

Novel Acetoxylation and C–C Coupling Reactions at Unactivated Positions in α -Amino Acid Derivatives

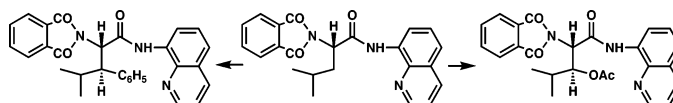
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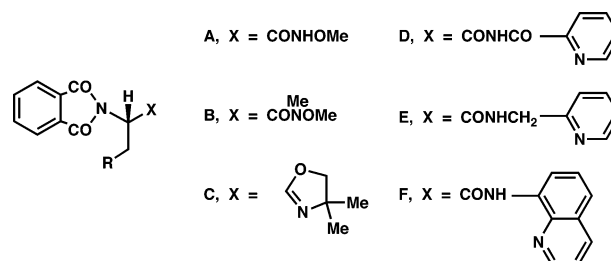
ABSTRACT



Under special conditions, N -phthaloyl- α -amino acid amides of 8-aminoquinoline can be either acetoxyated or arylated selectively at the β -carbon. In certain cases, arylation can be effected at the γ -carbon.

We recently described a method for the selective functionalization of various α -amino acids at the δ -carbon that is very useful because it provides rapid and efficient access to many useful chiral unnatural α -amino acids.¹ In that study, we utilized the α -amino function to direct the replacement of δ -C–H by δ -C–Br. Encouraged by the success of this approach, we turned our attention to the possibility of using the carboxyl function to direct replacement of a β -C–H atom of an α -amino acid by a β -C–OAc or β -C–OH group. β -Hydroxy- α -amino acids are important naturally occurring compounds. In addition to the two genetically coded proteinogenic amino acids serine and threonine, members of the β -hydroxy- α -amino acid class occur as constituents of many bioactive natural compounds. β -Hydroxyleucine is especially prominent as a building block for lactacystin, salinosporamide A, and various cyclodepsipeptides, including azinotricin, papuamides, and polyoxypeptins.² Although there have been several syntheses of β -hydroxyleucine, none of

these have used the conceptually simplest route, β -hydroxylation of leucine.² We report herein the development of such a process. The approach used was the carboxamide-directed Pd(OAc)₂-catalyzed oxidative conversion of β -CH₂ to β -CHO–Ac. This transformation has not previously been applied to α -amino acid derivatives as far as we are aware. The specific substrates employed in this research were amide derivatives of N -phthaloyl-protected leucine, alanine, β -methylalanine, β -ethylalanine, and β -phenylalanine. It should be noted that the overwhelming number of examples of sp³-C–H functionalization using Pd(OAc)₂ catalysis involves insertion into methyl groups;³ insertion into methylene groups generally does not occur under the conditions that suffice for CH₃ insertion.^{3f}

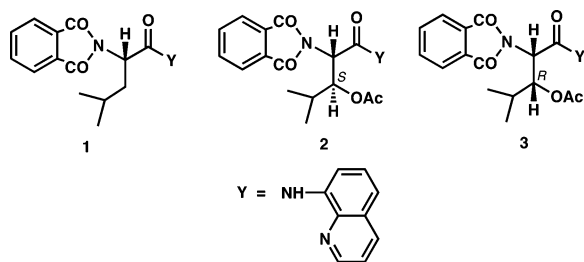


In our initial studies, a number of N -phthaloylamino acid amides of types A–F were screened using the t -BuOOH–Ac₂O–Pd(OAc)₂ system in toluene (C₇H₈) at 110 °C. Of

(1) Reddy, L. R.; Reddy, B. V. S.; Corey, E. J. *Org. Lett.* **2006**, *8*, 2819.

(2) For synthesis of β -hydroxyleucines, see: (a) Schollkopf, U.; Groth, U.; Gull, M.-R.; Nozulak, J. *Liebigs Ann. Chem.* **1983**, 1133. (b) Jung, M. E.; Jung, Y. H. *Tetrahedron Lett.* **1989**, *30*, 6637. (c) Caldwell, C. G.; Bondy, S. S. *Synthesis* **1990**, 34. (d) Blaser, D.; Seebach, D. *Liebigs Ann. Chem.* **1991**, 1067. (e) Sunazuka, T.; Nagamitsu, T.; Tanaka, H.; Omura, S.; Sprengler, P. A.; Smith, A. B., III. *Tetrahedron Lett.* **1993**, *34*, 4447. (f) Panek, J. S.; Masse, C. E. *J. Org. Chem.* **1998**, *63*, 2382. (g) Horikawa, M.; Busch-Petersen, J.; Corey, E. J. *Tetrahedron Lett.* **1999**, *40*, 3843. (h) MacMillan, J. B.; Molinsky, T. F. *Org. Lett.* **2002**, *4*, 1883. (i) Saravanan, P.; Corey, E. J. *J. Org. Chem.* **2003**, *68*, 2760. (j) Makino, K.; Hamada, Y. *J. Synth. Chem. Jap.* **2005**, *63*, 1198.

