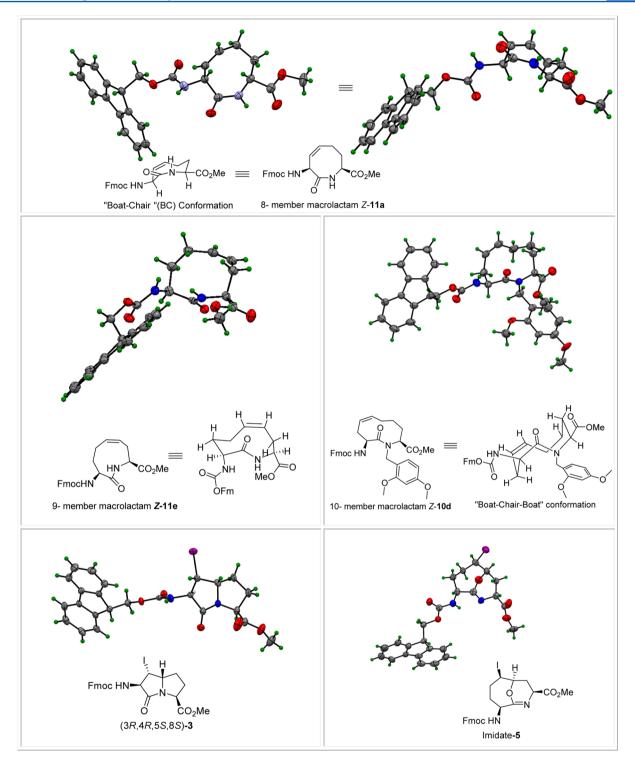


Figure 3. X-ray structures of macrocyclic dipeptide lactams and iodoazabicyclo[X.Y.Z]alkanes (C, gray; H, green; N, blue; O, red; I, magenta).

acid—base theory considering the iodine—olefin π -complex to be a hard electrophile.^{43,44} However, transannular attack of oxygen would result in an anti-Bredt bridgehead heterocycle,⁴⁵ which may be relatively disfavored due to ring strain. Given the choices between placing iodide on the nonlactam ring, forming a strained *N*-acylazetidine or forming the anti-Bredt imidate, the latter product **5** was favored.

Notably, the respective proximities of the amide nitrogen and oxygen to the double bond carbons on the amine (C^N) and carbonyl (C^{CO}) sides of the macrocycle did not correlate with the

mode of cyclization except in the case of the 8-membered lactam going to the fused 5,5-bicycle (Table 2).

A working hypothesis for the selectivity of the transannular iodolactamization begins with a favored macrocycle conformation that minimizes ring strain. Approach of iodine to the less sterically hindered face of the olefin provides the iodonium intermediate, which is typically attacked on the carbon (C^N) on the amide nitrogen side of the macrocycle to provide a bicycle having the iodide on the lactam ring. Attack of the iodonium carbon (C^{CO}) on the carbonyl side of the macrocycle is likely

Table 2. Comparison of Transannular Distances (Å) in Macrocyclic Lactams

bond	8-membered (Z)- 11a	9-membered (Z)- 11e	10-membered (Z)- 10d
N to C^{N}	3.11	3.23	3.45
N to C ^{CO}	3.40	3.01	3.47
O to C^{N}	3.14	3.33	3.29
O to C ^{CO}	4.12	3.38	3.63

disfavored due to the need to avoid allylic strain between the carboxylate and the lactam carbonyl as well as potential diaxial interactions between the carboxylate and the iodide. In the case of **5**, the combination of the factors mentioned above and ring strain to make a 4-membered azetidine lead to a preferred attack by oxygen, giving the imidate. Computational analyses are currently being pursued to provide additional support of these mechanistic considerations.

CONCLUSIONS

A variety of azabicyclo[X.Y.0] alkanone amino acids have been prepared using a common approach featuring effective assembly of macrocyclic lactams of 8-10-membered ring sizes by the synthesis and coupling of ω -unsaturated amino acids and RCM of their dipeptide derivatives, followed by regio- and stereoselective electrophilic transannular cyclization. Employing X-ray crystallographic and NMR spectroscopic analyses, we have demonstrated that the macrocyclic and bicyclic structures offer ample potential to mimic peptide secondary structures, in particular types I, II', and VI β -turn geometries contingent on ring size. Moreover, such conformational analyses have also provided insight into the mechanism of transannular iodolactamization. Considering the potential of this method for effectively making a series of related dipeptide mimics to explore peptide conformation-activity relationships, as well as the opportunity for replacing the iodide with other side chains to examine functional group influences on activity, this method should have a significant impact on peptidomimetic approaches for studying biologically active peptides, particularly for the discovery of therapeutic agents.

EXPERIMENTAL SECTION

General Methods. Unless otherwise noted, all reactions were performed under an argon atmosphere and distilled solvents were transferred by syringe. Anhydrous CH₂Cl₂ (DCM), THF, and diethyl ether were obtained by passage through a solvent filtration system. Final reaction mixture solutions were dried over anhydrous MgSO4 or Na₂SO₄, filtered, and rotary-evaporated under reduced pressure. Flash chromatography⁴⁶ was on 230-400 mesh silica gel, and thinlayer chromatography was performed on silica gel 60 F254 plates. Specific rotations, $[\alpha]_{\rm D}$ values, were calculated from optical rotations measured at 20 °C in CHCl₃ or MeOD at the specified concentrations (c, g/100 mL) using a 1 dm cell (l) on a polarimeter, using the general formula $[\alpha]_{D}^{20} = 100\alpha/lc$. Accurate mass measurements were performed on an LC-MSD instrument in electrospray ionization (ESI-TOF) mode. Sodium adducts $[M + Na]^+$ were used for empirical formula confirmation. ¹H NMR spectra were measured in CDCl₃ (7.26 ppm), CD₃OD (3.31 ppm), or DMSO-*d*₆ (2.50 ppm). ¹³C NMR spectra were measured in CDCl₃ (77.16 ppm) or DMSO-d₆ (39.52 ppm). ¹H NMR spectra for the dipeptides 7a, 7b, 6d, 7d, 6g, 11g, 11d, and 20 were recorded at 100 °C to coalesce signals due to conformational isomers. Coupling constant J values were measured in hertz and chemical shift values in parts per million. Infrared spectra were recorded for neat samples on an FT-IR apparatus. Syntheses and characterization of compounds 7e, 10e, 11e, 17a, 17b, 16a, and 16b have been previously reported.¹⁴

$$Br (0.6 eq), 60 °C$$

$$Br (0.6 eq), 60 °C$$

$$TMSCI (0.2 eq), rt-35 °C$$

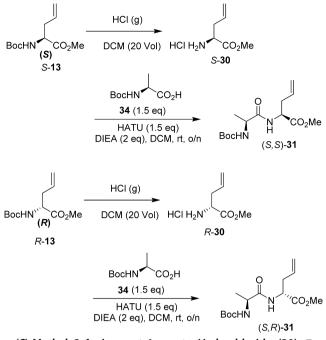
$$BocHN 2. Br (1 M in THF)$$

$$Pd_2(dba)_3 (0.02 eq)$$

$$P(o-tol)_3 (0.1 eq)$$

$$-78 °C-rt, 12 h$$

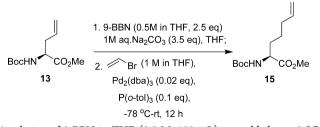
To a stirred solution of zinc dust (11.92 g, 182.3 mmol, 6 equiv) in dry DMF (20 mL) was added 1,2-dibromoethane (1.57 mL, 3.42 g, 18.2 mmol, 0.6 equiv) via a syringe. The resulting mixture was heated to 60 °C, stirred for 45 min, and cooled to room temperature, and the slurry was treated with chlorotrimethylsilane (0.77 mL, 6.0 mmol). After being stirred for 40 min at room temperature, the activated zinc was treated with a solution of methyl (R)-N-Boc-iodoalaninate (12; 10 g, 30.39 mmol) in dry DMF (20 mL), heated to 35 °C, and stirred for 60 min, when insertion was judged complete by TLC analysis [2:1 hexanes/ethyl acetate, using 254 nm UV light to visualize the starting material 12 ($R_f = 0.7$) and organozinc reagent ($R_f = 0.1$)]. After complete zinc insertion, the reaction mixture was cooled to room temperature, charged with Pd₂(dba)₃ (779 mg, 0.85 mmol, 0.028 equiv) and tri-o-tolylphosphine (925 mg, 3.03 mmol, 0.1 equiv), cooled to -78 °C, and treated dropwise via a cannula with a solution of vinyl bromide in THF (1 M, 42.5 mL, 42.5 mmol, 1.4 equiv). After complete addition of the vinyl bromide solution, the cold bath was removed, and the reaction mixture was allowed to warm to room temperature with stirring for 12 h. The reaction mixture was diluted with ethyl acetate (200 mL) and water (200 mL), and filtered through a pad of Celite. The pad was washed with ethyl acetate (300 mL). The filtrate and washings were combined and transferred to a separating funnel. The organic layer was separated. The aqueous layer was extracted with ethyl acetate $(2 \times 200 \text{ mL})$. The combined organic layers were washed with brine (400 mL), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure to give 8.2 g of brown oil, which was purified by column chromatography on silica gel (8-10% EtOAc in hexane) to give olefin 13 (4.50 g, 19.64 mmol, 65%) as a brown oil: $R_f = 0.29$ (9:1 hexanes/ethyl acetate, visualized as a UV inactive and KMnO₄ active spot on heating); $[\alpha]_{\rm D}^{20}$ +20.2 (c 1.5, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 2978, 1744, 1712, 1499, 1437, 1365, 1159, 1021, 868, 779 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 5.63–5.71 (m, 1H), 5.09-5.12 (m, 2H), 5.03-5.04 (br d, 1H), 4.34-4.37 (m, 1H), 3.71 (s, 3H), 2.50–2.55 (m, 1H), 2.42–2.47 (m, 1H), 1.41 (s, 9H); ¹³C NMR (125 MHz, CDCl₃) δ 172.7, 155.3, 132.4, 119.1, 79.9, 53.0, 52.3, 36.9, 28.4; HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₁₁H₁₉NO₄Na 252.1215, found 252.1217. Anal. Calcd for C11H19NO4: C, 57.62; H, 8.35; N, 6.11. Found: C, 57.13; H, 8.51; N, 5.93. Attempts to ascertain the enantiomeric purity of olefin 13, as well as the hydrochloride 30 obtained on treating 13 with HCl gas in dichloromethane, were both unsuccessful using supercritical fluid chromatography on a chiral column. To assess enantiomeric purity, diastereomeric amides were synthesized as described below. The crude residue was examined by ¹H NMR spectroscopy in CD₃CN at 700 MHz. Incremental addition of (S,R)-31 to (S,S)-31 and observation of the methyl ester singlets at 3.697 and 3.691 ppm demonstrated the diastereomers were of >99:1 dr. Hence, olefin 13 is assumed to be of >98% enantiomeric purity.



