An Efficient Synthesis of Aminocyclopentitols via the Stereoselective Amination of Polybenzyl Ethers Using Chlorosulfonyl Isocyanate

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Abstract: An efficient stereoselective route for the preparation of stereoisomeric aminocyclopentitols **3–5** was achieved using a six-step sequence starting from the corresponding D-sugars. Key steps of the synthesis of the title compounds involved the regioselective and diastereoselective amination of cinnamylic 1,2-*anti*- or 1,2-*syn*-tribenzyl ethers using chlorosulfonyl isocyanate (CSI), intramolecular olefin metathesis, and diastereoselective dihydroxylation.

Key words: aminocyclopentitol, chlorosulfonyl isocyanate, amination, mannostatin, dihydroxylation

Glycosidases are involved in a wide range of important biological processes, such as intestinal digestion, the posttranslational processing of glycoproteins, and the lysosomal catabolism of glycoconjugates.¹ Inhibitors of these enzymes have received much attention and are currently being used for treating diabetes,² viral infections (e.g., HIV and influenza),³ malaria,⁴ and cancer.⁵ The majority of these inhibitors are iminosugars, in which an oxygen atom in a monosaccharide is replaced by a nitrogen atom, and they have been often found in natural plants and microorganisms. While these iminosugars and their derivatives have been extensively studied, the development of aminocyclopentitol analogues as glycosidase inhibitors has received little attention.

The aminocyclopentitols have been found in a number of natural products, e.g. mannostatin A (1)⁶ and trehazoline (2).⁷ In particular, mannostatin A (1), isolated from the soil microorganism *Streptoverticillus* is the most potent and specific class II α -mannosidase inhibitor reported to date,⁸ which stimulated us to develop an efficient synthetic route for the manufacture of various aminocyclopentitol analogues as new α -mannosidase inhibitors (Figure 1).

Recently, we described a new method for the regioselective and diastereoselective introduction of an N-protected



Figure 1 Mannostatin A (1) and trehazoline (2)

SYNLETT 2007, No. 11, pp 1711–1714 Advanced online publication: 25.06.2007 DOI: 10.1055/s-2007-984518; Art ID: U02907ST © Georg Thieme Verlag Stuttgart · New York amine group into various allylic ethers using CSI⁹ and demonstrated the application of this methodology to the total synthesis of polyhydroxylated alkaloids.¹⁰ As a part of our ongoing investigation of iminosugars, we considered preparing aminocyclopentitol precursors with a 3-amino-1,2-diol functional group assembly.

We report herein a convenient synthetic route for the preparation of stereoisomeric 5-amino-1,2,3,4-cyclopentanetetrols **3–5** via the regioselective and stereoselective amination of polybenzyl ethers using CSI (Figure 2). In addition, we describe interesting results for the diastereoselective dihydroxylation of olefin in polyfunctionalized carbocycles **10a–c**.



Figure 2 Structure of various aminocyclopentitols

In initial studies, we investigated the regioselectivity and diastereoselectivity of the reactions of stereoisomeric tribenzyl ethers **6a–c** with CSI, which were prepared from commercially available D-sugars (D-lyxose, D-ribose and D-arabinose), to provide the corresponding products **7a–c** (Table 1).

As shown in entries 1 and 2, treatment of 1,2-*anti*-tribenzyl ethers **6a** and **6b** with CSI in toluene at 0 °C furnished the corresponding 1,2-*anti*-amino alcohols **7a** and **7b** with excellent diastereoselectivities of 24:1 and 18:1, respectively, in favor of the 1,2-*anti*-isomer. However, the reaction of 1,2-*syn*-tribenzyl ethers **6c** with CSI afforded the desired 1,2-*syn*-amino alcohol **7c** as a major product with moderate diastereoselectivity of 1:5 in a 72% combined yield (entry 3). Table 1 shows that the stereochemistries of major products **7a–c** were identical to those of the corresponding substrates **6a–c**. These results can be interpreted by the neighboring group effect leading to the retention of the original configuration via a double configuration inversion.^{10,11}

The carbocycles 10a-c were straightforwardly prepared in high overall yields (27–33%) from the benzylprotected lactols **8a–c** using a five-step sequence, as shown in Scheme 1. The Wittig olefination of **8a–c** using the DMSO anion¹² and benzyltriphenylphosphonium

Table 1 Diastereoselective CSI Reactions of Various Cinnamylic Polybenzyl Ethers 7a-c



^a Isolated yield of pure materials.

