

LIGNANOLIDES FROM *BUPLEURUM SALICIFOLIUM*

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(Received in revised form 2 December 1991)

Key Word Index—*Bupleurum salicifolium*; Umbelliferae; roots; lignans; $^1\text{H NMR}$; $^{13}\text{C NMR}$.

Abstract—Three new lignans isolated from the roots of *Bupleurum salicifolium* were characterised as guamarol, isoguamarol and guamarolin on the basis of chemical and spectral evidence. The known lignans, kaerophyllin, isokaerophyllin and matairesinol, were also obtained.

INTRODUCTION

Bupleurum salicifolium, a member of the Umbelliferae and endemic to the Canary Islands, is rich in lignans. Seven new lignans from this plant, (one from the branches [1], four from the leaves [2, 3] and two from the seeds [4]), have already been described and now three more new lignans have been obtained from the roots. Two belong to the benzylidenebenzyl- γ -butyrolactone group, viz guamarol (1) [*E* 2(3'-hydroxy-4'-methoxybenzylidene)-3(3'',4''-methylenedioxybenzyl)-3*R*- γ -butyrolactone] and isoguamarol (2) [*Z* 2(3'-hydroxy-4'-methoxybenzylidene)-3(3'',4''-methylenedioxybenzyl)-3*R*- γ -butyrolactone] while the third, guamarolin (3) [2*R*(3'-hydroxy-4'-methoxybenzyl)-3*R*-(3'',4''-methylenedioxybenzyl)- γ -butyrolactone] is a dibenzyl- γ -butyrolactone. The structures and stereochemistry of all these compounds were determined from their spectral data and chemical transformations.

The most distinctive structural feature of 1–3 is the presence of a 3-hydroxy-4-methoxyphenyl group, which is very unusual in naturally occurring lignans. To date it has only been found in prestegane A and prestegane B, dibenzyl- γ -butyrolactones from *Steganotaenia araliacea* [5], and provides yet another instance of the intriguing properties of *B. salicifolium*.

The known lignans, kaerophyllin (1b) [*E* 2(3',4'-dimethoxybenzylidene)-3(3'',4''-methylenedioxyphenylmethyl)-3*R*- γ -butyrolactone], isokaerophyllin (2b) [*Z*-2(3',4'-dimethoxybenzylidene)-3(3'',4''-methylenedioxyphenylmethyl)-3*R*- γ -butyrolactone] and matairesinol [2(3'-methoxy-4'-hydroxyphenylmethyl)-3(3''-methoxy-4''-hydroxyphenylmethyl)-2*R*,3*R*- γ -butyrolactone], were also obtained. *B. salicifolium* exhibits an interesting specificity for elaborating lignans, which are substances that have been shown to inhibit topo-isomerase II, an enzyme associated with the replication of the HIV-1 virus responsible for AIDS [6].