(S)-Methyl 2-Aminopent-4-enoate Hydrochloride (30). Dry HCl gas was bubbled into a stirred solution of (S)-13 (70 mg, 0.30 mmol) in dichloromethane at room temperature. Consumption of (S)-13 was observed by TLC after 3 h. The resulting solution was concentrated under reduced pressure to give (S)-30 as a brown solid: ¹H NMR (400 MHz, CD₃OD) δ 2.68–2.75 (m, 2H), 3.86 (s, 3H), 4.16–4.19 (m, 1H), 5.27–5.33 (m, 2H), 5.77–5.81 (m, 1H). (R)-30 was made from (R)-13 by a method analogous to that used to prepare (S)-30.

(S,S)-Methyl 2-[(N-Boc-alaninyl)aminopent-4-enoate [(S,S)-31]. A stirred solution of Boc-L-Ala (34; 86 mg, 0.45 mmol, 1.5 equiv) in DCM (5 mL) was treated with amine hydrochloride (S)-30 (50 mg, 0.30 mmol, 1 equiv), DIEA (78 mg, 0.6 mmol, 2 equiv), and HATU (173 mg, 0.45 mmol, 1.5 equiv), stirred at room temperature for 16 h, diluted with DCM (~10 mL), and washed with saturated aqueous NaHCO₃. The layers were separated. The aqueous layer was extracted with DCM (~10 mL). The combined organic layers were washed with brine, dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure to give (S,S)-31 as a brown oil, which was analyzed without further purification: ¹H NMR (400 MHz, CD₃CN) δ 1.27–1.28 (d, J = 7.2, 3H), 1.44 (s, 9H), 2.43–2.51 (m, 1H), 2.53–2.60 (m, 1H), 3.71 (s, 3H), 4.06-4.10 (m, 1H), 4.44-4.49 (m, 1H), 5.10-5.19 (m, 2H), 5.63 (br s, 1H), 5.72-5.82 (m, 1H), 6.91 (br s, 1H). (S,R)-31 was made using (R)-30 by a method analogous to that used to prepare (S,S)-31: ¹H NMR (400 MHz, CD₃CN) δ 1.26–1.28 (d, J = 7.2, 3H), 1.44 (s, 9H), 2.42–2.50 (m, 1H), 2.53–2.60 (m, 1H), 3.70 (s, 3H), 4.05-4.09 (m, 1H), 4.44-4.49 (m, 1H), 5.10-5.18 (m, 2H), 5.62 (br s, 1H), 5.68-5.81 (m, 1H), 6.90 (br s, 1H).

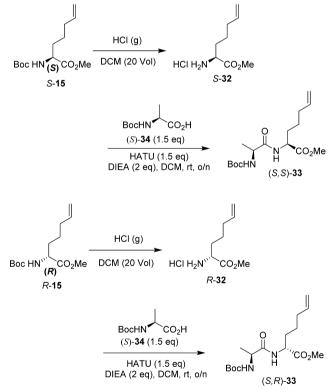
(S)-Methyl 2-(N-Boc-amino)hept-6-enoate (15).



A solution of 9-BBN in THF (0.5 M, 109 mL) was added to a 0 °C solution of methyl pent-4-enoate 13 (5 g, 21.8 mmol) in anhydrous THF (40 mL). The mixture was warmed to room temperature, stirred for 3 h, and quenched with 1 M aqueous Na₂CO₃ in H₂O (76.3 mL) with agitation using argon bubbles for 20 min. To a second round-bottomed flask containing a stirred suspension of $Pd_2(dba)_3$ (393 mg,

0.43 mmol) and tri-o-tolylphosphine (663 mg, 2.18 mmol) in THF at -78 °C was added a 1 M solution of vinyl bromide in THF (87 mL, 87 mmol), followed by a degassed solution of boronate. The resulting mixture was stirred at room temperature overnight, quenched with water (100 mL), and extracted using ethyl acetate (150 mL \times 2). The combined organic extracts were washed with brine (100 mL), dried over anhydrous Na₂SO₄, filtered, and concentrated under reduced pressure to give 7.8 g of brown oil, which was twice purified by column chromatography on silica gel using 4-6% EtOAc in hexane. Evaporation of the collected fractions gave olefin 15 (3.81 g, 14.81 mmol, 68%) as a light brown oil: $R_{f} = 0.29$ (9:1 hexanes/ethyl acetate, visualized as a UV inactive and KMnO₄ active spot on heating); $[\alpha]_{D}^{22}$ +14.6 (*c* 1, CHCl₃); FT-IR (neat) ν_{max} 2977, 1742, 1712, 1501, 1365, 1160. 1050, 909 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.70–5.80 (m, 1H), 4.93–5.01 (m, 3H), 4.27-4.28 (m, 1H), 3.71 (s, 3H), 2.00-2.09 (m, 2H), 1.74-1.82 (m, 1H), 1.56–1.65 (m, 1H), 1.37–1.48 (m, 11H); ¹³C NMR (100 MHz, CDCl₃) δ 173.5, 155.5, 138.1, 115.2, 79.9, 53.4, 52.3, 33.2, 32.3, 28.3, 24.7; HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₁₃H₂₃NO₄Na 280.1519, found 280.1533.

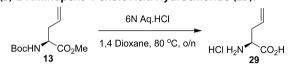
Attempts to ascertain the enantiomeric purity of olefin **15**, as well as the hydrochloride **32** that was obtained on treating **15** with HCl gas in dichloromethane, were both unsuccessful using SFC on a chiral column. To assess enantiomeric purity, diastereomeric amides were synthesized as described below. Crude residue was examined by ¹H NMR spectroscopy in CD₃CN at 700 MHz. Incremental addition of (*S*,*R*)-**33** to (*S*,*S*)-**33** and observation of the methyl ester singlets at 3.6578 and 3.6622 ppm demonstrated the diastereomers were of >99:1 dr. Hence, olefin **15** is assumed to be of >98% enantiomeric purity.



(S)-Methyl 2-Aminohept-6-enoate Hydrochloride [(S)-32]. Dry HCl gas was bubbled into a stirred solution of (S)-15 (100 mg, 0.38 mmol) in dry dichloromethane at room temperature. Consumption of (S)-15 was observed by TLC after 3 h. The resulting solution was concentrated under reduced pressure to give (S)-32 as a white solid: ¹H NMR (500 MHz, CDCl₃) δ 5.78–5.86 (m, 1H), 4.99–5.09 (m, 2H), 4.06–4.09 (t, 1H, J = 6.4 Hz), 3.85 (s, 3H), 2.11–2.16 (m, 2H), 1.86–1.99 (m, 2H), 1.45–1.63 (m, 2H). (R)-32 was made from (R)-15 by a method analogous to that used to prepare (S)-32.

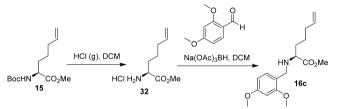
(*S*,*S*)-Methyl 2-[(*N*-Boc-alaninyl)amino]hept-6-enoate [(*S*,*S*)-33]. A stirred solution of 34 (87 mg, 0.46 mmol, 1.5 equiv) in DCM (5 mL) was treated with amine hydrochloride (*S*)-32 (60 mg, 0.31 mmol,

1 equiv), DIEA (80 mg, 0.62 mmol, 2 equiv), and HATU (180 mg, 0.45 mmol, 1.5 equiv), stirred at room temperature for 16 h, diluted with DCM (~10 mL), and washed with saturated aqueous NaHCO₃. The layers were separated. The aqueous layer was extracted with DCM (~10 mL). The combined organic layers were washed with brine, dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure to give (S,S)-33 as a brown oil, which was analyzed without further purification: ¹H NMR (700 MHz, CD₃CN) δ 6.86 (br s, 1H), 5.77-5.83 (m, 1H), 5.57 (br s, 1H), 4.99-5.03 (m, 1H), 4.94-4.96 (m, 1H), 4.34-4.37 (m, 1H), 4.03-4.05 (m, 1H), 3.65 (s, 3H), 2.03-2.06 (m, 2H), 1.75-1.80 (m, 1H), 1.61-1.68 (m, 1H), 1.38-1.42 (m, 11H), 1.24–1.25 (d, J = 7.1 Hz, 3H). Amide (S,R)-33 was made using (R)-32 by a method analogous to that used to prepare (S,S)-33: ¹H NMR (700 MHz, CD₃CN) δ 6.86 (br s, 1H), 5.76–5.82 (m, 1H), 5.57 (br s, 1H), 5.01-5.02 (m, 1H), 4.99-5.00 (m, 1H), 4.33-4.36 (m, 1H), 4.01-4.03 (m, 1H), 3.66 (s, 3H), 2.01-2.08 (m, 2H), 1.75-1.80 (m, 1H), 1.60–1.66 (m, 1H), 1.39–1.40 (m, 11H), 1.24–1.25 (d, J = 7.1 Hz, 3H). (S)-2-Aminopent-4-enoic Acid Hydrochloride (29).