^b Isomeric ratio as determined by ¹H NMR analysis.

chloride in THF at 45 °C afforded the alcohols **9a–c** in high yields. Oxidation of the alcohols **9a–c** with Dess– Martin reagent yielded the corresponding aldehydes, which were then subjected to the Wittig reaction using NaHMDS and methyltriphenylphosphonium bromide, to furnish the dienic compounds **6a–c**. Reactions between the polybenzyl ethers **6a–c** and CSI were carried out in toluene at 0 °C. These were followed by desulfonylation using aqueous 25% sodium sulfite solution, to furnish the cinnamylic amine products **7a–c** with high diastereoselectivity.¹³ Finally, treatment of **7a–c** with second-generation Grubbs' catalyst [(IMesH₂)(PCy₃)(Cl)₂Ru=CHPh]¹⁴ in toluene at 80 °C provided the carbocyclic compounds **10a–c** in good yields.¹⁵

Next, we investigated the dihydroxylation of the stereoisomeric cyclic allylic amines 10a-c using a catalytic amount of OsO4 in 80% acetone (Scheme 2).16 Ratios of stereoisomers were determined by HPLC analysis of the inseparable reaction mixture, and the relative configurations of 11a-13a and 11b were assigned by NOE studies of the isolated pure materials obtained by additional purification processes, involving introduction and removal of the acetonide group.¹⁷ In all cases, dihydroxylation occurred preferentially from the anti-direction with respect to the allylic benzyloxy group at C-4 to give 11a-13a as major products, although the level of diastereoselectivity depended upon the ring substitution pattern. Interestingly, osmylation of **10b** with its bottom face completely shielded by three substituents did not produce a higher level of diastereoselectivity than that observed for 10c. The excellent diastereoselectivity (21:1) of **10c** could be explained by steric interaction between the bulky benzyloxy groups at C-4/C-5 and the oxidant (OsO_4), the directing ability of the NHCO group¹⁸ (as reasonably good hydrogen donor), and allylic strain¹⁹ in the five-membered ring system. Finally, hydrogenolysis of the diols 11a-13a and 11b with



Scheme 1 Reagents and conditions: (a) NaH, DMSO, BnPPh₃Cl, THF, 45 °C; (b) (i) Dess–Martin periodinane, CH_2Cl_2 , 0 °C; (ii) NaH-MDS, MePPh₃Br, THF, 0 °C; (c) (i) CSI, Na₂CO₃, toluene, 0 °C; (ii) 25% Na₂SO₃, r.t.; (d) (IMesH₂)(PCy₃)(Cl)₂Ru=CHPh (5 mol%), toluene, 80 °C.

10% Pd/C and 6 N HCl in methanol followed by purification using ion-exchange resin provided the free amines 3-5 as viscous liquids.²⁰

In conclusion, an efficient stereoselective route for the preparation of stereoisomeric aminocyclopentitols 3-5 was accomplished from the readily available D-sugars via the regio- and diastereoselective amination of polybenzyl ethers using chlorosulfonyl isocyanate, intramolecular olefin metathesis leading to carbocyclic ring skeleton



Scheme 2 Reagents and conditions: (a) cat. OsO_4 , 50% NMO, 80% acetone, r.t.; (b) (i) 2,2-dimethoxypropane, PPTS, CH_2Cl_2 , r.t.; (ii) separation; (iii) *p*-TsOH, MeOH, 40 °C; (iv) 10% Pd/C, H₂, 6 N HCl, MeOH, r.t.; (v) DOWEX-50WX8 (H⁺ form, 0.5 M NH₄OH as eluent).

formation, and diastereoselective dihydroxylation. Moreover, we have provided useful information for the dihydroxylation of olefins in 1-amino-4,5-diol cyclopentenes. Currently, we are preparing other aminocyclitols and aminocyclopentitols for evaluation as potent α -mannosidase inhibitors.

