TOTAL SYNTHESIS OF MIMOCIN

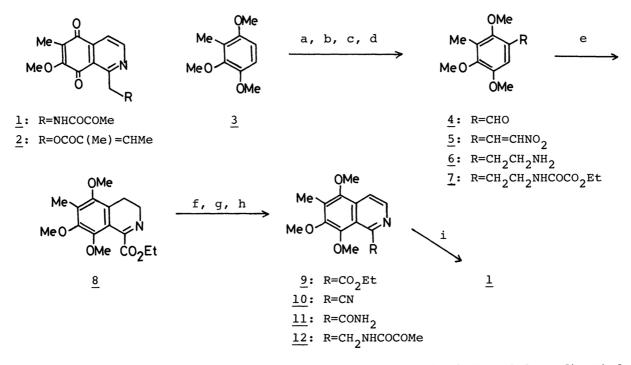
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Mimocin (1), an isoquinolinequinone antibiotic, is synthesized in ten steps starting from 2,3,6-trimethoxytoluene.

Mimocin (1), an isoquinolinequinone antibiotic exhibiting strong activity against <u>B. subtilis</u> and <u>C. albicans</u>, was isolated by Kubo and co-workers¹⁾ from the strain of <u>Streptomyces lavendulae</u>, and has been synthesized starting from 7-methoxy-6-methyl-5,8-isoquinolinedione.

In our course of synthetic studies on marine natural products,²⁾ we interested in the structural similarity between mimocin (<u>1</u>) and renierone (<u>2</u>),³⁾ isolated as an antimicrobial substance from a marine sponge, and their biological activity, and have carried out an alternate synthesis of the former.

Our synthetic approach toward mimocin $(\underline{1})$ was initiated by formylation of 2,3,6-trimethoxytoluene $(\underline{3})^{4}$ with hexamethylenetetramine and trifluoroacetic acid

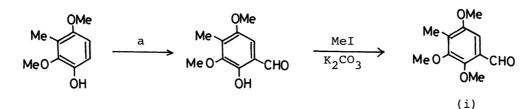


a) hexamethylenetetramine- CF_3CO_2H , b) $MeNO_2$ -NaOAc, c) $LiAlH_4$ -AlCl₃, d) ethyl oxalyl chloride- Et_3N , e) polyphosphate ester, f) Pd/C, heat, g) Me_2AlNH_2 , h) l) H_2 , Pd/C, 2) Cl₂CHOMe-MeCOCO₂H, i) AgO-HNO₃,

giving an aldehyde (4)^{5),6)} (58%). The aldehyde (4) was condensed with nitromethane in the presence of sodium acetate to furnish α, β -unsaturated nitro derivative $(5)^{6}$ (mp 121-123°C) in 77% yield which was then reduced to an amine ($\underline{6}$) with LiAl₄-AlCl₃ in ether in 84% yield. Treatment of the amine ($\underline{6}$) with ethyl oxalyl chloride in the presence of triethylamine afforded an amide ester (7)⁶⁾ in 91% yield, which was submitted to Bischler-Napieralski reaction. The amide ester (7) was allowed to react with polyphosphate ester without solvent at 110-120°C for 9.5 hr furnishing a ring closed ester (8)⁶⁾ in 55% yield, although the reaction with phosphoryl chloride resulted in poor yield (18%). Heating of the ester (8) in the presence of 5% Pd/C in decalin at 160-170°C gave an isoquinoline ester $(9)^{6}$ in 60% yield. A nitrile $(10)^{6}$ (mp 95.5-97.5°C) was obtained in 83% yield by treatment of the ester (9) with dimethylaluminum amide⁸⁾ in benzene at refluxing temperature for 1.5 days accompanying an amide (11)(9%). The nitrile (10) was then hydrogenated over 5% Pd/C in methanol containing hydrogen chloride to give an amine hydrochloride which, without isolation, was treated with α, α -dichloromethyl methyl ether and pyruvic acid⁹ to afford an amide (12)⁶ (mp 148-151°C) in 43% yield. Finally mimocin (<u>1</u>) (mp 189-190°C, lit.¹⁾ 189-191°C) was obtained by AgO-HNO₃ oxidation¹⁰⁾ of the amide (<u>12</u>) in 53% yield. IR and NMR spectra of natural mimocin $(\underline{1})$ and synthetic one were identical.¹¹⁾

References

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- 5) The structure of <u>4</u> was determined by inspection of NMR spectral data of <u>4</u> and its isomer (i). cf. W.Baker, J.F.W.McOmie, and D.Miles, J.Chem.Soc., 1953, 820.



NMR: δ(CDCl₃) <u>4</u>: 7.25 (lH, s, arom-H), 10.32 (lH, s, CHO); (i): 7.30 (lH, s, arom-H), 10.34 (lH, s, CHO).

- 6) Satisfactory IR, NMR and microanalyses or MS data were obtained.
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- 11) We are indebted to Prof. A.Kubo, Meiji College of Pharmacy, for IR and NMR spectra of natural mimocin.

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