A stirred solution of (*S*)-13 (4 g, 17.4 mmol) in 1,4-dioxane at 0 °C was treated with 6 N HCl (80 mL), heated to 80 °C overnight, cooled, and evaporated to a reside that was dissolved in water and washed with dichloromethane. The aqueous layer was concentrated to off-white solid 29 (2.51 g, 16.61 mmol, 95%): mp 206–208 °C, $[\alpha]_D^{20}$ +7.6 (*c* 1, CH₃OH); FT-IR (neat) ν_{max} 2914, 1730, 1490, 1427, 1215, 1174, 1124, 993, 933, 873 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 5.76–5.87 (m, 1H), 5.26–5.34 (m, 2H), 4.06–4.09 (m, 1H), 2.63–2.79 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 171.1, 131.8, 121.4, 53.4, 35.7 ; HRMS (ESI-TOF) *m*/*z* [M – H]⁻ calcd for C₅H₉ClNO₂ 150.0327, found 150.0330.

(S)-Methyl 2-[(2,4-Dimethoxybenzyl)amino]hept-6-enoate (16c).

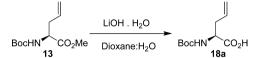


A stirred solution of (*S*)-**15** (3 g, 11.7 mmol) in dichloromethane (60 mL) was treated with HCl gas bubbles for 3 h. The volatiles were removed by rotary evaporation to give (*S*)-**32** (2.19 g, 11.34 mmol, 97%) as a white solid: mp 112–114 °C; $[\alpha]_{2}^{2h}$ +24.6 (*c* 1, CHCl₃); FT-IR (neat) ν_{max} 2922, 2134, 1745, 1438, 1234, 912 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) δ 5.78–5.86 (m, 1H), 4.99–5.09 (m, 2H), 4.06–4.09 (t, 1H, *J* = 6.4 Hz), 3.85 (s, 3H), 2.11–2.16 (m, 2H), 1.86–1.99 (m, 2H), 1.45–1.63 (m, 2H); ¹³C NMR (125 MHz, CD₃OD) δ 170.9, 138.6, 116.0, 53.9, 53.6, 34.0, 30.9, 25.1; HRMS (ESI-TOF) *m*/*z* [M + H]⁺ calcd for C₈H₁₆NO₂ 158.1175, found 158.1179.

Amino ester hydrochloride 32 (2 g, 10.35 mmol) was partitioned between a saturated NaHCO₃ solution (50 mL) and CH_2Cl_2 (50 mL). The aqueous phase was extracted with CH_2Cl_2 (2 × 50 mL), and the combined organic layers were washed with 30 mL of brine, dried over Na_2SO_4 , filtered, and concentrated to a volume of ~30 mL. The solution of free base was treated with 2,4-dimethoxybenzaldehyde (2.23 g, 13.45 mmol) and NaBH(OAc)₃ (3.28 g, 15.5 mmol), stirred for 18 h at room temperature, treated with 40 mL of saturated NaHCO₃, and stirred for 30 min. The aqueous layer was separated and washed with 20 mL of CH₂Cl₂. The combined organic layers were washed with brine, dried over MgSO₄, filtered, and concentrated to a residue that was purified by chromatography on silica gel (20-25% EtOAc in hexane) to give N-Dmb-amino ester 16c (2.09 g, 6.80 mmol, 65%) as a colorless oil: $R_f = 0.35$ (7:3 hexanes/ethyl acetate, visualized by UV), $[\alpha]_D^{20} = -3.1$ (c 1, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 2943, 1732, 1612, 1587, 1505, 1457, 1437, 1287, 1260, 1206, 1154, 1035, 913 cm⁻¹; ¹H NMR (500 MHz, CDCl₂) δ 7.11-7.12 (d, 1H, J = 8.25 Hz), 6.40-6.42 (m, 2H), 5.71-5.78 (m, 1H),

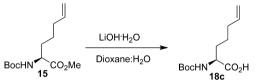
4.91–4.99 (m, 2H), 3.79 (s, 3H), 3.78 (s, 3H), 3.68–3.71 (d, 1H, J = 13.2 Hz), 3.64 (s, 3H), 3.61–3.64 (d, 1H, J = 13.2 Hz), 3.22–3.25 (t, 1H, J = 6.65 Hz), 1.99–2.01 (m, 2H), 1.59–1.65 (m, 2H), 1.39–1.46 (m, 2H); ¹³C NMR (125 MHz, CDCl₃) δ 175.9, 160.1, 158.6, 138.3, 130.3, 120.5, 114.7, 103.7, 98.5, 60.7, 55.4, 55.3, 51.5, 47.1, 33.5, 33.0, 25.0; HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₁₇H₂₅NO₄Na 330.1675, found 330.1691.

(S)-2-(N-Boc-amino)pent-4-enoic Acid (18a).



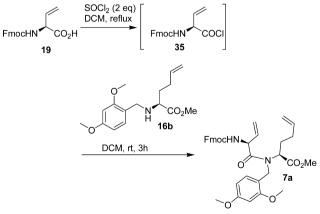
A solution of (S)-13 (1.2 g, 5.23 mmol) in 1:1 H₂O/dioxane (48 mL) was treated with LiOH·H₂O (219 mg, 5.23 mmol), stirred for 3 h, and evaporated to a residue that was partitioned between H₂O (20 mL) and EtOAc (20 mL). The aqueous phase was acidified with 1 M HCl to pH 4 and extracted twice with EtOAc (20 mL). The combined organic extracts were washed with brine, dried over Na₂SO₄, filtered, and concentrated to afford acid **18a** (1.09 g, 5.06 mmol, 97%) as a colorless oil: $[\alpha]_{D^2}^{D^2}$ +12.3 (*c*1, CHCl₃); FT-IR (neat) ν_{max} 2978, 1698, 1699, 1507, 1393, 1367, 1248, 1158, 1050, 869, 754 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) δ 5.74–5.83 (m, 1H,), 5.07–5.15 (m, 2H), 4.13–4.16 (m, 1H), 2.52–2.53 (m, 1H), 2.38–2.42 (m, 1H), 1.44 (s, 9H); ¹³C NMR (125 MHz, CD₃OD) δ 175.4, 157.9, 134.7, 118.5, 80.5, 54.6, 37.1, 28.7; HRMS (ESI-TOF) *m*/*z* [M + Na]⁺ calcd for C₁₀H₁₇NO₄Na 238.1049, found 238.1050.

(S)-2-(N-Boc-amino)hept-6-enoic Acid (18c).



Employing the procedure described for acid **8a**, methyl hept-6-enoate **15** (500 mg, 1.94 mmol) was converted to acid **18c** (445 mg, 1.83 mmol, 94%) as a colorless oil: $[\alpha]_{20}^{D}$ +11.5 (*c* 1, CHCl₃); FT-IR (neat) ν_{max} 2977, 2929, 1711, 1508, 1393, 1367, 1245, 1160, 910 cm⁻¹; ¹H NMR (500 MHz, CD₃OD) δ 5.76–5.84 (m, 1H,), 4.94–5.04 (m, 2H), 4.05–4.08 (m, 1H), 2.03–2.12 (m, 2H), 1.77–1.84 (m, 1H), 1.59–1.67 (m, 1H), 1.51–1.46 (m, 2H), 1.44 (s, 9H); ¹³C NMR (125 MHz, CD₃OD) δ 176.3, 158.1, 139.4, 115.3, 80.4, 54.7, 34.3, 32.3, 28.7, 26.2; HRMS (ESI-TOF) m/z [M – H]⁻ calcd for C₁₂H₂₀NO₄ 242.1397, found 242.1409.

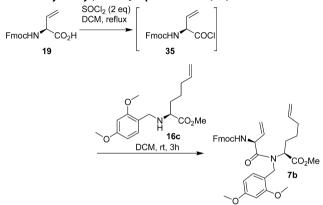
(*S*,*S*)-Methyl 2-[*N*-[2-(*N*-Fmoc-amino)but-3-enoyl]-*N*-(2,4-dimethoxybenzyl)amino]hex-5-enoate (7a).



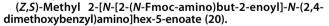
(S)-N-Fmoc-vinylglycine (19; 2.47 g, 7.64 mmol) was treated with thionyl chloride (1.1 mL, 15.32 mmol) in DCM at 0 °C. The resulting solution was then heated to reflux for 3 h and cooled, and the volatiles were evaporated. The residue was dissolved and evaporated three times from DCM. Without further purification, acid chloride **35** was dissolved in DCM (20 mL), cooled to 0 °C, and treated with *N*-Dmb-homoallylglycine (16b; 1.5 g, 5.12 mmol). The resulting mixture was stirred at room temperature for 3 h. The volatiles were evaporated, and

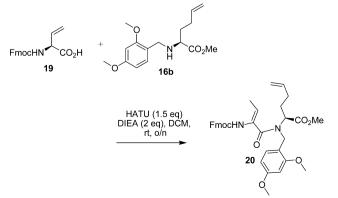
the residue was taken up in a minimum volume of DCM, applied onto a silica gel column, and eluted with 30-40% EtOAc/hexanes to give 7a (1.42 g, 2.37 mmol, 46%) as a colorless oil that solidified on standing: $R_f =$ 0.52 (4:6 EtOAc/hexanes, visualized by UV); $\left[\alpha\right]_{D}^{24}$ -40.1 (c 1, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 2947, 1719, 1647, 1612, 1588, 1506, 1207, 1156, 1032. 759, 738 cm⁻¹; ¹H NMR (500 MHz, DMSO- d_{61} 100 °C) δ 7.84–7.85 (d, 2H, J = 7.5 Hz, 7.67 - 7.70 (t, 2H, J = 6.7 Hz), 7.38 - 7.42 (m, 2H), 7.29 - 7.20 Hz7.32 (m, 2H), 7.18 (br s, 1H), 7.04 (br s, 1H), 6.55 (d, 1H, J = 2.0 Hz), 6.46-6.48 (dd, 1H, J = 2.2 Hz), 5.87-5.95 (m, 1H), 5.63-5.71 (m, 1H), 5.23-5.29 (m, 2H), 5.18-5.20 (br t, 1H), 4.92-4.93 (m, 1H), 4.90-4.91 (m, 1H), 4.55–4.58 (d, 1H, J = 16.4 Hz), 4.39–4.43 (m, 1H), 4.28–4.35 (m, 2H), 4.20-4.23 (m, 2H), 3.77 (s, 3H), 3.75 (s, 3H), 3.50 (s, 3H), 1.91-2.04 (m, 3H), 1.65-1.71 (m, 1H); ¹³C NMR (125 MHz, DMSO-d₆, 100 °C) δ 170.1, 169.8, 160.1, 157.9, 154.6, 143.4, 143.3, 140.2, 137.1, 133.5, 129.5, 127.2, 126.4, 124.6, 124.5, 119.3, 117.0, 114.5, 104.5, 98.2, 78.6, 65.6, 57.5 54.9, 54.8, 53.7, 50.9, 46.4, 29.4, 27.6; HRMS (ESI-TOF) $m/z [M + Na]^+$ calcd for C₃₅H₃₈N₂O₇Na 621.2571, found 621.2562.