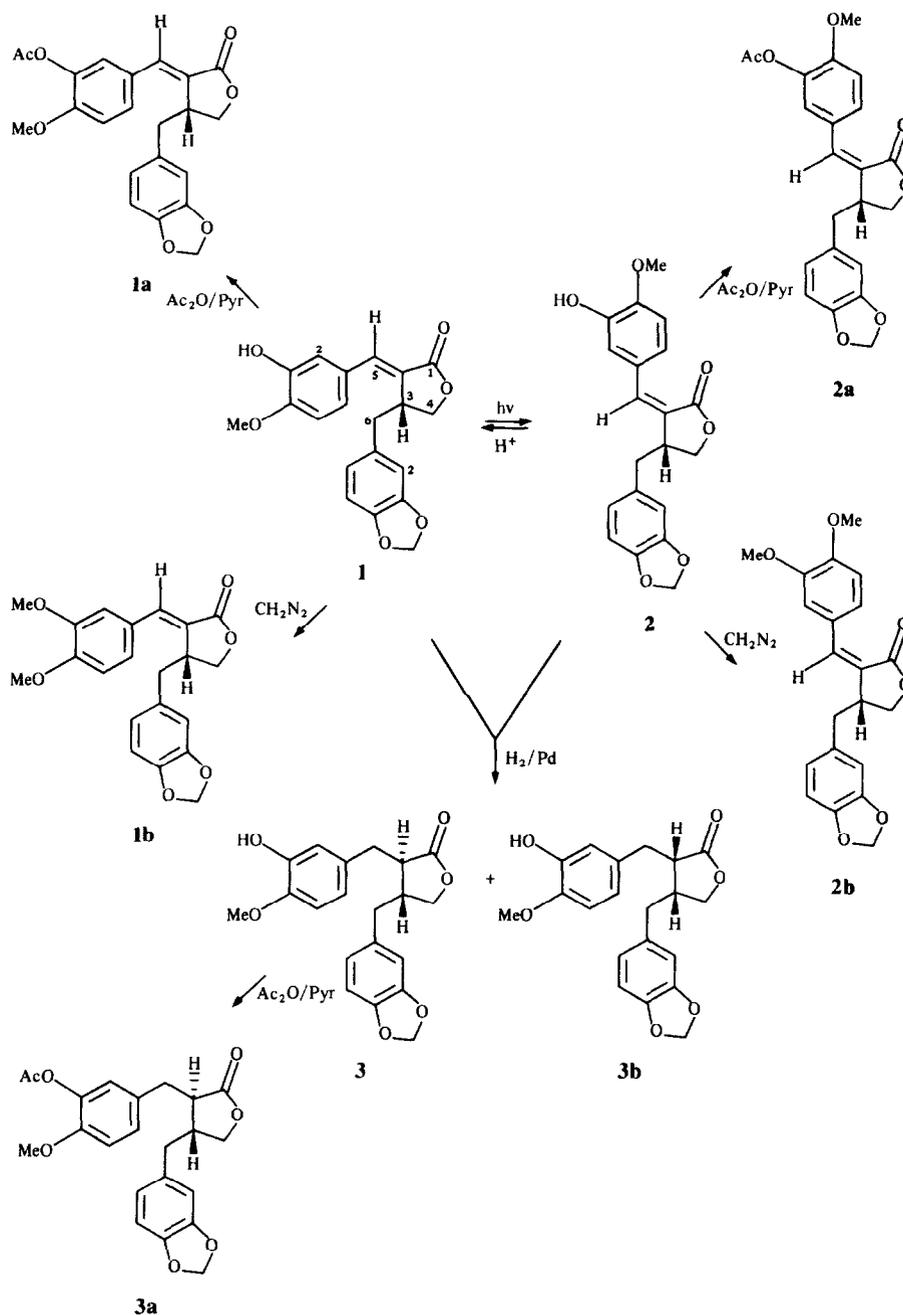
RESULTS AND DISCUSSION

Six lignans were separated from an ethanol extract of the roots of *B. salicifolium* and treated and chromatographed as described in the Experimental. Three of the lignans were new to the literature. One, guamarol (1), had

