Spring and Young: Sulphanilamide Derivatives.

64. Sulphanilamide Derivatives.

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A number of sulphanilamide derivatives carrying an alkyl group substituent at N^1 have been prepared for test as tuberculocides. Preliminary results show that they are inactive in this respect.

ALTHOUGH there is considerable evidence showing that sulphanilamide drugs have no appreciable effect upon the spread of experimental tuberculosis in animals infected in various ways (Smithburn, Proc. Soc. Exp. Biol. Med., 1938, 38, 574; Kolmer, Raiziss, and Rule, ibid., 1938, 39, 581; Steinbach and Dillon, ibid., 1939, 41, 613; Dietrich, Amer. Rev. Tuberc., 1938, 38, 388; Flippin, Forrester, and Fitz-Hugh, ibid., 1940, 42, 821; Heise and Steenken, *ibid.*, p. 801), yet it has been repeatedly claimed that sulphanilamide in relatively large doses will retard the dvelopment of experimental tuberculosis (Rich and Follis, Bull. Johns Hopkins Hosp., 1938, 62, 77; 1939, 65, 466; Buttle and Parish, Brit. Med. J., 1938, ii, 776; Greey, Boddington, and Little, Proc. Soc. Exp. Biol. Med., 1939, 40, 418). It is also claimed that sulphapyridine retards the development of experimental tuberculosis, and similar claims have been made for various sulphanilamide derivatives and sulphones (Ballon, Guernon, and Simon, Amer. Rev. Tuberc., 1942, 45, 217; Ballon and Guernon, ibid., p. 212; Barach, Molomut, and Soroka, ibid., 1942, 46, 268; Feldman, Hindshaw, and Moses, Proc. Staff Meet. Mayo Clinic, 1940, 15, 695; 1941, 16, 187; Amer. Rev. Tuberc., 1942, 45, 212, etc.).

In the present study, a group of sulphanilamide derivatives containing an alkyl group attached to N^1 * have been prepared in order to test their tuberculocidal properties. It was hoped that the introduction of such an alkyl group would lead to greater penetration by the drug of the waxy structure of the tubercle bacillus (cf. Crossley, Northey, and Hultquist, J. Amer. Chem. Soc., 1939, 61, 2952; Robinson, J., 1940, 505; Steinbach and Duca, Proc. Soc. Exp. Biol. Med., 1940, 44, 133; Muschenheim, Forkner, and Duerschner, ibid., 1940, 45, 556; Bergmann and Haskelberg, J. Amer. Chem. Soc., 1941, 63, 2243). It was appreciated that the bactericidal properties of the parent sulphanilamide would be reduced by the introduction of an alkyl group, but it was hoped that this effect would be offset by the increase in lipoid solubility.

 N^1 -Heptadecylsulphanilamide and NN'-disulphanilyltetramethylenediamine were each prepared by the action of acetylsulphanilyl chloride upon the requisite omine, followed by hydrolysis of the N⁴-acetyl group, and also

by the condensation of p-nitrobenzenesulphonyl chloride with the amine, followed by reduction of the nitro-group. Various derivatives of sulphapyridine of the general formula (I), in which $R = n - C_3 H_7$, $n - C_5 H_{11}$, $n - C_{16} H_{33}$, $n - C_{18} H_{37}$, and geranyl, together with N1-2-(6-methylpyridyl)-N1-octadecylsulphanilamide have been prepared.

Not one of these sulphanilamides has any action upon the tubercle bacillus in vitro; furthermore, there appears to be a decline in their inhibitory action upon hæmolytic streptococci as the size of the alkyl group increases.

EXPERIMENTAL.