(S,S)-Methyl 2-[N-[2-(N-Fmoc-amino)but-3-enoyl]-N-(2,4-dimethoxybenzyl)amino]hept-6-enoate (7b).



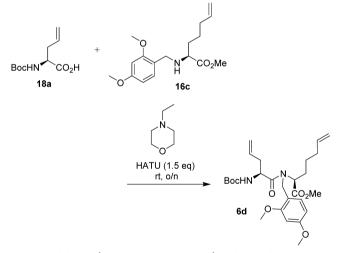
As described for the synthesis of dipeptide 7a, (S)-2-(N-Fmoc-amino)but-3-enoic acid (19; 785 mg, 2.43 mmol) was coupled with 16c (500 mg, 1.62 mmol). The residue was purified by chromatography on silica gel (30-40% EtOAc in hexane) to give dipeptide 7b (540 mg, 0.88 mmol, 54%) as a colorless oil: $R_f = 0.55$ (4:6 EtOAc/hexanes, visualized by UV); $[\alpha]_D^{20} - 30.6$ (c 1, CHCl₃); FT-IR (neat) ν_{max} 2947, 1719, 1647, 1612, 1588, 1506, 1207, 1156, 1032, 916, 758, 738 cm⁻¹; ¹H NMR (500 MHz, DMSO- d_{o} 100 °C) δ 7.84–7.85 (d, 2H, J = 7.5 Hz), 7.67–7.70 (t, 2H, J = 6.6 Hz), 7.38–7.42 (m, 2H), 7.29-7.32 (m, 2H), 7.18-7.19 (m, 1H), 7.03 (br s, 1H) 6.55-6.56 (m, 1H), 6.45-6.47 (m, 1H), 5.84-5.96 (m, 1H), 5.62-5.71 (m, 1H), 5.24-5.29 (m, 2H), 5.17-5.20 (m, 1H), 4.88-4.94 (m, 2H), 4.53-4.57 (d, 1H, J = 16.4), 4.41-4.44 (m, 1H), 4.20-4.35 (m, 4H), 3.75 (s, 3H), 3.78 (s, 3H), 3.51 (s, 3H), 1.81-1.97 (m, 3H), 1.55-1.67 (m, 1H), 1.17-1.37 (m, 2H); ¹³C NMR (125 MHz, DMSO-d₆, 100 °C) δ 170.2, 160.1, 157.7, 154.6, 143.4, 143.3, 140.3, 137.6, 133.6, 129.4, 127.0, 126.4, 124.6, 124.5, 119.3, 117.0, 114.0, 104.5, 98.2, 78.6, 65.6, 57.9, 54.9, 54.8, 53.7, 50.9, 46.4, 45.3, 32.1, 27.7, 24.7; HRMS (ESI-TOF) *m*/*z* [M + Na]⁺ calcd for C36H40N2O7Na 635.2727, found 635.2738.





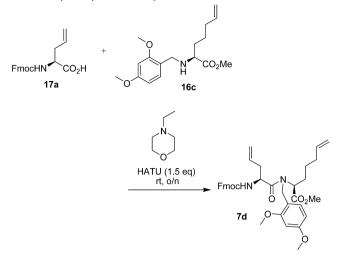
N-Fmoc-vinylglycine (19; 274 mg, 0.85 mmol) and methyl 2-(N-Dmbamino)hex-5-enoate (16b; 170 mg, 0.57 mmol) were dissolved in DCM (10 mL), treated with DIEA (147 mg, 1.14 mmol) and HATU (323.1 mg, 0.85 mmol), stirred for 24 h, and diluted with water. The aqueous layer was extracted with DCM (3×50 mL). The combined organic layers were washed with brine (50 mL), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The residue was purified by column chromatography on silica gel (30-40% EtOAc in hexane) to give dipeptide 20 (302 mg, 0.50 mmol, 87%) as a colorless gummy oil: $R_f = 0.27$ (4.5:5.5 EtOAc/hexanes, visualized by UV); $[\alpha]_D^{20}$ -63.2 (c 1, CHCl₃); FT-IR (neat) ν_{max} 2945, 1716, 1611, 1506, 1448, 1206, 1108, 1032, 757, 738 cm⁻¹; ¹H NMR (500 MHz, DMSO-*d*₆, 100 °C) δ 8.68 (br, s, 1H), 7.85–7.86 (d, 2H, J = 7.6 Hz), 7.71–7.72 (m, 2H), 7.40-7.43 (t, 2H, J = 7.4 Hz), 7.31-7.34 (m, 2H), 7.25-7.27 (d, 1H, J = 8.5 Hz), 6.52–6.53 (d, 1H, J = 2.4 Hz), 6.44–6.46 (dd, 1H, J = 2.4 Hz), 5.60-5.68 (m, 1H), 5.45-5.49 (m, 1H), 4.84-4.89 (m, 2H), 4.56–4.59 (d, 1H, J = 16 Hz), 4.47–4.50 (d, 1H, J = 15.9 Hz), 4.355– 4.356 (d, 1H, J = 0.8 Hz), 4.34 (s, 1H), 4.23–4.26 (t, 1H, J = 13.9 Hz), 4.16 (br s, 1H), 3.77 (s, 3H), 3.7 (s, 3H), 3.51 (s, 3H), 1.88-2.07 (m, 3H), 1.71–1.81 (m, 1H), 1.64–1.66 (d, 3H, *J* = 7 Hz); ¹³C NMR (125 MHz, DMSO-d₆, 100 °C) δ 170.5, 168.1, 159.6, 157.5, 153.3, 143.3, 143.3, 140.3, 137.3, 130.6, 129.3, 127.0, 126.4, 124.6, 119.4, 118.4, 117.2, 114.2, 104.6, 98.0, 78.6, 65.7, 58.2, 54.9, 54.8, 50.8, 46.4, 29.6, 27.9, 11.2; HRMS (ESI-TOF) $m/z [M + Na]^+$ calcd for $C_{35}H_{38}N_2NaO_7$ 621.2571, found 621.2573.

(*S*,*S*)-Methyl 2-[*N*-[2-(*N*-Boc-amino)pent-4-enoyl]-*N*-(2,4-dimethoxybenzyl)amino]hept-6-enoate (6d).



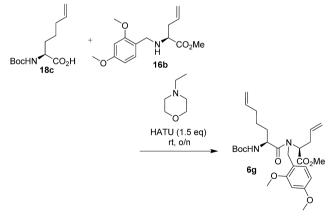
N-Boc-allylglycine (18a; 522 mg, 2.43 mmol) and N-Dmb-amino ester 16c (500 mg, 1.62 mmol) were dissolved in DCM (10 mL), treated with N-ethylmorpholine (373 mg, 3.24 mmol) and HATU (923 mg, 2.43 mmol), stirred for 24 h, and diluted with water. The aqueous layer was extracted with DCM (3×50 mL). The combined organic layers were washed with brine (50 mL), dried over anhydrous sodium sulfate, filtered, and concentrated under reduced pressure. The residue was purified by chromatography on silica gel (30-35% EtOAc in hexane) to give dipeptide 6d (610 mg, 1.21 mmol, 74%) as a colorless gummy oil: $R_f = 0.67$ (4:6 EtOAc/hexanes, visualized by UV); $[\alpha]_D^{20} - 33.1$ (c 1, CHCl₃); FT-IR (neat) ν_{max} 2975, 2932, 1739, 1708, 1639, 1613, 1507, 1437, 1208, 1158, 1033, 753 cm⁻¹; ¹H NMR (500 MHz, DMSO-*d*₆, 100 °C) δ 7.16 (br s, 1H), 6.56 (s, 1H), 6.47–6.49 (d, 1H, J = 7.7 Hz), 6.20 (brs, 1H), 5.63-5.77 (m, 2H), 5.01-5.09 (m, 2H), 4.88-4.95 (m, 2H), 4.61-4.62 (m, 2H), 4.43-4.48 (m, 1H), 4.23 (br s, 1H), 3.80 (s, 3H), 3.77 (s, 3H), 3.51 (s, 3H), 2.30-2.39 (m, 2H), 1.86-1.95 (m, 3H), 1.58-1.70 (m, 1H), 1.39 (s, 9H), 1.20-1.26 (m, 2H); ¹³C NMR (125 MHz, DMSO-d₆, 100 °C) δ 171.6, 170.3, 160.1, 157.9, 154.2, 137.6, 133.5, 129.6, 116.8, 116.6, 114.0, 104.5, 98.2, 78.6, 77.9, 58.5, 54.9, 54.8, 50.8, 50.1, 36.1, 32.2, 28.1, 27.6, 24.6; HRMS (ESI-TOF) m/z $[M + Na]^+$ calcd for $C_{27}H_{40}N_2NaO_7$ 527.2727, found 527.2732.