a $[\text{M}]^+$ m/z 354, molecular formula $\text{C}_{20}\text{H}_{18}\text{O}_6$, ν_{max} 1740 cm^{-1} (C=O) and 3420 cm^{-1} (OH); its mass fragmentation pattern showed peaks for two aromatic groups, a piperonylmethyl group on C-3 (m/z 135, $[\text{C}_6\text{H}_3(\text{OCH}_2\text{O})\text{CH}_2]^+$), (m/z 162, $[\text{C}_6\text{H}_3(\text{OCH}_2\text{O})\text{CH}_2\text{C}_2\text{H}_3]^+$) and an isoguayacyl group on the olefinic C-5 (m/z 123, 9%, $[\text{C}_6\text{H}_3(\text{OH})(\text{OMe})]^+$). The $^1\text{H NMR}$ spectrum of guamarol (Table 1) had the characteristic 2-benzylidene-3-benzyl- γ -butyrolactone-type *E* isomer signals [7], notably a multiplet centred at δ 3.79 corresponding to H-3 and two doublets at δ 4.25 (2H, H-4) and δ 7.50 (1H) ascribed to the olefinic H-5. The $^1\text{H NMR}$ also had signals for six benzene protons, three of which were those of a piperonylmethyl group. The signals of the other three benzene protons must be those of a 3-hydroxy-4-methoxyphenyl group because they were not consonant with those of the 3-methoxy-4-hydroxyphenyl (guayacyl) group common to many lignans of the benzylidene benzyl- γ -butyrolactone type such as 3''-deoxy- γ -thujaplicatin-4-*O*-methyl ether [δ 7.04 *d* (H-2'); 6.99 *d* (H-5'); 7.21 *dd* (H-6')] [8]. The equivalent values in guamarol (see Table 1) differed by 34 Hz for H-2' and 20 Hz for H-6', which, when taken in conjunction with the results of COSY experiments, sited the OH at C-3' in the benzene ring. ROESY experiments [9] revealed the couplings shown in Fig. 1, which unequivocally confirmed the relative configuration proposed for compound 1. The $^{13}\text{C NMR}$ spectrum of guamarol (Table 2) is very similar to that of kaerophyllin (1b) [10] but with a δ 6 downfield shift for the C-2' signals and a δ 4.8 upfield shift for those of C-3'.

The acetyl derivative, 1a, was also formed from guamarol: ($[\text{M}]^+$ m/z 398, $\text{C}_{22}\text{H}_{22}\text{O}_7$, m/z 135 (base peak) and m/z 165, 4% $[\text{C}_6\text{H}_3(\text{OMe})(\text{OCOMe})]^+$). Its $^1\text{H NMR}$ spectrum (Table 1) featured signals for H-6' 62 Hz downfield from those of H-6' in guamarol confirming that the OH group was at C-3'. Had it been at C-4', the shift of the H-6' signals would be very slight and less than that of H-2' and H-5', as can be seen from a comparison of the literature data for this group with a *para* hydroxyl group [8].

The negative optical activity of guamarol indicated a *b* orientation of the H-3 hydrogen, which was confirmed by hydrogenation of the prochiral double bond of guamarol



with H_2 and Pd/C when more of the diastereoisomer **3b** (75%) than of the diastereoisomer **3** (25%) was formed due to the steric hindrance of the piperonylmethyl group. The transformation of guaramarol to kaerophyllin (**1b**) when treated with diazomethane confirmed its structure as **1**.

The other benzylidene benzyl- γ -butyrolactone isolated from the roots of *B. salicifolium*, isoguamarol (**2**), showed a $[\text{M}]^+$ m/z 354, formula $\text{C}_{20}\text{H}_{18}\text{O}_6$, negative optical activity and spectroscopic data corresponding to the structure of the *Z*-isomer of guaramarol. The mass spectrum of isoguamarol had the same fragments as that of

guamarol, while its ^1H NMR spectrum displayed the signals of a *Z*-isomer, *viz.* a multiplet centred at δ 3.26 due to H-3, two double doublets centred at δ 4.09 (H-4a) and δ 4.32 (H-4b), and a singlet at δ 6.65 attributed to the olefinic H-5. In the ^{13}C NMR spectra, the signals of the C-2 are shifted δ 5.1 upfield, those of C-5, δ 3.1 downfield and those of C-6, δ 3.3 downfield, compared to guaramarol (Table 2). Isoguamarol also formed an acetate, **2a** ($[\text{M}]^+$ m/z 398, $\text{C}_{22}\text{H}_{22}\text{O}_7$), with signals in its ^1H NMR spectrum for a 42 Hz shift of H-6' in relation to those of the H-6' in isoguamarol.