these, only the 8-aminoquinoline derivatives (**F**) appeared to be promising for further study. Even with this most favorable derivative for functionalization, we were not able to obtain useful yields of β -acetoxy- α -phthaloylamino acid 8-aminoquinoline amide products under previously employed conditions for Pd-catalyzed acetoxylation. For instance, when *N*-phthaloyl-leucine 8-aminoquinoline amide **1** and 0.2 equiv of Pd(OAc)₂ were heated with either 2 equiv of C₆H₅I(OAc)₂ and Ac₂O in ClCH₂CH₂Cl at 80 °C for 12 h or 5 equiv of *t*-BuOOH and 5 equiv of Ac₂O in C₇H₈ at 110 °C for 8 h, none of the desired β -acetoxyated product **2** could be detected and, in fact, only the starting material **1** was recovered. These conditions have been shown to effect functionalization at β -methyl groups in ketone *O*-methyloximes.^{3f,h,j} However, we were pleased to find that treatment of **1** with 20 mol % of Pd(OAc)₂, 5 equiv of *t*-BuOOH, and 5 equiv of Ac₂O in benzene at 80 °C in the presence of 1.2 equiv of Mn(OAc)₂ gave **2** in 30% yield. This noteworthy acceleration



of reaction rate by Mn(II) has not previously been reported. Although AgOAc has been found to be beneficial to certain Pd(OAc)₂-catalyzed C–C couplings with aryl iodides,⁴ no reaction was observed when Mn(OAc)₂ was replaced by AgOAc in the acetoxylation experiment with **1**. Additional experimentation demonstrated that Cu(OAc)₂ and Co(OAc)₂ were also ineffective. With Mn(OAc)₂ as a promoter, we found that Oxone (2KHSO₅/KHSO₄/K₂SO₄) (5 equiv) was superior as an oxidant to *t*-BuOOH and that CH₃NO₂ and ClCH₂CH₂Cl (especially the former) served as the most effective solvents for the β -acetoxylation of the α -phthaloylamino acid amides investigated in the present work. Thus, the reaction of **1** with 20 mol % of Pd(OAc)₂, Oxone (5 equiv), acetic anhydride (10 equiv), and Mn(OAc)₂ (1.2 equiv) in CH₃NO₂ at 80 °C for 22 h (under air) afforded the crystalline (3*S*)-acetate **2** in 60% isolated yield along with the (3*R*)-diastereomer **3** (ca. 4%). The structure of **2** was

(3) (a) Constable, A. G.; McDonald, W. S.; Sawkins, L. C.; Shaw, B. L. *J. Chem. Soc., Chem. Commun.* **1978**, 1061. (b) Carr, K.; Sutherland, J. K. *J. Chem. Soc., Chem. Commun.* **1984**, 1227. (c) Baldwin, J. E.; Nájera, C.; Yus, M. *J. Chem. Soc., Chem. Commun.* **1985**, 126. (d) Jun, C.-H. *Chem. Soc. Rev.* **2004**, 610. (e) Jun, C.-H.; Moon, C. W.; Lee, D.-Y. *Chem.–Eur. J.* **2002**, 8, 2423. (f) Desai, L. V.; Hull, K. L.; Sanford, M. S. *J. Am. Chem. Soc.* **2004**, 126, 9542. (g) Giri, R.; Chen, X.; Yu, J.-Q. *Angew. Chem., Int. Ed.* **2005**, 44, 2112. (h) Giri, R.; Liang, J.; Lei, J.-G.; Li, J.-J.; Wang, D.-H.; Naggar, I. C.; Guo, C.; Foxman, B.; Yu, J.-Q. *Angew. Chem., Int. Ed.* **2005**, 44, 7420. (i) Dick, A. R.; Hull, K. L.; Sanford, M. S. *J. Am. Chem. Soc.* **2004**, 126, 2300. (j) Desai, L. V.; Malik, H. A.; Sanford, M. S. *Org. Lett.* **2006**, 8, 1141. (k) Zaitsev, V. G.; Shabashov, D.; Daugulis, O. *J. Am. Chem. Soc.* **2005**, 127, 13154. (l) Chen, X.; Hao, X.-S.; Goodhue, C. E.; Yu, J.-Q. *J. Am. Chem. Soc.* **2006**, 128, 6790.