(S,S)-Methyl 2-[N-[2-(N-Fmoc-amino)pent-4-enoyl]-N-(2,4-dimethoxybenzyl)amino]hept-6-enoate (7d).



Employing the protocol described for the synthesis of dipeptide 6d, (S)-2-(N-Fmoc-amino)pent-4-enoic acid (17a; 1.147 g, 3.40 mmol) was coupled with 16c (700 mg, 2.27 mmol). The residue was purified by chromatography on silica gel (30-40% EtOAc in hexane) to give dipeptide 7d (1.080 g, 1.72 mmol, 76%) as a colorless gummy oil: R_f = 0.57 (4:7 EtOAc/hexanes, visualized by UV); $[\alpha]_{D}^{20}$ -39.7 (c 1, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 2923, 1718, 1638, 1612, 1588, 1506, 1437, 1207, 1156, 758, 739538 cm⁻¹; ¹H NMR (500 MHz, DMSO- d_{6} , 100 °C) δ 7.82 - 7.85 (d, 2H, J = 7.5 Hz), 7.67 - 7.68 (m, 2H), 7.38 - 7.42 (m, 2H), 7.29-7.32 (t, 2H, J = 7.4 Hz), 7.15 (br s, 1H), 6.96 (br s, 1H), 6.54-6.55 (m, 1H), 6.44-6.46 (m, 1H), 5.62-5.75 (m, 2H), 5.02-5.10 (m, 2H), 4.87-4.93 (m, 2H), 4.67 (br s, 2H), 4.41 (br s, 1H), 4.28-4.34 (m, 2H), 4.19-4.22 (t, 2H, J = 13.9), 3.77 (s, 3H), 3.74 (s, 3H), 3.49 (s, 3H), 2.36-2.41 (m, 2H), 1.81-1.93 (m, 3H), 1.56-1.67 (m, 1H), 1.16-1.27 (m, 2H); ¹³C NMR (125 MHz, DMSO- d_6 , 100 °C) δ 171.5, 170.3, 160.1, 158.0, 154.9, 143.4, 140.2, 137.7, 133.4, 129.5, 127.0, 126.4, 124.6, 119.4, 116.9, 114.0, 104.5, 98.2, 78.6, 66.2, 65.4, 57.8, 54.9, 54.8, 50.8, 50.5, 46.4, 45.1, 36.0, 32.1, 27.8, 24.6. HRMS (ESI-TOF) m/z $[M + Na]^+$ calcd for $C_{37}H_{42}N_2O_7Na$ 649.2884, found 649.2873.

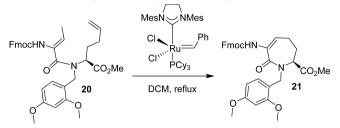
(*S*,*S*)-Methyl 2-[*N*-[2-(*N*-Boc-amino)hept-6-enoyl]-*N*-(2,4-dimethoxybenzyl)amino]pent-4-enoate (6g).



Employing the protocol described for the synthesis of dipeptide **6d**, **18c** (521 mg, 2.14 mmol) was coupled with **16b** (400 mg, 1.43 mmol). The residue was purified by chromatography on silica gel (30–40% EtOAc in hexane) to give dipeptide **6g** (561 mg, 1.11 mmol, 78%) as a colorless gummy oil: $R_f = 0.72$ (4:6 EtOAc/hexanes, visualized by UV); $[\alpha]_{D2}^{12}$ –52.5 (*c* 1, CHCl₃); FT-IR (neat) ν_{max} 2975, 2932, 1739, 1708, 1639, 1613, 1507, 1437, 1208, 1158, 1033, 753 cm⁻¹; ¹H NMR (500 MHz, DMSO- d_6 , 100 °C) δ 7.19 (br s, 1H), 6.56 (s, 1H), 6.47–6.49 (m, 1H), 6.16 (br s, 1H), 5.73–5.81 (m, 1H), 5.60–5.70 (m, 1H), 4.91–5.02 (m, 4H), 4.42–4.55 (m, 3H), 4.19–4.21 (m, 1H), 3.80 (s, 3H), 3.77 (s, 3H), 3.50 (s, 3H), 2.60–2.68 (m, 1H), 2.40–2.46 (m, 1H),

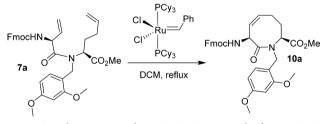
1.95–2.04 (m, 2H) 1.59–1.65 (m, 1H), 1.49–1.57 (m, 1H), 1.44–1.35 (m, 11H); ¹³C NMR (125 MHz, DMSO- d_{6r} 100 °C) δ 172.1, 169.8, 160.1,154.3, 137.8, 134.5, 129.3, 116.3, 114.0, 104.4, 98.2, 78.6, 77.8, 57.8, 54.9, 54.8, 50.9, 50.3, 32.8, 32.7, 32.1, 31.5, 27.6, 23.8, 23.7. HRMS (ESI-TOF) m/z [M + H]⁺ calcd for C₂₇H₄₁N₂O₇ 505.2908, found 505.2920.

(S)-Methyl 3-(*N*-Fmoc-amino)-1-(2,4-dimethoxybenzyl)-2oxo-5,6,7-tetrahydro-1*H*-azepine-7-carboxylate (21).



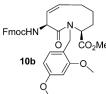
Dipeptide 20 (150 mg, 0.25 mmol) was dissolved in DCM (300 mL), treated with the Grubbs second-generation catalyst (42 mg, 0.050 mmol), heated, and stirred at reflux for 2 days. The volatiles were removed under reduced pressure, and the residue was taken up in a minimum volume of DCM, applied onto a silica gel column, and eluted with 35-40% EtOAc in hexanes. Evaporation of the collected fractions gave macrocycle 21 (95 mg, 68%) as a white foam: $R_f = 0.55$ (4:6 EtOAc/hexanes); $[\alpha]_D^{20}$ +31.8 (c 0.5, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 2947, 1718, 1654, 1612, 1505, 1206, 1154, 1123, 1032, 758, 738 cm⁻¹; ¹H NMR (300 MHz, CDCl₂) δ 7.75-7.77 (m, 2H); 7.59-7.63 (m, 2H), 7.37-7.43 (m, 2H), 7.27-7.34 (m, 3H), 7.05 (br s, 1H), 6.66 (br s, 1H), 6.45–6.49 (m, 2H), 5.16–5.20 (d, 1H, J = 14), 4.39-4.41 (m, 2H), 4.18-4.24 (m, 3H) 3.81 (s, 3H), 3.80 (s, 3H), 3.58 (s, 3H), 2.55-2.60 (m, 1H), 2.13-2.18 (m, 2H), 1.84-1.91 (m, 1H); 13 C NMR (75 MHz, CDCl₃) δ 171.8, 167.9, 161.0, 158.8, 153.6, 143.9, 143.9, 141.4, 141.4, 132.2, 131.5, 127.8, 127.8, 127.2, 125.2, 120.1, 120.0, 117.2, 116.2, 104.3, 98.5, 66.8, 59.5, 55.5, 55.3, 52.7, 47.2, 47.1, 32.8, 21.7; HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₃₂H₃₂N₂NaO₇ 579.2107, found 579.2101.

(35,85,Z)-Methyl 3-(N-Fmoc-amino)-1-(2,4-dimethoxybenzyl)-2-oxo-1,2,3,6,7,8-hexahydroazocine-8-carboxylate (10a).



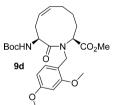
Dipeptide 7a (1 g, 1.67 mmol) was dissolved in DCM (2 L), treated with the Grubbs first-generation catalyst (412 mg, 0.50 mmol), heated, and stirred at reflux for 2 days. After 2 days, TLC showed remaining 7a, and more catalyst (274 mg, 0.33 mmol) was added to the reaction mixture, which was heated at reflux and stirred for 2 days. The volatiles were removed under reduced pressure. The residue was taken up in a minimum volume of DCM, applied onto a silica gel column, and eluted with 35-40% EtOAc in hexanes. Evaporation of the collected fractions afforded macrocycle 10a (605 mg, 1.06 mmol, 63%) as a colorless gummy oil: $R_f = 0.45$ (4:6 EtOAc/hexanes); $[\alpha]_D^{24} + 8.7$ (c 1, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 2947, 1718, 1654, 1612, 1505, 1447, 1206, 1154, 1123, 738 cm⁻¹; ¹H NMR (300 MHz, CDCl₃) δ 7.75–7.77 (d, 2H, J = 7.5 Hz), 7.61–7.63 (d, 2H, J = 7.4 Hz), 7.37–7.42 (t, 2H, J = 7.5 Hz), 7.28–7.33 (m, 2H), 7.16–7.19 (d, 1H, J = 8.3 Hz), 6.40–6.45 (m, 2H), 6.17-6.19 (d, 1H, J = 7 Hz), 5.67-5.79 (m, 2H), 5.28-5.31 (d, 1H, J = 7 Hz), 4.71-4.81 (m, 2H), 4.33-4.43 (m, 2H), 4.12-4.26 (m, 2H), 3.79 (s, 3H), 3.77 (s, 3H), 3.45 (s, 3H), 1.97-2.44 (m, 2H), 1.82-1.91 (m, 2H); ¹³C NMR (75 MHz, CDCl₃) δ 171.4, 169.9, 160.0, 157.7, 155.9, 144.1, 143.9, 141.4, 131.3, 129.2, 128.1, 127.8, 127.2, 125.7, 120.0, 116.8, 104.1, 98.2, 67.3, 57.6, 55.5, 55.4, 53.7, 52.2, 47.2, 41.0, 28.3, 21.4. HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₃₃H₃₄N₂O₇Na 593.2258, found 593.2255.

(35,95,*Z*)-Methyl 3-(*N*-Fmoc-amino)-1-(2,4-dimethoxybenzyl)-2-oxo-1,2,3,6,7,8,9-heptahydro-1*H*-azonine-9-carboxylate (10b).