Table 1. ^1H NMR data of compounds **1**, **1a**, **2**, **2a**, **3** and **3a** (200 MHz, CDCl_3)*

H	1	1a	2	2a	3	3a
2	—	—	—	—	2.30–2.60 <i>m</i>	2.40–2.60 <i>m</i>
3	3.79 <i>m</i>	3.72 <i>m</i>	3.26 <i>m</i>	3.32 <i>m</i>	2.30–2.60 <i>m</i>	2.40–2.60 <i>m</i>
4a			4.09 <i>dd</i> (3.8, 9.0)	4.09 <i>dd</i> (3.8, 9.1)	3.94 <i>dd</i> (2.9, 6.5)	3.83 <i>dd</i> (7.0, 10.2)
	4.25 <i>t</i> (2.6, 6.1)	4.26 <i>m</i>				
4b			4.32 <i>dd</i> (7.3, 9.0)	4.32 <i>dd</i> (6.9, 9.1)	4.10 <i>dd</i> (6.5, 9.2)	4.10 <i>dd</i> (7.0, 9.2)
5a					2.01 <i>dd</i> (5.2, 14.0)	
	7.50 <i>d</i> (1.9)	7.51 <i>d</i> (1.8)	6.65 <i>s</i>	6.60 <i>br s</i>		2.93 <i>d</i> (5.7)
5b					2.82 <i>dd</i> (6.9, 14.0)	
6a	2.61 <i>dd</i> (10.3, 14.2)	2.59 <i>dd</i> (10.1, 14.2)	2.72 <i>dd</i> (8.9, 13.7)	2.76 <i>dd</i> (8.9, 13.2)		
					2.30–2.60 <i>m</i>	2.40–2.60 <i>m</i>
6b	3.03 <i>dd</i> (4.2, 14.2)	2.99 <i>dd</i> (4.1, 14.2)	2.92 <i>dd</i> (6.9, 13.7)	2.92 <i>dd</i> (6.3, 13.2)		
2'	7.21 <i>d</i> (2.1)	7.28 <i>d</i> (2.2)	7.44 <i>d</i> (2.0)	7.78 <i>d</i> (2.0)	6.73 <i>d</i> (2.0)	6.79 <i>d</i> (2.0)
5'	6.91 <i>d</i> (8.4)	7.03 <i>d</i> (8.6)	6.81 <i>d</i> (8.4)	6.94 <i>d</i> (8.6)	6.70 <i>d</i> (8.2)	6.71 <i>d</i> (8.2)
6'	7.11 <i>dd</i> (2.1, 8.4)	7.42 <i>dd</i> (2.2, 8.6)	7.15 <i>dd</i> (2.1, 8.4)	7.70 <i>dd</i> (2.0, 8.6)	6.66 <i>dd</i> (2.0, 8.2)	6.99 <i>dd</i> (2.1, 8.2)
2''	6.71 <i>d</i> (1.5)	6.77 <i>d</i> (1.6)	6.59 <i>d</i> (1.7)	6.68 <i>br s</i> (1.7)	6.46 <i>d</i> (1.7)	6.48 <i>br s</i>
5''	6.75 <i>d</i> (7.8)	6.73 <i>d</i> (7.8)	6.76 <i>d</i> (7.8)	6.76 <i>d</i> (7.9)	6.78 <i>d</i> (8.0)	6.88 <i>d</i> (8.3)
6''	6.66 <i>dd</i> (1.6, 7.8)	6.63 <i>dd</i> (1.7, 7.8)	6.62 <i>dd</i> (1.7, 7.8)	6.62 <i>dd</i> (1.8, 7.9)	6.46 <i>dd</i> (1.8, 8.0)	6.48 <i>dd</i> (2.0, 8.3)
OMe	3.95 <i>s</i>	3.89 <i>s</i>	3.92 <i>s</i>	3.86 <i>s</i>	3.88 <i>s</i>	3.81 <i>s</i>
OCH ₂ O	5.94 <i>s</i>	5.94 <i>s</i>	5.96 <i>s</i>	5.96 <i>s</i>	5.93 <i>s</i>	5.94 <i>s</i>
OH	5.78 <i>s</i>	—	5.98 <i>s</i>	—	6.60 <i>s</i>	—
OAc	—	2.36 <i>s</i>	—	2.32 <i>s</i>	—	2.31 <i>s</i>

*Values in δ ; coupling constants (Hz) in parentheses.

