

Total Synthesis of (–)-Strychnine

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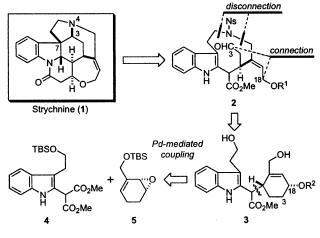
Strychnine (1), a well-known poison first isolated as far back as 1818,¹ has generated considerable attention among synthetic chemists mainly because of its architecturally complex structural features, including the unique heptacyclic framework as well as the six contiguous chiral centers.² While the first total synthesis was accomplished in 1954 by Woodward,^{3a} strychnine remains a popular target for demonstrating new reactions and novel synthetic strategies.^{3,4} In this communication, we report a concise stereo-controlled total synthesis of strychnine wherein efficient synthetic methodologies developed in our laboratories played crucial roles.

As illustrated in Scheme 1, our retrosynthesis relies upon an efficient construction of the core skeleton of 1 from the ninemembered cyclic intermediate 2, which would be derived from the corresponding diol by means of our 2-nitrobenzenesulfonamide (NsNH₂) strategy.⁵ To secure the geometry of the trisubstituted olefin, we planned to cleave the cyclohexene ring at the C3–18 bond of 3 to generate the two side chains. The precursor 3 should be available from the palladium-mediated coupling reaction of indolylmalonate 4 and vinyl epoxide 5.

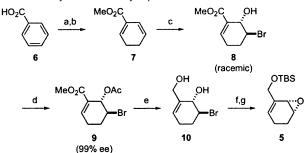
The synthesis of vinyl epoxide **5** commenced with methyl 1,5cyclohexadienecarboxylate (**7**),⁶ which is readily available from benzoic acid (**6**) (Scheme 2). Treatment of **9** with NBS in the presence of water gave bromohydrin **8**, which was subjected to an enzymatic resolution with lipase AYS to provide the desired chiral bromohydrin acetate **9** (46%, 99% ee) along with the unreacted enantiomer (50%, 99% ee). After reduction of **9** with DIBAL, the resultant bromohydrin **10** was treated with base to form the epoxide, and the subsequent protection of the primary alcohol as the TBS ether gave the desired vinyl epoxide **5**.

With the requisite vinyl epoxide **5** in hand, the next task was to perform the palladium-mediated coupling reaction⁷ with indolylmalonate **4**,⁸ which was efficiently prepared in large quantities by modification of our protocol involving radical cyclization of 2-alkenylthioanilides. While initial attempts at the coupling reaction with Pd(PPh₃)₄ in THF provided the desired product **11**, the yield was only 5% at best. After extensive investigation, we have found that the choice of catalyst and ligand is crucial for this process. Thus, when treated with Pd₂(dba)₃ and P(2-furyl)₃ in toluene, the yield increased dramatically to afford **11** in 86% yield with complete control of regio- and stereoselectivity (Scheme 3). The coupling product **11** was then converted into diol **12** by a four-step sequence involving protection of the secondary alcohol as the MOM ether, decarbomethoxylation, installation of a Boc group on the indole nitrogen, and deprotection of the two TBS groups.

For the construction of the nine-membered cyclic amine intermediate, we devised a ring-closing double *N*-alkylation of NsNH₂ with diol **12** (Scheme 3).⁹ Thus, upon subjection of diol **12** to the Mitsunobu reaction¹⁰ with NsNH₂, the desired nine-membered Nsamide **13** was obtained in 95% yield as the sole product. At this stage, the epimeric mixture at C16 was equilibrated to the thermodynamically more stable β -ester by treatment with DBU at Scheme 1. Retrosynthetic Analysis



