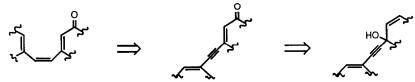
PYRIDINIUM DICHROMATE-INDUCED 1,3-OXIDATIVE REARRANGEMENTS OF ENYNOLS.

Dennis Liotta*, David Brown, William Hoekstra and Robert Monahan III Department of Chemistry, Emory University, Atlanta, Georgia 30322

Summary: The pyridinium dichromate-induced oxidative rearrangements of various 1-ene-4-yne-3ols have been examined. These reactions occur regiospecifically at the olefin site and are only synthetically useful with Z-substituted enynols.

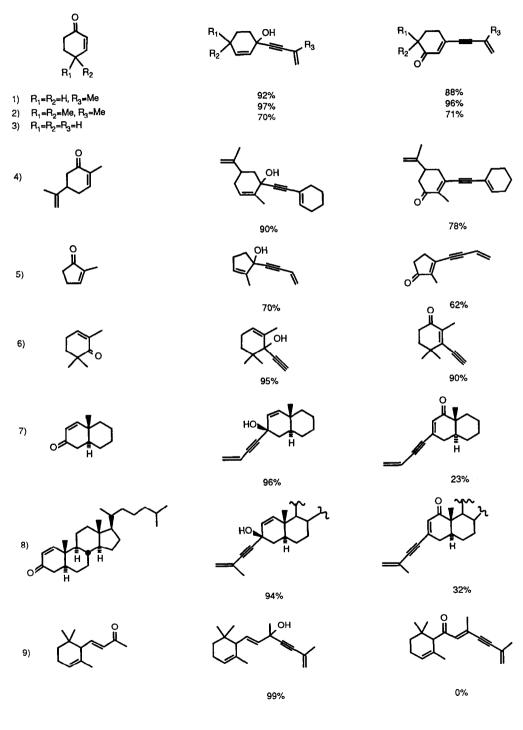
The frameworks of many natural products lend themselves to synthetic strategies which have as their key step a 6π thermal electrocyclization process. However, the viability of this approach depends in large part on the availability of general and reliable methods for the preparation of the required trienes. Retrosynthetically, these trienes can be envisaged as being prepared by reduction of dienynes, which, in turn, might conceivably be prepared by oxidative rearrangement of suitablyfunctionalized tertiary allylic alcohols^{1,2} (vide infra). Implicit to the successful use of the latter process is the notion that the oxidative rearrangement will occur regioselectively at the olefinic site.



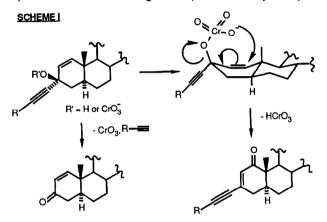
As a means of testing the viability of this approach , we have: (a) prepared a series of enynols and (b) studied their oxidative rearrangement with pyridinium dichromate (PDC).³ Each of the enynols was synthesized by addition of a lithium acetylide to an enone in dry THF at -78°C.⁴ The oxidative rearrangements were carried out using 1.5 equivalents of PDC in methylene chloride at room temperature for times ranging between 16 and 44 hours. The products and yields for each of these sequences are given in Table 1.

For all cases studied these PDC-induced oxidative rearrangements proceeded with complete regiospecificity. In six of the cases (Entries 1-6), the yields of transposed enones were good to excellent. Apparently, this type of rearrangement is capable of tolerating a moderate degree of steric hindrance around either the starting tertiary alcohol (e.g., Entry 6) or the allylic carbon which is ultimately oxygenated (e.g., Entry 2). Since products derived from oxygenation at the acetylenic site were never observed, we conclude that these reactions probably do not involve the intermediacy of any discrete cationic species, but instead proceed through a [3,3]-sigmatropic rearrangement.

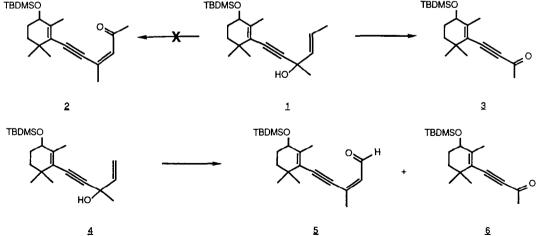
TABLE I



In order to help define the structural limitations of these processes, we have prepared a number of substrates for study. For example, in Entries 7 and 8, the trans A-B ring juncture locks the C-3 hydroxyl group in a pseudo-equatorial orientation, thereby requiring the [3,3]-sigmatropic rearrangment to proceed via a boat-like transition state. Although in each case the rearranged enones were produced in fair yields, significant amounts of the starting enones were also observed. Chromate ester formation and subsequent ring flip to an A-ring boat, necessarily creates a severe 1,4-steric (flagpole) interaction. As a consequence of this, the relative population of the required reactive intermediate should be significantly reduced, thereby allowing processes to compete which are not normally competitive with the rearrangement (such as acteylide ejection, see Scheme 1).