The structure of the new lignan, isoguamarol, was confirmed by the following reactions. It could be transformed to isokaerophyllin (**2b**) by reaction with diazomethane; it was isomerised to guaramarol (**1**) by UV irradiation reaching a photostationary equilibrium; it could be hydrogenated to form the γ -butyrolactones **3** and **3b**, and it was obtained by isomerisation of **1** in an acid medium.

The absolute configuration of the only chiral centre, C-3, in guaramarol and isoguamarol was settled beyond question because treatment with CH_2N_2 converted these compounds into kaerophyllin and isokaerophyllin, respectively. The latter are lignans with a known absolute configuration [2, 11, 12].* The physical and spectral data of these methylation products proved to be totally compatible with those of natural kaerophyllin and isokaerophyllin.

The third new lignan, guaramarolin (**3**), was an oil with a $[\text{M}]^+$ m/z 356, $\text{C}_{20}\text{H}_{20}\text{O}_6$, ν_{max} 1760 cm^{-1} (C=O),

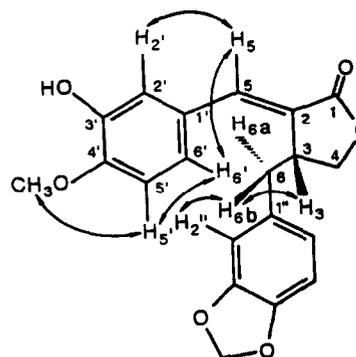


Fig. 1. NOEs observed in compound **1**.

3440 cm^{-1} (OH) and negative optical activity. Its mass spectrum suggested the presence of a hydroxymethoxybenzyl group bonded with C-2 (m/z 137, base peak) and a piperonylmethyl group (m/z 135, 67%) attached to C-3, which was confirmed by the presence of the fragment at

*A typographical error in ref. [2] stated isokaerophyllin to be 3S and not 3R.

Table 2. ^{13}C NMR data of compounds **1**, **1a**, **2**, **2a**, **3** and **3a** (50 MHz, CDCl_3)*

C	1	1a	2	2a	3	3a
1	172.83	173.94	172.70	174.24	178.44	178.13
2	131.83	129.60	126.76	126.68	46.44	46.61
3	40.11	40.03	44.35	44.16	41.50	41.30
4	69.60	69.62	69.93	69.71	71.08	71.06
5	137.46	136.27	140.52	139.52	34.88	34.18
6	37.56	37.57	40.87	40.69	38.39	38.51
1'	131.83	131.58	131.70	130.84	130.96	130.40
2'	115.21	112.54	117.30	116.03	115.37	123.97
3'	146.70	147.23	146.91	146.63	145.75	151.70
4'	146.09	146.05	146.70	146.63	145.56	144.12
5'	110.90	109.07	112.84	112.56	110.83	112.97
6'	124.09	123.95	124.33	123.94	121.56	123.97
1''	124.09	131.58	124.13	132.52	131.75	131.73
2''	108.66	108.46	108.55	108.22	108.86	109.04
3''	148.30	149.61	148.13	148.05	147.87	148.09
4''	148.13	148.36	147.80	148.05	146.36	146.24
5''	110.90	112.54	110.22	111.65	108.38	108.46
6''	122.28	122.08	122.44	122.09	120.69	127.43
OMe	56.22	56.06	56.10	56.06	55.98	56.09
OCH ₂ O	101.19	101.05	101.19	101.04	100.98	101.02
OAc	—	171.10	—	170.50	—	171.22
		20.56		20.61		20.54

*Chemical shifts are given in δ .