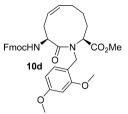
Dipeptide 7b (200 mg, 0.32 mmol) was dissolved in DCM (400 mL), treated with the Grubbs first-generation catalyst (79 mg, 0.096 mmol), heated, and stirred at reflux for 2 days, when TLC indicated remaining 7b, and another portion of catalyst (53 mg, 0.065 mmol) was added to the mixture, which was heated at reflux and stirred for 1 day. The volatiles were removed under reduced pressure. The residue was taken up in a minimum volume of DCM, applied onto a silica gel column, and eluted with 35-40% EtOAc in hexanes. Evaporation of the collected fractions afforded macrocycle 10b (131 mg, 0.22 mmol, 69%) as a colorless gummy oil: $R_f = 0.37$ (4:6 EtOAc/hexanes); $[\alpha]_D^{20}$ +7.1 (c 0.22, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 2946, 1718, 1637, 1612, 1587, 1505, 1447, 1288, 1206, 1034, 739 cm⁻¹; ¹HNMR (400 MHz, CDCl₃) δ 7.75–7.77 (d, 2H, J = 7.4 Hz), 7.60-7.63 (m, 2H), 7.38-7.42 (t, 2H, J = 7.4 Hz),7.30-7.33 (t, 2H, J = 7.4), 7.08-7.10 (d, 1H, J = 8.3 Hz), 6.35-6.42 (m, 3H), 5.69-5.73 (t, 1H, J = 7.8 Hz), 5.56-5.63 (m, 1H), 5.34-5.39 (t, 1H, J = 10 Hz), 4.87-4.91 (d, 1H, J = 15.1 Hz), 4.69-4.71 (d, 1H, J = 7.6 Hz), 4.36–4.38 (m, 2H), 4.22–4.31 (m, 2H), 3.76 (s, 3H), 3.78 (s, 3H), 3.59 (s, 3H), 2.62-2.73 (m, 1H), 2.23-2.2 6 (m, 1H), 2.06-2.11 (m, 1H), 1.63–1.69 (m, 1H), 1.47–1.57 (m, 1H), 1.31–1.39 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 170.3, 170.0, 160.5, 158.3, 155.8, 144.0, 144.0, 141.4, 130.8, 130.7, 129.1, 127.7, 127.1, 125.3, 125.3, 120.1, 120.0, 117.5, 104.1, 98.1, 67.1, 55.4, 55.3, 55.2, 52.4, 50.4, 47.2, 41.6, 28.3, 24.5, 23.7; HRMS (ESI-TOF) $m/z [M + Na]^+$ calcd for $C_{34}H_{36}N_2O_7Na$ 607.2414, found 607.2425

(3*S*,10*S*,*Z*)-Methyl 3-(*N*-Boc-amino)-1-(2,4-dimethoxybenzyl)-2-oxo-1,2,3,4,7,8,9,10-octahydroazecine-10-carboxylate (9d).



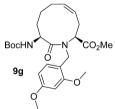
Dipeptide 6d (600 mg, 1.18 mmol) was dissolved in DCM (1300 mL), treated with the Grubbs first-generation catalyst (189 mg, 0.23 mmol), heated, and stirred at reflux for 2 days. The volatiles were removed under reduced pressure, and the residue was taken up in a minimum volume of DCM, applied onto a silica gel column, and eluted with 30-35% EtOAc in hexanes to afford macrocycle 9d (490 mg, 1.02 mmol, 86%) as a white solid: $R_f = 0.55$ (4:6 EtOAc/hexanes); mp 81–83 °C; $[\alpha]_D^{20}$ +15.1 (c 1, CHCl₃); FT-IR (neat) ν_{max} 2944, 1706, 1644, 1612, 1490, 1433, 1363, 1209, 1157, 1130, 1030, 833 cm $^{-1};$ $^1\mathrm{H}$ NMR (500 MHz, CDCl_3) δ 7.04-7.06 (d, 1H, J = 8.2 Hz), 6.42-6.43 (d, 1H, J = 2.2 Hz), 6.34-6.39 (m, 1H), 5.85–5.86 (d, 1H, J = 7.1 Hz), 5.53–5.58 (m, 1H), 5.44–5.46 (m, 1H), 5.23 (br s, 1H), 4.97-4.99 (d, 1H, J = 14.2), 3.93-3.98(m, 1H), 3.78 (s, 3H), 3.75 (s, 3H), 3.37-3.39 (d, 1H, J = 10.5 Hz), 3.30 (s, 3H), 2.79-2.81 (m, 1H), 2.44-2.45 (m, 1H), 2.25-2.37 (m, 2H), 2.02-2.05 (m, 1H), 1.93-1.96 (m, 1H), 1.65-1.66 (m, 1H), 1.46–1.41(m, 10H); ¹³C NMR (125 MHz, CDCl₃) δ 170.2, 169.9, 160.5, 159.0, 154.2, 132.1, 130.5, 123.1, 115.2, 102.3, 97.4, 78.3, 76.3, 61.2, 54.5, 54.1, 50.7, 50.4, 49.8, 28.5, 27.5, 25.0, 22.3; HRMS (ESI-TOF) $m/z [M + Na]^+$ calcd for C₂₅H₃₆N₂O₇Na 499.2414, found 499.2409.

(3*S*,10*S*,*Z*)-Methyl 3-(*N*-Fmoc-amino)-1-(2,4-dimethoxybenzyl)-2-oxo-1,2,3,4,7,8,9,10-octahydroazecine-10-carboxylate (10d).



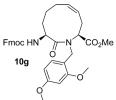
Dipeptide 7d (1 g, 1.60 mmol) was dissolved in DCM (2000 mL), treated with the Grubbs first-generation catalyst (263.34 mg, 0.32 mmol), heated, and stirred at reflux for 2 days. The volatiles were removed under reduced pressure, and the residue was taken up in a minimum volume of DCM, applied onto a silica gel column, and eluted with 35-40% EtOAc in hexanes to afford macrocycle 10d (872 mg, 1.45 mmol, 91%) as a white solid: $R_f = 0.40$ (4:6 EtOAc/hexanes); mp 89–92 °C; $[\alpha]_D^{20}$ –5.8 (c 1, CHCl₃); FT-IR (neat) v_{max} 3005, 1717, 1640, 1611, 1588, 1505, 1207, 1156, 1034, 758, 739, 542 cm⁻¹; ¹H NMR (400 MHz, CDCl₃) δ 7.76–7.78 (d, 2H, J = 7.5 Hz), 7.62–7.65 (m, 2H), 7.38–7.42 (t, 2H, J = 7.4), 7.30–7.33 (m, 2H), 7.06–7.08 (d, 1H, J = 8.2 Hz), 6.38–6.43 (m, 2H), 6.17-6.19 (d, 1H, J = 6.5 Hz), 5.49-5.60 (m, 2H), 5.26-5.28 (m, 1H), 4.97-5.01 (d, 1H, J = 14.4), 4.34-4.47 (m, 2H), 4.22-4.26 (m, 1H), 3.99-4.02 (m, 1H), 3.79 (s, 3H), 3.73 (s, 3H), 3.39-3.44 (m, 1H), 3.33 (s, 3H), 2.80-2.90 (m, 1H), 2.40-2.52 (m, 2H), 2.27-2.36 (m, 1H), 1.84–2.09 (m, 3H), 1.67–1.74 (m, 1H); ¹³C NMR (100 MHz, CDCl₃) δ 171.1, 170.5, 161.5, 159.9, 155.5, 144.1, 144.0, 141.4, 131.5, 127.8, 127.2, 127.1, 125.4, 125.3, 123.8, 120.1, 116.0, 103.5, 98.4, 66.8, 62.3, 55.5, 55.4, 55.4, 55.3, 52.0, 51.7, 47.4, 47.3, 26.0; HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₃₅H₃₈N₂O₇Na 621.2571, found 621.2588.

(35,105,*Z*)-Methyl 3-(*N*-Boc-amino)-1-(2,4-dimethoxybenzyl)-2-oxo-1,2,3,4,5,6,9,10-octahydroazecine-10-carboxylate (9g).



Dipeptide 6g (550 mg, 1.09 mmol) was dissolved in DCM (1000 mL), treated with the Grubbs first-generation catalyst (173 mg, 0.21 mmol), heated, and stirred at reflux for 2 days. The volatiles were removed under reduced pressure, and the residue was taken up in a minimum volume of DCM, applied onto a silica gel column, and eluted with 30-35% EtOAc in hexanes to afford macrocycle 9g (460 mg, 0.96 mmol, 79%) as a white solid: $R_f = 0.62$ (4:6 EtOAc/hexanes); mp 79-82 °C; $[\alpha]_D^{22}$ -40.5 (c 1, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 2943, 1735, 1706, 1637, 1612, 1589, 1506, 1438, 1363, 1208, 1156, 1034, 750, 533 cm⁻¹; ¹H NMR (400 MHz, $CDCl_3$) δ 7.03–7.05 (d, 1H, J = 8.16 Hz), 6.37–6.42 (m, 2H), 5.87– 5.88 (d, 1H, J = 6.76 Hz), 5.48-5.60 (m, 2H), 5.02-5.05 (t, 1H, J = 6.36 Hz), 4.81–4.85 (d, 1H, J = 14.2 Hz), 3.78 (s, 3H), 3.75 (s, 3H), 3.44– 3.49 (m, 1H), 3.42 (m, 3H), 3.26-3.35 (m, 1H), 2.46-2.55 (m, 1H), 2.25-2.31 (m, 1H), 1.98-2.04 (m, 1H), 1.84-1.93 (m, 2H), 1.74-1.78 (m, 1H), 1.39–1.53 (m, 11H); 13 C NMR (100 MHz, CDCl₃) δ 172.5, 170.8, 161.3, 159.9, 155.2, 135.3, 131.4, 125.0, 115.9, 103.4, 98.5, 79.2, 57.9, 55.4, 55.2, 51.8, 50.7, 50.1, 28.5, 27.7, 25.0, 24.1, 21.5; HRMS (ESI-TOF) $m/z [M + Na]^+$ calcd for C₂₅H₃₆N₂O₇Na 499.2414, found 499.2409.