(4) The use of AgOAc in Pd-catalyzed C–C bond formation with aryl iodides has been reported; see: Shabashov, D.; Daugulis, O. *Org. Lett.* **2005**, 7, 3657.

confirmed by single-crystal X-ray diffraction analysis (Figure 1).⁵ The diastereoselectivity of the reaction was estimated

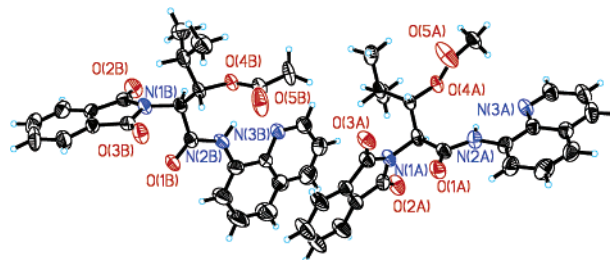
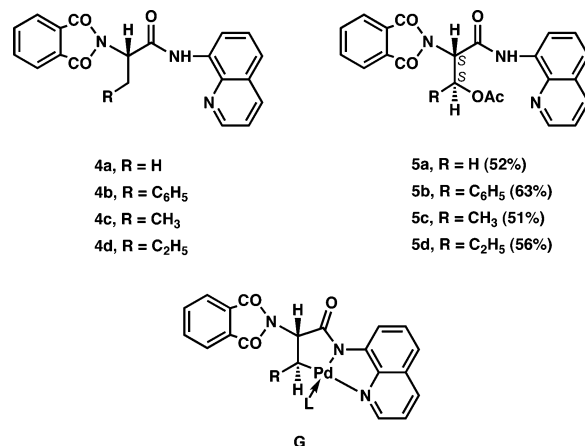


Figure 1. ORTEP representation of **2**.

by ¹H NMR analysis of the total reaction product as 20:1.

Under the same conditions as those described just above for the transformation of **1** to **2**, the alanine derivative **4a** was converted into the serine derivative **5a** in 52% yield. These conditions also resulted in the analogous conversion with the β -phenylalanine series of **4b** to **5b** (63%). In the case of reactant **4b**, the reaction was completely diastereoselective for **5b**, whereas for **4c** and **4d** diastereoselectivities were on the order of 5:1 and 8:1, respectively, favoring in each case the (3*S*)-diastereomers (**5c** and **5d**). The observed stereochemistry of the β -functionalization can be understood in terms of a preference for forming the sterically more favored intermediate *trans*-palladacycle **G**.



The palladacycle **G**, R = *i*-Pr, could be generated, trapped, and defined structurally by the reaction of **1** with *p*-iodoanisole (4 equiv), Pd(OAc)₂ (20 mol %), and AgOAc (1.5 equiv, to remove HI) at 110 °C for 30 min (without solvent) which afforded the crystalline 2*S*,3*S*-3-*p*-anisyl derivative **6** in 95% yield.⁶ The structure of **6** was verified by single-crystal X-ray diffraction analysis (Figure 2).⁵

Our results are generally supportive of the most recent mechanistic discussions of Pd(II)-mediated sp³-C–H inser-

(5) Carried out by Dr. Richard Staples; see Supporting Information for details.

(6) For precedent, see ref 3k.

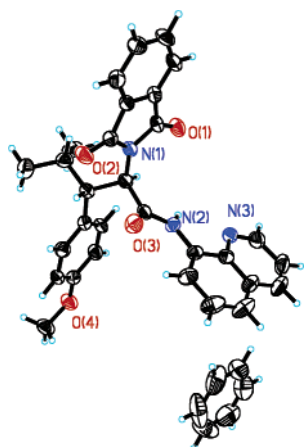
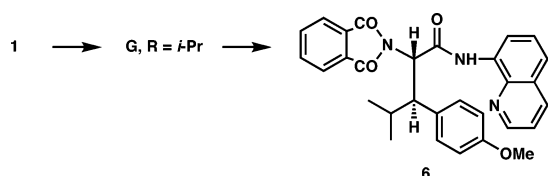


Figure 2. ORTEP representation of **6**.