m/z 162 $[\text{C}_6\text{H}_3(\text{OCH}_2\text{O})\text{CH}_2 + \text{C}_2\text{H}_3]^+$. The ^1H NMR spectrum of **3** had signals for a 2,3-*trans*- γ -butyrolactone (Table 1) and for two aromatic groups, one of which coincided in all respects with piperonylmethyl. The signals of the other group did not match those of the 4-hydroxy-3-methoxybenzyl group present in pluviatolide [13], a γ -butyrolactone found in the seeds of *B. salicifolium* [4], which had signals at δ 6.66 (H-2'), 6.69 (H-5'), 6.62 (H-6') and 3.85 (OMe). The signals of guararolin were 14 Hz downfield for H-2', 8 Hz for H-6' and 6 Hz upfield for the OMe group and thus this group in **3** must be a *meta*-phenolic hydroxyl. This was confirmed by the ^{13}C NMR spectrum (Table 2) where peaks were observed which were similar to those shown by prestegane A and B, the only natural lignans with an isoguaiacyl group [5].

Guamarolin formed an acetate, **3a**, as an oil with $[\text{M}]^+ m/z$ 398, $\text{C}_{22}\text{H}_{22}\text{O}_7$, in the ^1H NMR spectrum of which (Table 1) the H-6' signals appeared 67.4 Hz downfield as had been the case with **1** and **2** and their respective acetates. The stereochemistry of guararolin was established as 2*R*,3*R* since guararolin was formed in 25% yield when guararol and isoguararol were hydrogenated. The physical and spectral data of the hydrogenation product, guararolin, were exactly the same as those of the natural product.

Kaerophyllin (**1b**) [10], isokaerophyllin (**2b**) [2] and matairesinol [14] were also isolated from the roots of *B. salicifolium*.

EXPERIMENTAL

Mps are uncorr. IR spectra were recorded as films, UV spectra in EtOH. ^1H and ^{13}C NMR spectra were run at 200 and 50 MHz, respectively, with TMS as int. standard.

Schleicher-Schüll F-100/LS 254 and prep. TLC 1510/LS 254 foils were used for TLC while silica gel (0.2–0.63 mm) and Sephadex LH-20 were used for CC.

Isolation. Root bark of wild specimens of *B. salicifolium* Soland (0.61 kg) gathered in the Barranco Rio Badajoz de Güimar, Tenerife were extd with EtOH. This ext. was treated successively with H_2O , Me_2CO and *n*-hexane to afford a dark residue (27.5 g) which was chromatographed on a Sephadex column (78.5 cm \times 5 cm) eluting with *n*-hexane- CHCl_3 -MeOH (2:2:1). There frs, A–C, were sep'd and studied. They were all chromatographed on a silica gel column using mixts of *n*-hexane EtOAc of increasing polarity as eluants. Isokaerophyllin (**2b**) (20.8 mg) and kaerophyllin (**1b**) (16.1 mg) were sep'd from fr. A (0.5 g). Fr. B (0.28 g) afforded guararolin (**3**) (28.2 mg) and fr. C (0.24 g) yielded matairesinol (72.3 mg) ($[\text{M}]^+ m/z$ 358, $\text{C}_{20}\text{H}_{22}\text{O}_6$, base peak m/z 137), guararol (**1**) (24.2 mg), and isoguararol (**2**) (14.8 mg).

Guamarol (1). Yellow oil. UV λ_{max} nm: 232, 291 and 331. IR ν_{max} cm^{-1} : 3420 (br), 1740, 1685, 1662, 1640, 1360, 1285, 1250, 1195. $[\alpha]_{\text{D}}^{25}$ $^{\circ}$ $-\text{44}$ (CHCl_3 ; c 0.04). ^1H NMR, see Table 1. ^{13}C NMR, see Table 2. MS m/z 354.1110 (13%, $[\text{M}]^+$, calc. for $\text{C}_{20}\text{H}_{18}\text{O}_6$, 354.1098), 219 (35, $\text{C}_{12}\text{H}_{11}\text{O}_4$), 177 (2, $\text{C}_{10}\text{H}_9\text{O}_3$), 135.0466 (100, calc. for $\text{C}_8\text{H}_7\text{O}_2$, 135.0444), 123 (2, $\text{C}_7\text{H}_7\text{O}_2$), 121 (2, $\text{C}_7\text{H}_5\text{O}_3$).