Attempted oxidative rearrangement of the alcohol derived from α -ionone (Entry 9) resulted in a complex mixture of products, implying that the procedure may be limited to cyclic allylic alcohols. This is consistent with Dauben's findings that pyridinium chlorochromate (PCC)-induced oxidative rearrangements of simple tertiary allylic alcohols work best in systems containing Z-double bonds.^{1a,5} To probe this, we examined the oxidative rearrangements of two additional substrates. TBDMSO TBDMSO TBDMSO



Exposure of 1⁶ to PDC gave none of the desired oxidative rearrangement product 2. Instead, ketone 3 was quantitatively produced.⁷ This material appears to be the result of a vinyl hydrogen abstraction, followed by carbon-carbon bond scission. PDC treatment of the unmethylated alcohol 4^{6} resulted in a 3:1 mixture of 5 and 6, suggesting that in situations where both E- and Z-vinyl hydrogens are availabile, the rearrangement and fragmentation processes are competitive.

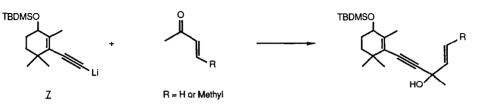
In conclusion, PDC-induced oxidative rearrangements of enynols have been shown to be completely regiospecific and therefore probably involve a [3,3]-signatropic rearrangement. These processes are synthetically useful with Z-substituted envnols. However, with E-substituted envnols. oxidative rearrangement competes with a variety of carbon-carbon bond scission processes.

Acknowledgement: Financial support by the National Institutes of Health is gratefully acknowledged.

References:

- 1. (a) Dauben, W. G.; Michno, D. M. J. Org. Chem. 1977, 42, 682. (b) Sundararaman, P.; Herz, W. J. Org. Chem. 1977, 42, 813.
- 2. Oxidative rearrangements of various sorts have also been utilized to effect 1.3 enone transpositions. For a brief survey of these, see:
 - (a) Trost, B. M.; Shanton, J. L. J. Am. Chem. Soc. 1975, 97, 4018.
 - (b) Trost, B. M.; Hiroi, K.; Holy, N. J. Am. Chem. Soc. 1975, 97, 5873.
 - (c) Sharpless, K. B.; Lauer, R. F. J. Am. Chem. Soc. 1972, 94, 7154...
 (d) Evans, D. A.; Andrews, G. C. Acc. Chem. Res. 1974, 6, 147.

 - (e) Moriwaki, M.; Sawada, S; Inoyue, Y. Chem Commun. 1970, 419.
 - (f) Wharton, P. S.; Bohler, D. H. J. Org. Chem. 1961, 26, 3615.
 - (g) Liotta, D.; Zima, G. J. Org. Chem. 1980, 45. 3884.
 - (h) Liotta, D.; Zima, G.; Saindane, M. J. Org. Chem. 1982, 47, 1258.
 - (i) Stork, G.; Danheiser, R. L. J. Org. Chem. 1973, 38, 1775.
- (a) Corey, E. J.; Schmidt, G. *Tetrahedron Lett.* **1979**, 399;
 (b) Corey, E. J.; Schmidt, G. *Tetrahedron Lett.* **1980**, 731.
- 4. Liotta, D.; Saindane; M.; Sunay, U.; Jamison, W. C. L.; Grossman, J.; Phillips, P. J. Org. Chem. 1982, 47, 1258.
- 5. For studies involving the use of PCC as an oxidant, see:
- (a) Corey, E. J.; Suggs, J., W. Tetrahedron Lett. 1975, 2644.;
- (b) Corey, E. J.; Suggs, J., W. J. Org. Chem. 1975, 40, 2554.
- 6. Compounds 1 and 4 were prepared by addition of lithium acetylide 7 to propenvl methyl ketone and methyl vinyl ketone, respectively (vide infra).



7. The structure of ketone 3 was unequivocally established by independent synthesis.

(Received in USA 26 December 1986)