tion and functionalization.⁷ In terms of the substrates and conditions involved in the reactions reported herein, any mechanistic proposal needs to take into account the rate (and yield) enhancing effects of the additive $\text{Mn}(\text{OAc})_2$ and the solvent CH_3NO_2 . One possibility for the role of $\text{Mn}(\text{OAc})_2$ is that it undergoes oxidation to $\text{Mn}_3\text{O}(\text{OAc})_7$ which functions as a Lewis acid to increase the positive charge on Pd in the first reaction intermediate (**7**, in Scheme 1),⁸ thereby lowering the barrier for C–H insertion and allowing concerted formation of the palladacycle **8** and HOAc .⁹ Successful β -acetoxylation requires that the oxidation of **8** to a Pd(IV) species be rapid so that it can compete with decomposition pathways such as Pd–C homolysis or β -C–H elimination of Pd(0). Oxone appears to be the oxidant that is best suited to this task. A reasonable role of Ac_2O could be its functioning to produce the diacetate **9** by acetylation of the initial Pd(IV) intermediate. This intermediate is the logical precursor of the reaction product **2**, with regeneration of $\text{Pd}(\text{OAc})_2$. The efficacy of CH_3NO_2 as solvent is due in part to the fact that it dissolves the various reactants and in part to its noncoordinating, polar nature which maximizes the electrophilicity (δ^+) of the Pd species responsible for C–H insertion.⁹

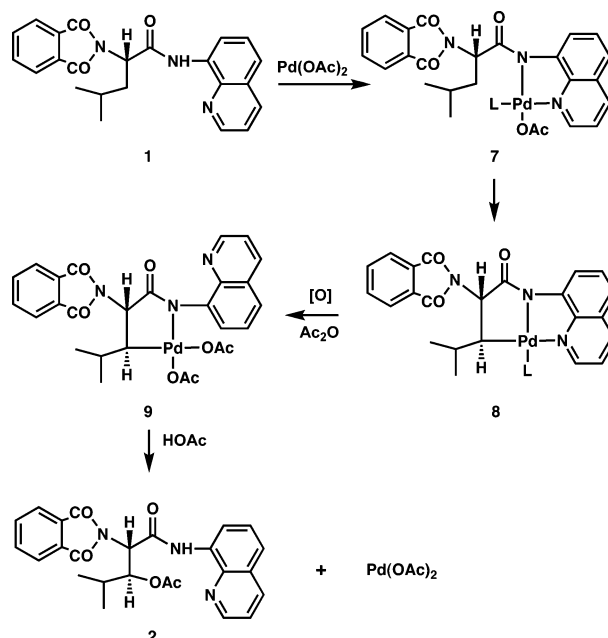
Although the initial objective of this research was the development of a methodology for the β -acetoxylation of α -amino acid derivatives, the ease and high yield of the conversion of **1** to **6** provided encouragement to examine

(7) See, especially, refs 3h, 3j, and 3k.

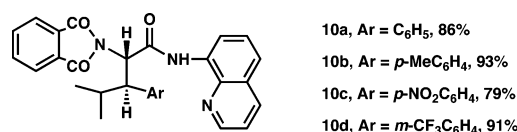
(8) This intermediate which is formed by heating **1** with $\text{Pd}(\text{OAc})_2$ has been isolated.

(9) The greater the positive charge is on Pd, the lower the energy of the vacant d-orbital that interacts with the C–H σ -bond and the more favorable the agostic interaction leading to C–H insertion.

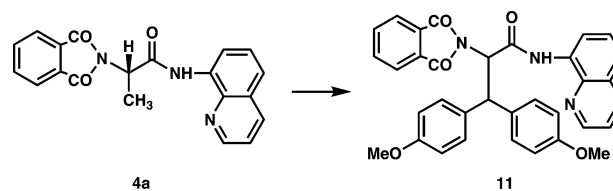
Scheme 1. Possible Pathway for the β -Acetoxylation of *N*-Phthaloyl-leucine 8-Aminoquinoline Amide (**1**) by Pd(II) Catalysis



this β -arylation reaction to assess scope. With regard to the arylation agent, several other aryl iodides were examined with outstanding results. Thus, the series of β -arylated leucine derivatives **10a–d** was readily prepared in the isolated yields indicated.