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- (13) Characterization data for 7a-c: **7a**: (*Z*:*E* = 2.5:1, *anti:syn* = 24:1); $[\alpha]_D^{30}$ -14.9 (*c* = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃): $\delta = 3.50-3.53$ (t, J =4.5 Hz, 0.71 H), 3.60-3.61 (br, 0.29 H), 3.91-4.01 (m, 1 H), 4.32 (d, J = 11.5 Hz, 0.29 H), 4.33 (d, J = 12.0 Hz, 0.71 H), 4.60–4.77 (m, 3.29 H), 5.01–5.15 (m, 4.13 H), 5.29 (d, J = 17.0 Hz, 0.29 H), 5.38 (d, J = 10.0 Hz, 0.29 H), 5.53 (t, J = 10.0, 12.0 Hz, 0.71 H), 5.70-5.84 (m, 1.71 H), 5.90-5.99 (m, 0.29 H),6.00 (dd, J = 7.5, 15.5 Hz, 0.29 H), 6.52 (d, J = 15.5 Hz, 0.29 H), 6.53 (d, J = 15.5 Hz, 0.71 H), 7.23–7.36 (m, 20 H). ¹³C NMR (125 MHz, CDCl₃): δ = 49.69, 66.67, 70.64, 73.65, 81.66, 81.94, 119.76, 120.06, 126.80, 127.51, 127.94, 127.99, 128.09, 128.17, 128.40, 128.55, 128.66, 128.69, 132.55, 132.85, 134.88, 135.12, 136.58, 137.05, 138.14, 138.33, 155.99. HRMS (FAB): m/z [M + H⁺] calcd for C₃₅H₃₆NO₄: 534.2644; found: 534.2639. **7b**: (*Z*:*E* = 2.6:1, *anti*:*syn* = 18:1); $[\alpha]_D^{25}$ -79.6 (*c* = 0.1, CHCl₃). ¹H NMR (500 MHz, CDCl₃): $\delta = 3.68-3.73$ (br, 0.56 H), 3.80 (t, J = 7.0 Hz, 0.72 H), 3.88–3.92 (m, 0.72 H), 4.29 (d, J = 11.5 Hz, 0.28 H), 3.39 (d, J = 11.5 Hz, 0.72 H), 4.47 (d, J = 11.5 Hz, 0.28 H), 4.54 (d, J = 11.5 Hz, 0.72 H),

4.57 (d, *J* = 11.5 Hz, 0.28 H), 4.61 (d, *J* = 11.5 Hz, 0.72 H), 4.63 (d, *J* = 11.5 Hz, 0.28 H), 4.67 (d, *J* = 11.5 Hz, 0.72 H), 4.89 (br, 0.28 H), 5.06–5.16 (m, 2 H), 5.25 (t, J = 8.0 Hz, 0.72 H), 5.35–5.44 (m, 3 H), 5.76 (dd, J = 10.0, 10.5 Hz, 0.72 H), 5.85–5.94 (m, 1 H), 6.17 (dd, *J* = 8.0, 15.5 Hz, 0.28 H), 6.49 (d, J = 15.5 Hz, 0.28 H), 6.63 (d, J = 10.5 Hz, 0.72 H), 7.03–7.46 (m, 20 H). ¹³C NMR (125 MHz, CDCl₃): $\delta =$ 50.32, 66.87, 70.47, 70.53, 74.49, 81.12, 81.34, 83.20, 83.50, 119.70, 120.25, 126.78, 127.53, 127.60, 127.85, 128.01, 128.15, 128.23, 128.37, 128.58, 128.65, 128.68, 128.76, 128.90, 132.74, 132.82, 136.18, 136.58, 138.25, 155.60. HRMS (FAB): m/z [M + H⁺] calcd for C₃₅H₃₆NO₄: 534.2644; found: 534.2640. **7c**: $(Z:E = 2.8:1, anti:syn = 1:5); [a]_D^{25} + 1.2 (c = 1.0, CHCl_3).$ ¹H NMR (500 MHz, CDCl₃): $\delta = 3.56$ (t, J = 7.5 Hz, 0.74 H), 3.71 (dd, J = 3.5, 6.5 Hz, 0.26 H), 3.88–3.93 (m, 1 H), 4.30 (d, J = 11.5 Hz, 0.26 H), 4.45–4.64 (m, 3.46 H), 4.76 (d, J = 11.5 Hz, 0.26 H), 4.91-4.93 (br, 0.26 H), 5.09-5.14 (m, 2.74 H), 5.31–5.64 (m, 3 H), 5.84 (dd, *J* = 10.0, 11.0 Hz, 0.74 H), 5.84–5.87 (m, 1 H), 6.18 (dd, J = 8.0, 16.0 Hz, 0.26 H), 6.53 (d, J = 11.0 Hz, 0.74 H), 6.56 (d, J = 16.0 Hz, 0.26 H), 7.06– 7.49 (m, 20 H). ¹³C NMR (125 MHz, CDCl₃): δ = 54.78, 66.88, 70.52, 74.50, 74.66, 81.11, 83.47, 120.33, 126.06, 126.81, 127.82, 127.87, 127.97, 128.14, 128.17, 128.27, 128.35, 128.53, 128.67, 128.69, 128.72, 128.79, 132.76, 136.18, 136.86, 137.01, 138.30, 155.80. HRMS (FAB): m/z [M + H⁺] calcd for C₃₅H₃₆NO₄: 534.2644; found: 534.2654.