NN'-Di-(p-nitrobenzenesulphonyl)tetramethylenediamine.—Adipamide (4.5 g.) was added to a solution prepared by addition of bromine (10 g.) to a mixture of sodium hydroxide (27.5 c.c.; 33%) and ice (50 g.). The mixture was heated on the steam-bath for 3 hours. The cold solution was then shaken with p-nitrobenzenesulphonyl chloride (13.85 g.) in ether (140 c.c.). The aqueous solution was separated, and acidified with 10% hydrochloric acid; the crude product was collected, washed with water, and purified by solution in hot potassium hydroxide solution (12%), followed by precipitation of the filtered solution with acid. After three crystallisations from aqueous acetone, NN'-di-(p-nitrobenzenesulphonyl)tetramethylenediamine was obtained as golden plates, m. p. 201° (Found : C, 41.9; H, 4.2. C.-N.S. requires C 41.9: H 3.9%)

hollowed by precipitation of the hittered solution with acid. After three crystansations from aqueous accord, i.e., di-(p-figOgN45g requires C, 41-9; H, 3-9%).
NN'-Disulphanilyltetramethylenediamine.—(a) A suspension of the nitro-compound (0.5 g.) in a mixture of alcohol (225 c.c.) and hydrochloric acid (d 1.15; 25 c.c.) was heated under reflux with tin (1 g.) for 1 hour. After standing overnight, the separated tin complex was collected and decomposed by heating with aqueous sodium carbonate solution (10%). The product was collected and crystallised from 80% alcohol (charcoal), yielding the disulphanilyl compound as needles, m. p. 205°, identical with the product prepared by method (b).
(b) A cooled solution of putrescine, prepared from adjamide (9 g.) as described above, was shaken with acetyl-sulphanilyl choride (30 g.) in ether (300 c.c.). The solid separating at the interface was collected (5 g.) and crystallised from methyl alcohol, from which NN'-di(acetylsulphanilyl)tetramethylenediamine separated as platelets, m. p. 233° (sintering at 218°) (Found : C, 49-8; H, 5-6. C₂₉H₂₉O₆N₃S₂ requires C, 49-8; H, 5-4%). Hydrolysis was effected by heating under reflux for 1 hour with alcoholic hydrochloric acid (90 c.c. alcohol; 10 c.c. co.c. acid). The hydrochloride was collected (microneedles, m. p. 241°), dissolved in aqueous acetone, and the solution neutralised by addition of sodium carbonate solution. Crystallisation of the product from 80% alcohol gave the disulphanilyl compound as needles, m. p. 205°, either alone or when mixed with the specimen prepared by method (a) (Found : C, 48-2; H, 5-5%).
N¹-n-Hepiadecylsulphanilamide.—(a) Heptadecylamine (5 g.) in ether (100 c.c.) was shaken with p-nitrobenzene-sulphonyl choride (4-75 g.) in ether (100 c.c.). After removal of the ether, the residue was crystallised from aqueous accone, from which heptadecyl-p-nitrobenzenesulphonamide (3 g.) separated as cream-coloured plates, m. p. 90-5° (Found : C, 62-7; H, 9-2. C₂₃H₄

choice active as described above gave N⁴-*n*-neptadecylsulphanianide (0.25 g.) as small needles, in. p. 118° either alone or when mixed with the specimen prepared by method (b). (b) Condensation of heptadecylamine (25 g.) and acetylsulphanilyl chloride (23 g.) in ether gave N⁴-acetyl-N¹-n-heptadecylsulphanilamide (22 g.), which separated as fine needles from alcohol, m. p. 128° (Found : C, 66·4; H, 9·8. $C_{25}H_{44}O_3N_2S$ requires C, 66·4; H, 9·7%). Hydrolysis of this acetyl derivative with alcoholic hydrochloric acid, followed by treatment with sodium carbonate, gave N¹-n-heptadecylsulphanilamide</sup> as needles, m. p. 118° (Found : C, 66·9; H, 9·9. $C_{23}H_{42}O_2N_2S$ requires C, 67·3; H, 10·2%).

* The nomenclature is that of Crossley, Northey, and Hultquist, J. Amer. Chem. Soc., 1938, 60, 2217.

p-NH2·C6H4·SO2·NR-

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N¹-2-Pyridyl-N¹-n-propylsulphanilamide.—A solution of freshly distilled 2-n-propylaminopyridine (Slotta and Franke, Ber., 1930, **63**, 690) (10 g.) in dry pyridine (50 c.c.) was treated with acetylsulphanilyl chloride (17 g.), added during 30 minutes with stirring. The mixture was heated on the steam-bath for 3 hours and poured in water. The oil was collected

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