(3*S*,10*S*,*Z*)-Methyl 3-(*N*-Fmoc-amino)-1-(2,4-dimethoxybenzyl)-2-oxo-1,2,3,4,5,6,,9,10-octahydroazecine-10-carboxylate (10g).



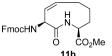
A stirred solution of Boc-protected dipeptide lactam 9g (200 mg, 0.42 mmol) in 10 mL of DCM was treated with TFA (4 mL) at 0 °C and stirred for 2 h, and the volatiles were removed under reduced pressure to provide the trifluoroacetate salt, which was dissolved in 15 mL of a 1:1 water/acetone solution and treated with Fmoc-OSu (138 mg, 0.42 mmol) and Na₂CO₃ (90 mg, 0.84 mmol). After being stirred for 18 h at room temperature, the mixture was diluted with water and acidified to pH 3-4 with 10% KHSO4 solution. This aqueous solution was extracted with ethyl acetate $(3 \times 20 \text{ mL})$. The combined organic layers were washed with brine, dried over MgSO4, filtered, and concentrated to a residue that was purified by chromatography on silica gel using 25-30% EtOAc in hexane as the eluent. Evaporation of the collected fractions gave Fmoc-protected dipeptide lactam 10g (151 mg, 0.25 mmol, 60%) as a colorless oil: $R_f = 0.50$ (4:6 EtOAc/hexanes, visualized by UV); $[\alpha]_{D}^{20}$ –45.2 (*c* 0.5, CHCl₃); FT-IR (neat) ν_{max} 2944, 1715, 1636, 1612, 1588, 1506, 1440, 1207, 1156, 1130, 1035, 758, 739, 535 cm^{-1} ; ¹H NMR (400 MHz, CDCl₃) δ 7.76–7.78 (d, 2H, J = 7.5 Hz), 7.63–7.66 (t, 2H, J = 6.0 Hz), 7.39–7.42 (t, 2H, J = 7.4 Hz), 7.30–7.34 (m, 2H), 7.05–7.07 (d, 1H, J = 8.1 Hz), 6.39–6.43 (m, 2H), 6.21–6.23 (d, 1H, J = 6.7), 5.52–5.64 (m, 2H), 5.08–5.12 (m, 1H), 4.83–4.87 (d, 1H, J = 14.3 Hz), 4.41-4.46 (m, 1H), 4.33-4.39 (m, 1H), 4.22-4.26 (t, 1H, J = 7.2 Hz), 3.79 (s, 3H), 3.75 (s, 3H), 3.48–3.54 (m, 1H), 3.45 (s, 3H), 3.30-3.40 (m, 1H), 2.50-2.62 (m, 1H), 2.30-2.35 (m, 1H), 2.05–211 (m, 1H), 1.80–1.98 (m, 3H), 1.75–1.49 (m, 2H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ 172.1, 170.7, 161.4, 159.8, 155.5, 144.1, 144.1, 141.5, 135.3, 131.4, 127.8, 127.1, 125.4, 125.3, 125.1, 120.1, 120.0, 115.7, 103.5, 98.5, 66.9, 58.0, 55.4, 55.4, 51.9, 50.7, 50.6, 47.4, 27.5, 25.0, 24.2, 21.5; HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₃₅H₃₈N₂O₇Na 621.2571, found 621.2576

(35,85,Z)-Methyl 3-(N-Fmoc-amino)-2-oxo-1,2,3,6,7,8-hexa-hydroazocine-8-carboxylate (11a).



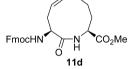
Lactam 10a (155 mg, 0.27 mmol) was treated with TFA (2 mL) in DCM (5 mL) overnight. The volatiles were removed under vacuum, and the residue was purified by chromatography on silica gel (40-50% EtOAc in hexane) to give macrocycle 11a (97 mg, 0.23 mmol, 85%) as a white solid: $R_f = 0.2$ (1:1 EtOAc/hexanes, visualized by UV); mp 183–185 °C, $[\alpha]_{\rm D}^{24}$ +29.5 (c 1, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 3377, 2955, 1728, 1696, 1663, 1528, 1432, 1255, 1050, 984, 721, 549 cm⁻¹; ¹H NMR (400 MHz, $CDCl_3$) δ 7.74–7.76 (d, 2H, J = 7.5 Hz), 7.59–7.61 (m, 2H), 7.37–7.41 (t, 2H, J = 7.4 Hz), 7.29–7.32 (m, 2H), 6.17–6.19 (d, 1H, J = 7.7 Hz), 6.06–6.08 (d, 1H, J = 7.3 Hz), 5.77–5.84 (m, 1H), 5.63–5.68 (m, 1H), 5.15-5.19 (t, 1H, J = 6.5 Hz), 4.44-4.49 (m, 1H), 4.33-4.41 (m, 2H), 4.21-4.24 (t, 1H, J = 7.23 Hz), 3.79 (s, 3H), 2.46-2.51 (m, 2H), 2.16-2.21 (m, 1H), 1.56–1.65 (m, 1H); 13 C NMR (100 MHz, CDCl₃) δ 172.0, 171.8, 155.9, 144.0, 143.9, 141.4, 132.7, 129.2, 127.8, 127.2, 125.3, 125.2, 120.0, 67.3, 55.9, 53.2, 52.2, 47.2, 33.0, 25.5; HRMS (ESI-TOF) $m/z [M + Na]^+$ calcd for C₂₄H₂₄N₂O₅Na 443.1577, found 443.1581.

(3*S*,9*S*,*Z*)-Methyl 3-(*N*-Fmoc-amino)-2-oxo-1,2,3,6,7,8,9-hep-tahydro-1*H*-azonine-9-carboxylate (11b).



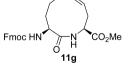
Lactam 10b (53 mg, 0.09 mmol) was treated as described for the synthesis of 11a above with TFA (1 mL) in DCM overnight. Chromatography (30-40% EtOAc in hexane) gave 11b (34 mg, 0.08 mmol, 87%) as a white solid: $R_f = 0.2$ (1:1 EtOAc/hexanes, visualized by UV); mp 202–204 °C; $[\alpha]_D^{20}$ +21.1 (*c* 1, CHCl₃); FT-IR (neat) ν_{max} 3325, 2922, 1736, 1660, 1521, 1434, 1318, 1240, 1204, 1032, 985, 882, 758 cm⁻¹; ¹H NMR (500 MHz, CDCl₃) δ 7.75–7.76 (d, 2H, I = 7.5 Hz), 7.58–7.60 (m, 2H), 7.37–7.41 (t, 2H, I = 7.4 Hz), 7.29– 7.32 (t, 2H, J = 7.4 Hz), 6.37–6.39 (d, 1H, J = 9.0 Hz), 6.13–6.14 (d, 1H, J = 5.9 Hz), 5.68–5.75 (m, 1H), 5.39–5.43 (m, 2H), 4.31–4.39 (m, 3H), 4.20-4.23 (t, 1H, J = 7.2 Hz), 3.80 (s, 3H), 2.80-2.91 (m, 1H), 2.29-2.40 (m, 1H), 1.93-2.01 (m, 2H), 1.76-1.82 (m, 1H), 162-1.71 (m, 1H); $^{13}\mathrm{C}$ NMR (100 MHz, CDCl₃) δ 172.4, 169.8, 155.6, 144.0, 143.9, 141.4, 131.5, 128.1, 127.8, 127.2, 125.2, 120.0, 67.2, 53.1, 50.2, 50.2, 47.2, 34.1, 29.8, 25.6, 23.4; HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₂₅H₂₆N₂O₅Na 457.1733, found 457.1745.

(3*S*,10*S*,*Ž*)-Methyl 3-(*N*-Fmoc-amino)-2-oxo-1,2,3,4,7,8,9,10-octahydroazecine-10-carboxylate (11d).



Lactam 10d (100 mg, 0.17 mmol) was treated with TFA (2 mL) in DCM (5 mL) for 3 days. The volatiles were removed under vacuum, and the residue was purified by chromatography on silica gel (40-50% EtOAc in hexane) to give lactam 11d (52 mg, 0.11 mmol, 69%) as a white solid: $R_f = 0.2$ (1:1 EtOAc/hexanes, visualized by UV); $[\alpha]_{\rm D}^{20}$ +48.1 (c 0.16, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 3327, 2920, 1723, 1649, 1505, 1437, 1349, 1221, 1022, 756 cm⁻¹; ¹H NMR (500 MHz, 100 °C DMSO- d_6) δ 7.86–7.87 (d, 2H, J = 7.5 Hz), 7.81–7.82 (m, 1H), 7.86– 7.89 (d, 2H, J = 7.5 Hz), 7.40–7.43 (t, 2H, J = 7.4), 7.32–7.35 (m, 2H), 6.86 (br s, 1H), 5.35-5.46 (m, 2H), 4.34-4.40 (m, 2H), 4.23-4.26 (m, 1H), 4.08-4.18 (m, 2H) 3.65 (s, 3H), 2.36-2.39 (m, 1H), 2.27-2.28 (m, 1H), 1.97-2.15 (m, 2H), 1.66-1.91 (m, 3H), 1.55-1.57 (m, 1H); ¹³C NMR (125 MHz, 100 °C DMSO-*d*₆) δ 172.0, 169.8, 144.4, 141.3, 134.3, 128.0, 127.5, 125.5, 123.8, 121.7, 120.6, 120.4, 120.3, 79.6, 66.4, 55.4, 54.3, 52.1, 47.5, 30.0, 26.1, 25.9, 25.6; HRMS (ESI-TOF) m/z $[M + Na]^+$ calcd for $C_{26}H_{28}N_2O_5Na$ 471.1890, found 471.1898.