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- (15) Characterization data for 10a-c: **10a**: $[\alpha]_D^{29}$ –32.9 (*c* = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃): $\delta = 4.09$ (br t, J = 5.5 Hz, 1 H), 4.52–4.61 (m, 5 H), 4.90-4.93 (br, 1 H), 5.12 (s, 2 H), 5.19 (d, J = 7.5 Hz, 1 H), 5.95 (d, J = 6.0 Hz, 1 H), 6.03 (d, J = 5.5 Hz, 1 H), 7.27–7.40 (m, 15 H). ¹³C NMR (125 MHz, CDCl₃): δ = 55.67, 66.99, 71.92, 72.79, 82.56, 86.81, 128.00, 128.09, 128.17, 128.22, 128.34, 128.61, 128.67, 128.72, 128.74, 132.89, 134.65, 136.76, 137.77, 138.37, 156.37. HRMS (FAB): m/z [M + H^+] calcd for $C_{27}H_{28}NO_4$: 430.2018; found: 430.2020. **10b**: $[\alpha]_D^{25}$ -79.6 (*c* = 0.1, CHCl₃). ¹H NMR (500 MHz, CDCl₃): $\delta = 4.02$ (t, J = 5.5 Hz, 1 H), 4.30 (dd, J = 2.5, 5.0 Hz, 1 H), 4.64 (d, J = 12.0 Hz, 1 H), 4.65 (d, J = 12.0 Hz, 1 H), 4.69–4.71 (br, 1 H), 4.72 (d, *J* = 12.0 Hz, 1 H), 4.79 (d, J = 12.0 Hz, 1 H), 5.12 (s, 2 H), 5.29 (d, J = 9.0 Hz, 1 H), 6.05 (dd, J = 2.0, 6.0 Hz, 1 H), 6.09 (dd, J = 3.0, 6.0 Hz, 1 H), 7.27–7.38 (m, 15 H). ¹³C NMR (125 MHz, CDCl₃): $\delta =$ 54.65, 66.88, 72.31, 72.58, 78.42, 80.03, 127.86, 127.90, 128.02, 128.05, 128.22, 128.61, 128.69, 133.06, 135.08, 136.90, 138.30, 138.79, 156.36. HRMS (FAB): m/z [M + H⁺] calcd for C₂₇H₂₈NO₄: 430.2018; found: 430.2013. **10c**: $[\alpha]_{D}^{29}$ +6.2 (*c* = 1.0, CHCl₃). ¹H NMR (500 MHz, CDCl₃): $\delta = 4.02$ (t, J = 5.5 Hz, 1 H), 4.30 (dd, J = 2.5, 5.0 Hz, 1 H), 4.64 (d, J = 12.0 Hz, 1 H), 4.65 (d, J = 12.0 Hz, 1 H), 4.70–4.73 (br m, 2 H), 4.79 (d, J = 12.0 Hz, 1 H), 5.12 (s, 2 H), 5.28 (d, J = 9.0 Hz, 1 H), 6.05 (dd, J = 1.5, 6.0 Hz, 1 H), 6.08 (dd, J = 2.5, 6.5 Hz, 1 H), 7.27–7.41 (m, 15 H). ¹³C NMR (125 MHz, CDCl₃): δ = 54.65, 66.87, 72.31, 72.58, 78.44, 80.03, 127.84, 127.90, 128.01, 128.03, 128.21, 128.60, 128.69, 133.05, 135.08, 136.91, 138.30, 143.67, 156.35. HRMS (FAB): m/z [M + H⁺] calcd for C₂₇H₂₈NO₄: 430.2018; found: 430.2019.
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- (17) Characterization data for 11a-13a and 11b:
 - **11a**: $[\alpha]_D^{23}$ +1.0 (c = 0.22, CHCl₃). ¹H NMR (500 MHz, CDCl₃): δ = 3.92–4.25 (m, 5 H), 4.46 (d, J = 11.0 Hz, 1 H), 4.53 (d, J = 12.0 Hz, 1 H), 4.57 (d, J = 11.0 Hz, 1 H), 4.65 (d, J = 12.0 Hz, 1 H), 5.11 (d, J = 12.5 Hz, 1 H), 5.16 (d, J = 12.5 Hz, 1 H), 5.54–5.59 (br, 1 H), 7.22–7.36 (m, 15 H). ¹³C NMR (125 MHz, CDCl₃): δ = 53.83, 67.33, 72.06, 72.71, 73.41, 80.22, 82.63, 88.40, 128.04, 128.09, 128.24, 128.39, 128.45, 128.71, 128.78, 128.81, 136.52, 137.20, 137.92, 154.56. HRMS (FAB): m/z [M + H⁺] calcd for C₂₇H₃₀NO₆: 464.2073; found: 464.2068.