Guamarol acetate (1a). Guamarol (2 mg) was dissolved in 1 drop of pyridine, Ac_2O (0.2 ml) added and the soln left at room temp. for 24 hr, giving an acetyl derivative, **1a** (2.2 mg), as an oil. UV λ_{max} nm: 228, 294 and 312. IR ν_{max} cm^{-1} : 1748, 1642, 1609, 1442, 1185, 1097. ^1H NMR, see Table 1. ^{13}C NMR, see Table 2. MS m/z 396.1239 (10%, $[\text{M}]^+$, calc. for $\text{C}_{22}\text{H}_{20}\text{O}_7$, 396.1210), 354 (2, $\text{C}_{20}\text{H}_{18}\text{O}_6$), 165.0496 (4, calc. for $\text{C}_9\text{H}_9\text{O}_3$, 165.0479), 135 (36, $\text{C}_8\text{H}_7\text{O}_2$).

Methylation. Guamarol (2 mg) was dissolved in Et_2O , and CH_2N_2 in Et_2O (10 drops) was added to give kaerophyllin (**1b**)

which was identified by comparison with natural kaerophyllin by TLC, ^1H NMR, ^{13}C NMR, MS and optical activity.

Hydrogenation. Guamarol (10 mg) was dissolved in dry benzene (2.5 ml), Pd/C added and the mixt. stirred under an H_2 atm. at room temp. for 4 hr. The mixt. of products obtained was resolved by GCC using benzene-EtOAc (4:1). The hydrogenation product with the greatest R_f (25%) was identified as guamarolin (3) and that of least R_f (75%) as *cis*-2-(3-hydroxy-4-methoxyphenyl)-3(2,3-dioxymethylenephenyl)- γ -butyrolactone (3b): $[\text{M}]^+$ m/z 356, base peak m/z 137, $\text{C}_{20}\text{H}_{20}\text{O}_6$; ^1H NMR (CDCl_3) δ : 2.27 (1H, t, $J = 13.3$, H-5a), 2.59 (1H, m, H-3), 2.89 (1H, m, H-5b), 3.06 (1H, m, H-2), 3.25 (1H, m, H-6b), 3.90 (3H, s, OMe), 4.02 (2H, m, H-4), 5.64 (1H, br s, OH), 5.92 (2H, s, OCH_2O), 6.48 (1H, br s, H-2''), 6.51 (1H, m, H-6''), 6.68 (1H, d, $J = 8.1$, H-5'), 6.75 (1H, m, H-6), 6.82 (1H, d, $J = 8.3$, H-5''), 6.85 (1H, d, $J = 2.0$, H-2').

Isomerisation. Guamarol (3 mg) dissolved in Me_2CO (10 ml) was irradiated with UV light (254 nm) for 68 hr and was partially (26%) transformed to isoguamarol (2).

Isoguamarol (2). Yellow oil. UV λ_{max} nm: 236, 288, 334. IR ν_{max} cm^{-1} : 3542, 1732, 1648, 1614, 1284, 1249, 1038. ^1H NMR, see Table 1. ^{13}C NMR, see Table 2. MS m/z 354.1108 ($[\text{M}]^+$, calc. for $\text{C}_{20}\text{H}_{18}\text{O}_6$, 354.1098), 219 (8, $\text{C}_{12}\text{H}_{11}\text{O}_4$), 177 (8, $\text{C}_{10}\text{H}_9\text{O}_3$), 135.0451 (100, calc. for $\text{C}_8\text{H}_7\text{O}_2$, 135.0444), 123 (4, $\text{C}_7\text{H}_7\text{O}_2$), 121 (5, $\text{C}_7\text{H}_5\text{O}_2$).