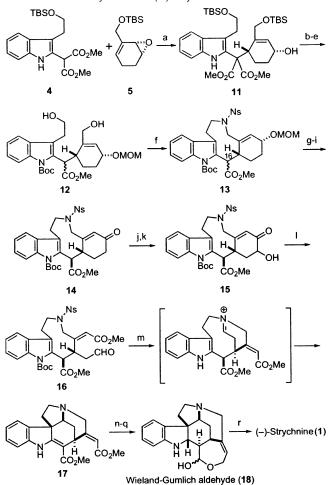
Scheme 2. Synthesis of Vinyl Epoxide^a



^{*a*} Reagents and conditions: (a) Na, liq NH₃-EtOH, -78 °C; (b) AcCl, MeOH, room temp; NaOMe; (c) NBS, H₂O, DMSO, room temp, 62% (three steps); (d) Lipase AYS, vinyl acetate, 40 °C, 46%, 99% ee; (e) DIBAL, CH₂Cl₂, 0 °C, 73%; (f) NaOMe, MeOH, room temp; (g) TBSCl, imidazole, CH₂Cl₂, room temp, 61% (two steps).

100 °C. Removal of the MOM group followed by Dess–Martin oxidation¹¹ of the resultant alcohol gave ketone **14**, which was converted into α -hydroxyketone **15** according to Rubottom's protocol.¹²

The crucial construction of the core skeleton was next achieved through a transannular formation of the iminium ion and ensuing cyclization. Oxidative cleavage of the α -hydroxyketone **15** was best performed by treatment with Pb(OAc)₄ in methanol benzene to furnish aldehyde **16** bearing $\alpha_{,}\beta$ -unsaturated ester with the desired geometry. Removal of the Ns group from **16** using the standard conditions (PhSH,Cs₂CO₃), followed by treatment with TFA and Me₂S in one-pot induced a smooth transannular cyclization^{3b,13-15} to give the pentacyclic compound **17** in 84% yield from **15**, which is an intermediate of the Kuehne's total synthesis of **1**.^{3e} Finally, **17** was converted into (–)-strychnine(**1**) via the Wieland–Gumlich aldehyde (**18**)^{16,17} by way of the five-step sequence reported earlier.^{3d,e} All spectroscopic data of synthetic **1** were identical to those of natural strychnine.



^a Reagents and conditions: (a) Pd₂(dba)₃, P(2-furyl)₃, toluene, room temp, 86%; (b) MOMCl, *i*-Pr₂NEt, CH₂Cl₂, room temp; (c) LiI, collidine, 80 °C; (d) Boc₂O, DMAP, MeCN, room temp; (e) NH₄F·HF, DMF-NMP, room temp, 72% (four steps); (f) NsNH₂, PPh₃, DEAD, toluene, room temp, 95%; (g) DBU, toluene, 100 °C; (h) aq HCl, THF, 50 °C; (i) Dess-Martin periodinane, CH₂Cl₂, 0 °C, 69% (three steps); (j) TMSOTf, Et₃N, CH₂Cl₂, 0 °C; (k) mCPBA, aq NaHCO₃-CH₂Cl₂, 0 °C; aq HCl, MeOH, room temp, 66% (two steps); (l) Pb(OAc)₄, MeOH-benzene, 0 °C; (m) PhSH, Cs₂CO₃, MeCN; TFA, Me₂S, CH₂Cl₂, 50 °C, 84% (two steps); (n) DIBAL, BF₃•OEt₂, CH2Cl2, -78 °C, 93%; (o) NaBH3CN, AcOH, 10 °C; (p) NaOMe, MeOH-THF, room temp; (q) DIBAL, CH₂Cl₂, -98 °C; (r) CH₂(CO₂H)₂, NaOAc, Ac₂O, AcOH, 110 °C, 42% (four steps).

In conclusion, we have completed an stereocontrolled total synthesis of (-)-strychnine (1), demonstrating the uniqueness of the nitrobenzenesulfonamide chemistry in constructing the mediumsized cyclic amine. Finally, it should be noted that the facile construction of the polycyclic core skeleton was made possible by the removal of the nosyl group under very mild conditions. We believe that the chemistry described herein would be useful for the preparation of a variety of alkaloids.

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Supporting Information Available: Experimental details and spectroscopic data (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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