11b: $[\alpha]_D^{22}$ –18.2 (*c* = 0.08, CHCl₃). ¹H NMR (500 MHz, CDCl₃): δ = 1.58–1.64 (br, 1 H), 2.84–2.88 (br, 1 H), 3.86 (dd, J = 4.0, 4.5 Hz, 1 H), 3.93 (t, J = 6.5 Hz, 1 H), 4.01–4.21 (m, 3 H), 4.78 (d, J = 11.5 Hz, 1 H), 4.52–4.60 (m, 2 H), 4.71 (d, J = 11.5 Hz, 1 H), 5.12 (s, 2 H), 5.52 (br s, 1 H), 7.24– 7.48 (m, 15 H). ¹³C NMR (125 MHz, CDCl₃): δ = 57.16, 67.41, 69.13, 72.46, 73.06, 77.49, 80.21, 82.78, 128.22, 128.28, 128.33, 128.42, 128.49, 128.79, 136.41, 137.42, 137.57, 157.66. HRMS (FAB): *m*/*z* [M + Na⁺] calcd for C27H29NO6Na: 486.1893; found: 486.1896. **12a**: $[\alpha]_D^{23}$ +7.3 (*c* = 0.44, CHCl₃). ¹H NMR (500 MHz, $CDCl_3$): $\delta = 1.82$ (br s, 1 H), 3.09 (br s, 1 H), 3.91–3.99 (m, 3 H), 4.17 (dd, J = 4.5, 5.5 Hz, 1 H), 4.23 (br d, J = 3.0 Hz, 1 H), 4.52 (d, *J* = 12.0 Hz, 1 H), 4.68 (d, *J* = 12.0 Hz, 1 H), 4.76 (d, J = 12.0 Hz, 1 H), 4.79 (d, J = 12.0 Hz, 1 H), 5.14 (d, J = 12.5 Hz, 1 H), 5.10 (d, J = 12.5 Hz, 1 H), 5.62 (d, J =

