A Ready, Convergent Synthesis of the Heptasaccharide GPI Membrane Anchor of Rat Brain Thy-1 Glycoprotein¹

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That certain proteins are covalently attached to cell surfaces via glycosylphosphatidylinositol (GPI) anchors was first established eight years ago by Ferguson and co-workers for the variant surface glycoprotein (VSG) of the parasitic protozoan Trypanosoma brucei.3 Since then over 100 GPI-anchored proteins have been identified in eukaryots4 with diverse functions, including adhesion molecules, hydrolases, receptors, and transmembrane signal inducers.^{4,5} Common structural features in the complex oligosaccharide anchoring domain, indicative of a high degree of evolutionary conservation, would seem to suggest functional interrelationships.6 However, Ferguson has cautioned that "'x does y rules' may be hard to realize".4 Furthermore, a structure/ activity debate may be premature since complete structural details⁷ are available for only two GPI anchors, VSG3 and the rat brain Thy-1 glycoprotein,8 the latter being of special interest since it was the first to be isolated from a mammalian source. In this communication we describe the first synthesis of the complete glycan moiety of Thy-1, by a route which is convergent, provides the material in multimilligram amounts, and makes provisions for attaching the protein components.

Although the structures of the Thy-1 (1) and VSG anchors (Scheme I) may seem similar biologically, their laboratory syntheses^{9,10} require vastly different strategies. We envisaged a triply convergent approach based on subunits 2 and 3 in the form of n-pentenyl glycosides (NPGs)11 and the pseudodisaccharide

The retrosynthetic plan in Scheme I gives subunits 2, 3, and 4. The trisaccharide 2 presented an opportunity to explore the advantages of the pentenyl activating group for reducing the numbers of "starting materials" required for oligosaccharide syntheses. As described in a recent report from this laboratory, the ability to readily protect and deprotect pentenyl double bonds enabled the preparation of a nonamannan from only two starting materials.¹² In this communication we focus attention on the divergent pathways followed by pentenyl ortho esters upon

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Scheme I

In VSG, M4 is missing and M1 has a tetragalactan at O3 instead of O4 GalNAc residue.

treatment with different electrophiles (Scheme II). For example, reaction of 5a11 with a protic acid induces transfer of the pentenyloxy group to the anomeric center of 10a.13 However, when titrated with bromine, pentenyl ortho esters yield glycosyl bromides such as 9.11 These reaction pathways can be rationalized by invoking the protonated intermediate 7 and the furanylium ion 6, both of which proceed to the dioxolenium ion 8, which then reacts with the best nucleophile available to give 9 or 10. Coupling of the former is best carried out in situ after discharging excess bromine with a stream of argon. In this way, 9b and 10b furnished disaccharide 11a in 89% yield and, in turn, acceptor 11b reacted with 9a to give trisaccharide 12a. The protecting groups of the latter were then changed to those in 12b to provide for future phosphorylations.

With respect to the disaccharide segment 3 (Scheme I), the galactosyl donor is disarmed by virtue of the phthalimide residue, and hence the armed/disarmed strategy employing two NPGs11 would be problematic. Hence a different glycosyl donor, the trichloroacetimidate,14 was chosen since it had been tested in a comparable coupling. 15 Accordingly, compound 1316 was coupled with 14, obtained from n-pentenyl mannopyranoside11 by standard transformations, 17 to give disaccharide 15a (Scheme III). Cleavage of the phthalimide¹⁸ and selective acylation then led to 15b.

Synthetic routes to variously protected forms of the NI pseudodisaccharide 4 (Scheme I) have emerged from this and other 10,19 laboratories, and so there were several options that could be retooled to fit our present needs. A critical factor in our choice was the desire to protect the future site of inositol phosphorylation with a functionality that would be refractory to the conditions of NPG activation¹¹ (vide infra). An allyl group fit this need, and hence the known²⁰ derivative 17 was chosen for coupling

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Scheme IIa

(i) Br₂, 0 °C; (ii) TsOH, CH₂Cl₂, 60 °C; (iii) NaOMe; (iv) AgOTf, 4 Å, CH₂Cl₂, -30 °C, 30 min; (v) NaOMe, then BnBr, then Bu₄NF/THF, then (ClCH2CO)2O.

Scheme IIIa

(i) TMSOTf, 4 Å, toluene, -20 °C, 2 h; (ii) MeNH₂, EtOH, 70 °C, 35 h, then Ac₂O, then (ClCH₂CO)₂O (1.1 equiv), Et₃N, then Ac₂O; (iii) AgClO₄, Et₂O₅, -35 °C → rt, 30 min, then NaOMe; (iv) Ac₂O (1.2 equiv), then DHP, PPTS, CH₂Cl₂, 40 °C, 15 h, then NaOMe, then BnBr, then PPTS/MeOH, CH₂Cl₂, 60 °C, 2 h; (v) NIS, Et₃SiOTf, CH₂Cl₂, rt, 15 min; (vi) thiourea. * represents sites for future couplings.

with the glycosyl bromide 16,21 the product 18a being obtained in 63% yield.

Assembly of the heptasaccharide was now undertaken. Coupling of acceptor 18b and donor 15b in 1.2:1 ratio²² was complete within 15 min to give tetrasaccharide 19a in 55% yield based on 15b, substantial amounts of acceptor 18b also being recovered.²³ Dechloroacetylation then readied 19b for coupling to trisaccharide12b. For this task, compounds 12b (250 mg) and

19b (200 mg) were used in 1.5:1 ratio, affording 115 mg of heptasaccharide 20 (39% yield based on recovered acceptor 19b).

The sites for future couplings in 20 have been provided with temporary protecting groups that can be removed selectively. These tasks are now being carried out as a prelude to biological testing.

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Supplementary Material Available: Listings of experimental procedures for the preparation of all key compounds and their NMR data (16 pages). Ordering information is given on any current masthead page.

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that the material was highly prone to silylation by the NIS/Et₃SiOTf promoter. (23) The recovered silylated glycosyl acceptor was treated with tetra-n-butylammonium fluoride in THF to regenerate the corresponding alcohol.