Isoguamarol acetate (2a). Isoguamarol (2.4 mg), when treated with excess Ac_2O in pyridine at room temp. for 24 hr, formed an acetyl derivative, 2a, (2.6 mg) as an oil. UV λ_{max} nm: 223, 289, 311. IR ν_{max} cm^{-1} : 1743, 1651, 1609, 1491, 1443, 1369, 1277, 1194, 1128, 1092, 1041. $[\alpha]_{\text{D}}^{25} = -10^\circ$ (CHCl_3 ; c 0.02). ^1H NMR, see Table 1. ^{13}C NMR, see Table 2. MS m/z 396.1221 (22%, $[\text{M}]^+$, calc. for $\text{C}_{22}\text{H}_{20}\text{O}_7$, 396.1210), 354 (5, $\text{C}_{20}\text{H}_{18}\text{O}_6$), 165.0599 (4, calc. for $\text{C}_9\text{H}_9\text{O}_3$, 165.0479), 261 (22, $\text{C}_{14}\text{H}_{13}\text{O}_5$), 135.0450 (100, calc. for $\text{C}_8\text{H}_7\text{O}_2$, 135.0444).

Methylation. Isoguamarol (2.8 mg), when treated with excess CH_2N_2 in Et_2O at room temp. formed isokaerophyllin (2b), identified by TLC, UV, IR, ^1H NMR, ^{13}C NMR, MS and optical activity comparison with an authentic sample.

Isomerisation. Isoguamarol (1.9 mg) was dissolved in CHCl_3 (2 ml) and 2 drops of HCl added. The soln was left at room temp. for 72 hr when 2 was transformed partially into 1 (84%).

Hydrogenation. Isoguamarol was hydrogenated under the same conditions as described for 1 to give a mixt. of 3 and 3b in similar proportions to those obtained when guamarol was hydrogenated.

Guamarolin (3). Yellow oil. UV λ_{max} nm: 237, 290. IR ν_{max} cm^{-1} : 3440, 1760, 1590, 1510, 1440, 1350, 1230, 1225. $[\alpha]_{\text{D}}^{25} = -27.4^\circ$ (CHCl_3 ; c 0.06). ^1H NMR, see Table 1. ^{13}C NMR, see

Table 2. MS m/z 356.1278 (20%, $[\text{M}]^+$, calc. for $\text{C}_{20}\text{H}_{20}\text{O}_6$, 356.1254), 162.0687 (13, calc. for $\text{C}_{10}\text{H}_{10}\text{O}_2$, 162.0678), 137.0598 (100, calc. for $\text{C}_8\text{H}_7\text{O}_2$, 137.0600), 135 (52, $\text{C}_8\text{H}_7\text{O}_2$).

Guamarolin acetate (3a). Guamarolin (2.2 mg), when treated with pyridine (1 drop) and excess Ac_2O (10 drops) at room temp. for 24 hr, formed an acetate, 3a, (2 mg). UV λ_{max} nm: 228, 280. IR ν_{max} cm^{-1} : 1766, 1603, 1444, 1369, 1269, 1201, 1125, 1040. $[\alpha]_{\text{D}}^{25} = -96^\circ$ (CHCl_3 ; $c = 0.05$). ^1H NMR, see Table 1. ^{13}C NMR, see Table 2. MS m/z 398.1360 (33%, $[\text{M}]^+$, calc. for $\text{C}_{22}\text{H}_{22}\text{O}_7$, 398.1359), 356 (100, $\text{C}_{20}\text{H}_{20}\text{O}_6$), 179.0702 (6, calc. for $\text{C}_{10}\text{H}_{11}\text{O}_3$, 179.0715), 162.0687 (12, calc. for $\text{C}_{10}\text{H}_{10}\text{O}_2$, 162.0678), 135.0494 (52, calc. for $\text{C}_8\text{H}_7\text{O}_2$, 135.0444).

Acknowledgements—We are indebted to the National Programme of Pharmacological Research and Development (Project No. FAR 88-0501) for financial help and to the Ministerio de Educación y Ciencias for a grant to A. E.-B.

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