6.0 Hz, 1 H), 7.27–7.40 (m, 15 H). ¹³C NMR (125 MHz, CDCl₃): δ = 58.34, 67.35, 72.63, 73.27, 73.59, 76.42, 77.58, 84.59, 127.94, 128.01, 128.21, 128.37, 128.50, 128.69, 128.76, 128.82, 136.45, 138.04, 138.29, 157.59. HRMS (FAB): *m*/*z* [M + H⁺] calcd for C₂₇H₃₀NO₆: 464.2073; found: 464.2083.

13a: $[a]_D^{29}$ –2.5 (c = 1.0, CHCl₃). ¹H NMR (300 MHz, CDCl₃): δ = 2.98–3.22 (br, 2 H), 3.90–3.98 (br m, 2 H), 4.21–4.26 (br m, 3 H), 4.51 (d, J = 11.7 Hz, 1 H), 4.71 (d, J = 12.0 Hz, 1 H), 4.79 (d, J = 12.0 Hz, 1 H), 4.82 (d, J = 11.7 Hz, 1 H), 5.17 (s, 2 H), 5.63 (br d, J = 5.7 Hz, 1 H), 7.23–7.65 (m, 15 H). ¹³C NMR (125 MHz, CDCl₃): δ = 58.41, 67.34, 72.66, 73.31, 73.61, 76.46, 77.52, 84.55, 127.91, 127.98, 128.17, 128.34, 128.47, 128.66, 128.74, 128.80, 136.44, 138.00, 138.29, 157.69. HRMS (FAB): m/z [M + H⁺] calcd for C₂₇H₃₀NO₆: 464.2073; found: 464.2072.

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- (20) Characterization data for **3**–**5**: **3**: $[\alpha]_D^{23}$ +6.3 (c = 0.79, MeOH). ¹H NMR (500 MHz, D₂O): $\delta = 3.22$ (t, J = 3.0 Hz, 1 H), 3.73 (dd, J = 4.0, 7.5 Hz, 1 H), 3.80 (dd, J = 3.0, 7.5 Hz, 1 H), 3.91–3.94 (m, 2 H). ¹³C NMR (125 MHz, D₂O): $\delta = 51.91$, 73.32, 75.91, 76.01, 83.35. HRMS (FAB): m/z [M + H⁺] calcd for C₅H₁₂NO₄: 150.0766; found: 150.0766. **4**: $[\alpha]_D^{23}$ –11.9 (c = 0.23, MeOH). ¹H NMR (500 MHz, D₂O): $\delta = 3.35$ (t, J = 8.0 Hz, 1 H), 3.86–3.92 (m, 2 H), 3.99 (t, J = 4.5 Hz, 1 H), 4.07 (dd, J = 6.0, 8.0 Hz, 1 H). ¹³C NMR (125 MHz, D₂O): $\delta = 56.37$, 71.18, 73.37, 74.80, 76.84. HRMS (FAB): m/z [M + H⁺] calcd for C₅H₁₂NO₄: 150.0766; found:

150.0767. **5**: $[α]_D^{22}$ -2.3 (*c* = 1.0, MeOH). ¹H NMR (500 MHz, D₂O): δ = 3.46 (t, *J* = 6.0 Hz, 1 H), 3.96 (t, *J* = 5.0 Hz, 1 H), 4.00 (dd, *J* = 6.0, 6.5 Hz, 1 H), 4.19 (t, *J* = 7.0 Hz, 1 H), 4.22 (dd, *J* = 4.5, 5.5 Hz, 1 H). ¹³C NMR (125 MHz, D₂O): δ = 57.19, 67.83, 71.83, 73.85, 76.39. HRMS (FAB): *m*/*z* [M + H⁺] calcd for C₅H₁₂NO₄: 150.0766; found: 150.0767. Copyright of Synlett is the property of Georg Thieme Verlag Stuttgart and its content may not be copied or emailed to multiple sites or posted to a listserv without the copyright holder's express written permission. However, users may print, download, or email articles for individual use.