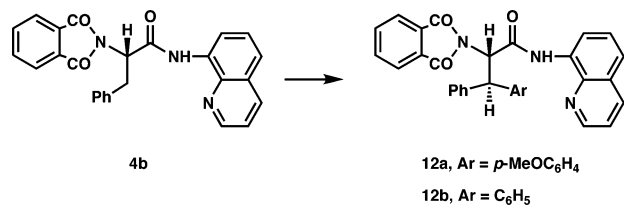


An interesting result was obtained when the alanine 8-aminoquinoline amide **4a** was treated with 20 mol % of $\text{Pd}(\text{OAc})_2$ and 1.5 equiv of AgOAc in *p*-iodoanisole (4 equiv) as solvent at 110 °C for 1.5 h. The product was the unusual diarylated alanine **11** (92% yield), which can also be viewed as a β -*p*-anisylated tyrosine derivative.

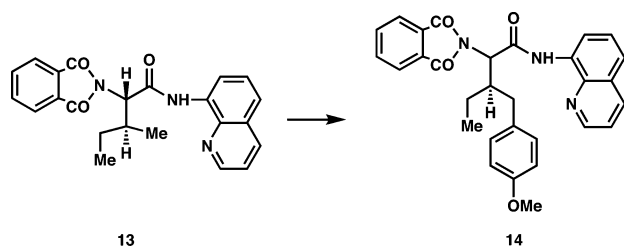


β -Diarylated alanine derivatives were also made from the *N*-phthaloylated phenylalanine amide **4b**, which was efficiently converted into **12a** (91%) or **12b** (89%) with *p*-iodoanisole or iodobenzene, respectively. Clearly, **12a** would be much more difficult to synthesize stereoselectively by other routes.

An equally fascinating outcome was seen in the reaction of the isoleucine derivative **13** with *p*-iodoanisole under the



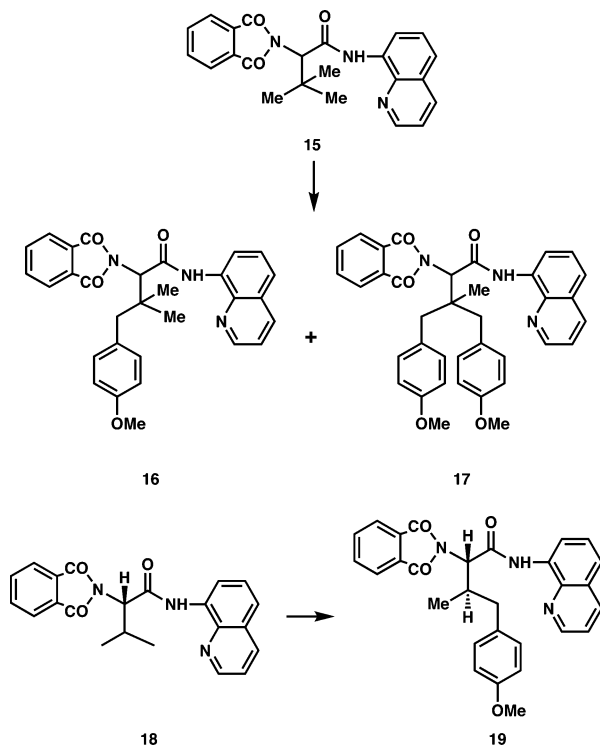
arylation conditions described just above, but with a modestly longer reaction time (2.5 h at 110 °C). The isoleucine substrate was cleanly transformed into the γ -CH₃ arylation product **14** in 87% yield. This result indicates that when the



β -hydrogen is attached to a tertiary carbon the rate of C–H activation (i.e., insertion) by Pd is so attenuated that insertion into a γ -methyl C–H can compete. Obviously, the arylated isoleucine analogue **14** would not be easy to synthesize by other means. In extension of this result, we found that the *t*-leucine derivative **15** was smoothly transformed into a mixture of mono- and diarylated products **16** and **17** with *p*-iodoanisole (4 equiv) under the standard conditions after 3.5 h.

Finally, when the coupling reaction was applied to *N*-phthaloylvaline 8-aminoquinoline amide **18** and *p*-iodoanisole under the standard conditions (1.5 equiv of AgOAc, 20 mol % of Pd(OAc)₂, 4 equiv of *p*-iodoanisole at 110 °C for 3.5 h; no solvent), the γ -arylated product **19** was obtained in 85% yield. It is difficult to imagine another synthesis of **19** as simple as this.

The β -acetoxylation and γ -C–C coupling reactions described herein provide easy access to a broad range of



unnatural (*S*)- α -amino acids using readily available, naturally occurring (*S*)- α -amino acids as starting materials. This methodology and the previously described process of selective δ -halogenation of (*S*)- α -amino acid derivatives¹ open up new avenues of research in synthetic and biological chemistry.

Supporting Information Available: Experimental procedures and characterization data for the new compounds reported herein. X-ray crystallographic data for compounds **2** and **6**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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