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LETTERS TO THE EDITOR

Synthesis of Diacylals of Amino-Substituted Aldehydes

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It is known that the reaction of aldehydes and carboxylic acid anhydrides affording diacylals (1,1-dialkanoates) occurs only in the presence of acid catalysts BF₃ and PCl₃ [1].

We first found that dialkylamino-substituted aldehydes I react with carboxylic acids anhydrides II under mild conditions without any catalyst.

$$R_{2}^{1}NCH_{2}CMe_{2}CHO + (R^{2}CO)_{2}O$$

$$I$$

$$II$$

$$\rightarrow R_{2}^{1}NCH_{2}CMe_{2}CH(OCOR^{2})_{2}$$

$$III$$

III, $R^1 = R^2 = Me(\mathbf{a})$, $R^1 = Me$, $R^2 = Et(\mathbf{b})$, $R^1 = Me$, $R^2 = Ph(\mathbf{c})$, $R^1 = Et$, $R^2 = Me(\mathbf{d})$.

Obviously, the starting aldehyde I with tertiary amine moiety acts as a nucleophilic catalyst B (Scheme 1).

An equimolar mixture of compounds I and II was kept at room temperature for two weeks till the disappearance of the aldehyde proton signal at δ 10 ppm in the ¹H NMR spectrum of the reaction mixture. After the removal of volatile substances the residue was distilled to give diacylals **IIIa–IIIb** as individual compounds. Product **IIIc** is a thick liquid non-distillable in a vacuum. It was identified as a quaternary ammonium salt **IV** (Scheme 2).

To a solution of the compound **IIIc** in acetonitrile was added dropwise a slight excess of methyl iodide at room temperature under stirring. After 6 h the solvent and volatiles were removed, the residue was recrystallized from ethanol.

3-(Dimethylamino)-2,2-dimethylpropane-1,1diethanoate (IIIa). A mixture of 5 g (0.038 mol) of 3-(dimethylamino)-2,2-dimethylpropanal **Ia** and 3.95 g (0.038 mol) of acetic anhydride was kept in a sealed ampule for 14 days and then distilled in vacuum. Yield 8 g (54%), bp 51–52°C (0.065 mm Hg), n_D^{20} 1.4336. ¹H NMR spectrum (acetone- d_6), δ , ppm: 0.8 s (6H, CMe₂), 1.93 s (6H, COMe), 2.07 s (2H, CH₂N), 2.17 s

Scheme 1.

$$\mathbf{I} + \mathbf{\ddot{B}} \stackrel{\mathbf{I}}{\longleftrightarrow} \mathbf{R}_{2}^{1} \mathrm{NCH}_{2} \mathrm{CMe}_{2} \mathrm{CH} - \mathrm{O}^{-} \underbrace{\overset{(\mathrm{R}^{2}\mathrm{CO})_{2}\mathrm{O}}{\overset{|}_{\mathrm{B}^{+}}} \qquad \mathbf{R}_{2}^{1} \mathrm{NCH}_{2} \mathrm{CMe}_{2} \mathrm{CH} - \mathrm{O}^{-} \underbrace{\overset{|}_{\mathrm{O}^{-}}}{\overset{|}_{\mathrm{O}^{-}} \mathrm{COR}^{2}}$$

$$\longrightarrow \mathbf{R}_{2}^{1} \mathrm{NCH}_{2} \mathrm{CMe}_{2} \mathrm{CH} \mathrm{OCOR}^{2} \xrightarrow{\mathrm{III}} + \mathbf{\ddot{B}}$$

$$\overbrace{\mathbf{B}^{+}}^{|}{\mathbf{Scheme 2.}}$$

 $\label{eq:Me2NCH2CMe2CH(OCOPh)_2 + MeI \rightarrow \Gamma \ Me_3N^+CH_2CMe_2CH(OCOPh)_2} IV$

(6H, NMe₂), 6.5 s (1H, CH). Found N, %: 6.17. $C_{11}H_{21}NO_4$. Calculated N, %: 6.06.

3-(Dimethylamino)-2,2-dimethylpropane-1,1-dipropanoate (IIIb) was obtained from 5 g (0.038 mol) of **Ia** and 5.04 g (0.038 mol) of propionic anhydride. Yield 6.3 g (62.7%), bp 73–74°C (0.082 mm Hg), n_D^{20} 1.4352. ¹H NMR spectrum (acetone- d_6), δ , ppm: 6.55 s (1H, CH), 2.15 s (2H, CH₂N), 2.70 s (6H, NMe₂), 2.10 q (4H, COCH₂, ³J_{HH}7 Hz), 0.90 t (6H, COCH₂CH₃, ³J_{HH}7 Hz), 0.80 s (6H, CMe₂). Found N, %: 6.02. C₁₃H₂₅NO₄. Calculated N, %: 5.93.

3-(Dimethylamino)-2,2-dimethylpropane-1,1-dibenzoate (IIIc) was obtained from 5 g (0.038 mol) of **Ia** and 8.76 g (0.038 mol) of benzoic anhydride. Yield 12.66 g (92%). Ammonium salt of **IIIc** was obtained as follows. To a solution of 12.36 g (0.035 mol) of compound **IIIc** in 35 ml of anhydrous acetonitrile was added dropwise 5.93 g (0.042 mol) of methyl iodide at room temperature. The mixture was stirred at 25°C for 6 h. Yield 17.1 g (93.5%), mp 124–125°C (ethanol). ¹H NMR spectrum (CDCl₃), δ , ppm: 1.5 s (6H, CMe₂), 3.7 s (9H, NMe₃), 4.08 s (2H, CH₂N), 7.41 s (1H, CH), 7.15–8.1 m (10H, Ph). Found N, %: 6.22. C₁₃H₂₅NO₄. Calculated N, %: 5.93. **3-(Diethylamino)-2,2-dimethylpropane-1,1-diethanoate (IIId)** was obtained from 5 g (0.032 mol) of **Ib** and 3.25 g (0.032 mol) of acetic anhydride. Yield 4.7 g (57%), bp 72–73°C (0.01 mm Hg), n_D^{20} 1.4368. ¹H NMR spectrum (acetone- d_6), δ , ppm: 0.8 s (6H, CMe₂), 0.91 t (6H, NCH₂*Me*, ³*J*_{HH} 6.75 Hz), 1.93 s (6H, COMe), 2.07 s (2H, CH₂N), 2.5 q (4H, NCH₂Me, ³*J*_{HH} 6.75 Hz), 6.5 s (1H, CH). Found N, %: 6.22. C₁₃H₂₅NO₄. Calculated N, %: 5.93.

The ¹H NMR spectra were registered on a Tesla BS-567A spectrometer operating at 100 MHz, internal reference TMS. The ³¹P NMR spectra were recorded on a RYa-2303 instrument (21 MHz) relative to 85% H₃PO₄.

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