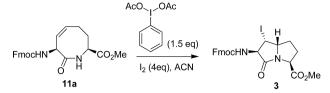
(35,105,Z)-Methyl 3-(*N*-Fmoc-amino)-2-oxo-1,2,3,4,5,6,9,10-octahydroazecine-10-carboxylate (11g).



Lactam **10g** (40 mg, 0.07 mmol) was treated as described for the synthesis of **11a** above with TFA (0.5 mL) in DCM overnight. Chromatography (40–50% EtOAc in hexane) gave lactam **11g** (24 mg, 0.05 mmol, 80%) as a white solid: $R_f = 0.2$ (1:1 EtOAc/hexanes, visualized by UV); mp 210–215 °C; $[\alpha]_D^{22}$ +2.4 (c 0.16, CHCl₃); FT-IR (neat) ν_{max} 3296, 2926, 1726, 1685, 1643, 1527, 1438, 1247, 1105, 1031, 755, 736 cm⁻¹; ¹H NMR (500 MHz, 100 °C, DMSO- d_6) δ 7.92 (br, s, 1H), 7.84–7.86 (d, 2H, J = 7.6 Hz), 7.67–7.69 (m, 2H), 7.39–7.42 (t, 2H, J = 7.4 Hz), 7.30–7.33 (t, 2H, J = 7.5), 6.74 (br s, 1H), 5.48–5.53 (m, 1H), 5.34–5.41 (m, 1H), 4.32–4.33 (m, 2H), 4.21–4.24 (t, 1H, J = 6.9), 4.06–4.09 (m, 1H), 3.90 (br s, 1H), 3.65 (s, 3H), 2.75–2.82 (m, 1H), 2.42–2.43 (m, 1H), 2.12–2.20 (m, 1H), 1.83–1.89 (m, 2H), 1.74–1.78 (m, 2H), 1.44–1.48 (m, 1H); ¹³C NMR (125 MHz, 100 °C, DMSO- d_6) δ 171.3, 170.4, 154.5, 143.4, 140.3, 134.4, 127.0, 126.5, 124.6, 124.2, 119.4, 65.4, 53.2, 51.7, 51.2, 46.5, 35.8, 29.4, 24.7, 23.7,

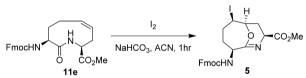
21.7, 16.5; HRMS (ESI-TOF) $m/z \,[M + Na]^+$ calcd for $C_{26}H_{28}N_2O_5Na$ 471.1890, found 471.1903.

(3*R*,4*R*,55,8*S*)-Methyl 3-(*N*-Fmoc-amino)-4-iodo-2-oxohexahydro-5*H*-pyrrolizine-8-carboxylate (3).



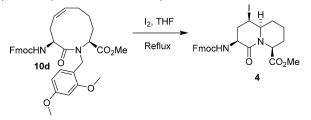
In the dark, a solution of macrocycle 11a (100 mg, 0.24 mmol) in acetonitrile (4 mL) was treated with iodine (243 mg, 0.96 mmol) followed by (diacetoxyiodo)benzene (116 mg, 0.36 mmol). The resulting mixture was heated to 80 °C for 30 min and cooled to room temperature, and the volatiles were evaporated under reduced pressure. The residue was chromatographed on silica gel (40-50% EtOAc in hexane) to give pyrrolizidinone 3 (81 mg, 0.15 mmol, 62%) as a light-sensitive white solid: $R_f = 0.45$ (3:2 EtOAc/hexanes, visualized by UV); mp 94–96 °C; $[\alpha]_{\rm D}^{20}$ –70.8 (c 1, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 3374,1754, 1739, 1692, 1522, 1444, 1380, 1254, 1220, 1166, 1076, 1049, 818, 755, 741, 532 cm⁻¹; ¹H NMR (700 MHz, C_6D_6) δ 7.57–7.58 (d, 2H, J = 7.7 Hz), 7.43–7.44 (m, 2H), 7.17 - 7.24 (m, 4H), 4.92 - 4.93 (d, 1H, J = 7.6 Hz), 4.49 - 4.51 (t, J = 7.6 Hz), 4.49 (t,1H, J = 7.4), 4.40–4.43 (t, 1H, J = 7.6 Hz), 4.34–4.35 (d, 2H, J = 6.5 Hz), 3.96–3.98 (t, 1H, J = 6.5 Hz), 3.90–3.92 (t, 1H, J = 6.6 Hz), 3.26–3.29 (m, 1H), 3.22 (s, 3H), 1.70–1.78 (m, 1H), 1.43–1.48 (m, 1H), 1.32–1.34 (m, 1H), 0.80–0.83 (m, 1H); 13 C NMR (175 MHz, C₆D₆) δ 172.0, 171.0, 156.2, 144.5, 144.4, 141.8, 127.4, 127.4, 125.5, 120.2, 67.0, 63.9, 62.6, 57.6, 51.8, 47.5, 33.5, 30.3, 30.0, 22.5; HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₂₄H₂₃IN₂O₅Na 569.0543, found 569.0559.

Imidate 5.



In the dark, a solution of macrocycle 11e (50 mg, 0.11 mmol) in acetonitrile (4 mL) was treated with NaHCO₃ (27 mg, 0.33 mmol) followed by iodine (83.7 mg, 0.33 mmol) in three portions, stirred for 1 h at rt, and treated with 1 M Na₂S₂O₃ until the purple solution became clear. The mixture was extracted quickly with ethyl acetate $(3 \times 10 \text{ mL})$. The organic extractions were washed with brine, dried, and concentrated under reduced pressure to a residue that was chromatographed on silica gel (20-30% EtOAc in hexane). Evaporation of the collected fractions gave imidate 5 (33 mg, 0.06 mmol, 51%) as a white solid: $R_f = 0.45$ (4.5:6.5 EtOAc/hexanes, visualized by UV); mp 152-154 °C; $[\alpha]_{\rm D}^{20}$ +48.4 (c 1, CHCl₃); FT-IR (neat) $\nu_{\rm max}$ 3410, 2923, 1738, 1715, 1657, 1490, 1449, 1352, 1277, 1208, 1162, 1020, 958, 903, 739, 541 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 7.76–7.77 (d, 2H, J = 7.4 Hz), 7.60-7.63 (m, 2H), 7.39-7.41 (m, 2H), 7.31-7.33 (m, 2H), 6.17-6.18 (d, 1H, J = 5.9 Hz), 4.89–4.93 (m, 1H), 4.67–4.69 (m, 1H), 4.38–4.39 (d, 2H, J = 7 Hz), 4.26–4.29 (m, 1H), 4.21–4.23 (t, 1H, J = 6.9), 3.95– 3.97 (m, 1H), 3.85 (s, 3H), 2.69–2.73 (m, 1H), 2.12–2.17 (m, 2H), 2.03-2.10 (m, 1H), 1.94-1.98 (m, 2H); ¹³C NMR (175 MHz, CDCl₃) δ 172.0, 164.8, 155.7, 144.1, 143.9, 141.4, 141.4, 127.8, 127.2, 125.3, 125.2, 120.1, 120.1, 67.1, 55.3, 52.7, 51.3, 47.3, 36.7, 33.2, 30.3, 30.1, 14.1; HRMS (ESI-TOF) m/z [M + Na]⁺ calcd for C₂₅H₂₅IN₂O₅Na 583.0700, found 583.0705.

(3*S*,*SR*,*6R*,10*S*)-Methyl 3-(*N*-Fmoc-amino)-5-iodo-2-oxooctahydro-6*H*-quinolizine-10-carboxylate (4).



In the dark, a solution of macrocycle 10d (300 mg, 0.50 mmol) in THF (10 mL) was treated with iodine (508 mg, 2 mmol), heated to 80 °C for 8 h, and cooled to room temperature, and the volatiles were removed under reduced pressure to a residue that was chromatographed on silica gel (40-50% EtOAc in hexane) to give quinolizidinone 4 (151 mg, 0.26 mmol, 53%) as a white solid: $R_f = 0.45$ (3:2 EtOAc/hexanes); mp 98– 101 °C; $[\alpha]_{D}^{20}$ +16.2 (c 1, CHCl₃); FT-IR (neat) ν_{max} 2947, 1717, 1655, 1508, 1446, 1324, 1226, 738, 538 cm⁻¹; ¹H NMR (700 MHz, CDCl₃) δ 7.75–7.76 (d, 2H, J = 7.1 Hz), 7.57–7.58 (d, 2H, J = 7.3 Hz), 7.38–7.40 (t, 2H, J = 7.4 Hz), 7.30-7.32 (t, 2H, J = 7.3 Hz), 5.75 (br s, 1H), 4.69-4.70 (m, 1H), 4.31-4.42 (m, 2H), 4.17-4.20 (m, 2H), 3.73 (s, 3H), 3.65-3.68 (m, 1H), 3.44-3.45 (m, 1H), 3.06-3.08 (m, 1H), 2.45-2.51 (m, 1H), 2.24-2.25 (m, 1H), 2.06-2.09 (m, 1H), 1.94-1.98 (m, 1H), $1.83-1.89 (m, 1H), 1.59-1.60 (m, 2H)^{13} C NMR (175 MHz, CDCl₃) \delta$ 169.6, 168.6, 156.2, 143.9, 143.8, 141.4, 141.4, 127.8, 127.2, 125.3, 120.1, 67.2, 62.9, 60.2, 52.6, 52.4, 47.2, 37.7, 30.1, 24.8, 22.9, 20.3; HRMS (ESI-TOF) $m/z [M + Na]^+$ calcd for C₂₆H₂₇IN₂O₅Na 597.0856, found 597 0861

ASSOCIATED CONTENT

Supporting Information

¹H and ¹³C NMR spectra for all new compounds and X-ray data for compounds **10d**, **11a**, **11e**, **3**, and **5** (PDF and CIF). This material is available free of charge via the Internet at http://pubs. acs.org.

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Notes

The authors declare no competing financial interest.

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